

## Supporting Information

### Impact of Aryl bulky group on One-pot Reaction of Aldehyde with Malononitrile and N-substituted 2-cyanoacetamide

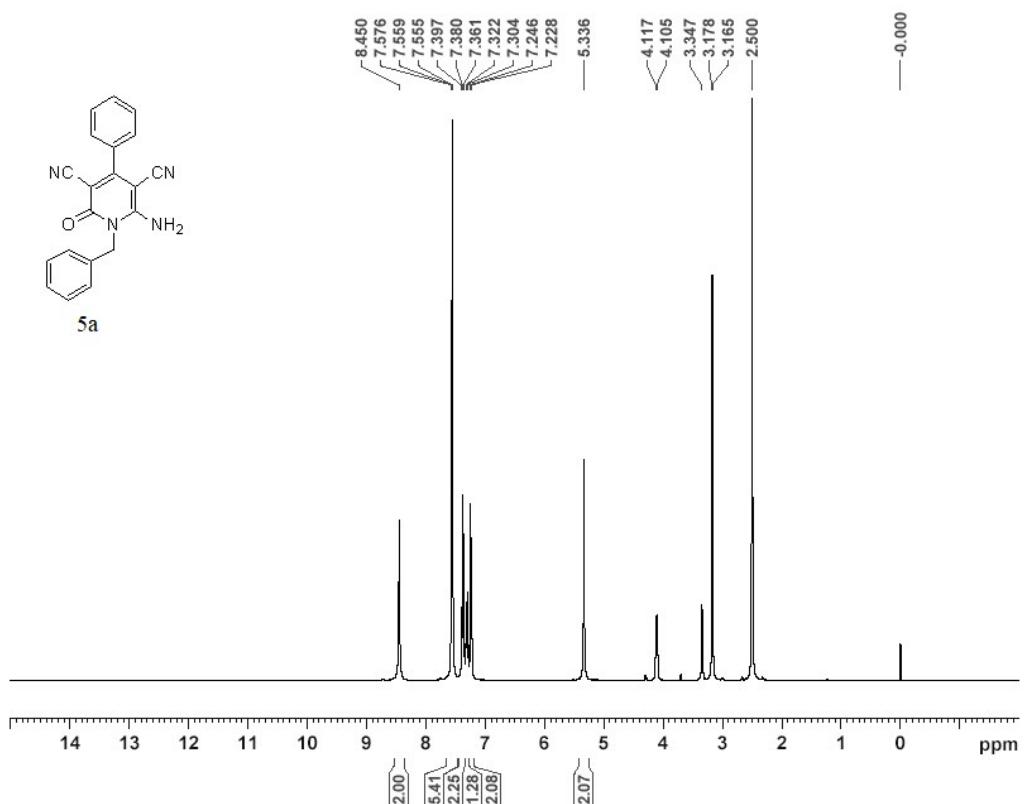
Ruturajsinh M. Vala<sup>a</sup>, Divyang M. Patel<sup>a</sup>, Mayank G. Sharma<sup>a</sup>, Hitendra M. Patel<sup>\*a</sup>

<sup>1</sup>Department of Chemistry, Sardar Patel University, Vallabh Vidyanagar-388120, Gujarat, India

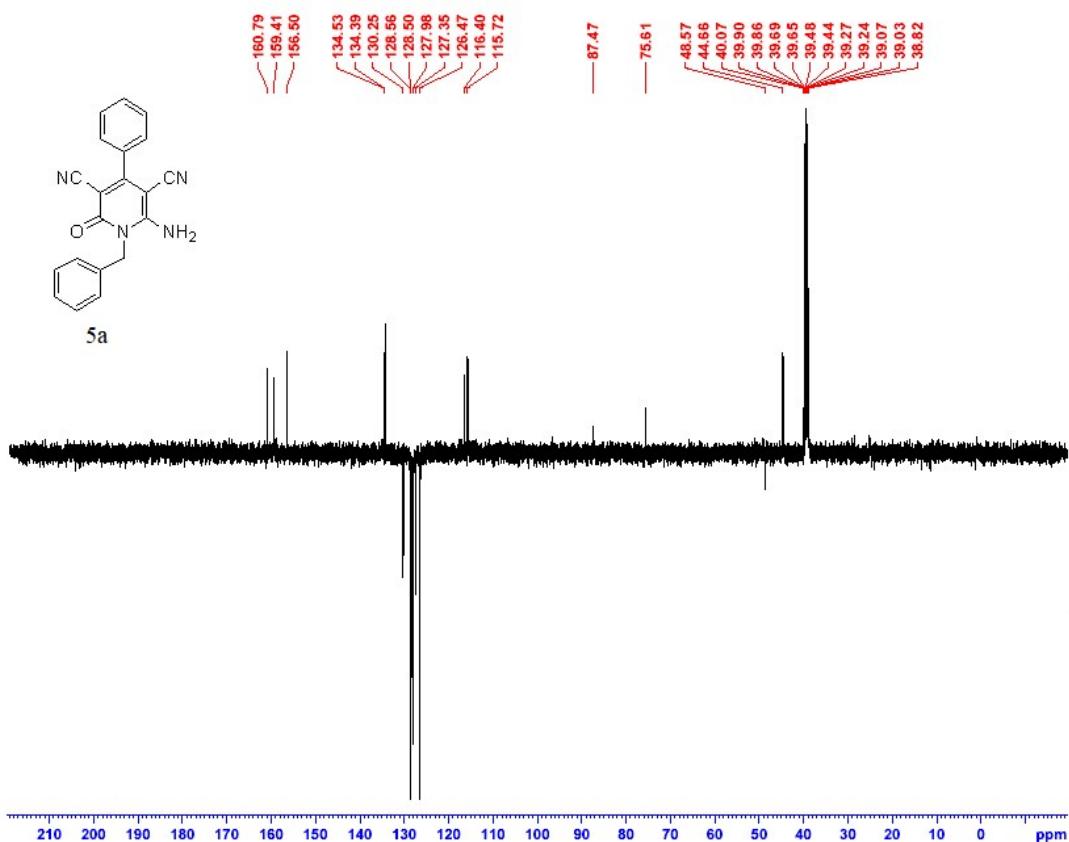
Email: hm\_patel@spuvn.edu

Copies of <sup>1</sup> H NMR, <sup>13</sup> C{ <sup>1</sup> H} APT spectra	S2-S20
Copies of Mass spectra of 5a-5l	S21-S26
Copies of LCMS spectra of 6a-6d	S27-S30
Crystallographic data	S31-S34

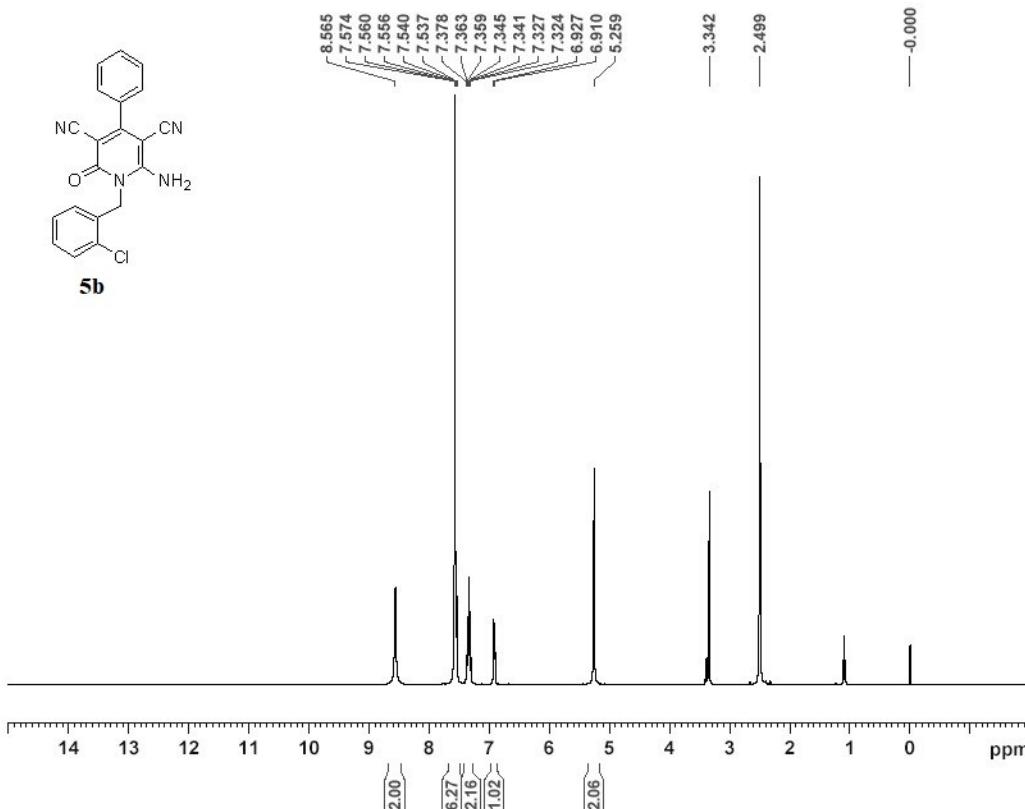
<sup>1</sup>H-NMR (400 MHz, DMSO-d<sub>6</sub>, CD<sub>3</sub>OD)



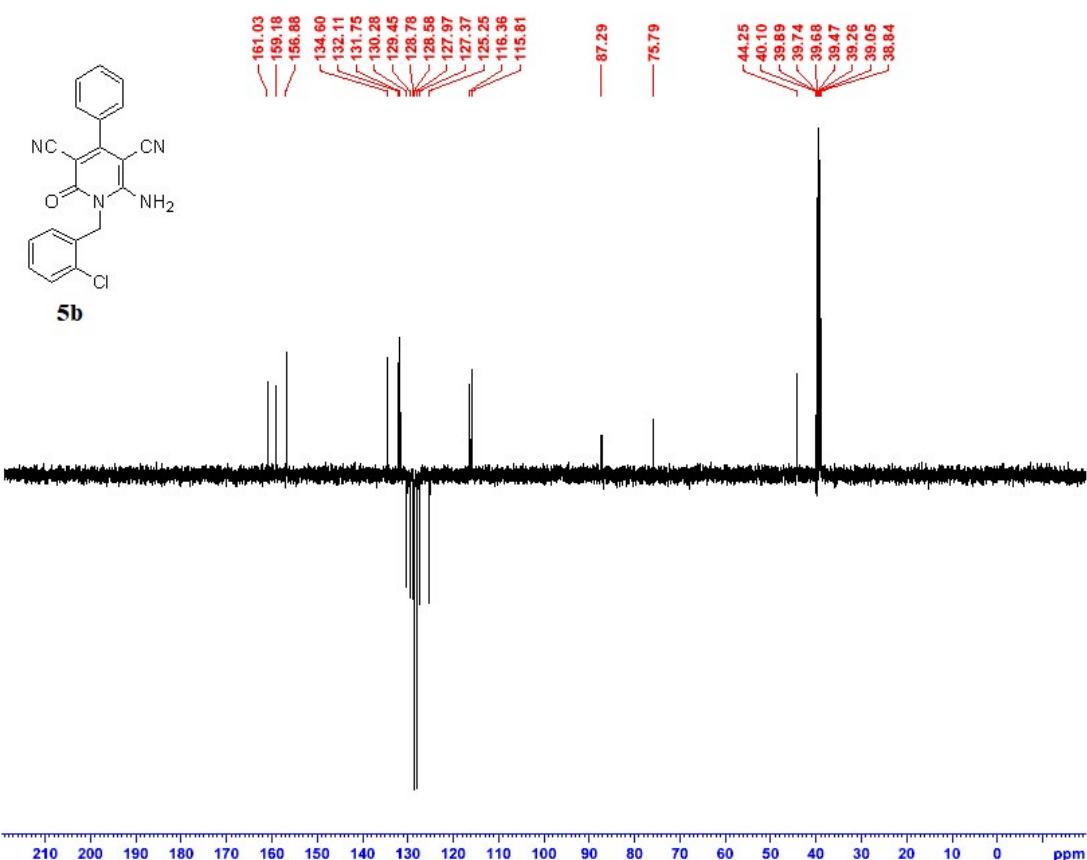
<sup>13</sup>C{<sup>1</sup>H}-APT (100 MHz, DMSO-d<sub>6</sub>, CD<sub>3</sub>OD)



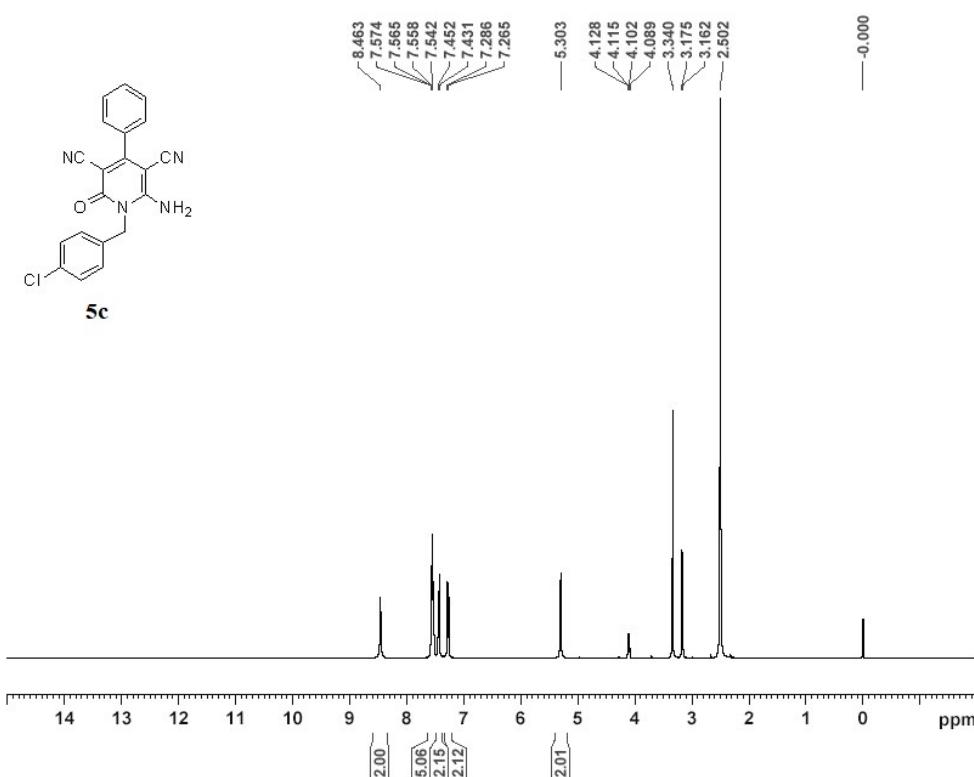
<sup>1</sup>H-NMR (400 MHz, DMSO-d<sub>6</sub>)



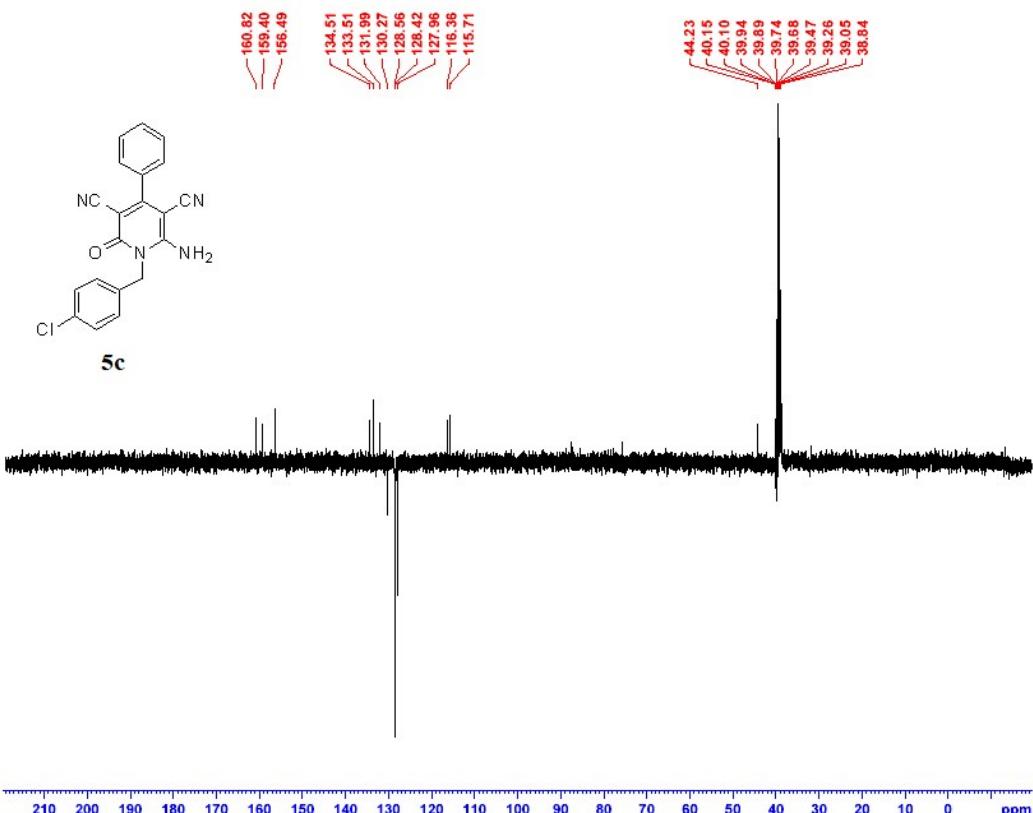
### <sup>13</sup>C{<sup>1</sup>H}-APT (100 MHz, DMSO-d6)



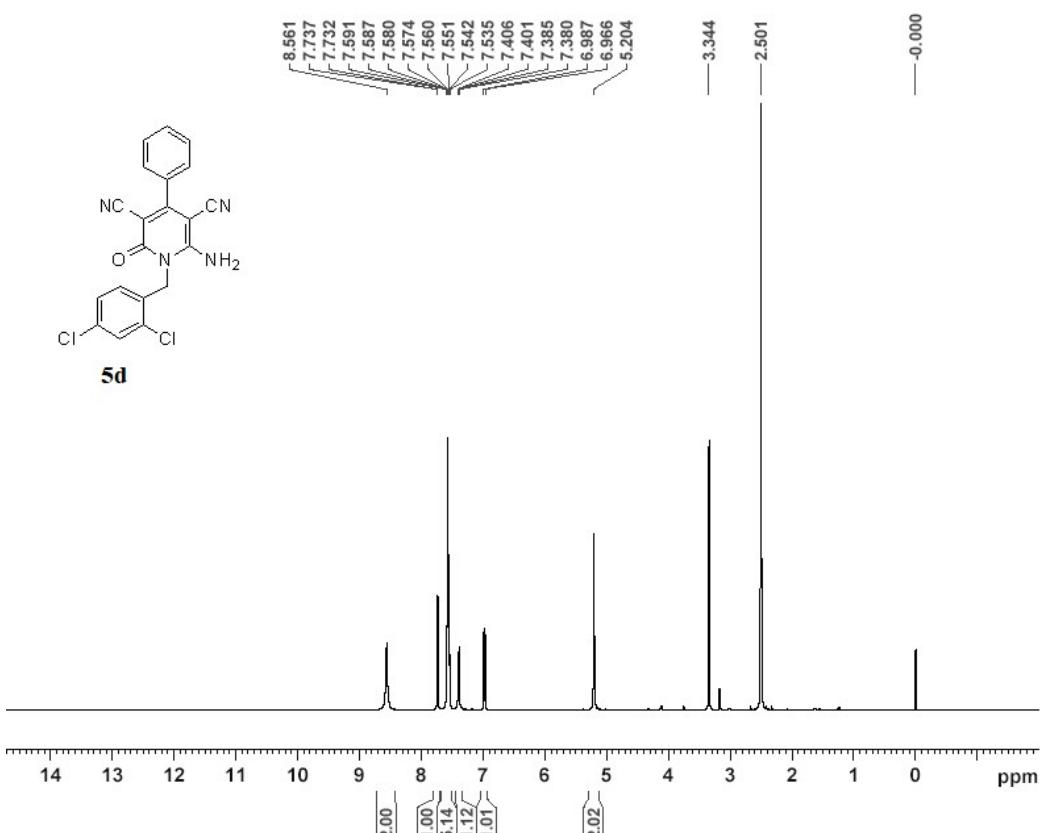
<sup>1</sup>H-NMR (400 MHz, DMSO-d<sub>6</sub>, CD<sub>3</sub>OD)



