

Supporting Information

Facile synthesis of ordered mesoporous zinc alumina catalysts and
dehydrogenation behavior

Ming Cheng ^{a,b}, Huahua Zhao ^a, Jian Yang ^a, Jun Zhao ^a, Liang Yan ^a, Huanling Song
^{a,*}, Lingjun Chou ^{a,c,*}

^a State Key Laboratory for Oxo Synthesis and Selective Oxidation, Lanzhou Institute
of Chemical Physics (LICP), Chinese Academy of Sciences, Lanzhou 730000, PR
China

^b University of Chinese Academy of Sciences, Beijing 100049, PR China

^c Suzhou Research Institute of LICP, Chinese Academy of Sciences, Suzhou 215123,
PR China

*Corresponding author. *E-mail address*: ljchou@licp.cas.cn (Lingjun Chou), Tel: +86
931 4968 066, Fax: +86 931 4968 129; songhl@licp.cas.cn (Huanling Song), Tel:
+86 931 4968 066, Fax: +86 931 4968 129;

S1. NH₃-TPD analysis

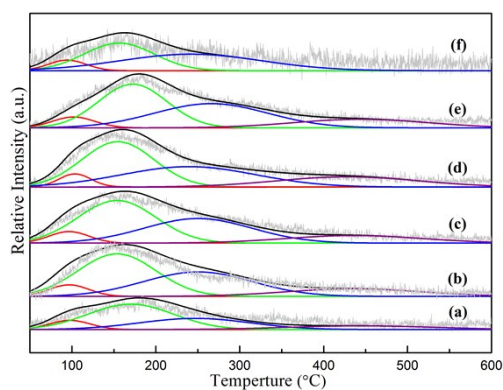


Fig. S1. The NH₃-TPD profiles of the as-synthesized xZn/Al₂O₃ catalysts and Al₂O₃: (a) Al₂O₃; (b) 3%Zn/Al₂O₃; (c) 5%Zn/Al₂O₃; (d) 7%Zn/Al₂O₃; (e) 10%Zn/Al₂O₃; (f) 15%Zn/Al₂O₃.

S2. NH₃-TPD profiles

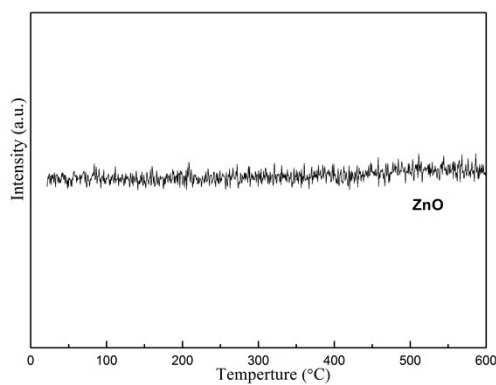


Fig. S2. The NH₃-TPD profile of ZnO.

S3. The catalytic dehydrogenation of isobutane over the ordered mesoporous Al₂O₃ and commercial ZnO.

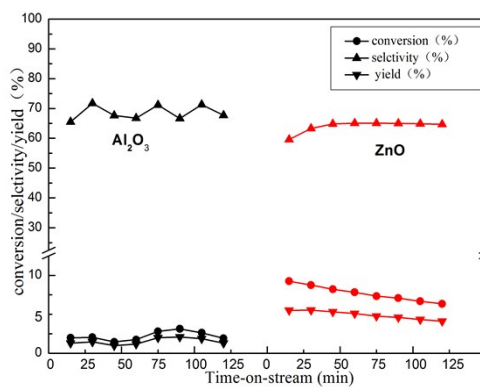


Fig. S3. The catalytic dehydrogenation of isobutane over the ordered mesoporous Al₂O₃ and commercial ZnO. Reaction condition: T = 580 °C, GHSV = 300 h⁻¹.

S4. Nitrogen adsorption–desorption analysis

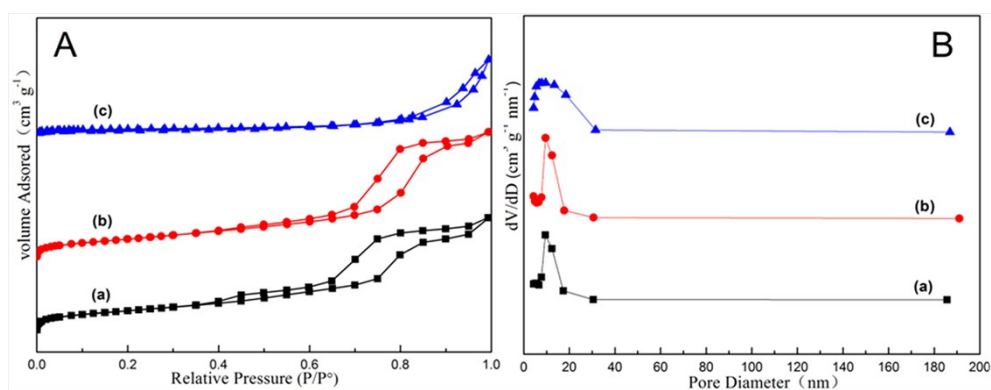


Fig. S4. The nitrogen adsorption–desorption analysis of the catalysts: (a) the spent 10%Zn/Al₂O₃; (b) the fifth regenerated 10%Zn/Al₂O₃; (c) the spent 15%Zn/Al₂O₃.