Supplementary information

New transition metal Complexes with Indole ring pendent: Insights into the Antifungal activity and Mode of Action

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Figure S1 Mass spectrum of schiff base ligand (L)

Figure S2 Mass spectrum of C1 complex

Figure S3 Mass spectrum of C2 complex

Figure S4 Mass spectrum of C3 complex

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Figure S7 FT-IR spectrum of C3 complex

Figure S8 FT-IR spectrum of C4 complex

Figure S9¹HNMR spectrum of Schiff base (L) in CDCl₃

Figure S10¹HNMR spectrum of C4 complex in DMSO-d₆

Figure S11¹³CNMR spectrum of Schiff base ligand(L)

Figure S12¹³CNMR spectrum of C4 complex

Table S1: Effect of the test compounds (L, C1-C4) on the rate of H⁺-efflux by various*Candida albicans* isolates at pH 7.0: Cells were suspended in 0.1 mM CaCl₂ and 0.1 MKCl at 25 °C

1. Mass spectra

The mass spectra of the Schiff base (L) and metal complexes (C1-C4) were recorded and found in good agreement with the molecular formulae and structures of these synthesized compounds. The mass spectra $[M+H]^+$ of Schiff base ligand (L) showed the molecular ion peak at m/z 315.0 and its corresponding metal complexes (C1, C2, C3 and C4) showed molecular ion peaks at 609.5, 605.9, 605.5 and 611.4 respectively. The mass spectra of the synthesized compounds are given in **FiguresS1-S5**.

2. ¹H NMR and ¹³C NMR spectra

The ¹H NMR of the Schiff base (**L**) and **C4** complex was recorded in CDCl₃ and DMSO-d6 respectively using tetramethylsilane (TMS) as internal standard. In Schiff base ligand (**L**) the CH=N (azomethinic) proton appeared as a singlet at 8.4 ppm and in **C4** complex the signal for the imine proton appeared at 8.53 ppm and this downfield shift with respect to the corresponding signal in the free ligand confirms the coordination of azomethine nitrogen with the Zn(II) metal ion . The Ph-OH proton of the Schiff base appeared as a singlet at 13.11 ppm. In **C4** complex the Ph-OH proton signal was found absent confirming the complexation through phenolic oxygen. The aromatic protons in Schiff base appeared as a set of multiplets in the region 6.7-8.2 ppm. The CH₂ protons of the Schiff base appeared as triplets in the region 2.9-3.5 ppm. The positions of the main signals in the ¹HNMR and ¹³CNMR are given in the experimental section. The attained ¹³C values of **C4** complex were compared with the corresponding ligands and were in good agreement with the proposed structure of the **C4** complex. In Schiff base the azomethine carbon atom was observed at 163.94 ppm which was shifted to 164.10 ppm in complex. The paramagnetic **C1**, **C2** and **C3** complexes did not show fine NMR spectra and were therefore not

included in this study. The ¹H and ¹³C NMR spectra of Schiff base (L) and C4 complex are shown in **Figures S9-S12**.



Figure S1 Mass spectrum of schiff base ligand (L)showing prominent parent ion peak at 315 (m/z) which corresponds to $[M+H]^+$



Figure S2 Mass spectrum of C1complex showing prominent parent ion peak at 609.5 (m/z) which corresponds to $[M+H]^+$



Figure S3 Mass spectrum of C2 complex showing prominent parent ion peak at 605.9 (m/z) which corresponds to $[M+H]^+$



Figure S4 Mass spectrum of C3 complex showing prominent parent ion peak at 605.5 (m/z) which corresponds to $[M+H]^+$



Figure S5 Mass spectrum of C4 complex showing prominent parent ion peak at 611.4 (m/z) which corresponds to $[M+H]^+$



Figure S6FT-IR spectrum of C2 complex



Figure S7FT-IR spectrum of C3 complex



Figure S8FT-IR spectrum of C4 complex



Figure S9¹HNMR spectrum of Schiff base (L) in CDCl₃



Figure S10¹HNMR spectrum of C4 complex in DMSO-d₆



Figure S11¹³CNMR spectrum of Schiff base ligand (L)



Figure S12¹³CNMR spectra of C4 complex

	Range of relative H ⁺ -efflux rate (×10 ⁻¹¹ mol/min/mg cells)										
Test compounds	FLC Susceptible strains								FLC Resistant strains		
	SC5314	4175	4179	4180	4251	4554	4563	4576	4085	4122	4135
Control	4.131	5.642	5.861	6.124	5.113	5.772	6.193	5.911	9.043	8.442	8.341
Glucose(5 mM)	11.34	11.89	12.71	12.48	11.22	12.32	13.57	12.61	22.99	21.37	21.54
L	3.038	3.923	4.577	4.673	4.003	4.237	4.787	4.451	5.362	5.259	5.246
	(27)	(31)	(22)	(24)	(19)	(27)	(23)	(25)	(41)	(38)	(37)
L+Glucose	9.331	9.429	10.68	9.772	10.01	10.01	11.61	10.61	18.23	17.51	6.181
	(18)	(21)	(16)	(12)	(11)	(19)	(15)	(16)	(21)	(18)	(26)
C1	3.033	3.886	4.588	4.673	4.029	4.219	4.156	4.451	6,520	6.441	6.781
	(27)	(31)	(22)	(24)	(19)	(27)	(23)	(25)	(28)	(24)	(19)
C1+Glucose	9.296	9.429	10.68	9.834	10.02	9.992	11.57	10.58	18.23	19.63	6.923
	(18)	(21)	(16)	(12)	(11)	(19)	(15)	(16)	(21)	(19)	(17)
C2	4.689	2.098	2.778	2,639	2.623	2.869	2.768	2.843	2.830	2.820	3.028
	(59)	(63)	(53)	(57)	(49)	(51)	(56)	(52)	(69)	(67)	(64)
C2+Glucose	7.563	6.325	5.783	6.402	6.698	4.078	6.391	7.226	8.989	6.582	2.94
	(44)	(47)	(55)	(49)	(41)	(67)	(53)	(43)	(61)	(69)	(65)
С3	0.983	1.602	2.174	2,652	1.416	2.372	2.558	2.288	1.709	1.849	1.843
	(67)	(72)	(63)	(57)	(53)	(59)	(59)	(62)	(81)	(78)	(78)
C3+Glucose	5.635	5.125	6.266	3.456	4.645	6.542	6.120	4.843	7.380	7.971	2.761
	(51)	(57)	(51)	(63)	(59)	(47)	(55)	(62)	(68)	(63)	(67)
C4	3.161	2.882	4.178	4,862	4.407	3.873	4.279	4.498	6.339	6.171	5.747
	(24)	(49)	(29)	(21)	(24)	(33)	(31)	(24)	(30)	(27)	(31)
C4+Glucose	9.632	7.027	10.11	10.75	9.391	9.400	10.62	10.78	17.95	17.83	6.056
	(15)	(41)	(21)	(14)	(17)	(24)	(22)	(15)	(22)	(17)	(27)
Fluconazole	3.571	4.585	5.233	5.469	4.566	4.571	5.456	5.385	8.130	7.758	7.582
	(14)	(19)	(11)	(11)	(13)	(21)	(12)	(09)	(10)	(08)	(10)

Table S1Effect of the test compounds (L, C1-C4)on the rate of H⁺-efflux by various*Candida albicans* isolates at pH 7.0: Cells were suspended in 0.1 mM CaCl₂ and 0.1 MKCl at 25 °C

Values in parentheses give the %-ageinhibition of H⁺-efflux w.r.t. control