

Supporting Information

Design of highly porous Fe₃O₄@reduced graphene oxide via a facile PMAA-induced assembly

Huan Wang,[†] Madumali Kalubowilage,[‡] Stefan H. Bossmann,[‡] and Placidus B. Amama^{*†}

[†]*Tim Taylor Department of Chemical Engineering, Kansas State University, Manhattan, KS 66506, USA*

[‡]*Department of Chemistry, Kansas State University, Manhattan, KS 66506, USA*

Table S1. Comparison of surface area of Fe₃O₄@RGO composites.

Ratio of Fe ₃ O ₄ (wt %)	Surface Area (m ² /g)	Ref.
84	114.5	12
70	221	28
75	262	37
60	52.84	38
46	110	39
60	338.8	This work

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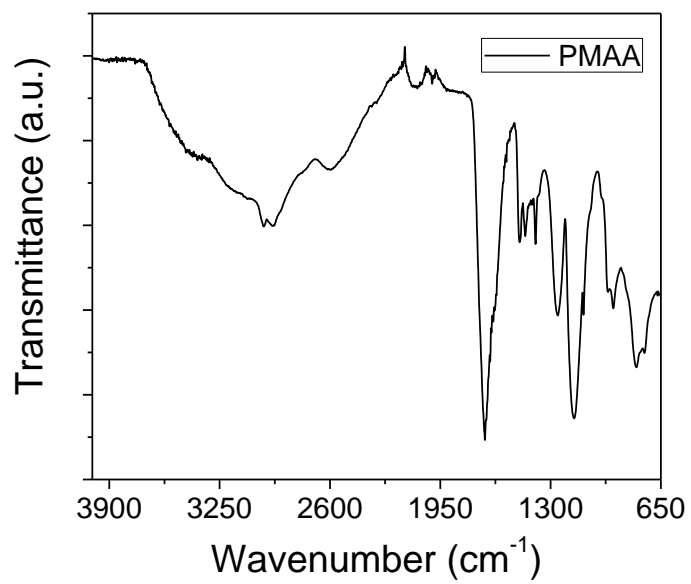


Fig. S1 FTIR spectrum of as-synthesized PMAA.

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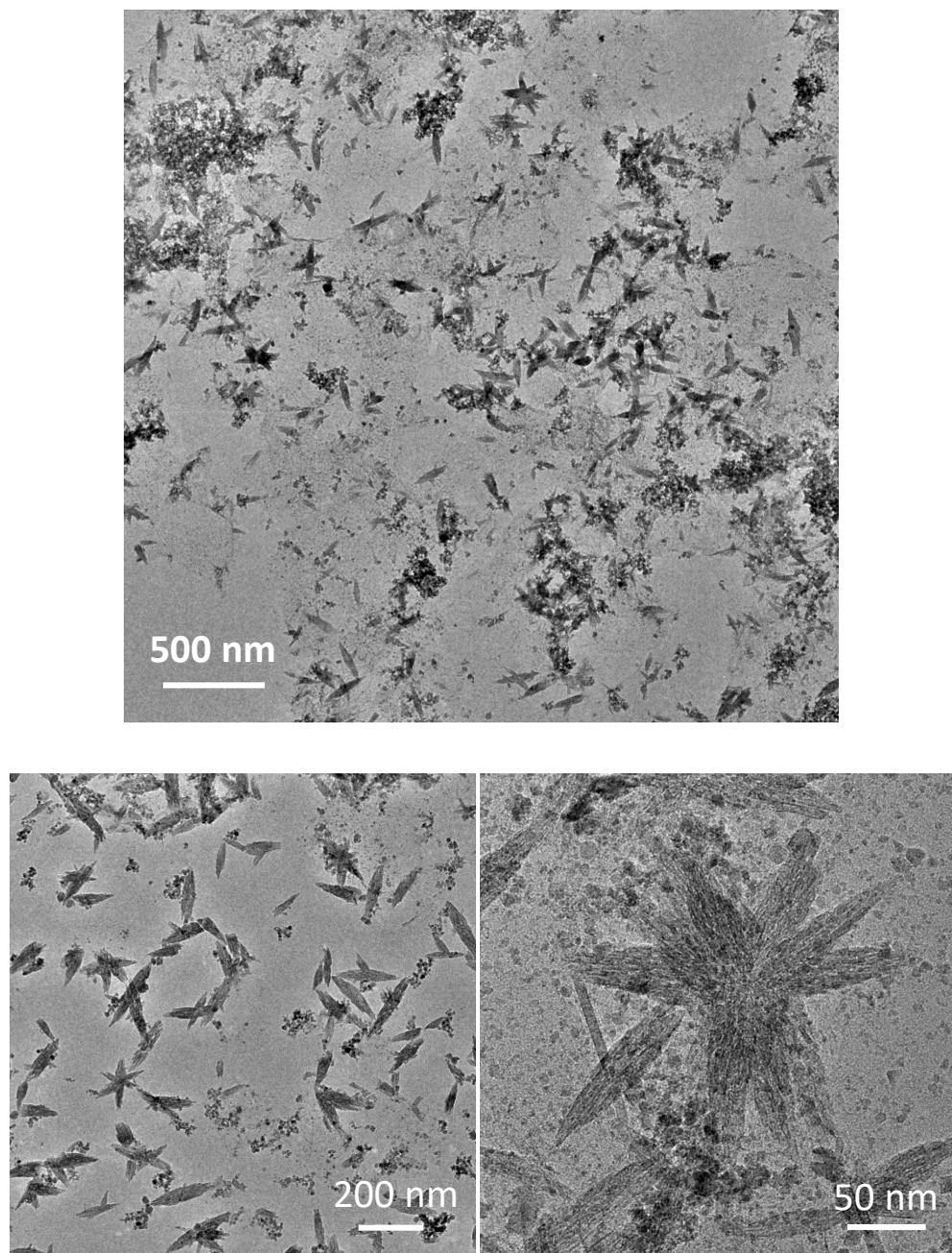


