

**Rationally designing coupling mechanism of physical
adsorption and chemical charge effect for high performance
Lithium sulfur batteries**

Guilin Feng^a, Xiaohong Liu^a, Yasai Wang^a, Zhenguo Wu^a, Chen Wu^a, Rong Li^a,
Yanxiao Chen^{a,*}, Xiaodong Guo^a, Benhe Zhong^a, Jianshu Li^b

^aSchool of Chemical Engineering, Sichuan University, Chengdu 610065, P. R. China.

*^bCollege of Polymer Science and Engineering, State Key Laboratory of Polymer Materials
Engineering, Sichuan University, Chengdu 610065, People's Republic of China.*

Corresponding Authors:

Tel: +86-28-85406702; Fax: +86-28-85406702;

E-mail address:

yxchen888@163.com

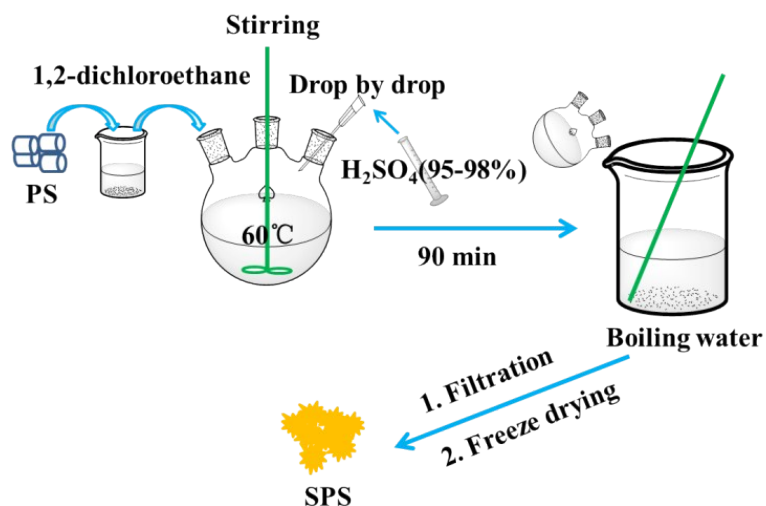


Fig.S1 The diagrammatic drawing of prepared SPS.

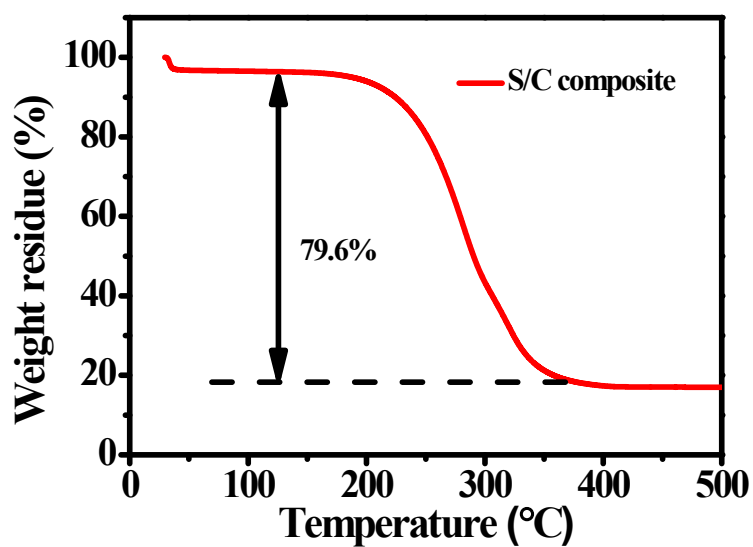


Fig.S2 TGA curve of S/C composite under Argon atmosphere

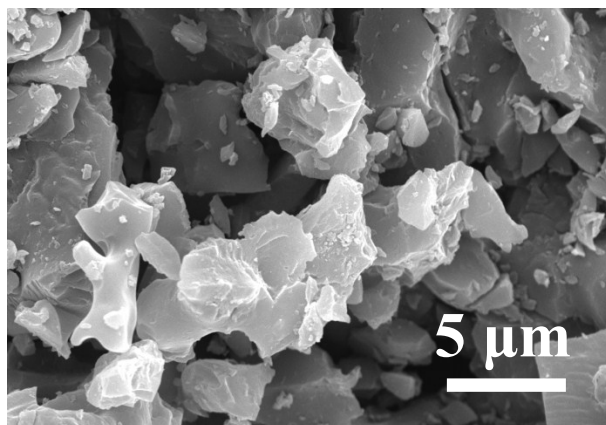


Fig.S3 SEM image of prepared NACC.

