Supporting information:

A novel water-soluble naked-eye probe with a large Stokes Shift for selective optical sensing of Hg²⁺ and its application in water samples and living cells

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Contents:

1. Synthesis of Hcy-pP

A solution of compound 2 (0.160 g, 0.5 mmol), 4-dihydroxy-1-benzaldehyd(0.249 g, 1.5 mmol) and p-methyl benzenesulfonic acid (0.175g, 1.0 mmoL) in anhydrous ethanol (10.0 mL) were refluxed for 3 h, then the mixture was cooled to room temperature. The precipitate was filtered off, washed with anhydrous ethanol and dried in vacuum to give **Hcy-pP** as a red solid mixture. (0.211 g, 73%). mp = $227\sim228$ °C. The structure of Hcy-mP was confirmed by ¹H NMR): ¹H NMR (600 MHz, DMSO-*d*₆): δ 1.61 (s, 6H, CH₃), 6.93-6.94 (d, J = 9.8 Hz, 1H, Ar-H), 7.10-7.12 (d, J = 15.5 Hz, 1H, CH=CH), 7.26-7.29 (d, J = 9.7 Hz, 1H, Ar-H), 7.47-7.48 (d, J = 8.9 Hz, 1H, Ar-H), 7.51-7.52 (d, J = 8.7 Hz, 1H, Ar-H), 7.72-7.73 (d, J = 9.1 Hz, 1H, ArH), 7.83-7.84 (d, J = 9.1 Hz, 1H, ArH), 7.90 (s, 1H, Ar-H), 8.18-8.21 (d, J = 15.6 Hz, 1H, CH=CH), 10.63 (s, 1H, OH).



Scheme S.1 The synthetic routes of probe Hcy-pP

2. Supplementary Figures

Figure S-1 Benesi-Hildebrand plot from fluorescence titration data of **Hcy-mP** (10 μ M) in HEPES buffer solution.

Figure S-2 The Stern-Volmer curve of Hcy-mP with different concentrations of Hg²⁺.

Figure S-3 The effect of pH on the fluorescence intensity changes of Hcy-mP and Hcy-mP + Hg^{2+} at 517 nm in HEPES buffer solution.

Figure S-4 ESI-MS of $[Hcy-mP + Hg^{2+}]^{-}$.

Figure S-5 ESI-MS of $[Hcy-mP + Hg^{2+} + K^{+} + H^{+}]^{+}$.

Figure S-6 ¹H NMR of Hcy-pP

Figure S-7¹H NMR of Hcy-mP

Figure S-8¹³C NMR of Hcy-mP

Figure S-9 IR of Hcy-mP

3. Table S1 Application in water samples

Table S1Analytical results of Hg^{2+} in water samples

Sample	Added (µM)	$Deteced(x^a \pm SD^b)(\mu M)$	Recovery(%)	RSD(%)
Tap Water	1.00	$\textbf{0.99} \pm \textbf{0.013}$	99	1.2
	3.00	$\textbf{3.11}\pm\textbf{0.018}$	103	0.9
	5.00	$\textbf{4.98} \pm \textbf{0.021}$	99.6	0.7
Drinking Water	1.00	$\textbf{1.01}\pm\textbf{0.014}$	100.5	0.8
	3.00	$\textbf{2.85} \pm \textbf{0.021}$	95	1.1
	5.00	5.11 ± 0.031	102.5	1.3
Lake Water	1.00	$\textbf{0.97} \pm \textbf{0.027}$	97	1.0
	3.00	$\textbf{3.08} \pm \textbf{0.011}$	102	1.3
	5.00	5.03 ± 0.027	100	2.3

^a Mean of three determination

^b SD: Standard deviation



Figure S.1 Benesi-Hildebrand plot from fluorescence titration data of **Hcy-mP** (10 μ M) in HEPES buffer solution.



Figure S.2 The Stern-Volmer curve of Hcy-mP with different concentrations of Hg²⁺.



Fig S.3 The effect of pH on the fluorescence intensity changes of **Hcy-mP** and **Hcy-mP**+Hg²⁺ at 517 nm in HEPES buffer solution.



Fig S.4 ESI-MS of [Hcy-mP + Hg²⁺]⁻



Figure S.5 ESI-MS of $[Hcy-mP + Hg^{2+} + K^+ + H^+]^+$



Figure S.6 ¹H NMR of Hcy-pP



Figure S.7 ¹H NMR of Hcy-mP



Figure S.8¹³C NMR of Hcy-mP



Figure S.9 IR of Hcy-mP