Fig. S2 TEM images of $\text{Fe}_3\text{O}_4@\text{RGO}$ sheets dispersed at pH 9 with 4.8 mg/ml PMAA in water.

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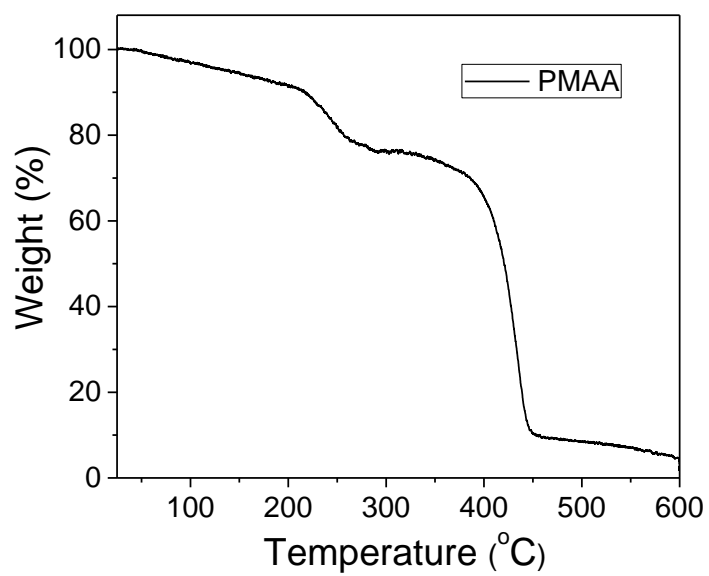


Fig. S3 TGA curve of PMAA under N₂ flow at a temperature ramp of 5 °C min⁻¹ in air.

