Electronic Supplementary Information

Tunable preparation of highly dispersed Ni_xMn-LDO catalysts derived from Ni_xMn-LDHs precursors and application in lowtemperature NH₃-SCR reaction

Benhui Hou, ^a Yali Du, ^b Xuezhen Liu, ^a Chao Ci, ^a Xu Wu*a and Xianmei Xie *a

^a College of Chemistry and Chemical Engineering, Taiyuan University of Technology, Taiyuan 030024, PR China.

^b College of Chemistry and Chemical Engineening, Jinzhong University, Jinzhong 030619, PR China.

* Corresponding authors. E-mail address: wuxu@tyut.edu.cn, xxmsxty@sina.com. Fax: +86-351-6018528, 86-351-6018564.

Figure captions

Table S1 Comparison of catalytic performance of different Ni-Mn oxides catalysts.

Fig. S1 SEM image and EDS mapping of Ni₃Mn-LDO catalyst.

Fig. S2 SEM image and EDS mapping of Ni_4Mn -LDO catalyst.

Fig. S3 SEM image and EDS mapping of $\rm Ni_5Mn\text{-}LDO$ catalyst.

Fig. S4 SEM image and EDS mapping of Ni₆Mn-LDO catalyst.

Table S1 Comparison of catalytic performance of different Ni-Mn oxides catalysts

Catalysts	Preparation method	Reaction conditions	NO _x conversion/%	Ref.	
			(Tem./°C)		
Ni-MnOx	co-precipitation	500 ppm NO,500 ppm NH ₃ ,	80% (100-300 °C)	[1]	
		5 vol % O_2 in He, GHVS=64.000 h ⁻¹			
NiMn ₂ O ₄	solvothermal	500 ppm NO, 500 ppm NH ₃ ,	80% (85-250°C)	[2]	
		5 vol % O ₂ in Ar, GHVS=68.000 h ⁻¹			
Ni-Mn-O-10	low temperature	400 ppm NO, 400 ppm NH_3 ,	80% (90-270°C)	[3]	
	crystal splitting	2 vol % O_2 in N_2 , GHVS=30.000 h	1 ⁻¹		
Ni-Mn(3DH-NM/NF) hydrothermal		500 ppm NO, 500 ppm NH ₃ ,	80% (250-360°C)	[4]	
		3 vol % O_2 in N_2 , GHVS=20.000 h ⁻¹			
NiMn-LDO	co-precipitation	500 ppm NO, 500 ppm NH ₃ ,	80% (160-390°C)	This work	
		5 vol % O_2 in He, GHSV=45.000 h ⁻¹			



Fig. S1 SEM image and EDS mapping of Ni₃Mn-LDO catalyst.

(b)			0		1	Vi			Mn
	Ni:Mn = 4.40	Elt.	Line	Intensity (c/s)	Conc	Units	Error 2-sig	MDL 3-sig	
9		0	Ka	2,798.23	25.877	wt.%	0.549	0.684	
Ni		Mn	Ka	868.80	13.010	wt.%	0.474	0.568	
Min	Ni	Ni	Ka	1,686.43	61.113	wt.%	1.562	1.914	
Mm M	In Ni _{Ni} n				100.000	wt.%			Total
5 Cursor= Vert=4128	10 Window 0.005 - 40.955= 165,1	15 84 ent		20	25	30	1 1 1 1	35	40 keV

Fig. S2 SEM image and EDS mapping of $\rm Ni_4Mn\text{-}LDO$ catalyst.

(C) 5.08003 CeV 37m x50 Cit 5(L)		0		1	Ni			Mn
Ni:Mn = 4.68	Elt.	Line	Intensity (c/s)	Conc	Units	Error 2-sig	MDL 3-sig	
Ni	0	Ka	685.18	25.436	wt.%	0.929	1.160	
Q	Mn	Ka	202.97	12.422	wt.%	0.763	0.911	
Ni Ni	Ni	Ka	443.80	62.141	wt.%	2.718	3.333	
Min Ni Ni Min Min Ni				100.000	wt.%			Total
5 10 Cursor= Vert=1835 Window 0.005 - 40.955= 80,87	15 9 ent		20	25	30		35	40 keV

Fig. S3 SEM image and EDS mapping of Ni $_5$ Mn-LDO catalyst.

(d) 3.6010 3.040 & 4mm sid 0x 5EU 1 1 Jum		0		Γ	Vi			Mn
	Elt.	Line	Intensity	Conc	Units	Error	MDL	
Ni:Mn = 5.91			(c/s)			2-sig	3-sig	
	0	Ka	1,167.61	19.679	wt.%	0.703	0.876	
	Mn	Ka	345.94	10.981	wt.%	0.552	0.662	
Ni Ni	Ni	Ka	959.86	69.340	wt.%	2.259	2.768	
Ma Ma Ni				100.000	wt.%			Total
S 10 Cursor= Vert=2198 Window 0.005 - 40.955= 90,5	15 15 ent		20	25	30	<u> </u>	35	40 keV

Fig. S4 SEM image and EDS mapping of Ni₆Mn-LDO catalyst.

Notes and references

- [1] Y. Wan, W. R. Zhao, Y. Tang, L. Li, H. J. Wang, Y. L. Cui, J. L. Gu, Y.S. Li and J. L. Shi, Appl. Catal. B, 148, (2014) 114-122.
- [2] Y. L. Han, J. C. Mu, X. Y. Li, J. S. Gao, S. Y. Fan, F. Tan, Q. D. Zhao, Chem. Commun. 54 (2018) 9797-9800.
- [3] B. Meng, Z. B. Zhao, Y. S. Chen, X. Z. Wang, Y. Li and J. S. Qiu, Chem. Commun. 50 (2014) 12396-12399.
- [4] S. X. Cai, D. S. Zhang, L. Y. Shi, J. Xu, L. Zhang, L. Huang, H. R. Li and J. P. Zhang, Nanoscale, 6 (2014) 7346.