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Supporting Information

An efficient synthesis of new imidazo[1,2-*a*]pyridine-6-carbohydrazide and pyrido[1,2-*a*]pyrimidine-7-carbohydrazide derivatives *via* a five-component cascade reaction

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Experimental Section

General remarks:

Melting points were measured on an Electrothermal 9100 apparatus. Mass spectra were recorded with an Agilent 5975C VL MSD with Triple-Axis Detector operating at an ionization potential of 70 eV. ¹H and ¹³C NMR spectra were measured (DMSO) with a Bruker DRX-300 AVANCE spectrometer at 300 and 75 MHz, respectively. IR spectra were recorded on a Bruker Tensor 27, \bar{v} in cm⁻¹. All NMR spectra at room temperature were determined in DMSO-*d*₆. Chemical shifts are reported in parts per million (δ) downfield from an internal tetramethylsilane reference. Coupling constants (*J* values) are reported in hertz (Hz), and spin multiplicities are indicated by the following symbols: s (singlet), d (doublet), t (triplet), q (quartet), m (multiplet). All chemicals were purchased from Merck or Aldrich and were used without further purification.



Figure 1. Structure of all products 6a-q.

The structures of all products **6a-q** were deduced from their IR, mass, ¹H NMR, and ¹³C NMR spectra (see the following images).

The ¹H and ¹³C NMR spectra are taken in DMSO- d_6 , but some of the products are slightly soluble in the solvent therefore have no clear carbon spectra such as **6g**, **6h**, **6m**, **6o**, **6p**, **6q**.



¹H NMR of 6a

¹³C NMR of 6a

IR of 6a

¹H NMR of 6b

¹³C NMR of 6b

IR of 6b

MS of 6b

¹H NMR of 6c

IR of 6c

¹H NMR of 6d

¹³C NMR of 6d

MS of 6d

¹H NMR of 6e

IR of 6e

IR of 6f

¹³C NMR of 6g

¹H NMR of 6h

¹³C NMR of 6h

¹H NMR of 6i

¹³C NMR of 6i

IR of 6i

MS of 6i

¹H NMR of 6j

¹³C NMR of 6j

IR of 6j

MS of 6j

¹H NMR of 6k

¹³C NMR of 6k

¹H NMR of 6l

¹³C NMR of 6l

IR of 6l

MS of 6l

¹H NMR of 6m

¹³C NMR of 6m

¹H NMR of 6n

¹³C NMR of 6n

¹³C NMR of 60

IR of 60

MS of 60

¹³C NMR of 6p

¹H NMR of 6q

¹³C NMR of 6q

MS of 6q