

## Electronic Supplementary Information (ESI)

for

### **Tunable pseudocapacitive contribution by dimension control in nanocrystalline-constructed $(\text{Mg}_{0.2}\text{Co}_{0.2}\text{Ni}_{0.2}\text{Cu}_{0.2}\text{Zn}_{0.2})\text{O}$ solid solutions to achieve superior lithium-storage properties**

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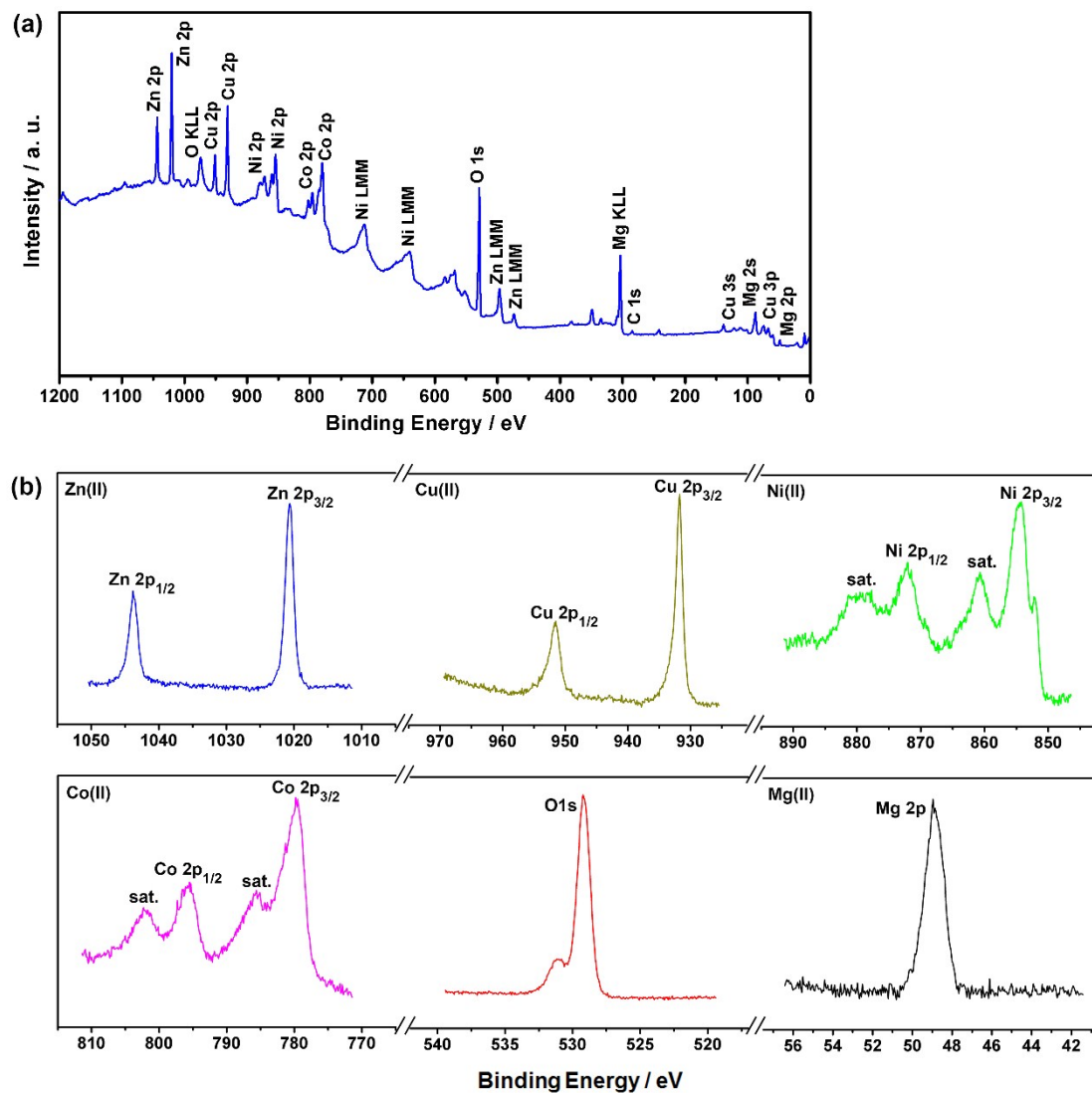
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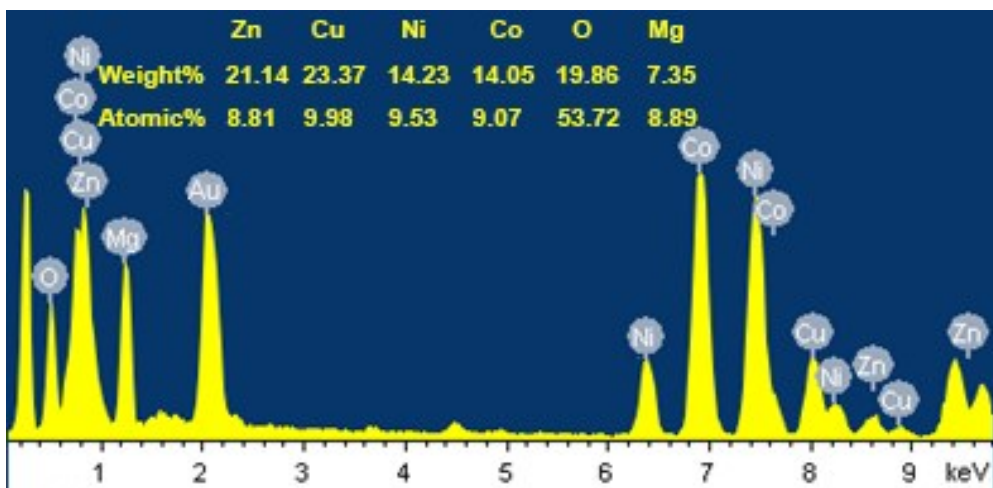
impedance spectroscopy (EIS) measurements.

