A Chiral Auxiliary-Based Synthesis of the C5-C17 *trans*-Decalin Framework of Anthracimycin

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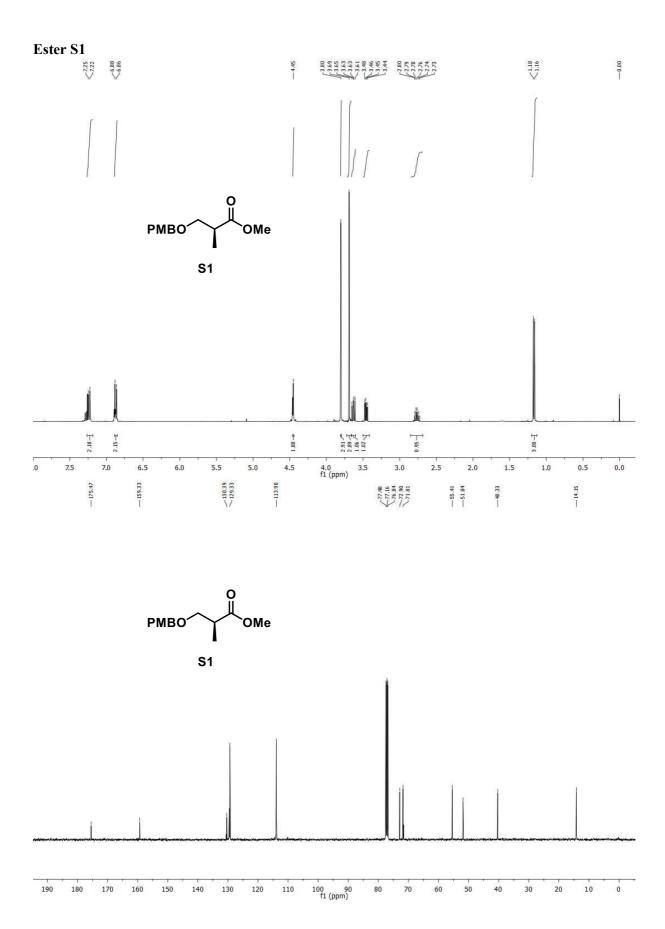
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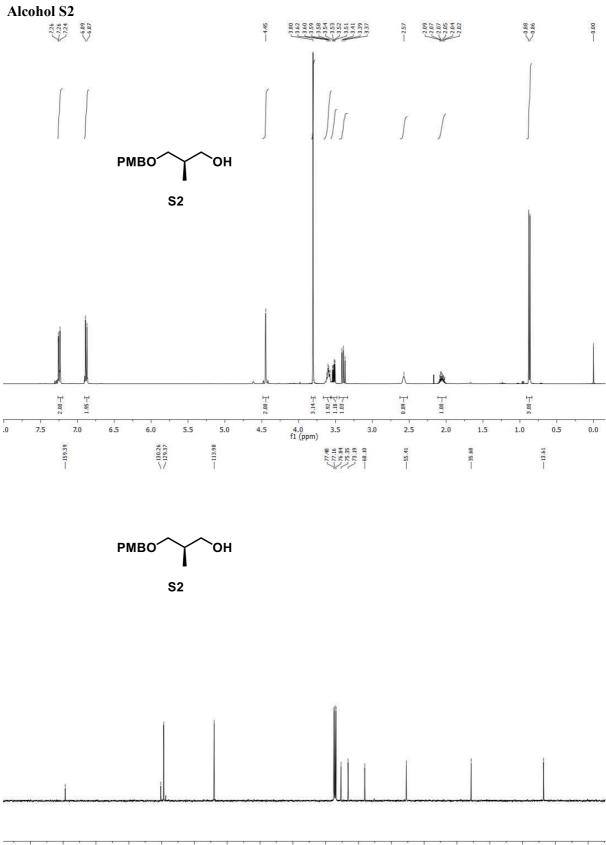
Supporting Information

- S2 General Information
- S3 ¹H and ¹³C NMR Spectra
- S21 NOESY Spectra for 17 + 18
- S22 *X*-ray structure for **22**

