

SUPPORTING INFORMATION

The Enzyme-like Catalytic Hydrogen Abstraction Reaction Mechanisms of Cyclic Hydrocarbons with Magnesium-Diluted Fe-MOF-74

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Table S1. Comparison of Fe-O bond lengths (\AA) of Fe/(Mg)-MOF-74 using B3LYP-D3 and M06-L-D3 methods.

Project Bond	EXP ^a	B3LYP-D3	Relative Error	M06-L-D3	Relative Error
Fe-O ₁	2.067	2.064	-0.15%	2.065	-0.10%
Fe-O ₂	2.166	2.168	0.10%	2.169	0.14%
Fe-O ₃	2.077	2.070	-0.34%	2.075	-0.10%
Fe-O ₄	2.022	2.016	-0.30%	2.017	-0.25%
Fe-O ₅	2.054	2.052	-0.10%	2.053	-0.05%

^a Parameters obtained from crystal structure of Fe/(Mg)-MOF-74.

Table S2. The Calculated ${}^5\Delta H$, ${}^3\Delta H$, ${}^5\Delta G$, ${}^3\Delta G$ for various species of the oxidation reaction of CHD catalyzed by Fe(Mg)-MOF-74.

project	species	A+N₂O +CHD	B	TS1	C	D	TS2	E	F	TS3	G	A+N₂+
												H₂O+b
${}^5\Delta H$ (kcal/mol)		0.0	-8.5	10.9	-24.1	-33.7	-30.2	-53.8	-56.1	-50.9	-88.4	-62.8
${}^3\Delta H$ (kcal/mol)		31.2	29.0	32.4	-5.2	-14.5	-9.8	-34.1	-35.6	-27.7	-61.7	-31.6
${}^5\Delta G$ (kcal/mol)		0.0	0.6	20.4	-17.0	-21.9	-17.5	-42.6	-45.1	-39.1	-78.5	-73.1
${}^3\Delta G$ (kcal/mol)		31.6	38.3	43.1	1.6	-2.2	4.7	-22.4	-23.8	-14.3	-50.3	-41.5

Table S3. The Calculated ${}^5\Delta H$ and ${}^5\Delta G$ for various species of the oxidation reaction of CHA catalyzed by Fe(Mg)-MOF-74.

project \ species	H	TS4	I	TS5	J
	${}^5\Delta H$ (kcal/mol)	0.0	8.1	-5.2	-4.3
${}^5\Delta G$ (kcal/mol)	0.0	9.2	-5.0	-4.0	-32.2

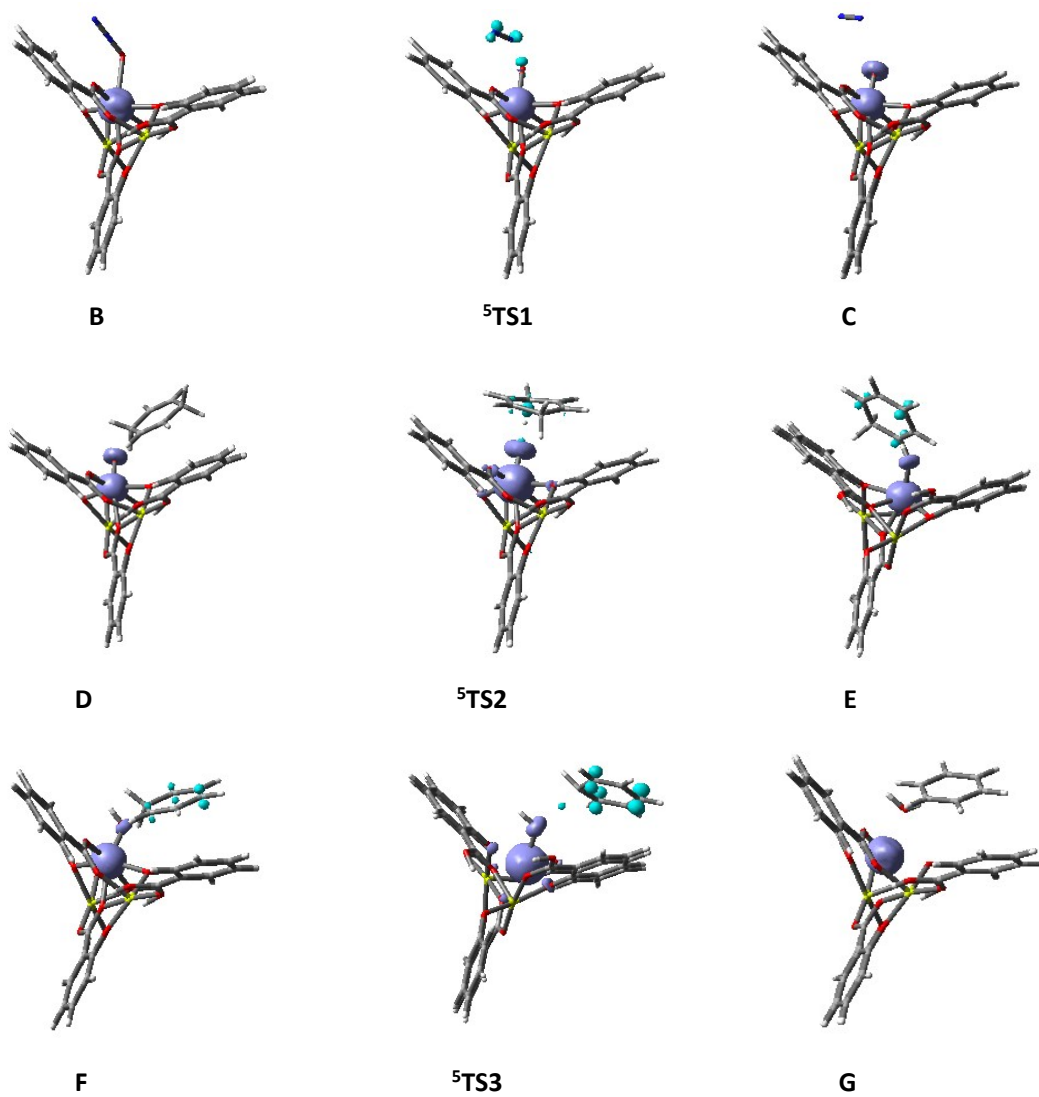
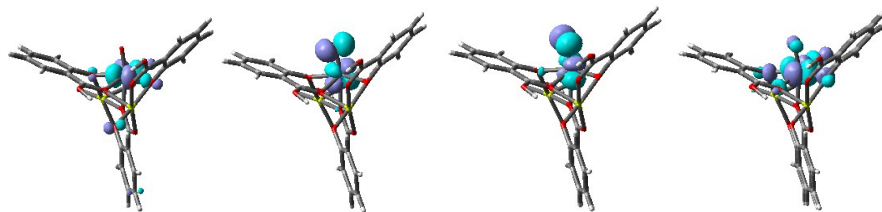