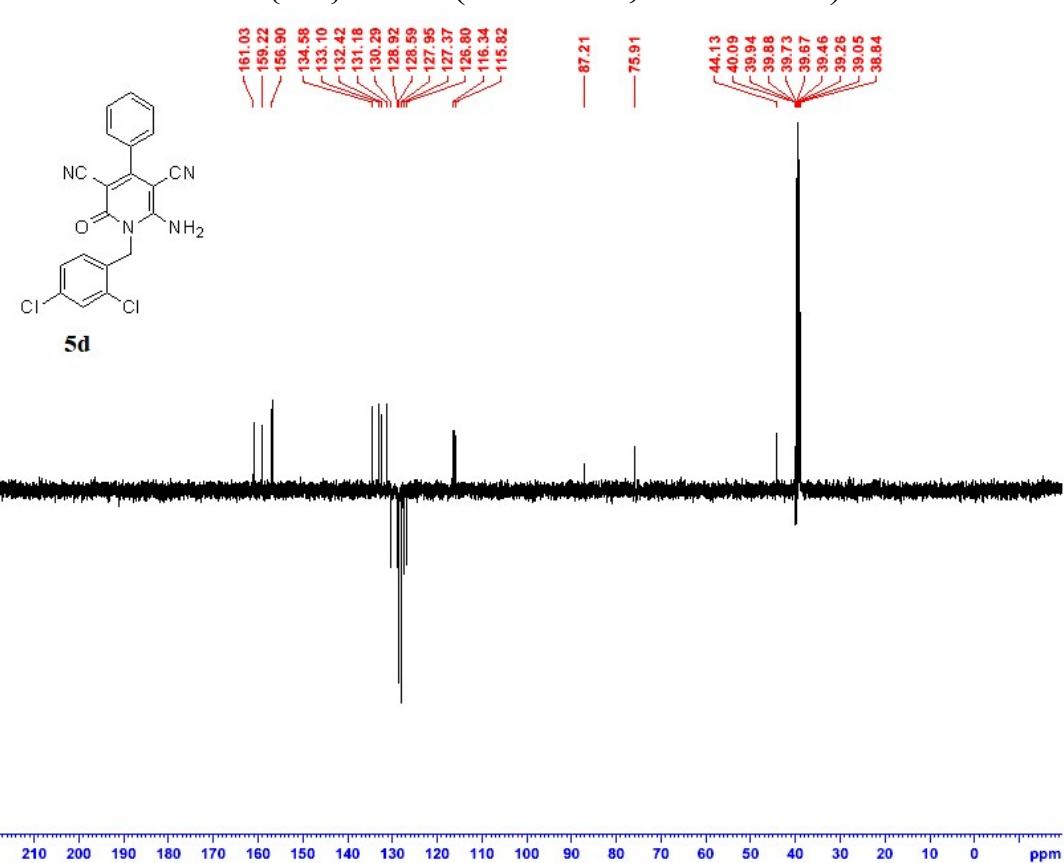
<sup>13</sup>C{<sup>1</sup>H}-APT (100 MHz, DMSO-d<sub>6</sub>, CD<sub>3</sub>OD)



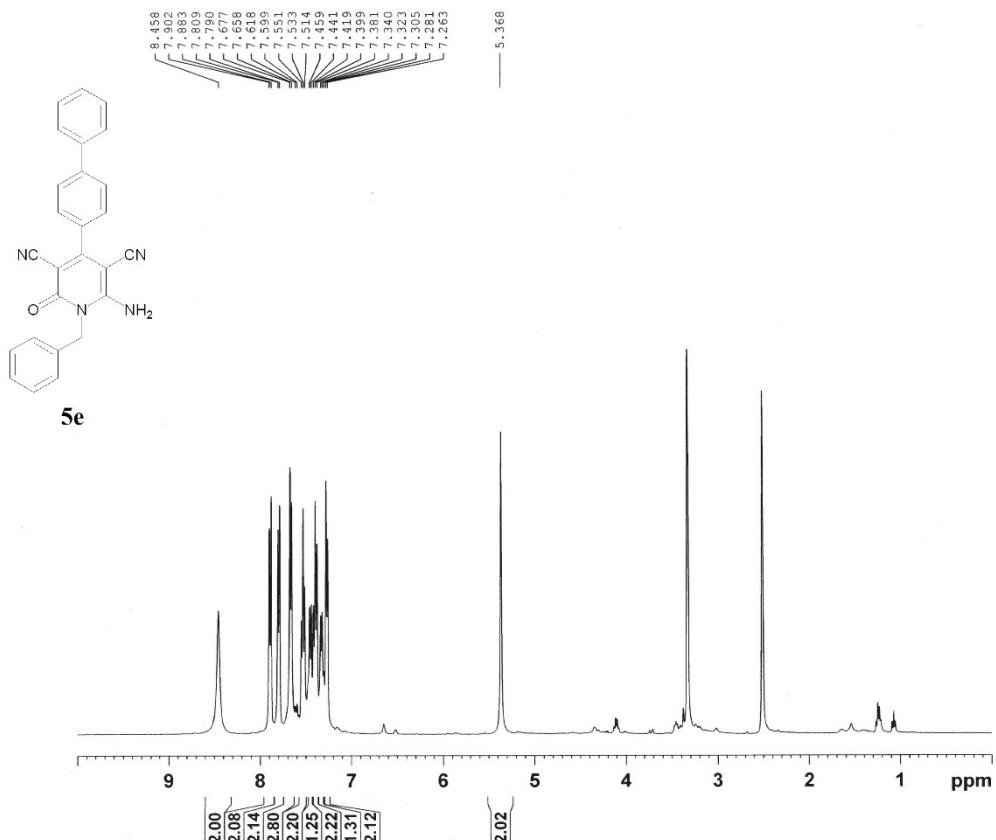
<sup>1</sup>H-NMR (400 MHz, DMSO-d6)



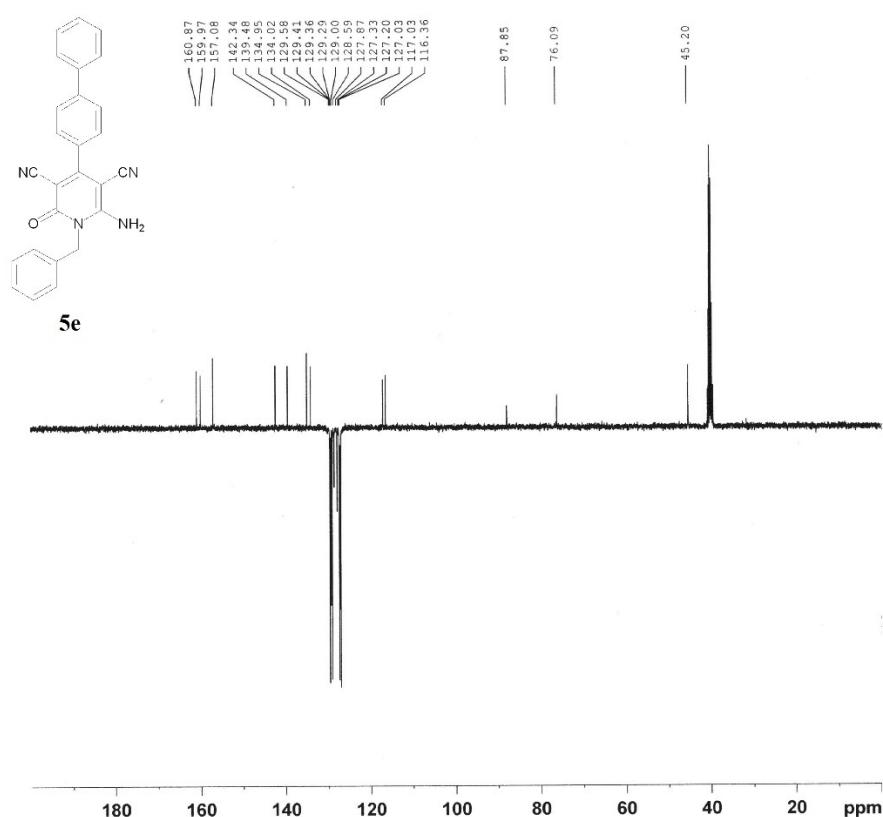
<sup>13</sup>C{<sup>1</sup>H}-APT (100 MHz, DMSO-d6)



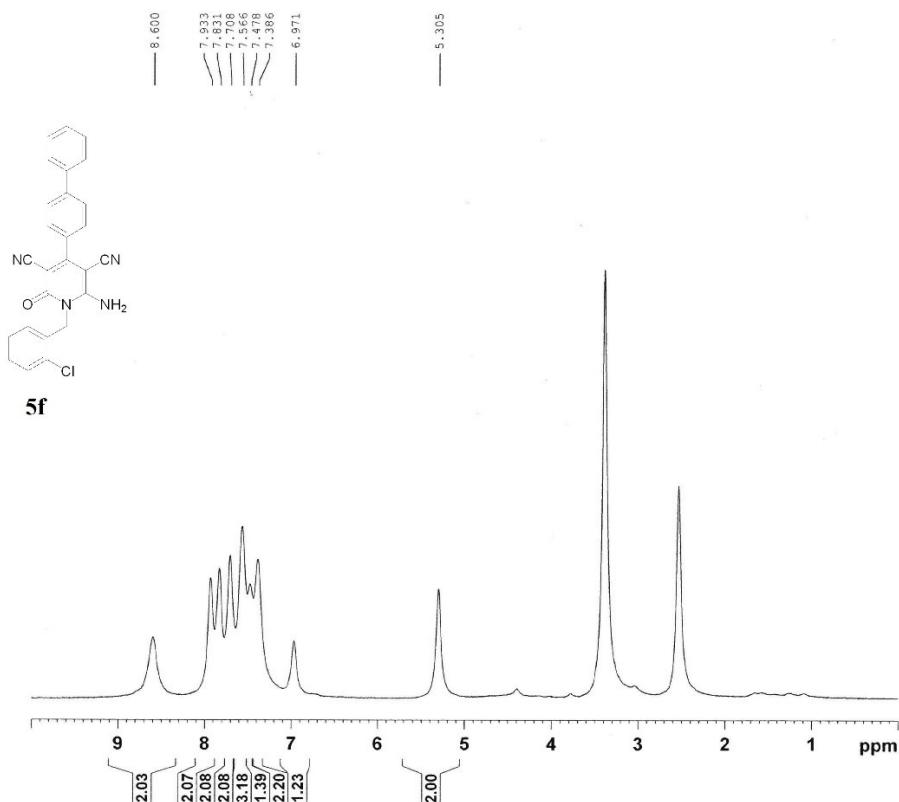
<sup>1</sup>H-NMR (400 MHz, DMSO-d6)



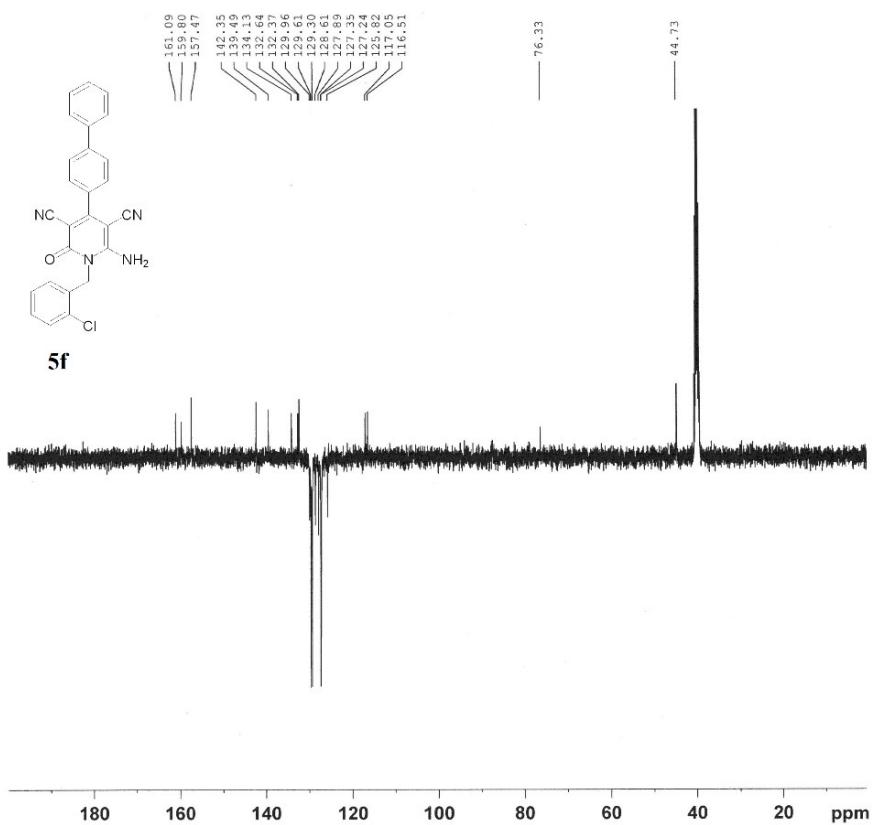
<sup>13</sup>C-APT (100 MHz, DMSO-d6)



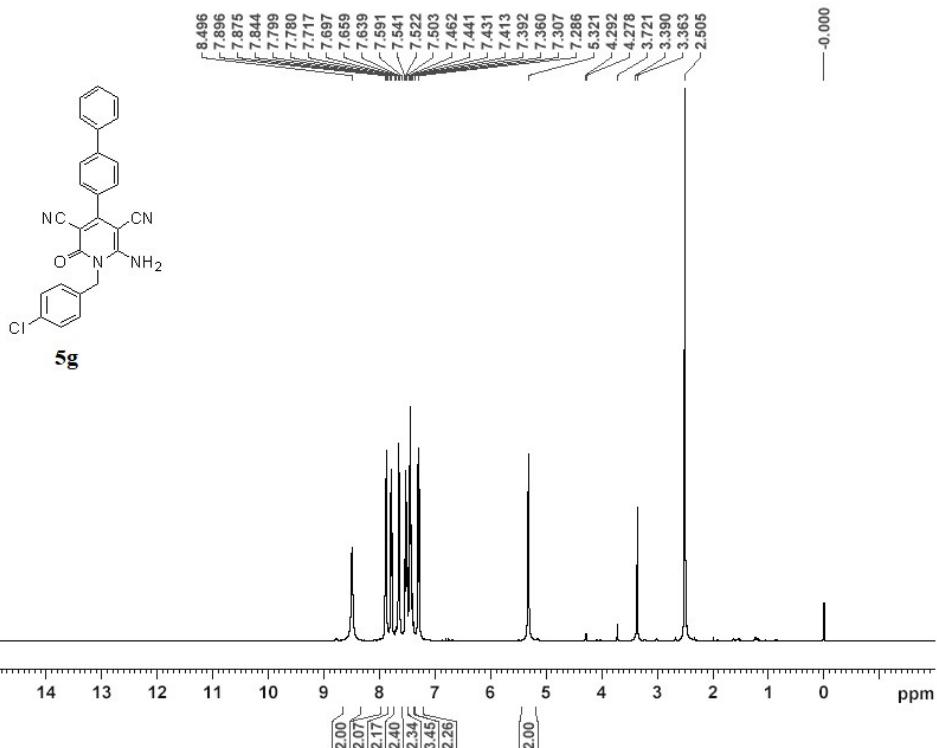
<sup>1</sup>H-NMR (400 MHz, DMSO-d6)



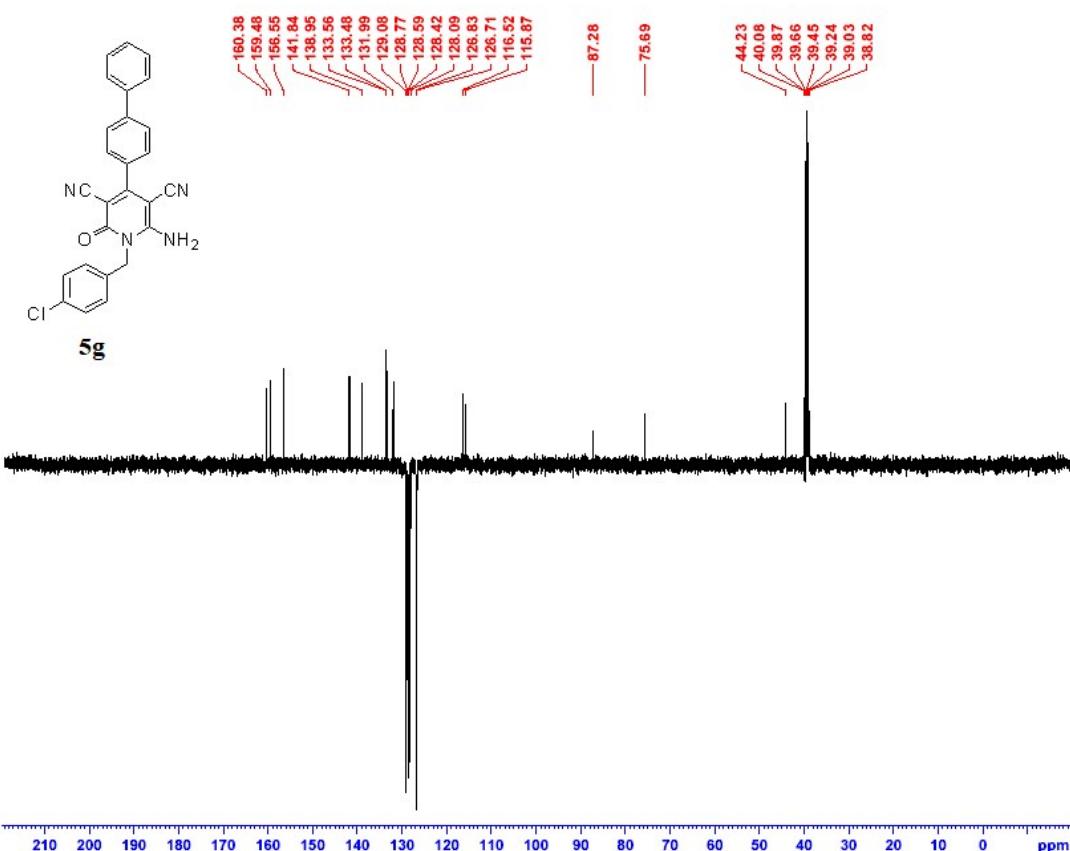
<sup>13</sup>C{<sup>1</sup>H}-APT (100 MHz, DMSO-d6)



