

Supporting Information

One-Pot Synthesis of Pyrrolo[3,2,1-*de*]phenanthridines from 7-Phenylindoles via Tandem C-H Olefination/Aza-Michael addition

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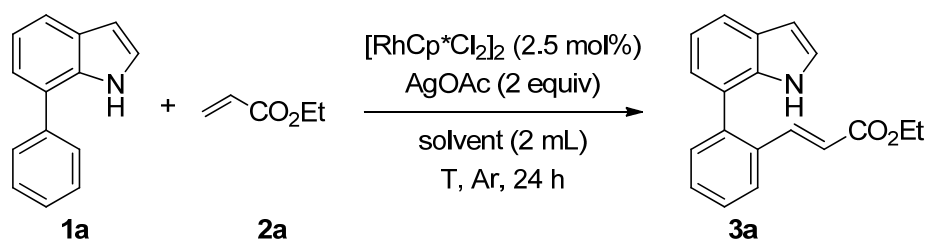
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1. Optimization of the C-H olefination of 7-phenyl-1*H*-indole **1a** with ethyl acrylate **2a**

Table S1 Optimization results of the C-H olefination of 7-phenyl-1*H*-indole **1a** with ethyl acrylate **2a**^a



entry	solvent	T(°C)	Yield ^b (%)
1	EtOAc	100°C	35
2	acetone	100°C	84
3	DCM	100°C	40
4	DCE	100°C	48
5	1,4-dioxane	100°C	75
6	DMSO	100°C	23
7	acetone	60°C	73
8	acetone	80°C	100
9	acetone	120°C	70
10	acetone	140°C	76
11 ^c	acetone	80°C	99(97)

[a] Conditions: **1a** (0.3 mmol), **2a** (0.75 mmol), [Cp*RhCl₂]₂ (2.5 mol%), AgOAc (2 equiv), solvent (2 mL), T °C, Ar, 24 h.

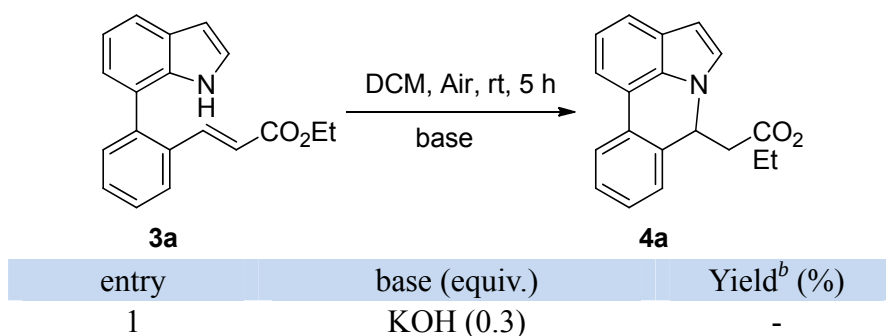
[b] ¹H NMR Yield on the basis of the amount of **1a** used, Number in parenthesis is isolated yield.

[c] Air atmosphere

2. Optimization of the intramolecular aza-Michael addition of (*E*)-ethyl 3-(2-(1*H*-indol-7-yl)phenyl)acrylate **3a**

3-(2-(1*H*-indol-7-yl)phenyl)acrylate **3a**

Table S2 Optimization results of the intramolecular aza-Michael addition^a



2	KO ^t Bu (0.3)	100
3	DMAP (0.3)	-
4	Et ₃ N (0.3)	-
5	Piperidine (0.3)	-
6	2,3-Dimethylquinoline (0.3)	-
7	ⁿ Bu ₄ NOAc (10)	-
8	Me ₄ NOAc (10)	-
9 ^c	Me ₄ NOAc (10)	-
10	ⁿ Bu ₄ NBF ₄ (10)	-
11	KPF ₆ (10)	-
12	KOAc (10)	-
13	NaOAc (10)	-
14	CsOAc (10)	-
15	LiOAc (10)	-

[a] Conditions: **3a** (0.2 mmol), base, DCM (1.5 mL), Air, 5 h.

[b] Isolate yield.

[c] 1.5 mL MeCN

3. Single Crystal X-ray Structure Determination. Data collection and structural analysis of crystal were collected on an Agilent Technologies SuperNova Single Crystal Diffractometer equipped with graphite monochromatic Cu K α radiation (λ = 1.54184 Å). The crystal was kept at 293 (10) K during data collection. Using Olex2^[1], the structure was solved with the Superflip structure solution program using Charge Flipping and refined with the ShelXL refinement package using Least Squares minimisation. The hydrogen atoms on the ligands were placed in idealized positions and refined using a riding model. The detailed crystallographic data and structure refinement parameters for these compounds are summarized in **Table S1-S2**. CCDC 1943455 and 1943456 contain the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.

(1) Single crystal structure of 3a (CCDC 1943455)

Table S1. Crystal data and structure refinement for **3i**

Identification code	Y-117
Empirical formula	C ₁₉ H ₁₆ FNO ₂
Formula weight	309.33
Temperature/K	294.50(10)
Crystal system	orthorhombic
Space group	Fdd2

a/Å	9.0639(3)
b/Å	37.0554(13)
c/Å	19.0348(9)
$\alpha/^\circ$	90
$\beta/^\circ$	90
$\gamma/^\circ$	90
Volume/Å ³	6393.2(4)
Z	16
$\rho_{\text{calc}}/\text{g/cm}^3$	1.286
μ/mm^{-1}	0.750
F(000)	2592.0
Radiation	CuK α ($\lambda = 1.54184$)
2 θ range for data collection/ $^\circ$	9.548 to 145.422
Index ranges	$-8 \leq h \leq 11$, $-45 \leq k \leq 34$, $-23 \leq l \leq 20$
Reflections collected	8640
Independent reflections	2807 [$R_{\text{int}} = 0.0226$, $R_{\text{sigma}} = 0.0182$]
Data/restraints/parameters	2807/1/209
Goodness-of-fit on F^2	1.068
Final R indexes [$I \geq 2\sigma(I)$]	$R_1 = 0.0482$, $wR_2 = 0.1273$
Final R indexes [all data]	$R_1 = 0.0540$, $wR_2 = 0.1336$
Largest diff. peak/hole / e Å ⁻³	0.23/-0.17

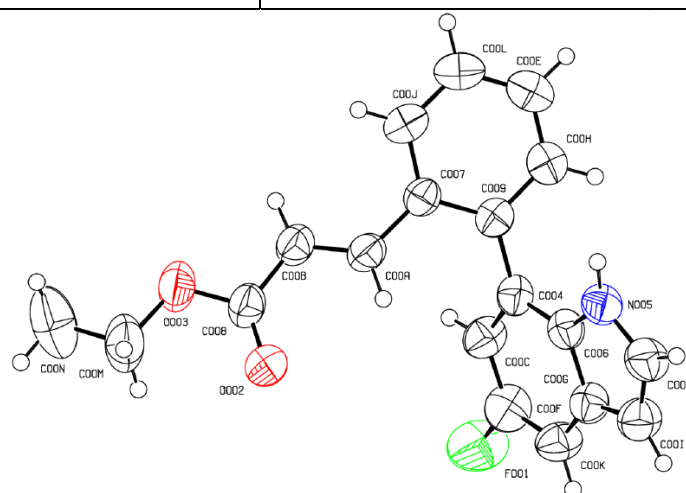


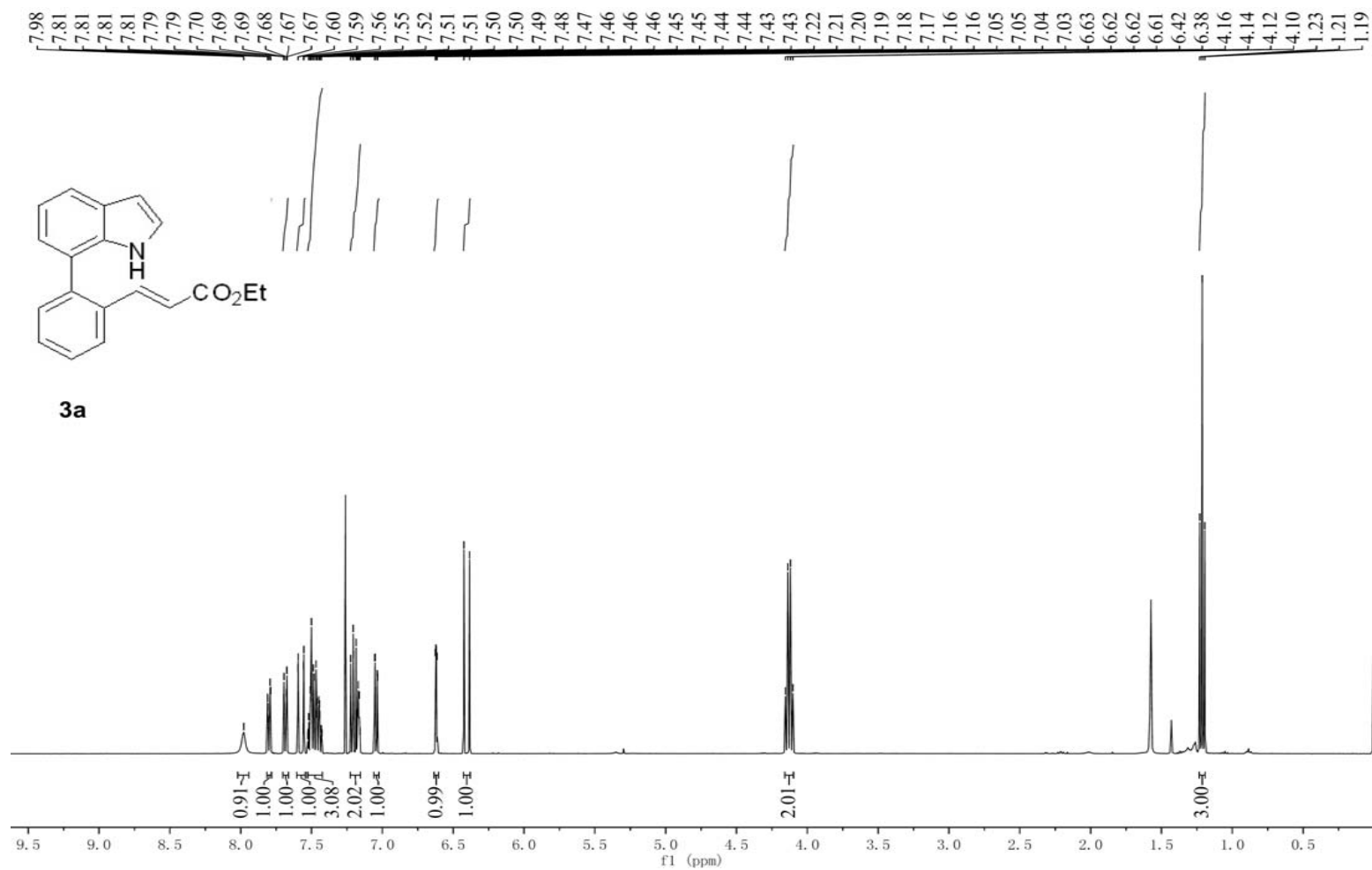
Figure S1 X-Ray structure of **3i** with 50% probability ellipsoids

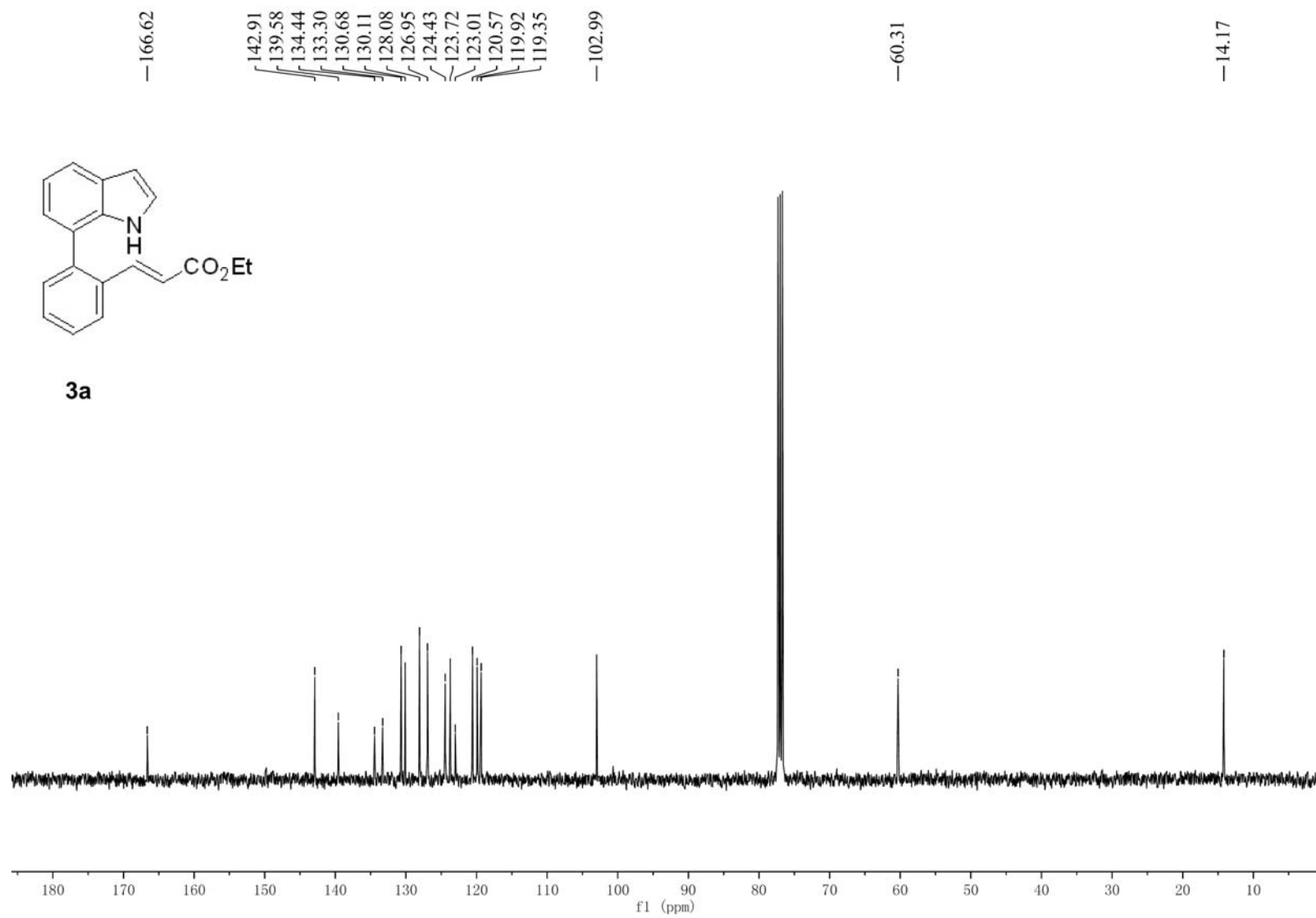
(2) Single crystal structure of **3m** (CCDC 1943456)

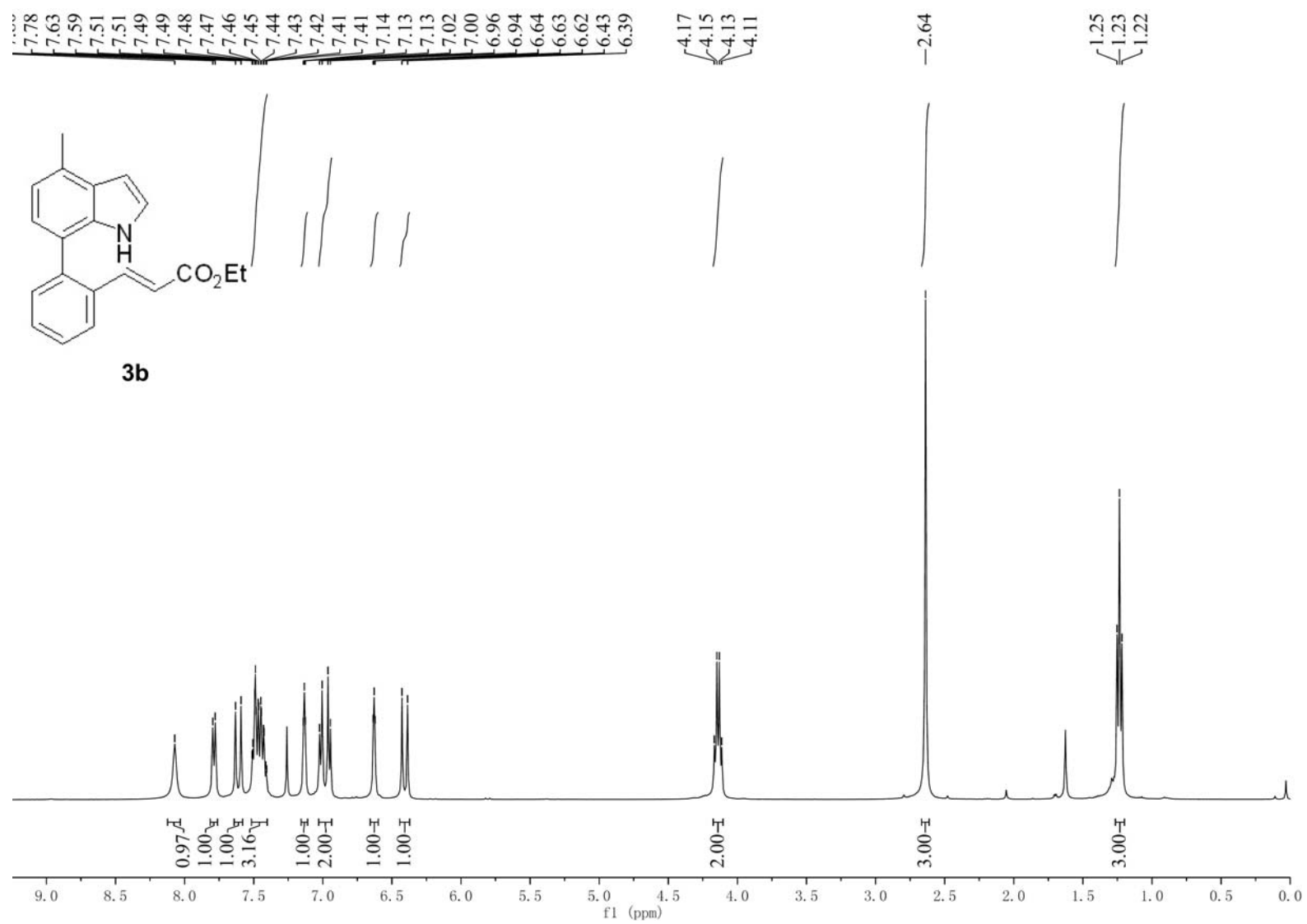
Table S2. Crystal data and structure refinement for **3m**

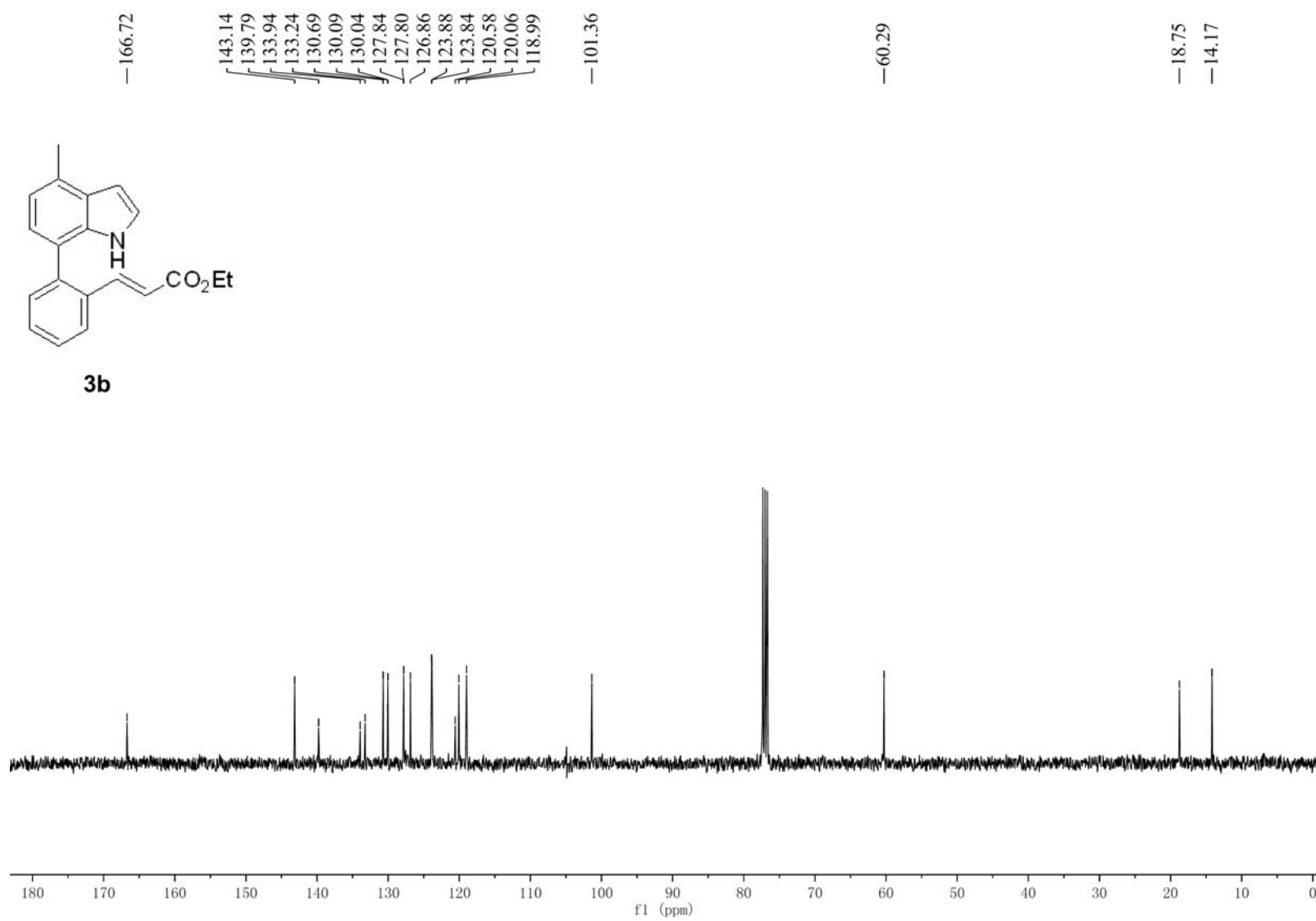
Identification code	Y-I_164
Empirical formula	2(C ₂₁ H ₁₉ NO ₄)
Formula weight	698.74
Temperature/K	295.42(10)
Crystal system	monoclinic

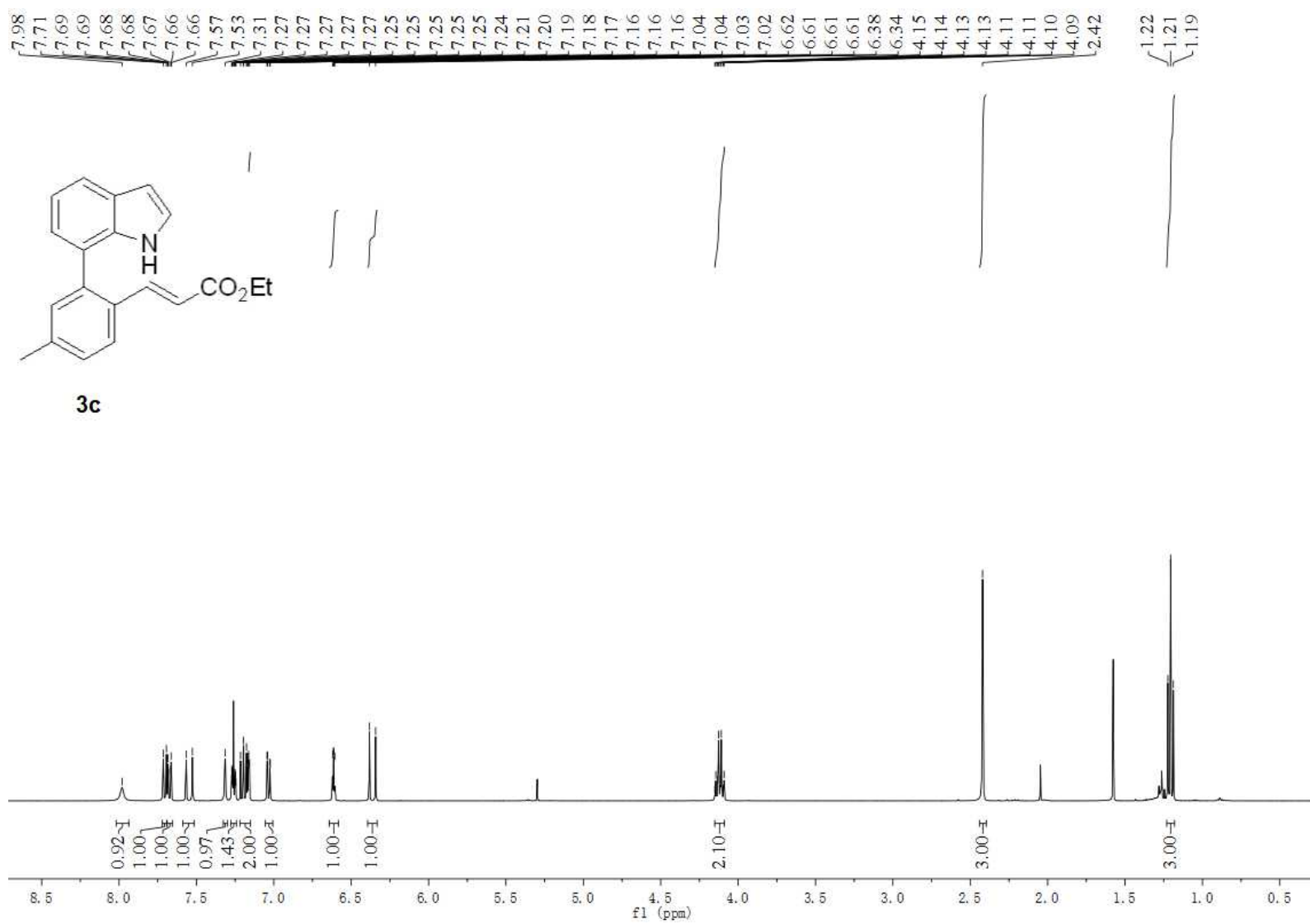
4. Copies of ^1H and ^{13}C -NMR charts of products

