

Supplementary Information

Electrodeposition of amorphous WO₃ on SnO₂-TiO₂ inverse opal nano- framework for highly transparent, effective and stable electrochromic smart window

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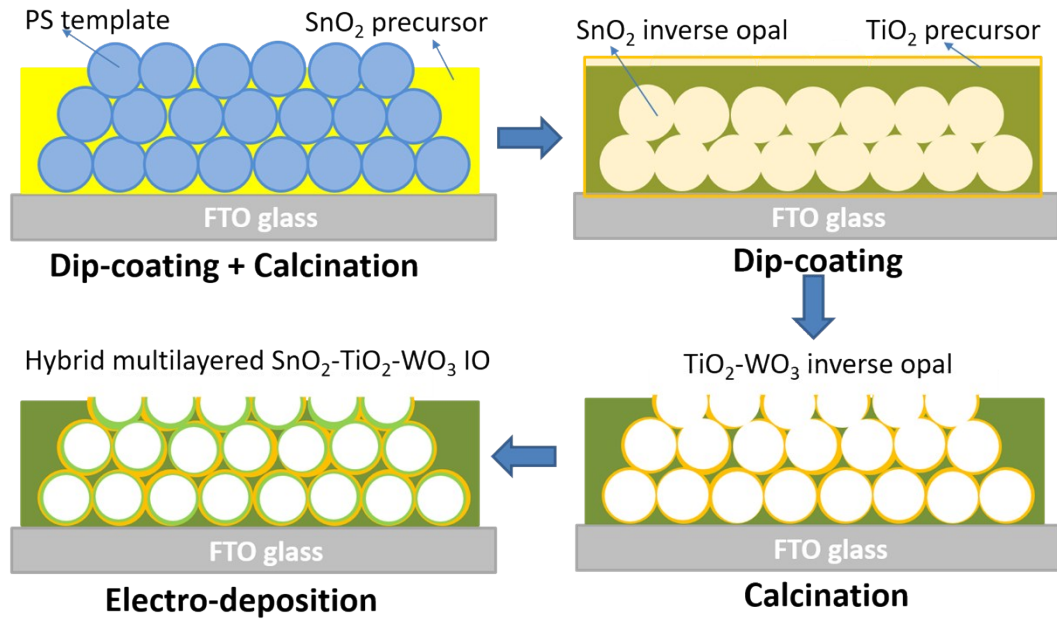


Figure S1: The synthesis procedure of hybrid multilayered SnO₂-TiO₂-WO₃ inverse opal nanostructure.

The hybrid multilayered SnO₂-TiO₂-WO₃ inverse opal nanostructure was fabricated onto a 1.5 x 2.5 cm² commercial FTO glass. The as-received FTO glass was pre-treated by a cleaning procedure using methanol, acetone and ethanol, and finally kept in the mixed solution of ethanol and DI water before sample fabrication. The PS spheres were dispersed in DI water solvent with the concentration of 0.02 wt.%. For each fabrication, three pieces of pre-cleaning FTO glass were immersed in 100 mL of PS dispersion solution.

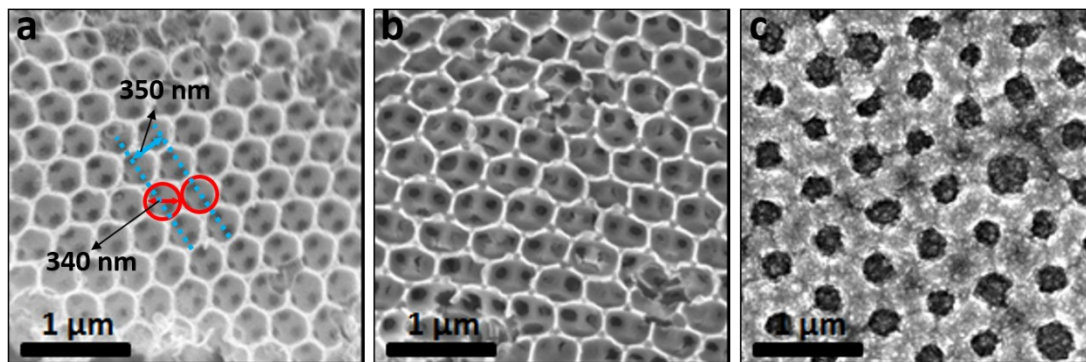


Figure S2: FESEM images of SnO₂ IO (a), SnO₂-TiO₂ (b) and hybrid multilayered SnO₂-TiO₂-WO₃ IO nanostructure (c) with initial pore-size of 520 nm and electro-deposition duration of 200 seconds. All images were acquired under the same magnification.

Figure S2 shows the FESEM images of SnO₂, SnO₂-TiO₂, and hybrid multilayered SnO₂-TiO₂-WO₃ IO nanostructure obtained from the PS solution with initial pore-size of 520 nm. The diameter of the SnO₂ IO pores is about 340 nm, which is smaller than that of the initial PS opal diameter (520 nm). The distance between two adjacent pores is measured to be approximately 350 nm. With the coating of a thin layer of TiO₂, there is no significant change in the morphology of the double-layered SnO₂-TiO₂ IO framework (Figure S2b). However, following WO₃ electrodeposition, a uniformly deposited WO₃ layer can be clearly observed on the surface of the double-layered SnO₂-TiO₂ IO framework (Figures S2c), forming a hybrid multilayered SnO₂-TiO₂-WO₃ IO nanostructure.