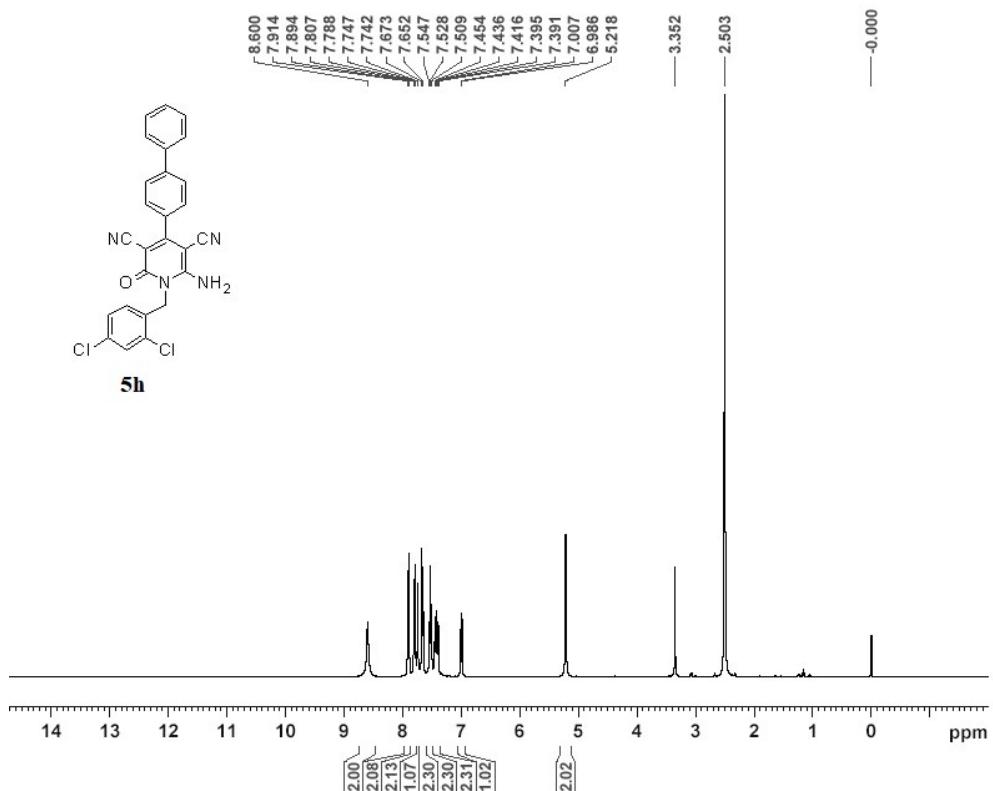
<sup>1</sup>H-NMR (400 MHz, DMSO-d6)



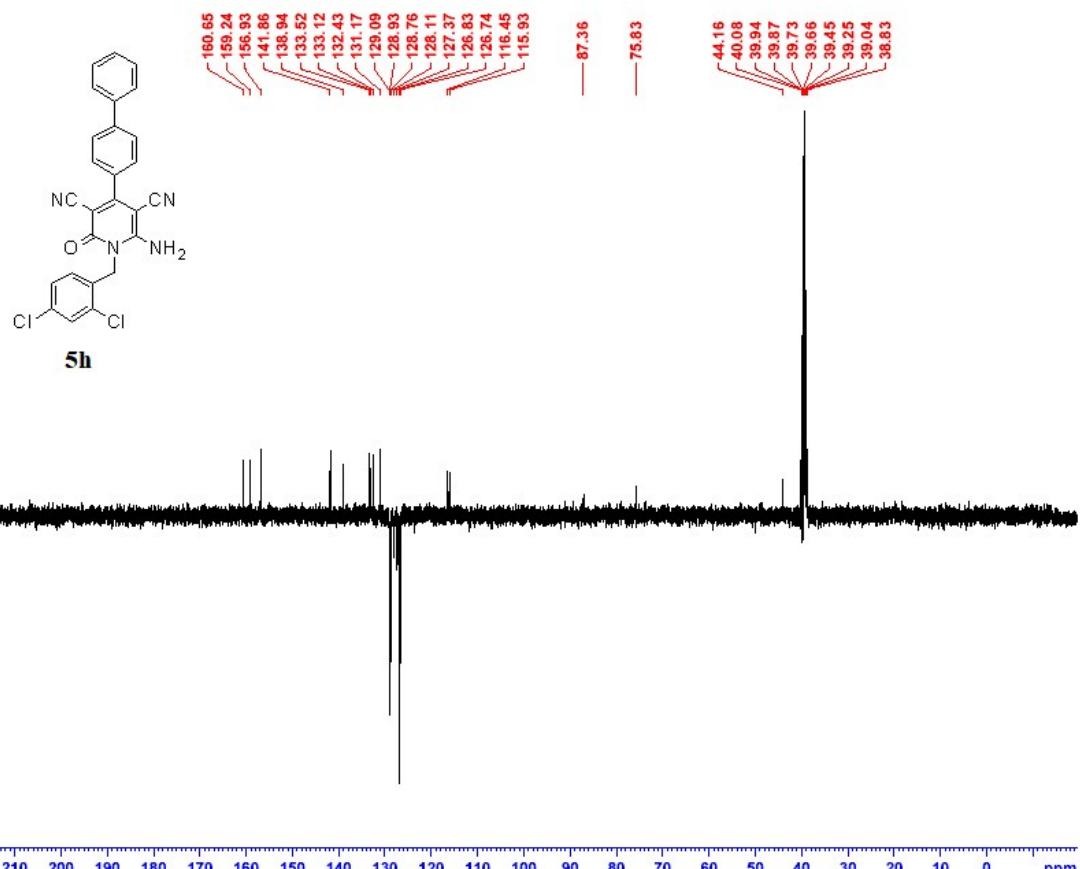
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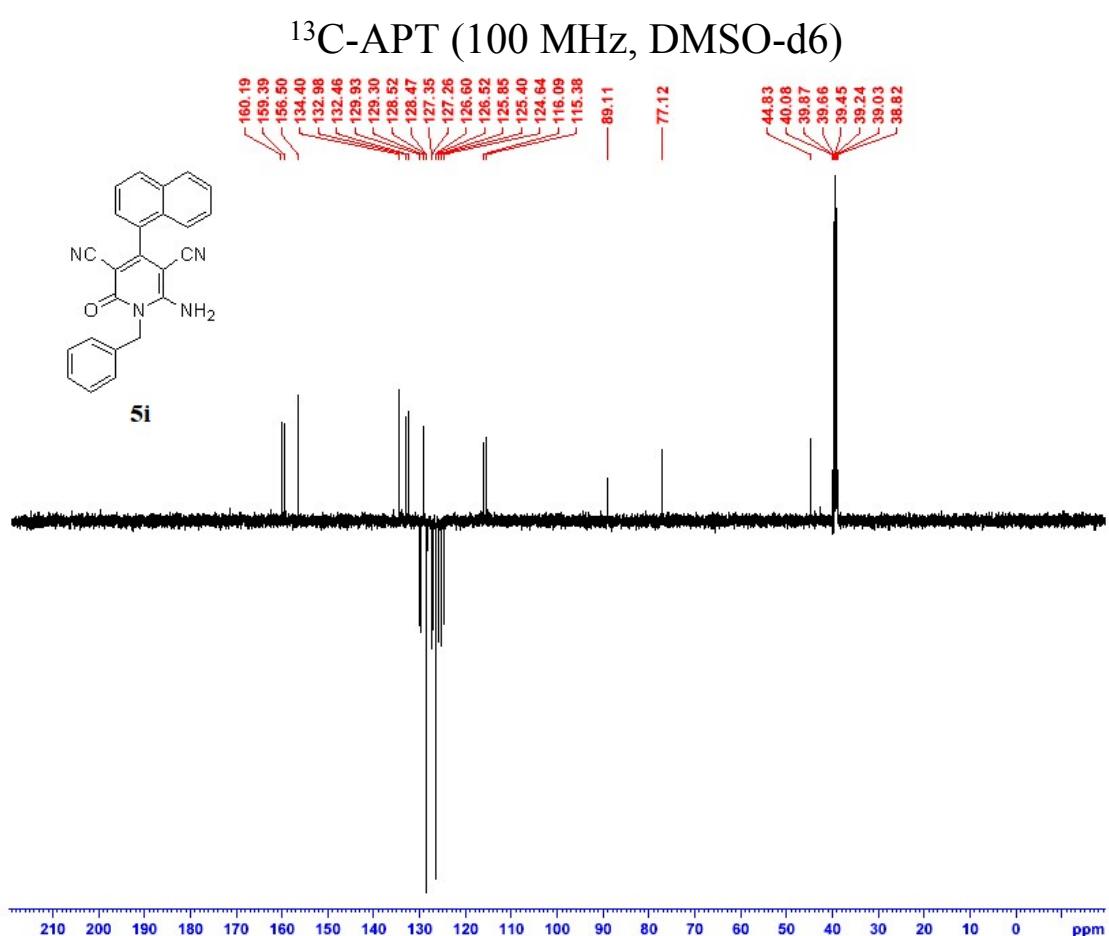
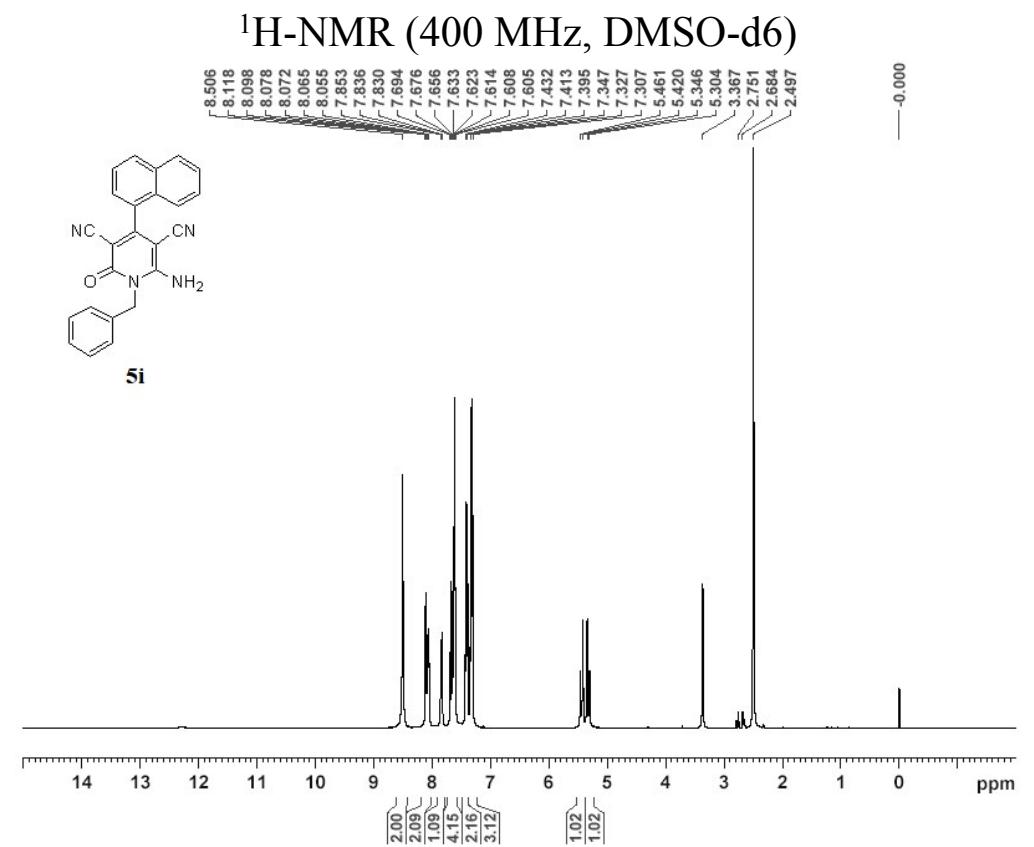


<sup>1</sup>H-NMR (400 MHz, DMSO-d6)

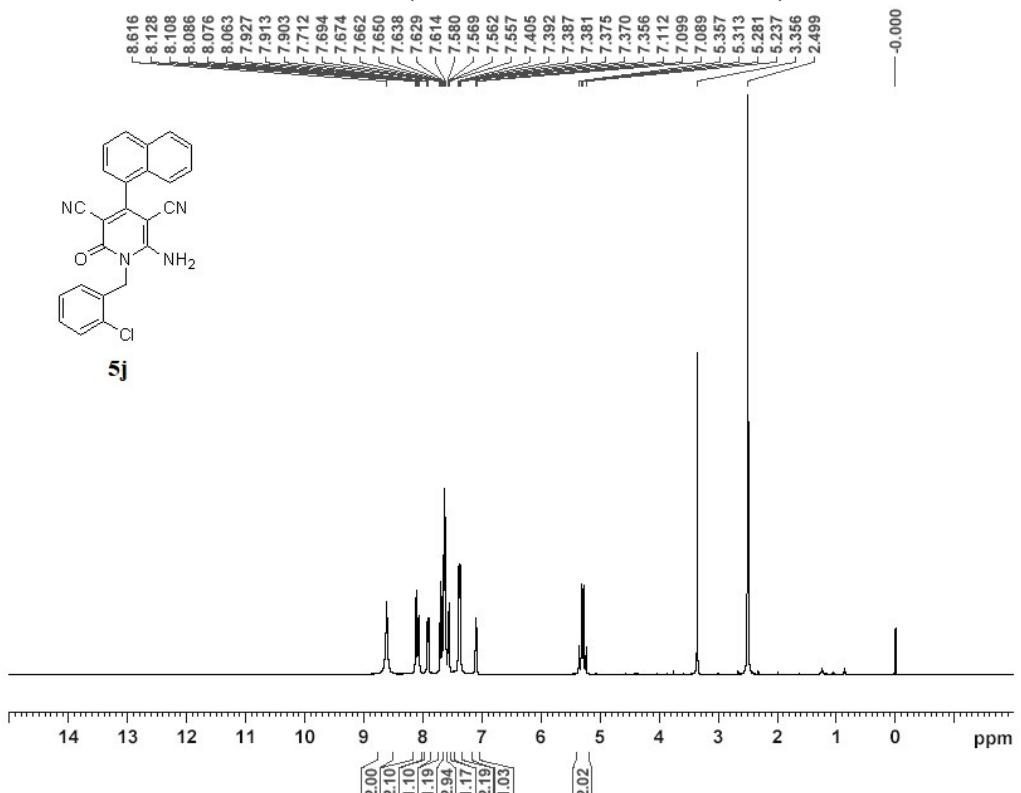


<sup>13</sup>C{<sup>1</sup>H}-APT (100 MHz, DMSO-d6)

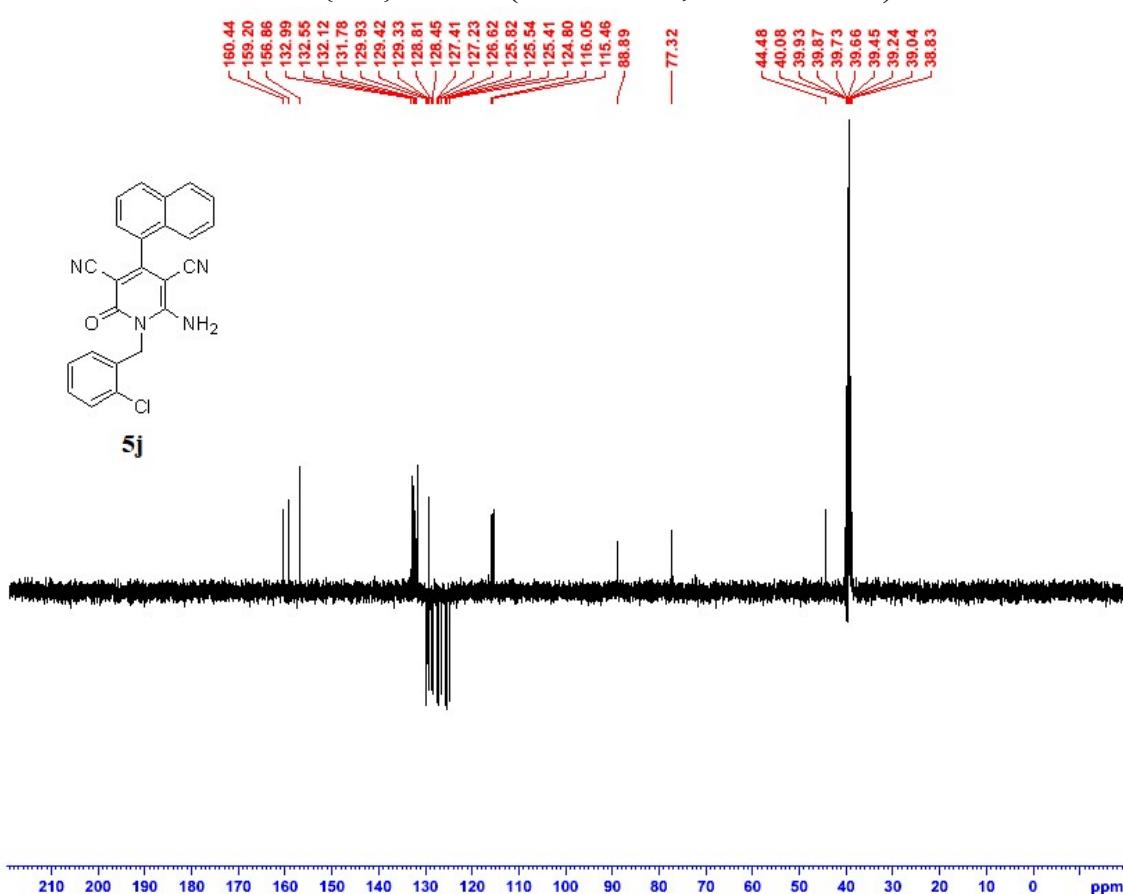




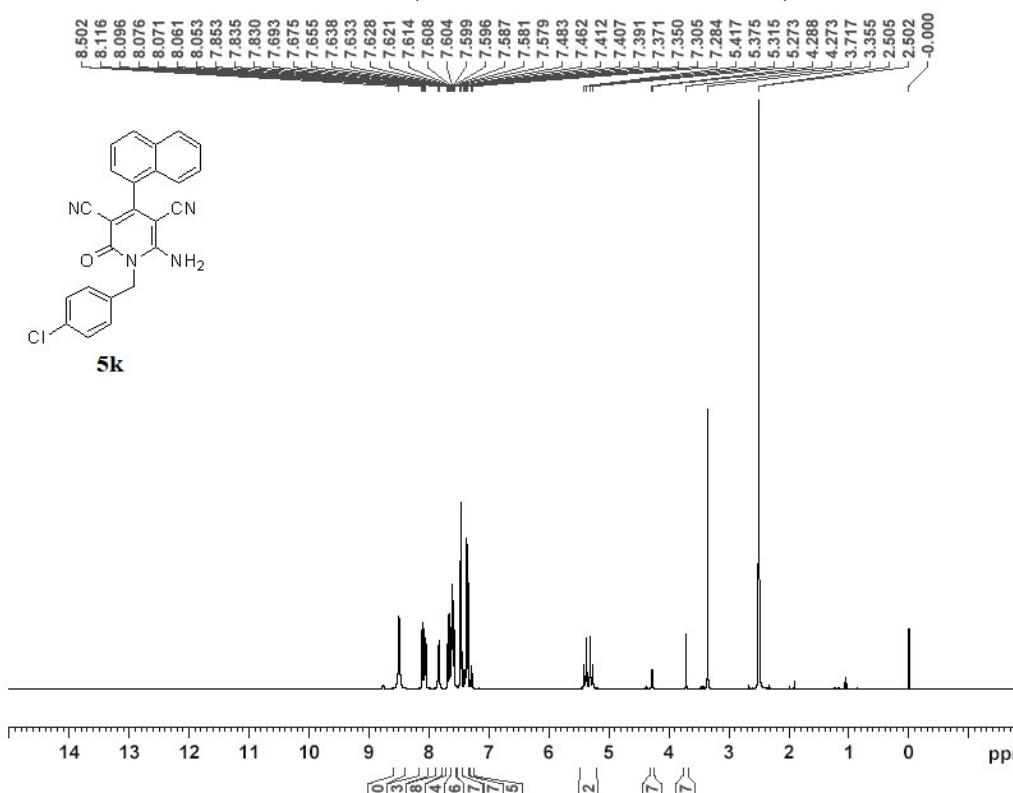
<sup>1</sup>H-NMR (400 MHz, DMSO-d6)