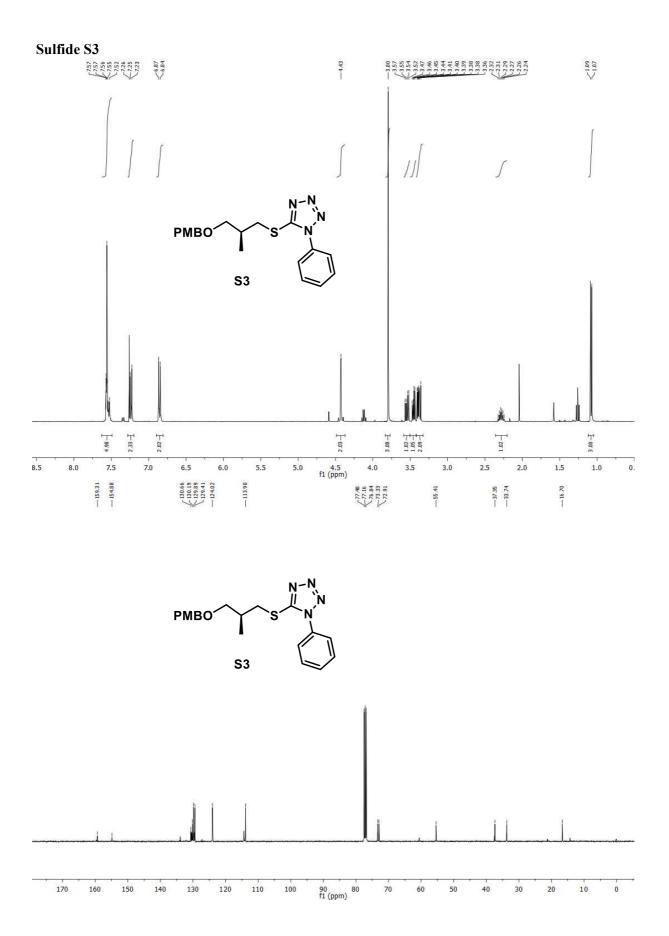
General Information

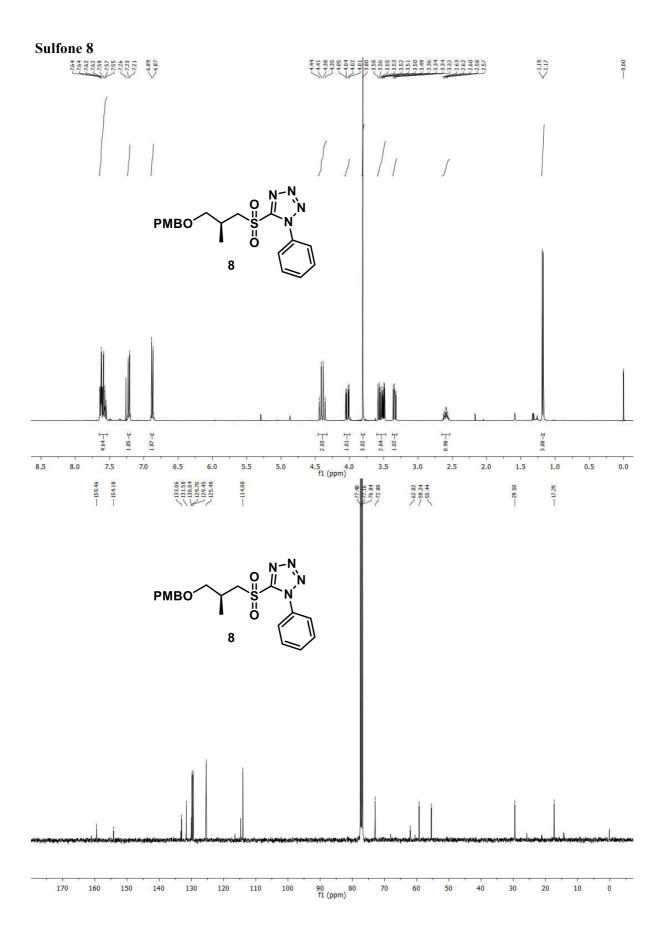
All reactions were carried out in flame- or oven-dried glassware under a dry nitrogen atmosphere unless otherwise stated. CH₂Cl₂, (CH₃)₂CO, DMF, THF CH₃CN and Et₂O were obtained from a solvent purifier from LC Technology Systems. Triethylamine was distilled from calcium hydride. All other reagents were used as received unless otherwise noted. Yields refer to chromatographically and spectroscopically (¹H NMR) homogenous materials, unless otherwise stated. Reactions performed at low temperature were either cooled with an acetone/dry ice bath to reach -78 °C, or acetone/brine/dry ice bath to reach -30 °C, or a water ice bath to reach 0 °C. Flash chromatography was carried out using 0.063-0.1 mm silica gel with the required solvent system. TLC was carried out using 0.2 mm Kieselgel F254 (Merck) silica plates and compounds were visualised using UV irradiation at 365 nM and/or staining with vanillin in methanolic sulphuric acid or potassium permanganate and potassium carbonate in aqueous sodium hydroxide. Melting points were measured with a Kofler hot-stage apparatus and are uncorrected. NMR spectra were recorded as indicated on the Bruker DRX-400 spectrometer operating at 400 MHz for ¹H nuclei, 100 MHz for ¹³C nuclei and 282 MHz for ¹⁹F nuclei. All chemical shifts are reported in ppm relative to tetramethylsilane ($\delta = 0$ for ¹H NMR) and CDCl₃ ($\delta = 77.0$ for ¹³C NMR). ¹H NMR data is reported as chemical shift, relative integral, multiplicity (s, singlet; d, doublet; t, triplet; q, quartet; dd, doublet of doublets; m, multiplet), coupling constant (J, Hz), and assignment, Assignments were made with the aid of COSY, NOESY and HSQC experiments where required. High resolution mass spectra were recorded using a VG-70SE spectrometer at a nominal accelerating voltage of 70 eV or on a Bruker micrOTOF-Q II mass spectrometer.





