

Supporting Information

Magnetically retrievable Ce-doped Fe_3O_4 nanoparticles as scaffolds for removal of azo dye

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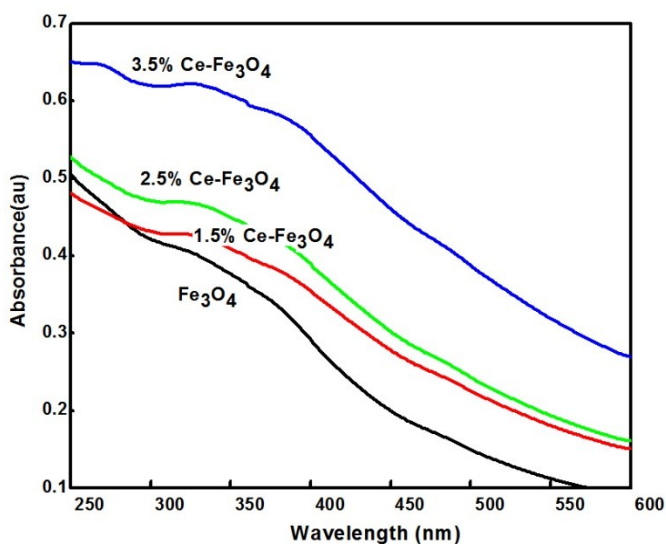


Figure S1. UV-visible spectra of synthesized samples

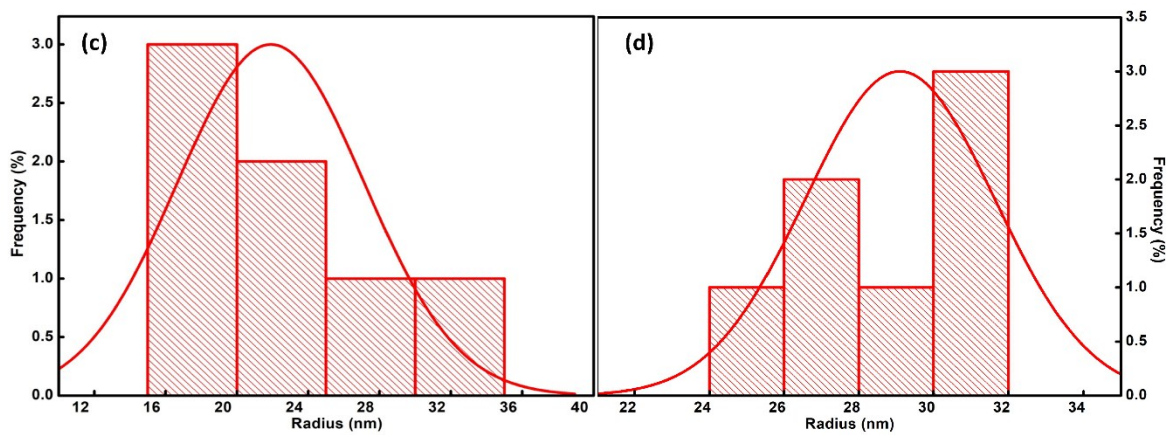
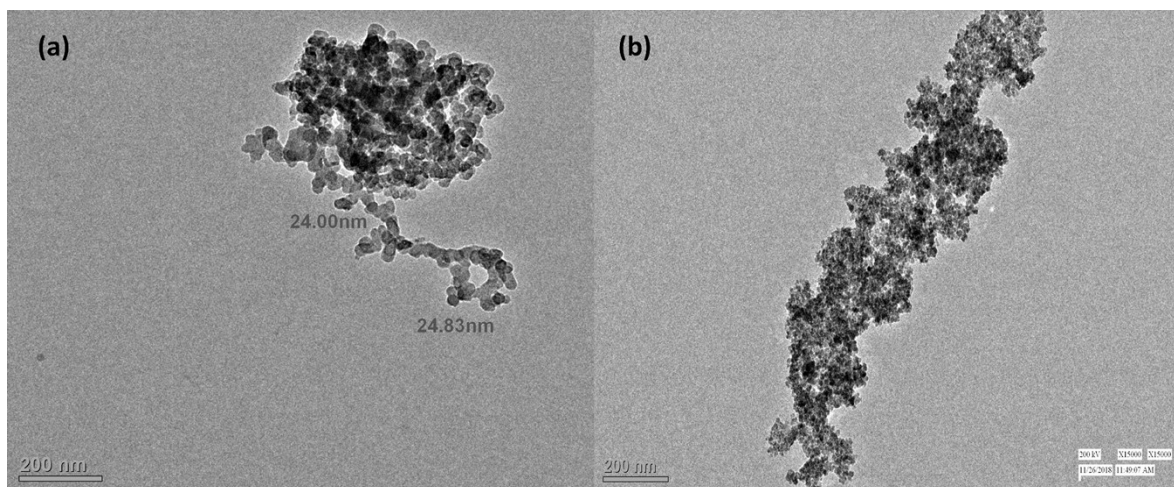