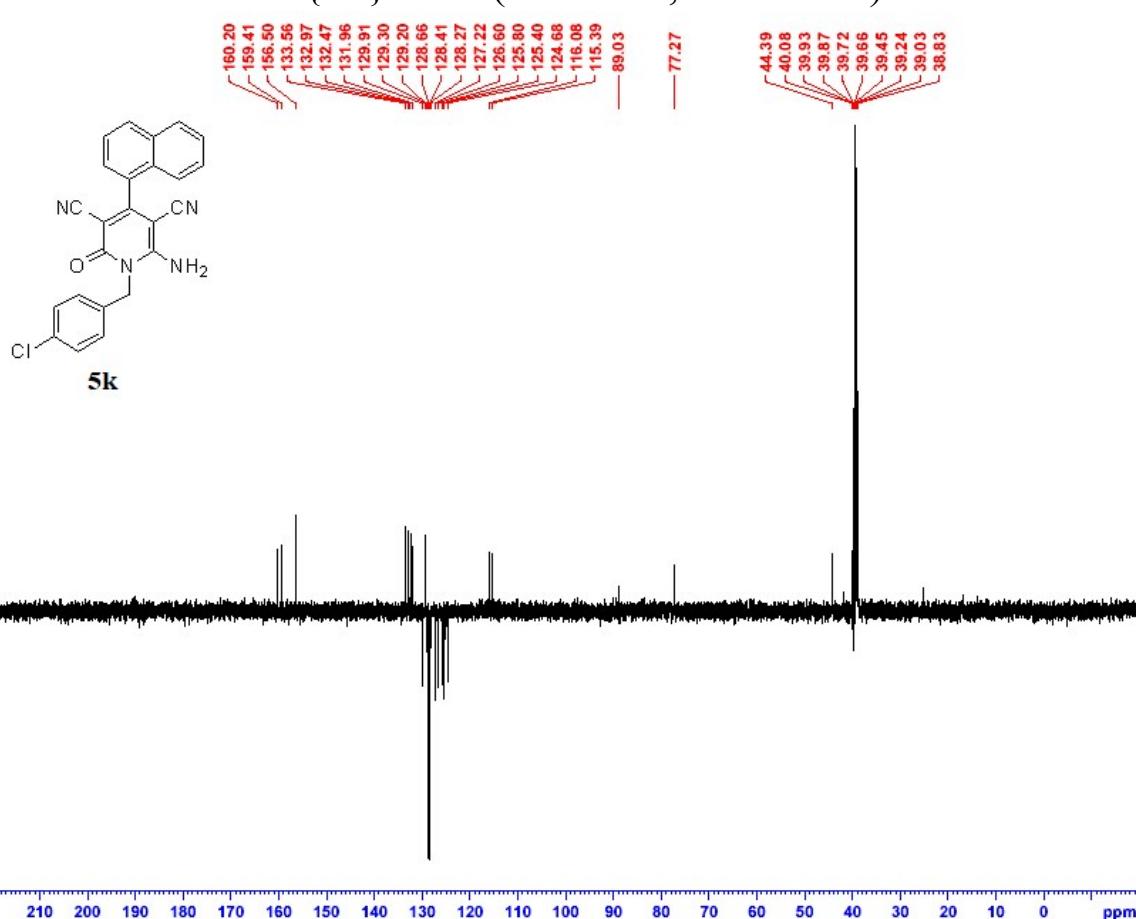
<sup>13</sup>C{<sup>1</sup>H}-APT (100 MHz, DMSO-d6)



<sup>1</sup>H-NMR (400 MHz, DMSO-d6)



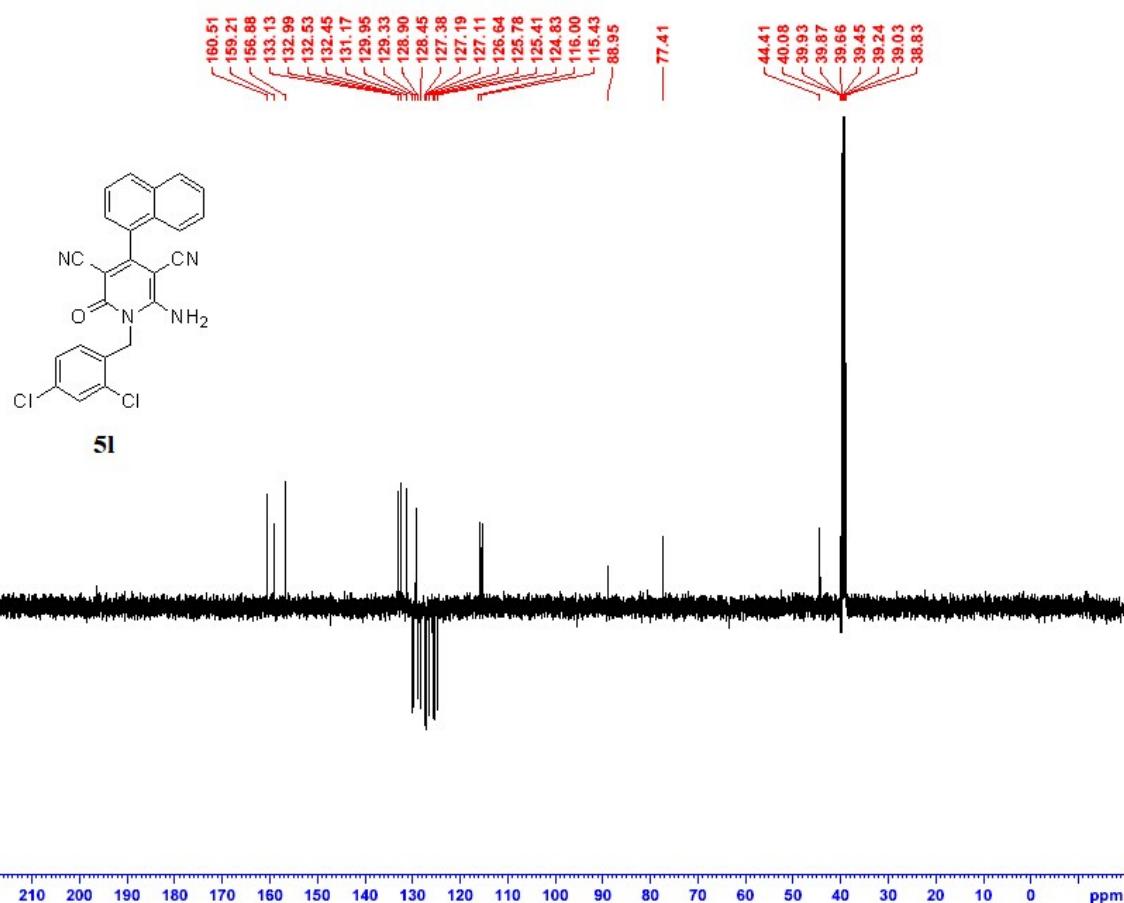
<sup>13</sup>C{<sup>1</sup>H}-APT (100 MHz, DMSO-d6)



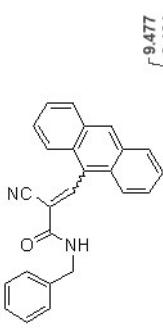
<sup>1</sup>H-NMR (400 MHz, DMSO-d6)



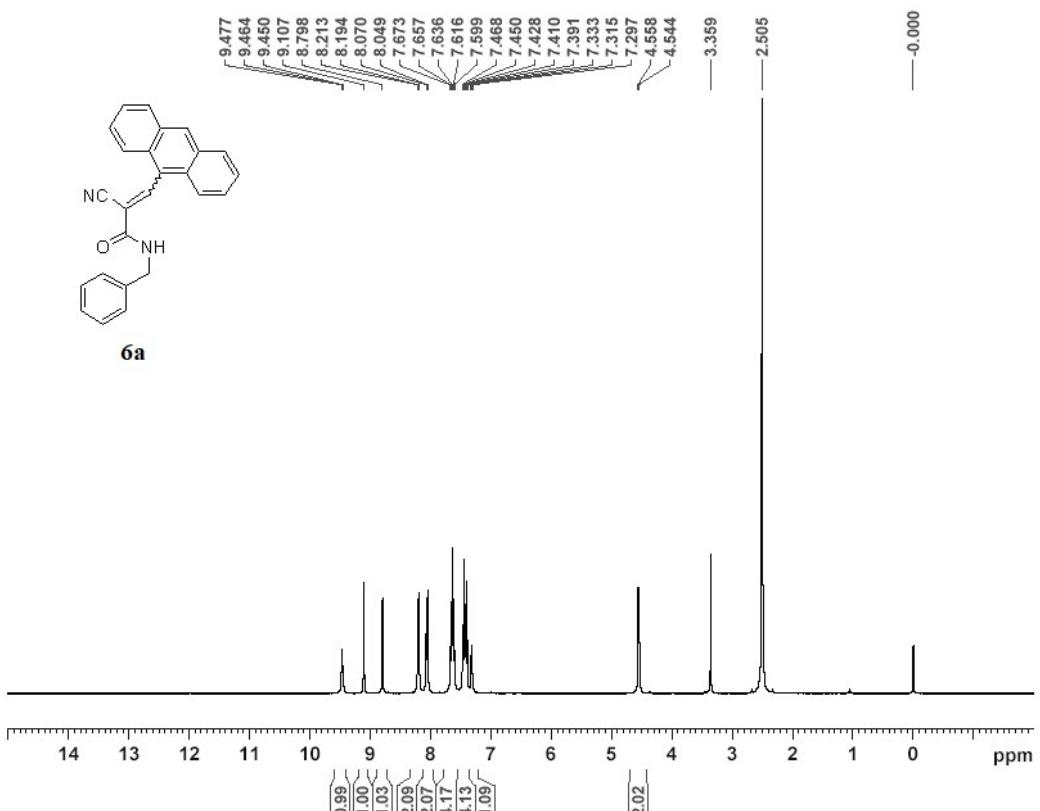
<sup>13</sup>C{<sup>1</sup>H}-APT (100 MHz, DMSO-d6)



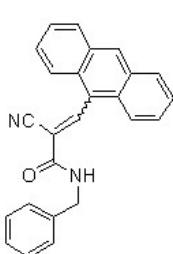
<sup>1</sup>H-NMR (400 MHz, DMSO-d<sub>6</sub>)



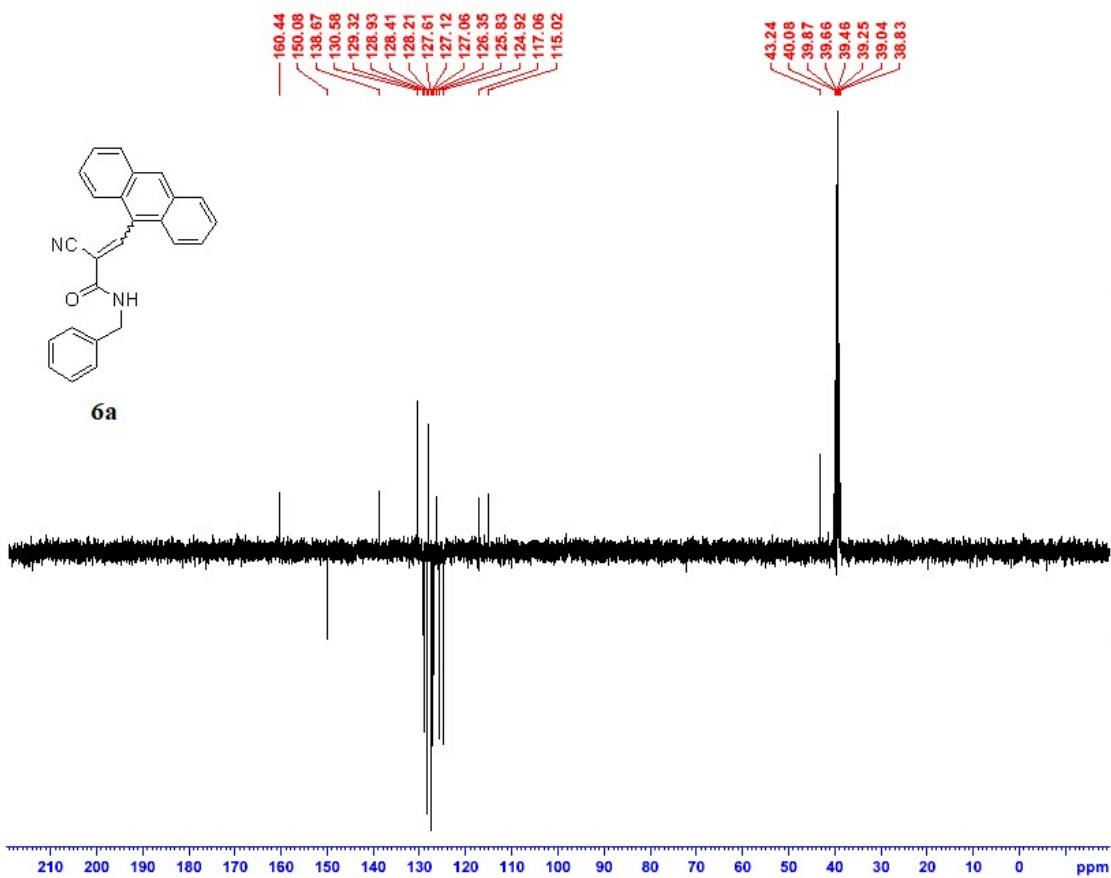
6а



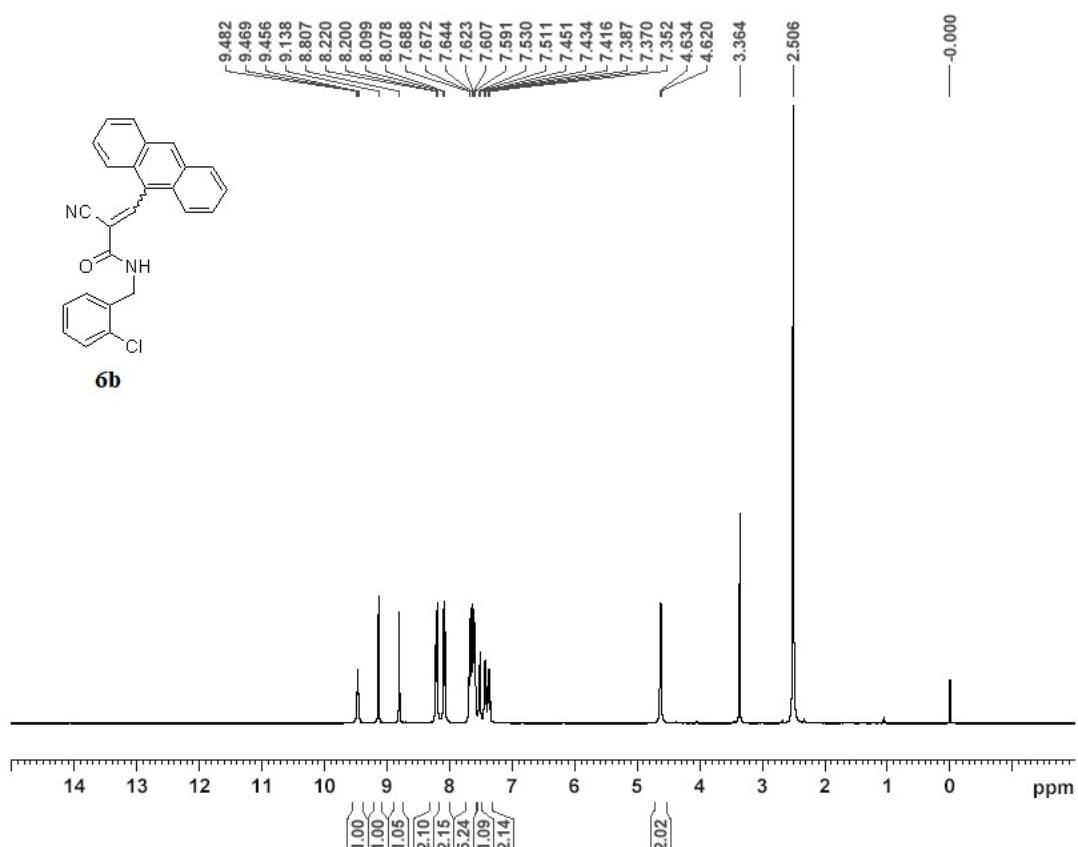
### <sup>13</sup>C{<sup>1</sup>H}-APT (100 MHz, DMSO-d<sub>6</sub>)