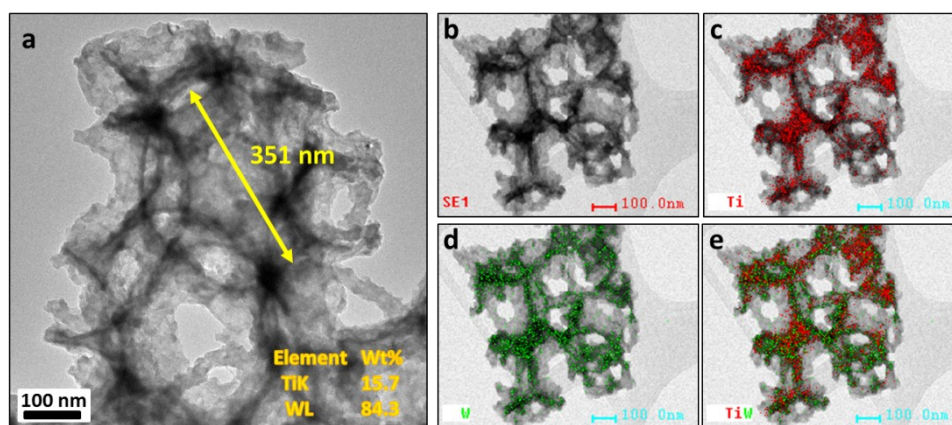


Figure S3: STEM images coupled with EDX mapping of hybrid multilayered SnO₂-TiO₂-WO₃ IO nanostructure obtained from the PS solution with initial pore-size of 392 nm.

Figure S3 indicates the STEM image coupled with EDX mapping of TiO₂-WO₃ IO nanostructure. It is obvious that the inner SnO₂ framework is surrounded by a layer of amorphous WO₃. By EDX analysis, the bulk elemental composition of TiO₂-WO₃ IO is measured to be 15.7 and 84.3 at.% for Ti and W, respectively.

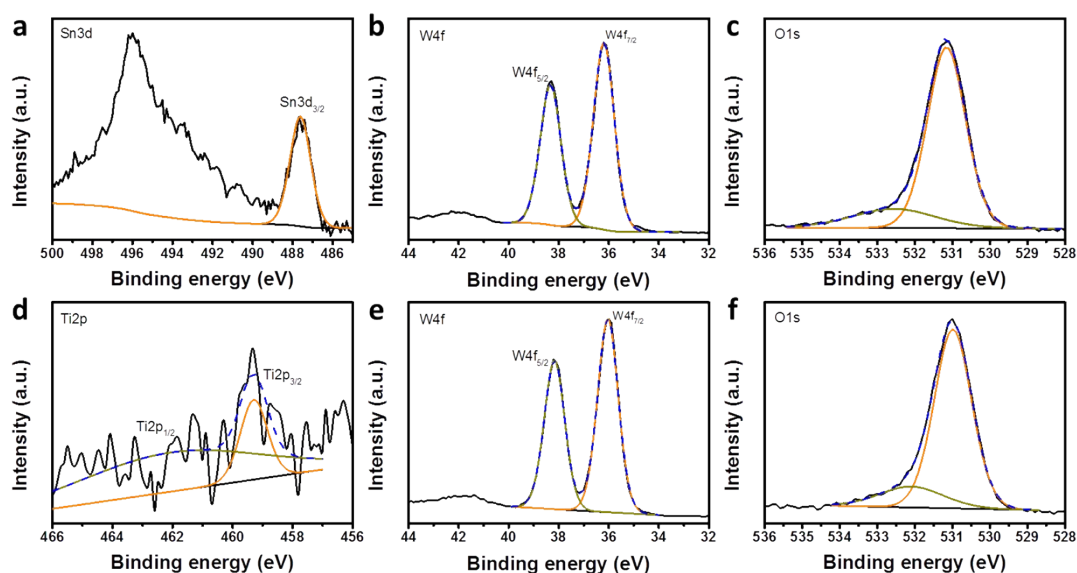


Figure S4: The fine XPS spectra of SnO₂-WO₃ (a-c) and TiO₂-WO₃ (d-f) IO nanostructures.

Figure S4 presents the fine scan XPS spectra of Sn3d, W4f, O1s components for SnO₂-WO₃ and Ti2p, W4f, O1s for TiO₂-WO₃ IO nanostructures. By curve-fitting analysis,

the O1s spectrum of both SnO₂-WO₃ and TiO₂-WO₃ IO is located in the metal oxide binding energy range (~530.8 eV). The W4f spectrum is located at the binding energy of approximately 36.0 eV, confirming the WO₃ chemical state, with well separated spin-orbit components ($\Delta_{\text{metal}}=2.17$ eV). The Sn3d and Ti2p spectra also confirms the SnO₂ (~486.6 eV) and TiO₂ chemical state (~459.1 eV), respectively.