

Electronic Supplementary Information (ESI)

Theoretical predictions for low-temperature phases, softening of phonons and elastic stiffnesses, and electronic properties of sodium peroxide under high pressure

Pornmongkol Jimlim,^{a,b,c} Pruttipong Tsuppayakorn-aeck,^{a,b} Teerachote Pakornchote,^{a,b}
Annop Ektarawong,^{a,b} Udomsilp Pinsook,^{a,b} and Thiti Bovornratanarak^{a,b}

^a Extreme Conditions Physics Research Laboratory, Physics of Energy Materials Research Unit,
Department of Physics, Faculty of Science, Chulalongkorn University, Bangkok 10330, Thailand

^b Thailand Center of Excellence in Physics, Commission on Higher Education, Bangkok 10400, Thailand

^c Department of Physics, Mahidol Wittayanusorn School, Nakhon Pathom 73170, Thailand

E-mail: thiti.b@chula.ac.th

Index	Page
Table S1. Structural data of Na ₂ O ₂ for the <i>Amm2</i> , the <i>P-62m</i> , the <i>P2₁/c</i> , the <i>Pbam</i> , the <i>Pbam</i> [‡] , the <i>C2/m</i> , the <i>Pmmm</i> , the <i>Immm</i> , and the <i>I4/mmm</i> structures at different pressures.....	S3
Table S2. Calculated bulk and shear moduli of the <i>Amm2</i> and the <i>P-62m</i> structures at ambient pressure.....	S4
Table S3. Vibrational modes, activity in the IR and Raman spectra for the <i>P-62m</i> structure at 0 GPa, the <i>Amm2</i> structure at 0 GPa, the <i>P2₁/c</i> structure at 22 GPa, and the <i>Pbam</i> structure at 30GPa.....	S4
Figure S1. Simulated XRD pattern of various structures: (a) the <i>Amm2</i> and <i>P-62m</i> structures at ambient pressure and the intensity difference between both structures, (b) the <i>P2₁/c</i> and <i>Pbam</i> structures at 22 GPa.....	S5
Figure S2. Phonon dispersion curves and the phonon density of states for the various structures with different pressures: (a) the <i>P2₁/c</i> structure at 22 GPa, (b) the <i>P2₁/c</i> structure at 30 GPa, (c) the <i>Pbam</i> structure at 28 GPa, and (d) the <i>Pbam</i> structure at 300 GPa.....	S5
Figure S3. Simulated XRD pattern of the crystal structures: (a) Na ₂ O ₂ -II at 550°C and ambient pressure obtained from the Tallman's experiment and (b) - (c) are the <i>I4/mmm</i> and <i>P4/mmb</i> structures at 0 K and ambient pressure obtained from the AIRSS searches.....	S6
Figure S4. Simulated XRD pattern of the crystal structures: (a) Na ₂ O ₂ -Q at room temperature and ambient pressure obtained from the Tallman's experiment and (b) - (d) are the <i>Pmmm</i> , <i>Immm</i> , and <i>C2/m</i> structures at 0 K and ambient pressure obtained from the AIRSS searches, respectively.....	S6
Figure S5. Relative enthalpy of the various structures that obtained from the AIRSS searches with respect to the <i>Pbam</i> structure in the pressure range of 0-100 GPa.....	S6

Figure S6. Plot of the angles in the same hexagonal Na layer of (a) $2\times2\times2$ supercell of the $P-62m$ structure and (b) $2\times2\times2$ supercell of the primitive $Amm2$ structure <i>versus</i> pressure from 0 to 20 GPa.....	S7
Figure S7. Plot of the differences in the angles between the $Amm2$ and the $P-62m$ structures at different pressures.....	S8
Figure S8. Plots of total energy (E), enthalpy (H), and work done (PI) of the $Amm2$ structure with respect to the $P-62m$ structure <i>versus</i> pressure.....	S8
Figure S9. (a) Plot of the normalized lattice constants <i>versus</i> pressure ranging from 0 to 300 GPa. The a_0 , b_0 , and c_0 are the lattice constants at 0 GPa of the $Amm2$ structure. (b) The volume change of the three structures of Na_2O_2 at the pressure range of 0-300 GPa.....	S9
Figure S10. (a) and (b) are the plots of normalized interatomic distances <i>versus</i> pressure for the O-O and Na-O in the $Amm2$ structure as (c), respectively. (d) Plot of the cell angle (γ) of the primitive $Amm2$ structure <i>versus</i> pressure from 0 to 21 GPa.....	S9
Figure S11. The peroxide group alignment in each phase of Na_2O_2 : (a) the $Amm2$ structure at 0 GPa, (b) the $P-62m$ structure at 0 GPa, (c) the $P2_1/c$ structure at 22 GPa, (d) the $Pbam$ structure at 28 GPa, (e) the $Pbam$ structure at 100 GPa, and (f) the $Pbam$ structure at 300 GPa.....	S10
Figure S12. Plot of Gibbs free energy <i>versus</i> temperature for the structures at different pressures: (a) 0 and 10 GPa, (b) 22 GPa, and (c) 28 GPa.....	S11
Figure S13. Comparison of the atomic displacement between the Tallman's and Föppl's structures using the COMPSTRU program.....	S12
Figure S14. (a) Simulated XRD pattern of the Tallman's and Föppl's structures, (b) – (c) the (100) plane of the Tallman's and Foppl's structures.....	S12
Figure S15. Relative Gibbs free energy (ΔG) of the $P-62m$ structure for the Tallman's and the Föppl's structures as a function of temperature at ambient pressure.....	S12
Figure S16. Plot of the universal elastic anisotropy index (A^U) for the $P-62m$ and the $Amm2$ structures <i>versus</i> pressure from 0 to 15 GPa.....	S13
Figure S17. Plot of the phonon free energy for the $Amm2$ structure <i>versus</i> volume at the different temperatures: (a) 55 K and (b) 296 K.....	S13

Table S1. Structural data of Na₂O₂ for the *Amm2*, the *P-62m*, the *P2₁/c*, the *Pbam*, the *Pbam*[‡], the *C2/m*, the *Pmmm*, the *Immm*, and the *4/mmm* structures at different pressures.

