

Electronic Supplementary Information (ESI)

Rod-like anhydrous V₂O₅ assembled by tiny nanosheets as a high-performance cathode material for aqueous zinc-ion batteries

Weijun Zhou,^a Jizhang Chen,*^a Minfeng Chen,^a Xinwu Xu,^a Qinghua Tian,^b Junling Xu*^c and Ching-Ping Wong^{c,d}

^a College of Materials Science and Engineering, Nanjing Forestry University, Nanjing 210037, China

^b Department of Chemistry, School of Sciences, Zhejiang Sci-Tech University, Hangzhou 310018, China

^c Department of Electronic Engineering, The Chinese University of Hong Kong, NT, Hong Kong, China

^d School of Materials Science and Engineering, Georgia Institute of Technology, Atlanta, United States

E-mail addresses: jizhang.chen@hotmail.com (J. Chen) and junlingxu@outlook.com (J. Xu)

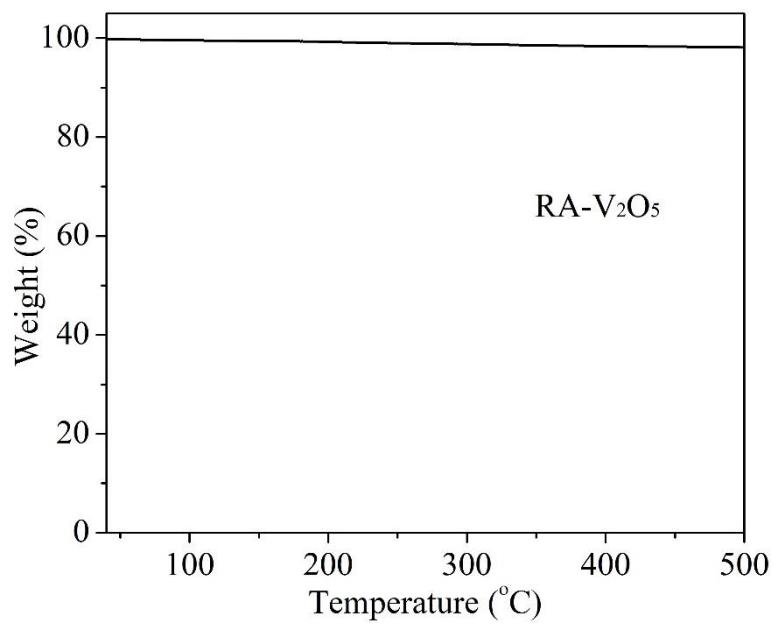


Fig. S1 TGA curve of RA-V₂O₅, measured in air at a heating rate of 10 °C min⁻¹.

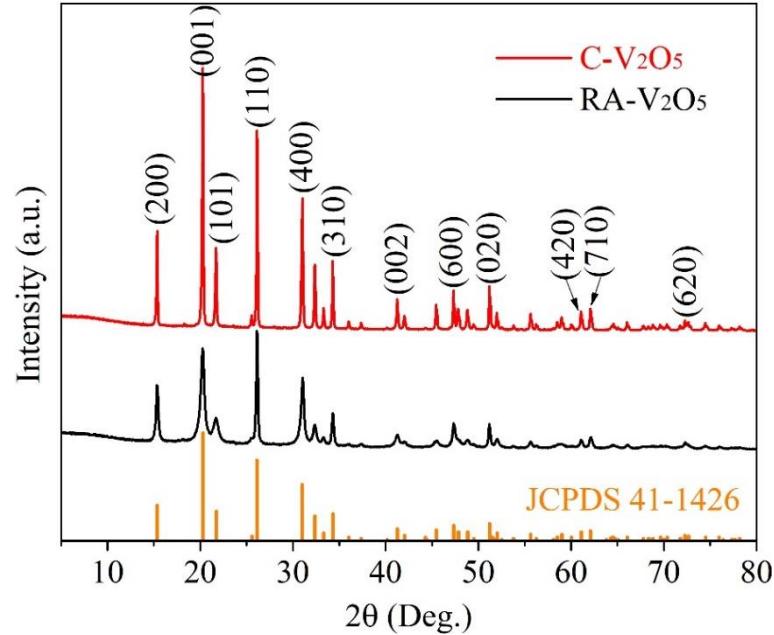


Fig. S2 XRD pattern of C-V₂O₅, in comparison with that of RA-V₂O₅.