**Table S1.** The kinetic parameters of different HEO samples at cut-off potential of 0.01, 1.0 1.5, 2.0, 2.5 and 3.0 V.

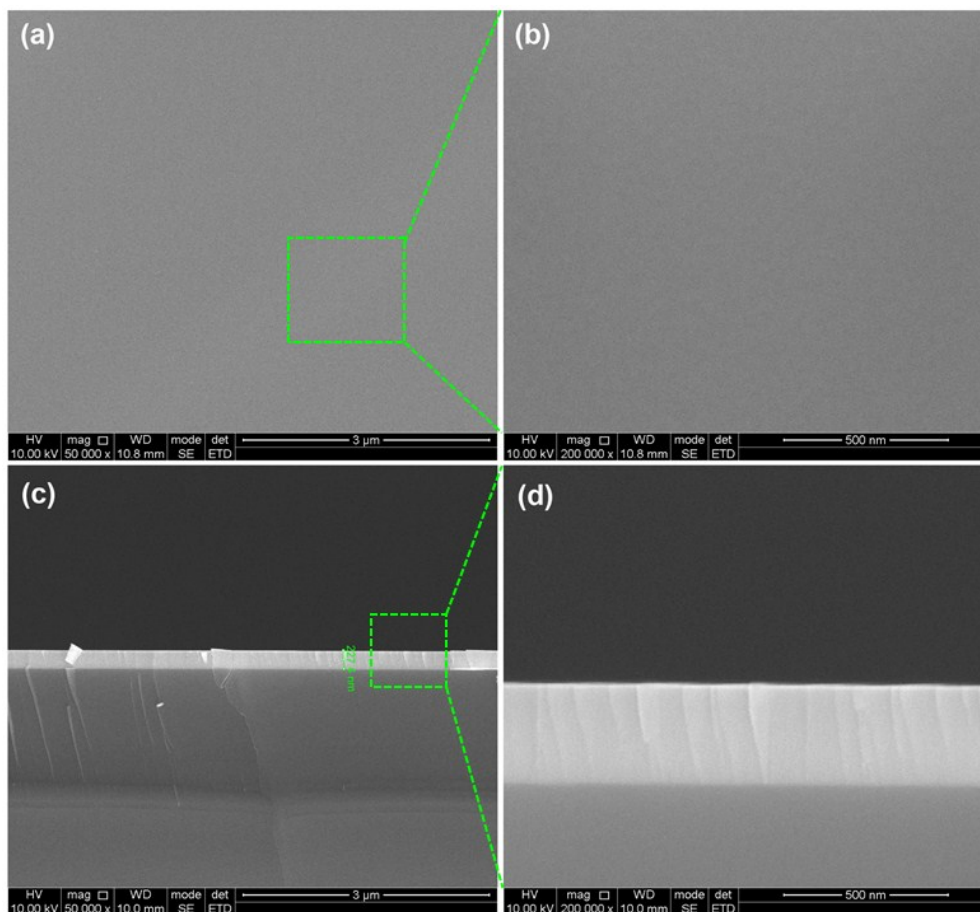
**Figure S15.** The lithium-ion diffusion coefficients (D) of different HEO samples at cut-off potential of 0.01, 1.0 1.5, 2.0, 2.5 and 3.0 V.



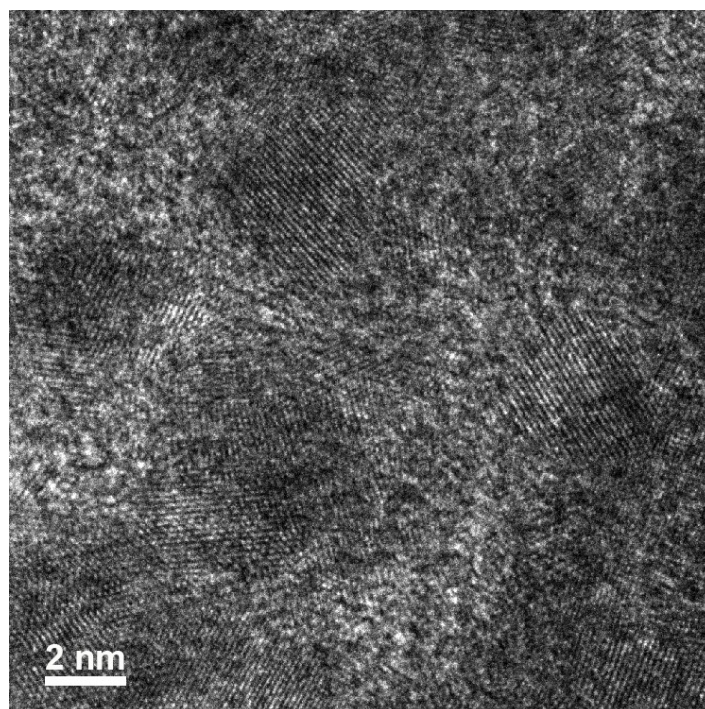
**Figure S1.** XPS spectrum of the as-prepared HEO-Film, (a) Survey scan; and (b) High-resolution XPS spectra of Zn 2p, Cu 2p, Ni 2p, Co 2p, O 1s, Mg 2p.



