

Synthesis of $\text{Ni}_{4.5}\text{Fe}_{4.5}\text{S}_8/\text{Ni}_3\text{S}_2$ Film on Ni_3Fe Foam as an Excellent Electrocatalyst for Oxygen Evolution Reaction

Shili Qin,^a Jinlong Lei,^a Yun Xiong,^b Xiaohu Xu,^c Xinhua Geng^{*a} and Jiahai Wang^{*a}

^a Department of Chemistry and Chemical Engineering, Guangzhou Key Laboratory for Environmentally Functional Materials and Technology, Guangzhou University, Guangzhou 510006, P.R. China.

^b Wuhan Economic and Technological Development Zone, Wuhan HydraV Fuel Cell Tech. Co., Ltd, Wuhan 430056, P.R. China.

^c Key Laboratory of Spectral Measurement and Analysis of Shanxi Province, Shanxi Normal University, Linfen 041004, P.R. China.

E-mail: ccgxinh1230@gzhu.edu.cn; jiahaiwang@gzhu.edu.cn

Tel: +86-18816801579

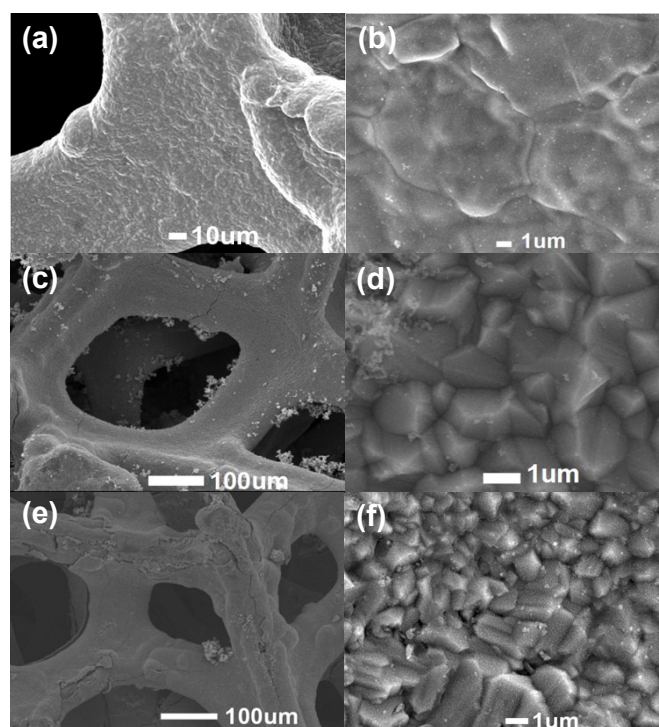


Fig. S1 SEM images of (a, b) Ni_3Fe foam. (c, d) $\text{Ni}_{4.5}\text{Fe}_{4.5}\text{S}_8/\text{Ni}_3\text{S}_2$ | Ni_3Fe sample. (e, f) $\text{Ni}_{4.5}\text{Fe}_{4.5}\text{S}_8/\text{Ni}_3\text{S}_2$ | Ni_3Fe sample after 1000 CV cycles.

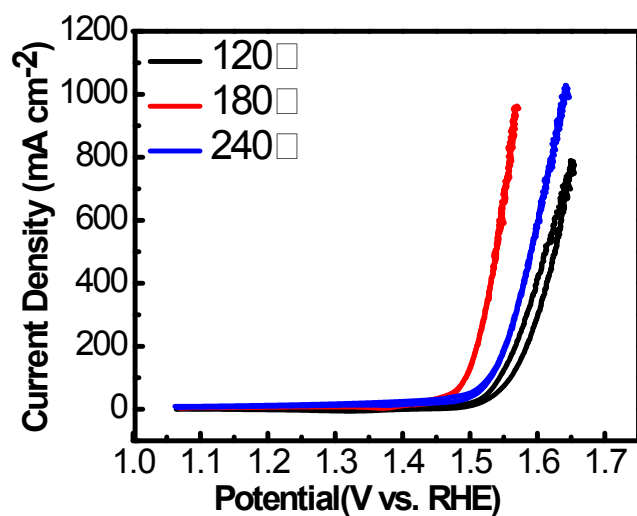


Fig. S2 The OER of $\text{Ni}_{4.5}\text{Fe}_{4.5}\text{S}_8/\text{Ni}_3\text{S}_2$ | Ni_3Fe prepared in different temperatures.