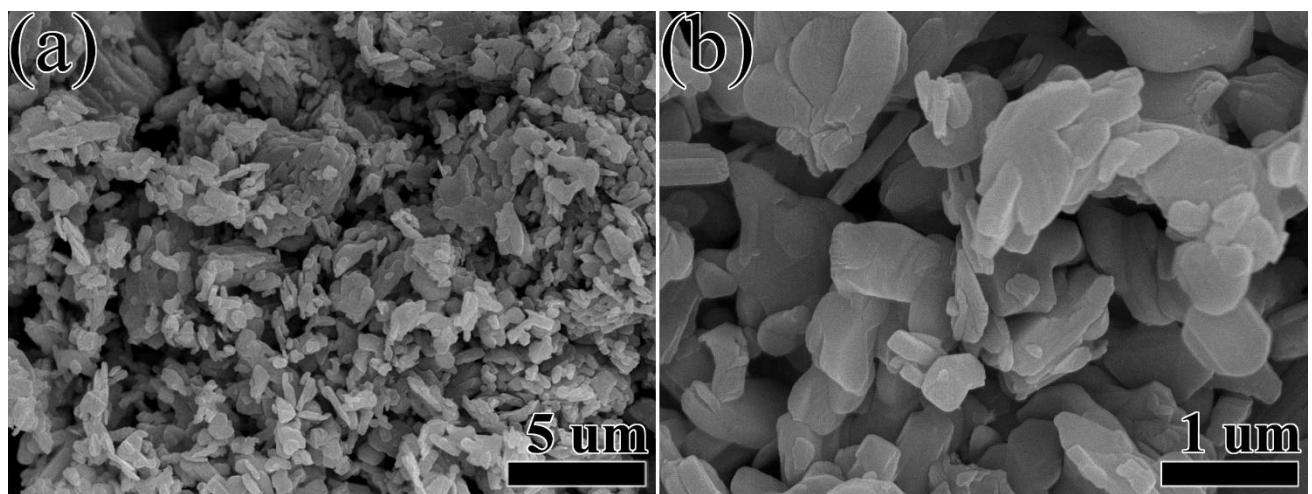


Fig. S3 (a, b) SEM images C-V₂O₅.

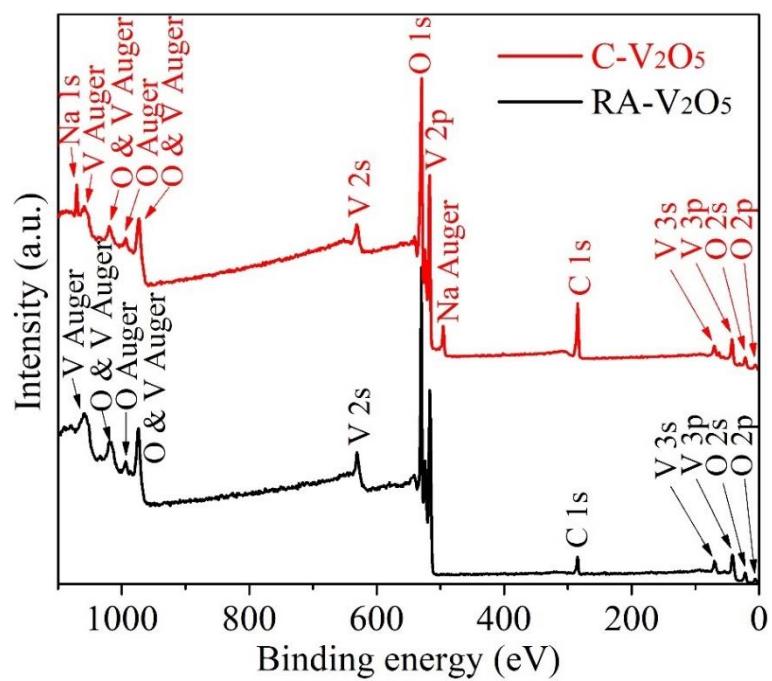


Fig. S4 XPS spectra of RA-V₂O₅ and C-V₂O₅.