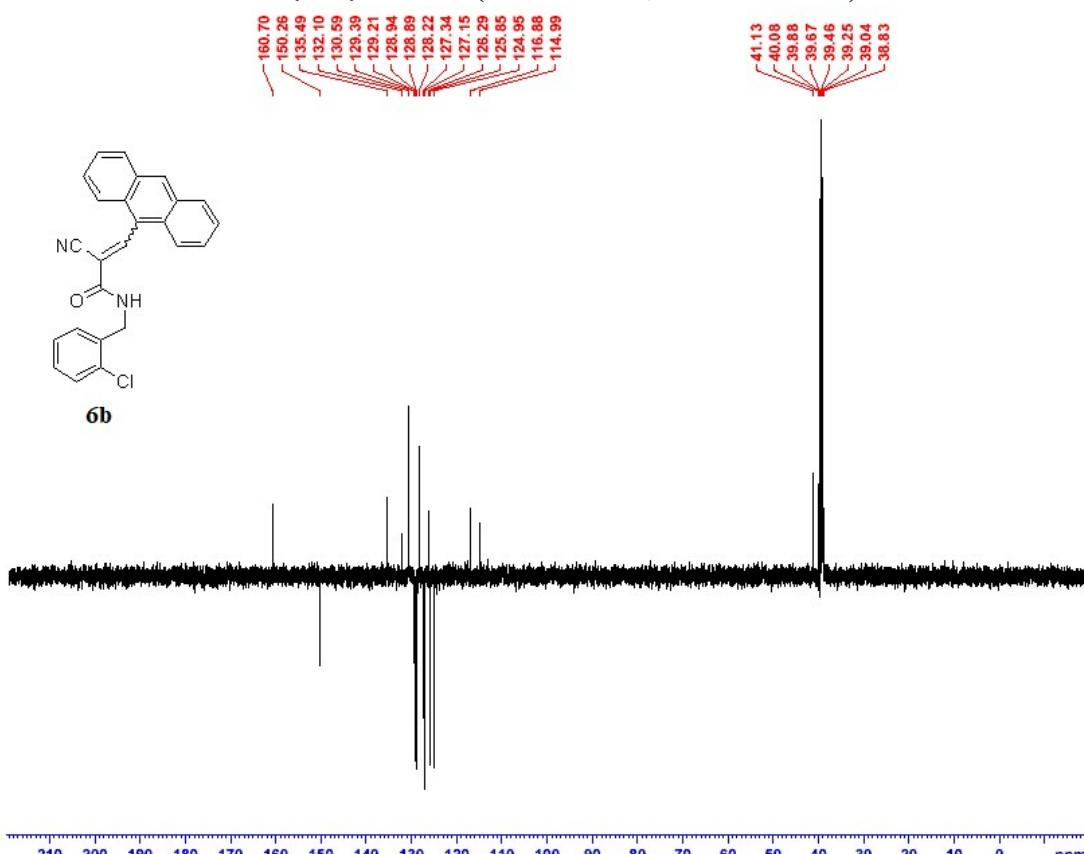
6а



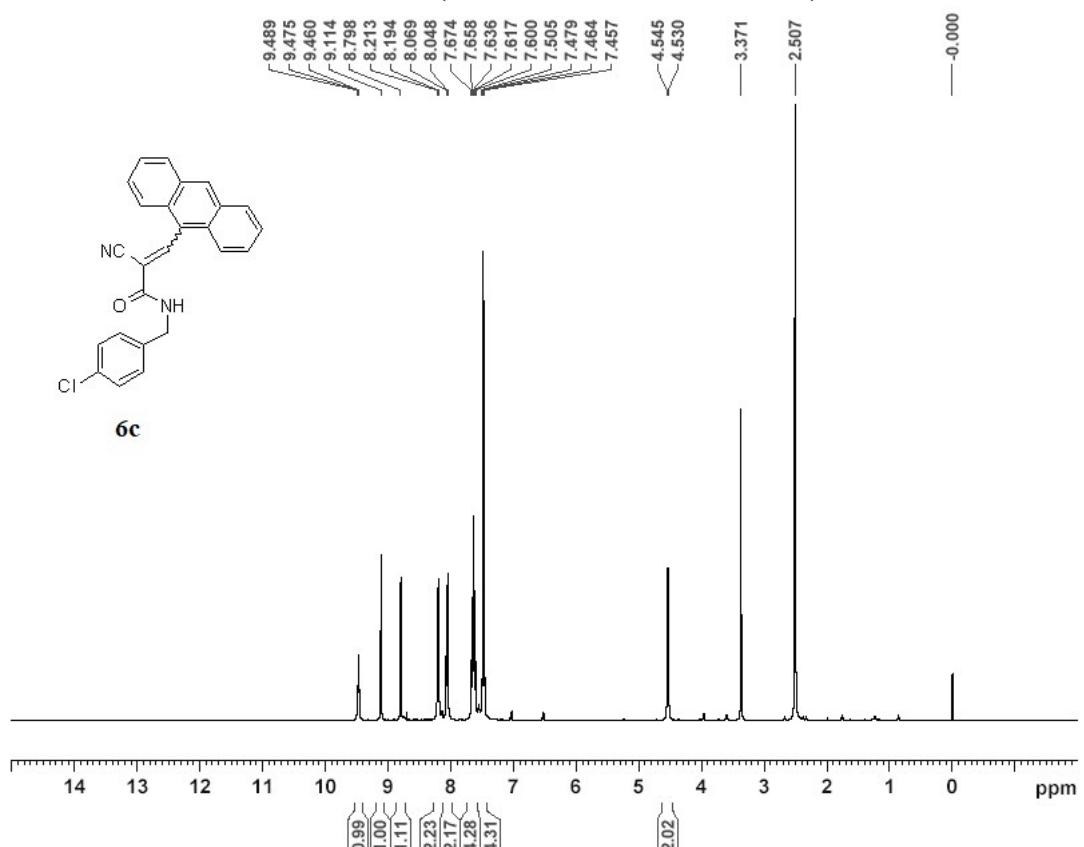
<sup>1</sup>H-NMR (400 MHz, DMSO-d6)



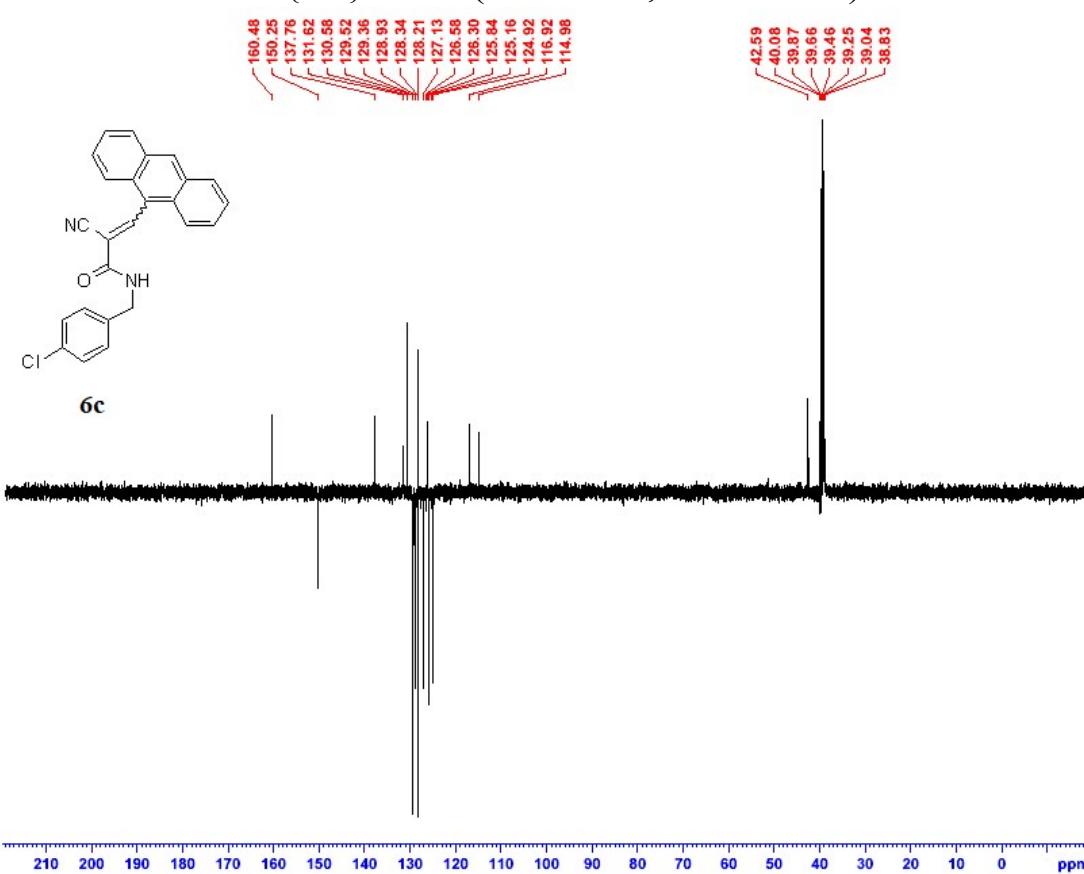
<sup>13</sup>C{<sup>1</sup>H}-APT (100 MHz, DMSO-d6)



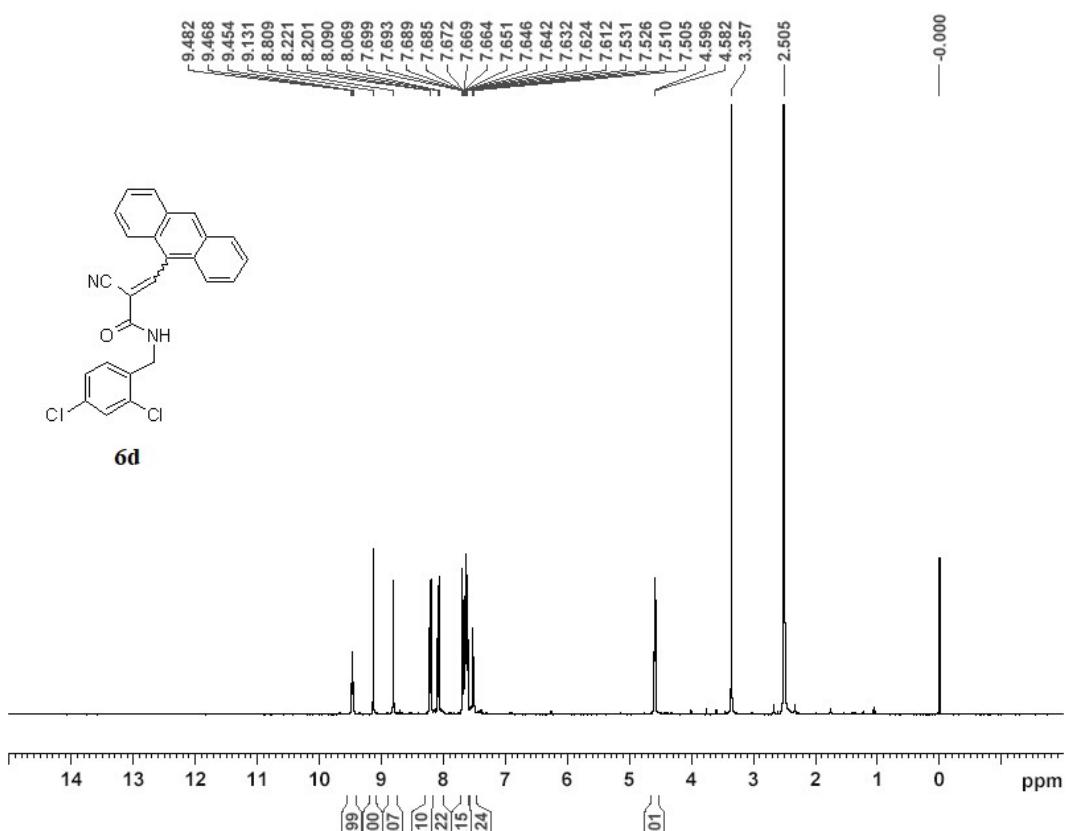
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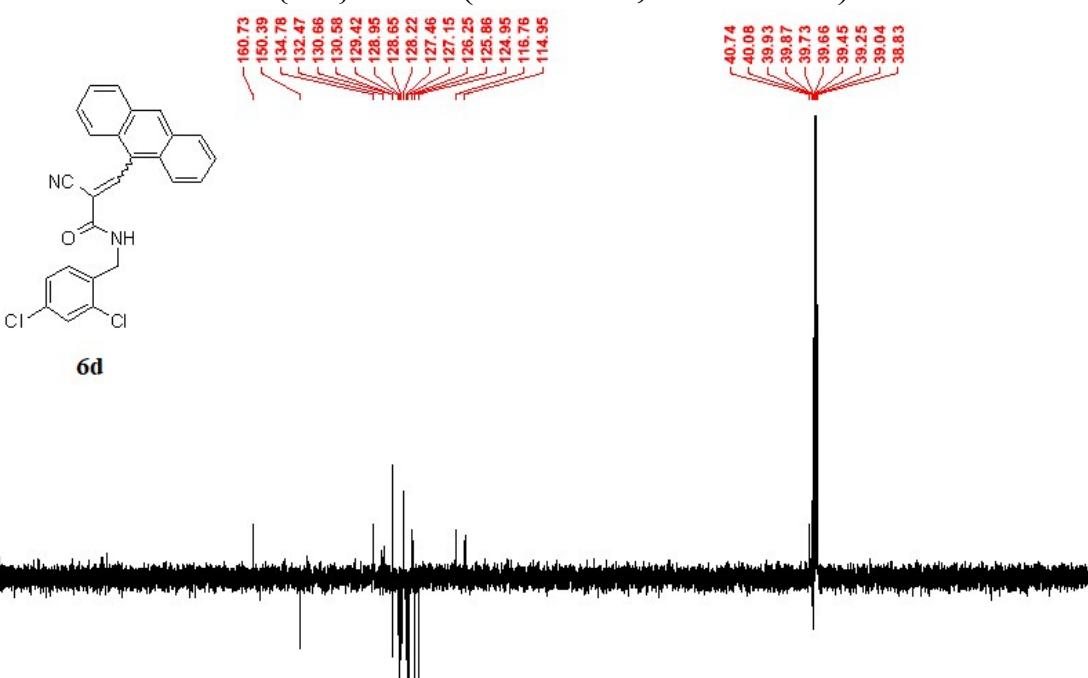
<sup>13</sup>C{<sup>1</sup>H}-APT (100 MHz, DMSO-d6)



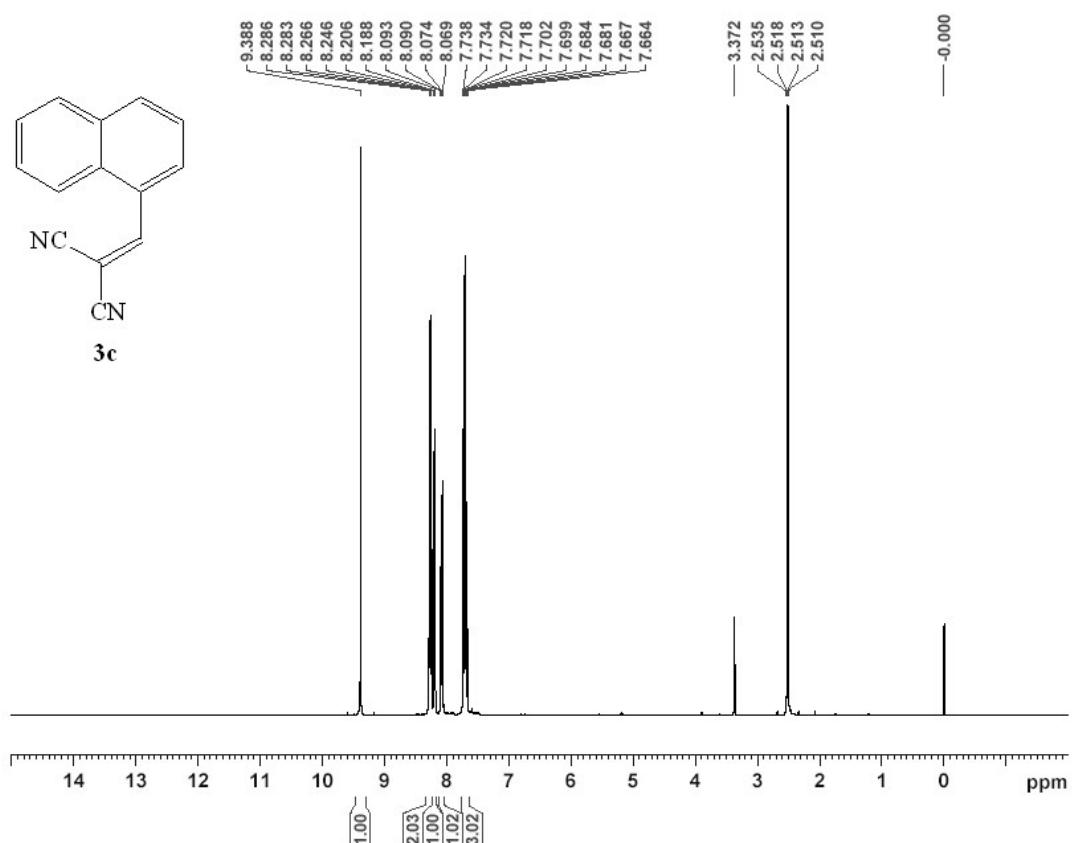
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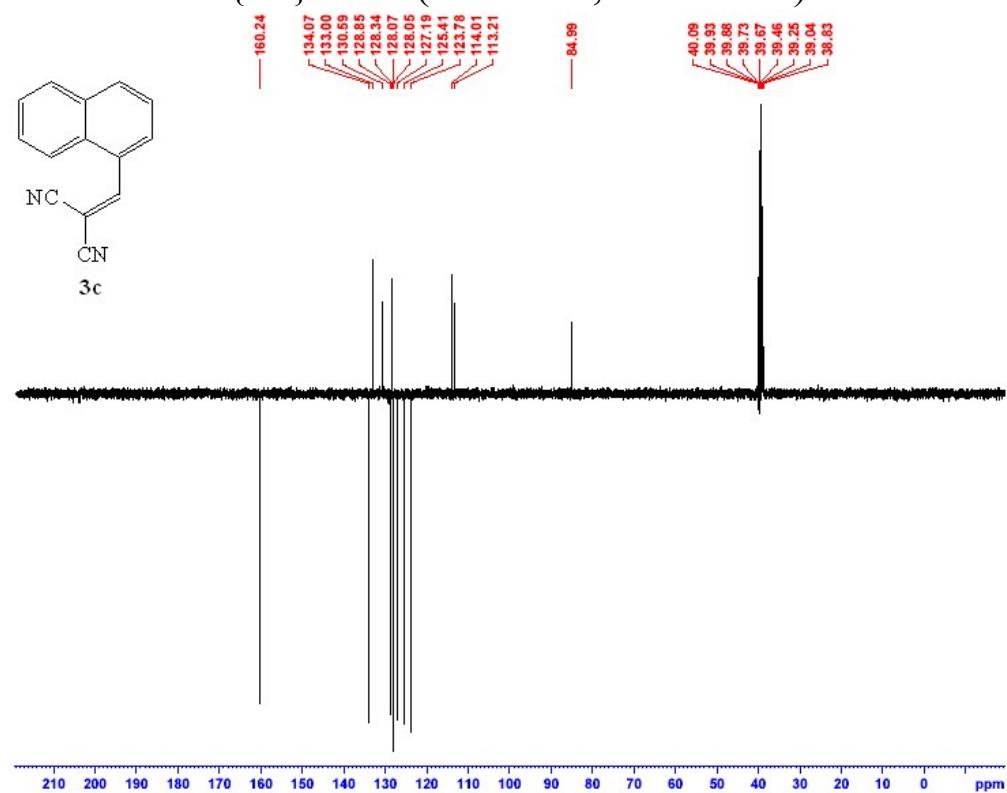
<sup>13</sup>C{<sup>1</sup>H}-APT (100 MHz, DMSO-d6)



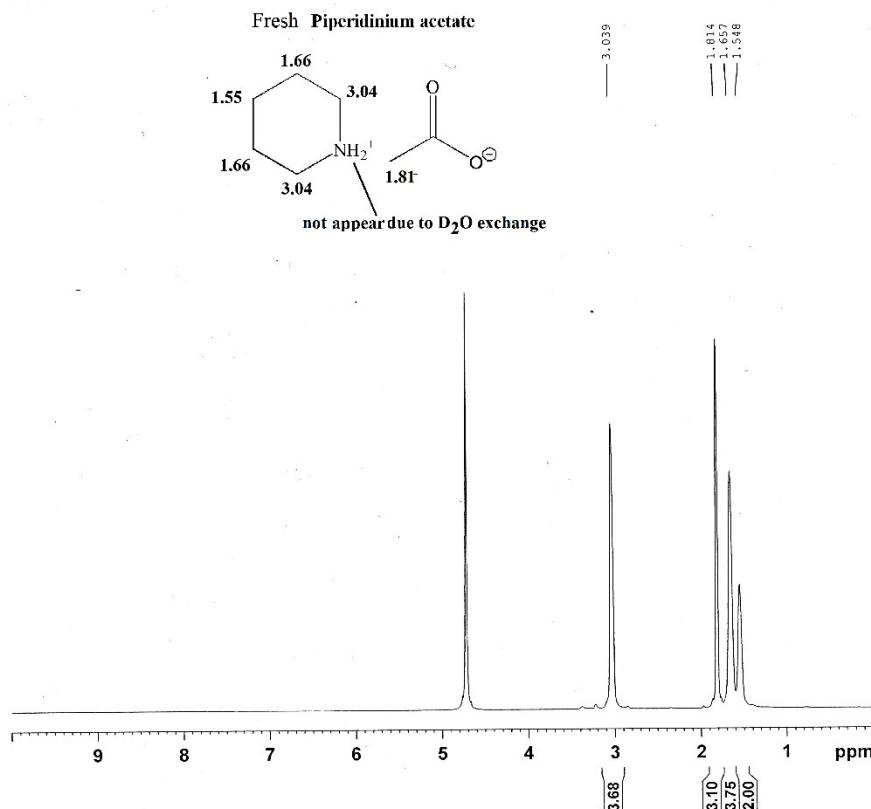
<sup>1</sup>H-NMR (400 MHz, DMSO-d6)