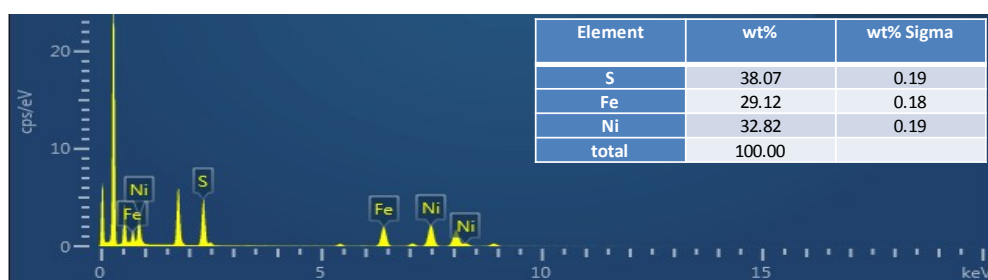


Fig. S3 EDS pattern of $\text{Ni}_{4.5}\text{Fe}_{4.5}\text{S}_8/\text{Ni}_3\text{S}_2$ | Ni_3Fe sample.

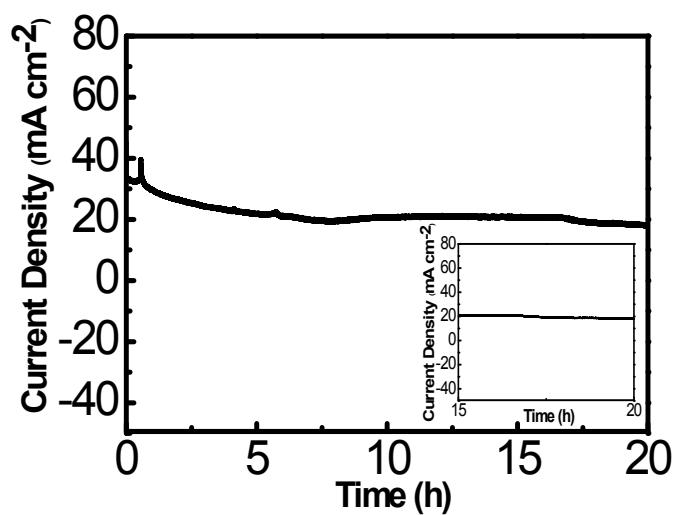


Fig.S4 Chronoamperometric durability test at a constant overpotential of 230 mV.

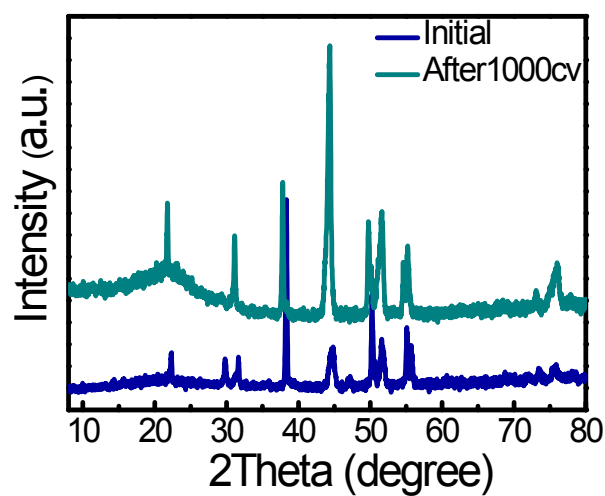


Fig. S5 XRD pattern of $\text{Ni}_{4.5}\text{Fe}_{4.5}\text{S}_8/\text{Ni}_3\text{S}_2 \mid \text{Ni}_3\text{Fe}$ and after 1000 CV cycles.

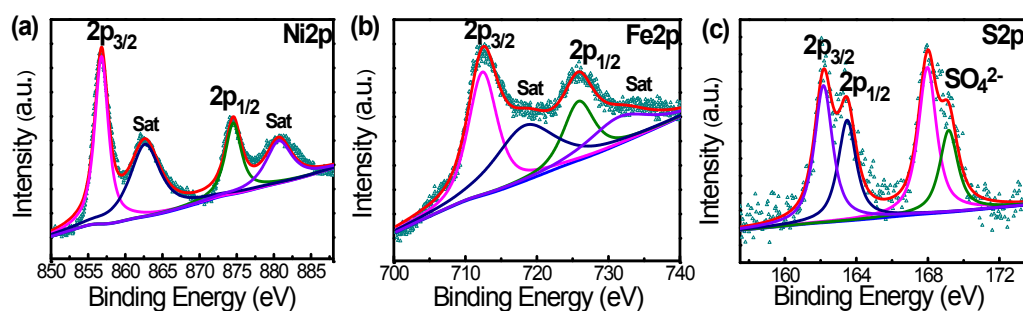


Fig. S6 XPS spectra for (a) $\text{Ni}2p$. (b) $\text{Fe}2p$. (c) $\text{S}2p$ of $\text{Ni}_{4.5}\text{Fe}_{4.5}\text{S}_8/\text{Ni}_3\text{S}_2 \mid \text{Ni}_3\text{Fe}$ after 1000 CV.