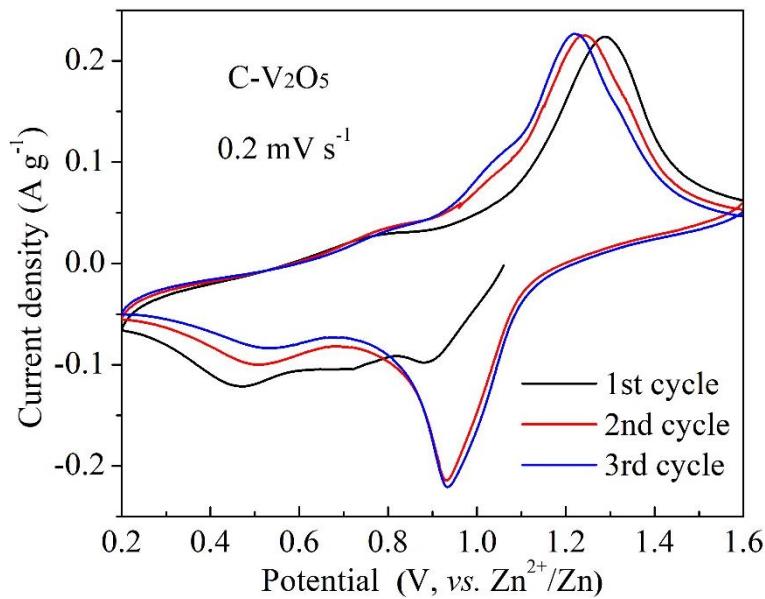


Fig. S5 CV curves of C-V₂O₅ in initial three cycles at 0.2 mV s^{-1} .

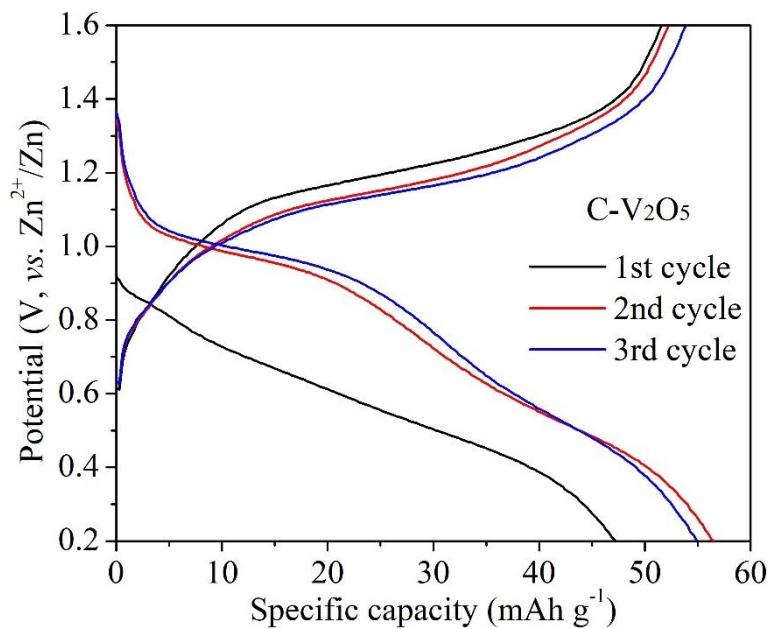


Fig. S6 GCD curves of C-V₂O₅ in initial three cycles at 0.1 A g^{-1} .