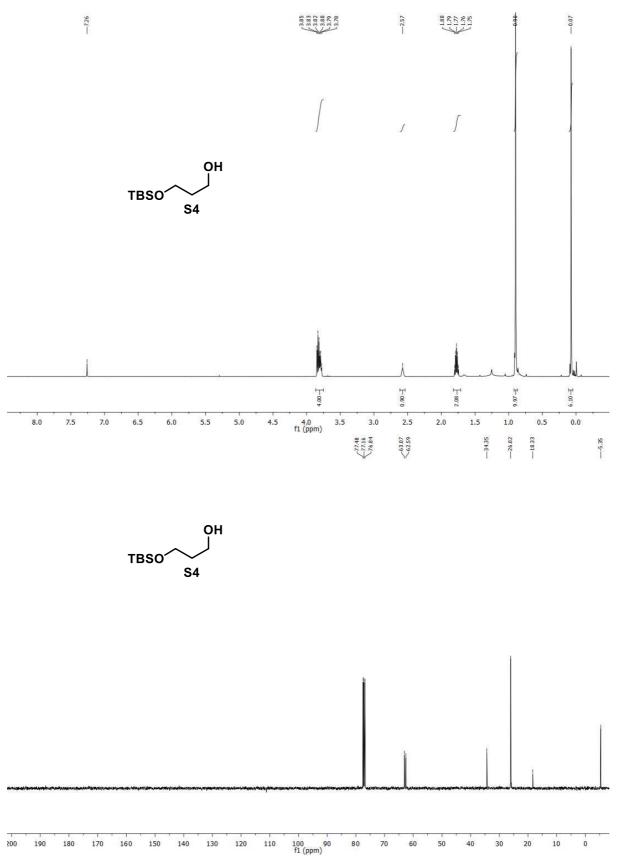
90 80 f1 (ppm)