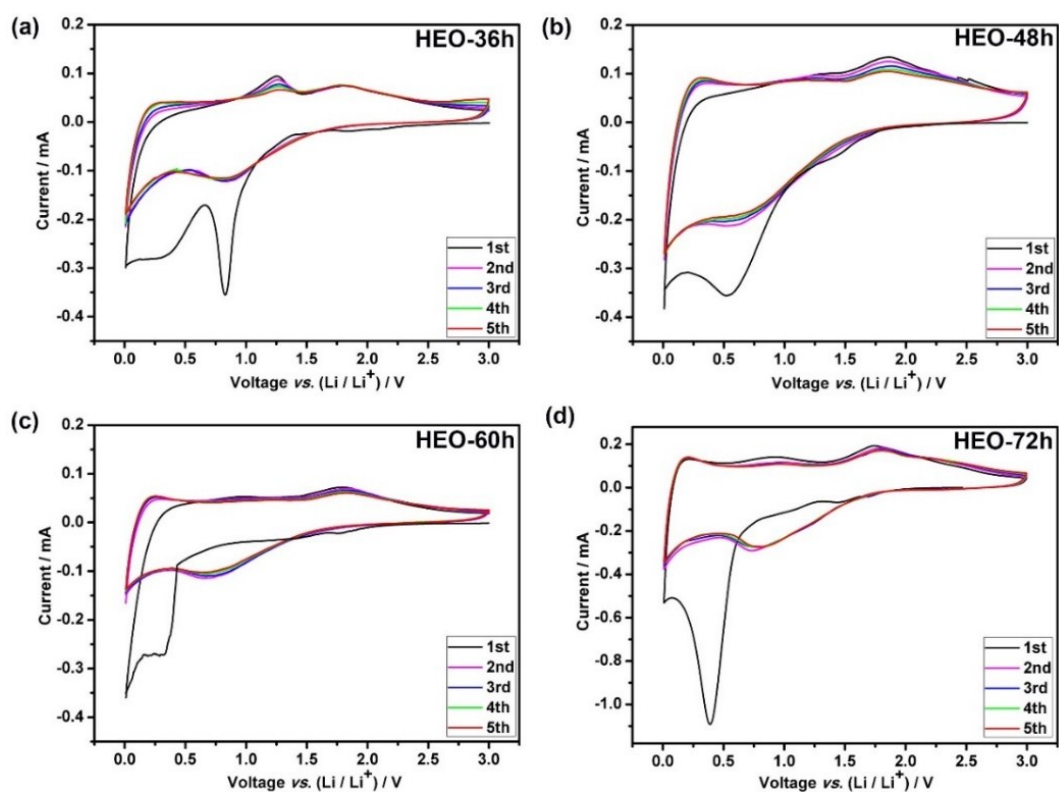
**Figure S2.** STEM-EDS analysis of the as-prepared HEO-Film on a gold grid. The corresponding element atomic ratio indicate the homogeneous spatial distributions of each element.



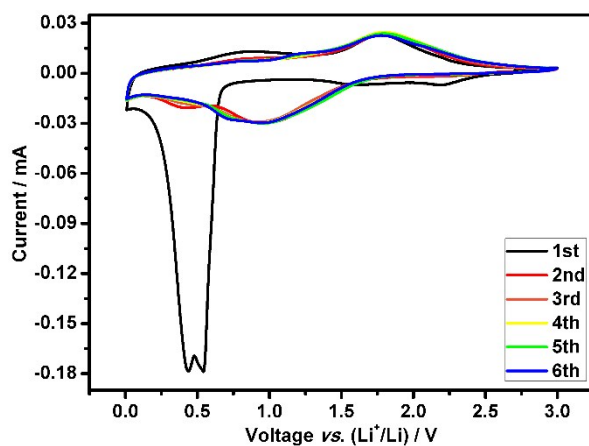
**Figure S3.** (a-b) Surface and (c-d) cross-section SEM images of the as-deposited HEO thin film.



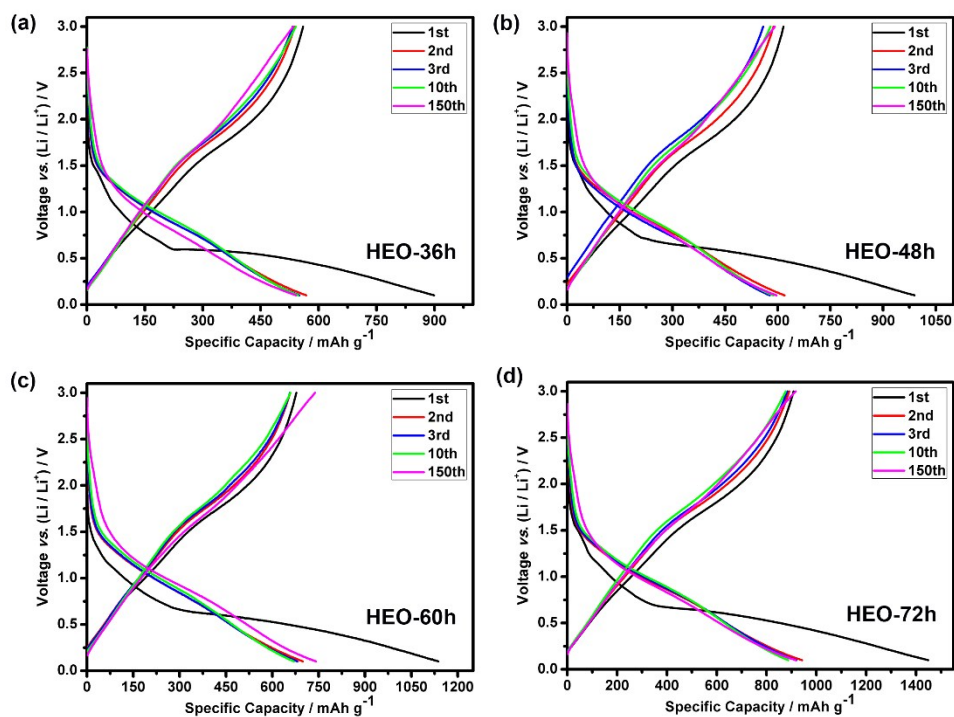
**Figure S4.** The enlarged HRTEM pattern of **Figure 4(c)**.