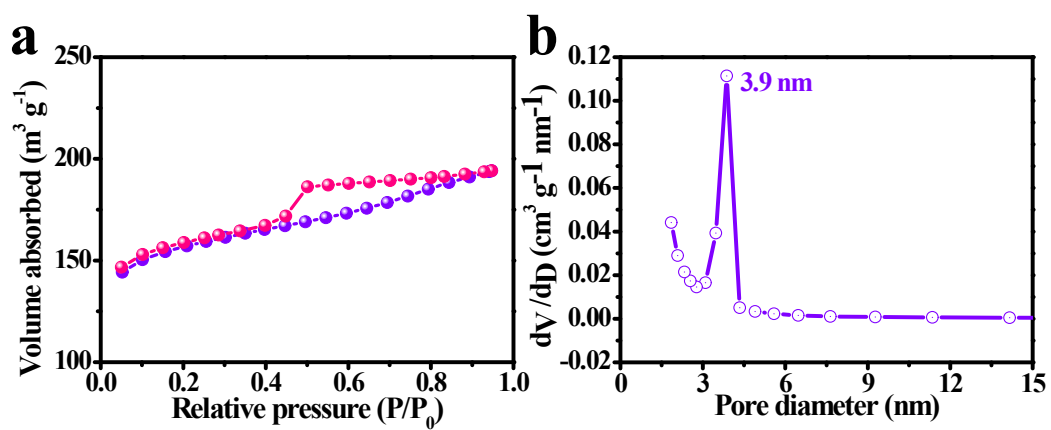


Fig.S4. (a) Nitrogen adsorption–desorption isotherms for prepared NACC, (b)

BJH pore size distribution profiles.

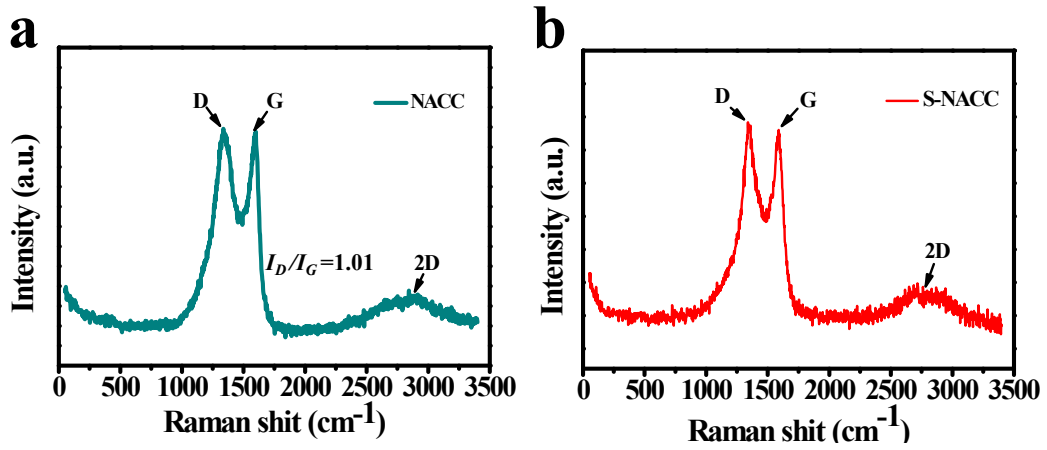


Fig.S5 The Raman spectra of (a) NACC, (b) S-NACC coated separator

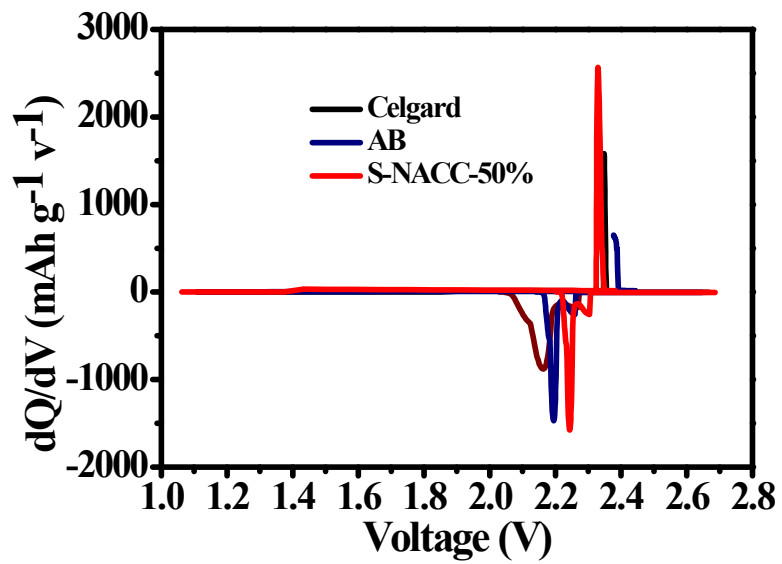


Fig.S6 The dQ/dV result of three different separators.