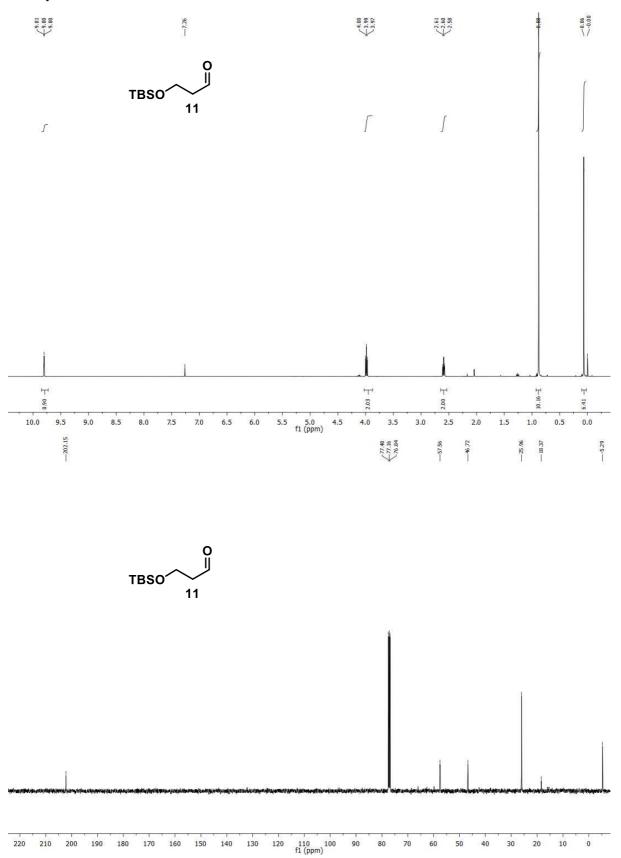


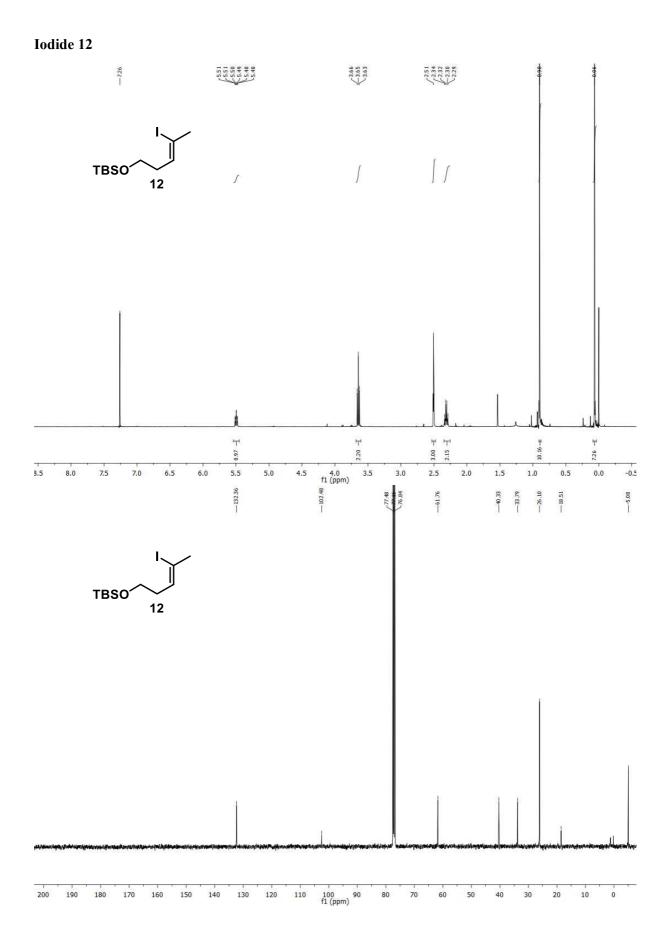
S6



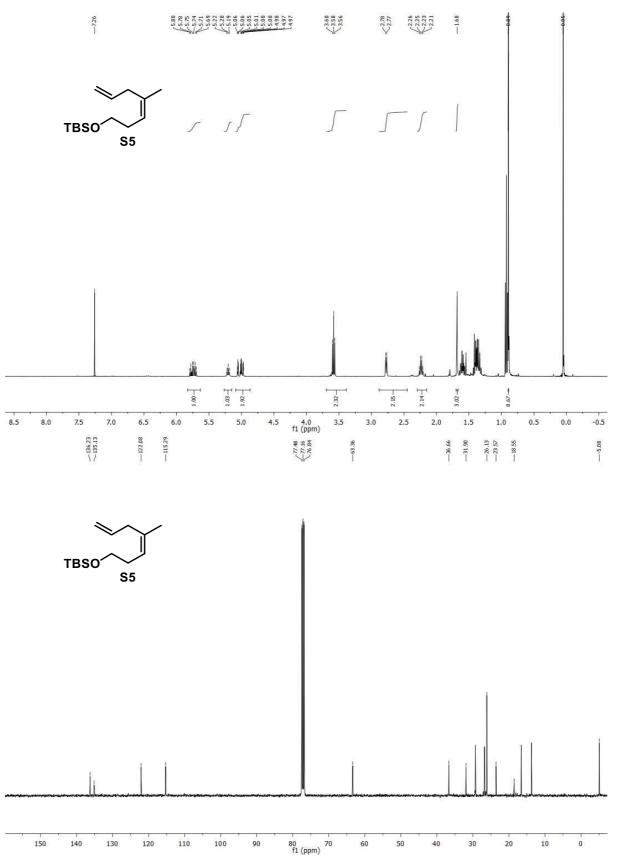


Aldehyde 11





Alkene S5

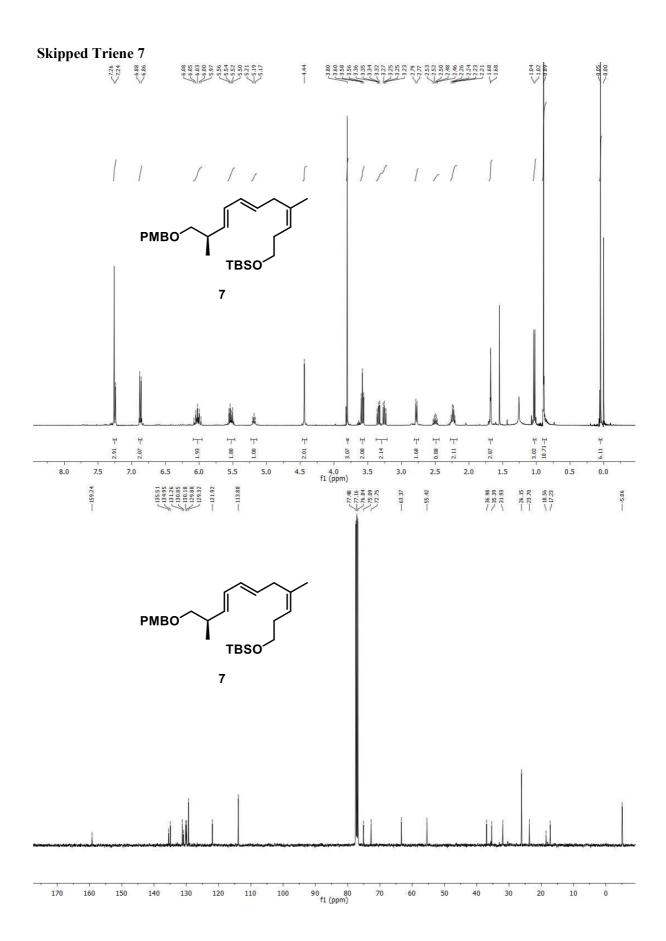


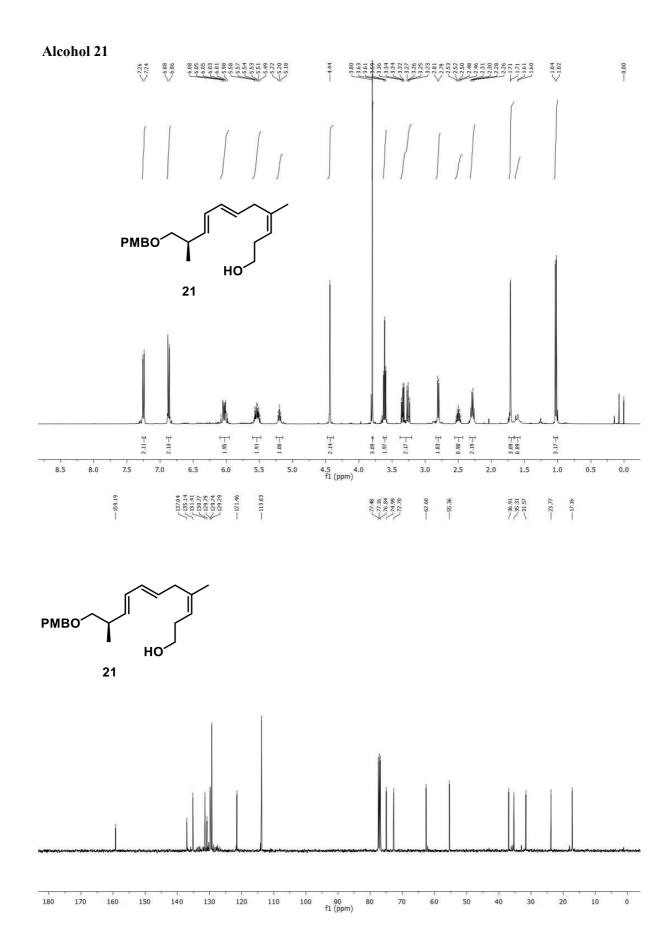
Aldehyde 9 <9.54 <9.52 362 358 358 306 306 224 223 221 219 -172 0.89 0.05 $\int \int \int$ ſ ſ ſ I 0 TBSO 9 F 56'0 1.00-1 1.04 2.12 J 1.00-I 2.10 I 2.14J 9.42 -* 3.05 H I-61.9 3.0 0.5 10.0 3.5 0.0 9.5 8.5 8.0 7.5 4.0 2.5 2.0 1.0 1.5 0.5 9.0 77.48 77.16 76.84 -62.96 0 TBSO 9

-0.