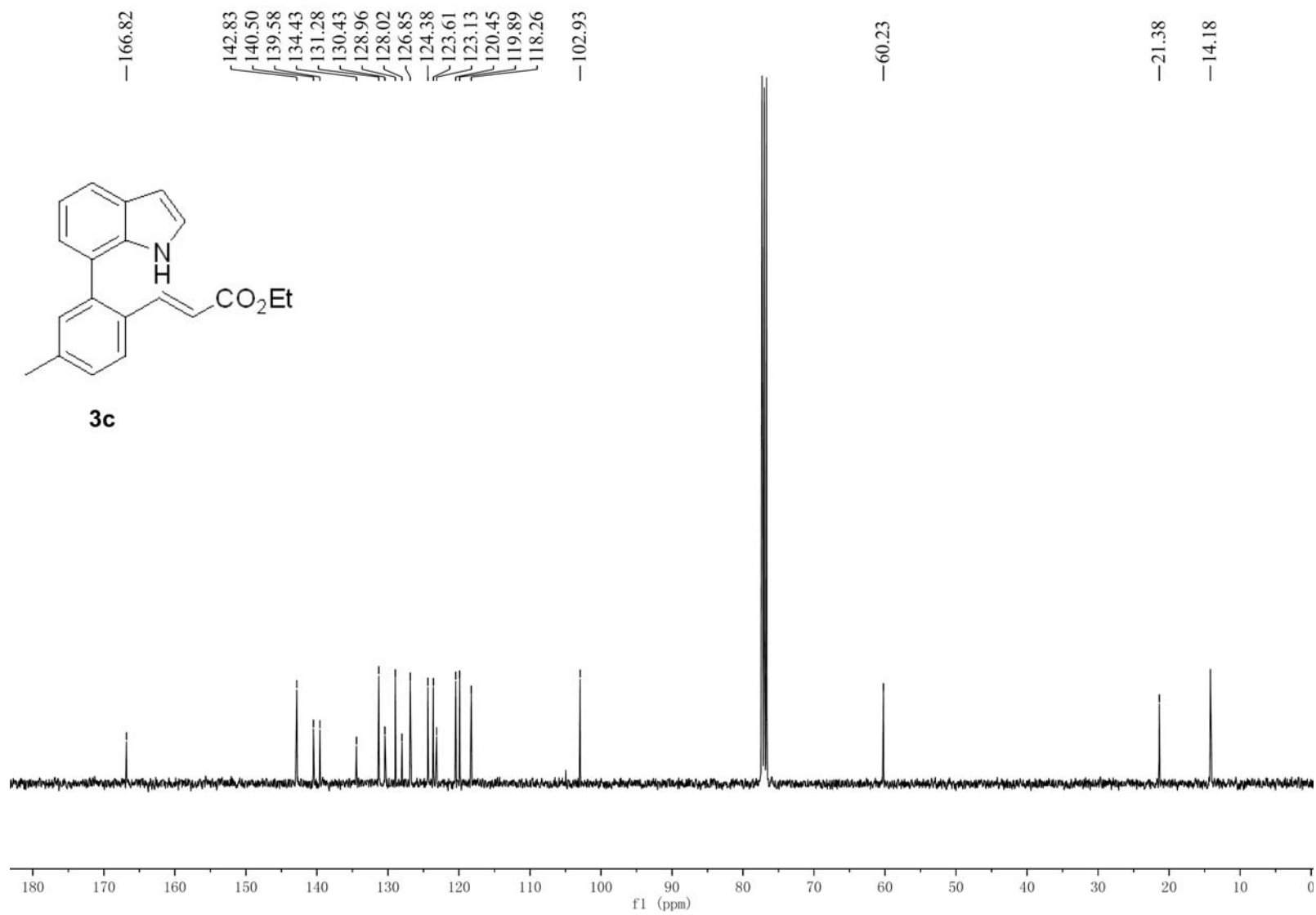


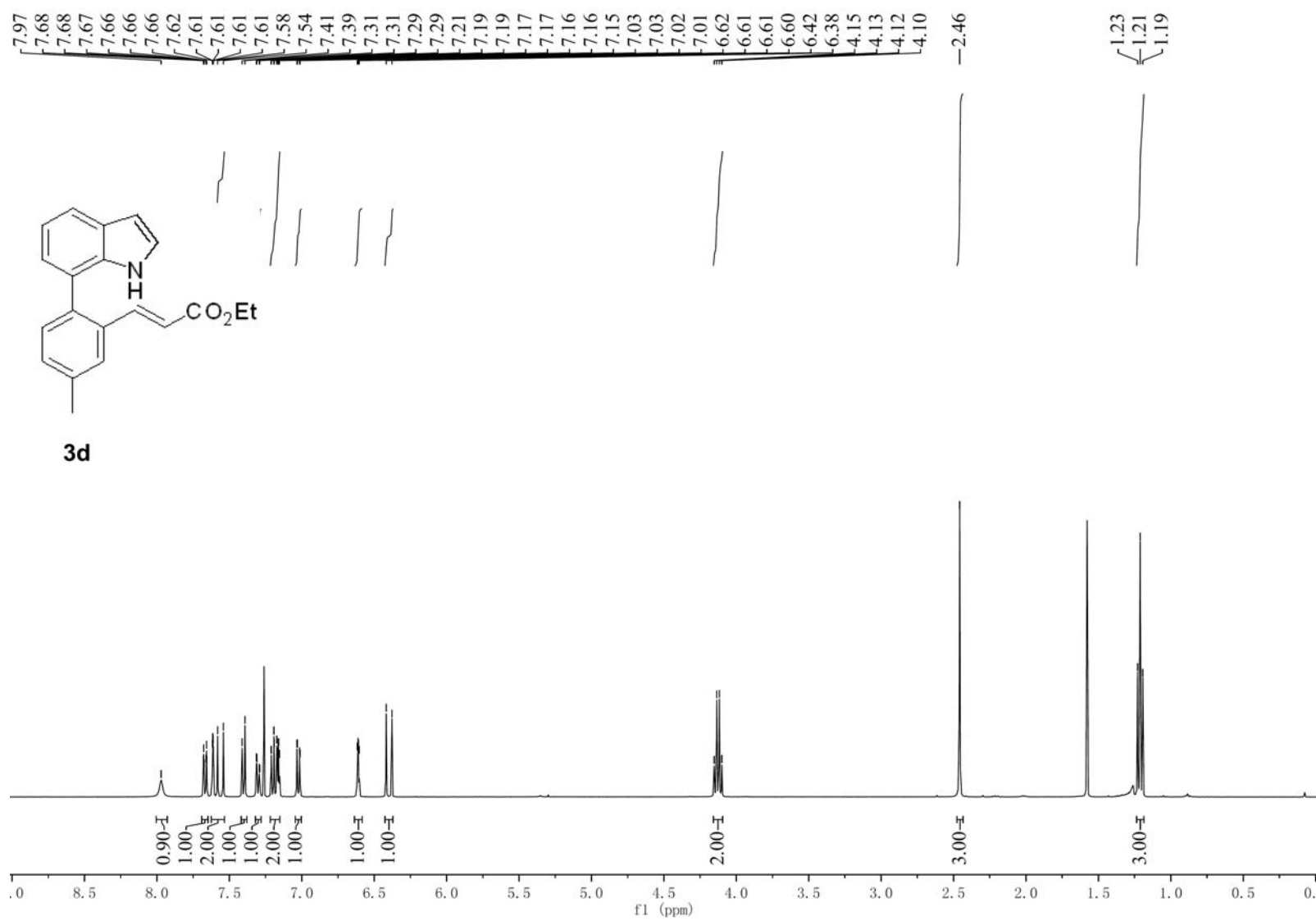


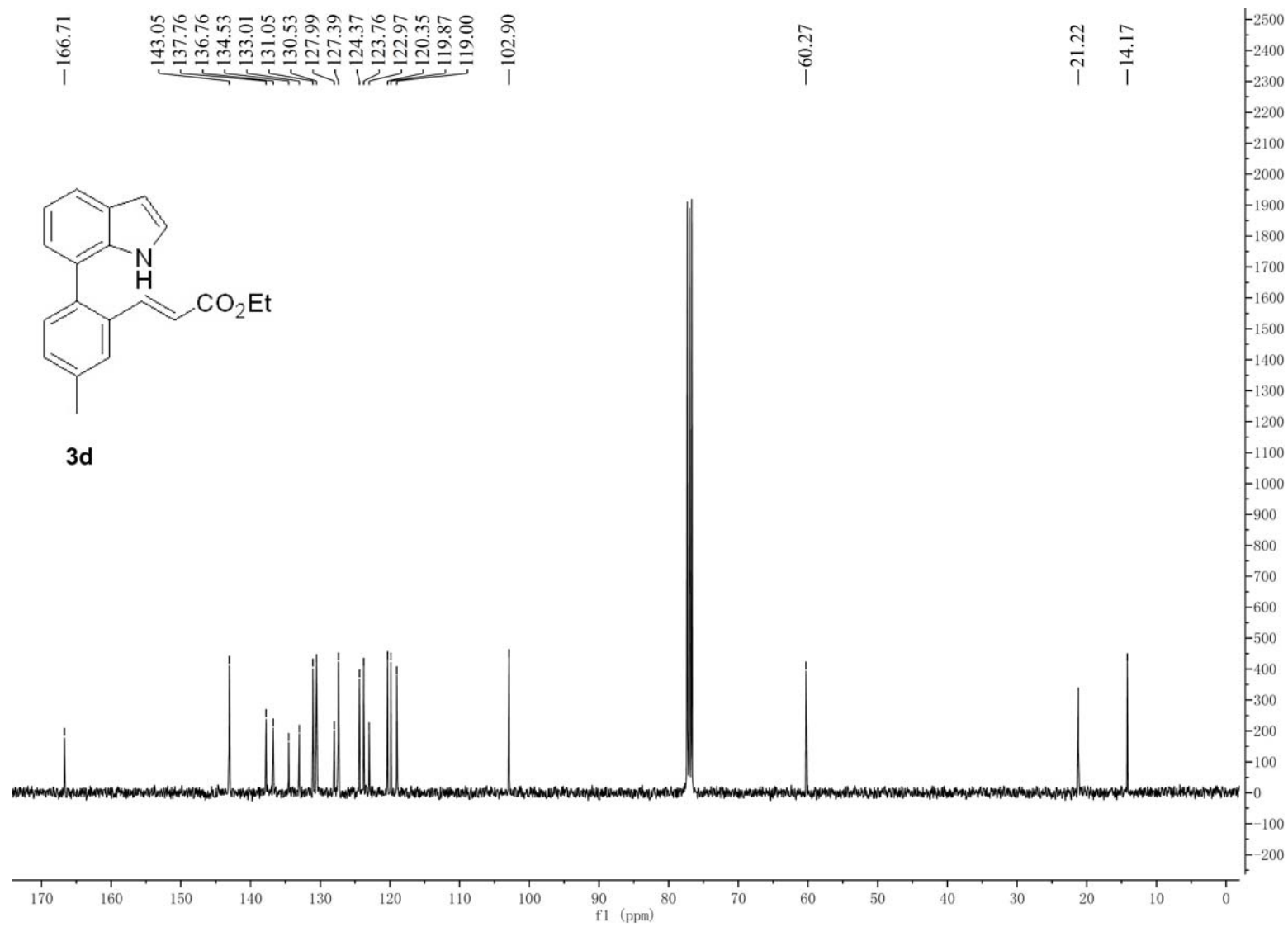


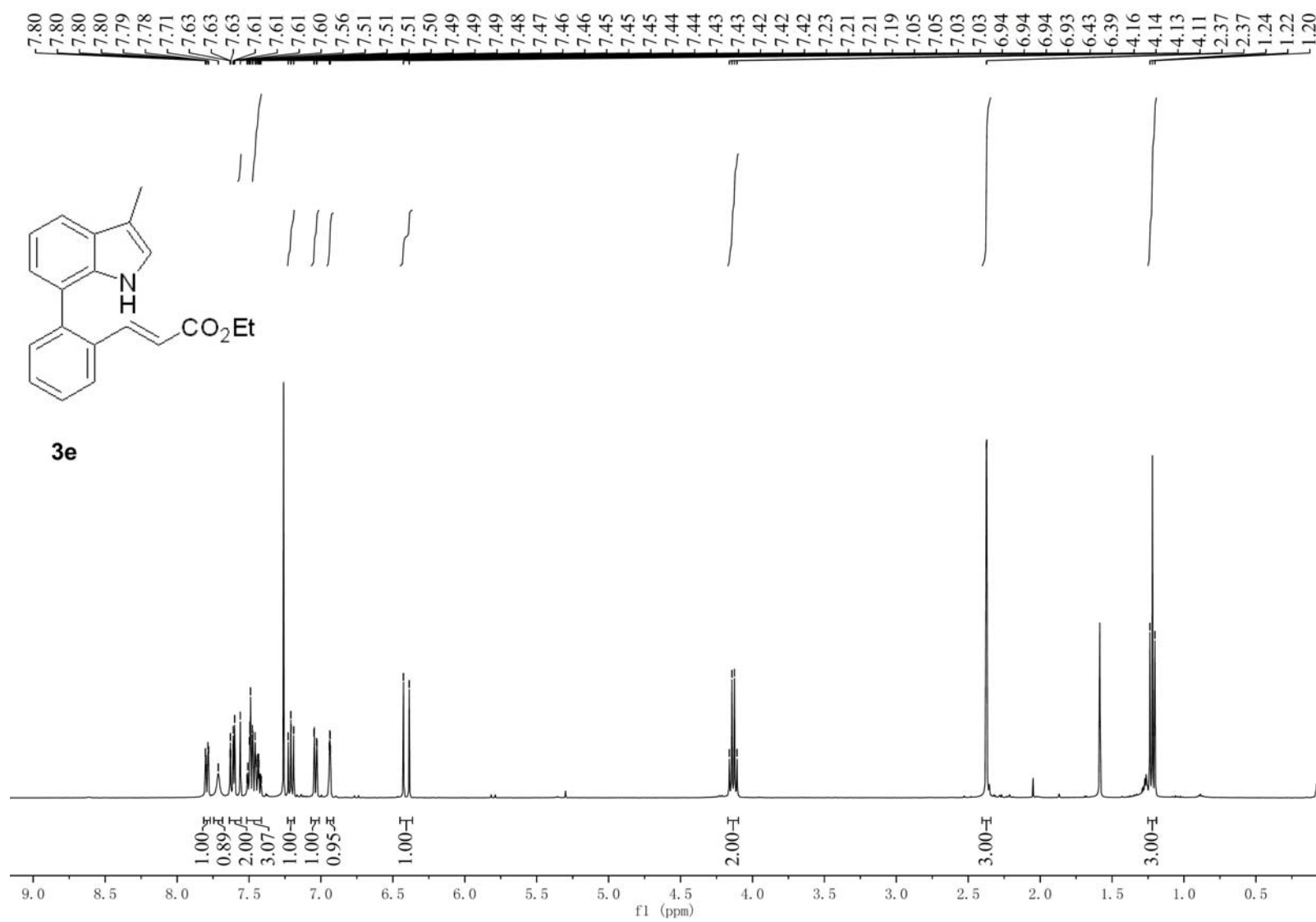


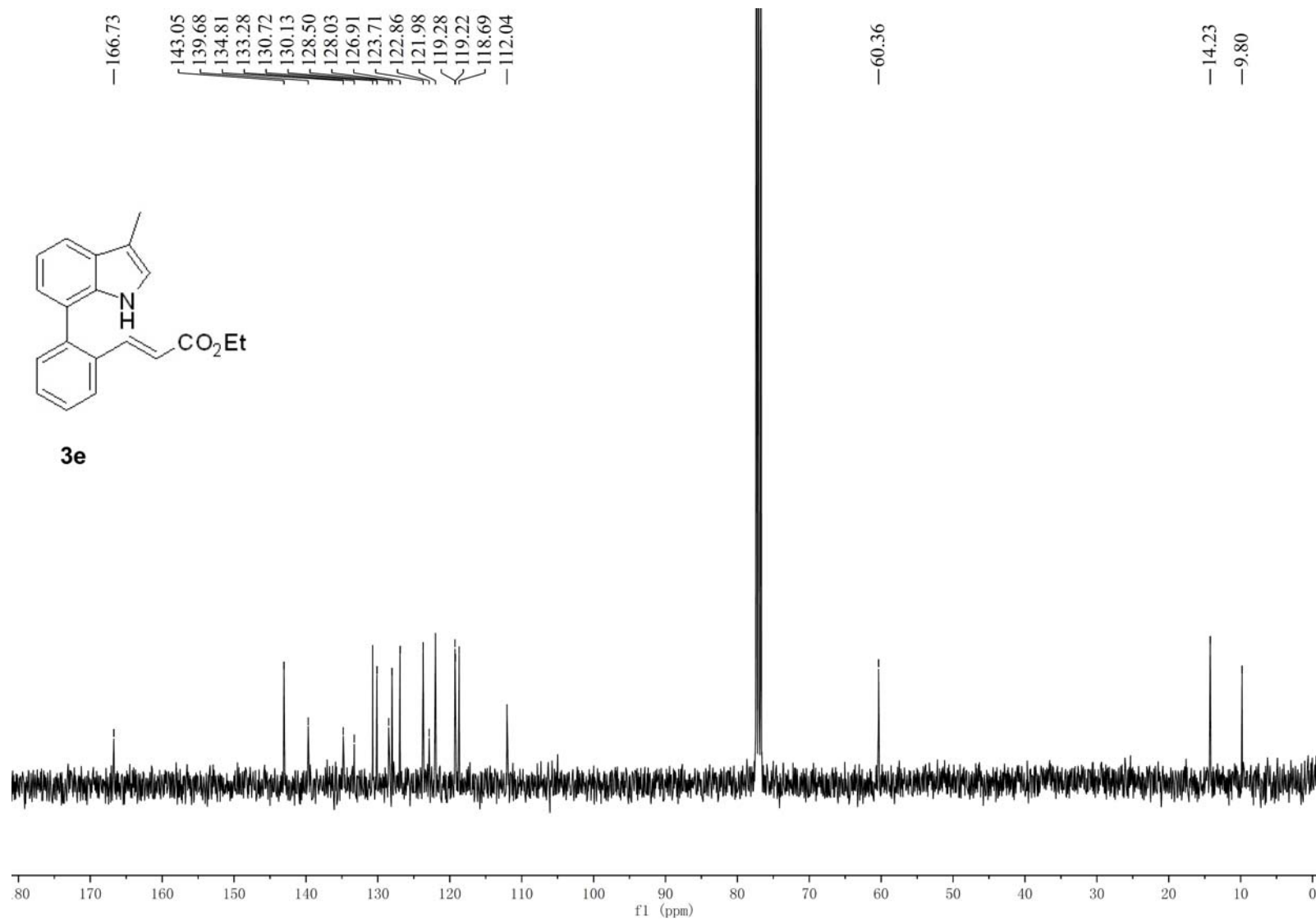


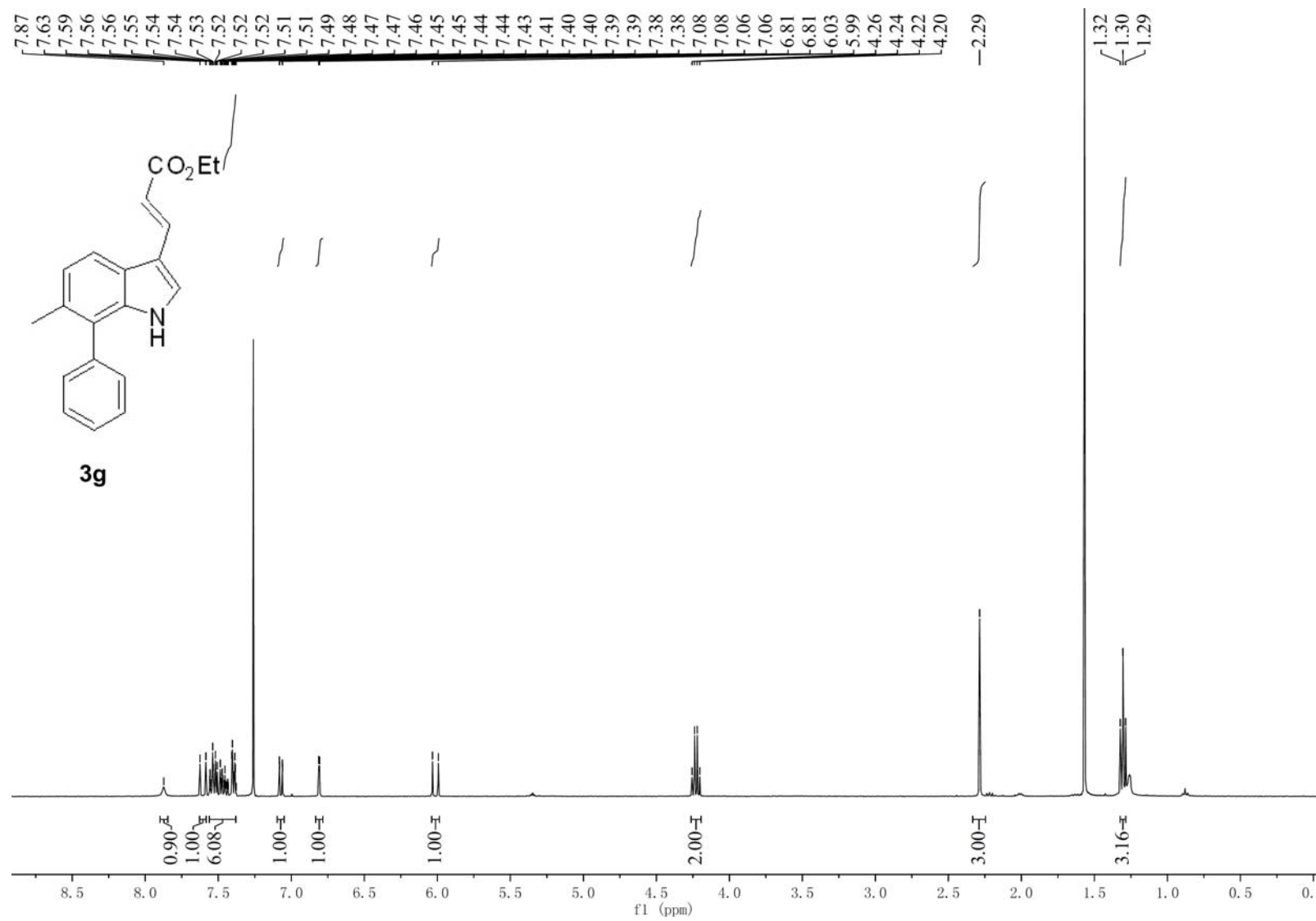


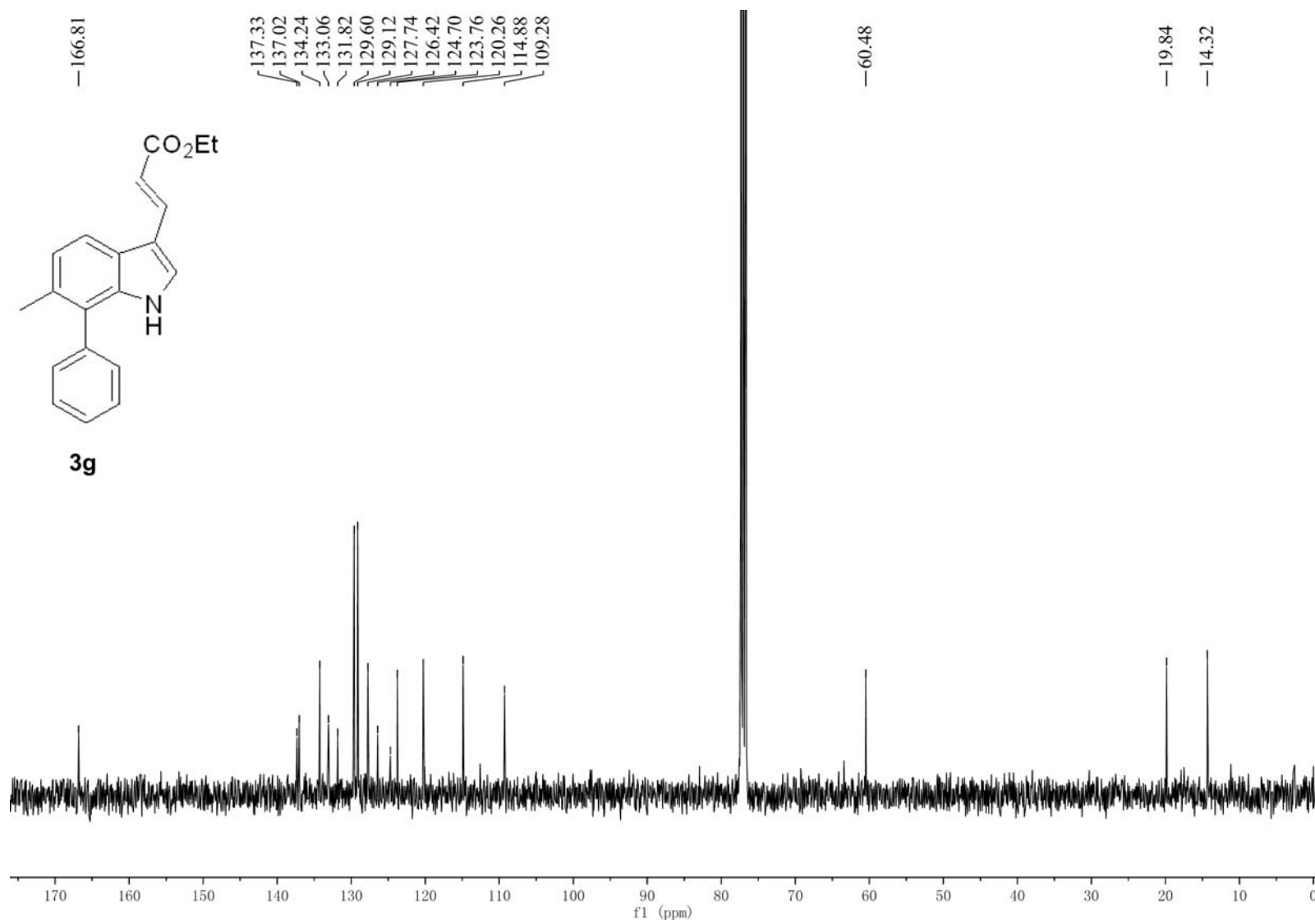


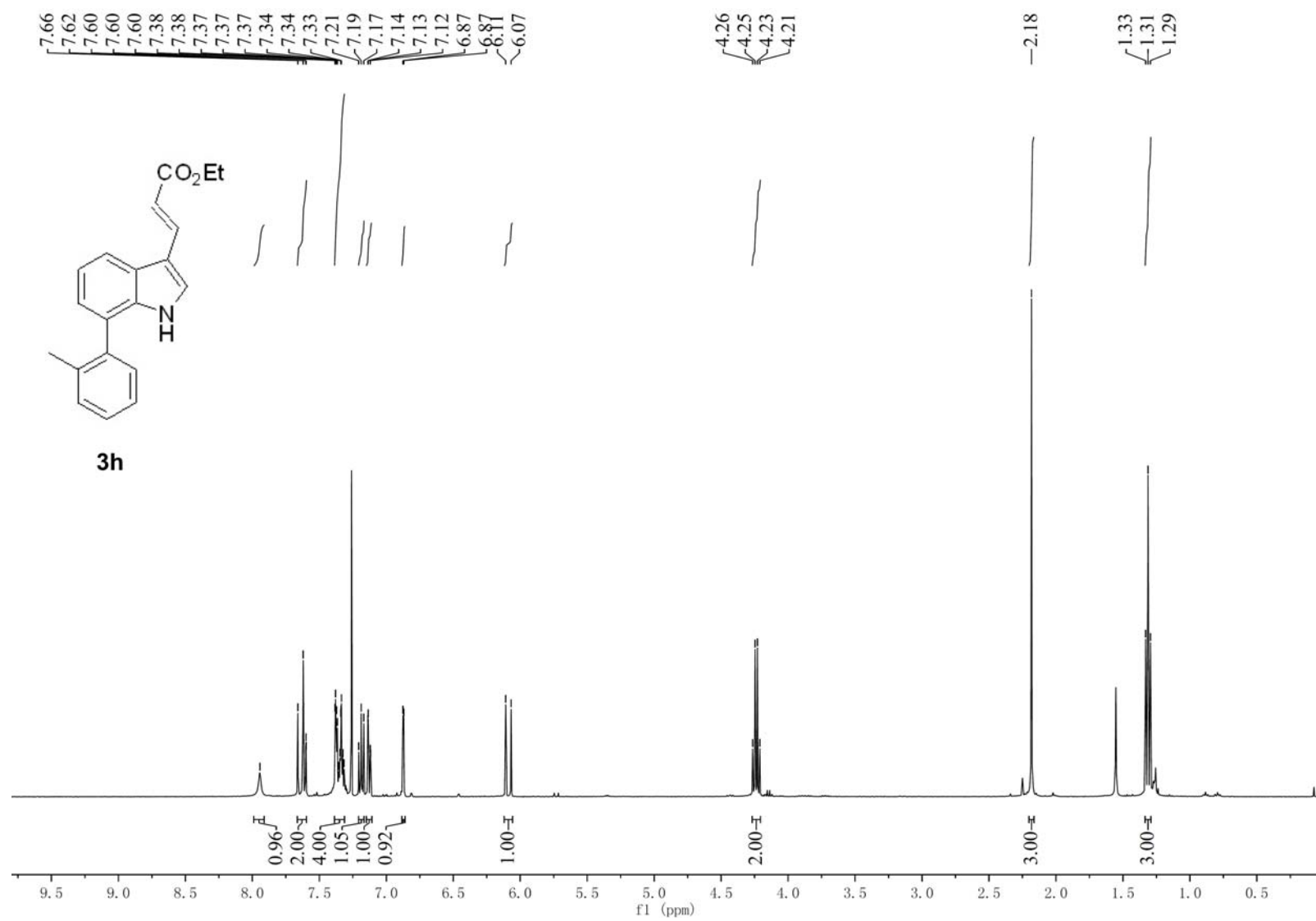