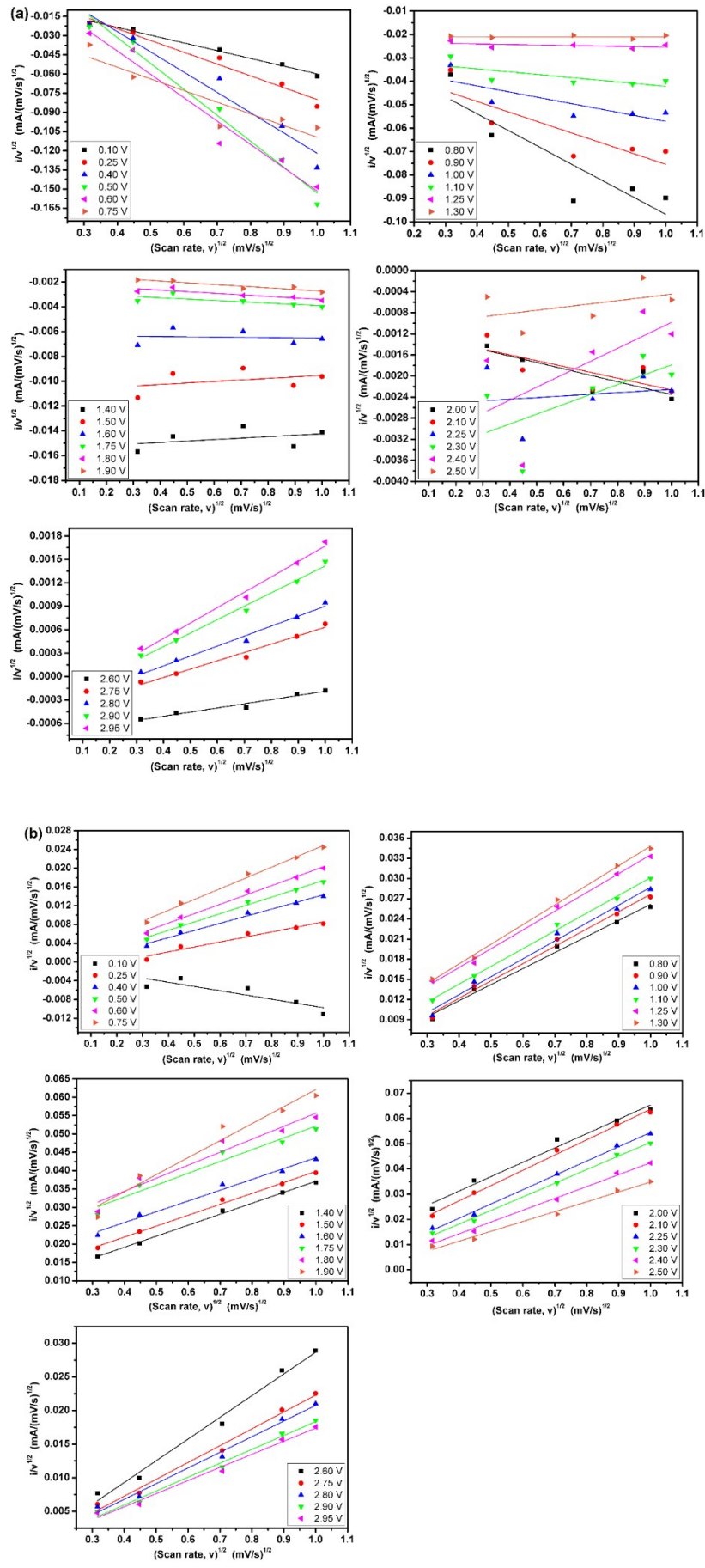
**Figure S5.** CV curves of (a) HEO-36h, (b) HEO-48h, (c) HEO-60h and (d) HEO-72h electrodes at a scan rate of  $0.2 \text{ mV s}^{-1}$ .