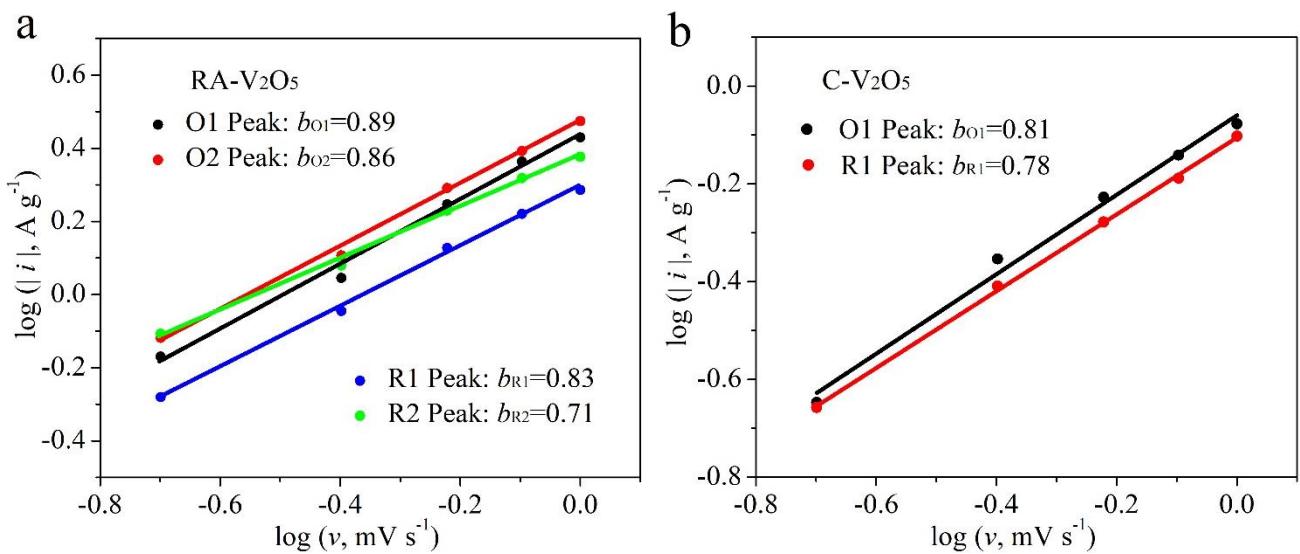


Fig. S7 Log (i) versus log (v) plots of different redox peaks of (a) RA-V₂O₅ and (b) C-V₂O₅ under CV measurements.

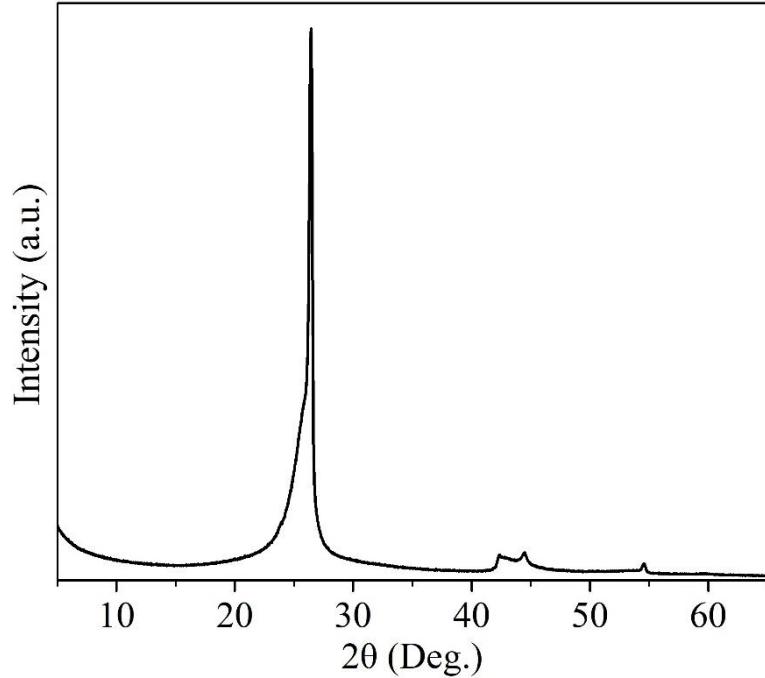


Fig. S8 XRD pattern of neat CNT power.