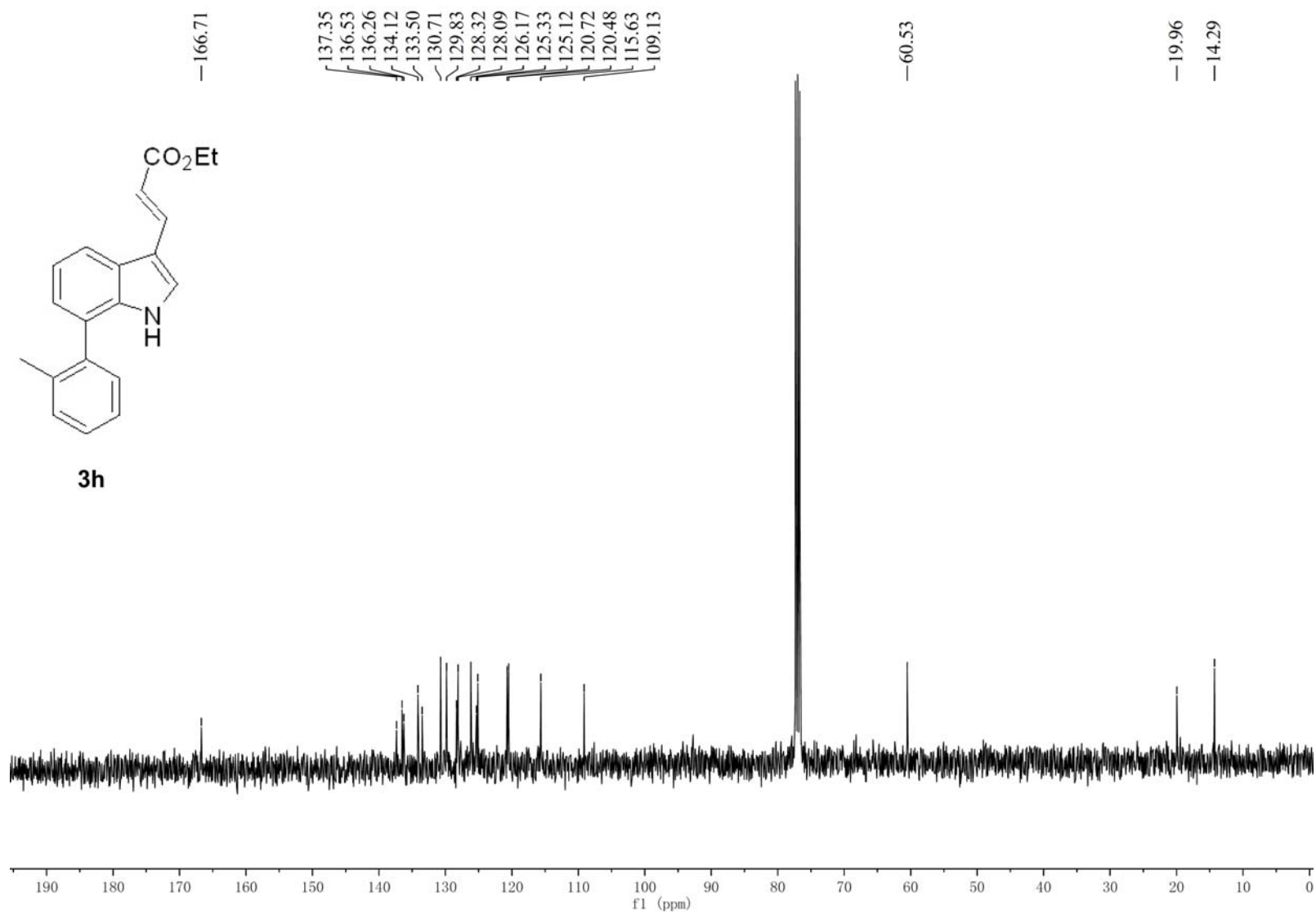


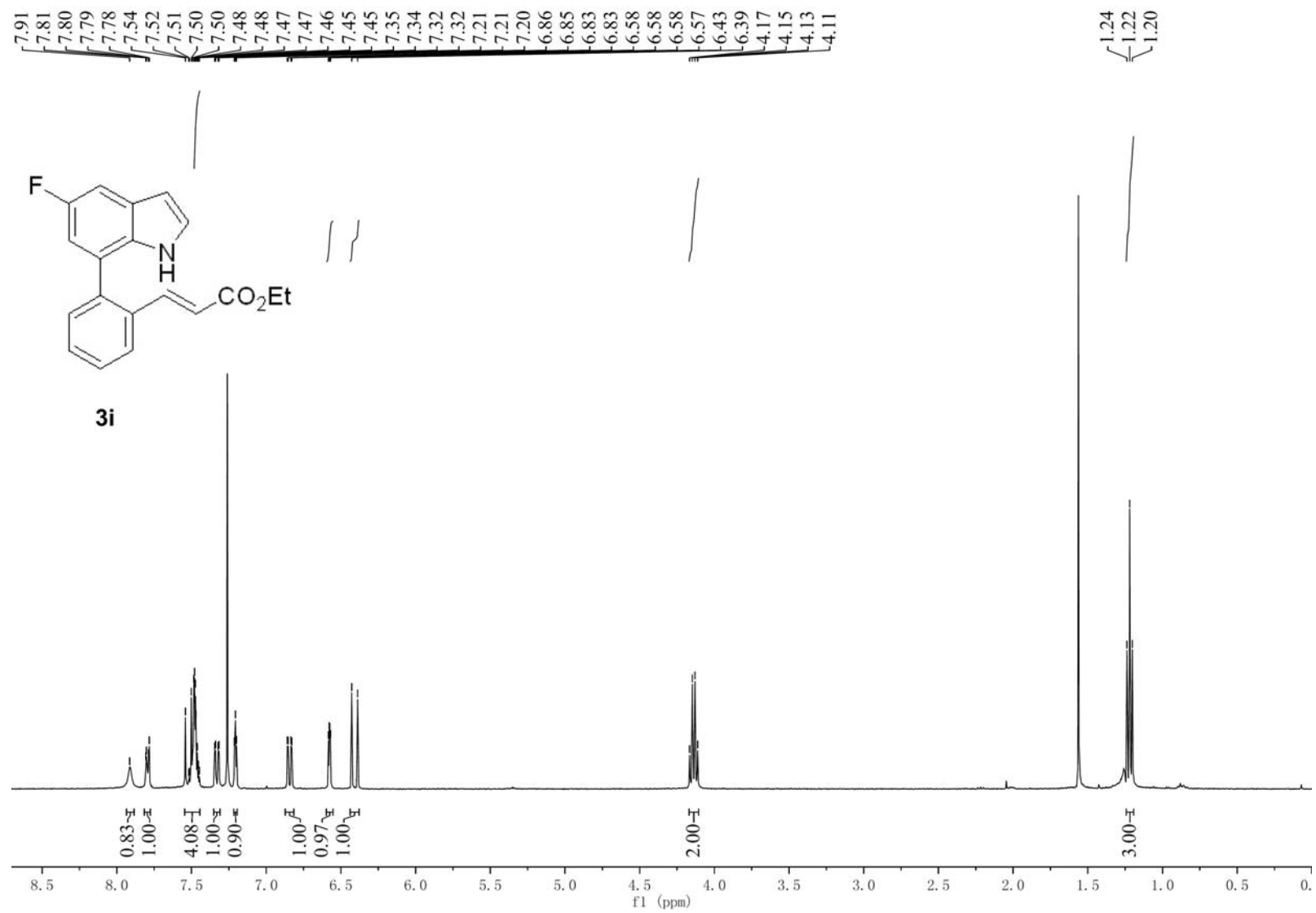


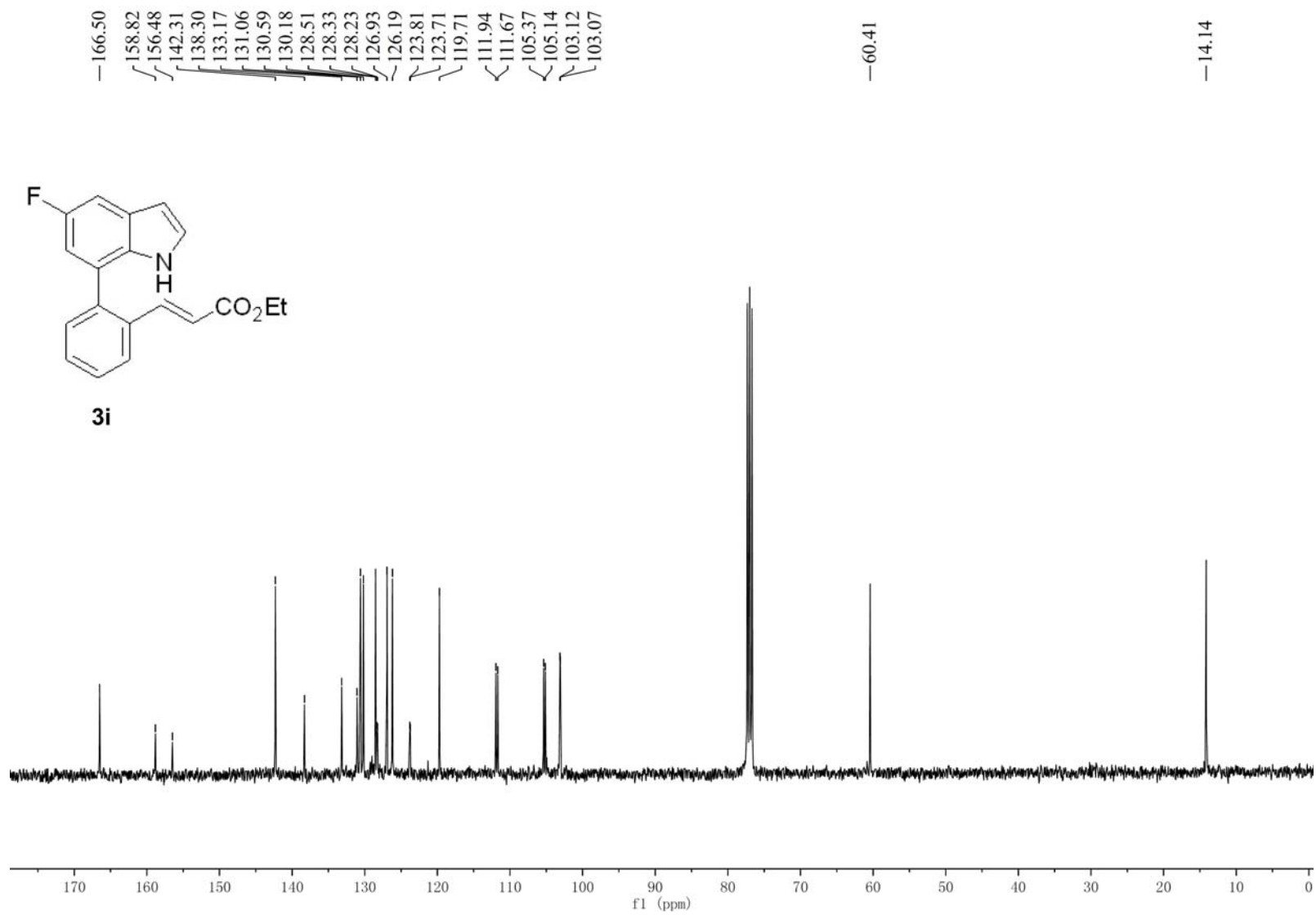


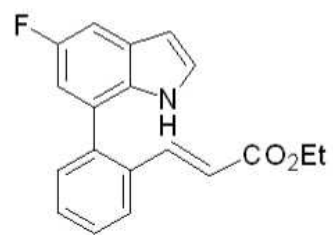




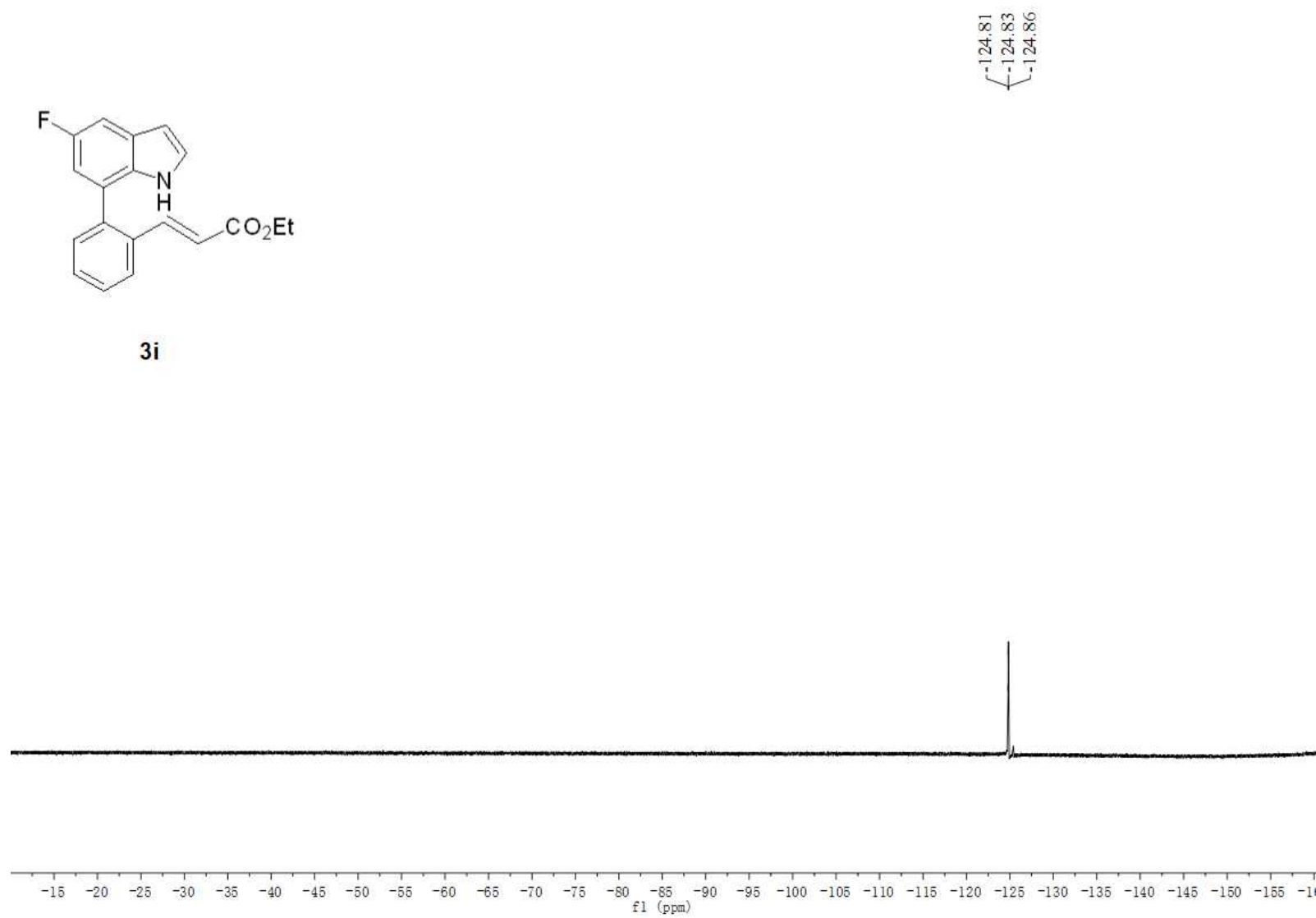


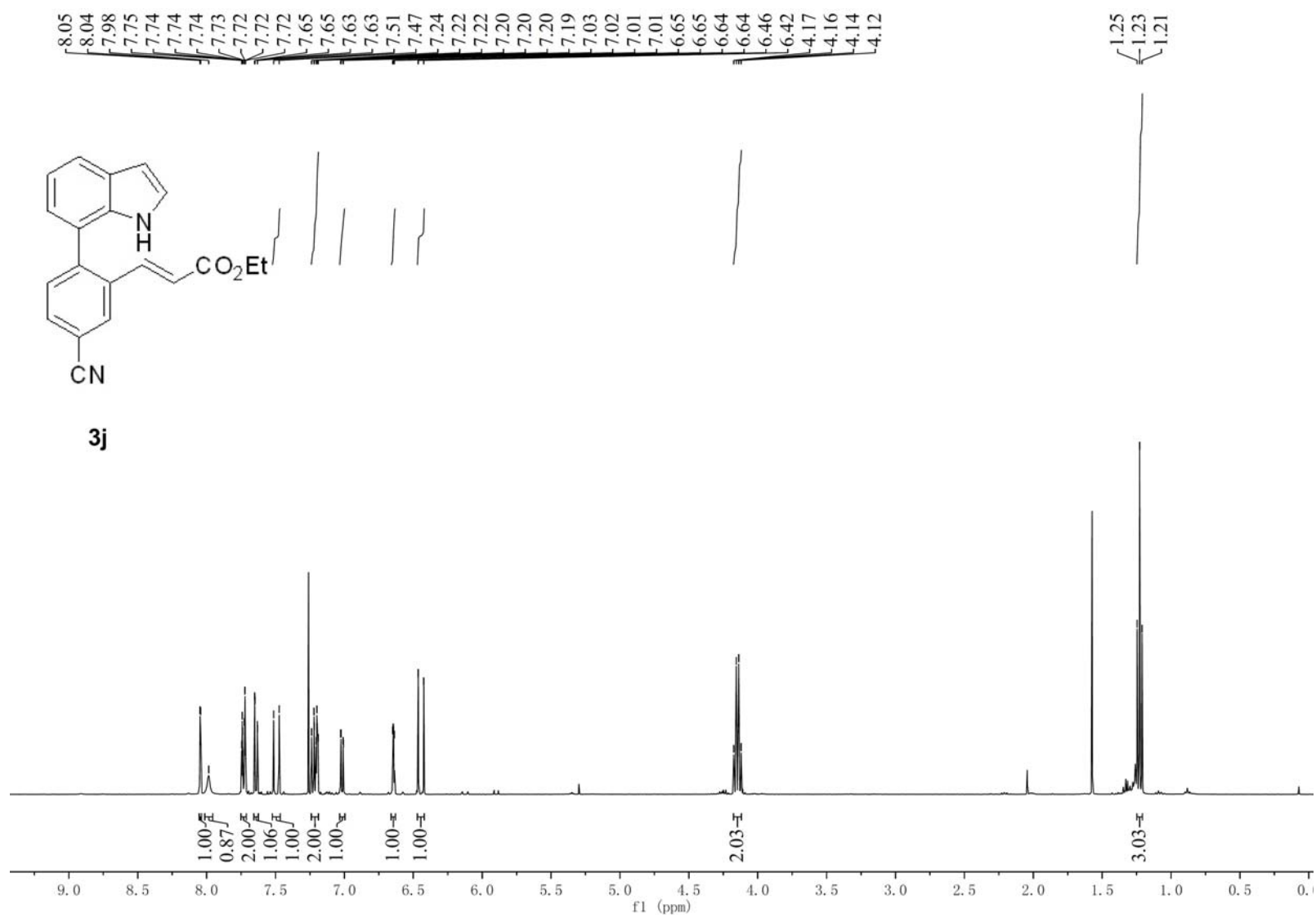


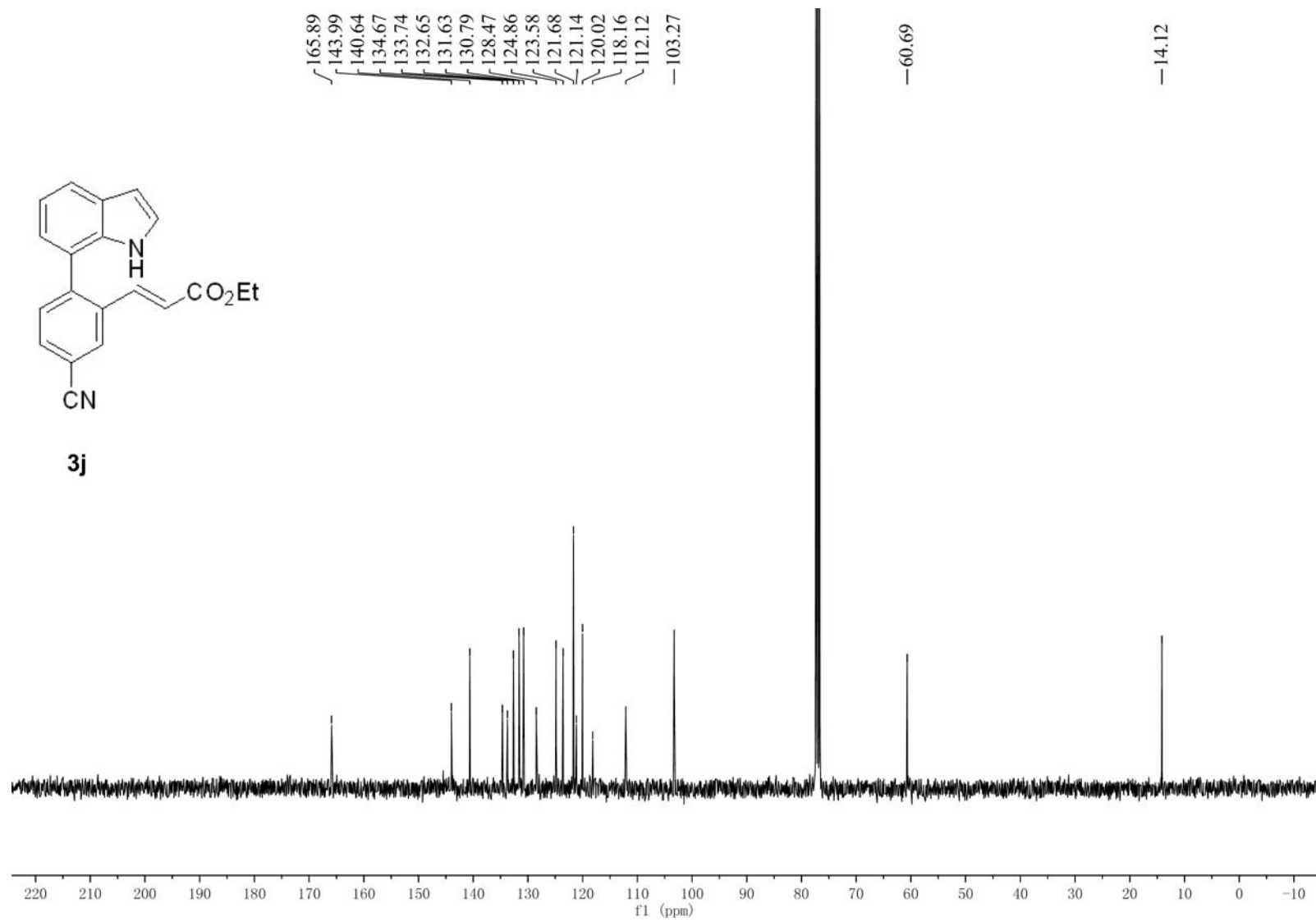


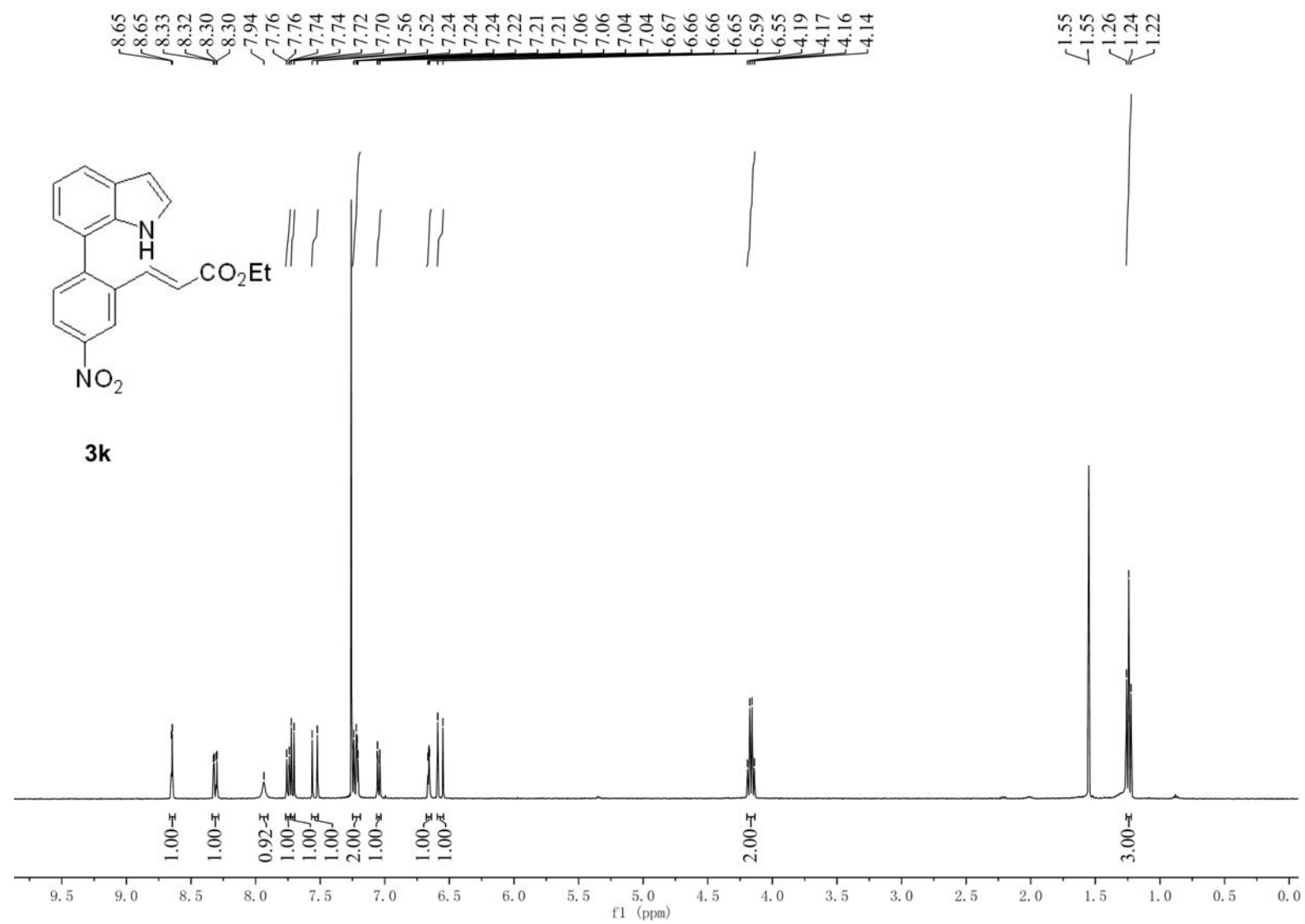