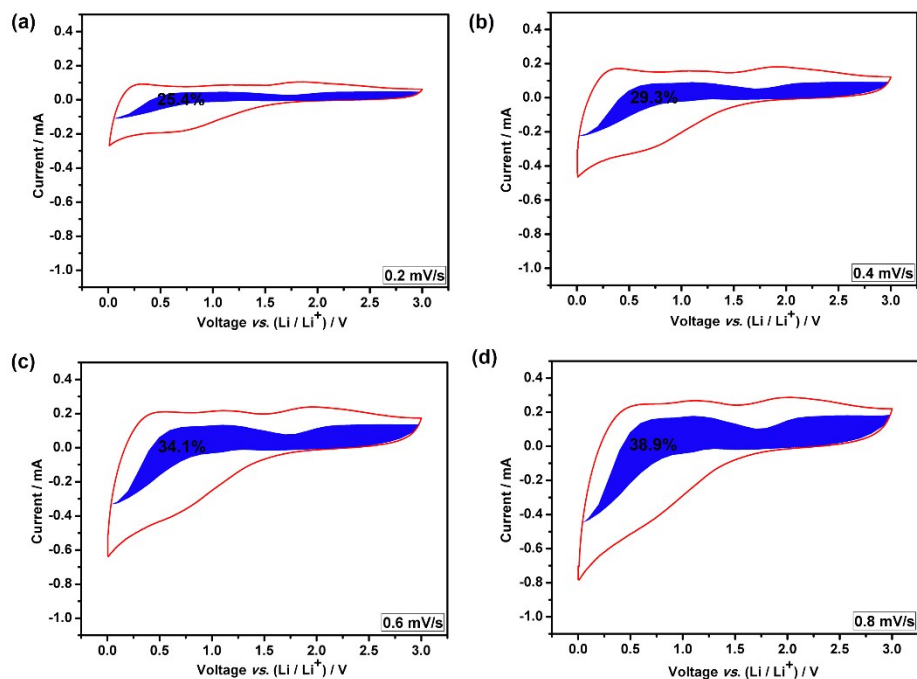
**Figure S6.** CV curve of HEO-Film electrode at a scan rate of  $0.2 \text{ mV s}^{-1}$ .



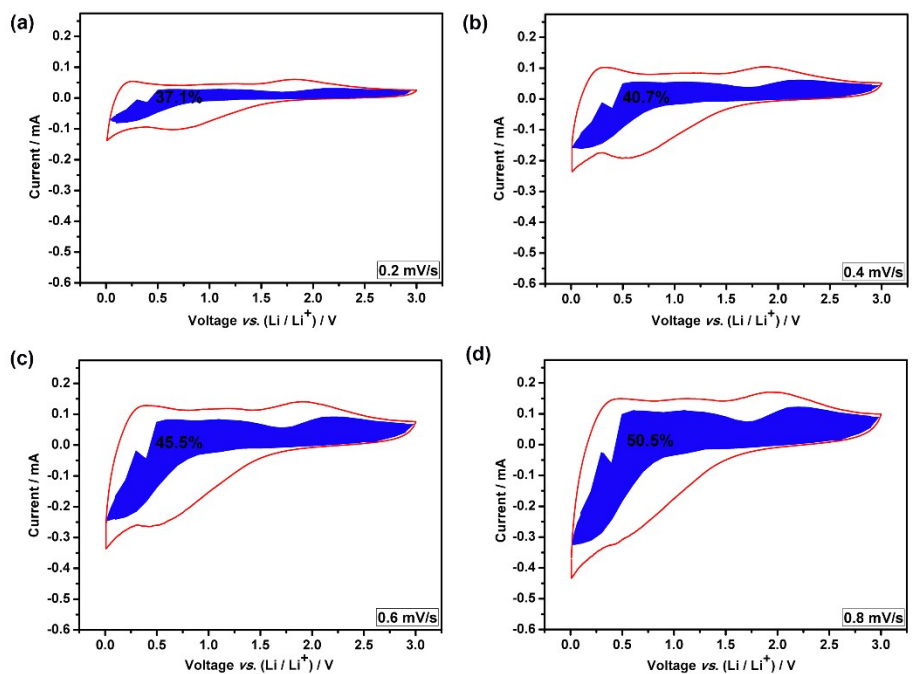
**Figure S7.** Discharge-charge voltage profiles of HEO electrodes for the 1st, 2nd, 3rd, 10th, and 150th cycles in the voltage range of 0.01-3.00 V at a current rate of  $200 \text{ mA g}^{-1}$ .



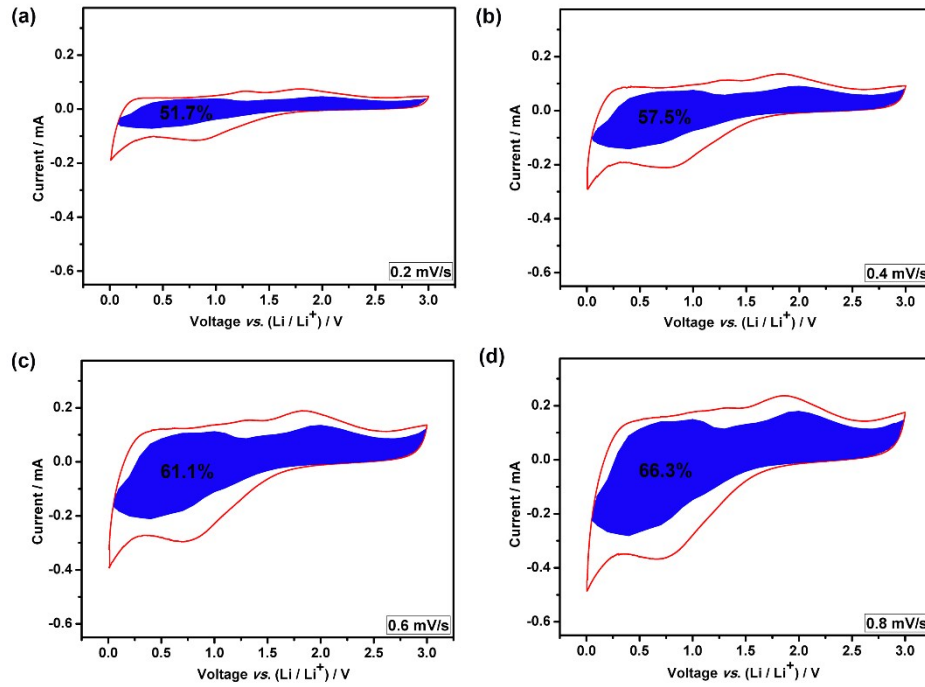
**Figure S8.** Correlation between the scan rates and corresponding currents of the HEO-Film electrode at (a) anodic and (b) cathodic scan according to equation:  $i(V)/v^{1/2} = k_1v^{1/2} + k_2$ .