Pressure (GPa)	Structure/ Space group	Lattice parameter				Atom	Site	Atomic coordinates (fractional)		
		a (Å)	b (Å)	c (Å)	β (degree)			x	y	z
0	<i>Amm2</i>	4.5103	10.8856	6.2819	90.000	O1	4c	-0.32927	0.00000	0.00000
						O2	8f	-0.17247	-0.16666	-0.50001
						Na1	4e	-0.50000	0.31748	-0.31745
						Na3	2b	-0.50000	0.00000	-0.36509
						Na2	4d	0.00000	0.15033	-0.15030
						Na4	2a	0.00000	0.00000	-0.69939
0	<i>P-62m</i>	6.2821	6.2821	4.5114	90.000	O1	2e	0.00000	0.00000	-0.17069
						O2	4h	0.33333	0.66667	0.32758
						Na1	3f	0.36521	0.00000	0.00000
						Na2	3g	0.69951	0.00000	0.50000
						O1	2e	0.00000	0.00000	0.16700
						O2	4h	0.33333	0.66667	0.33300
(RT)	<i>P-62m</i> [†]	6.22	6.22	4.47	90.000	Na1	3f	0.36600	0.00000	0.00000
						Na2	3g	0.72400	0.00000	0.50000
						O1	2e	0.00000	0.00000	0.33200
						O2	4h	0.33333	0.66667	0.16800
						Na1	3f	0.29500	0.00000	0.00000
						Na2	3g	0.63200	0.00000	0.50000
22	<i>Amm2</i>	4.1525	10.0080	5.7770	90.000	O1	4c	-0.32112	0.00000	-0.00004
						O2	8f	-0.18321	-0.16664	-0.50003
						Na1	4e	-0.50000	0.31479	-0.31479
						Na3	2b	-0.50000	0.00000	-0.37038
						Na2	4d	0.00000	0.14821	-0.14814
						Na4	2a	0.00000	0.00000	-0.70358
22	<i>P2₁/c</i>	4.6421	6.8657	3.6366	141.5836	O	4e	-0.49999	0.09003	-0.37268
						Na	4e	-0.00005	0.14092	0.66039
28	<i>P2₁/c</i>	4.5710	6.7955	3.5762	141.4831	O	4e	-0.49999	0.09082	-0.37198
						Na	4e	-0.00003	0.14129	0.65621
28	<i>Pbam</i>	3.5758	6.7976	2.8461	90.000	O	4g	0.12768	0.09089	0.00000
						Na	4h	0.15598	0.35872	0.50000
28	<i>Pbam</i> [‡]	3.5743	6.8001	2.8459	90.000	O	4h	-0.37247	0.40908	-0.50000
						Na	4g	-0.84429	0.35871	0.00000
100	<i>Pbam</i>	3.2502	6.2622	2.6121	90.000	O	4g	0.13333	0.09480	0.00000
						Na	4h	0.14474	0.35744	0.50000
200	<i>Pbam</i>	3.0575	5.9049	2.4648	90.000	O	4g	0.13688	0.09644	0.00000
						Na	4h	0.14323	0.35714	0.50000
300	<i>Pbam</i>	2.9413	5.6869	2.3741	90.000	O	4g	0.13887	0.09716	0.00000
						Na	4h	0.14311	0.35701	0.50000
0	<i>C2/m</i>	7.8720	3.6262	6.1470	141.4238	O	4i	0.84296	-0.00000	1.34321
						Na	4i	0.49998	-0.00000	0.72763
0	<i>Pmmm</i>	6.1415	3.6241	4.9218	90.000	O1	2s	0.50000	0.00000	0.15645
						O2	2r	0.00000	0.50000	0.65645
0	<i>Immm</i>	6.1414	3.6242	4.9213	90.000	Na1	2j	0.22799	0.00000	0.50000
						Na2	2k	0.72796	0.50000	0.00000
0	<i>4/mmm</i>	4.4082	4.4082	5.3359	90.000	O	4j	0.50000	0.00000	-0.84353
						Na	4f	0.22797	0.00000	-0.50000
0	<i>4/mmm</i>	4.4082	4.4082	5.3359	90.000	O	4e	-0.50000	-0.50000	0.35278
						Na	4c	0.00000	0.50000	1.00000

[†] The *P-62m* structure of Na₂O₂ at ambient conditions proposed by Tallman *et al.*²

[‡] The *P-62m* structure of Na₂O₂ at ambient conditions proposed by Föppl.³

[‡] The relaxed *Pbam* structure of Na₂O₂ predicted by Deng *et al.*¹

RT: Room temperature

Table S2. Calculated bulk and shear moduli of the *Amm*2 and the *P*-62*m* structures at ambient pressure.

Structure/Space group	Condition	Bulk modulus (GPa)		Shear modulus (GPa)	
		GGA-PBE	LDA	GGA-PBE	LDA
<i>Amm</i> 2		49.16	64.53	31.33	38.98
<i>P</i> -62 <i>m</i>		48.95	64.72	29.88	37.59
<i>P</i> -62 <i>m</i> ⁴	0 GPa, 0 K	56.40	-	-	-
<i>P</i> -62 <i>m</i> ⁵		49	-	29	-
<i>P</i> -62 <i>m</i> [†]	ambient	55.82	40.19	31.14	27.92
<i>P</i> -62 <i>m</i> [§]		57.16	41.37	32.11	29.08

[†]using the structural data proposed by Tallman *et al.*²

[§]using the structural data proposed by Föppl *et al.*³

Table S3. Vibrational modes, activity in the IR and Raman spectra for the *P*-62*m* structure at 0 GPa, the *Amm*2 structure at 0 GPa, the *P*2₁/*c* structure at 22 GPa, and the *Pbam* structure at 30 GPa.

Structure/ Space group	Vibrational mode	Frequency (cm ⁻¹)	Activity	Structure/ Space group	Vibrational mode	Frequency (cm ⁻¹)	Activity
<i>P</i> -62 <i>m</i> at 0 GPa	A'' ₂	129.3, 195.0, 292.3	IR	<i>Amm</i> 2 at 0 GPa	B ₁	223.4, 232.7, 260.4, 263.6, 271.8, 305.1, 314.5, 340.8, 779.1	IR / Raman
	E'	169.6, 203.4, 224.9, 286.4, 316.3, 338.0	IR / Raman			225.0, 227.9, 254.8, 261.4, 283.8, 294.0, 331.3, 337.2	
	A'' ₁	125.5	-			124.4, 183.2, 216.3, 219.7, 251.0, 227.9, 295.3, 310.9, 317.7, 331.4, 784.9, 821.5	
	A' ₁	169.6, 278.8, 780.8, 821.0	Raman		B ₂	108.2, 196.7, 210.9, 262.2, 307.5, 358.0	Raman
	A ₂	180.8, 275.5, 773.5	-		A ₁	405.9 262.5, 318.7, 449.9 305.9, 332.1, 432.6 199.3, 359.7 236.2, 361.1	IR / Raman
	E''	221.8, 226.5, 242.5, 293.2, 318.4	Raman		A ₂	255.8, 336.8	Raman
	A _u	194.9, 237.2, 295.1, 341.9, 418.5	IR		B _{1u}	405.9 262.5, 318.7, 449.9 305.9, 332.1, 432.6 199.3, 359.7 236.2, 361.1	IR
	B _u	262.3, 305.3, 380.5, 415.3	IR		B _{2u}	255.8, 336.8	Raman
	A _g	237.2, 268.2, 345.8, 372.8, 433.3, 780.8	Raman		B _{3u}	274.1, 378.0, 461.3, 815.3	Raman
	B _g	244.4, 337.4, 348.7, 412.6, 440.0, 766.6	Raman		A _g	357.6, 431.6, 461.9, 797.2	Raman
					B _{1g}		