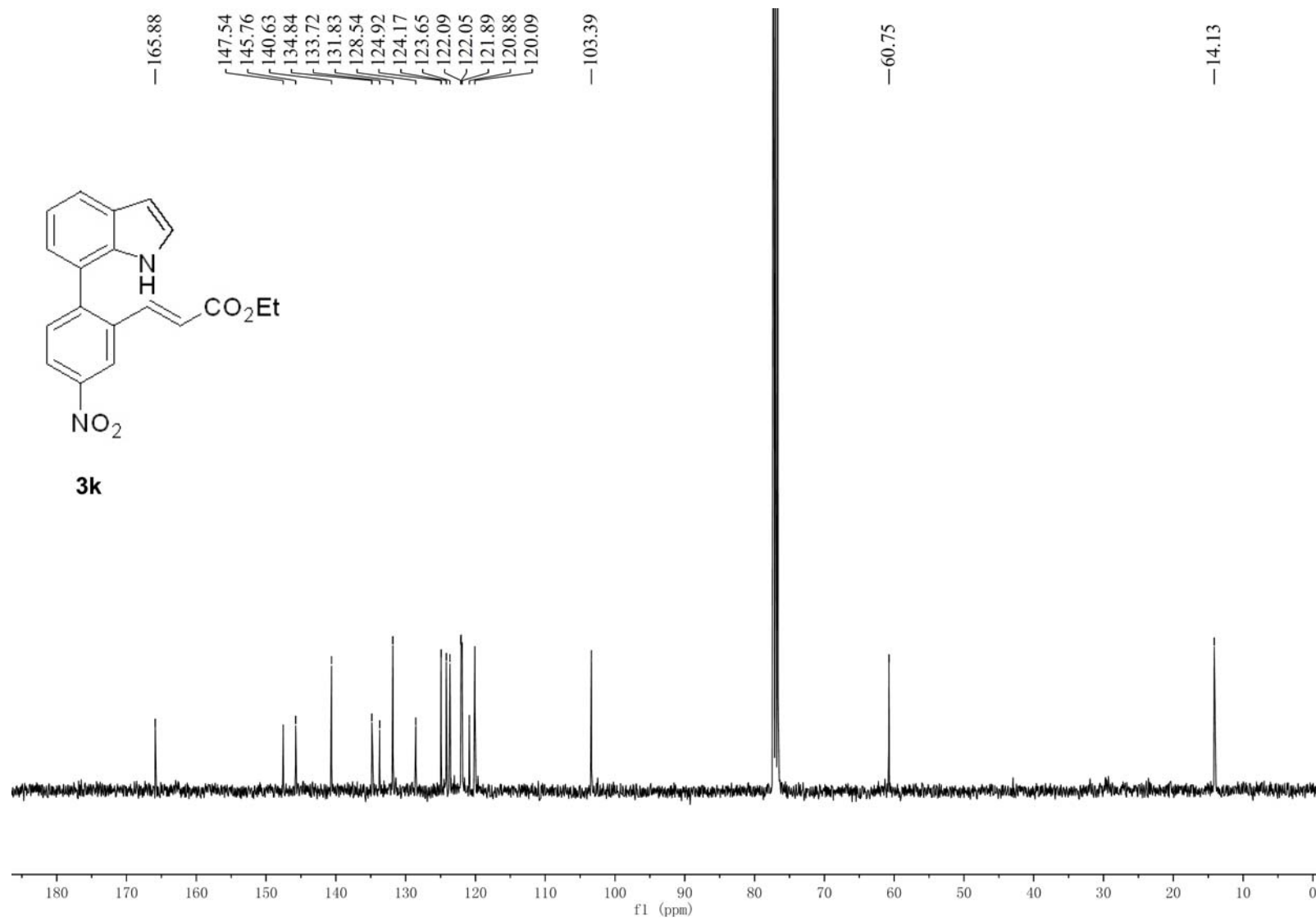
3i

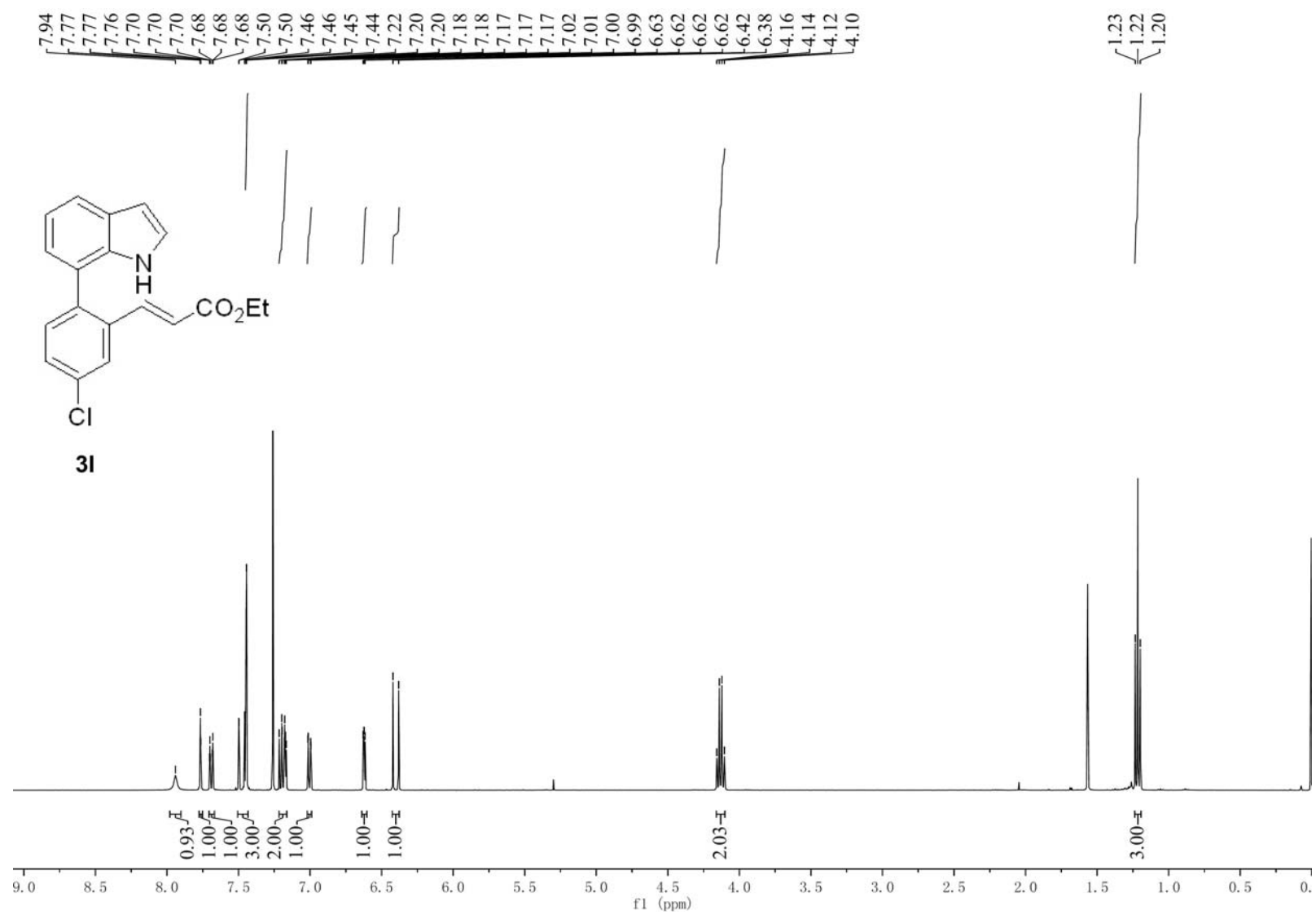


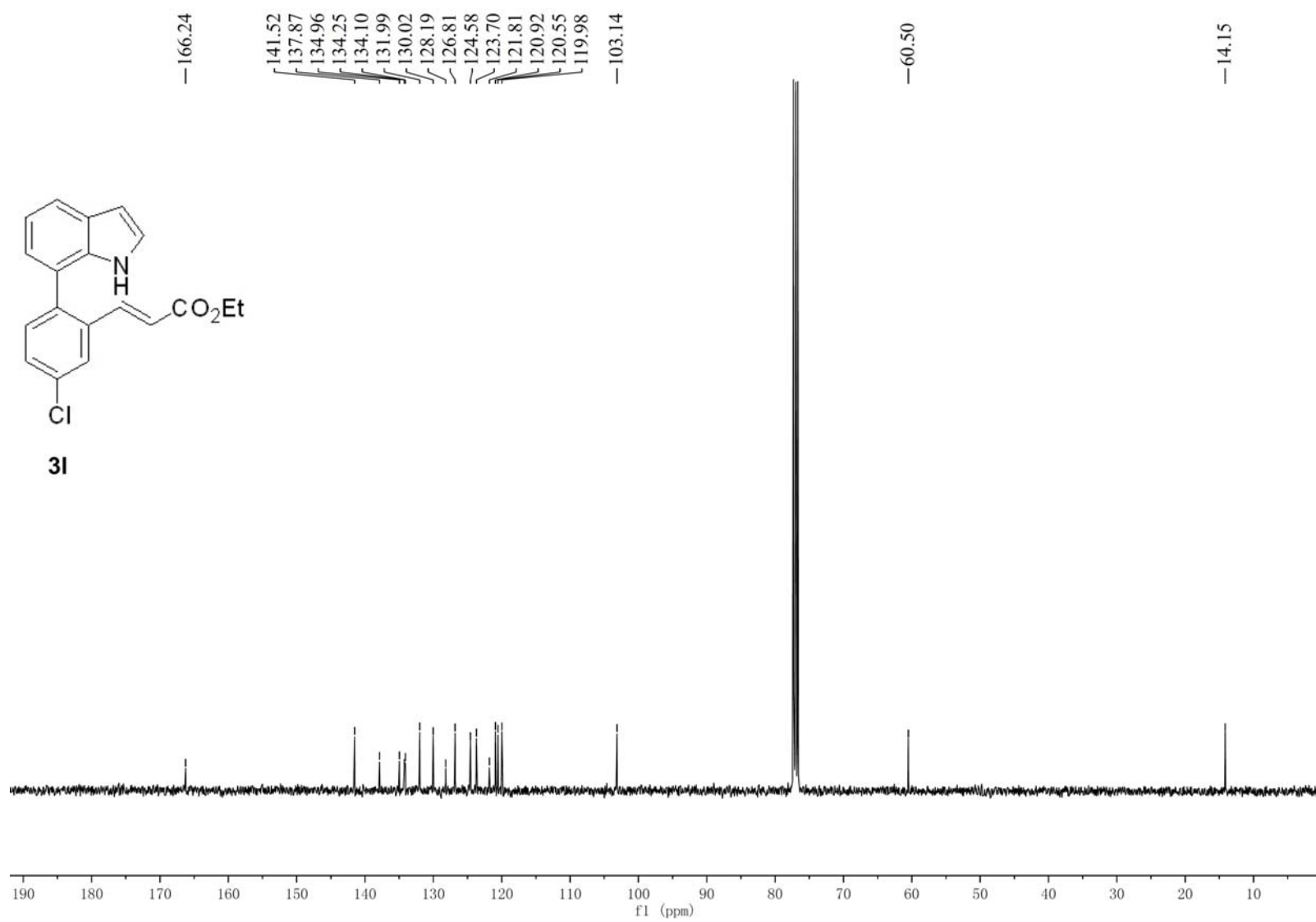


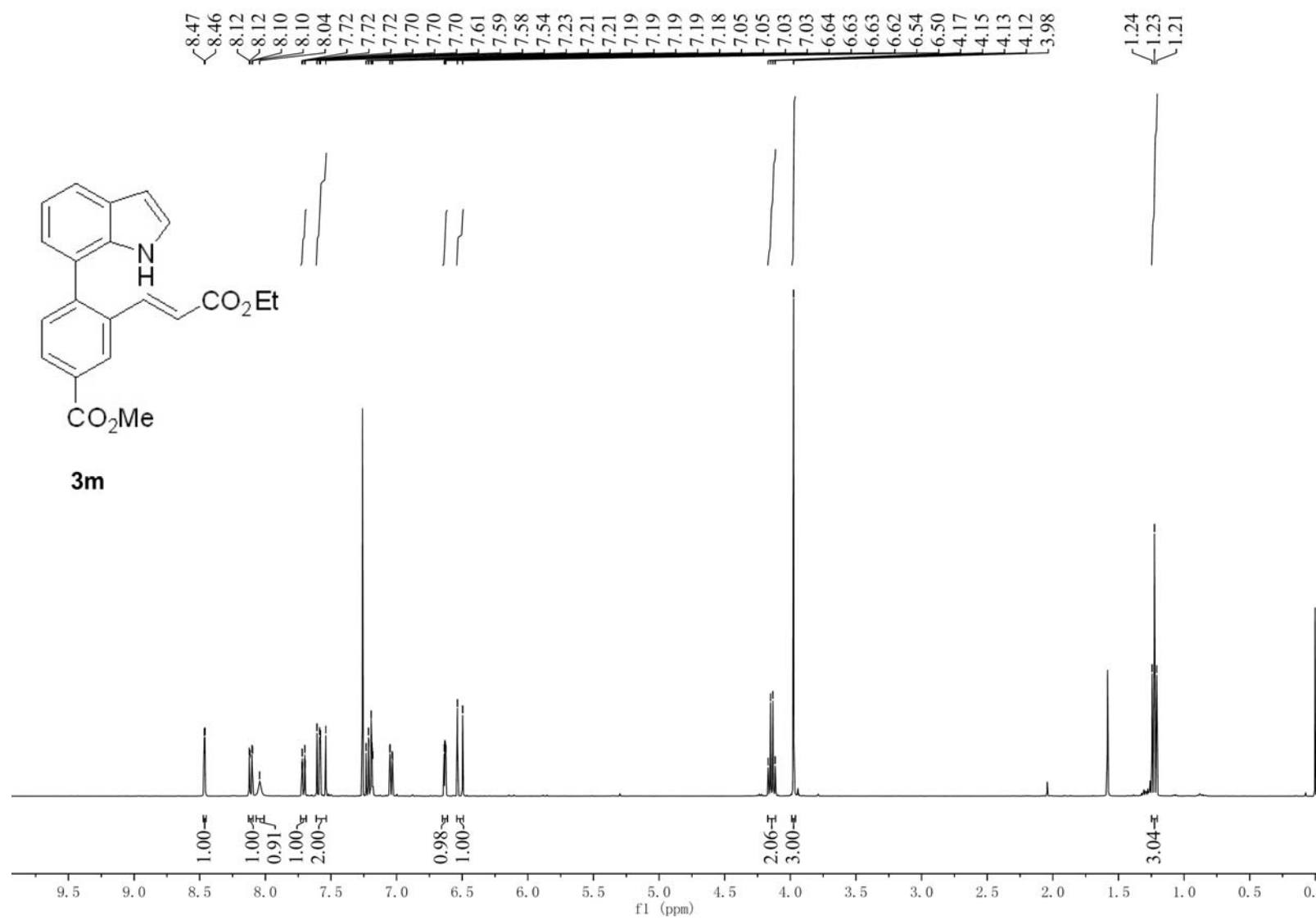


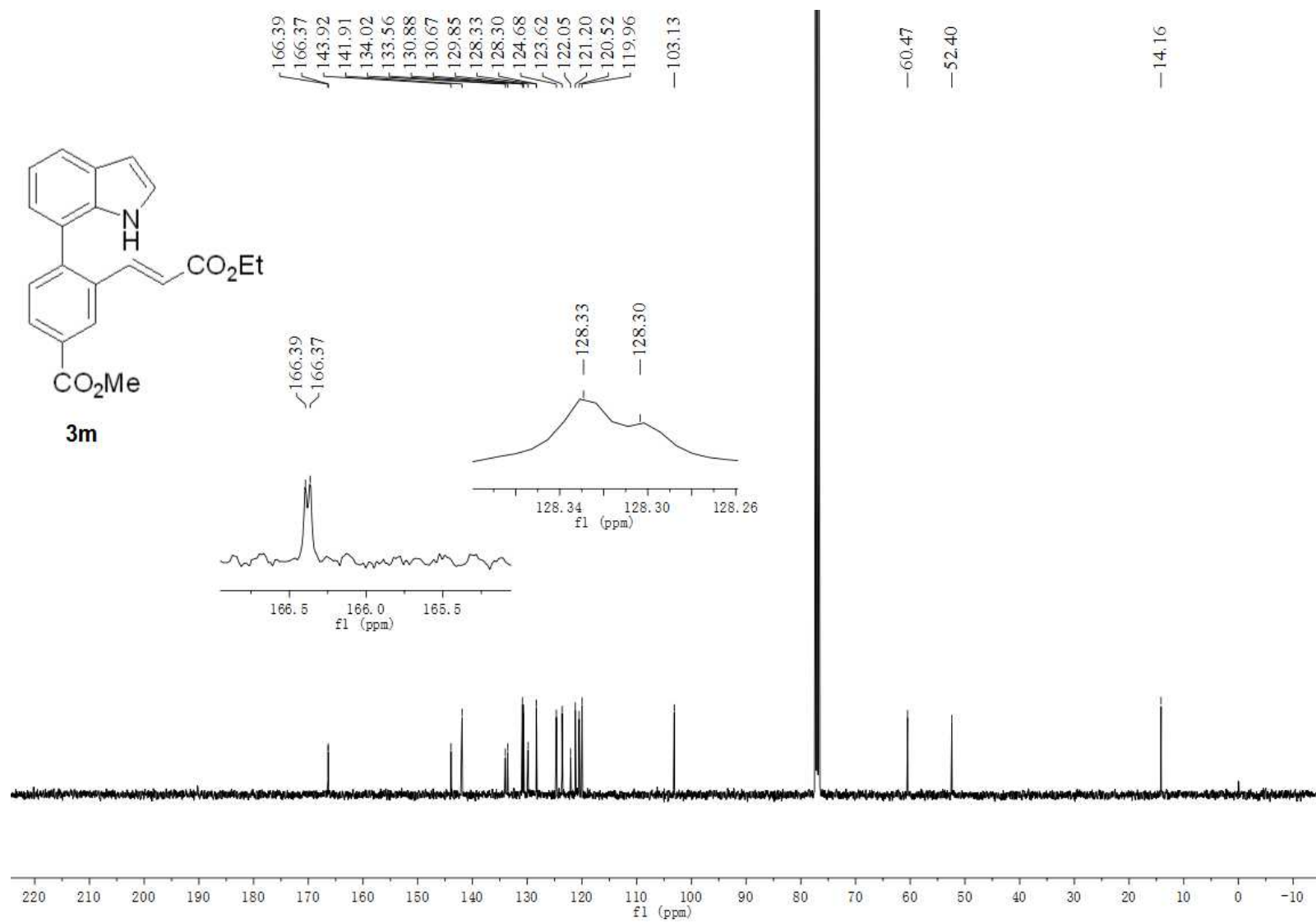


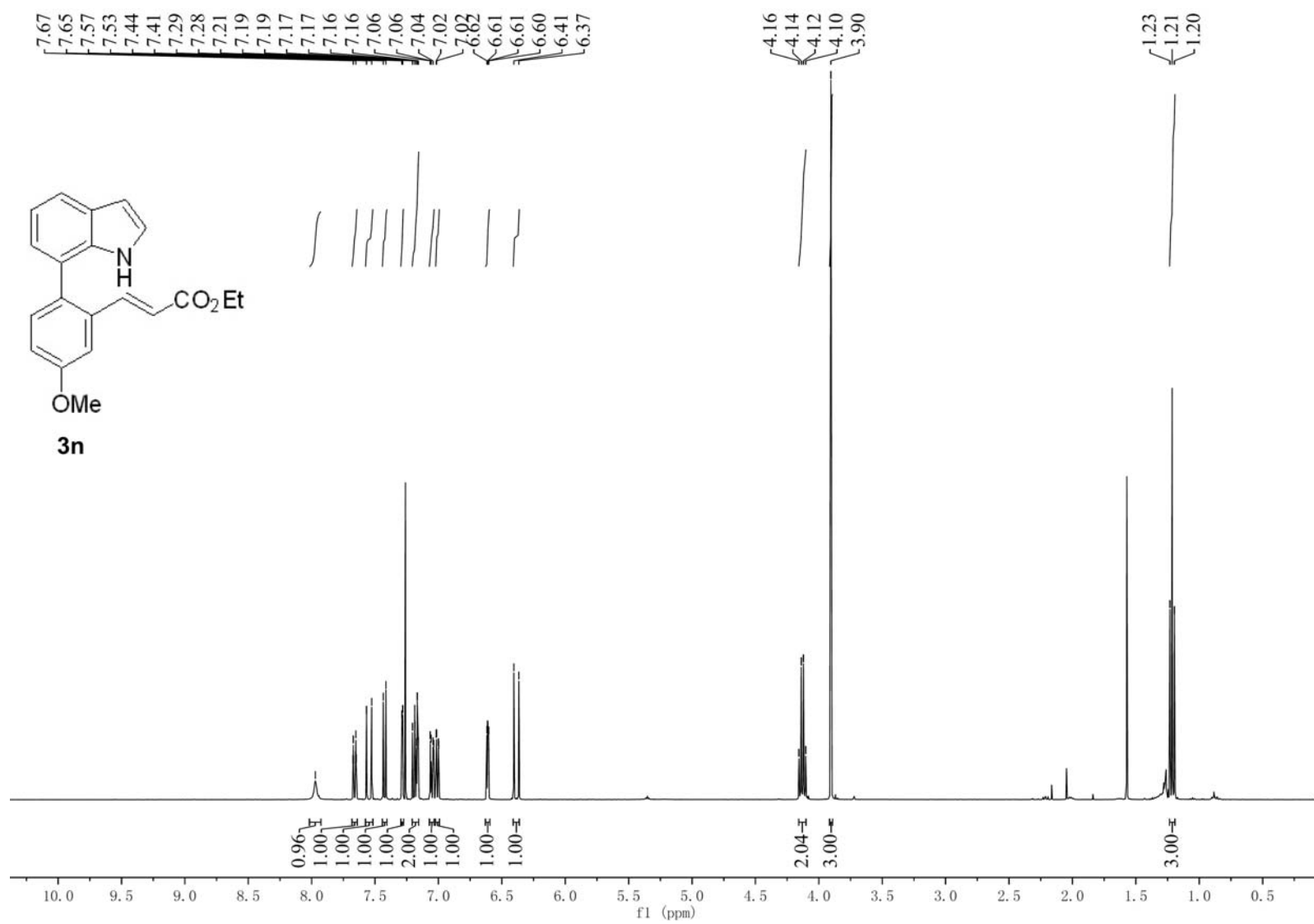


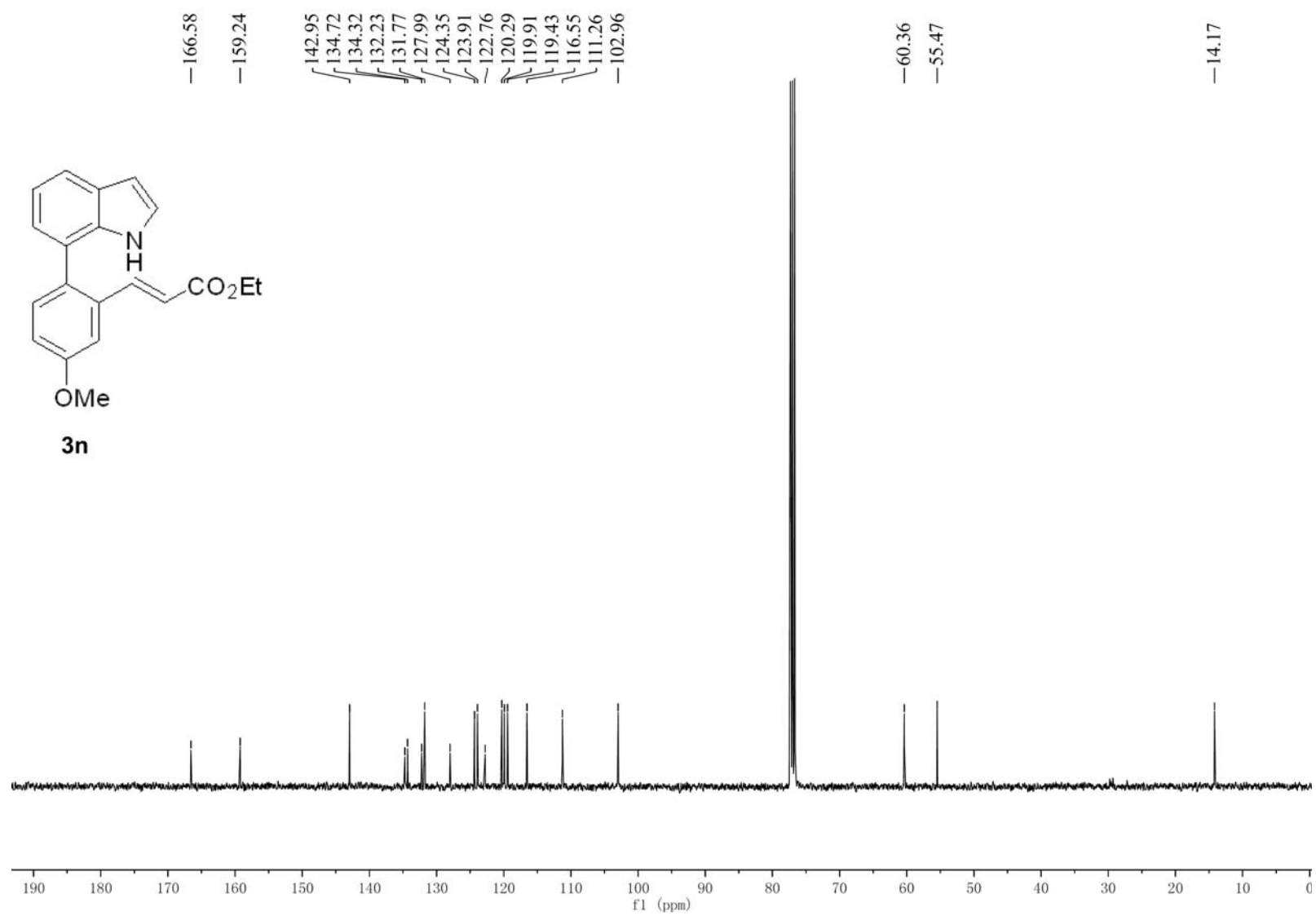


