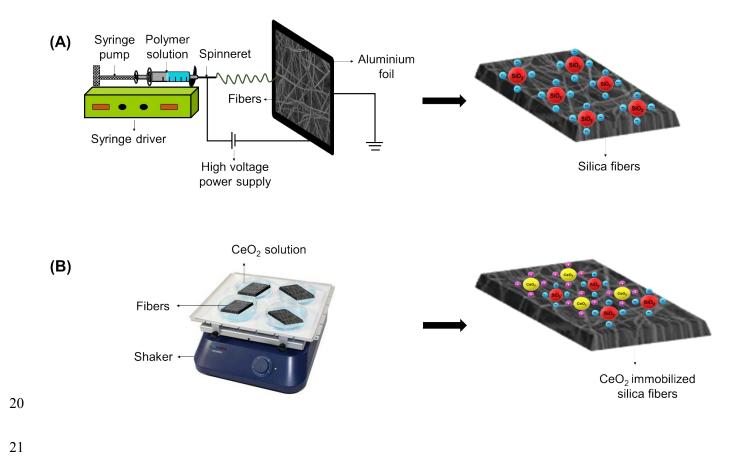
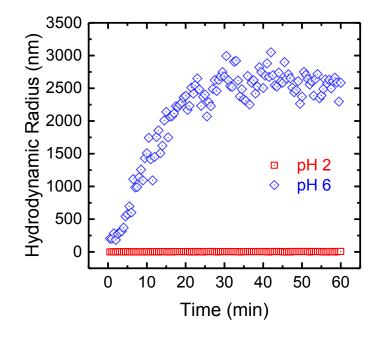
1	Electronic Supplementary Information
2	
3	Electrospun silica nanofiber mats
4	functionalized with ceria nanoparticles for
5	water decontamination
6	
7	Ines Zucker <sup>a,b,c*</sup> , Nadir Dizge <sup>a,d</sup> , Camrynn L. Fausey <sup>a</sup> , Evyatar Shaulsky <sup>a</sup> , Meng Sun <sup>a</sup> ,
8	Menachem Elimelech <sup>a</sup>
9	
10	
11	<sup>a</sup> Department of Chemical and Environmental Engineering, Yale University, New Haven,
12	Connecticut 06520-8286
13	<sup>b</sup> Porter School of Environmental Studies, Tel Aviv University, Israel
14	<sup>c</sup> School of Mechanical Engineering, Tel Aviv University, Israel
15	<sup>d</sup> Department of Environmental Engineering, Mersin University, Mersin, 33343, Turkey
16	
17	Enclosed:
18	Fig. S.1-S.6
19	Table S.1

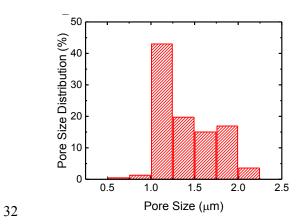


**Fig. S.1.** Schematic illustration of fabricating electrospun CeO<sub>2</sub> composite silica fibers. (A) Preparation of silica fibers by electrospinning. Experimental conditions: collector distance, 11 cm; rotating drum speed, 16 cm min<sup>-1</sup>; collection time, 10 h; flow rate, 0.5 mL h<sup>-1</sup>; applied voltage, 13 kV; temperature, 30°C; relative humidity, 30%. (B) CeO<sub>2</sub> immobilization on silica fibers. Experimental conditions: CeO<sub>2</sub> concentration, 1 mg mL<sup>-1</sup>; fiber weight, 10 mg; shaker speed, 75 rpm; volume, 10 mL, immobilization time, 16 h; temperature, 25 °C.