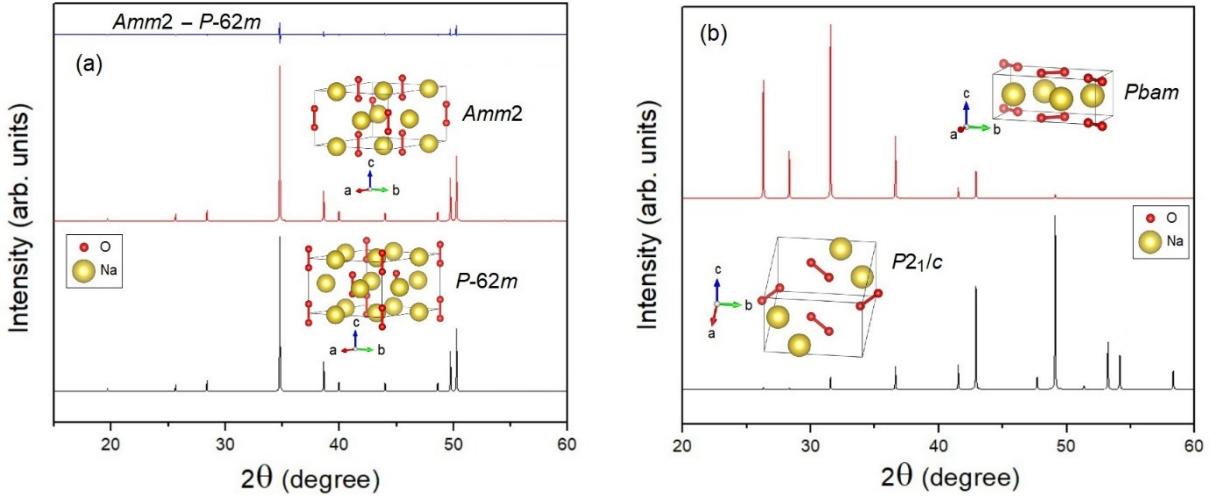


Figure S1. Simulated XRD pattern of various structures: (a) the *Amm2* and *P-62m* structures at ambient pressure and the intensity difference between both structures, (b) the *P21/c* and *Pbam* structures at 22 GPa.

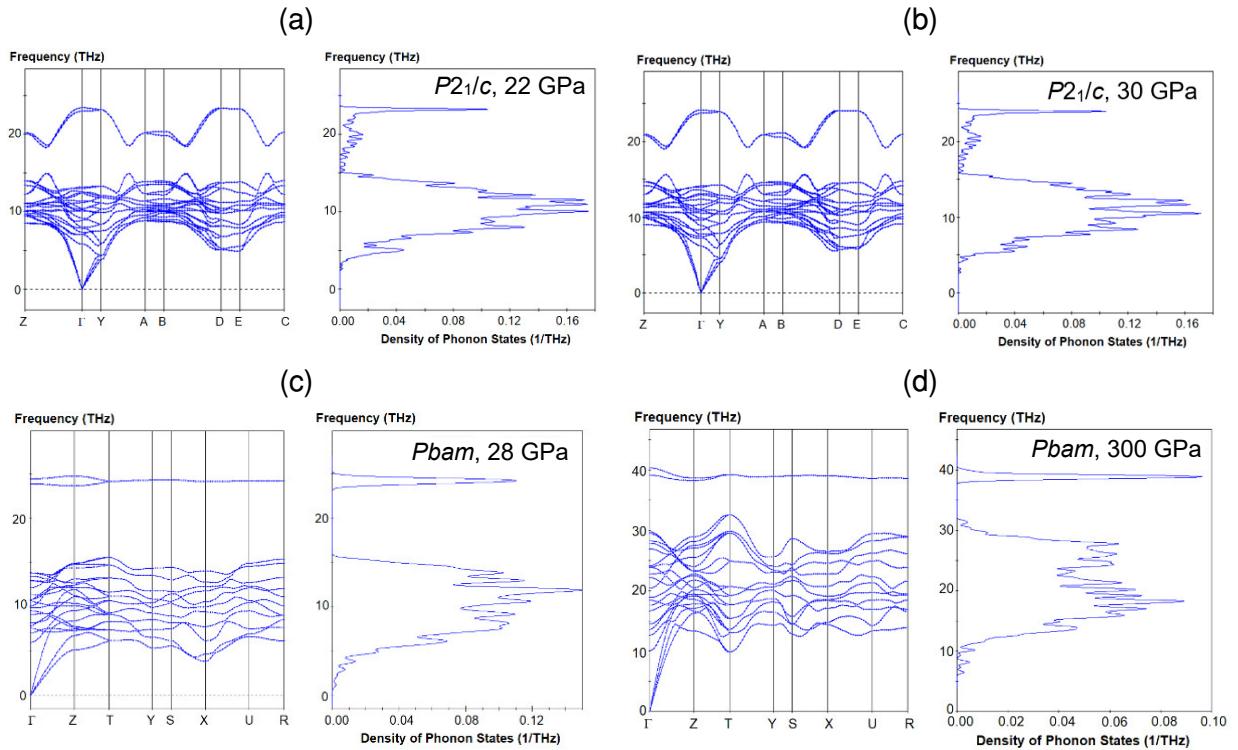


Figure S2. Phonon dispersion curves and the phonon density of states for the various structures with different pressures: (a) the *P21/c* structure at 22 GPa, (b) the *P21/c* structure at 30 GPa, (c) the *Pbam* structure at 28 GPa, and (d) the *Pbam* structure at 300 GPa.

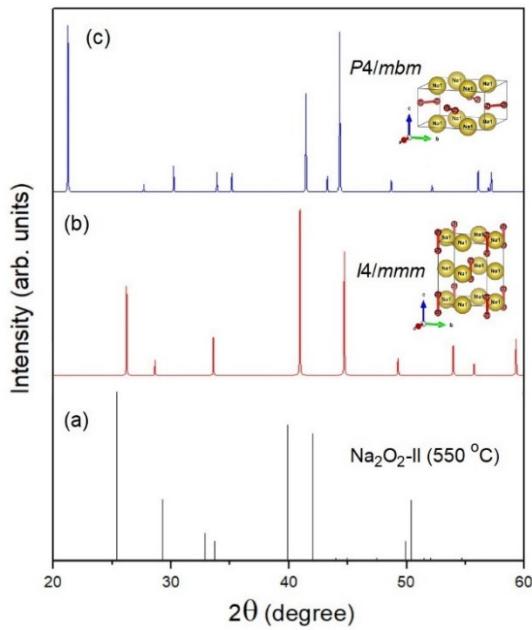


Figure S3. Simulated XRD pattern of the crystal structures: (a) Na_2O_2 -II at 550°C and ambient pressure obtained from the Tallman's experiment⁶ and (b) - (c) are the $\text{I}4/mmm$ and $\text{P}4/mbm$ structures at 0 K and ambient pressure obtained from the AIRSS searches.