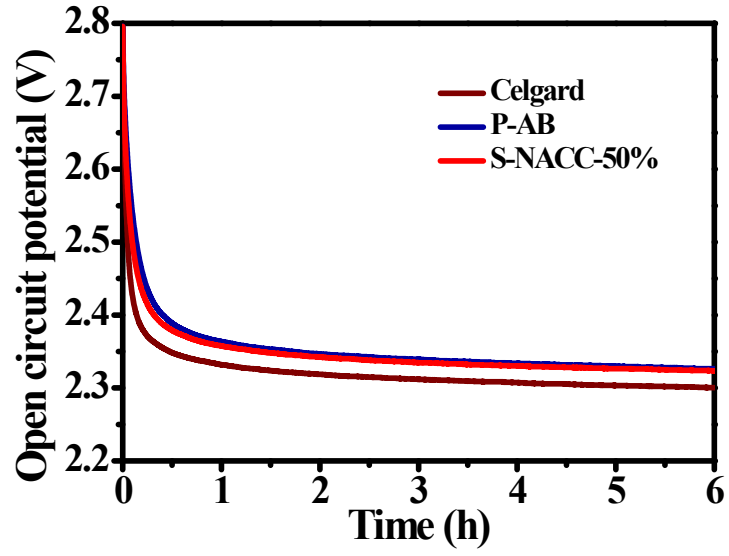


Fig.S7 Self-discharge behavior of Li-S batteries with different separators

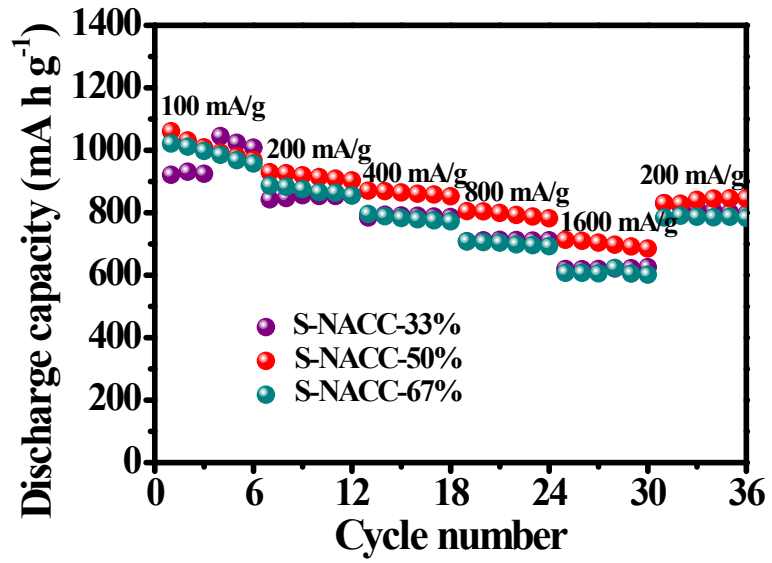


Fig.S8 The rate performance of different addition of NACC

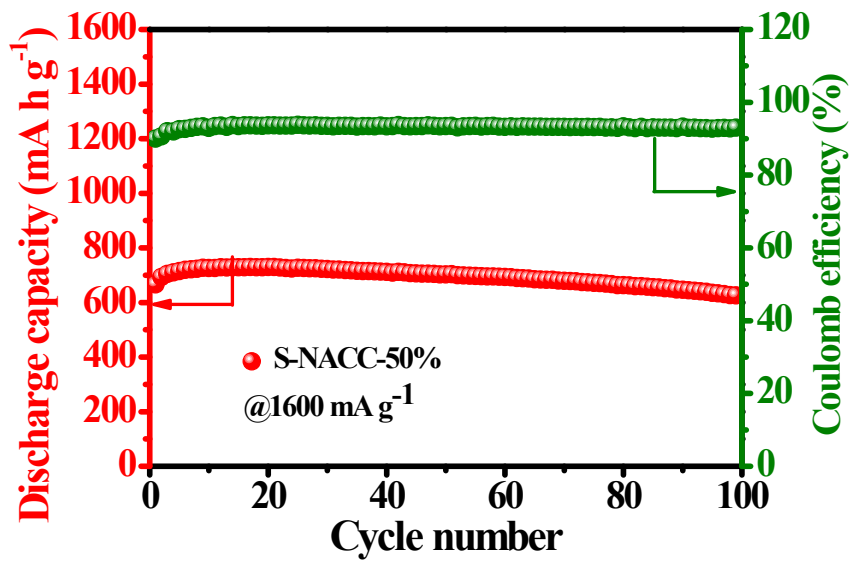


Fig.S9 the cycling performance of S-NACC-50% at 1600 mA g⁻¹