Supporting Information

Two-phase interface hydrothermal synthesis of binder-free SnS₂/graphene flexible paper electrodes for high-performance Li-ion batteries

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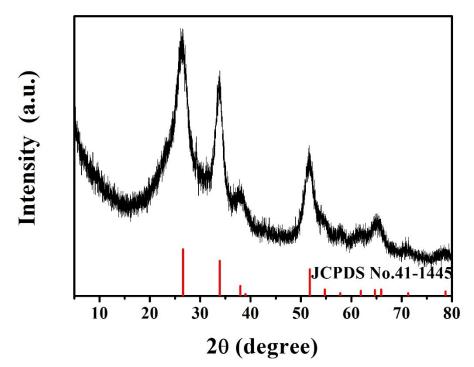


Figure S1. XRD pattern of the SnS₂/graphene oxide composite paper subjected to a hydrothermal process at the solid/gas interface with de-ionized water as a reducing agent.

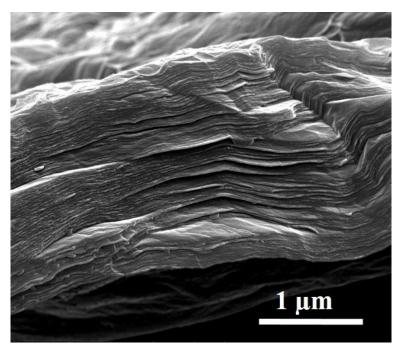
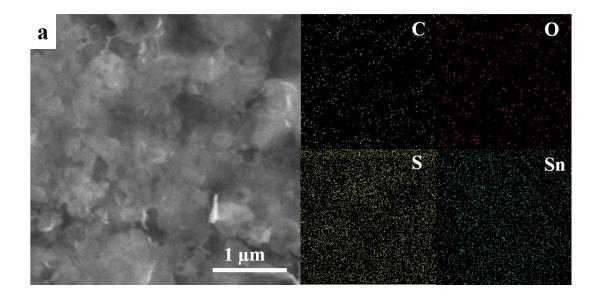


Figure S2. Cross-section view SEM image of the graphene paper.



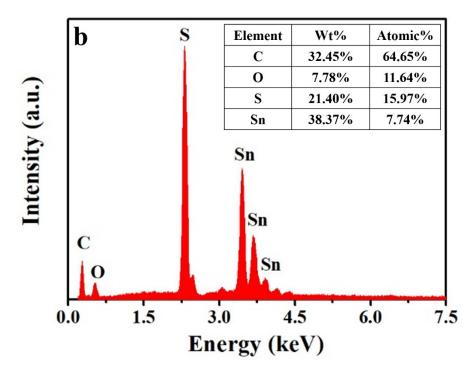


Figure S3. a) SEM image of the SGP with corresponding EDX mapping images; b) EDX spectrum of the SGP.

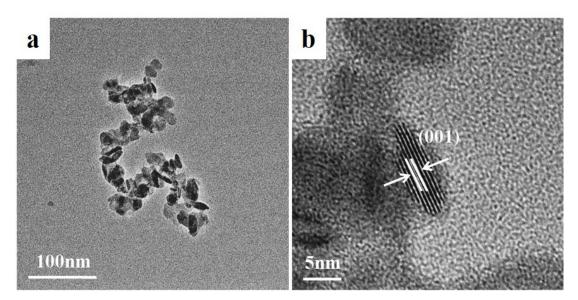


Figure S4. a) TEM and b) HRTEM images of the pristine SnS₂ nanocrystals.

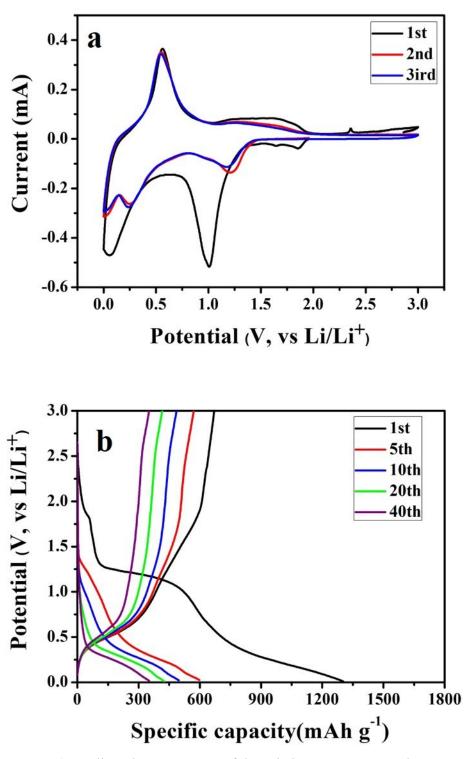


Figure S5. a) Cyclic voltammograms of the pristine SnS_2 measured at a scan rate of 0.1 mV s⁻¹; b) Galvanostatic charge-discharge profiles of the pristine SnS_2 measured at a current density of 100 mA g⁻¹ in the voltage range of 0.01 to 3.0 V (vs. Li/Li⁺).

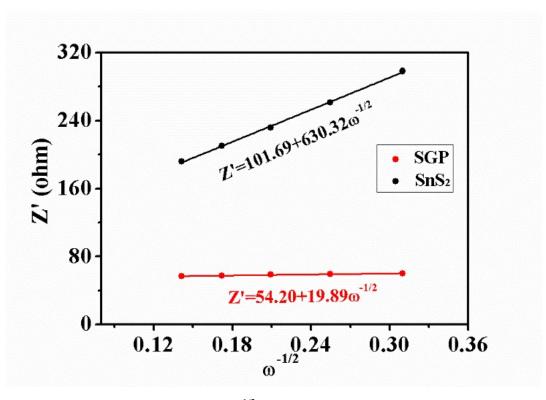


Figure S6. Z' as a function of the $\omega^{-1/2}$ plot in the low frequency region of the SGP and pristine SnS₂ nanocrystals electrodes (the slope of fitting curves is the Warburg factor, σ).

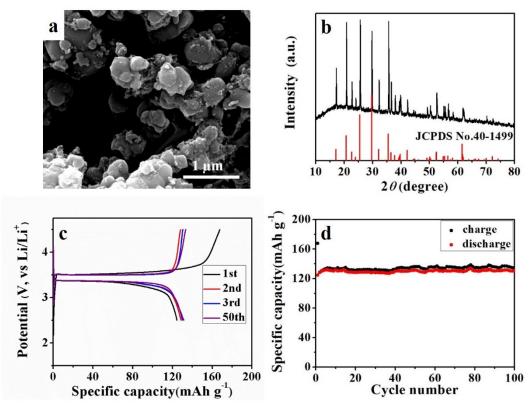


Figure S7. a) SEM image and b) XRD pattern of the commercial LiFePO₄; c) The galvanostatic charge-discharge curves, and cyclic performance of the LiFePO₄ half cell, measured at a current density of 100 mA g⁻¹ in the voltage range of 2.5 to 4.5 V (vs. Li/Li⁺).

	Before cycling		After 10 cycles	
	Pristine SnS ₂	SGP	Pristine SnS ₂	SGP
Re/Ω	6.78	2.02	6.05	1.66
Rs/Ω	195.69	139.2	141.7	50.38
Rct/Ω	2.3	1.94	2.1	0.28
R_{total}/Ω	204.77	143.16	149.85	50.38

Table S1. Values of the equivalent circuit elements used to fit the experimental d	ata