Figure S2. TEM analysis (a) Undoped (b) 3.5 % cerium doping, Histogram of (c) Undoped (d) 3.5 % cerium doping

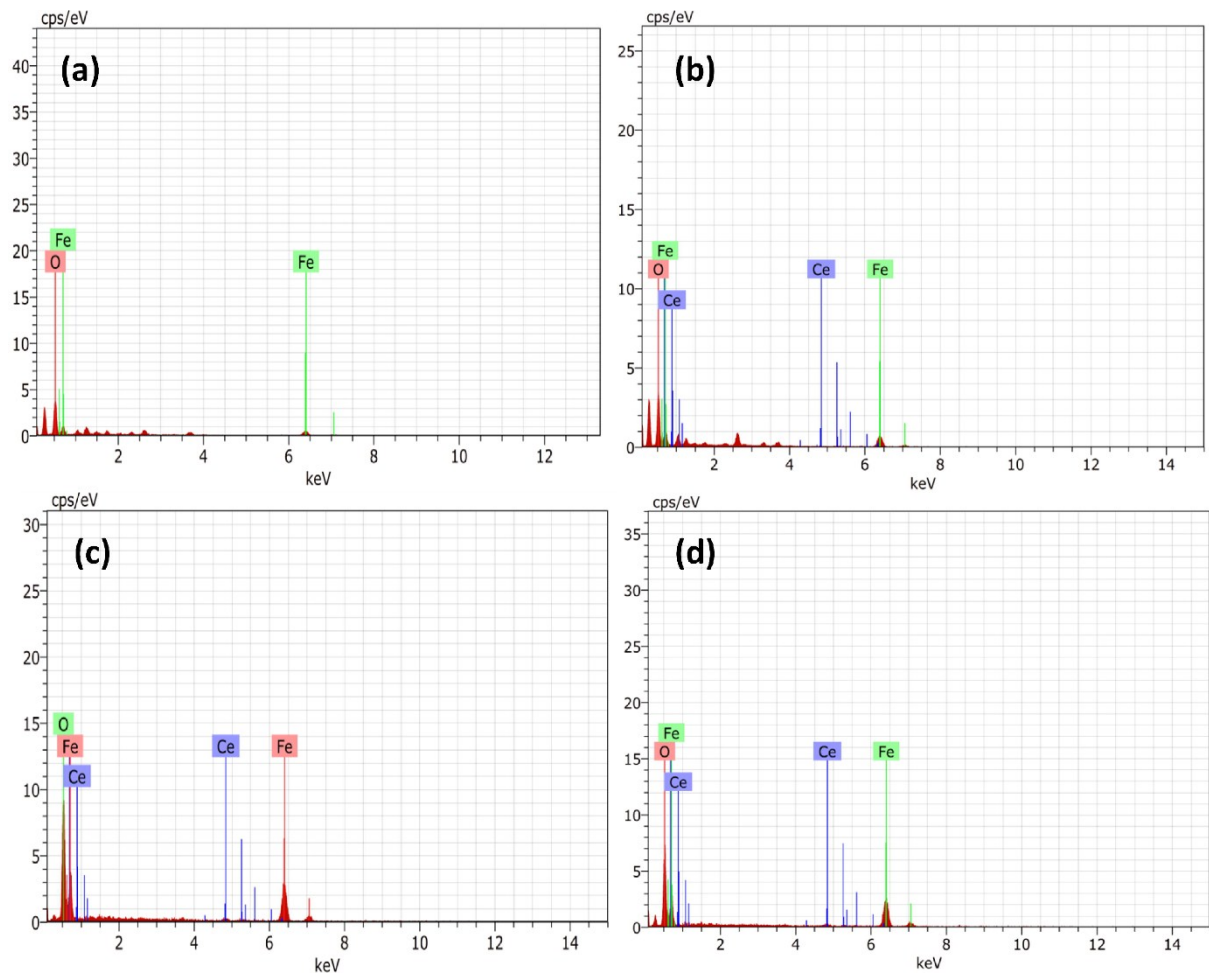


Figure S3. EDX (a) Undoped (b) 1.5 % (c) 2.5 % (d) 3.5 % cerium doping

Table S1 : Parameters calculated from BET analysis

System	Specific surface area (a) [$\text{cm}^3 \text{g}^{-1}$]	Total pore volume	Mean pore diameter [nm]	Amount of monolayer adsorption (V_m) [$\text{cm}^3 \text{g}^{-1}$]