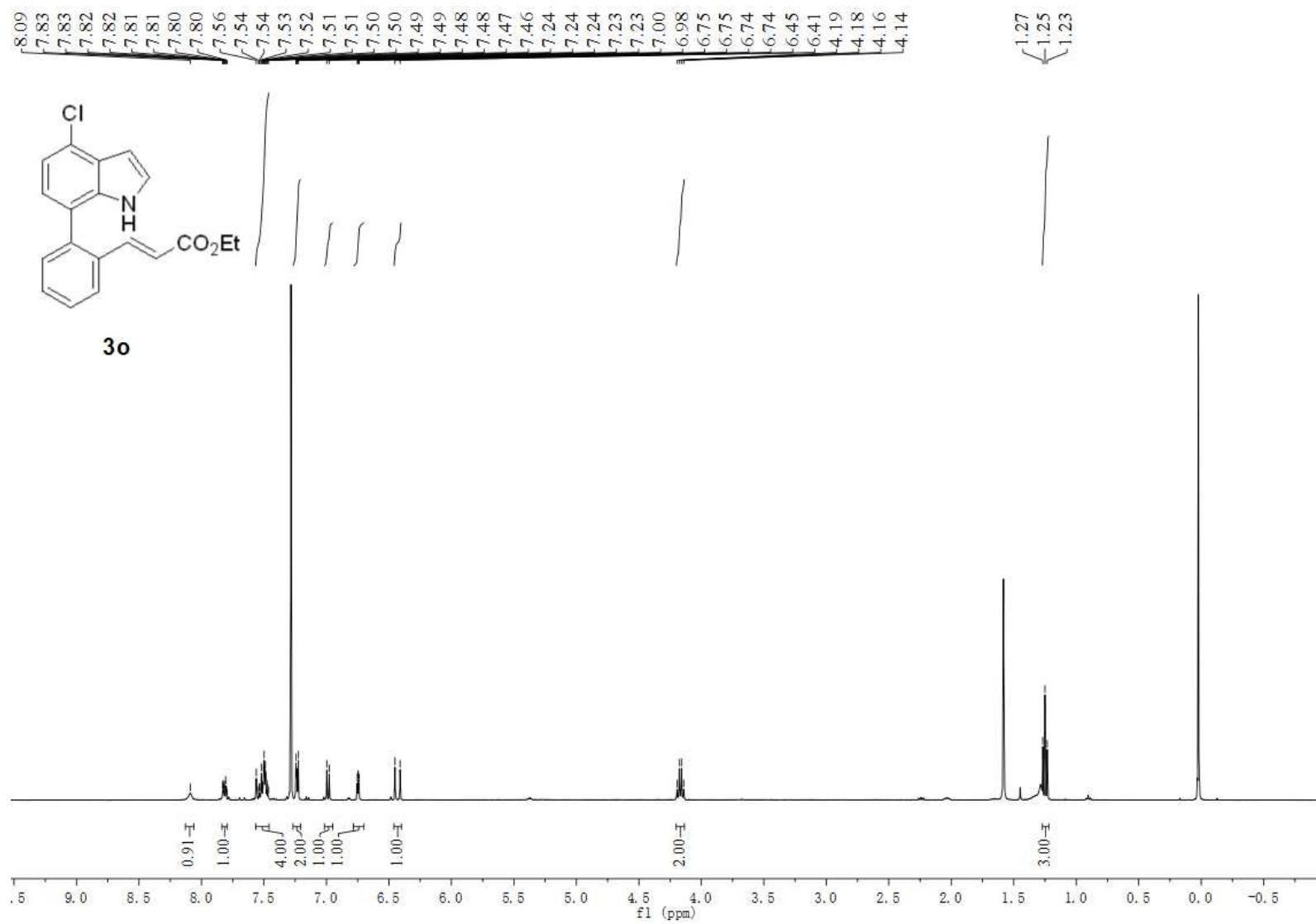


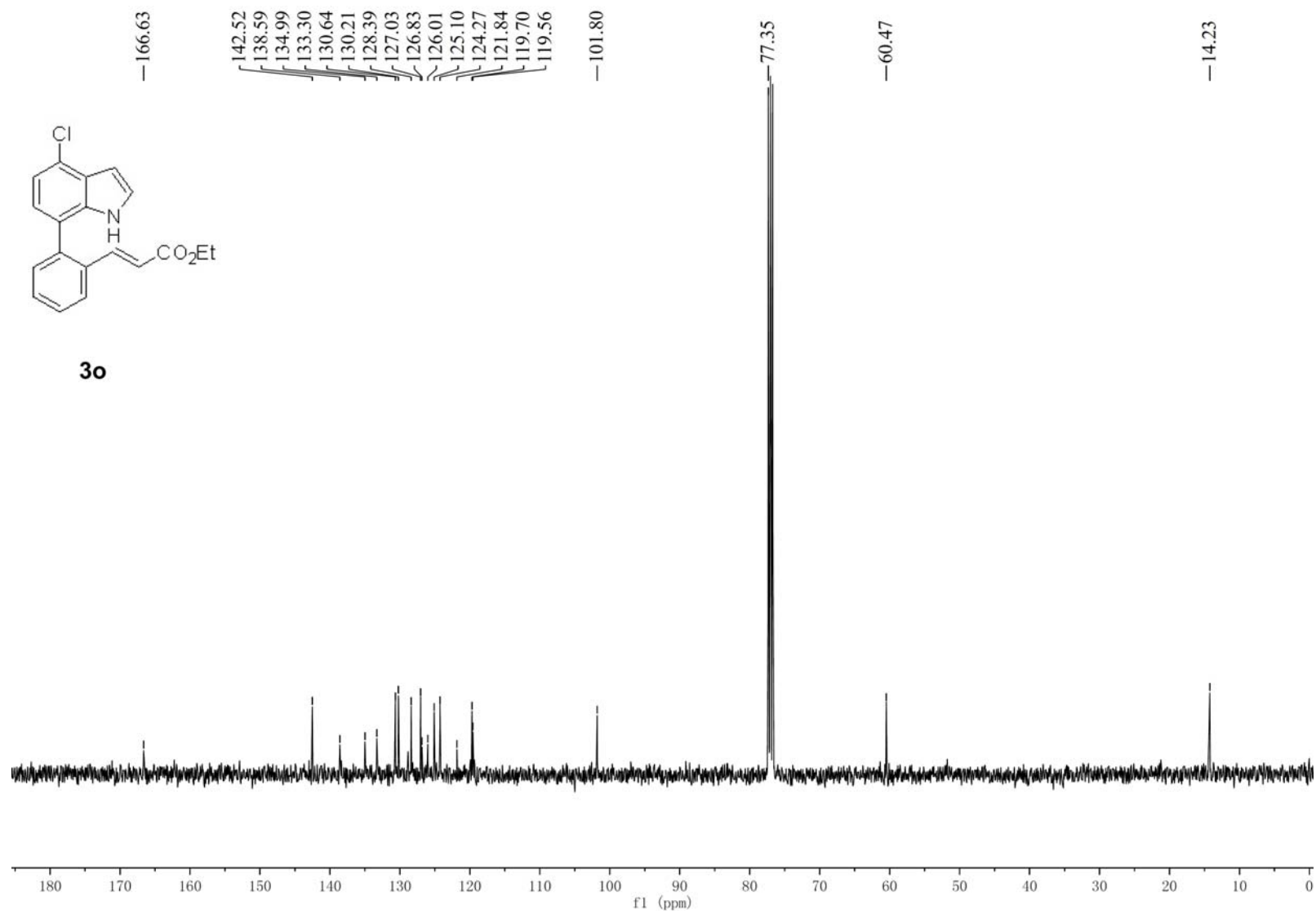


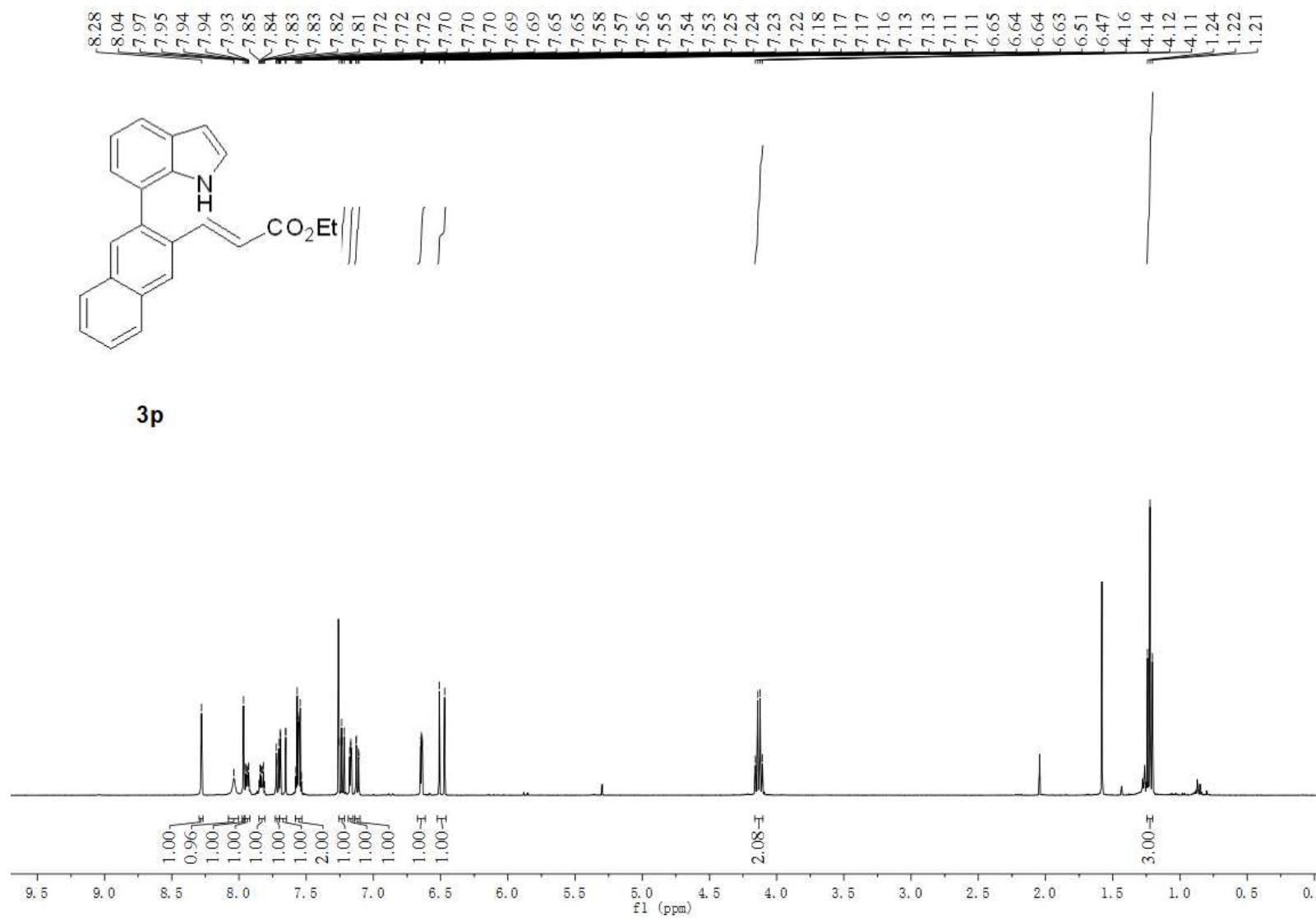


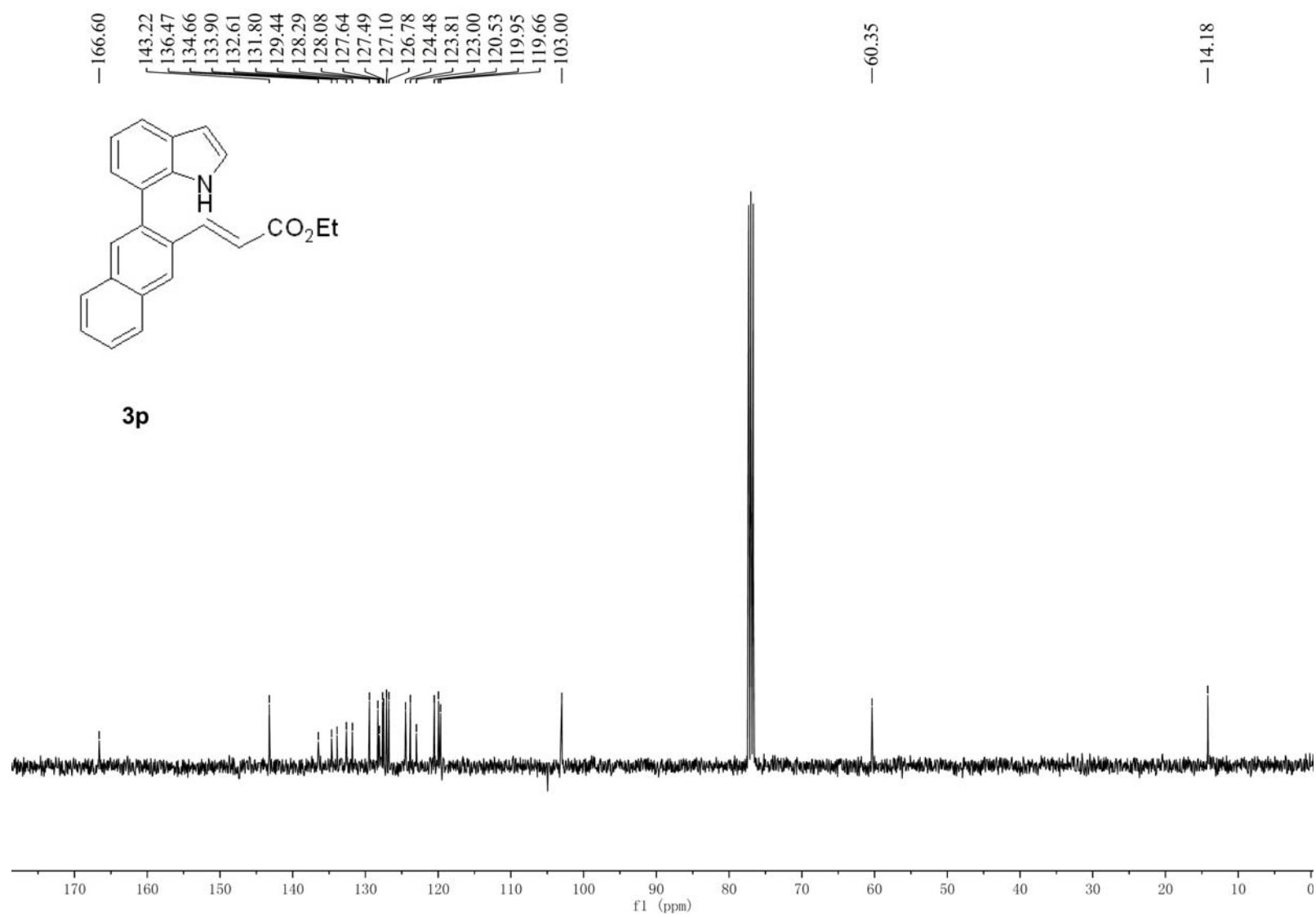


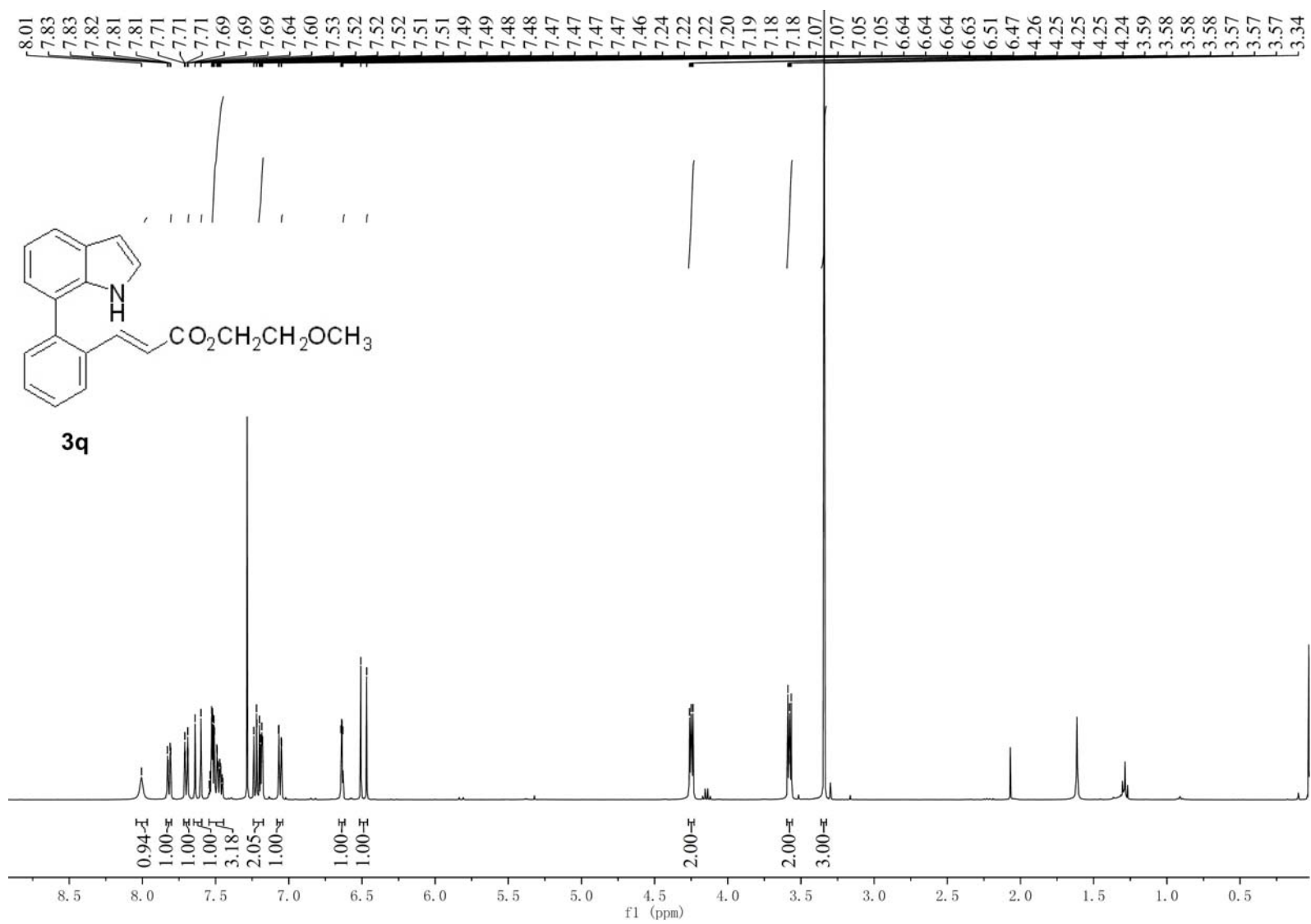


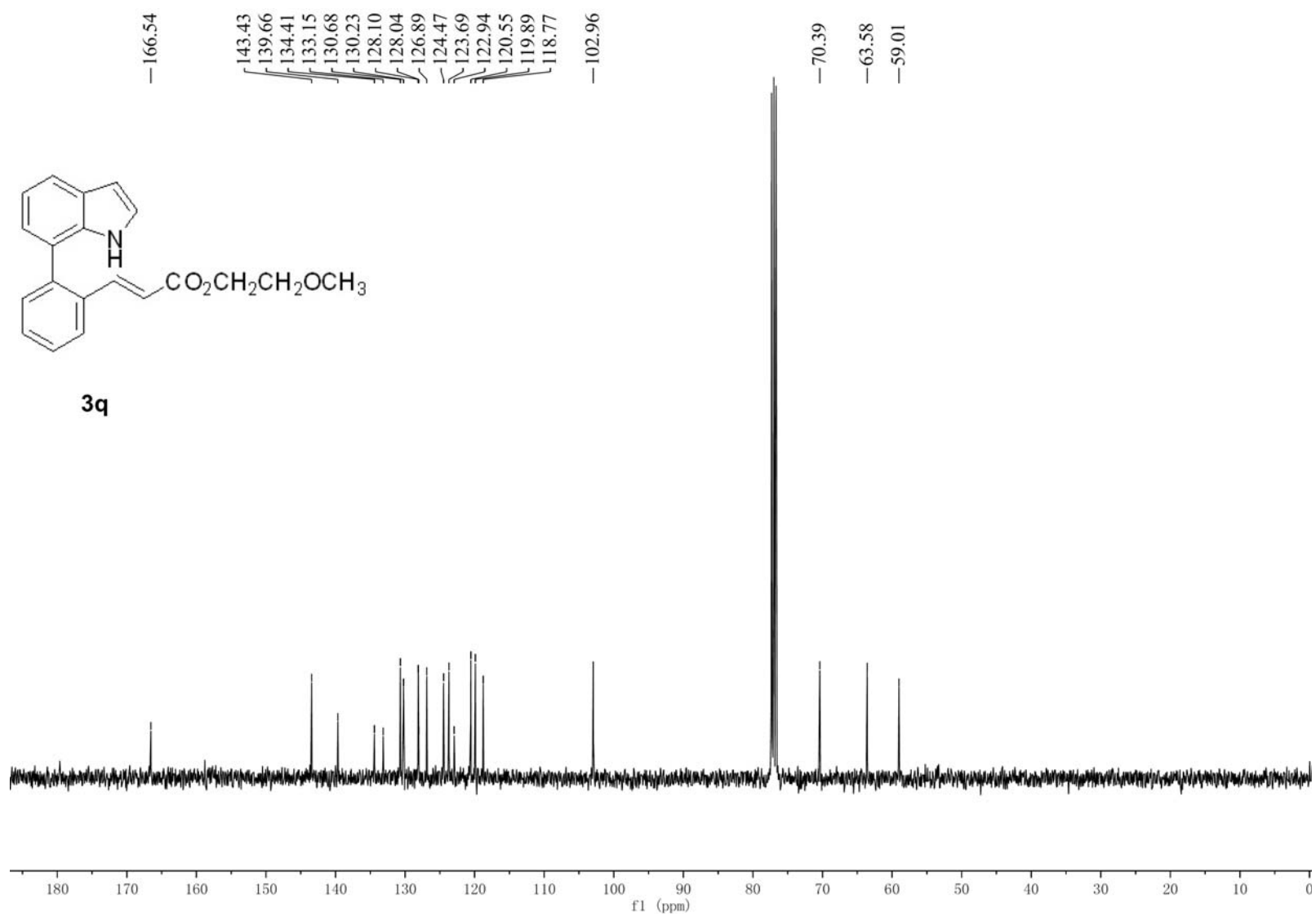


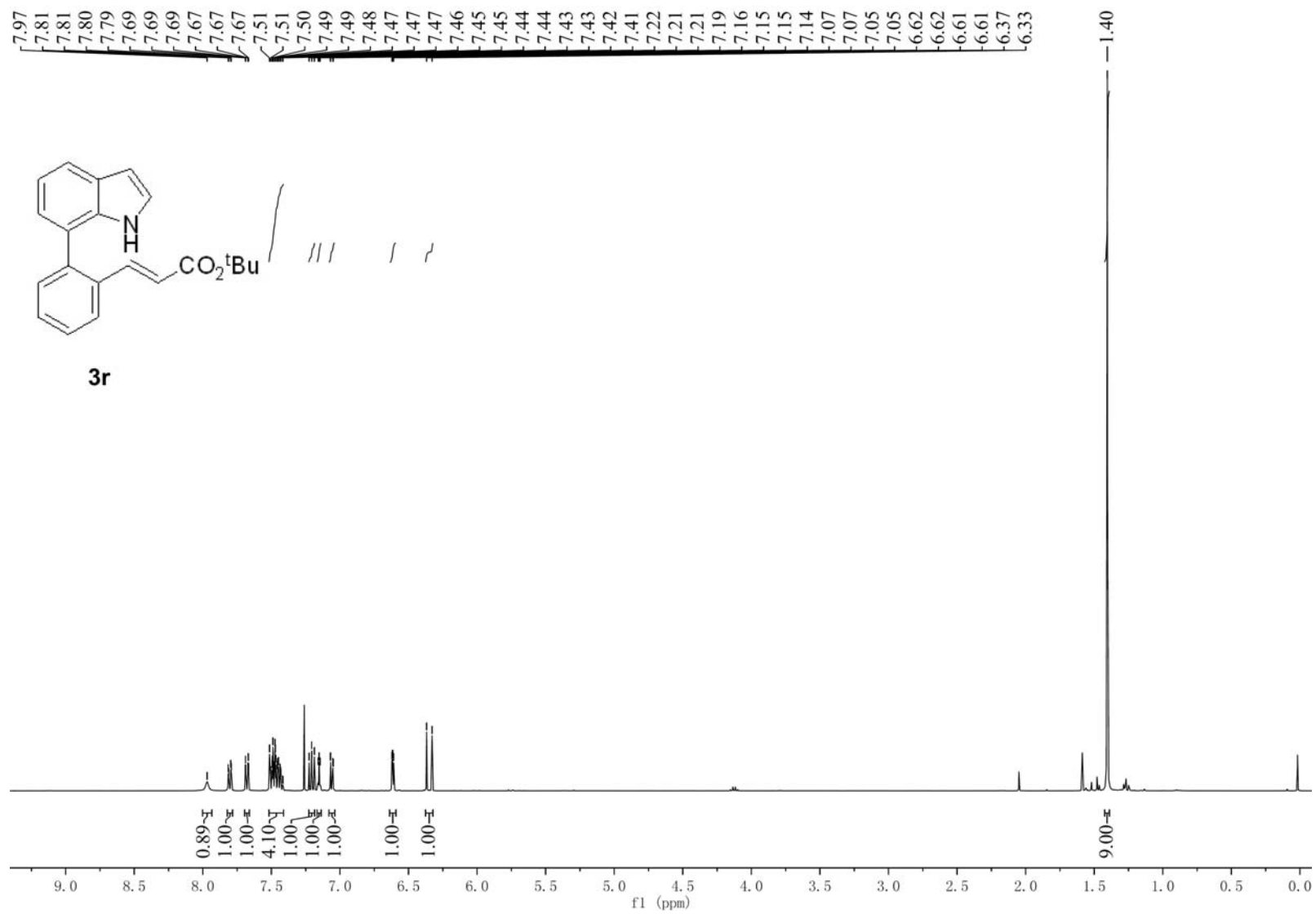


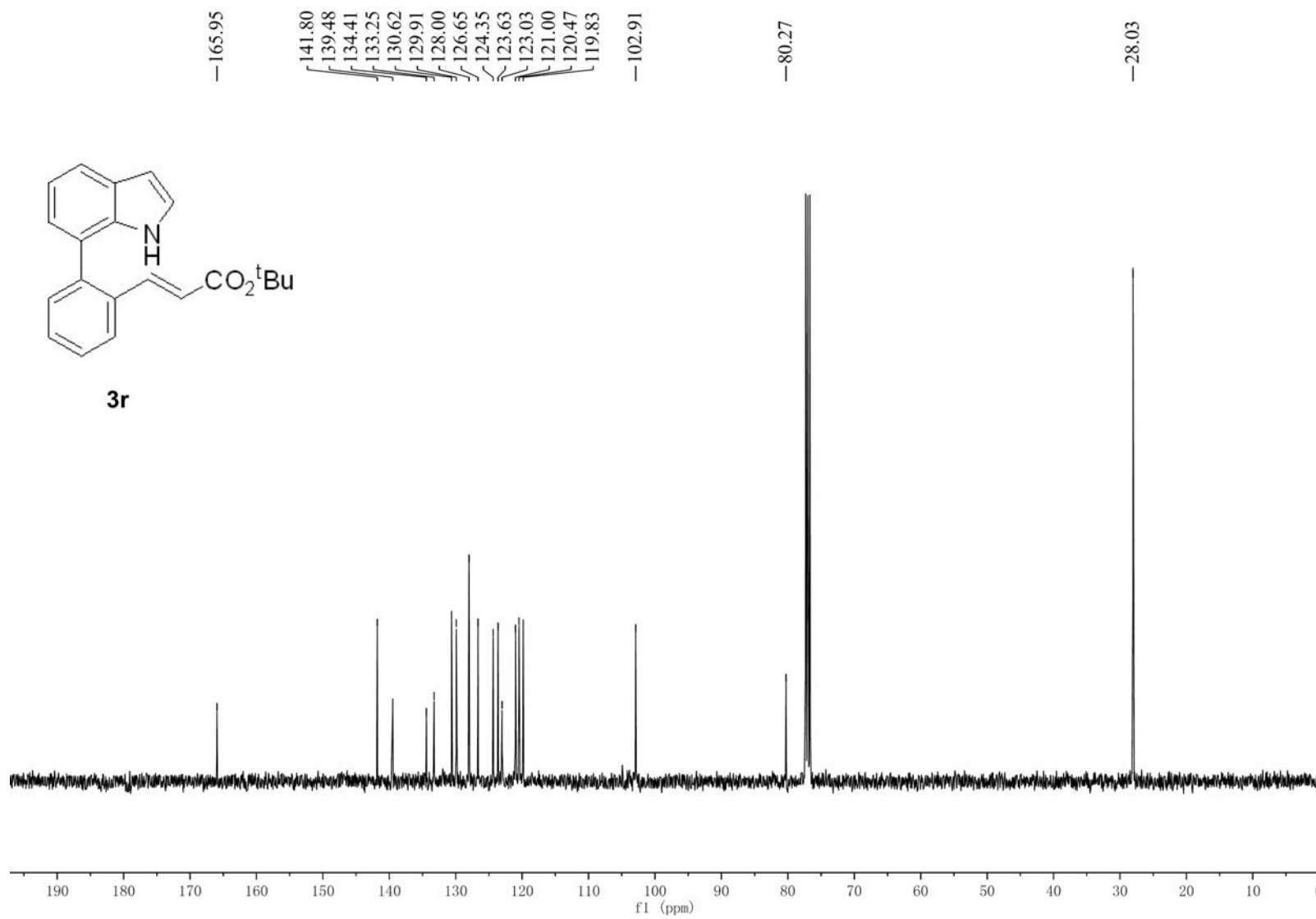