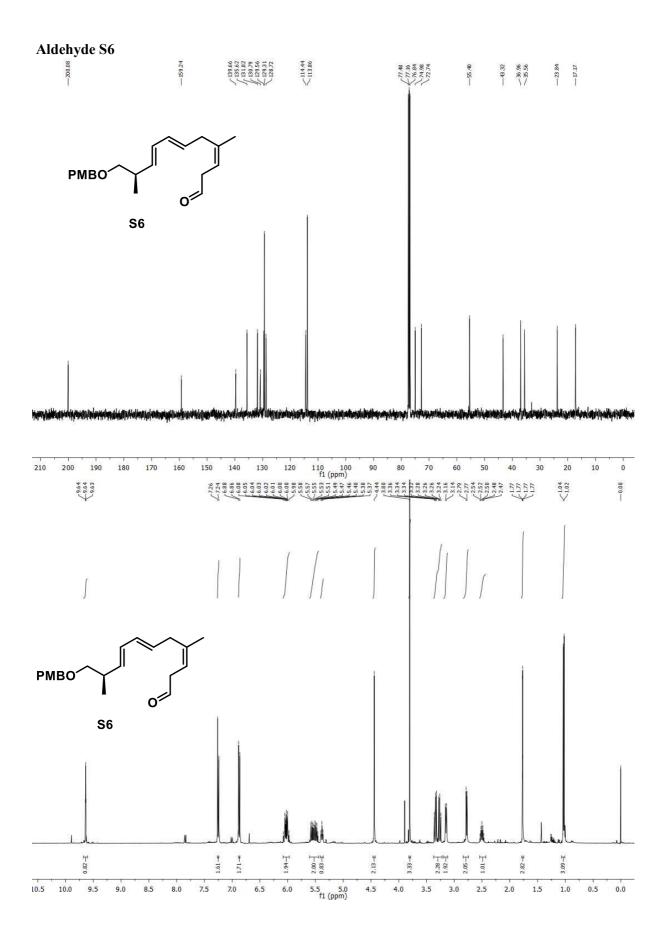
---5.10

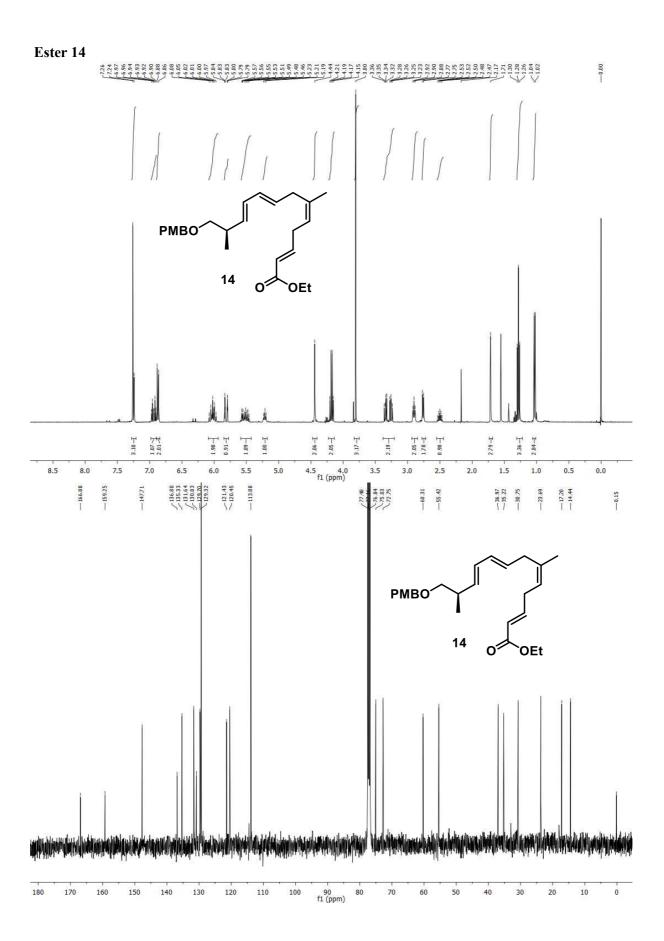
220 210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 f1 (ppm)