Fig. S1. The calculated spin density distribution of ⁵TS1, ⁵TS2, ⁵TS3 and B~G species.

Spin Natural Orbitals of ${}^5\mathbf{A-O}$



δ

(1.00 e)

π^*

(1.00 e)

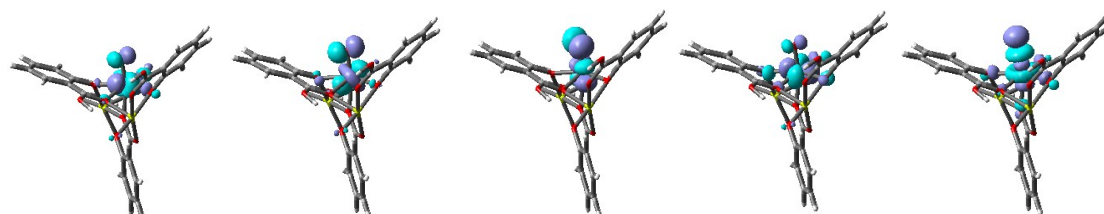
π^*

(1.00 e)

σ_{xy}^*

(1.00 e)

Natural Orbitals of ${}^5\mathbf{A-O}$



δ

(1.00 e)

π^*

(1.00 e)

π^*

(1.00 e)

σ_{xy}^*

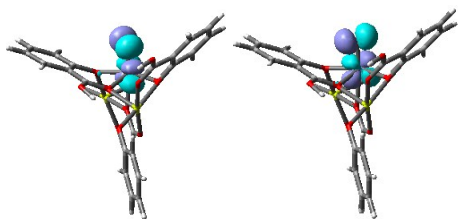
(1.00 e)

σ_z^{*2}

(0.00 e)

(a)

Spin Natural Orbitals of ${}^3\text{A-O}$



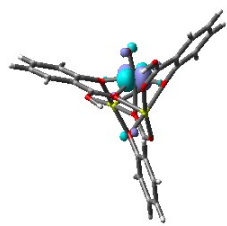
π^*

(1.00 e)

π^*

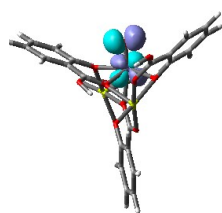
(1.00 e)

Natural Orbitals of ${}^3\text{A-O}$



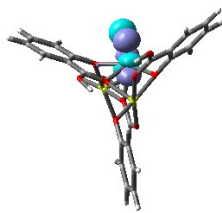
δ

(2.00 e)



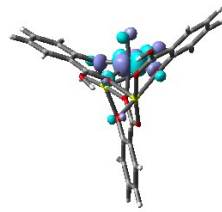
π^*

(1.00 e)