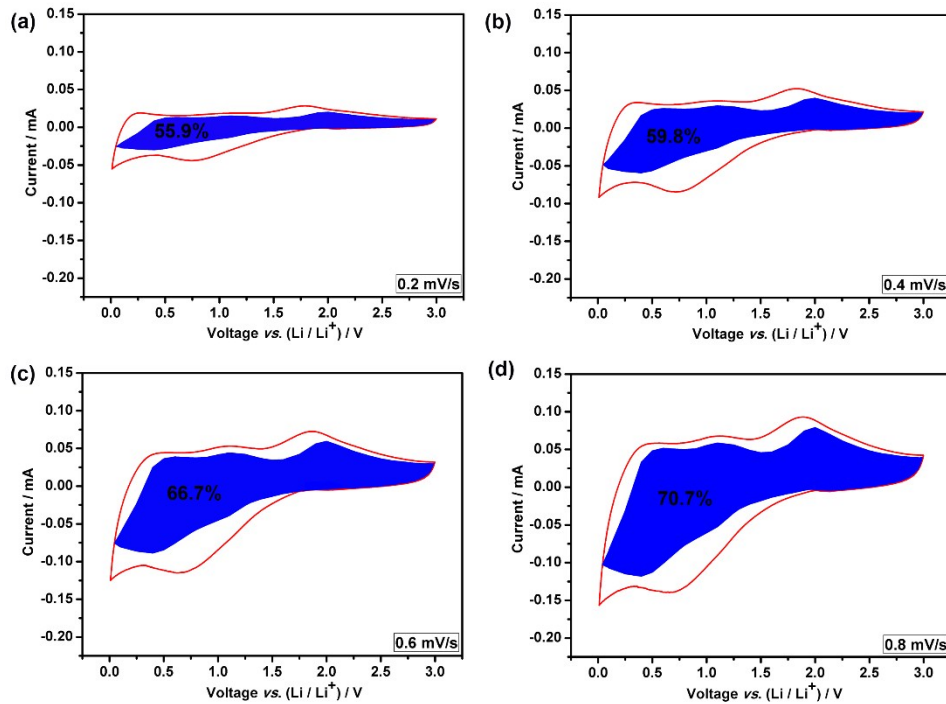
**Figure S9.** The capacitive contribution (blue shaded region) to the total current contribution (red line) for the HEO-36h electrode at (a) 0.2, (b) 0.4, (c) 0.6, (d) 0.8  $\text{mV s}^{-1}$  vs.  $\text{Li}^+/\text{Li}$ .



**Figure S10.** The capacitive contribution (blue shaded region) to the total current contribution (red line) for the HEO-48h electrode at (a) 0.2, (b) 0.4, (c) 0.6, (d) 0.8  $\text{mV s}^{-1}$  vs.  $\text{Li}^+/\text{Li}$ .