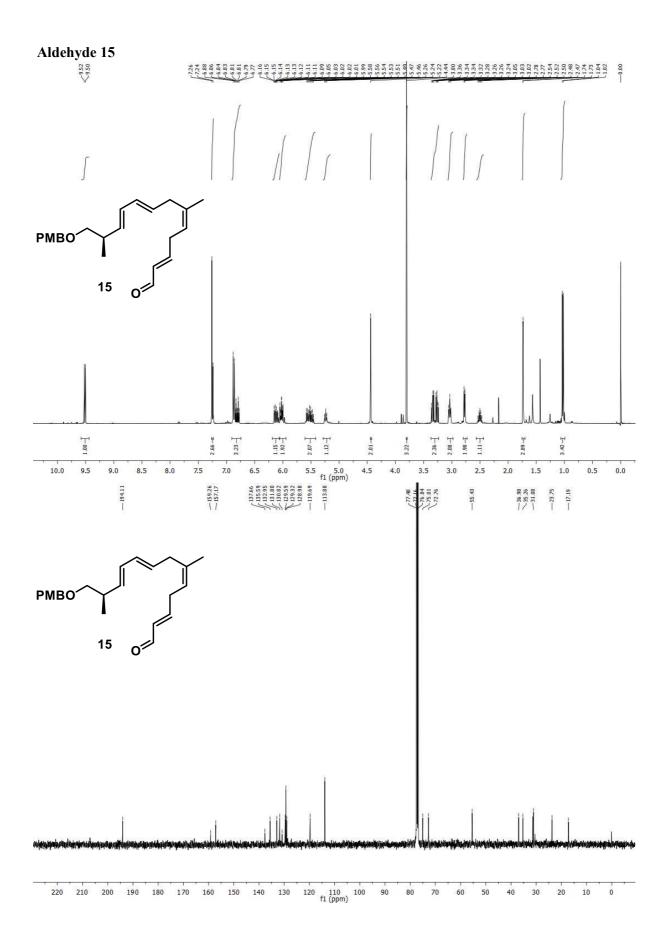


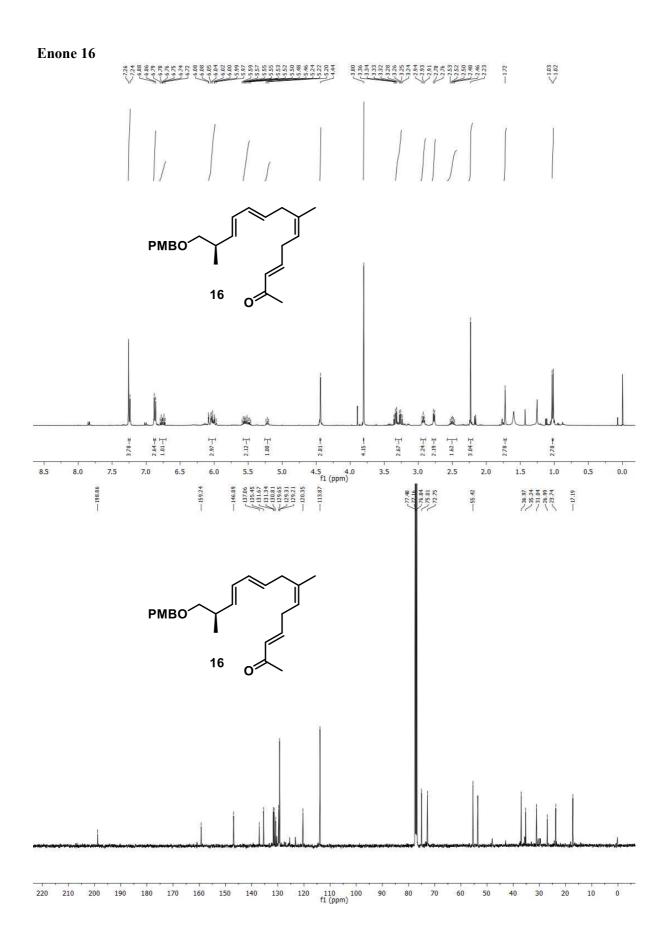


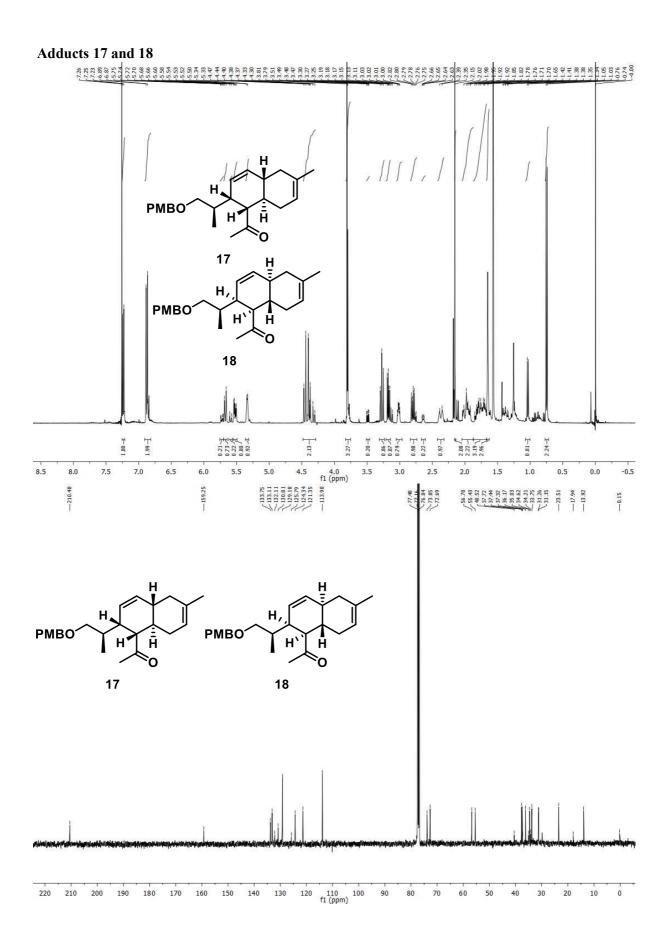
S13



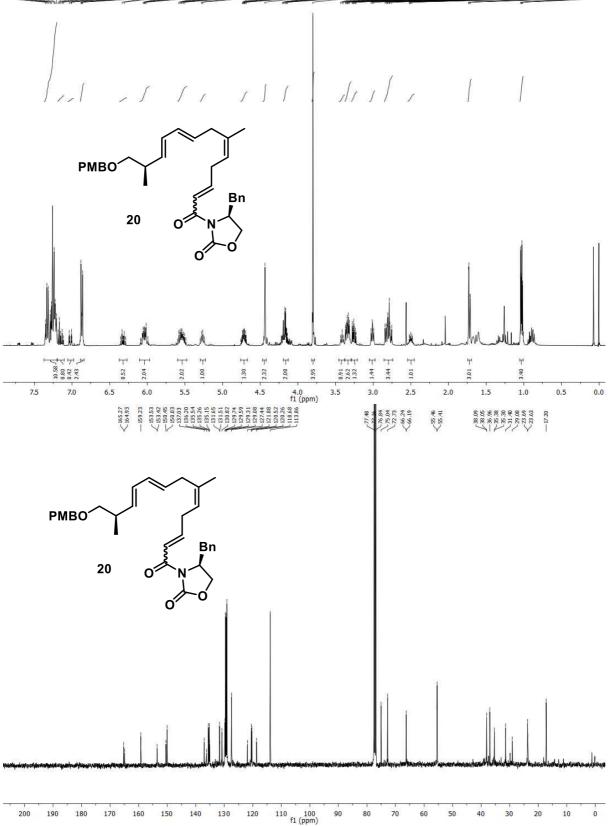




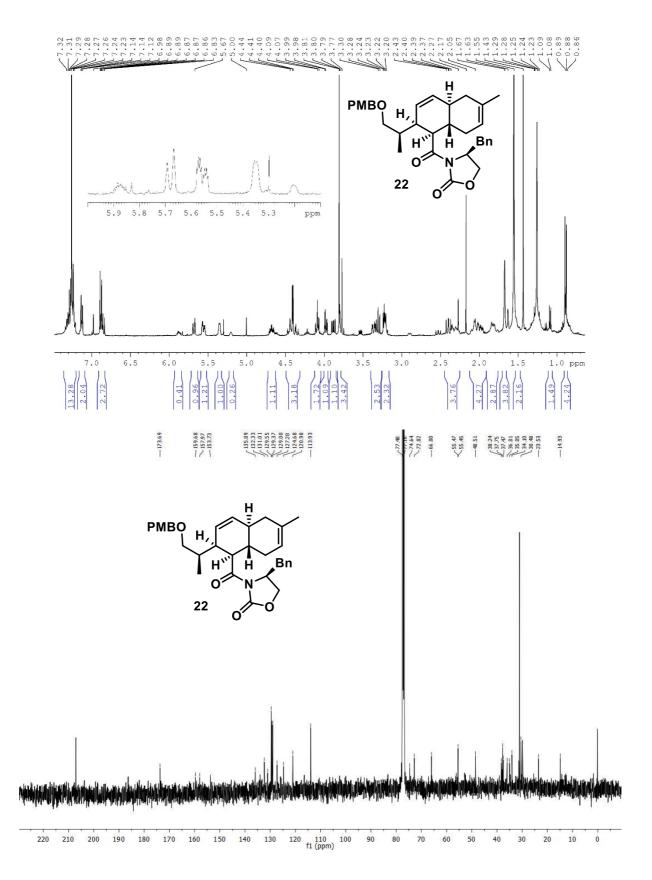




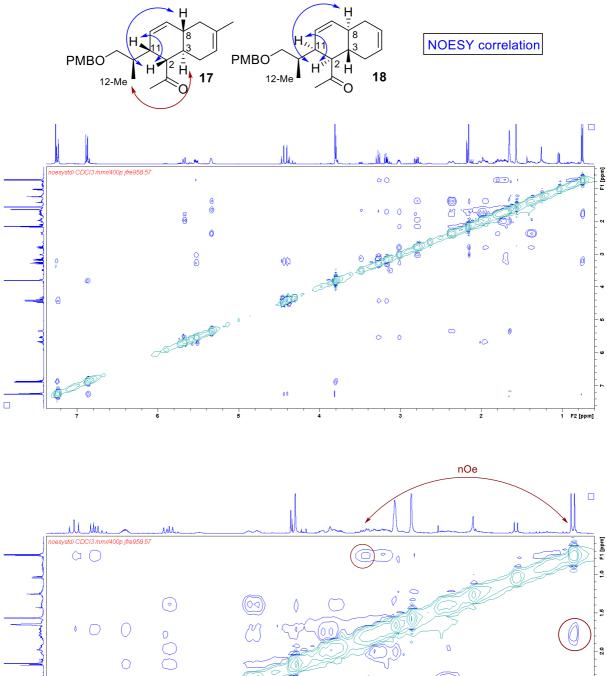
Imide 20



Adduct 22



Adducts 17-18





Crystallization. Single crystals of Adduct **22** were obtained by slow recrystallization of a solution of the compound in ethyl acetate.

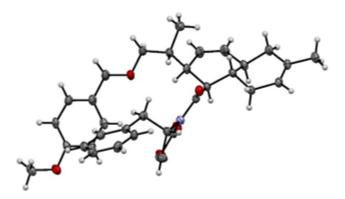


Figure S1. ORTEP diagram drawn with 50% ellipsoid probability of the crystal structure of Adduct **22**.

Table S1. Crystal data and structure refinement details for *N*-Ts bicyclic ketone 20.

Empirical formula	C ₃₃ H ₃₈ NO ₅
Formula weight	528.64
Temperature (K)	100
Wavelength (A)	1.54184
Crystal system	Triclinic
Space group	P 1
a (Å)	9.1302(3)
b (Å)	9.1578(5)
c (Å)	10.2260(4)
α (°)	108.986(4)
β (°)	98.417(3)
γ (°)	114.037
V (Å ³)	698.83(6)
Z	1
$D_c (Mg/m^3)$	1.256
F(000)	283
μ (mm ⁻¹)	0.670
θ_{max} (°)	68.241
Total reflections	16713
Unique reflections	4896
Reflections $[I > 2\sigma(I)]$	0.0595
Parameters	356
Rint	0.0595
Goodness-of-fit	1.054
$R[F_2 > 2\sigma(F_2)]$	0.0448
wR (F2, all data)	0.1169