# **RSC Advances Supporting Information**

### Development of in-situ synthesized Y-based nanoparticle/ polyethersulfone

### adsorptive membranes by adjusting the composition of coagulation bath for

#### enhanced removal of fluoride

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## The pH effect on the leakage of Y-based NPs

The pH effect on the leakage of Y-based NPs in batch adsorption was also studied by measuring the Y<sup>3+</sup> concentration via an inductively coupled plasma mass spectrometer (ICP-MS, Agilent 7900).

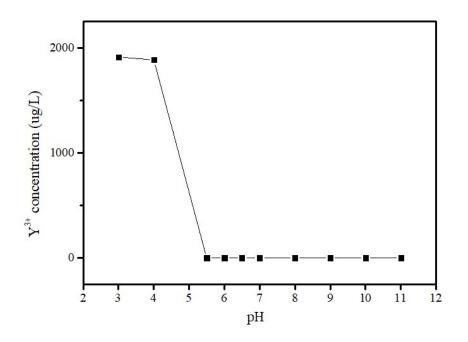


Fig. S1 pH effect on the leakage of Y-based NPs

### The XPS analysis

The surface chemistry of Y-based NPs before and after adsorption of F- was studied by

the X-ray photoelectron spectroscopy (XPS) (ESCALAB 250Xi, Thermo Scientific, USA), with a monochromatic Al K $\alpha$  X-ray source (1486.8 eV). The high resolution scans were conducted according to the peak being examined with a pass energy of 30 eV and step size of 0.10 eV. To compensate for the charging effects, all spectra were calibrated with graphitic carbon as the reference at a binding energy (BE) of 284.8 eV. The XPS results were collected in binding energy forms and fit using a nonlinear least-square curve fitting program (XPSPEAK41 software).

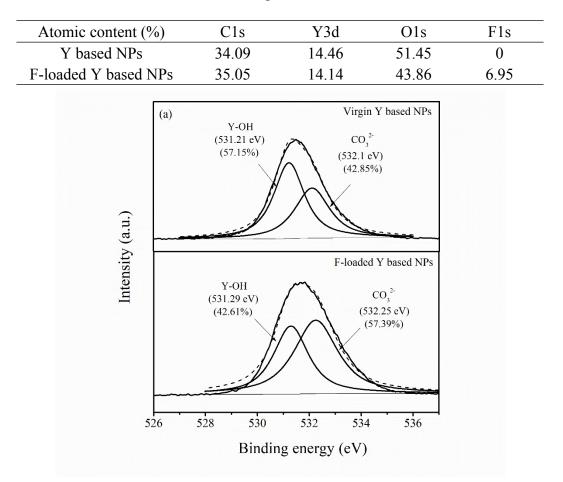
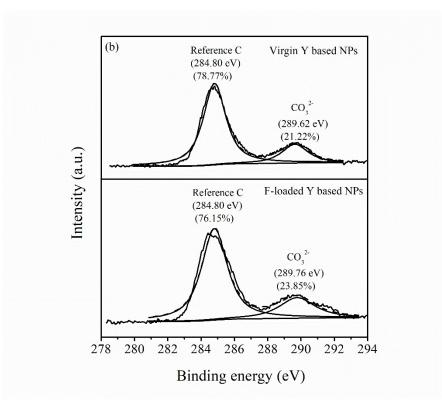


Table S1 The atomic content of vingin and F-loaded Y based NPs<sup>1</sup>



**Fig. S2** High resolution XPS spectra of (a) O1s and (b) C1s of Y-based NPs before and after adsorption<sup>1</sup>.

## **Regeneration Experimental**

0.05 g membrane M10-2 was first added into 100 mL of 50 mg/L fluoride solution for 24 h. After the adsorption, the used membrane was collected and desorbed into 100 mL of 0.01 M NaOH solution for 12 h. The regenerated membrane was then washed by DI water several times and reused for the subsequent adsorption experiment. Three adsorption-desorption cycles were conducted to evaluate the membrane reusability. The adsorption capacity in each cycle was determined and the regeneration rate was calculated.

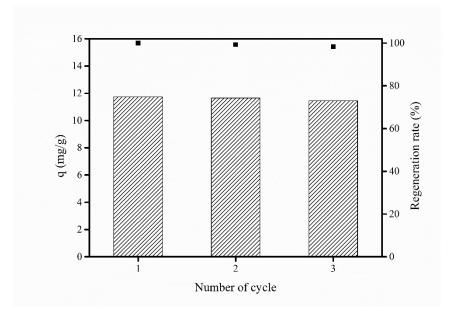


Fig. S3 Regeneration performance of M10-2 for fluoride removal.

### References

1. J. He, A. Cui, F. Ni, S. Deng, F. Shen, C. Song, L. Lou, D. Tian, C. Huang and L. Long, Journal of Colloid and Interface Science, 2019, 536, 710-721.