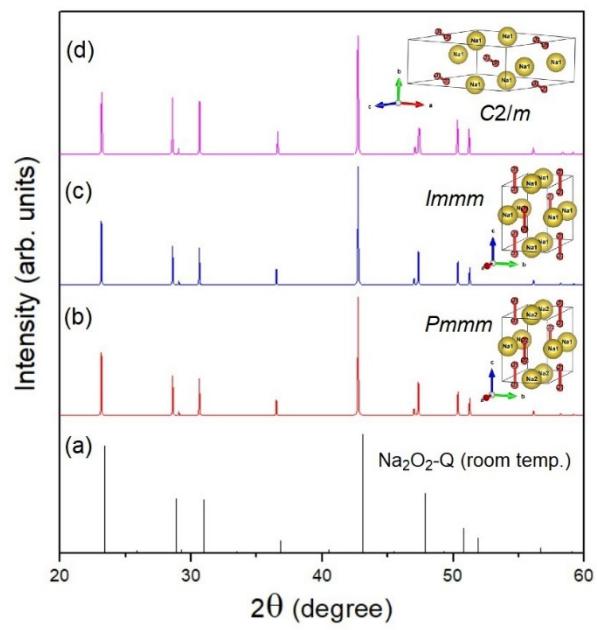


Figure S4. Simulated XRD pattern of the crystal structures: (a) Na_2O_2 -Q at room temperature and ambient pressure obtained from the Tallman's experiment⁶ and (b) - (d) are the Pmmm , Immm , and $\text{C}2/m$ structures at 0 K and ambient pressure obtained from the AIRSS searches, respectively.

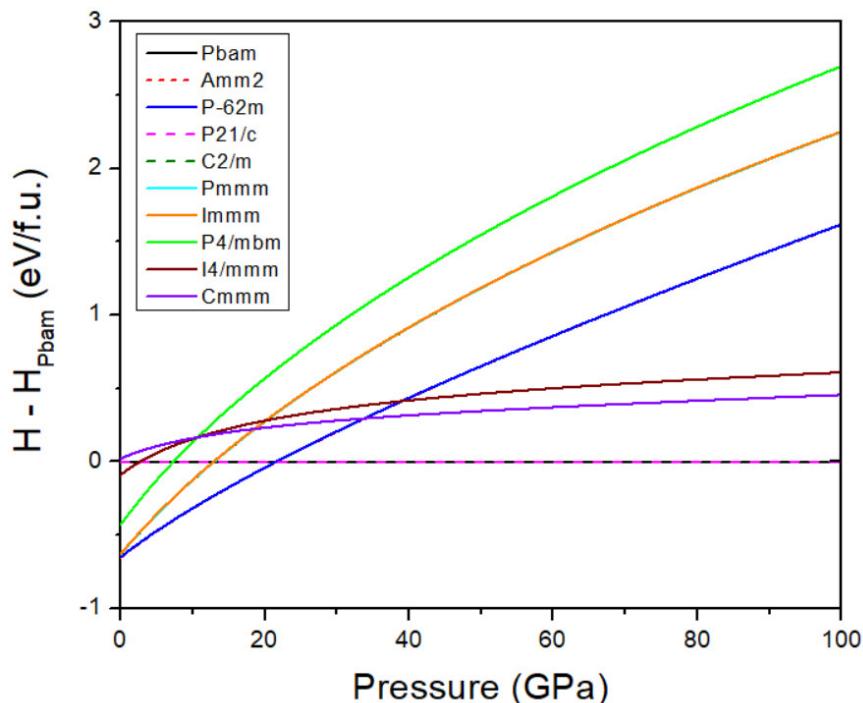


Figure S5. Relative enthalpy of the various structures that obtained from the AIRSS searches with respect to the Pbam structure in the pressure range of 0-100 GPa.

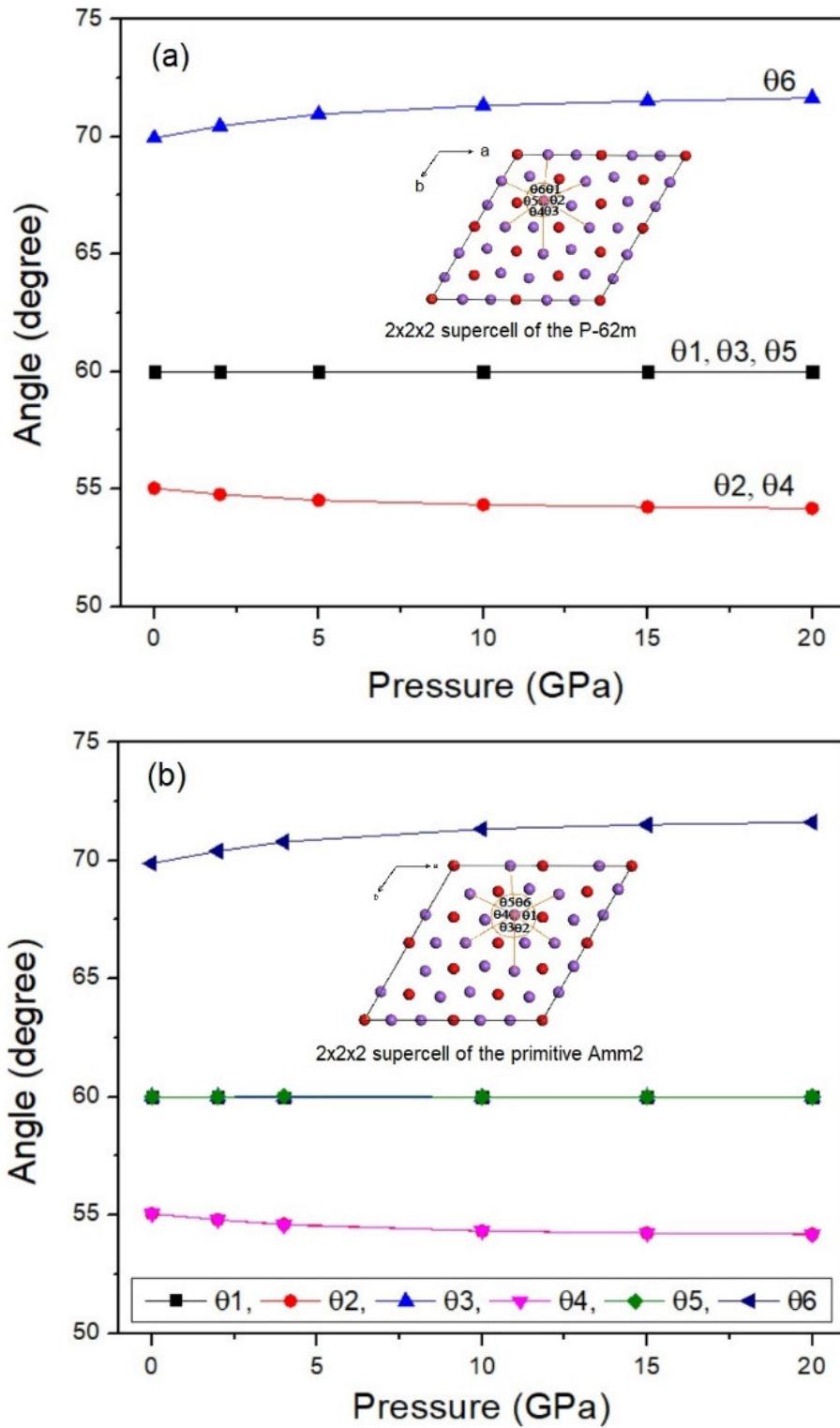


Figure S6. Plot of the angles in the same hexagonal Na layer of (a) $2\times2\times2$ supercell of the *P-62m* structure and (b) $2\times2\times2$ supercell of the primitive *Amm2* structure *versus* pressure from 0 to 20 GPa.