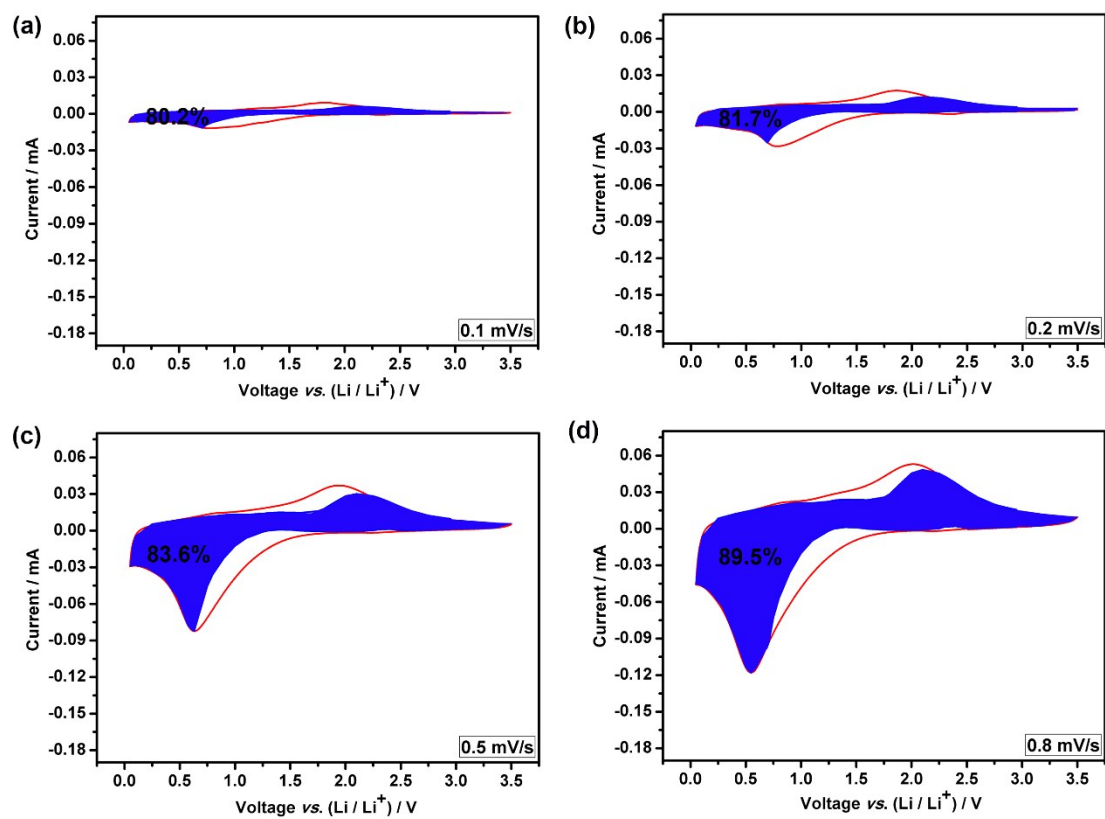


**Figure S11.** The capacitive contribution (blue shaded region) to the total current contribution (red line) for the HEO-60h electrode at (a) 0.2, (b) 0.4, (c) 0.6, (d) 0.8 mV s<sup>-1</sup> vs. Li<sup>+</sup>/Li.

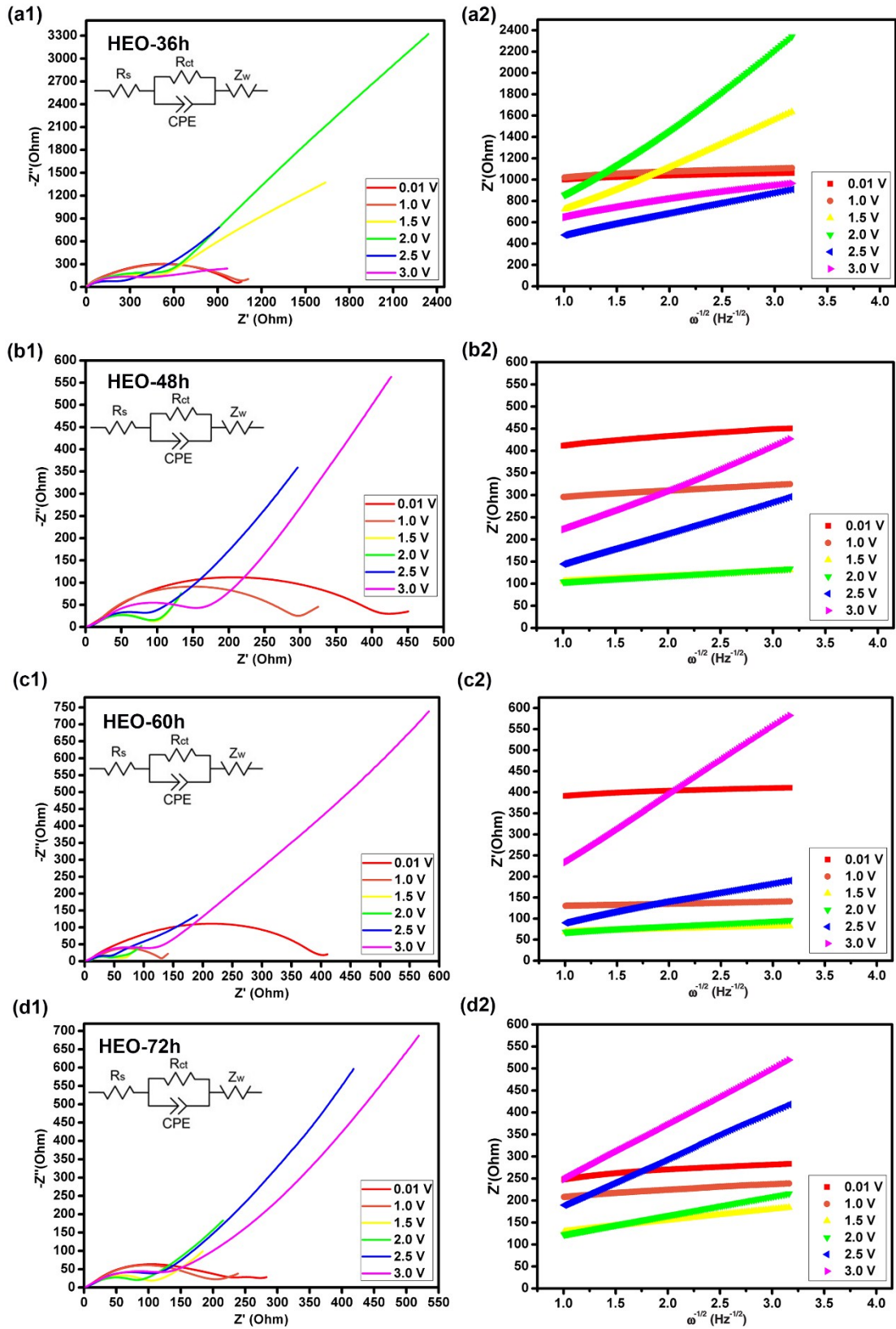


**Figure S12.** The capacitive contribution (blue shaded region) to the total current contribution (red line) for the HEO-72h electrode at (a) 0.2, (b) 0.4, (c) 0.6, (d) 0.8 mV s<sup>-1</sup> vs. Li<sup>+</sup>/Li.