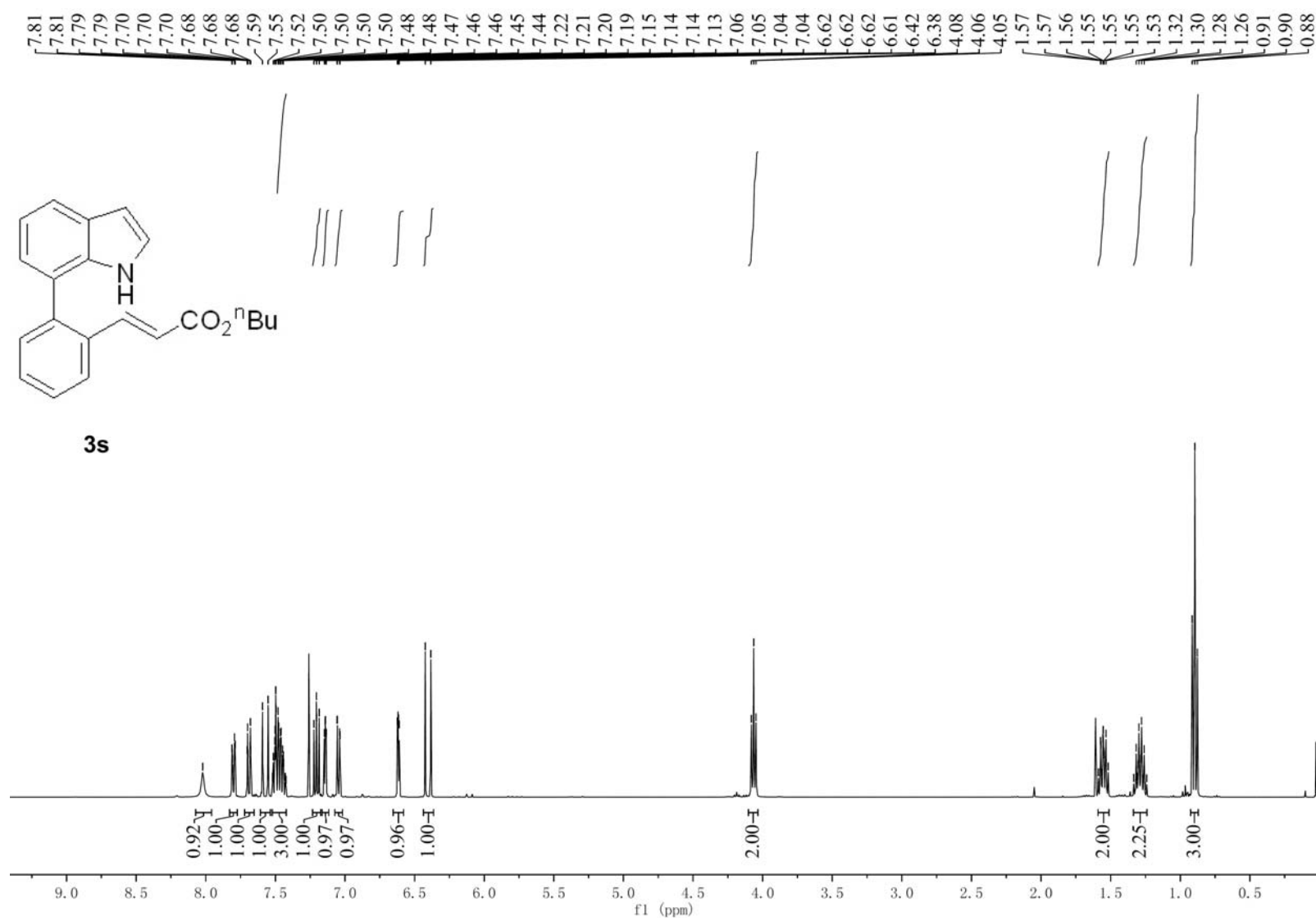


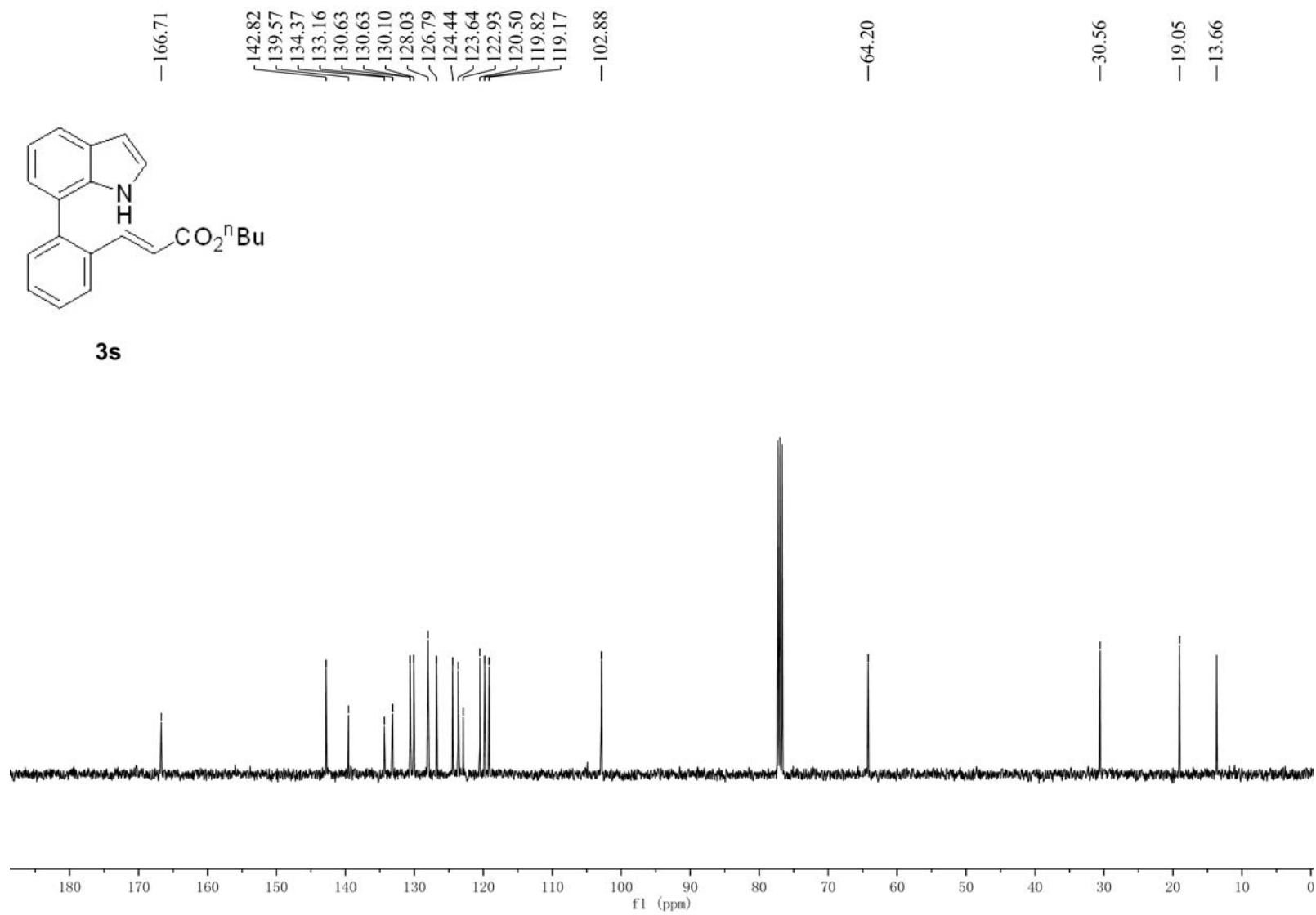


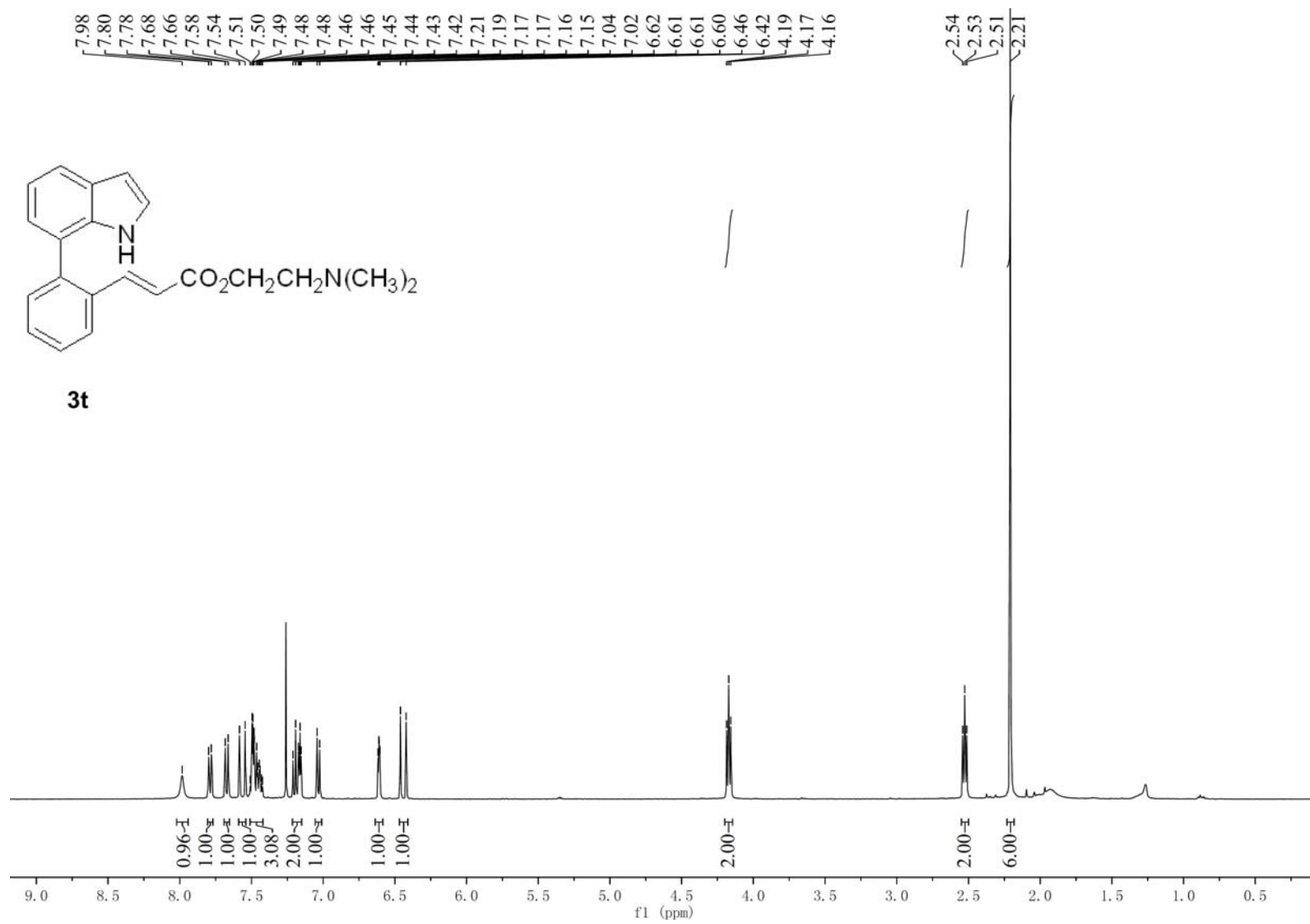


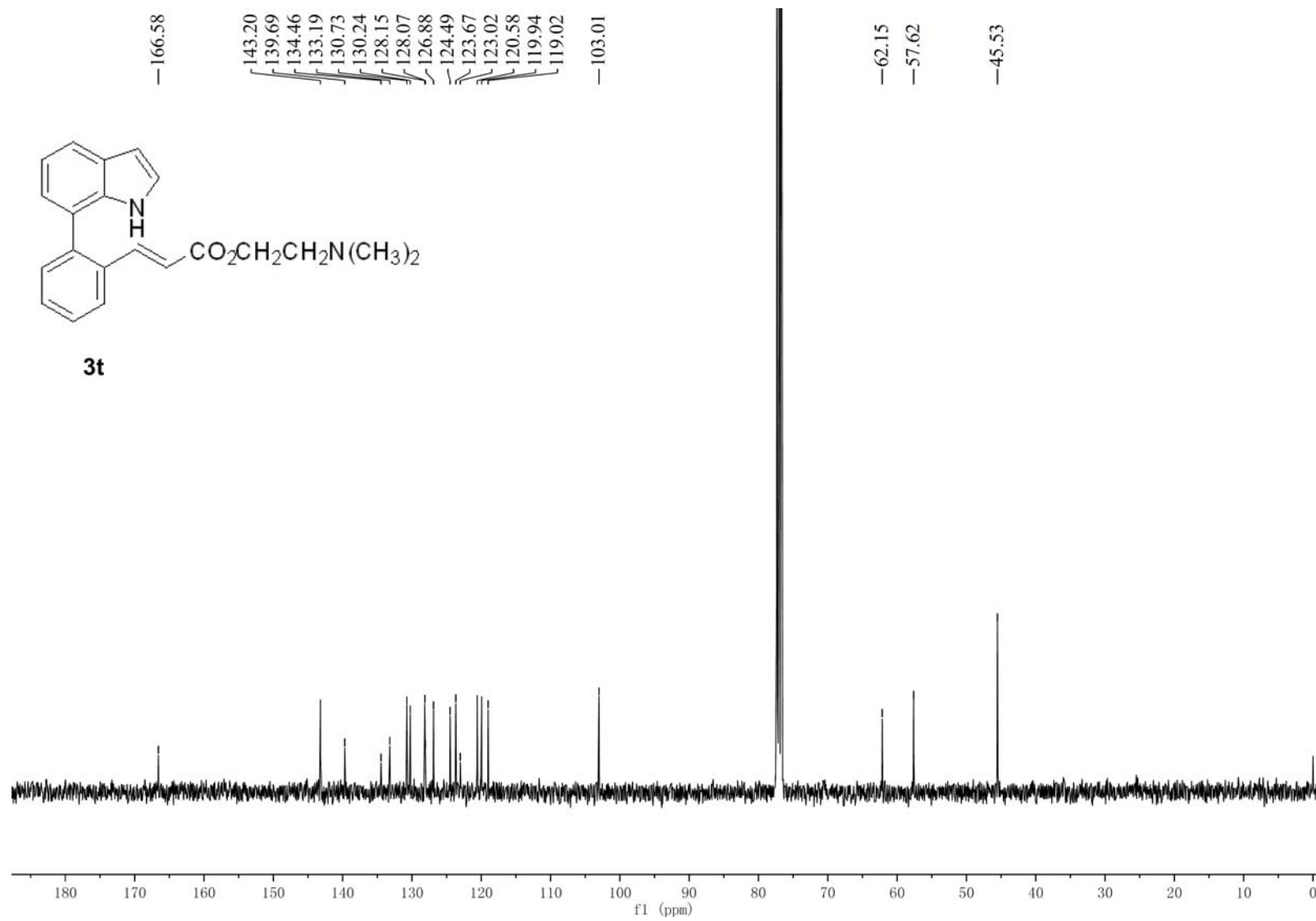


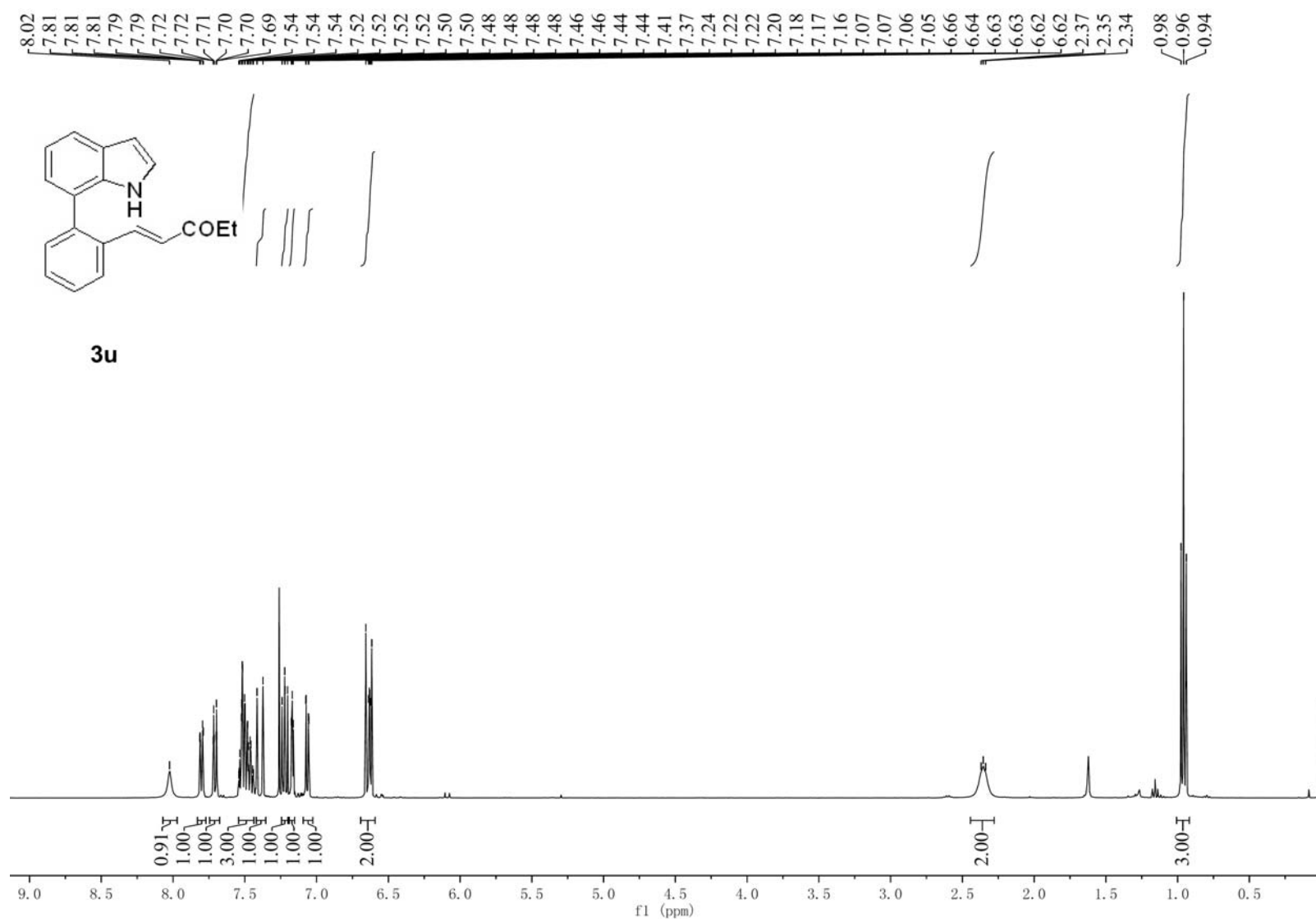


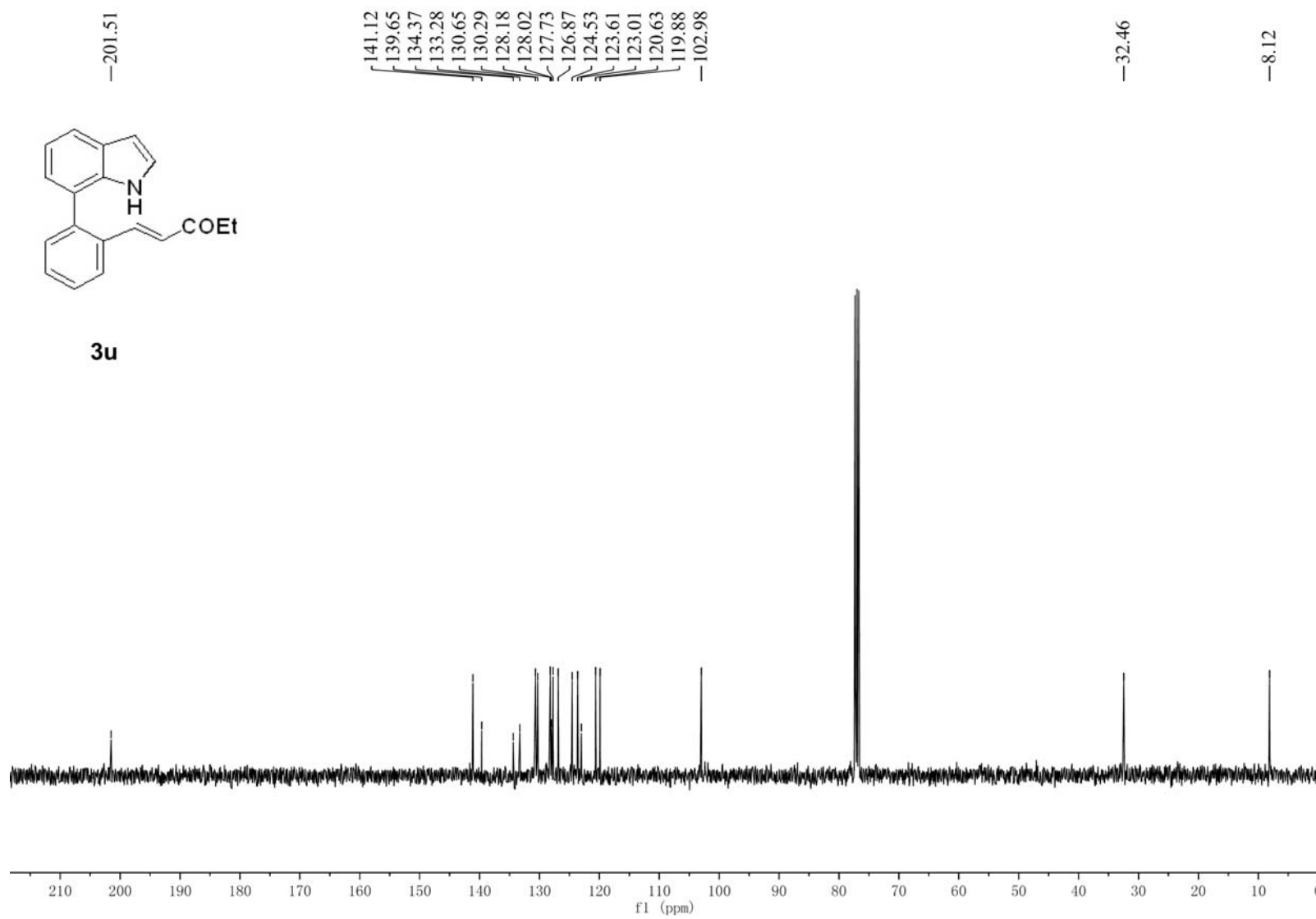


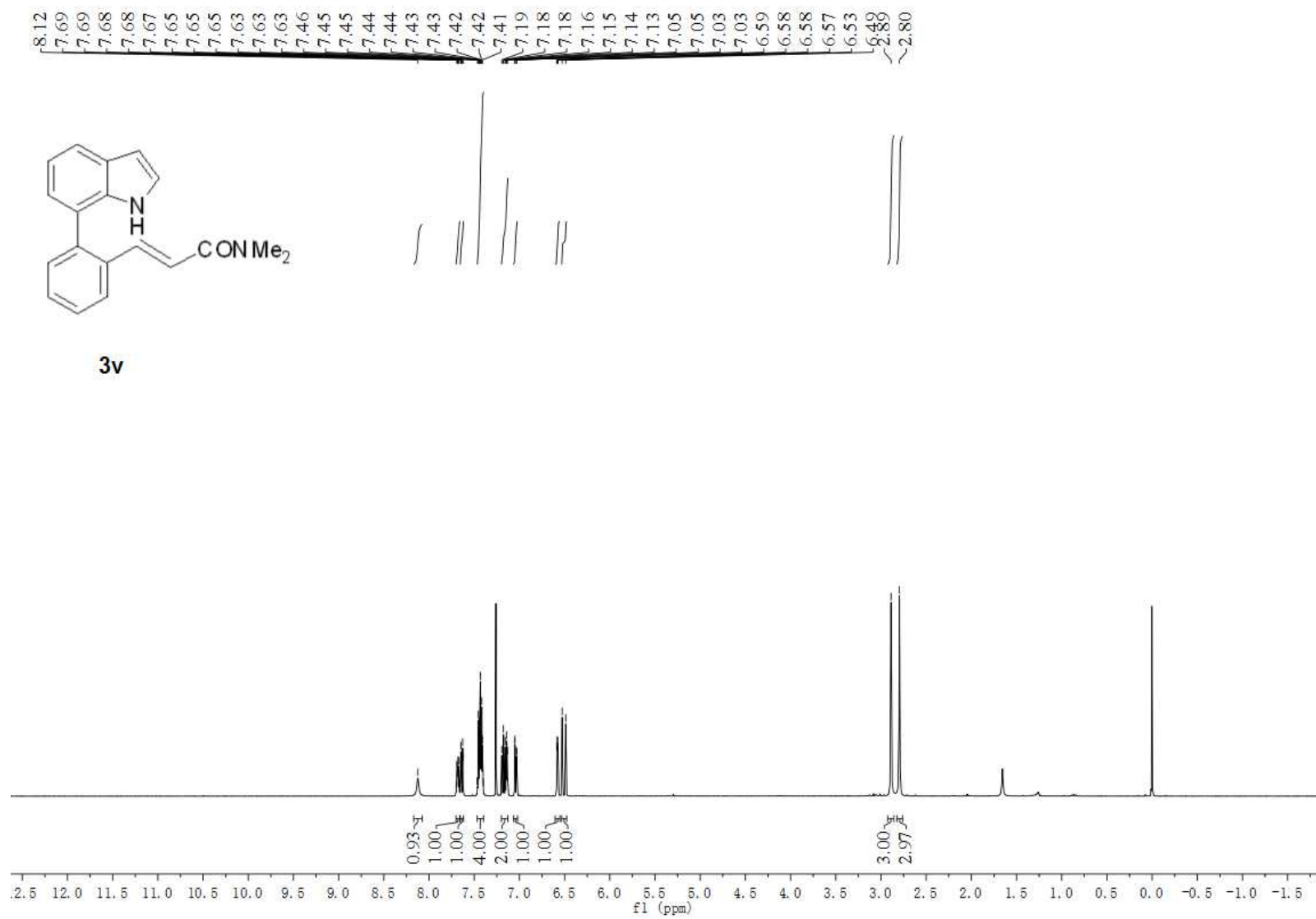


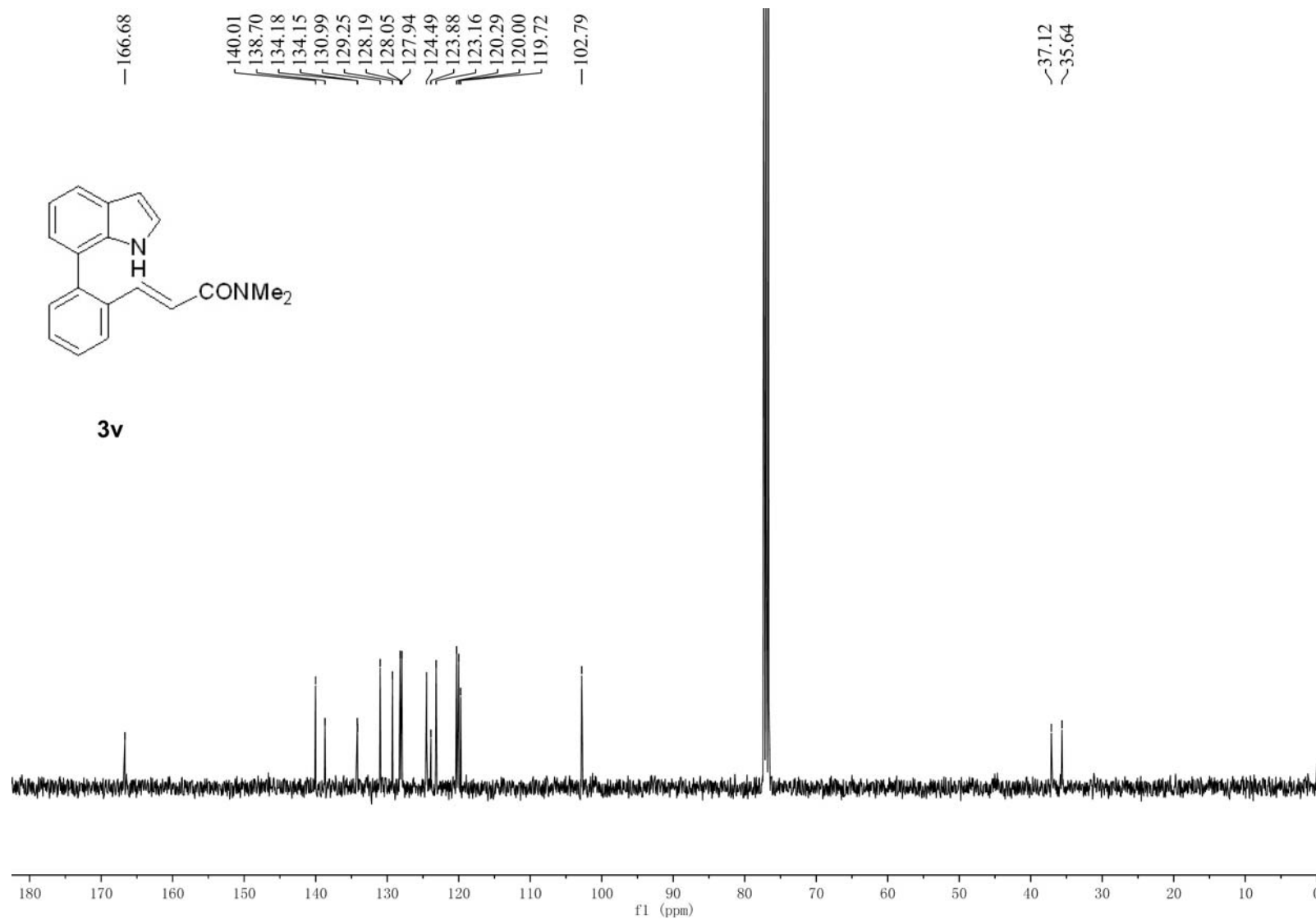