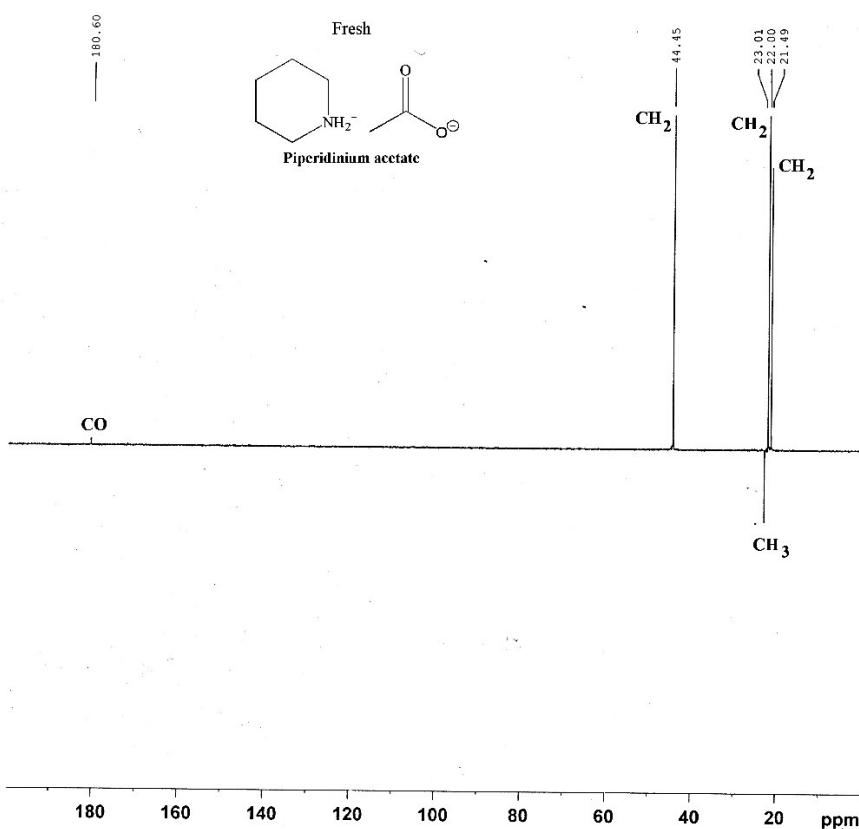
<sup>13</sup>C{<sup>1</sup>H}-APT (100 MHz, DMSO-d6)



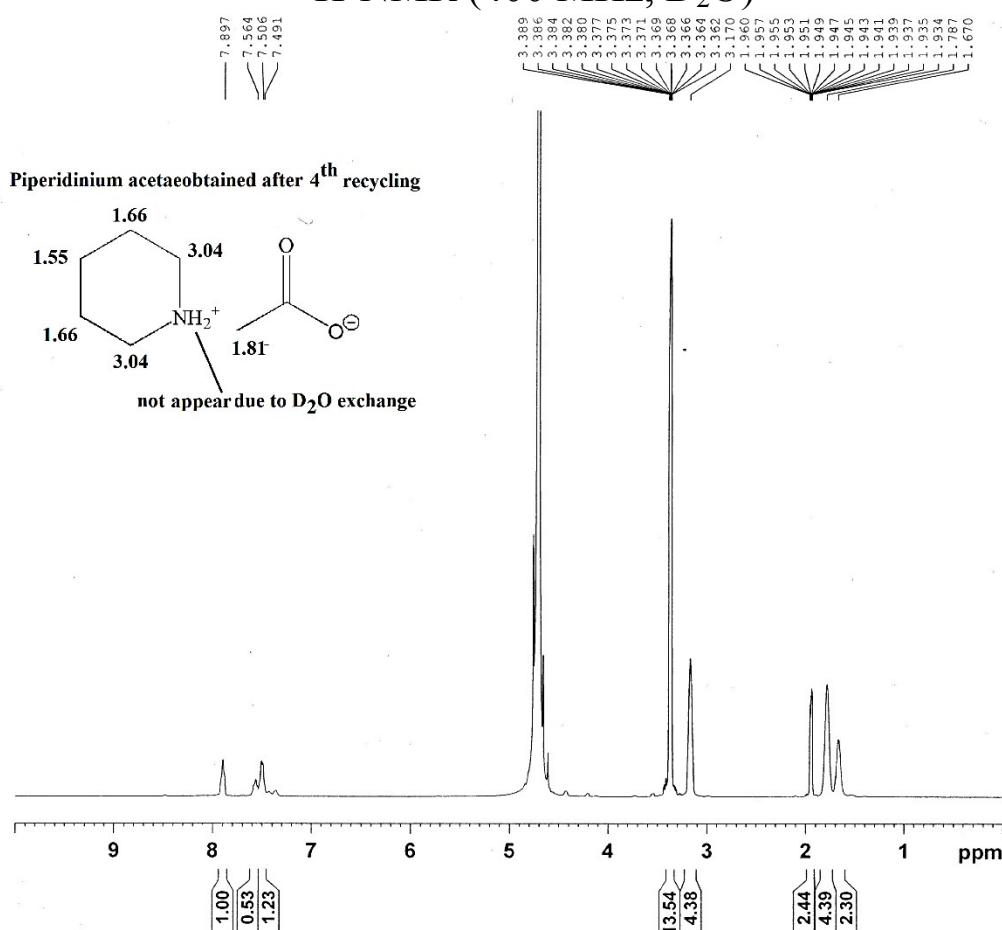
<sup>1</sup>H-NMR (400 MHz, D<sub>2</sub>O)



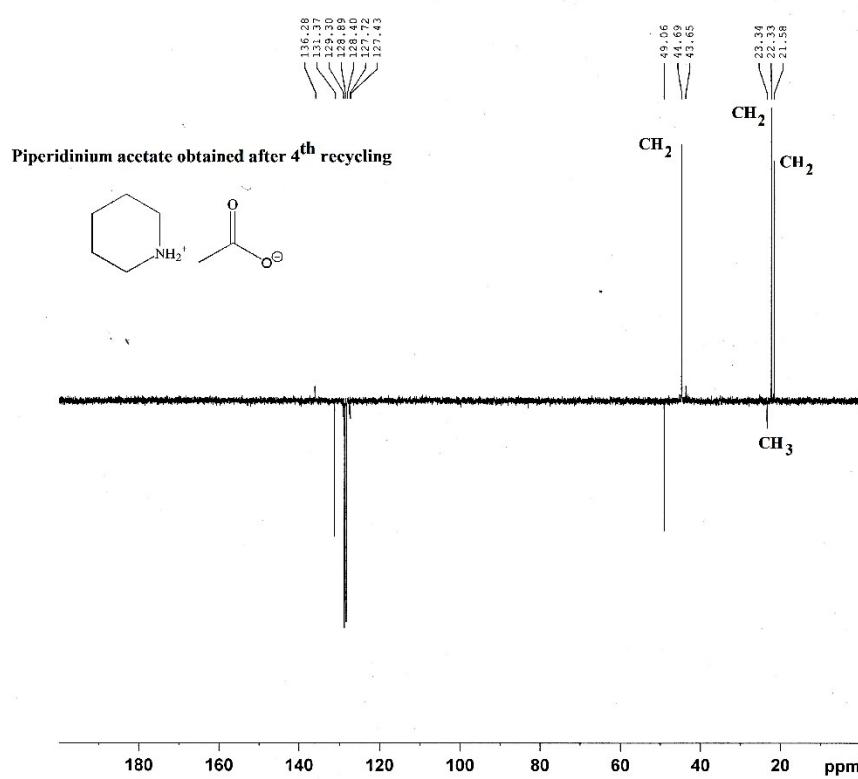
<sup>13</sup>C{<sup>1</sup>H}-APT (100 MHz, D<sub>2</sub>O)



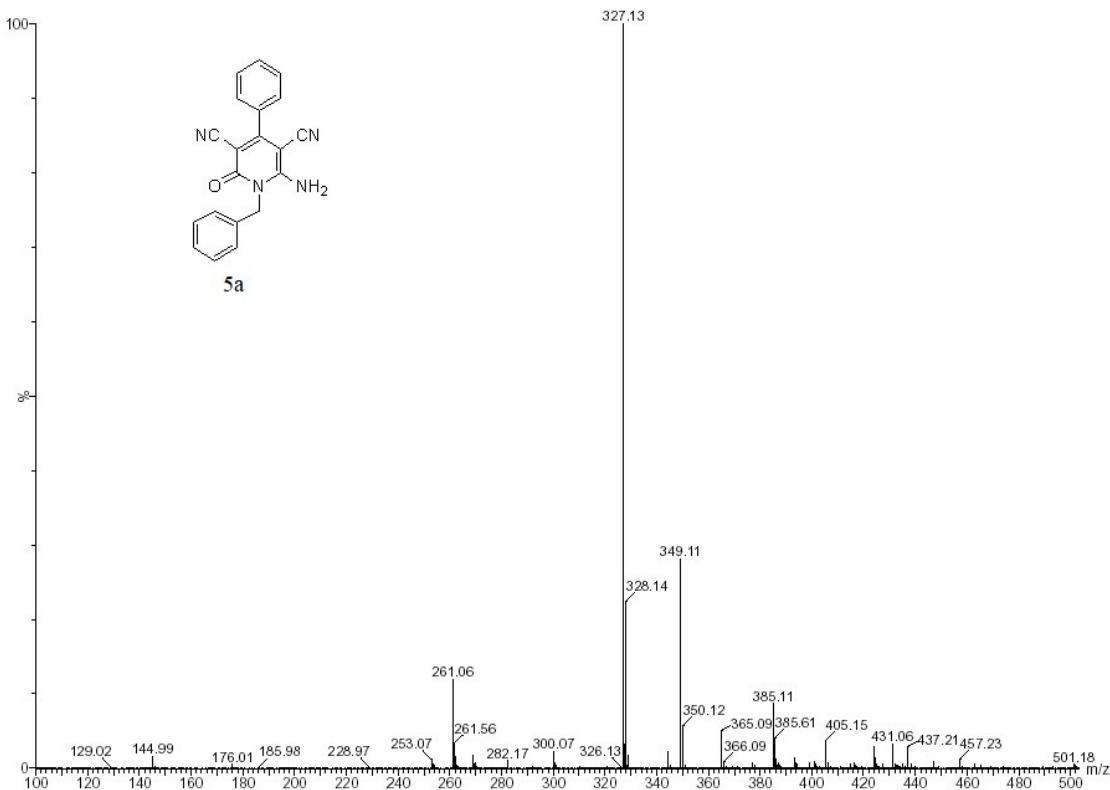
<sup>1</sup>H-NMR (400 MHz, D<sub>2</sub>O)



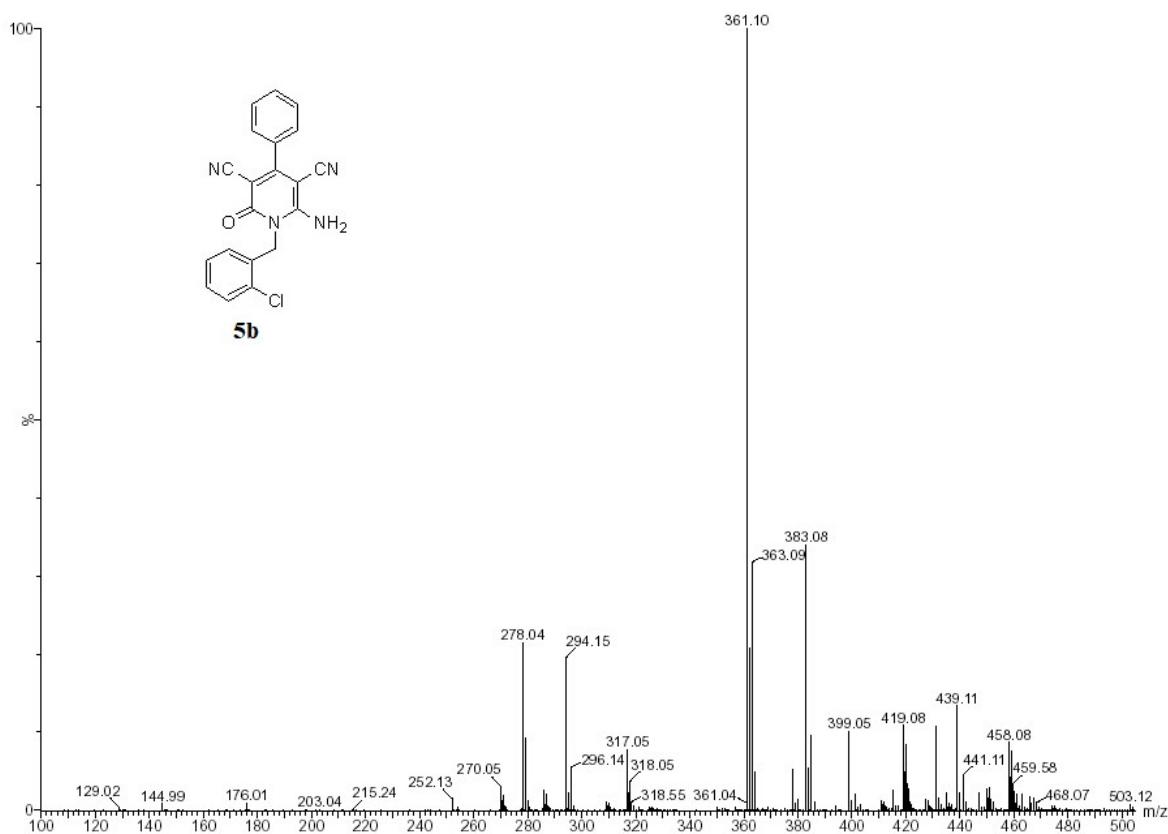
### <sup>13</sup>C{<sup>1</sup>H}-APT (100 MHz, D<sub>2</sub>O)



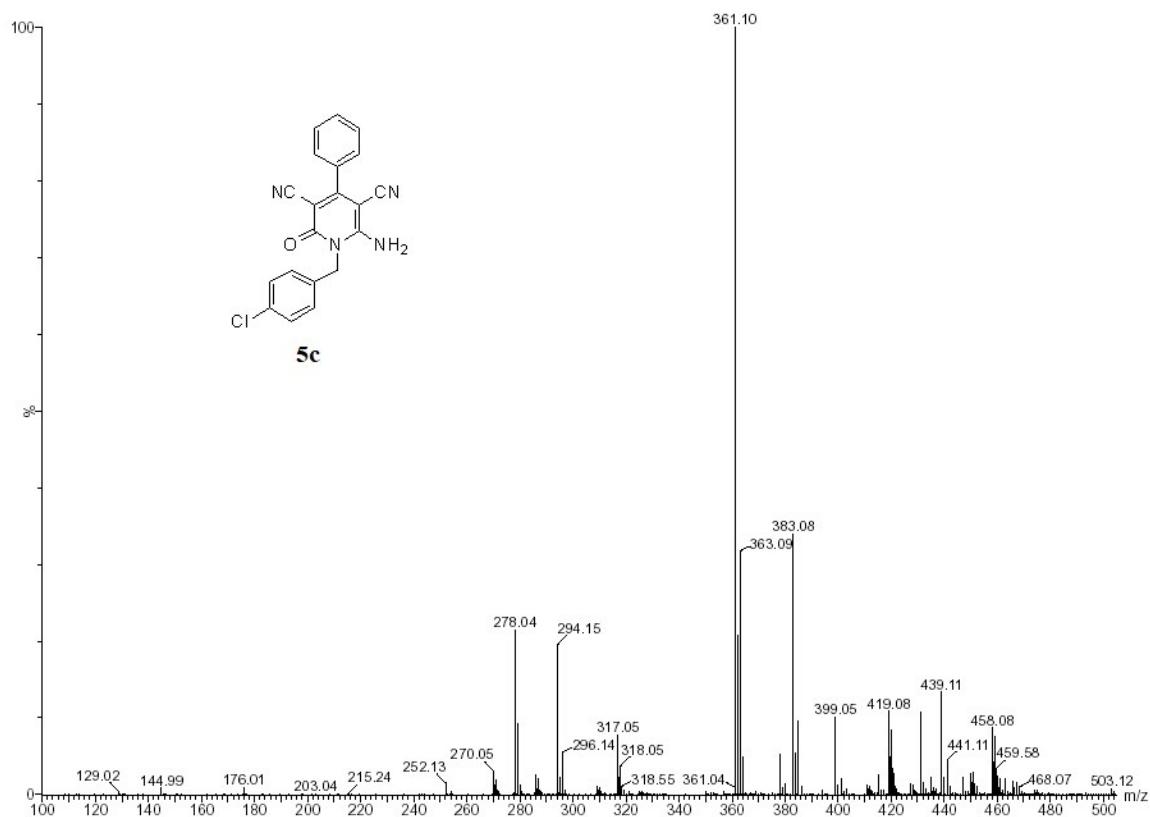
### Mass spectrum of 5a



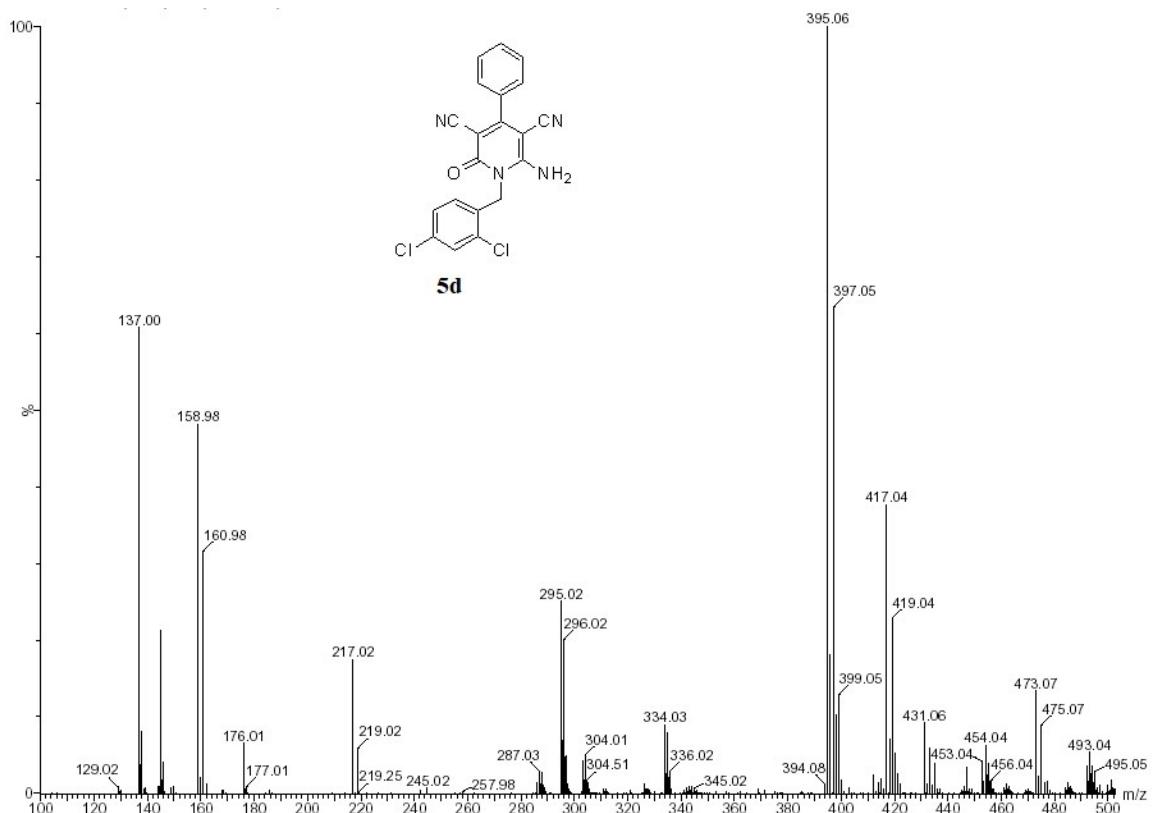
### Mass spectrum of 5b



Mass spectrum of 5c



### Mass spectrum of 5d



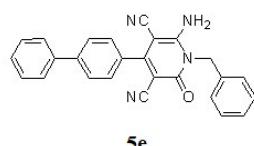
### Mass spectrum of 5e

Retention Time (MS)