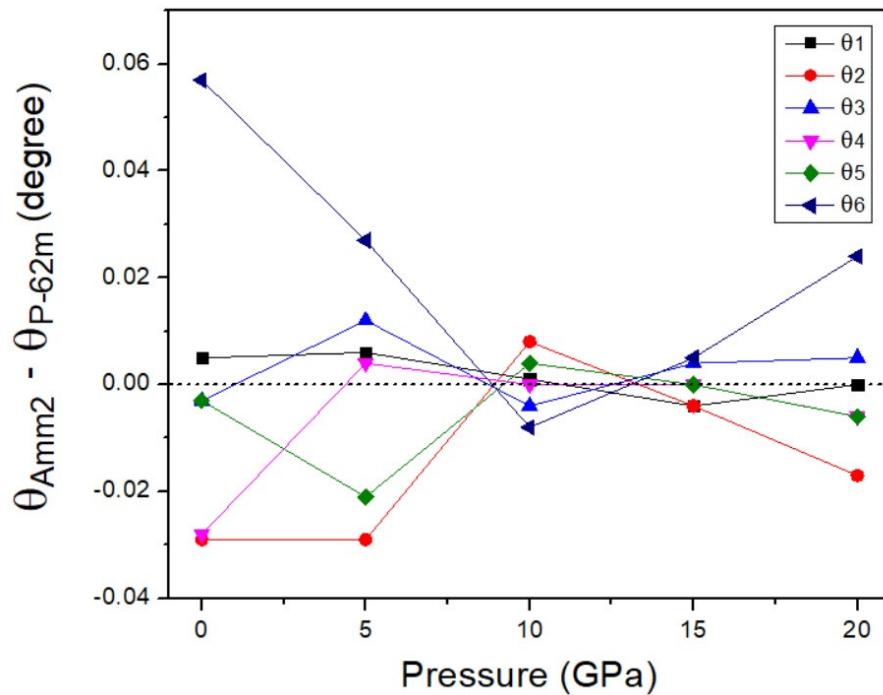


Figure S7. Plot of the differences in the angles between the Amm2 and the P-62m structures at different pressures.

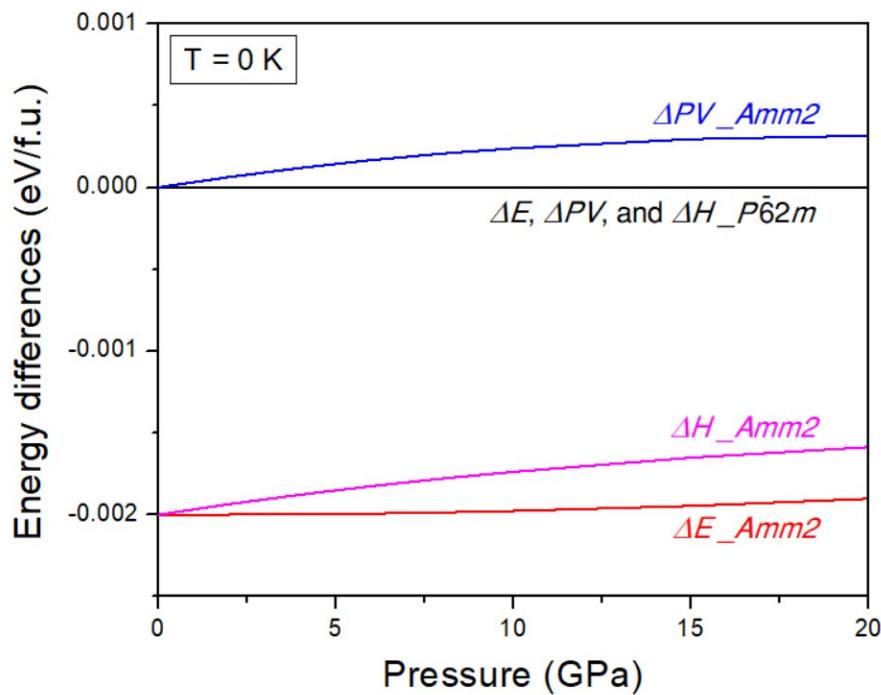


Figure S8. Plot of total energy (E), enthalpy (H), and work done (PV) of the Amm2 structure with respect to the P-62m structure *versus* pressure.

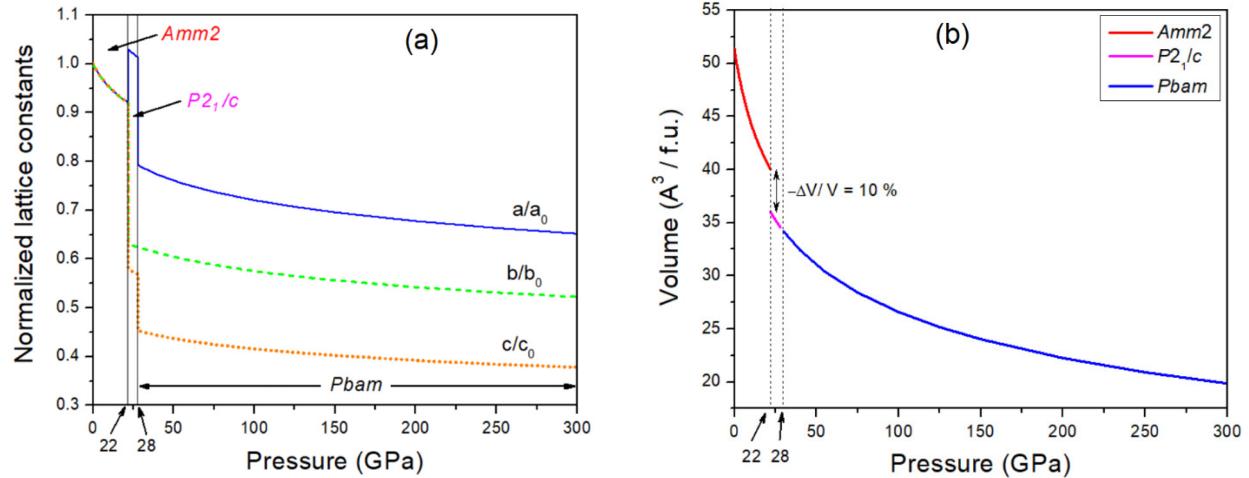


Figure S9. (a) Plot of the normalized lattice constants *versus* pressure ranging from 0 to 300 GPa. The a_0 , b_0 , and c_0 are the lattice constants at 0 GPa of the *Amm2* structure. (b) The volume change of the three structure of Na_2O_2 at the pressure range of 0-300 GPa: the *Amm2*, the *P2₁/c*, and the *Pbam* structures.