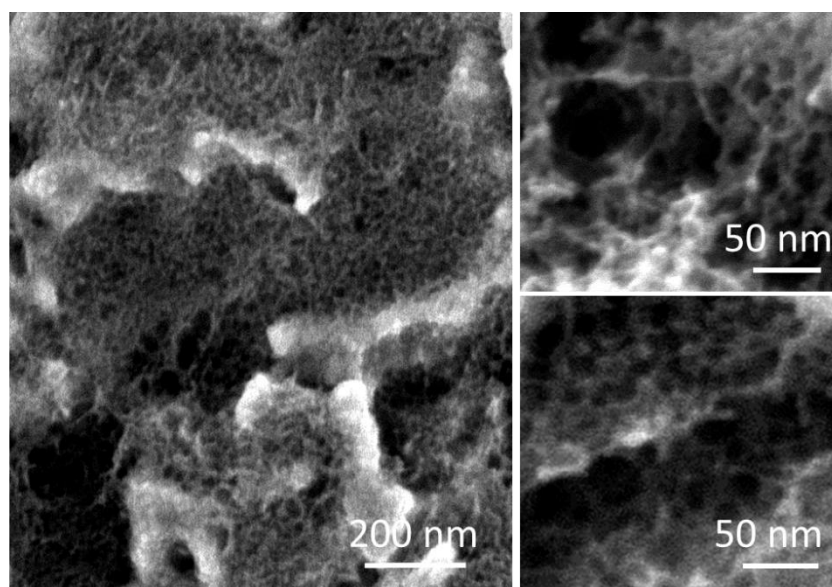


Fig. S4 SEM images of porous Fe₃O₄@RGO anodes.

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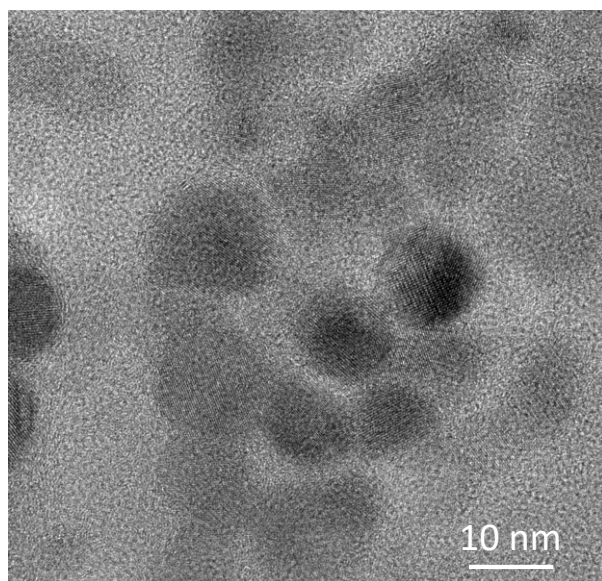


Fig. S5 TEM image of Fe₃O₄@RGO nanosheets

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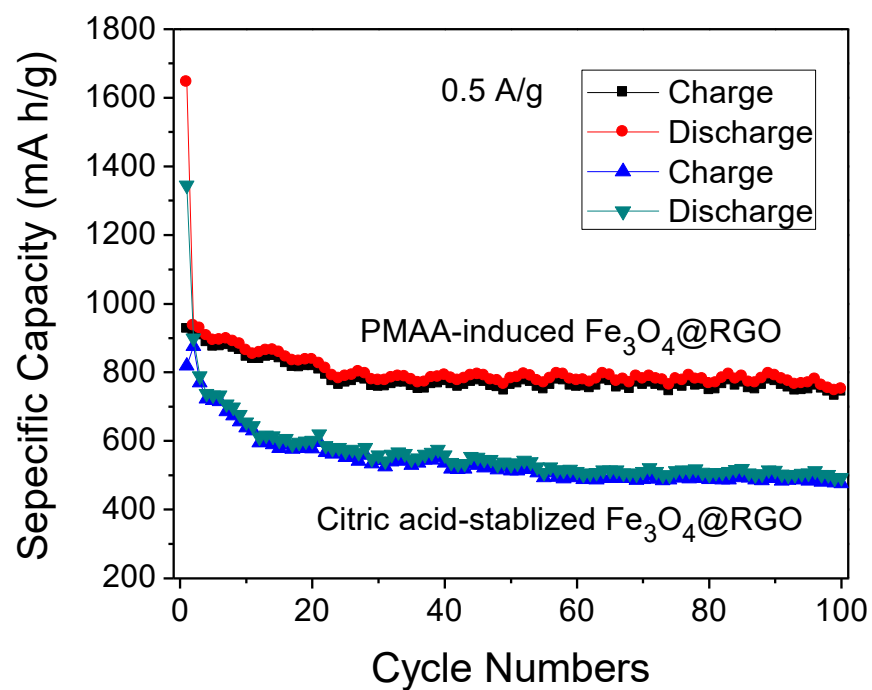


Fig. S6 Cycling performance of a PMAA-induced $\text{Fe}_3\text{O}_4/\text{RGO}$ anode and a citric acid-stabilized $\text{Fe}_3\text{O}_4@\text{RGO}$ anode at a current density of 0.5 A/g in the potential window of $3.0\text{--}0.002 \text{ V}$ (vs. Li/Li^+).