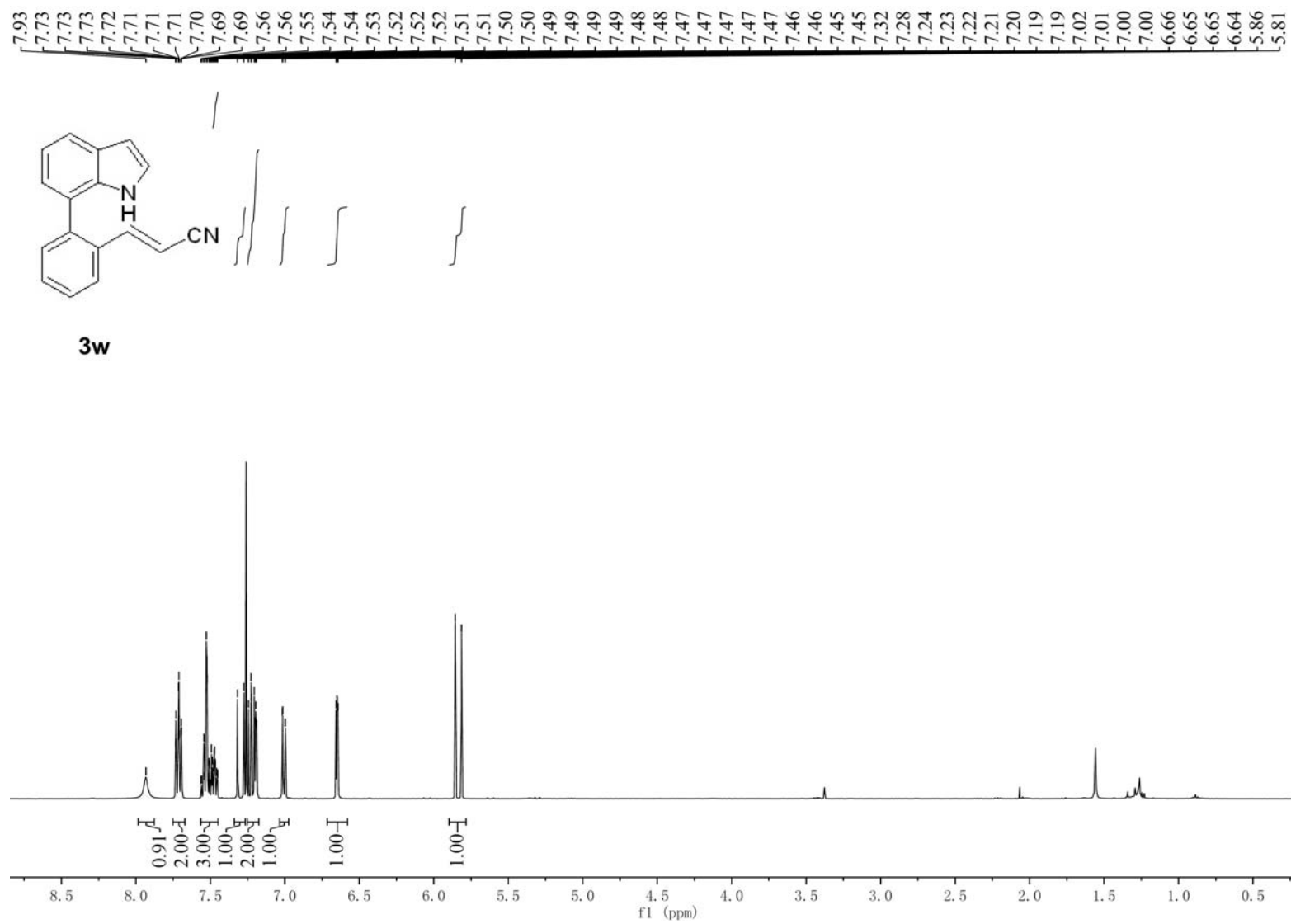


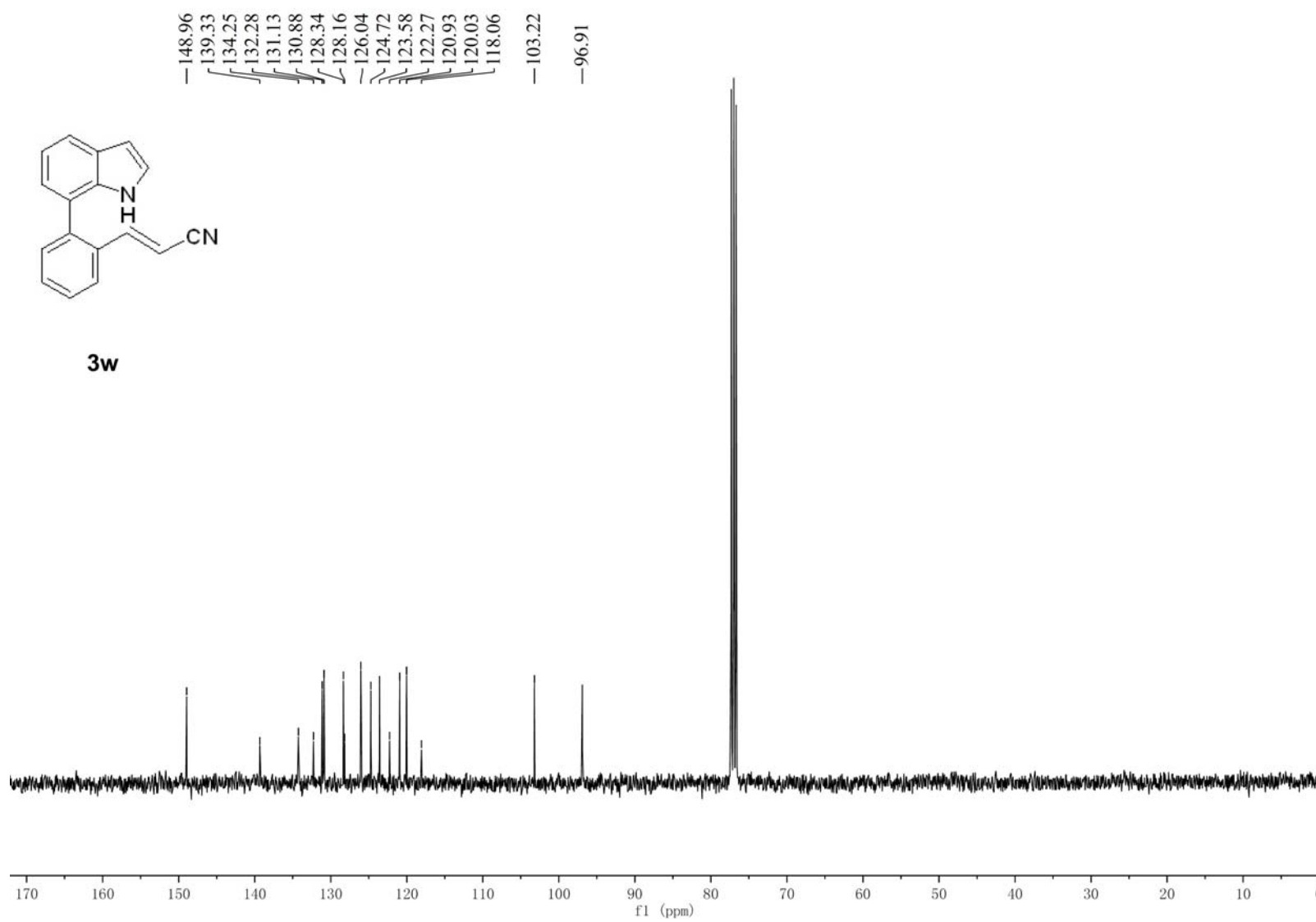


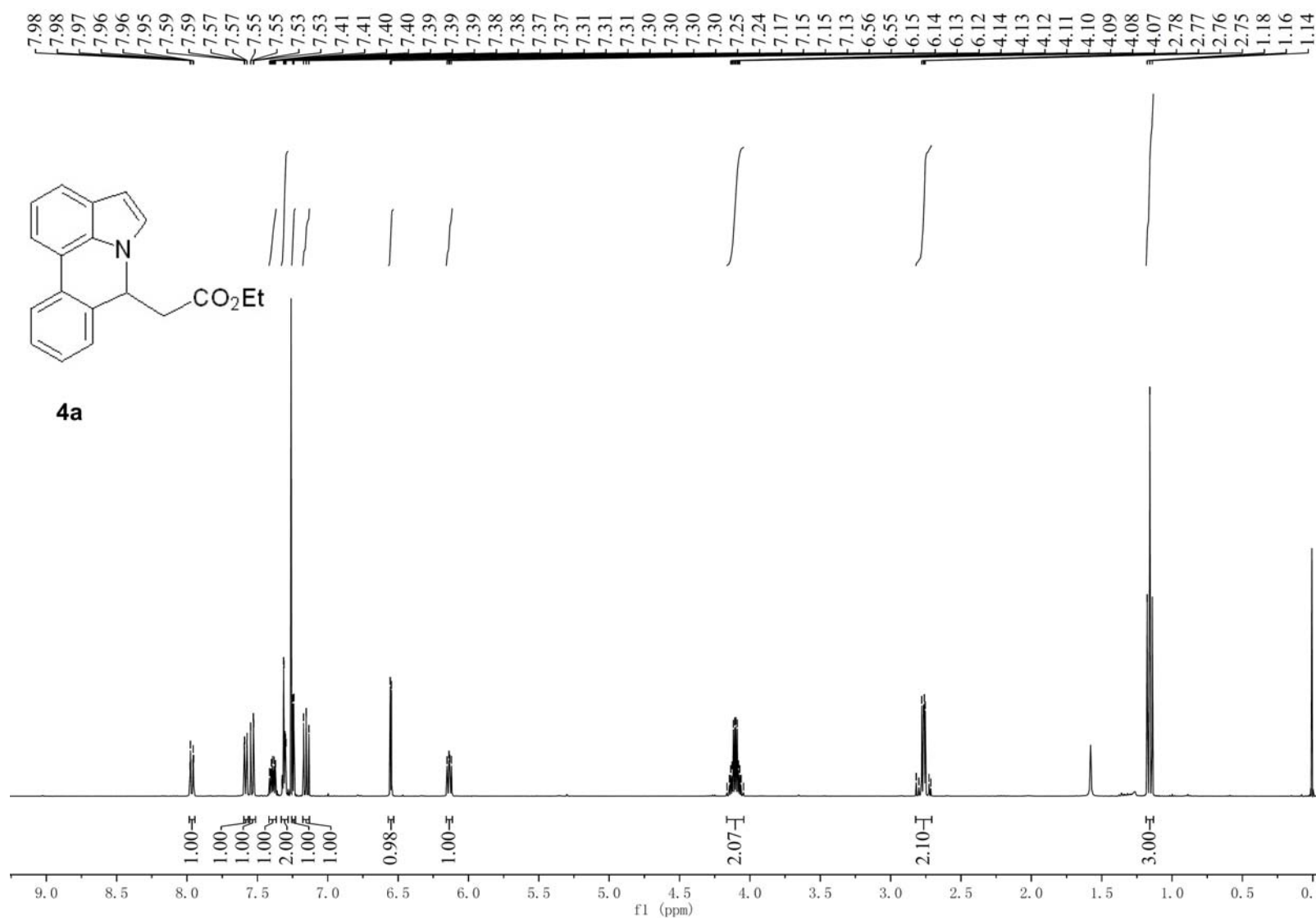


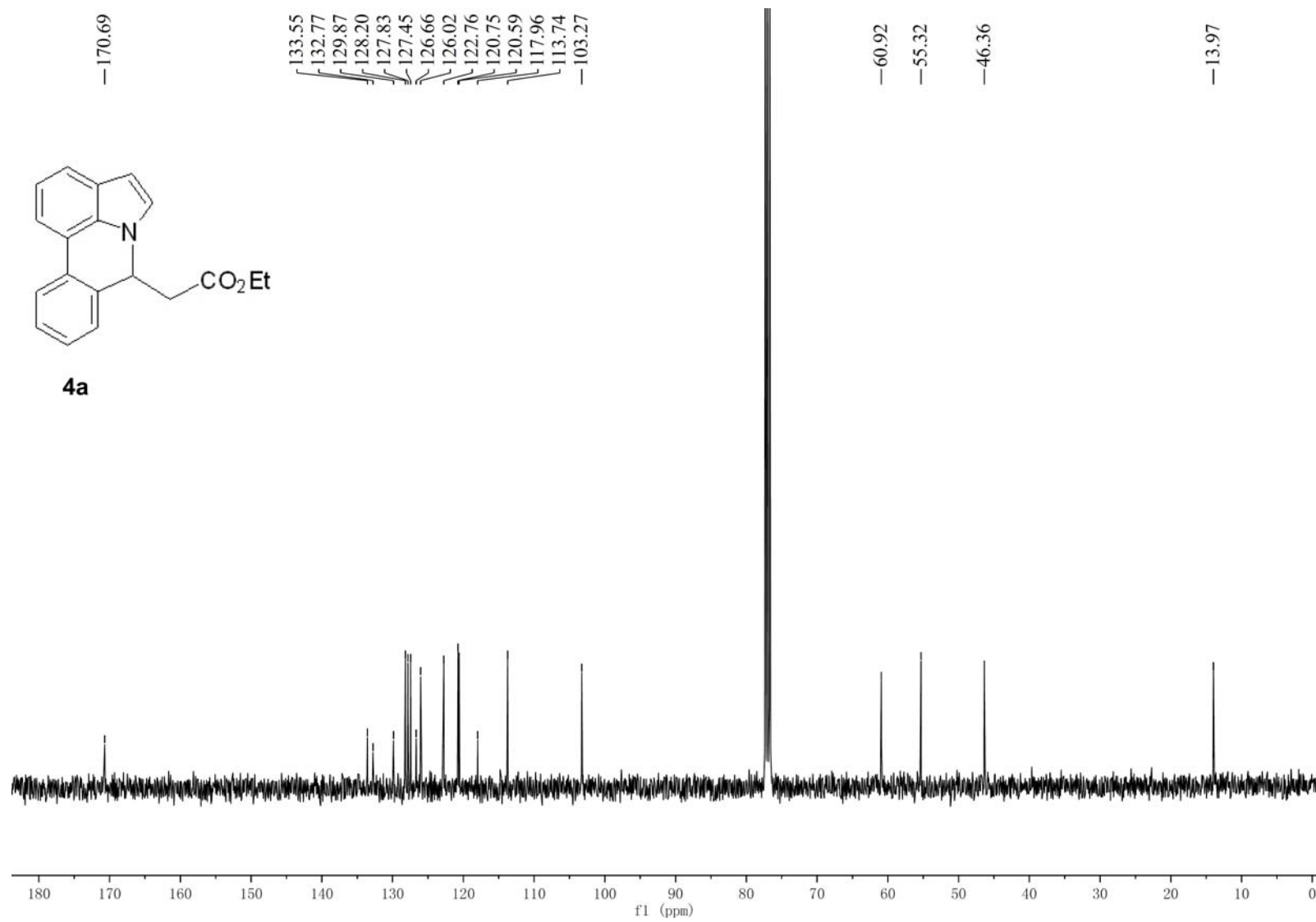


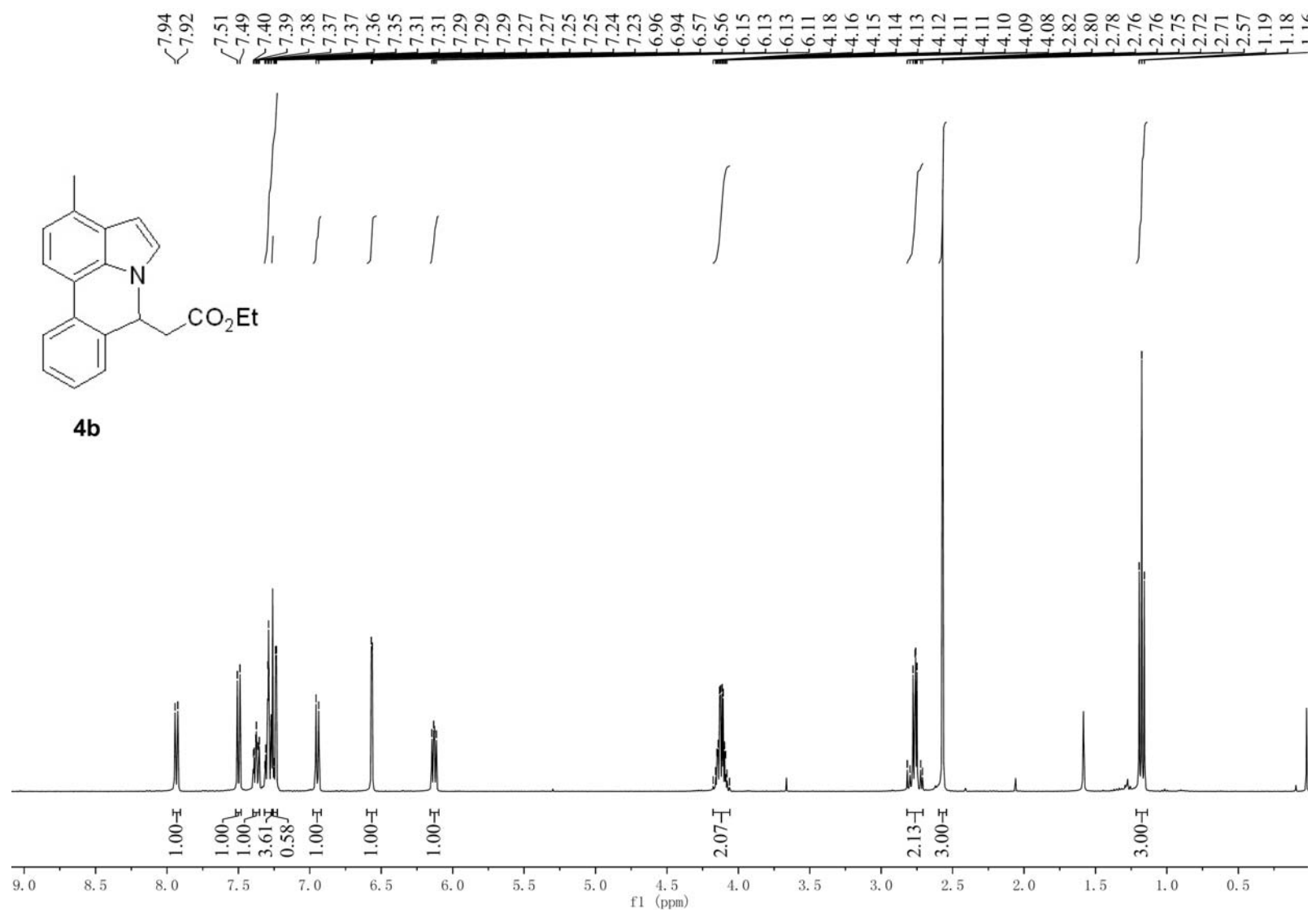


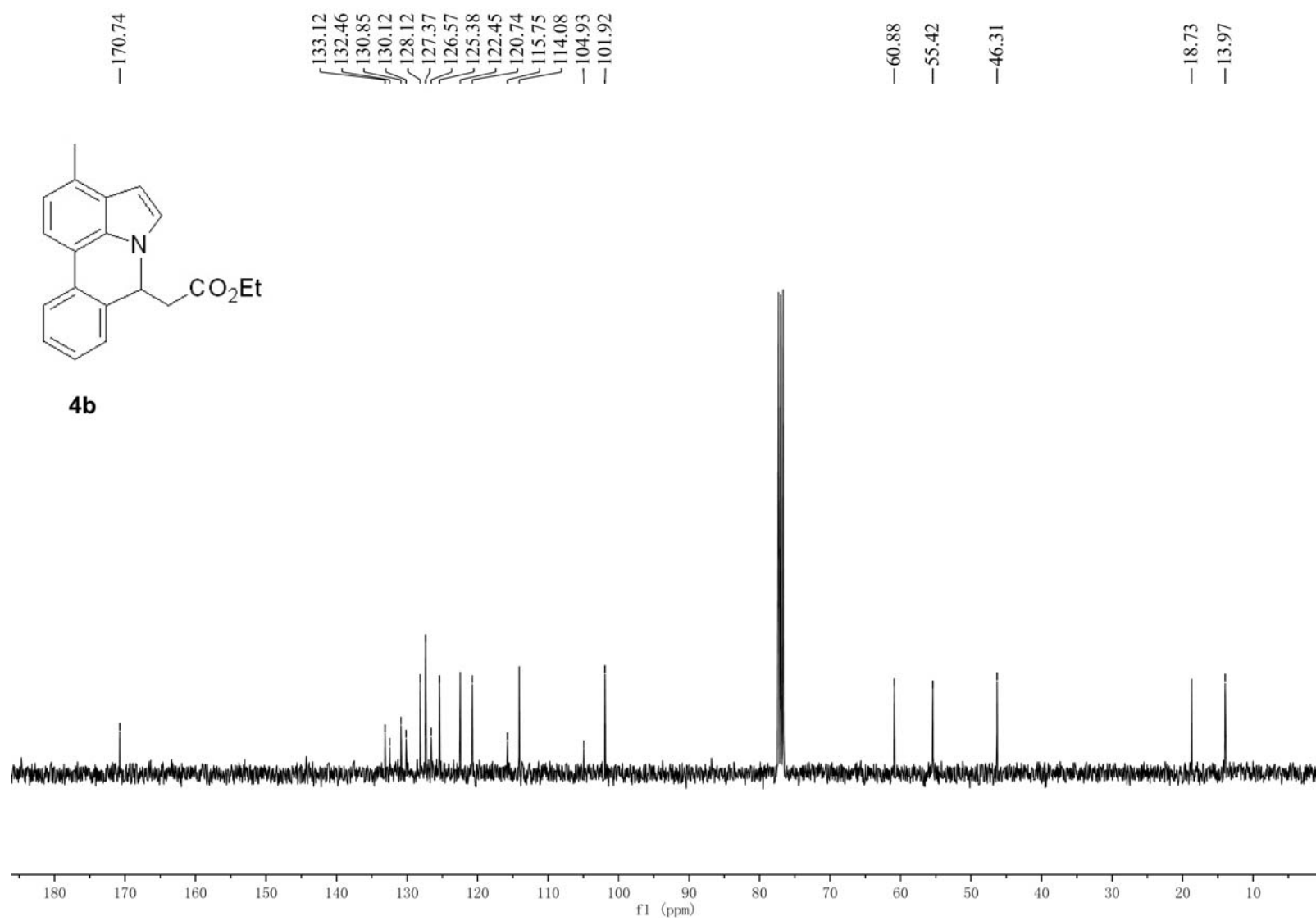


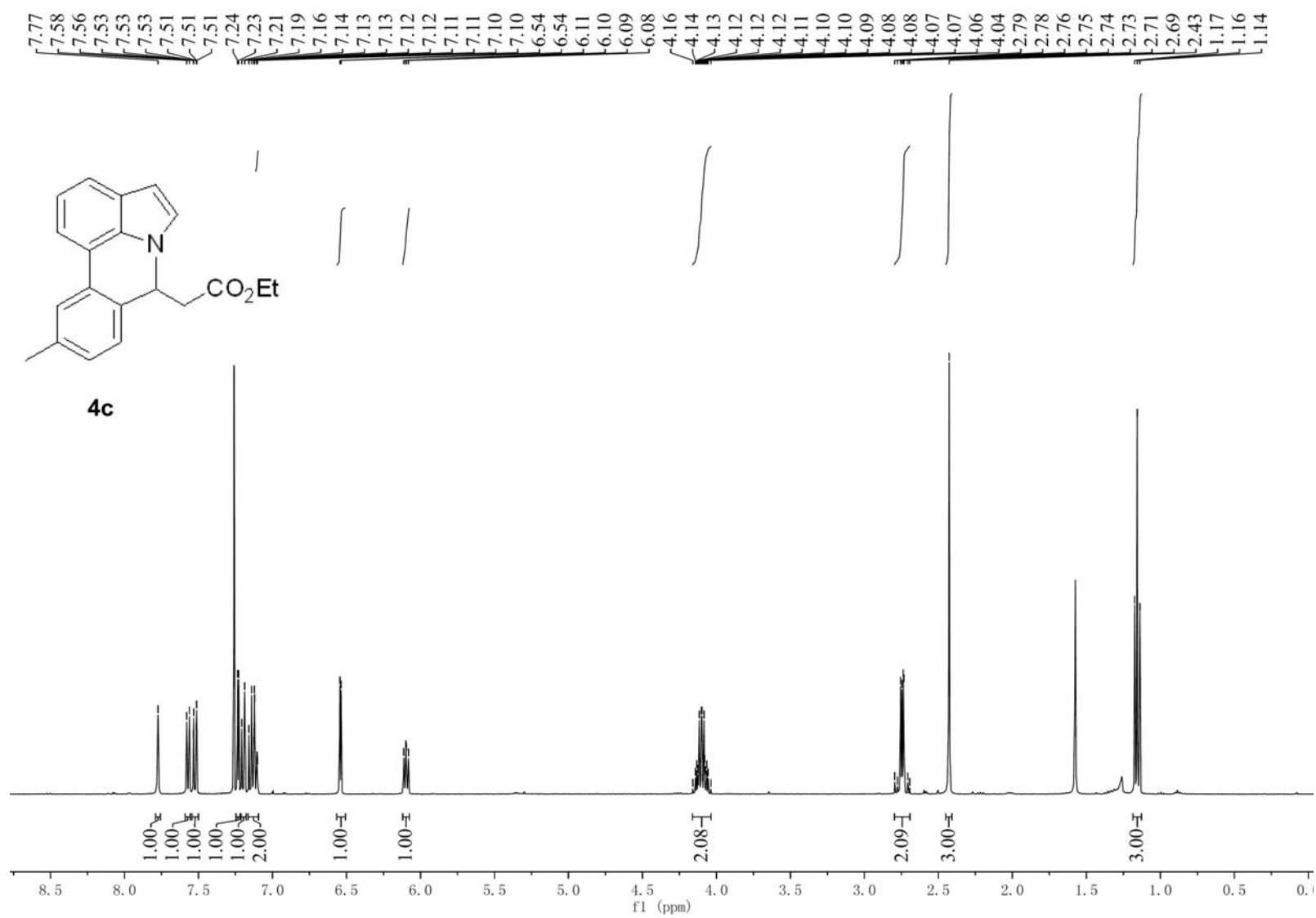


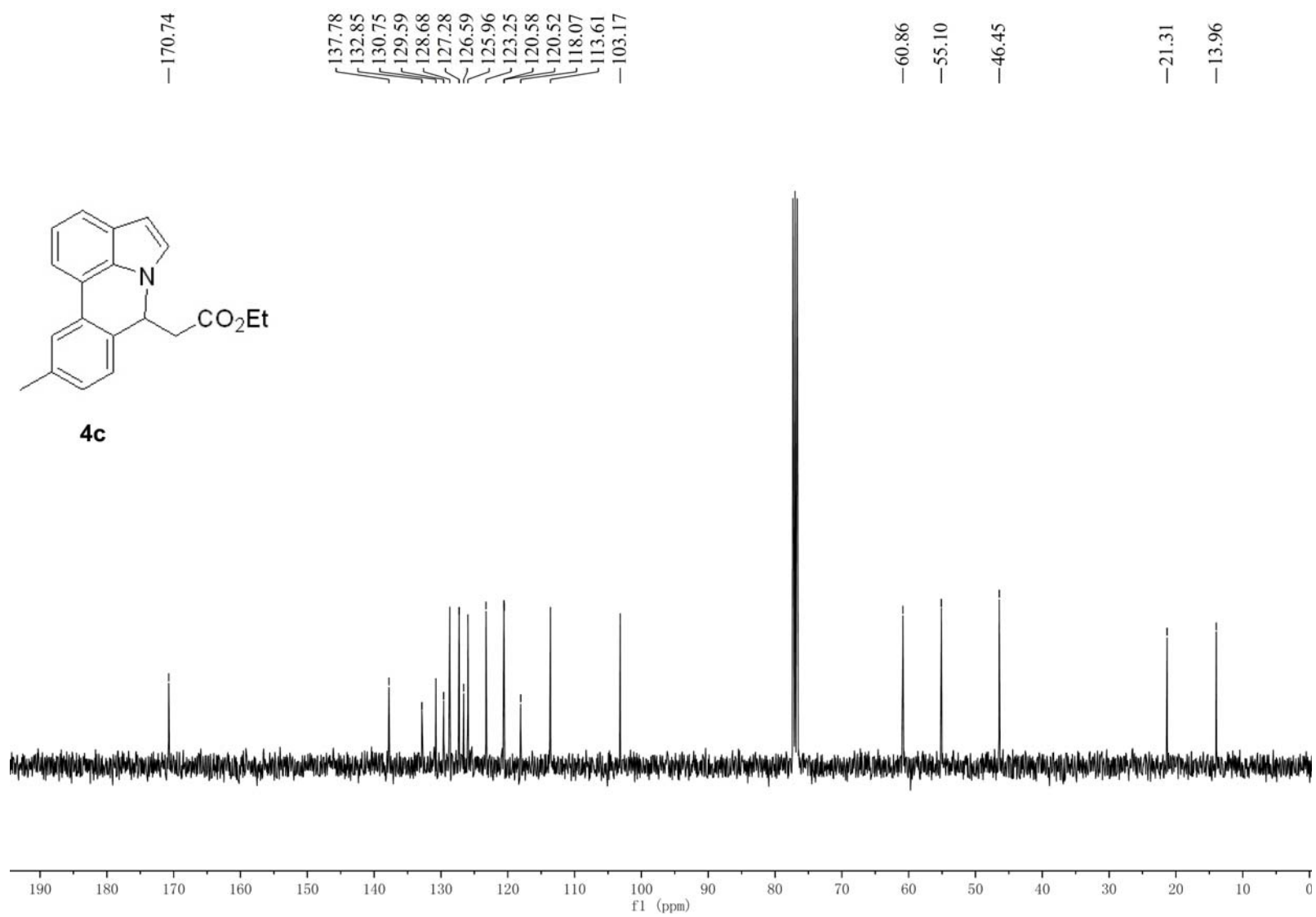