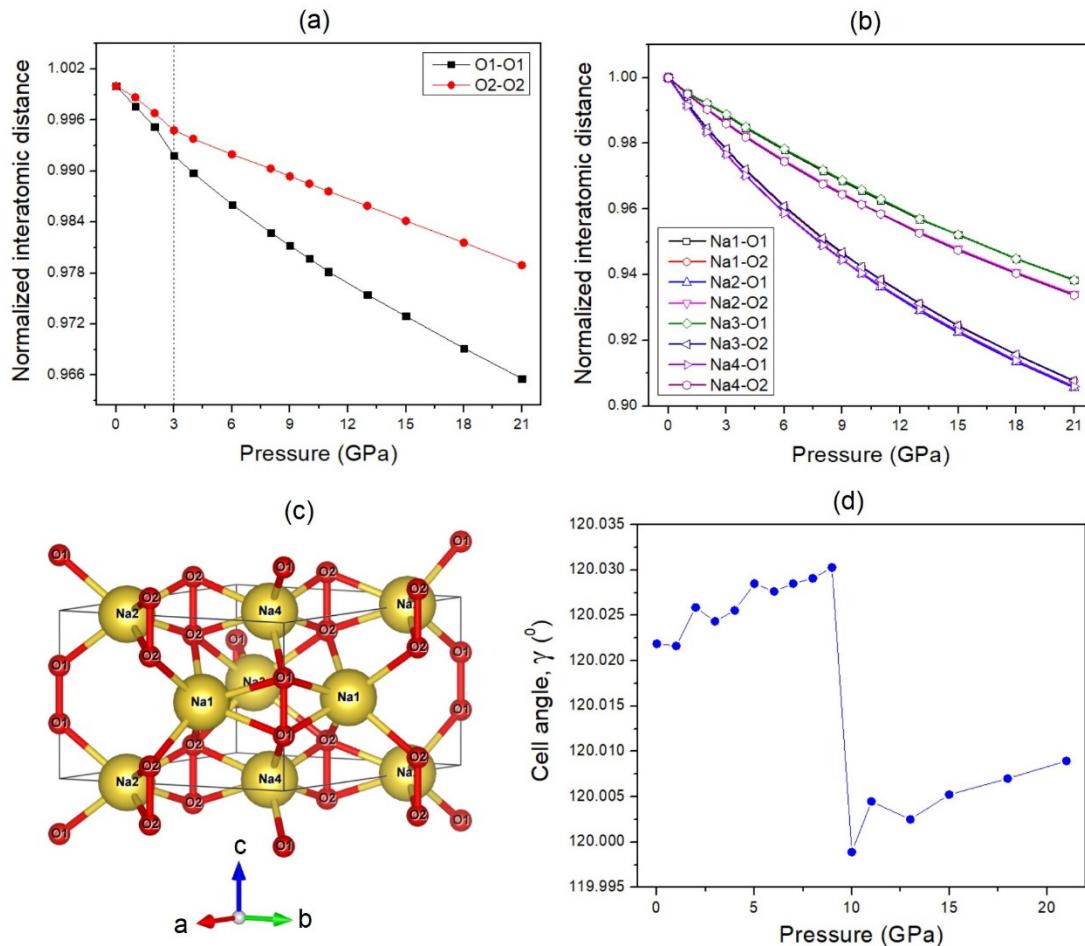


Figure S10. (a) and (b) are the plots of normalized interatomic distances *versus* pressure for the O-O and Na-O in the *Amm2* structure as (c), respectively. (d) Plot of the cell angle (γ) of the primitive *Amm2* structure *versus* pressure from 0 to 21 GPa.