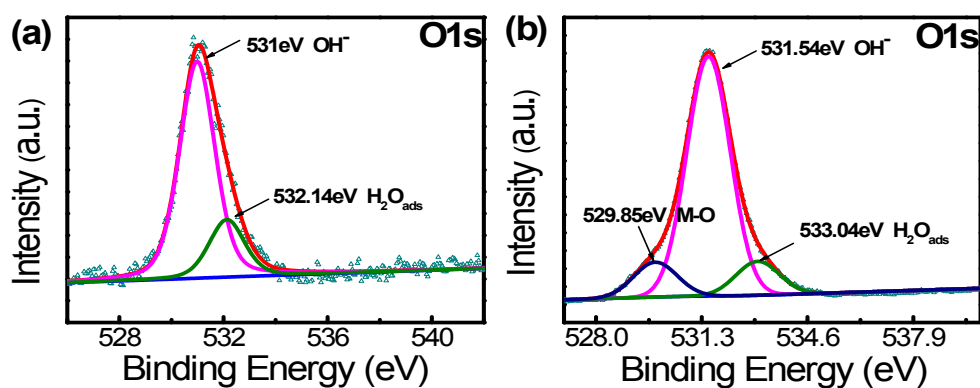


Fig. S7 XPS spectra for $\text{O}1s$ of (a) $\text{Ni}_{4.5}\text{Fe}_{4.5}\text{S}_8/\text{Ni}_3\text{S}_2 \mid \text{Ni}_3\text{Fe}$. (b) After 1000 CV.

Table S1 A comparison of the OER overpotentials for the reported electrocatalysts.

Catalysts	η at 10 mA cm ⁻² (mV)	η at 100 mA cm ⁻² (mV)	Tafel slope (mV dec ⁻¹)	Substrate	References
Fe- Ni ₃ S ₂	---	253	65.5	NF	J. Mater. Chem. A, 2015.3(46):p.23207-23212.
Fe- Ni ₃ S ₂	282	490	54	NiFe	Small, 2017. 13(18): p. 1604161.
N-Ni ₃ S ₂	---	330	70	NF	Adv. Mater.2017. 29(30).
NiFeS	65	189	119.4	NF	J. Mater. Chem. A, 2016 4(35):p.13499-13508.
NiFeS	---	180	55	NF	J. Mater. Chem. A, 2016.4(42):p.16394-16402.
MoS ₂ /Fe ₅ Ni ₄ S ₈	204	240	48	NiFe	Adv. Mater, 2018: p. e1803151.
NiFe-NC	271	340	48	---	ACS. Appl. Mater Interfaces.2017.9(48):p.41906-41915
NiFeOF	320	---	38	Stainless steel	ACS Catalysis.2017. 7(12): p. 8406-8412.
Fe-Ni(OH) ₂	235	---	51.5	NF	Chem. Commun (Camb). 2018. 54(5): p. 463-466.
Ni ₃ FeN	190	---	72	CC	ACS Appl. Mater Interfaces. 2018. 10(4): p. 3699-3706.
(NiFe)S ₂	260	410	56.2	---	Chem. Sci. 2018. 9(10): p. 2762-2767.
NiFe	283	---	53	NiFe	ACS Appl. Mater Interfaces. 2017. 9(34): p.28627-28634
Ni _x Co _{3-x} S ₄ /Ni ₃ S ₂	---	300	95	NF	Nano Energy. 2017. 35: p. 161-170.
Co-NiO/NiFe ₂ O ₄	186	220	35	NF	J. Mater. Chem. A, 2018. 6(1): p. 167-178.
Ni-Fe-P	271	---	53	---	ACS. Appl. Mater Interfaces.2017. 9(31): p. 26134-26142.
Ni ₂ P@NiFe-LDH	205	230	32	NF	Chem. Sci.2018. 9(5): p.1375-1384.
Ni _{4.5} Fe _{4.5} S ₈ /Ni ₃ S ₂	166	264	63.15	Ni ₃ Fe	This work