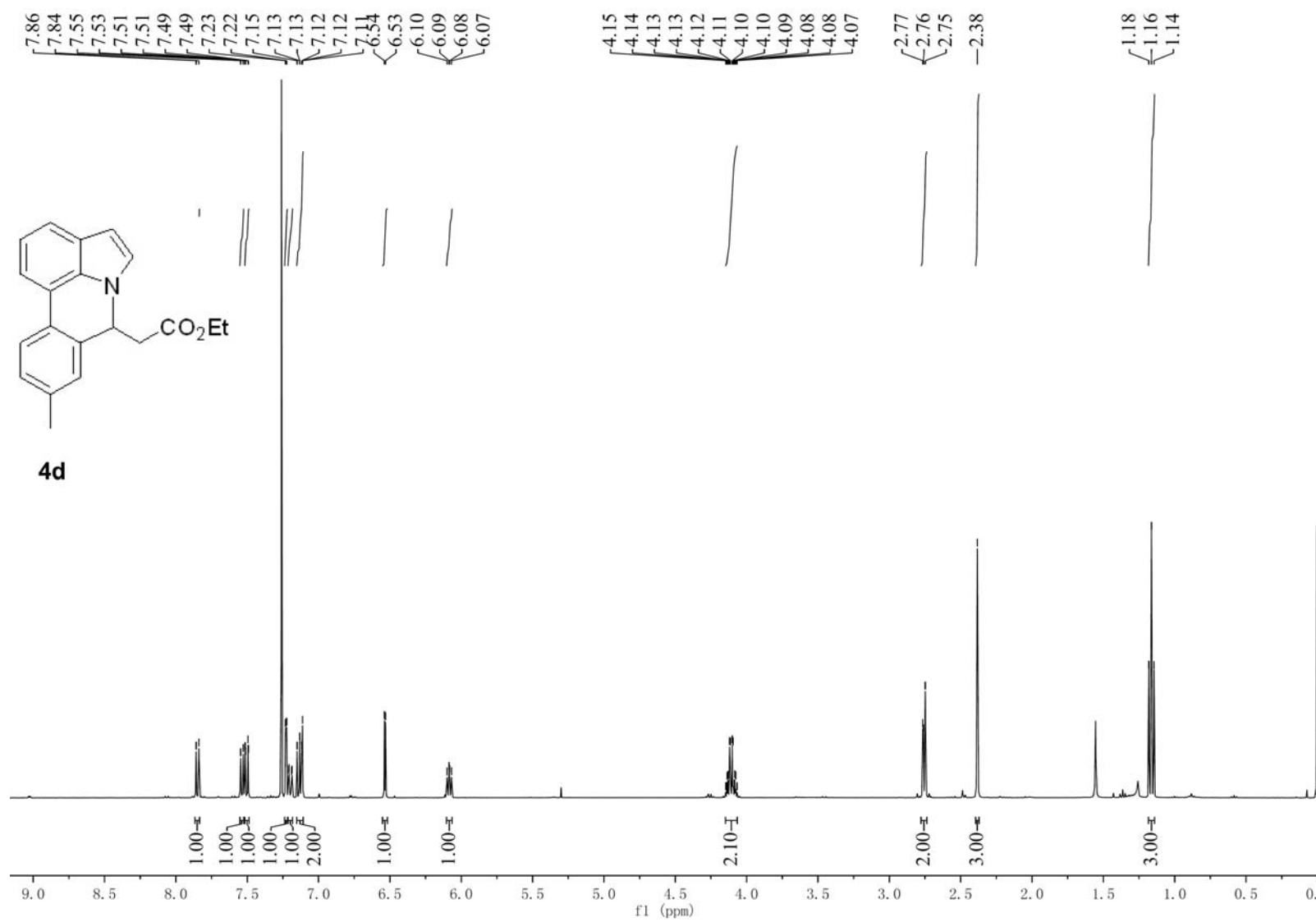


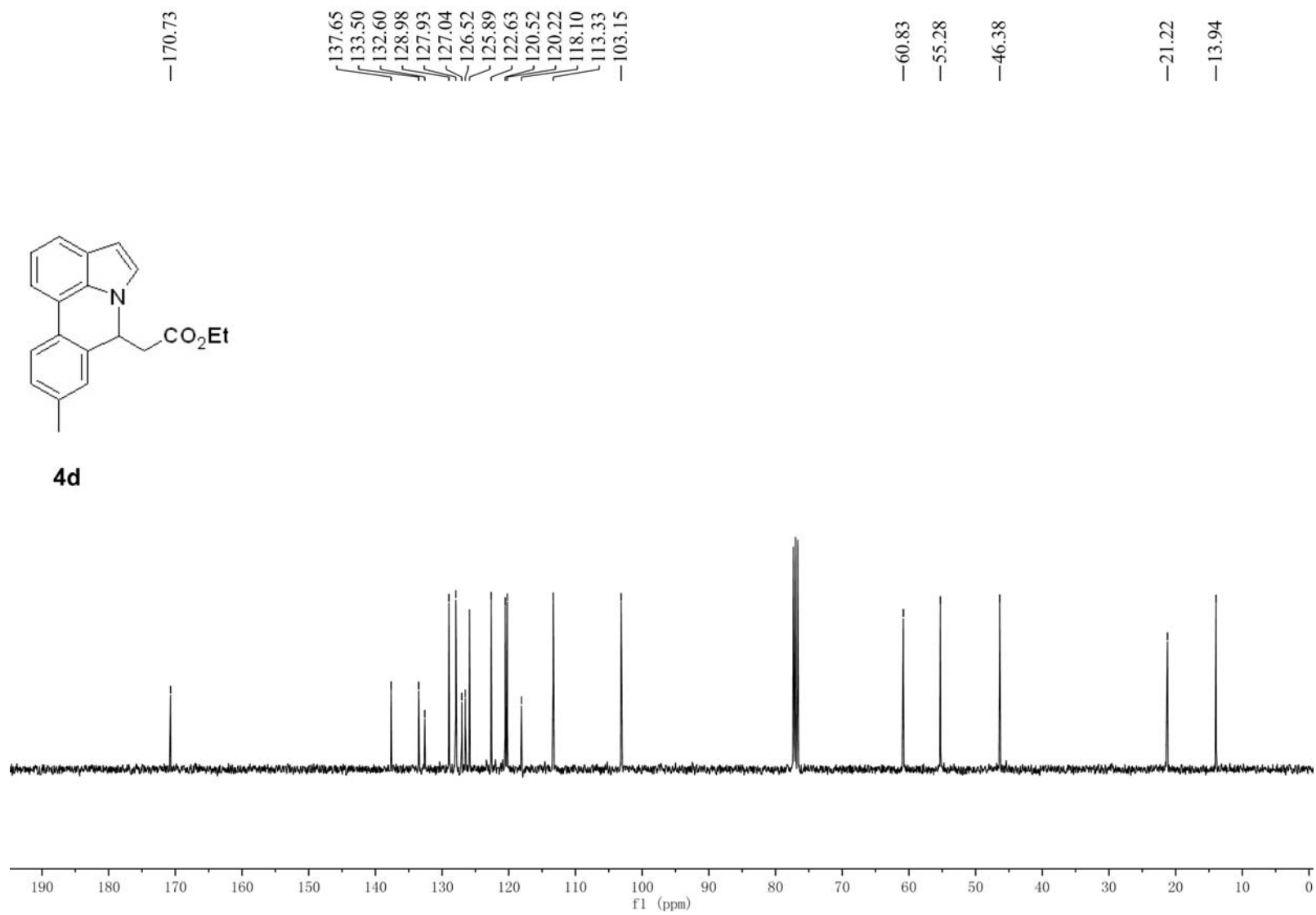


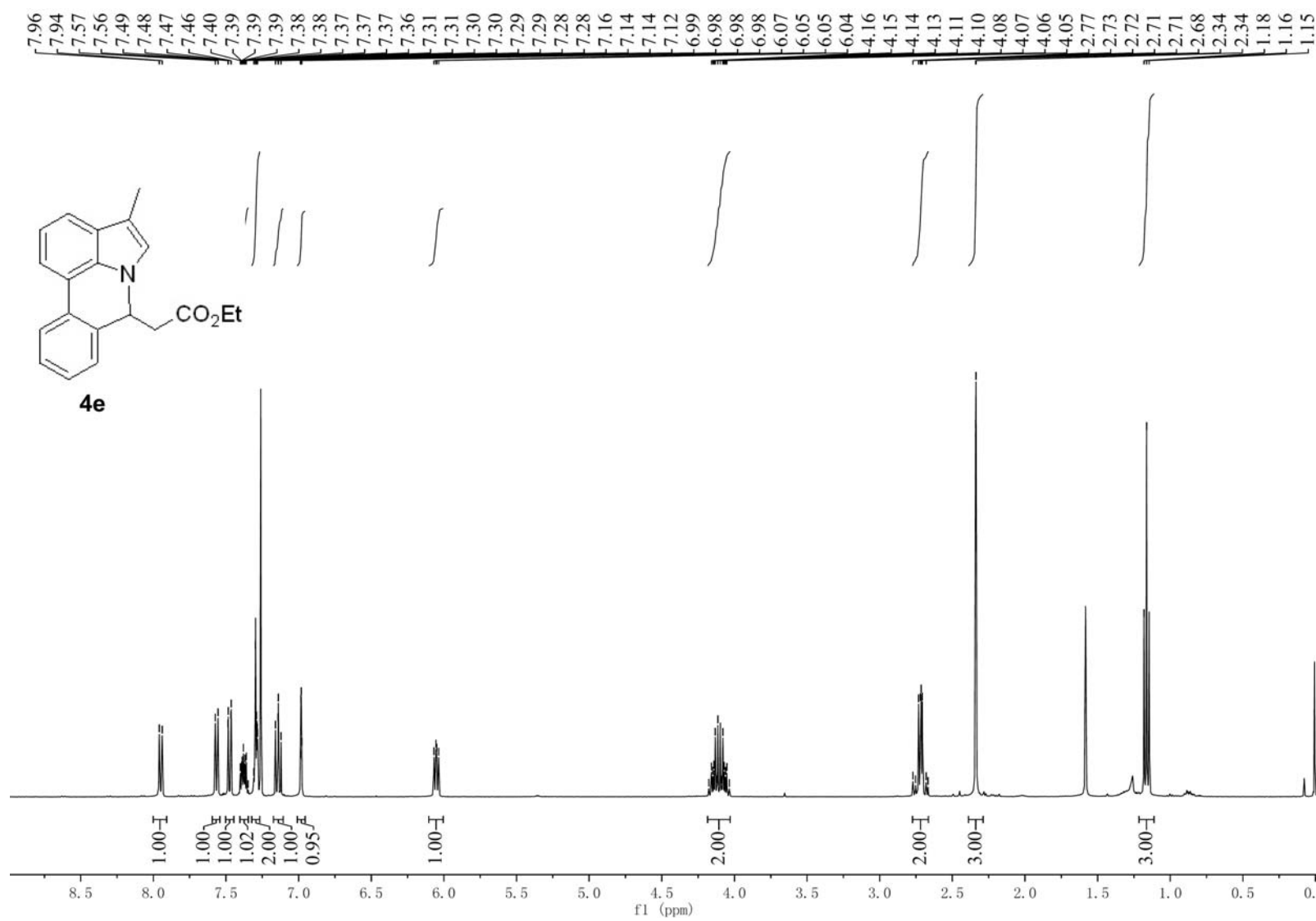


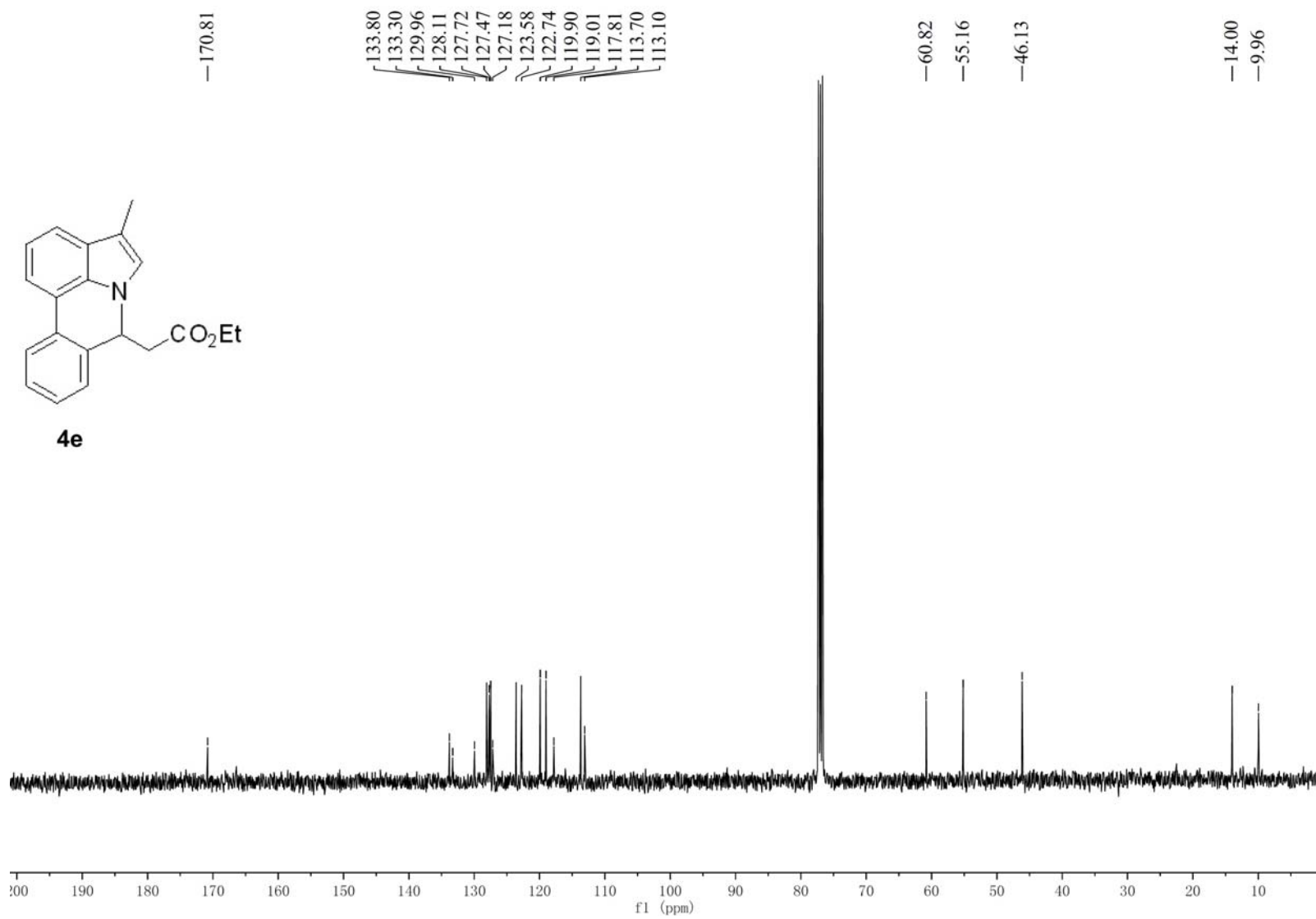


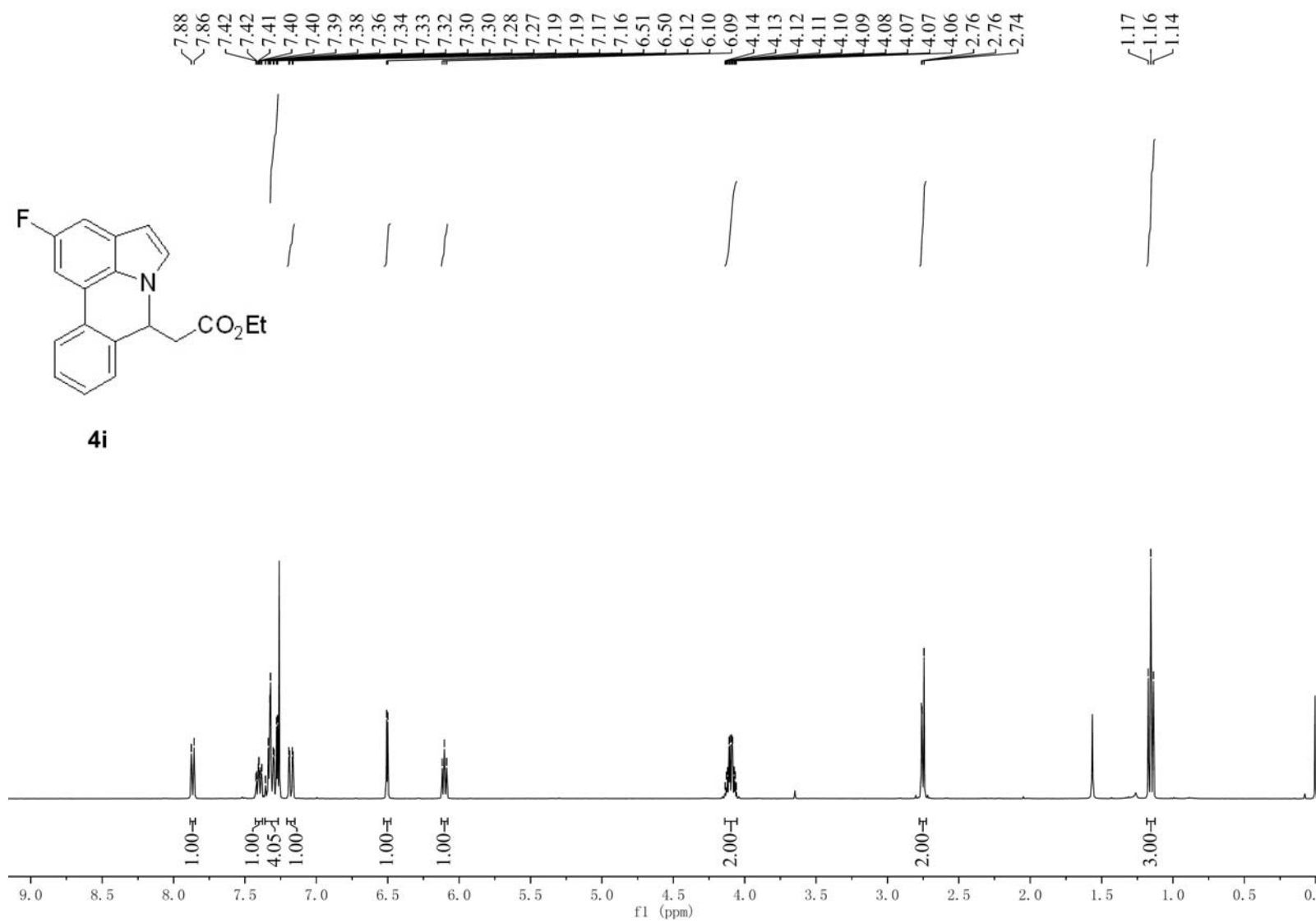


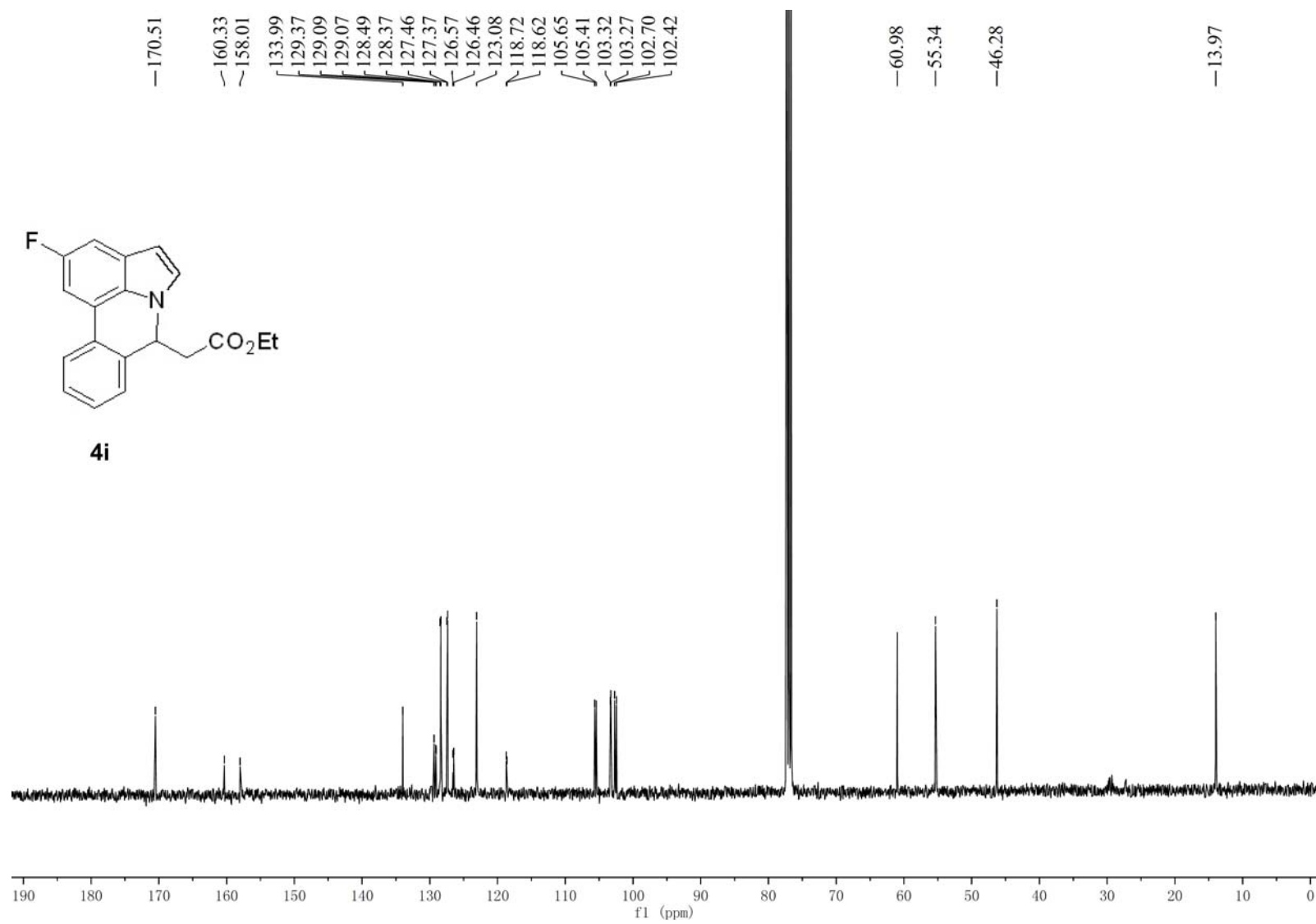


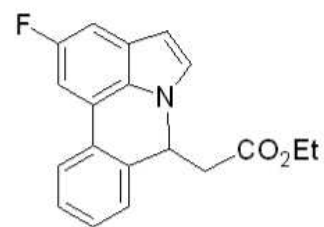












4i

123.44
123.46
123.49