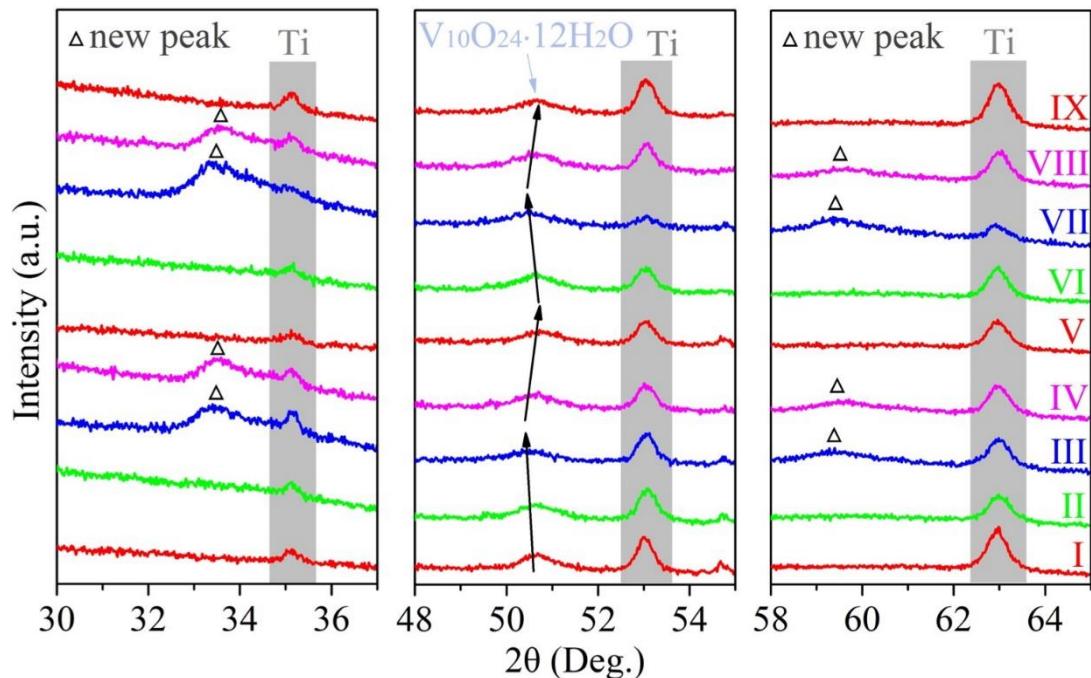


Fig. S9 The magnified ex situ XRD patterns of RA-V₂O₅ at nine different charge/discharge states at 21th and 22th cycles.