**Figure S13.** The capacitive contribution (blue shaded region) to the total current contribution (red line) for the HEO-Film electrode at (a) 0.1, (b) 0.2, (c) 0.5, (d) 0.8 mV s<sup>-1</sup> vs. Li<sup>+</sup>/Li.

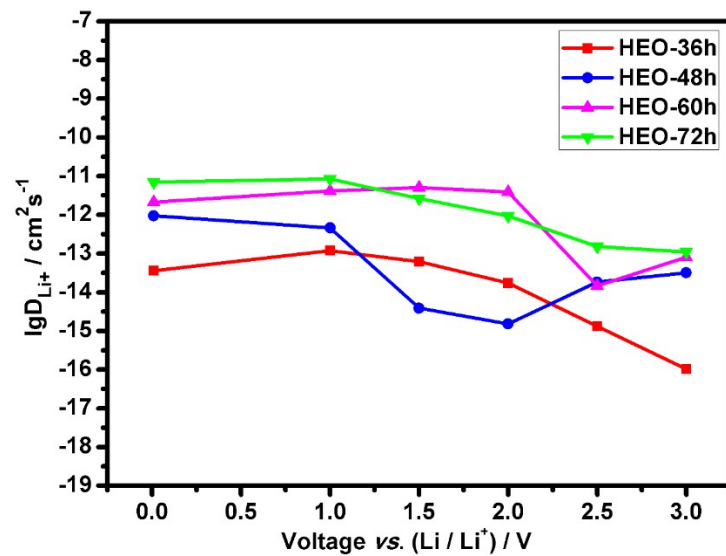


**Figure S14.** Determination of lithium-ion diffusion coefficients ( $D$ ) by electrochemical impedance spectroscopy (EIS) measurements according to  $D = R^2 T^2 / 2 A^2 n^4 F^4 C^2 \sigma^2$ , where  $R$ ,  $T$ ,  $A$ ,  $n$ ,  $F$ , and  $C$  are the gas constant, absolute temperature, electrode area, electron transfer

number, Faraday constant, and lithium-ion molar concentration, respectively. The  $\sigma$  value (Warburg factor) can be calculated by  $Z' = R_s + R_{ct} + \sigma\omega^{-1/2}$  from the low-frequency plots of EIS, and  $\omega$  is the corresponding angular frequency.

Samples	$R_s$ [ $\Omega$ ]	$R_{ct}$ [ $\Omega$ ]	$\sigma$ [ $\Omega \text{ Hz}^{1/2}$ ]	$D_{Li}$ [ $\text{cm}^2 \text{ s}^{-1}$ ]
HEO-36h (0.01 V)	1.89	979.21	27.3	$9.38 \times 10^{-13}$
HEO-36h (1.0 V)	1.83	991.15	39.1	$4.60 \times 10^{-13}$
HEO-36h (1.5 V)	1.75	274.32	425.0	$3.93 \times 10^{-15}$
HEO-36h (2.0 V)	1.41	117.23	682.4	$1.52 \times 10^{-15}$
HEO-36h (2.5 V)	1.45	291.33	195.7	$1.81 \times 10^{-14}$
HEO-36h (3.0 V)	1.87	511.79	148.6	$3.21 \times 10^{-14}$
HEO-48h (0.01 V)	3.87	391.54	18.3	$2.13 \times 10^{-12}$
HEO-48h (1.0 V)	3.02	280.48	13.1	$4.11 \times 10^{-12}$
HEO-48h (1.5 V)	2.89	90.03	11.9	$5.09 \times 10^{-12}$
HEO-48h (2.0 V)	3.05	87.89	13.4	$3.88 \times 10^{-12}$
HEO-48h (2.5 V)	2.71	71.75	69.7	$1.47 \times 10^{-14}$
HEO-48h (3.0 V)	2.39	122.08	93.1	$8.14 \times 10^{-14}$
HEO-60h (0.01 V)	4.41	380.46	8.7	$3.59 \times 10^{-14}$
HEO-60h (1.0 V)	3.49	122.21	4.9	$1.21 \times 10^{-13}$
HEO-60h (1.5 V)	3.11	58.85	6.7	$6.19 \times 10^{-14}$
HEO-60h (2.0 V)	3.10	53.61	12.4	$1.75 \times 10^{-14}$
HEO-60h (2.5 V)	2.50	44.71	45.9	$1.33 \times 10^{-15}$
HEO-60h (3.0 V)	2.30	64.47	163.8	$1.05 \times 10^{-16}$
HEO-72h (0.01 V)	2.84	234.00	15.9	$7.05 \times 10^{-12}$
HEO-72h (1.0 V)	2.40	192.65	14.2	$8.39 \times 10^{-12}$
HEO-72h (1.5 V)	2.20	101.20	25.9	$2.60 \times 10^{-12}$
HEO-72h (2.0 V)	2.29	77.21	43.3	$9.35 \times 10^{-13}$
HEO-72h (2.5 V)	2.25	78.90	106.9	$1.51 \times 10^{-13}$
HEO-72h (3.0 V)	2.30	119.95	124.9	$1.15 \times 10^{-13}$

**Table S1.** The kinetic parameters of different HEO samples at cut-off potential of 0.01, 1.0 1.5, 2.0, 2.5 and 3.0 V.



**Figure S15.** The lithium-ion diffusion coefficients ( $D$ ) of different HEO samples at cut-off potential of 0.01, 1.0 1.5, 2.0, 2.5 and 3.0 V.