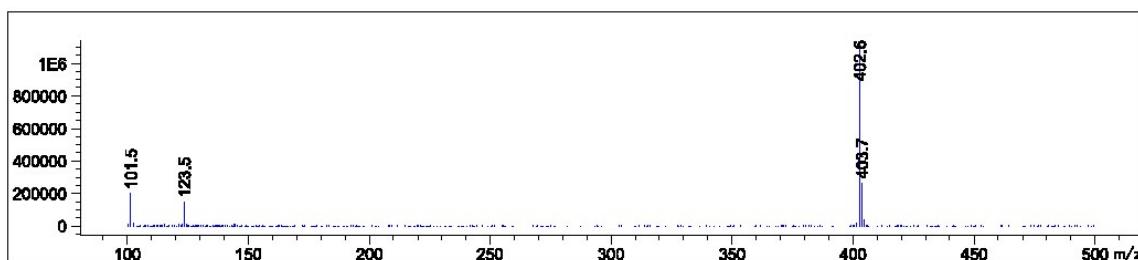
MS Area

Mol. Weight or Ion

5.595	<b>8492896</b>	<b>403.70 I</b>
		<b>402.60 I</b>
		<b>123.50 I</b>
		<b>101.50 I</b>

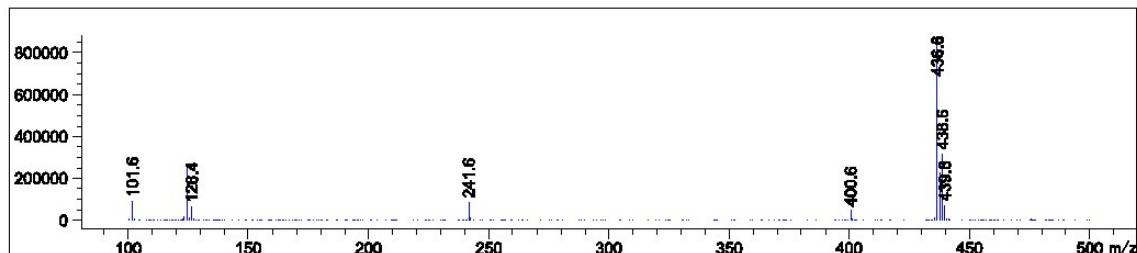
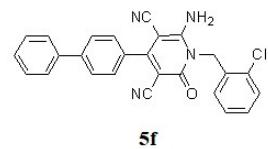


**5e**



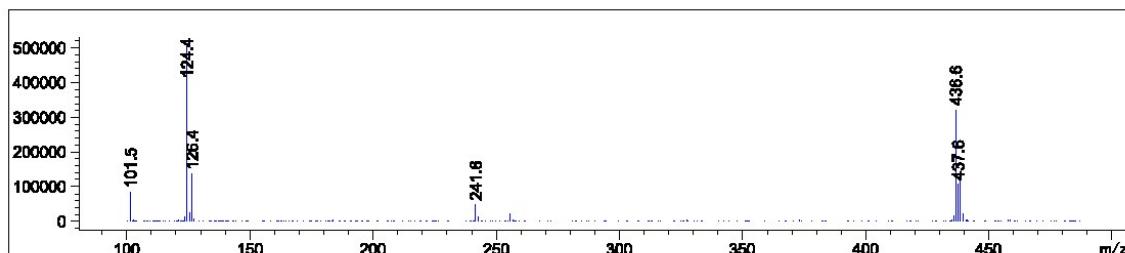
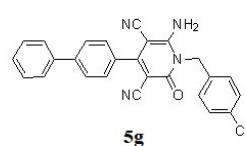
### Mass spectrum of 5f

Retention Time (MS)	MS Area	Mol. Weight or Ion
5.752	8504891	438.60 I 437.60 I 436.60 I 241.65 I 124.40 I 101.60 I



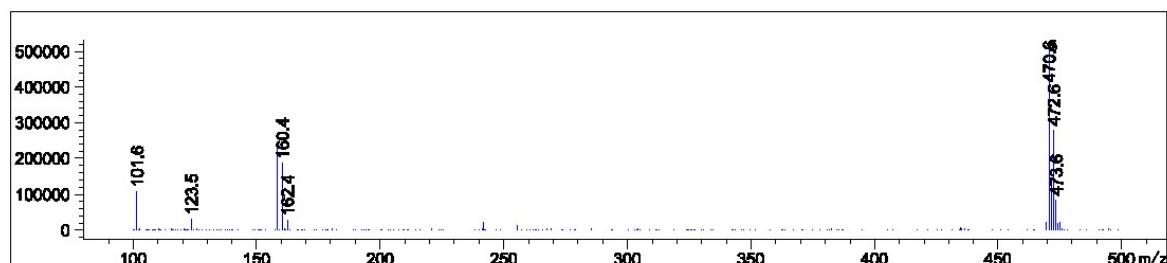
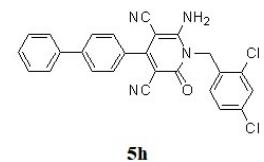
Mass spectrum of 5g

Retention Time (MS)	MS Area	Mol. Weight or Ion
5.826	5850985	438.60 I 437.60 I 436.60 I 126.40 I 124.40 I 101.50 I



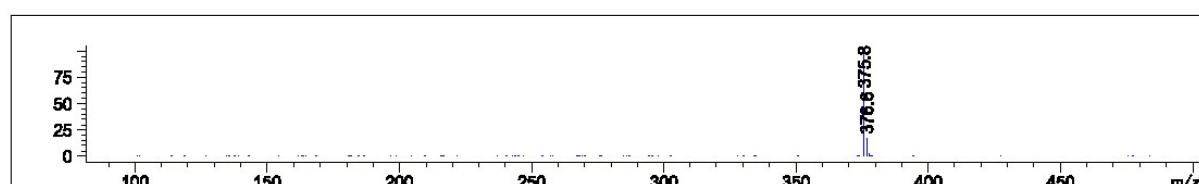
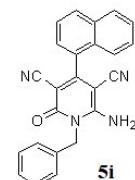
Mass spectrum of 5h

Retention Time (MS)	MS Area	Mol. Weight or Ion
5.989	5433183	473.65 I
		472.60 I
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		160.35 I
		158.40 I
		101.60 I



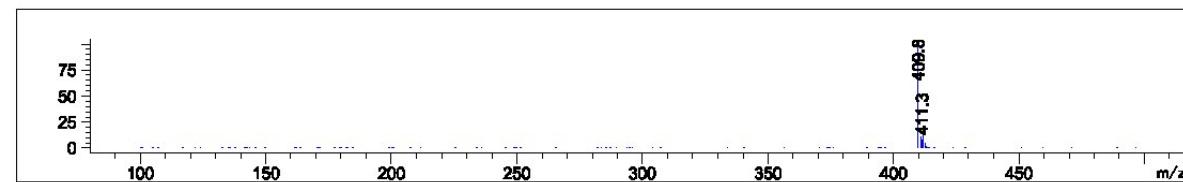
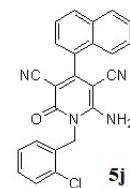
Mass spectrum of 5i

Retention Time (MS)	MS Area	Mol. Weight or Ion
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		375.80 I

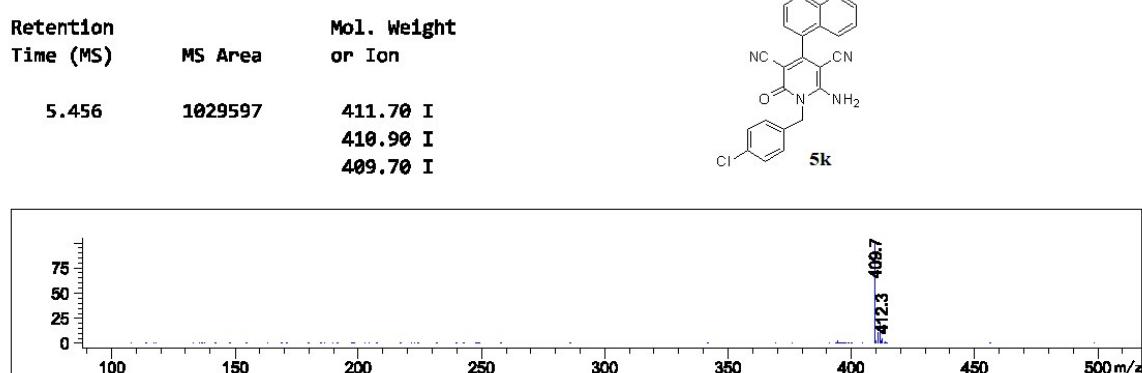


Mass spectrum of 5j

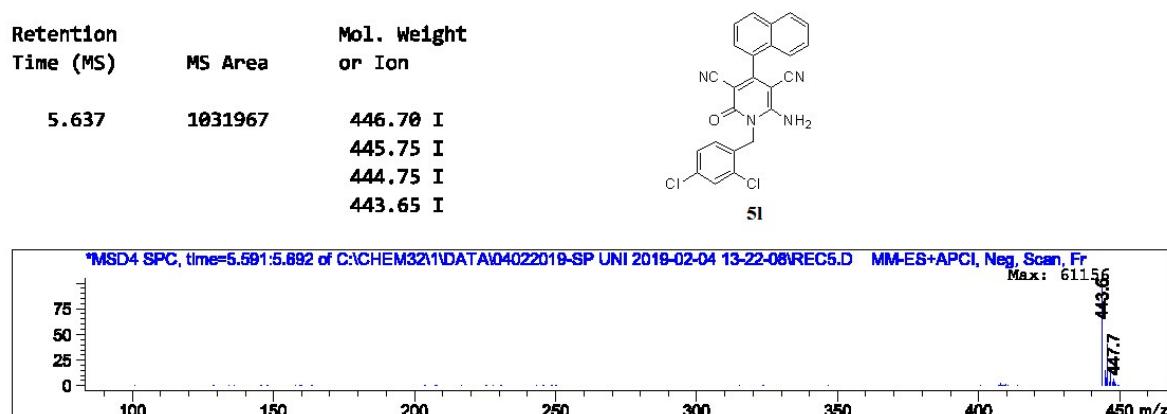
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5.372	973178	411.70 I
		410.95 I
		409.80 I



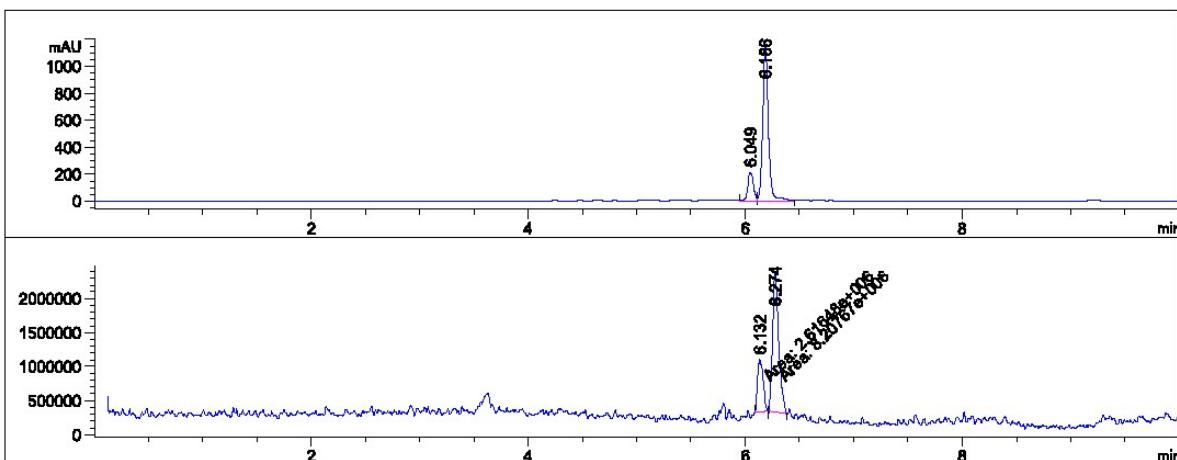
Mass spectrum of 5k



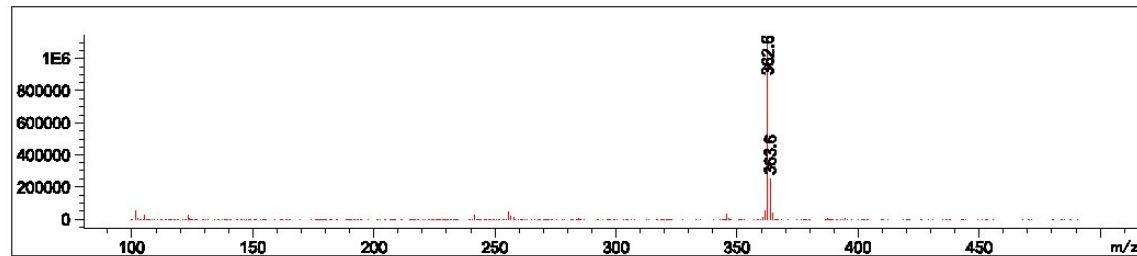
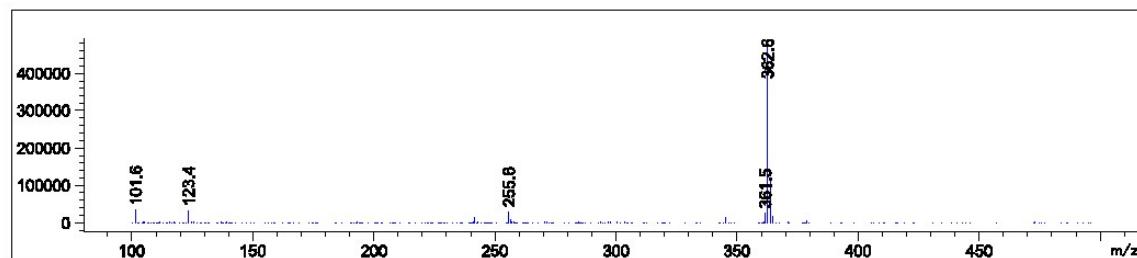
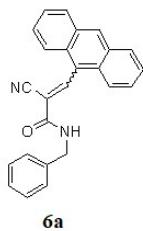
Mass spectrum of 5l



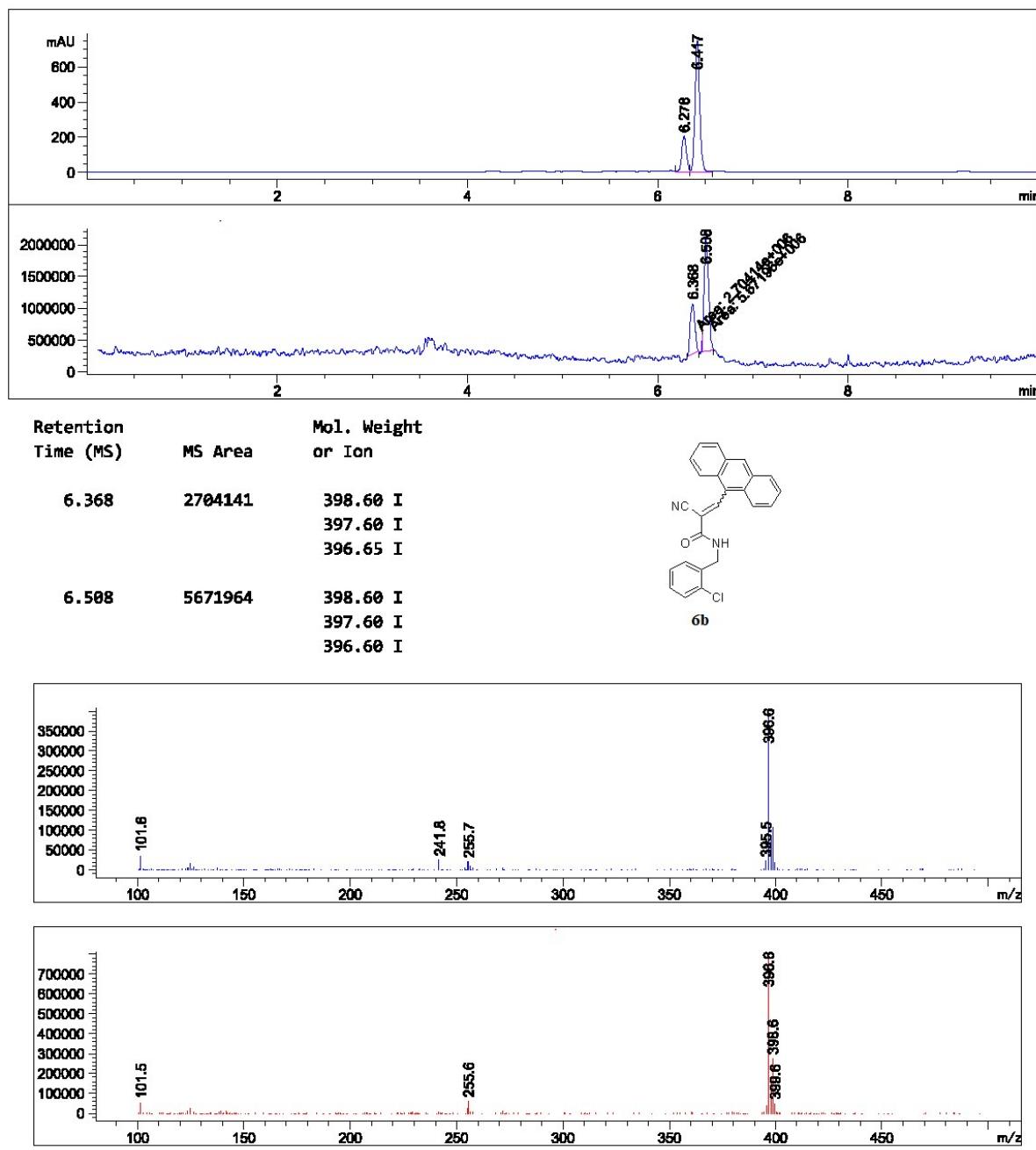
## LCMS of 6a



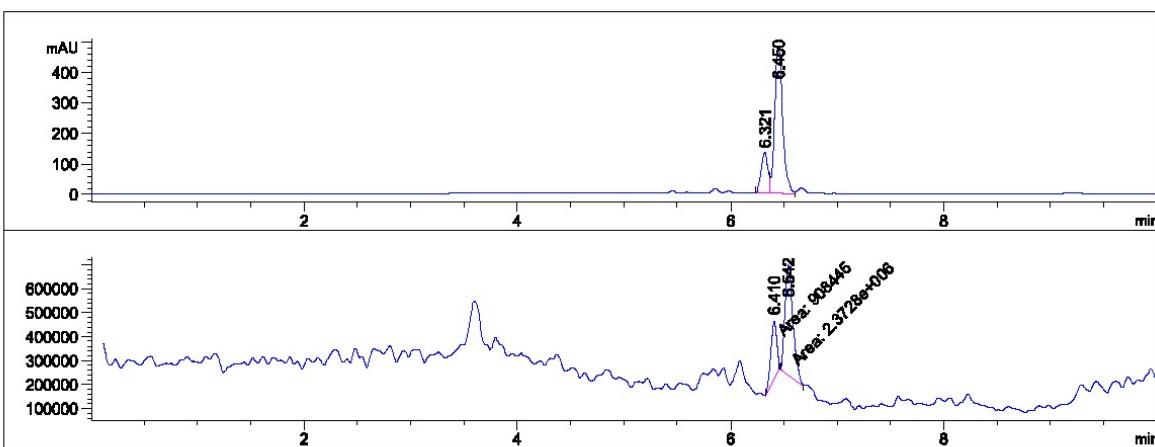
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6.132	2616479	363.60 I 362.60 I
6.274	8207667	363.60 I 362.60 I