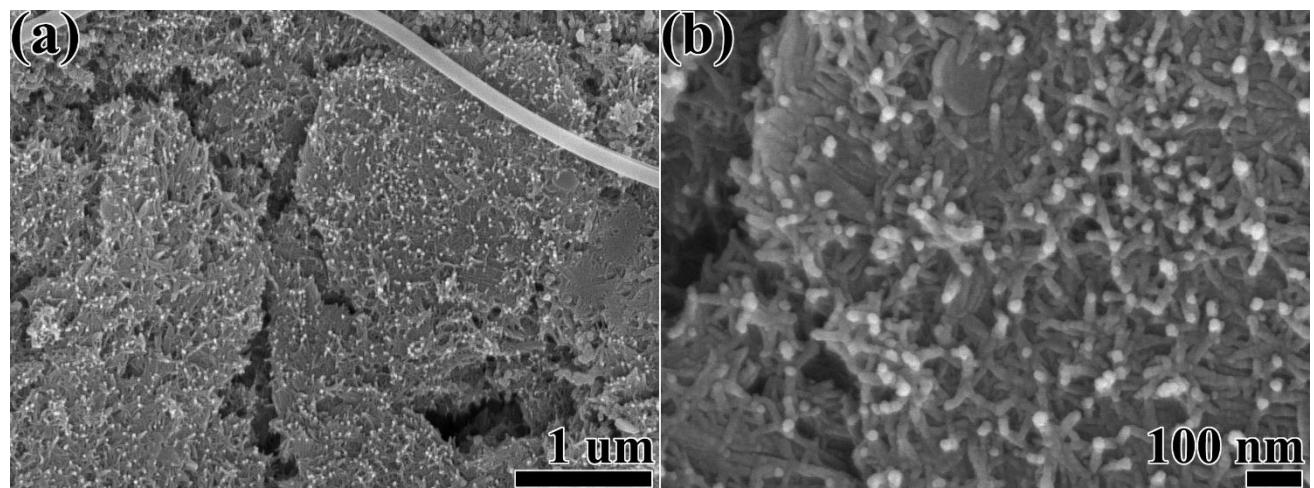


Fig. S10 SEM images of the RA-V₂O₅ electrode surface after 2000 cycles at 2 A g⁻¹.

Table S1 The capacities of RA-V₂O₅ in comparison with that of state-of-the-art vanadium-based cathode materials for AZIBs.

Cathode material	Capacity	Reference
RA-V ₂ O ₅	449.8 mA h g ⁻¹ at 0.1 A g ⁻¹	This report
	314.3 mA h g ⁻¹ at 2 A g ⁻¹	
	186.8 mA h g ⁻¹ at 5 A g ⁻¹	
Mg _{0.34} V ₂ O ₅ ·0.84H ₂ O	353 mA h g ⁻¹ at 0.1 A g ⁻¹	1
	81 mA h g ⁻¹ at 5 A g ⁻¹	
Ag _{0.4} V ₂ O ₅	340 mA h g ⁻¹ at 0.1 A g ⁻¹	2
	185 mA h g ⁻¹ at 2 A g ⁻¹	
Porous V ₂ O ₅ nanofibers	319 mA h g ⁻¹ at 0.02A g ⁻¹	3
	104 mA h g ⁻¹ at 3 A g ⁻¹	
V ₂ O ₅ nanosheets	224 mA h g ⁻¹ at 0.1 A g ⁻¹	4
	100 mA h g ⁻¹ at 2 A g ⁻¹	
V ₂ O ₅ nanospheres	188.7 mA h g ⁻¹ at 0.5 A g ⁻¹	5
	138.3 mA h g ⁻¹ at 5 A g ⁻¹	
V ₂ O ₅ hollow spheres	280 mA h g ⁻¹ at 0.2 A g ⁻¹	6
	147 mA h g ⁻¹ at 5 A g ⁻¹	
VO ₂	283 mA h g ⁻¹ at 0.1 A g ⁻¹	7
	72 mA h g ⁻¹ at 5 A g ⁻¹	
VO ₂	274 mA h g ⁻¹ at 0.1 A g ⁻¹	8
	170 mA h g ⁻¹ at 5 A g ⁻¹	
V ₁₀ O ₂₄ ·12H ₂ O	164.5 mA h g ⁻¹ at 0.2 A g ⁻¹	9
	90.4 mA h g ⁻¹ at 5 A g ⁻¹	
VS ₂	190.3 mA h g ⁻¹ at 0.05 A g ⁻¹	10
	115.5 mA h g ⁻¹ at 2 A g ⁻¹	
LiV ₃ O ₈	230 mA h g ⁻¹ at 0.033 A g ⁻¹	11
	29 mA h g ⁻¹ at 1.666 A g ⁻¹	
NaV ₃ O ₈ ·1.5H ₂ O	375 mA h g ⁻¹ at 0.1 A g ⁻¹	12
	165 mA h g ⁻¹ at 4 A g ⁻¹	
NaV ₆ O ₁₅ nanorods	427 mA h g ⁻¹ at 0.05 A g ⁻¹	13
	195 mA h g ⁻¹ at 1.6 A g ⁻¹	
Zn ₂ V ₂ O ₇	203.4 mA h g ⁻¹ at 0.3 A g ⁻¹	14
	155 mA h g ⁻¹ at 4 A g ⁻¹	
Zn ₂ (OH)VO ₄	204 mA h g ⁻¹ at 0.1 A g ⁻¹	15
	160 mA h g ⁻¹ at 2 A g ⁻¹	
Zn ₃ V ₂ O ₇ (OH) ₂ ·2H ₂ O	213 mA h g ⁻¹ at 0.05 A g ⁻¹	16
	76 mA h g ⁻¹ at 3 A g ⁻¹	
Fe ₅ V ₁₅ O ₃₉ (OH) ₉ ·9H ₂ O	385 mA h g ⁻¹ at 0.1 A g ⁻¹	17
	105 mA h g ⁻¹ at 5 A g ⁻¹	
VOPO ₄	139 mA h g ⁻¹ at 0.05 A g ⁻¹	18
	50 mA h g ⁻¹ at 5 A g ⁻¹	