Fe_3O_4	71.600	0.2976	14.491	16.450
1.5% Ce- Fe_3O_4	89.672	0.3249	15.039	20.603
2.5% Ce- Fe_3O_4	94.866	0.3567	16.625	21.796
3.5% Ce- Fe_3O_4	97.601	0.4611	18.898	22.424

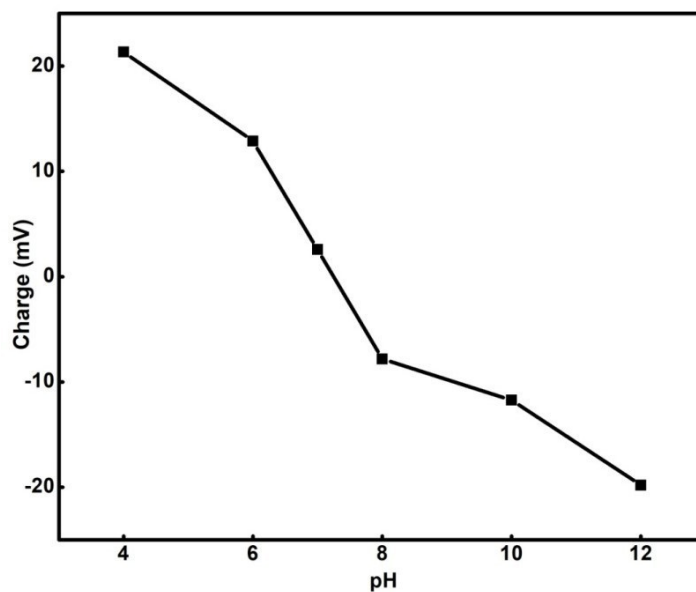


Figure S4. Variation of zeta potential charge as function of pH

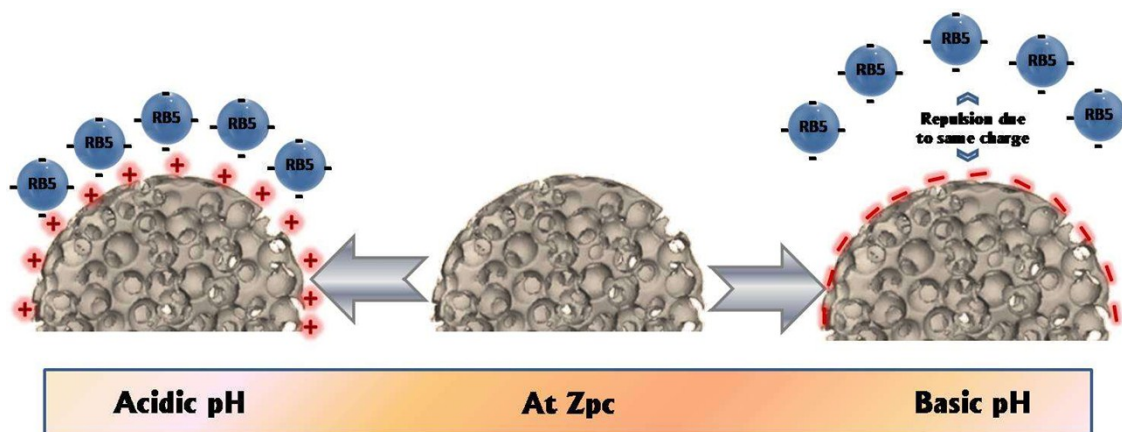


Figure S5. Role of point zero surface charge of Nanoparticles on adsorption of anionic dye RB5