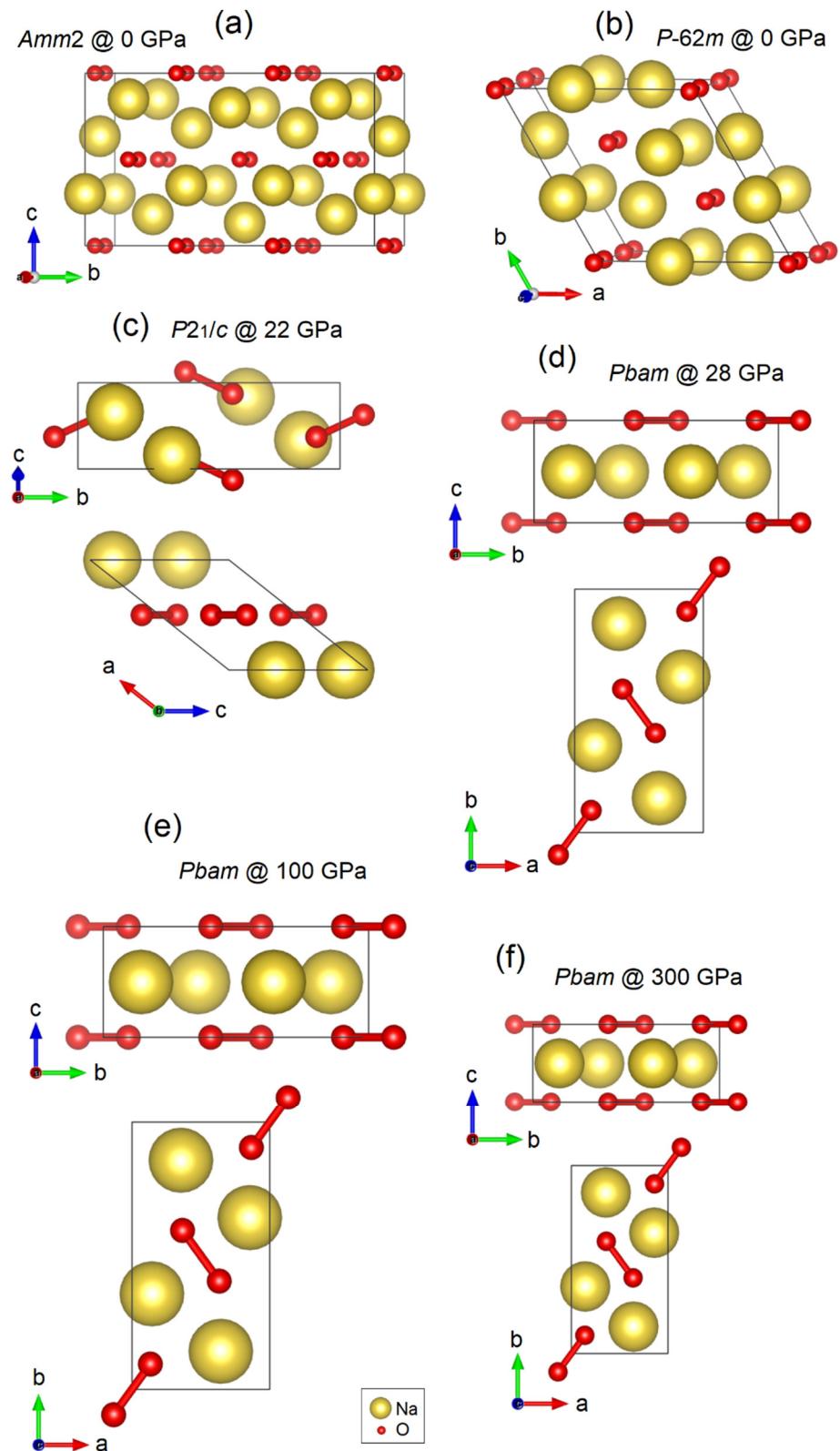


Figure S11. The peroxide group alignment in each phase of Na_2O_2 : (a) the $Amm\bar{2}$ structure at 0 GPa, (b) the $P-62m$ structure at 0 GPa, (c) the $P2_1/c$ structure at 22 GPa, (d) the $Pbam$ structure at 28 GPa, (e) the $Pbam$ structure at 100 GPa, and (f) the $Pbam$ structure at 300 GPa.

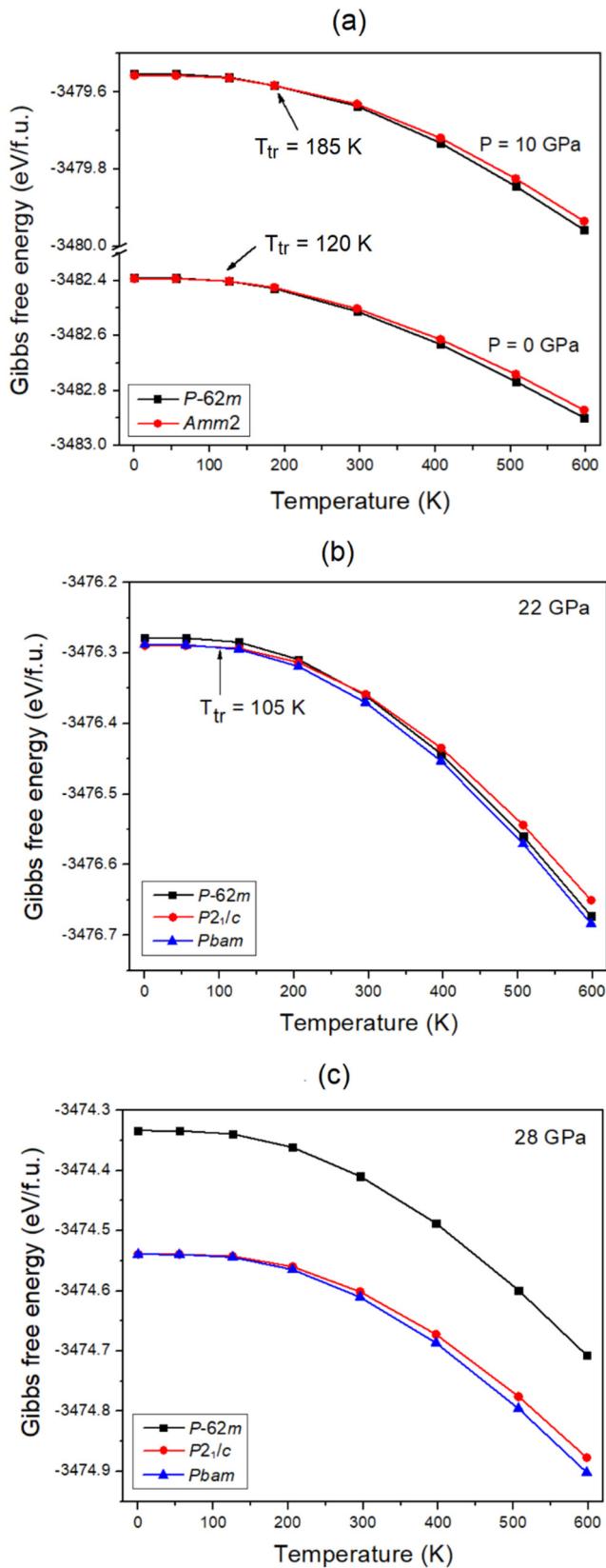


Figure S12. Plot of Gibbs free energy *versus* temperature for the structures at different pressures: (a) 0 and 10 GPa, (b) 22 GPa, and (c) 28 GPa.

WP	Atom	Atomic Displacements				
		u_x	u_y	u_z	$ u $	
3f	(x,0,0)	Na2	0.0020	0.0000	0.0000	0.0124
2e	(0,0,z)	O1	0.0000	0.0000	0.0010	0.0045
3g	(x,0,1/2)	Na1	-0.0190	0.0000	0.0000	0.1182
4h	(1/3,2/3,z)	O2	0.0000	0.0000	-0.0010	0.0045

NOTE: u_x , u_y and u_z are given in relative units. $|u|$ is the absolute distance given in Å

Available from: <http://www.cryst.ehu.es/cgi-bin/cryst/programs/nph-compSTRU>

Figure S13. Comparison of the atomic displacement between the Tallman's and Föppl's structures using the COMPSTRU program.⁷

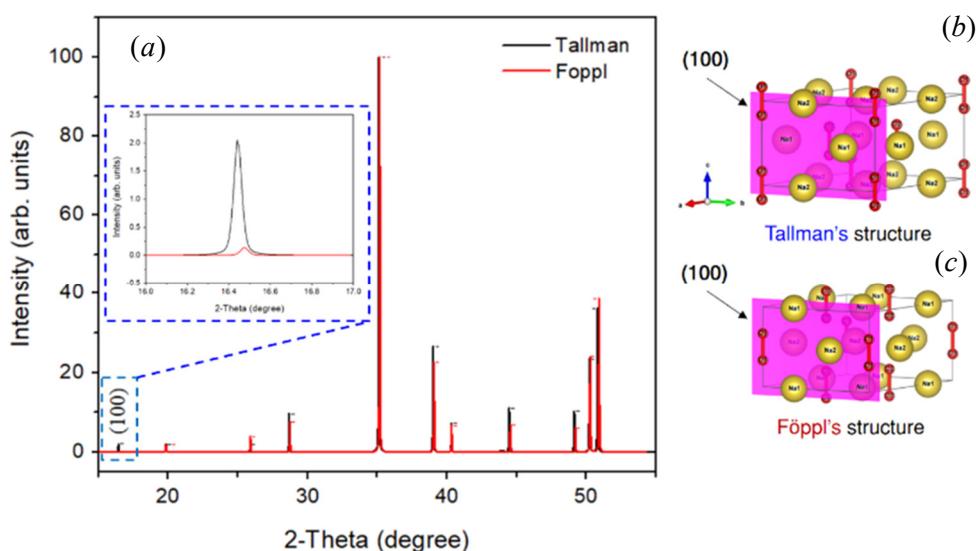


Figure R14. (a) Simulated XRD pattern of the Tallman's and Föppl's structures, (b) – (c) the (100) plane of the Tallman's and Föppl's structures.