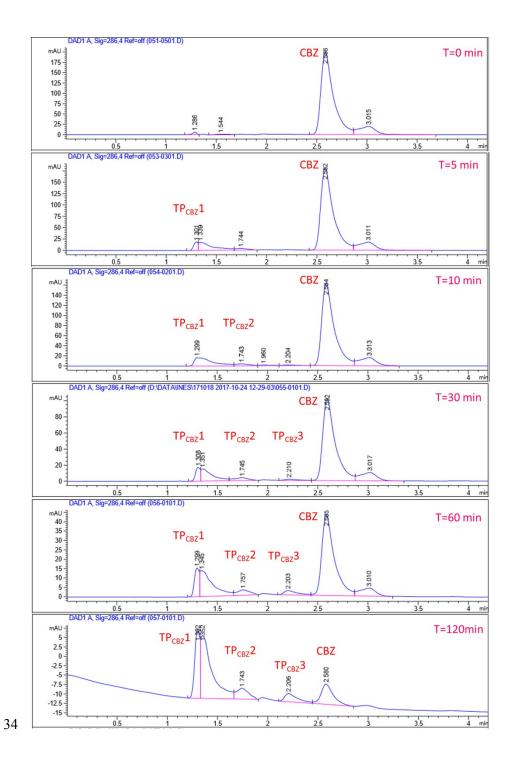


28

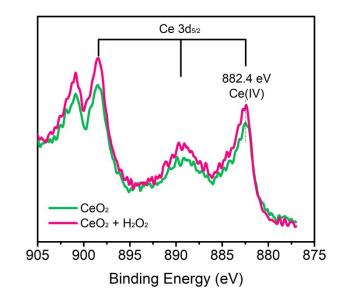
**Fig. S.2.** Aggregation kinetics for ceria nanoparticle suspension (in 10 mM acetate buffer solution) at pH 2 and 6 as determined by dynamic light scattering. Light scattering data are presented as the corresponding spherical hydrodynamic radius.



33 Fig. S.3. Pore size distribution of ceria/silica nanofiber composites.

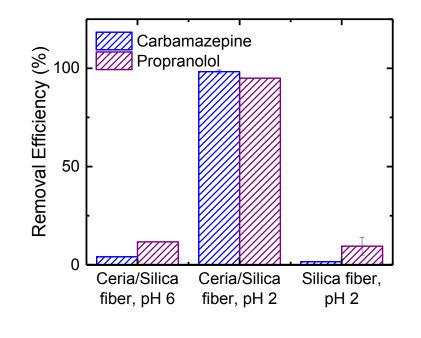


35 **Fig. S.4.** HPLC chromatographs taken during oxidation of 1 mg L<sup>-1</sup> carbamazepine (CBZ) by 36 ceria/silica fibers at pH 2 in the presence of 0.5 mM  $H_2O_2$ . Transformation products appeared 37 during oxidation at lower retention times, and were marked as  $TP_{CBZ}1$ , 2 and 3.



38

39 **Fig. S.5.** High resolution X-ray photoelectron spectroscopy (XPS) spectra of dried CeO<sub>2</sub> 40 nanoparticles prior to and following addition of 0.5 mM H<sub>2</sub>O<sub>2</sub>. XPS data were obtained with 41 a scanning XPS microprobe (PHI, VersaProbe II, Japan) using monochromatic Al K  $\alpha$ 42 radiation with a 0.47 eV system resolution.



43

44 **Fig. S.6.** Removal efficiency of 1 mg  $L^{-1}$  carbamazepine (blue) and propranolol (purple) by 45 ceria/silica fibers and pristine fibers at pH 2 and 6 in the presence of 0.5 mM  $H_2O_2$ .

46 Table S.1. Chemical structures and properties of the trace organic compounds used in this47 study.

48

Property	Carbamazepine (CBZ)	Propranolol (PRO)	Perfluorooctanesulfonic acid (PFOS)
Chemical structure	O NH <sub>2</sub>	OH H O N	O HO-S-CF <sub>2</sub> (CF <sub>2</sub> ) <sub>6</sub> CF <sub>3</sub> O
Molecular formula	$C_{15}H_{12}N_2O$	C <sub>16</sub> H <sub>21</sub> NO <sub>2</sub>	CF <sub>3</sub> (CF <sub>2</sub> ) <sub>7</sub> SO <sub>3</sub> H
Molecular weight (g mol <sup>-1</sup> )	236.27	295.80	500.13
pK <sub>a</sub>	13.9	9.42	-3.27

49