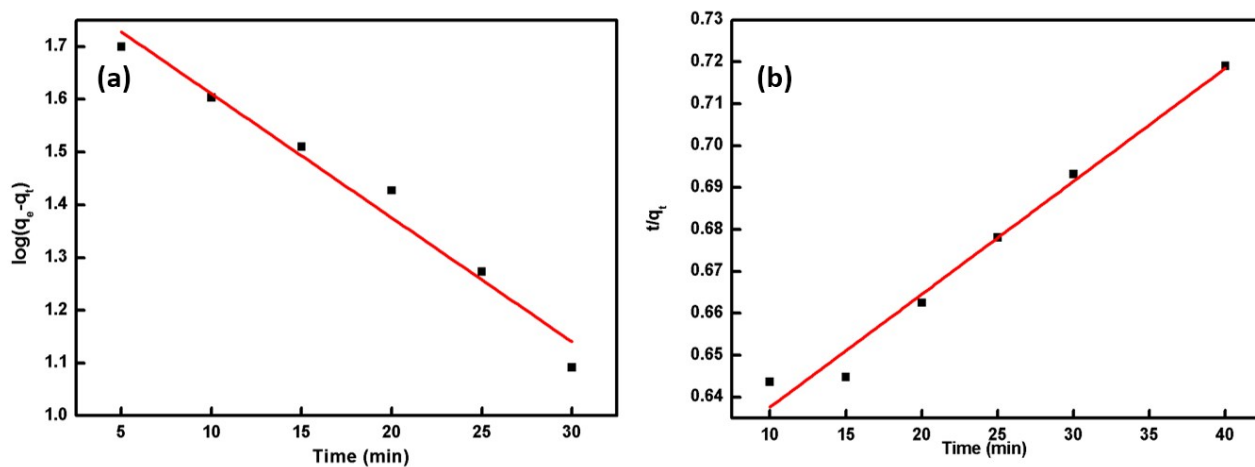


Figure S6. Kinetic models (a) Pseudo First order (b) Pseudo second order

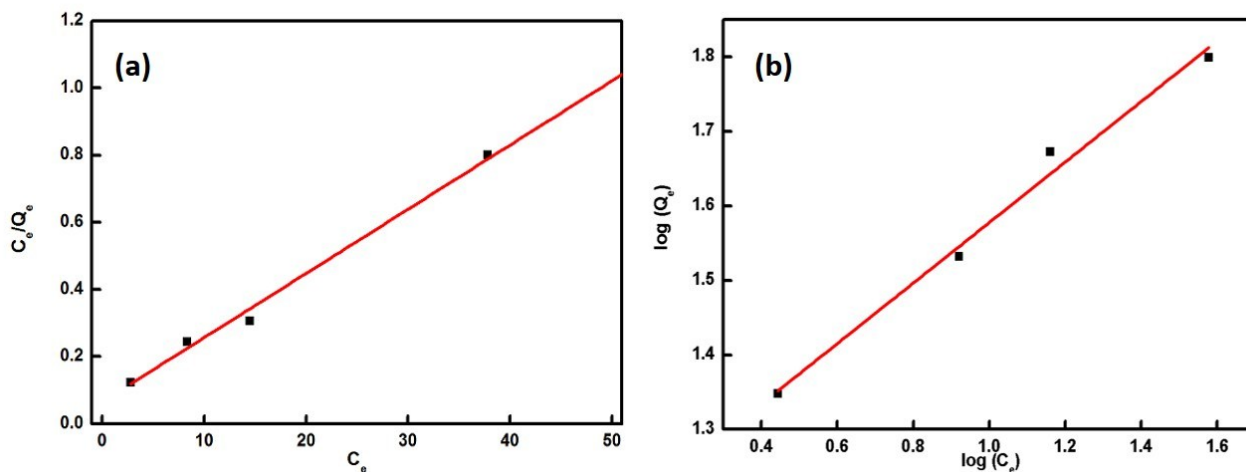


Figure S7. Adsorption isotherms of RB5 onto the Nps (a) Langmuir (b) Freundlich

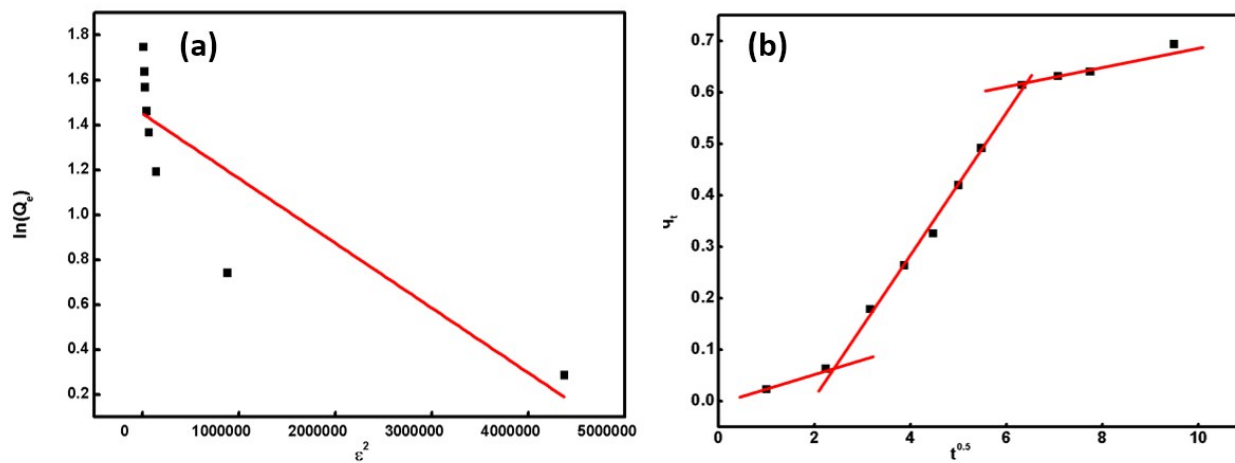