## LCMS of 6b

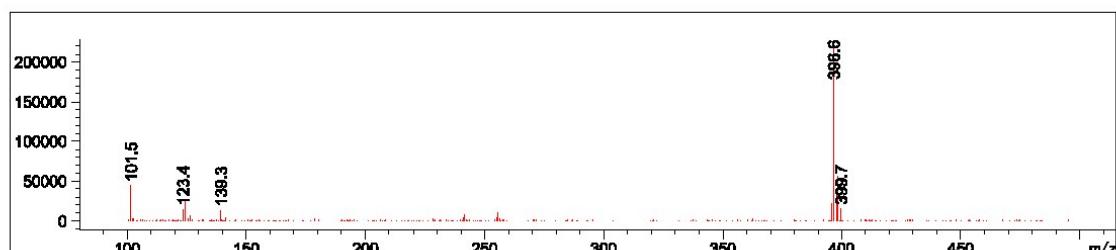
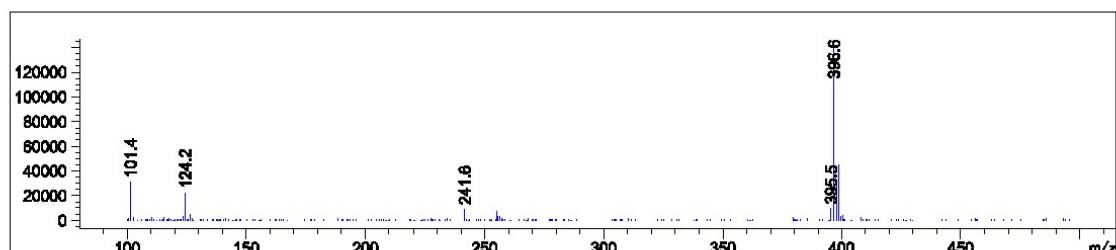
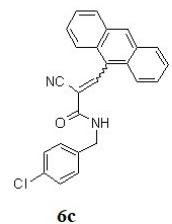


## LCMS of 6c

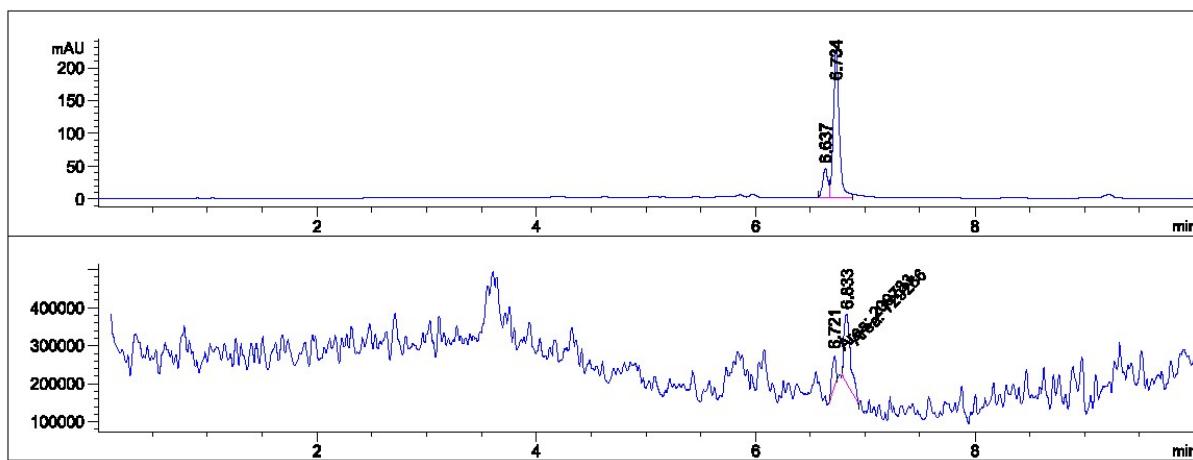


Retention Time (MS)	MS Area	Mol. Weight or Ion
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<b>6.410</b>	<b>908445</b>	<b>398.60 I</b> 397.60 I 396.60 I 124.20 I 101.45 I
<b>6.542</b>	<b>2372796</b>	<b>398.55 I</b> 397.65 I 396.60 I 124.40 I 101.50 I

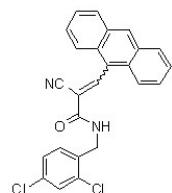


## LCMS of 6d

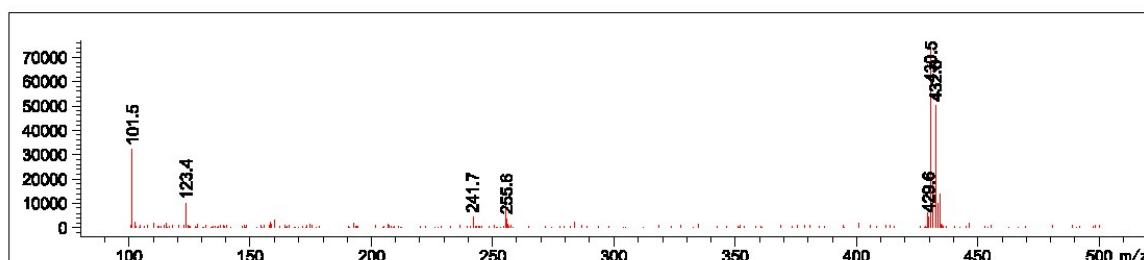
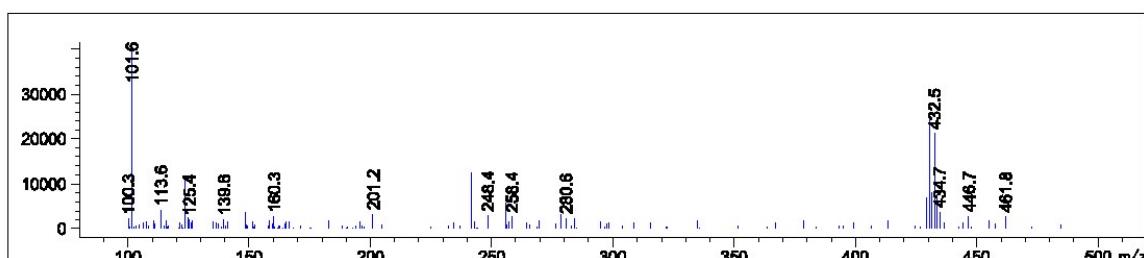


Retention Time (MS)	MS Area	Mol. Weight or Ion
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<b>6.721</b>	<b>209783</b>	<b>433.60 I</b> 432.50 I 431.45 I 430.55 I 429.40 I 255.60 I 241.65 I 123.60 I 113.60 I 101.55 I
<b>6.833</b>	<b>729286</b>	<b>434.35 I</b> 433.60 I 432.65 I 431.50 I 430.55 I 255.35 I 123.45 I 101.50 I



**6d**

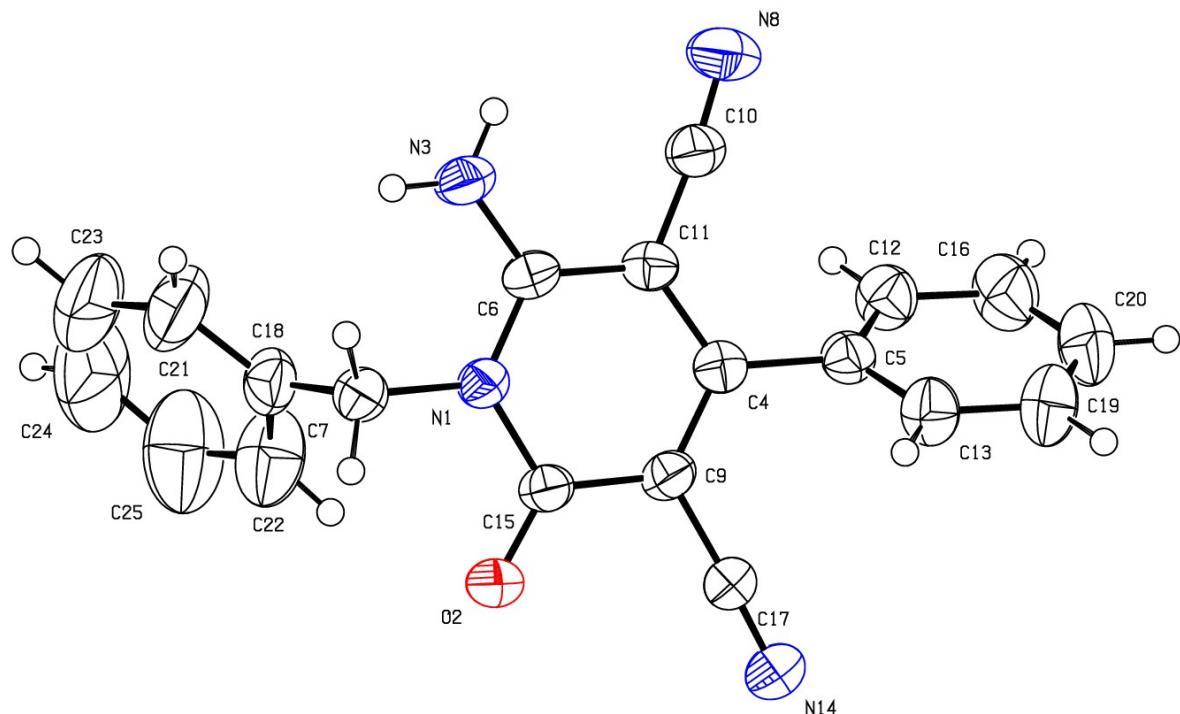


## X-ray crystallography

### Single crystal X-ray diffraction experimental

The crystal blocks of **5a** and **5i** were grown by the slow evaporation method using ethyl acetate as solvent. Single-crystal structural data of all of the compounds, **5a** and **5i**, were collected on a Bruker CMOS-based D8 Venture diffractometer equipped with a microfocus source with graphite-monochromated Mo K $\alpha$  radiation ( $\lambda = 0.71073 \text{ \AA}$ ). Using Olex2,<sup>1</sup> the structures were solved with the olex2.solve<sup>2</sup> structure solution program using Charge Flipping and refined with the olex2.refine<sup>2</sup> refinement package using Gauss-Newton minimisation. The crystallographic information files (CIFs) are also deposited as CCDC 19222235 and 1922233 for compounds **5a** and **5i**, respectively, with the Cambridge Crystallographic Data Centre. These can be obtained free of charge from the Cambridge Crystallographic Data Centre via [www.ccdc.cam.ac.uk/data\\_request/cif](http://www.ccdc.cam.ac.uk/data_request/cif).

### 6-amino-1-benzyl-2-oxo-4-phenyl-1,2-dihdropyridine-3,5-dicarbonitrile (**5a**)

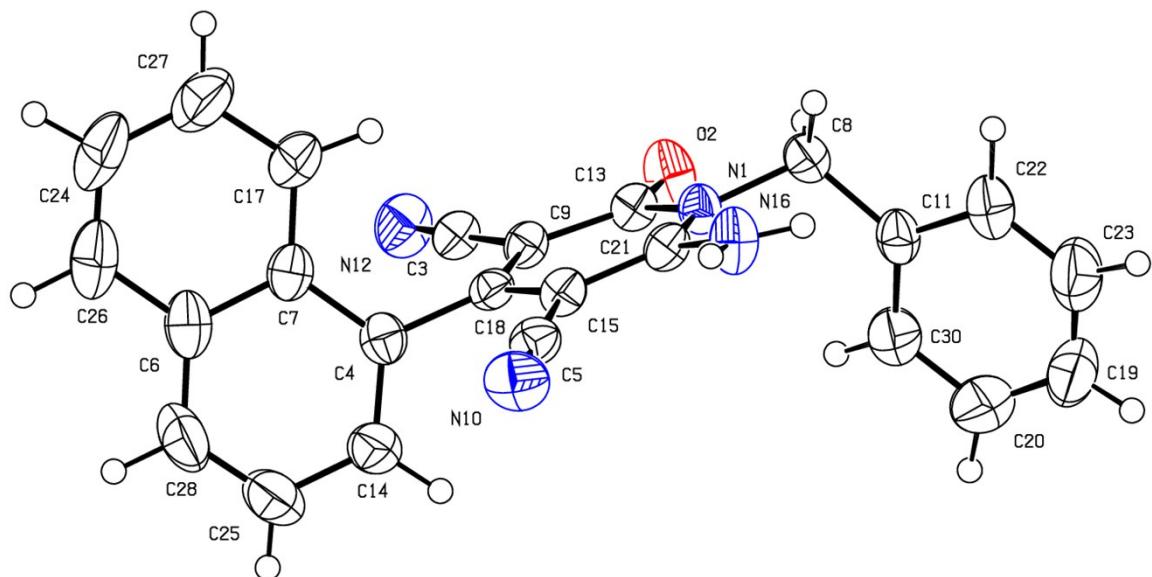


**Figure S1** ORTEP view of compound **5a** with displacement ellipsoids drawn at 50%. H atoms are shown as small spheres of arbitrary radii.

**Table S1 Crystal data and structure refinement for 5a.**

<b>Identification code</b>	5a
<b>Empirical formula</b>	C <sub>20</sub> H <sub>14</sub> N <sub>4</sub> O
<b>Formula weight</b>	326.36
<b>Temperature/K</b>	273.15
<b>Crystal system</b>	monoclinic
<b>Space group</b>	P2 <sub>1</sub> /c
<b>a/Å</b>	9.5352(4)
<b>b/Å</b>	16.8113(8)
<b>c/Å</b>	11.9674(5)
<b>α/°</b>	90
<b>β/°</b>	112.7581(13)
<b>γ/°</b>	90
<b>Volume/Å<sup>3</sup></b>	1769.01(14)
<b>Z</b>	4
<b>ρ<sub>calc</sub>g/cm<sup>3</sup></b>	1.2253
<b>μ/mm<sup>-1</sup></b>	0.079
<b>F(000)</b>	680.3
<b>Crystal size/mm<sup>3</sup></b>	9.5352 × 16.8113 × 11.9674
<b>Radiation</b>	Mo Kα ( $\lambda = 0.71073$ )
<b>2Θ range for data collection/°</b>	4.42 to 52.86
<b>Index ranges</b>	-11 ≤ h ≤ 11, -21 ≤ k ≤ 21, -14 ≤ l ≤ 14
<b>Reflections collected</b>	40629
<b>Independent reflections</b>	3615 [ $R_{\text{int}} = 0.0321$ , $R_{\text{sigma}} = 0.0146$ ]
<b>Data/restraints/parameters</b>	3615/0/226
<b>Goodness-of-fit on F<sup>2</sup></b>	2.093
<b>Final R indexes [I&gt;=2σ (I)]</b>	$R_1 = 0.0700$ , $wR_2 = 0.2503$
<b>Final R indexes [all data]</b>	$R_1 = 0.0767$ , $wR_2 = 0.2551$
<b>Largest diff. peak/hole / e Å<sup>-3</sup></b>	0.53/-0.39

**6-amino-1-benzyl-4-(naphthalen-1-yl)-2-oxo-1,2-dihdropyridine-3,5-dicarbonitrile (5i)**



**Figure S2** ORTEP view of compound **5i** with displacement ellipsoids drawn at 50%. H atoms are shown as small spheres of arbitrary radii.

**Table S2 Crystal data and structure refinement for 5i.**

<b>Identification code</b>	5i
<b>Empirical formula</b>	C <sub>24</sub> H <sub>16</sub> N <sub>4</sub> O
<b>Formula weight</b>	376.42
<b>Temperature/K</b>	273.15
<b>Crystal system</b>	orthorhombic
<b>Space group</b>	Aea2
<b>a/Å</b>	14.5818(13)
<b>b/Å</b>	28.597(2)
<b>c/Å</b>	9.1804(7)
<b>α/°</b>	90
<b>β/°</b>	90
<b>γ/°</b>	90
<b>Volume/Å<sup>3</sup></b>	3828.2(6)
<b>Z</b>	8
<b>ρ<sub>calc</sub>g/cm<sup>3</sup></b>	1.3061
<b>μ/mm<sup>-1</sup></b>	0.083
<b>F(000)</b>	1568.6

<b>Crystal size/mm<sup>3</sup></b>	9.1804 × 28.597 × 14.5818
<b>Radiation</b>	Mo K $\alpha$ ( $\lambda = 0.71073$ )
<b>2<math>\Theta</math> range for data collection/°</b>	5.44 to 53.32
<b>Index ranges</b>	-18 ≤ h ≤ 18, -36 ≤ k ≤ 36, -11 ≤ l ≤ 11
<b>Reflections collected</b>	42627
<b>Independent reflections</b>	4033 [ $R_{\text{int}} = 0.0705$ , $R_{\text{sigma}} = 0.0338$ ]
<b>Data/restraints/parameters</b>	4033/1/117
<b>Goodness-of-fit on <math>F^2</math></b>	2.703
<b>Final R indexes [I&gt;=2σ (I)]</b>	$R_1 = 0.0998$ , $wR_2 = 0.3099$
<b>Final R indexes [all data]</b>	$R_1 = 0.1024$ , $wR_2 = 0.3136$
<b>Largest diff. peak/hole / e Å<sup>-3</sup></b>	0.70/-0.48
<b>Flack parameter</b>	-0(3)

## REFERENCE:

1. Dolomanov, O. V.; Bourhis, L. J.; Gildea, R. J.; Howard, J. A. K.; Puschmann, H., OLEX2: a complete structure solution, refinement and analysis program. *Journal of Applied Crystallography* **2009**, *42* (2), 339-341.
2. Bourhis, L. J.; Dolomanov, O. V.; Gildea, R. J.; Howard, J. A. K.; Puschmann, H., The anatomy of a comprehensive constrained, restrained refinement program for the modern computing environment - Olex2 dissected. *Acta Crystallographica Section A* **2015**, *71* (1), 59-75.