Supplementary information

Improving the structural stability and electrochemical performance of $Na_2Li_2Ti_6O_{14}$ particles via MgF₂ coating Wei-Wei Ma^{*a*}, Hai-Tao Yu^{*a*}, Chen-Feng Guo^{*a*}, Ying Xie^{*a*,*}, Ning Ren^{*b*,*}, Ting-Feng Yi ^{*c*,*},

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Fig. S1 Reitveld refinements for (a) NLTO, (b) MgF₂-NLTO-1, (c) MgF₂-NLTO-3, (d)

MgF₂-NLTO-5, and (e) MgF₂-NLTO-7 samples.

	NLTO	MgF ₂ -	MgF ₂ -	MgF ₂ -	MgF ₂ -
		NLTO-1	NLTO-3	NLTO-5	NLTO-7
a (Å)	5.737	5.734	5.741	5.739	5.769
b (Å)	11.2245	11.226	11.226	11.21	11.221
c (Å)	16.485	16.475	16.459	16.471	16.521
Volume (Å ³)	1061.503	1060.463	1060.757	1060.171	1069.359
R_{wp}	5.6%	10.9%	9.5%	8.3%	11.3%
$(V_{mf}-V)/V(\%)^{a}$	0.02%	0.11%	0.09%	0.10%	0.66%

Table S1 Lattice constants (Å), volumes, and relevant $R_{wp.}$ calculated from Rietveld refinement for different samples.

^{*a*} The volume of NLTO is taken as the reference

Table S2 Potentials of the redox peaks (V) and the potential differences between the oxidation and reduction peaks (V) ^{*a*} for different samples.

	Redox-potention (V)(<i>a</i>)				
Cycle number	lst	2nd	3rd		
NLTO	1.43/1.21 (0.22)	1.42/1.17 (0.25)	1.44/1.15 (0.29)		
MgF ₂ -NLTO-1	1.33/1.21 (0.12)	1.35/1.19 (0.16)	1.35/1.21 (0.14)		
MgF ₂ -NLTO-3	1.32/1.22 (0.10)	1.32/1.21 (0.11)	1.32/1.21 (0.11)		
MgF ₂ -NLTO-5	1.35/1.27 (0.08)	1.36/1.27 (0.09)	1.35/1.27 (0.08)		
MgF ₂ -NLTO-7	1.38/1.18 (0.20)	1.36/1.17 (0.19)	1.30/1.14 (0.16)		

^{*a*} The values in bracket are the potential difference between the oxidation and reduction peaks

Sample	σ (Ω s ^{-0.5})	$D_{Li}(cm^2 s^{-1})$
NLTO	1471.14	3.53×10 ⁻¹⁷
MgF ₂ -NLTO-1	938.57	8.67×10 ⁻¹⁷
MgF ₂ -NLTO-3	655.30	1.78×10 ⁻¹⁶
MgF ₂ -NLTO-5	554.11	2.49×10 ⁻¹⁶
MgF ₂ -NLTO-7	1439.41	3.69×10 ⁻¹⁷

Table S3 Warburg factors and lithium diffusion coefficients for pristine and MgF_{2} -NLTO samples