Figure S8. Adsorption isotherms of RB5 onto the Nps (a) Dubinin-Radushkevich (D-R) (b) Intra-particle diffusion models

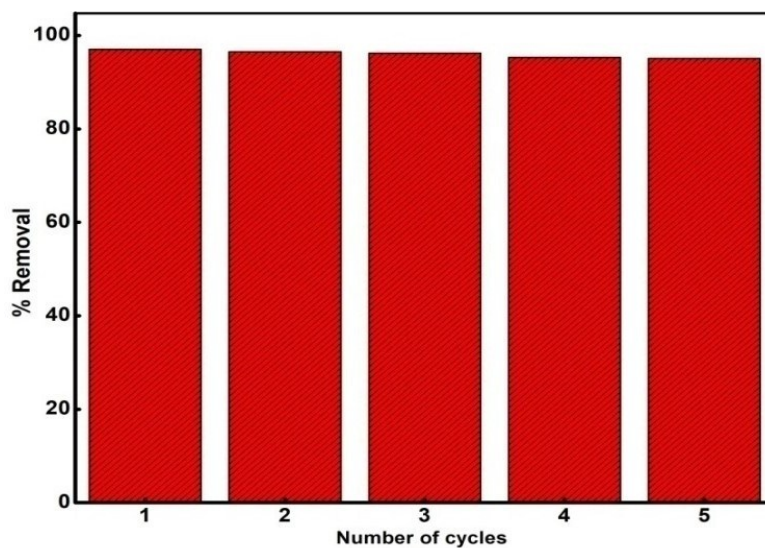


Figure S9. Recyclability of nanoabsorbent

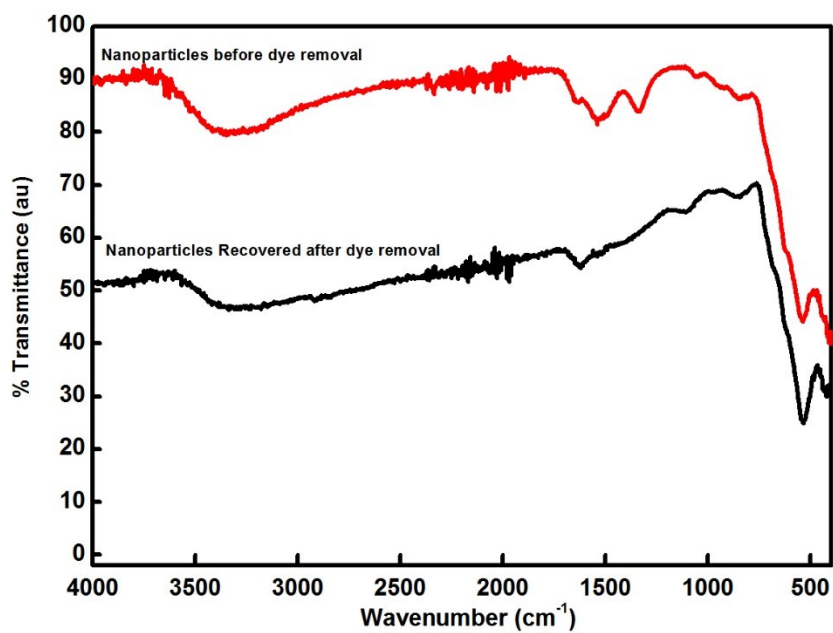


Figure S10. FTIR of recovered Nps compared with Nps before use