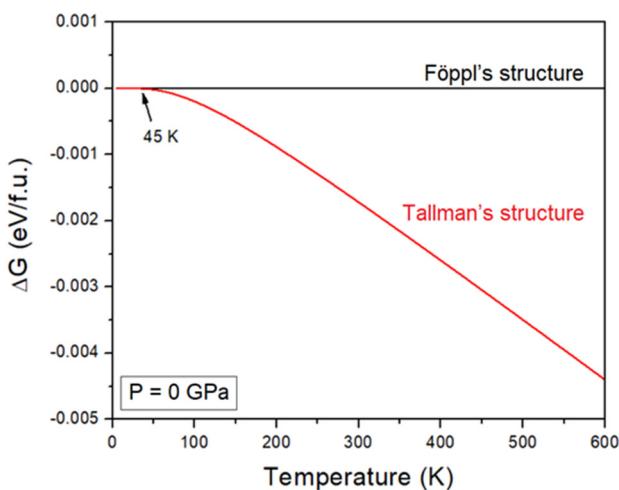


Figure S15. Relative Gibbs free energy (ΔG) of the $P-62m$ structure between the Tallman's and the Föppl's structures as a function of temperature at ambient pressure.

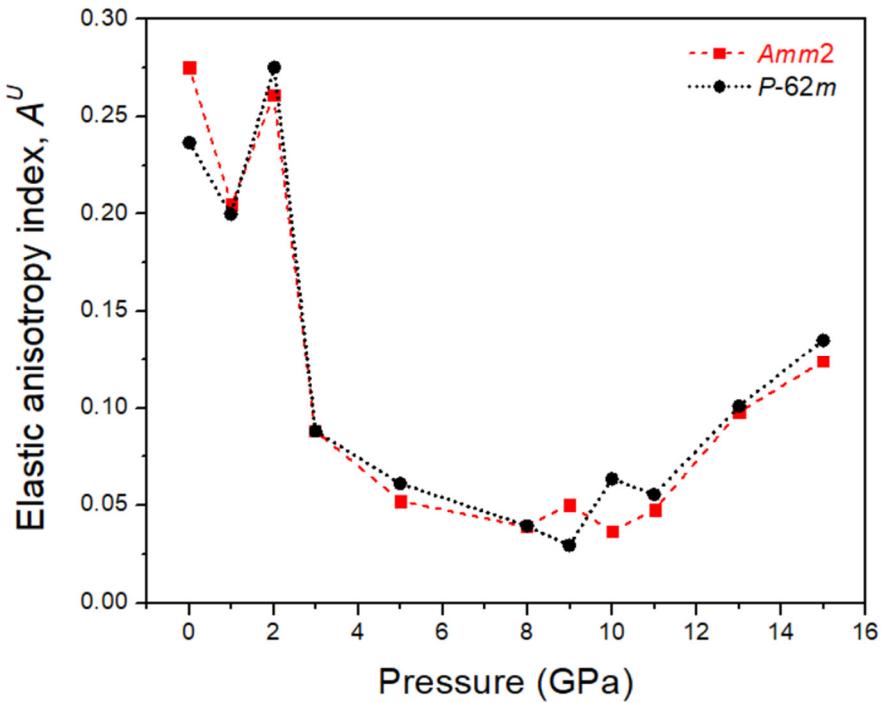


Figure S16. Plot of the universal elastic anisotropy index (A^U) for the *P-62m* and the *Amm2* structures *versus* pressure from 0 to 15 GPa.

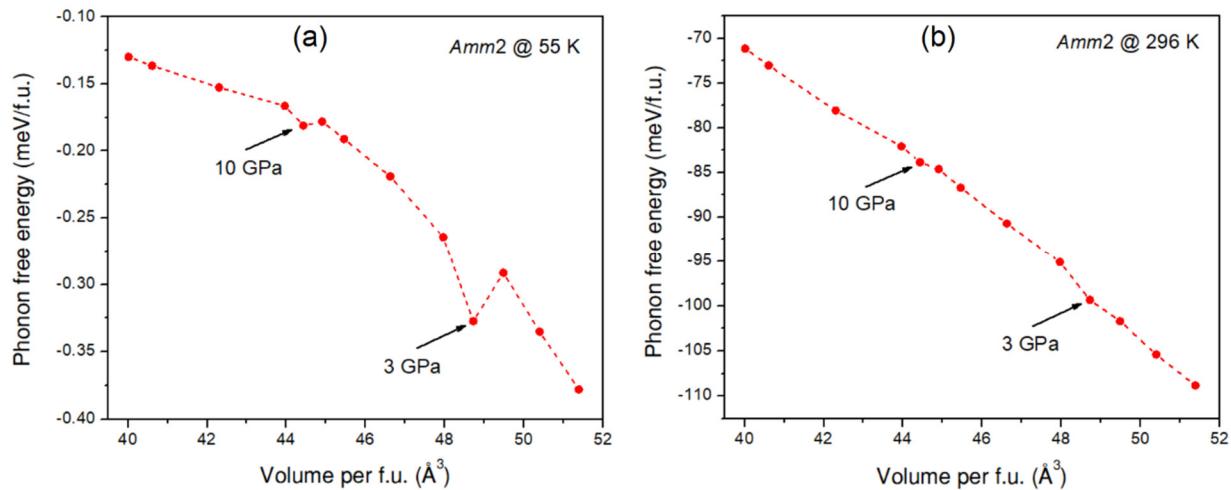


Figure S17. Plot of the phonon free energy for the *Amm2* structure *versus* volume at the different temperatures: (a) 55 K and (b) 296 K.

References

- [1] N. Deng, W. Wang, G. Yang and Y. Qiu, *RSC Adv.*, 2015, **5**, 104337-104342.
- [2] R. L. Tallman, J. L. Margrave and S. W. Bailey, *J. Am. Chem. Soc.*, 1957, **79**, 2979-2980.
- [3] H. Föppl, *Z. Anorg. Allg. Chem.*, 1957, **291**, 12-50.
- [4] M. V. Aleinikova, Y. N. Zhuravlev and D. V. Korabelnikov, *Russ. Phys. J.*, 2012, **55**, 495-500.
- [5] M. de Jong, W. Chen, T. Angsten, A. Jain, R. Notestine, A. Gamst, M. Sluiter, C. K. Ande, S. van der Zwaag, J. J. Plata, C. Toher, S. Curtarolo, G. Ceder, K. A. Persson and M. Asta, *Sci. Data.*, 2015, **2**, 150009.
- [6] R. L. Tallman and J. L. Margrave, *J. Inorg. Nucl. Chem.*, 1961, **21**, 40–44.
- [7] G. de la Flor, D. Orobengoa, E. Tasici, J. M. Perez-Mato and M. I. Aroyo, *J. Appl. Cryst.*, 2016, **49**, 653-664.