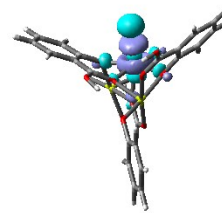
π^*

(1.00 e)



σ_{xy}^*

(0.00 e)

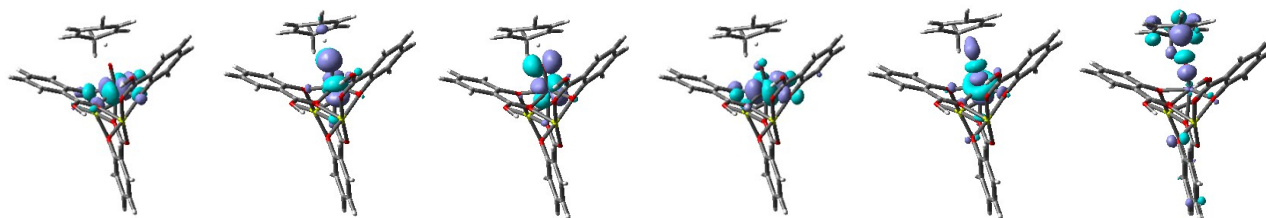


σ_z^{*2}

(0.00 e)

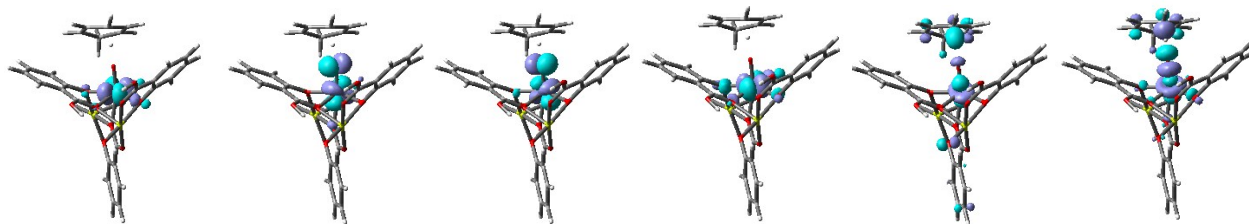
(b)

Spin Natural Orbitals of ${}^5\text{TS2}$



δ	π^*	π^*	σ_{xy}^*	σ_z^{*2}	$\phi_C+2P(O)$
(1.00 e)	(1.00 e)	(1.00 e)	(1.00 e)	(0.70 e)	(-0.70 e)

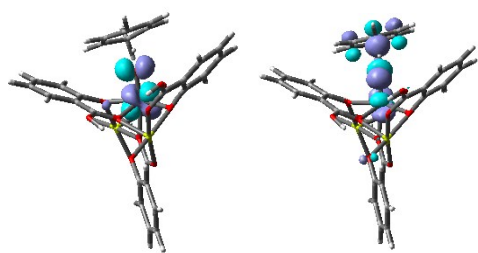
Natural Orbitals of ${}^5\text{TS2}$



δ	π^*	π^*	σ_{xy}^*	$\phi_C+\sigma_z^{*2}$	$\phi_C+\sigma_z^{*2}$
(1.00 e)	(1.00 e)	(1.00 e)	(1.00 e)	(1.77 e)	(0.23 e)

(c)

Spin Natural Orbitals of ${}^3\text{TS2}$



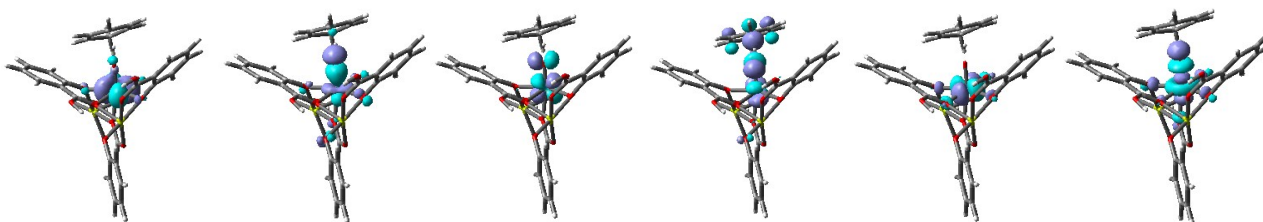
π^*

(1.00 e)

$\pi^* + \phi_C$

(1.00 e)

Natural Orbitals of ${}^3\text{TS2}$



δ

(2.00 e)

d_{yz}

(2.00 e)

π^*

(1.00 e)

$\pi^* + \phi_C$

(1.00 e)

σ_{xy}^*

(0.00 e)

σ_z^{*2}

(0.00 e)

(d)

Fig. S2. The calculated spin natural orbitals, natural orbitals and their occupations of (a) ${}^5\text{A-O}$; (b) ${}^3\text{A-O}$; (c) ${}^5\text{TS2}$; and (d) ${}^3\text{TS2}$ for the first hydrogen abstraction reaction. The negative occupation in ϕ_C denotes β spin electron.