Electronic Supplementary Material (ESI) for RSC Advances. This journal is © The Royal Society of Chemistry 2019

Supplementary Information

Enhanced plasma-catalytic decomposition of toluene over Co-Ce

binary metal oxide catalysts with high energy efficiency

Zheng Bo, Jinhui Zhu, Shiling Yang, Huachao Yang*, Jianhua Yan and Kefa Cen

State Key Laboratory of Clean Energy Utilization, Institute for Thermal Power Engineering, College of Energy Engineering, Zhejiang University, Hangzhou, Zhejiang Province 310027, China

Correspondence should be addressed to Huachao Yang (Email: huachao@zju.edu.cn; Tel: 86 571 87953290)

Catalysts	m (g) 2	Q (L min⁻¹)	GHSV (mL g ⁻¹ h ⁻¹) ³	^c _{in} (ppm)	SIE (J L ⁻¹)	η _{toluene} (%)	S _{CO2} (%) Bc (%)	$\eta_{ m energy}$ (g kWh ⁻¹)	Ref
$Co_3O_4/Al_2O_3/nickel$ foam	-	0.1	-	50	500	96	~72 -	1.32	1
MnO _x /alumina/nickel foam	-	0.1	-	50	756	95.9	81	0.88	2
MnO _x /γ-Al ₂ O ₃	-	2	-	186	700	96	-	3.52	3
TiO_2/γ -Al $_2O_3/nikel$ foam	-	0.2	-	50	1527	94	99 -	0.42	4
Cu/OMS-2	0.2	0.06	18000	800	-	80	-	-	5
$MnO_x/Al_2O_3/nickel foam$	0.45	0.2	26666.67	100	651	87	37 48	1.84	6
Ag/Mn-SBA-15	0.2	0.1	30000	21	317	88	- 36	0.80	7
Co-MCM-41	0.1	0.2	120000	100	226	~100	- 75%	6.11	8
CeO ₂ -MnO _x	-	0.25	-	1500	24 W	95.94	90.73 -	3.45	9
MnO _x -MCM41	0.5	1	120000	110	500	99.4	73 99.5%	3.02	10
Co-Ce	0.2	0.5	150000	210	401	98.5	87.3 97.8	7.12	This work

Table S1. An overview on the decomposition of toluene with various catalysts in IPC¹.

¹ Conditions: room temperature; atmospheric pressure; IPC; DBD reactor.

² m: catalyst mass.

³ GHSV: gas hourly space velocity. GHSV (mL g–1 h–1) = 60000 × Q / m.

Sample	Crystallite size (nm) ¹	Specific surface area (m ² g ⁻¹)	Total pore volume (cm ³ g ⁻¹)	Average pore size (nm)
CoO _x	-	12.1	0.048	18.1
Co _{0.75} Ce _{0.25} O _x	9.6	56.0	0.138	4.7
$Co_{0.5}Ce_{0.5}O_x$	11.9	51.1	0.108	6.7
$Co_{0.25}Ce_{0.75}O_x$	12.6	46.0	0.124	5.2
CeO _x	14.2	25.6	0.077	7.3
used Co _{0.75} Ce _{0.25} O _x	10.7	53.7	0.131	5.1

Table S2. Physicochemical properties of Co-Ce catalysts.

 1 The crystallite size is calculated from the characteristic peak of CeO₂ (111) crystal face located at 2ϑ

= 28.7°.

 Table S3. XPS analysis of Co-Ce binary metal oxide catalysts.

Sample	Ce ³⁺ / (Ce ³⁺ + Ce ⁴⁺)	$O_{s} / (O_{s} + O_{l})$	
$Co_{0.75}Ce_{0.25}O_{x}$	19.2%	31.0%	
$Co_{0.5}Ce_{0.5}O_x$	16.2%	26.9%	
$Co_{0.25}Ce_{0.75}O_{x}$	12.7%	20.6%	
used Co _{0.75} Ce _{0.25} O _x	17.6%	29.1%	



Fig. S1 (a) O₃ and (b) NO₂ concentrations of Co-Ce catalysts as a function of SIE.



Fig. S2 (a) The durability and (b) reusability tests of the $Co_{0.75}Ce_{0.25}O_x$ catalyst at 298 J L⁻¹.



Fig. S3 (a) XRD and (b-d) XPS characterization of the used $Co_{0.75}Ce_{0.25}O_x$ catalyst.

References

- 1 Y. F. Guo, D. Q. Ye and K. F. Chen, *Plasma Chem. Plasma Process.*, 2006, **26**, 237-249.
- 2 Y. F. Guo, D. Q. Ye, K. F. Chen, J. C. He and W. L. Chen, *J. Mol. Catal. A: Chem.*, 2006, **245**, 93-100.
- 3 T. Zhu, J. Li, W. Liang and Y. Jin, J. Hazard. Mater., 2009, **165**, 1258-1260.
- H. B. Huang, D. Q. Ye, D. Y. C. Leung, F. D. Feng and X. J. Guan, *J. Mol. Catal. A: Chem.*, 2011,
 336, 87-93.
- 5 H. T. Quoc An, T. Pham Huu, T. Le Van, J. M. Cormier and A. Khacef, *Catal. Today*, 2011, **176**, 474-477.
- 6 H. B. Huang, D. Q. Ye and D. Y. C. Leung, *IEEE Trans. Plasma Sci.*, 2011, **39**, 576-580.
- X. X. Xu, P. T. Wang, W. C. Xu, J. L. Wu, L. M. Chen, M. L. Fu and D. Q. Ye, *Chem. Eng. J.*, 2016, 283, 276-284.
- 8 X. X. Xu, J. L. Wu, W. C. Xu, M. L. He, M. L. Fu, L. M. Chen, A. M. Zhu and D. Q. Ye, *Catal. Today*, 2017, **281**, 527-533.
- 9 B. W. Wang, C. M. Chi, M. Xu, C. Wang and D. Meng, *Chem. Eng. J.*, 2017, **322**, 679-692.
- 10 X. Yao, J. Zhang, X. Liang and C. Long, *Chemosphere*, 2018, **208**, 922-930.