References

1. F. W. Ming, H. F. Liang, Y. J. Lei, S. Kandambeth, M. Eddaoudi and H. N. Alshareef, *ACS Energy Lett.*, 2018, **3**, 2602-2609.
2. L. Shan, Y. Yang, W. Zhang, H. Chen, G. Fang, J. Zhou and S. Liang, *Energy Storage Mater.*, 2018, **18**, 10-14.
3. X. Chen, L. Wang, H. Li, F. Cheng and J. Chen, *J. Energy Chem.*, 2019, **38**, 20-25.
4. J. Zhou, L. T. Shan, Z. X. Wu, X. Guo, G. Z. Fang and S. Q. Liang, *Chem. Commun.*, 2018, **54**, 4457-4460.
5. F. Liu, Z. Chen, G. Fang, Z. Wang, Y. Cai, B. Tang, J. Zhou and S. Liang, *Nano-Micro Lett.*, 2019, **11**, 25.
6. H. Qin, L. Chen, L. Wang, X. Chen and Z. Yang, *Electrochim. Acta*, 2019, **306**, 307-316.
7. L. N. Chen, Y. S. Ruan, G. B. Zhang, Q. L. Wei, Y. L. Jiang, T. F. Xiong, P. He, W. Yang, M. Y. Yan, Q. Y. An and L. Q. Mai, *Chem. Mater.*, 2019, **31**, 699-706.
8. T. Y. Wei, Q. Li, G. Z. Yang and C. X. Wang, *J. Mater. Chem. A*, 2018, **6**, 8006-8012.
9. T. Y. Wei, Q. Li, G. Z. Yang and C. X. Wang, *Electrochim. Acta*, 2018, **287**, 60-67.
10. P. He, M. Y. Yan, G. B. Zhang, R. M. Sun, L. N. Chen, Q. Y. An and L. Q. Mai, *Adv. Energy Mater.*, 2017, **7**, 1601920.
11. M. H. Alfaruqi, V. Mathew, J. Song, S. Kim, S. Islam, D. T. Pham, J. Jo, S. Kim, J. P. Baboo, Z. Xiu, K. S. Lee, Y. K. Sun and J. Kim, *Chem. Mater.*, 2017, **29**, 1684-1694.
12. F. Wan, L. L. Zhang, X. Dai, X. Y. Wang, Z. Q. Niu and J. Chen, *Nat. Commun.*, 2018, **9**, 1656.
13. S. Islam, M. H. Alfaruqi, B. Sambandam, D. Y. Putro, S. Kim, J. Jo, S. Kim, V. Mathew and J. Kim, *Chem. Commun.*, 2019, **55**, 3793-3796.
14. B. Sambandam, V. Soundharajan, S. Kim, M. H. Alfaruqi, J. Jo, S. Kim, V. Mathew, Y. K. Sun and J. Kim, *J. Mater. Chem. A*, 2018, **6**, 3850-3856.
15. D. L. Chao, C. Zhu, M. Song, P. Liang, X. Zhang, N. H. Tiep, H. F. Zhao, J. Wang, R. M. Wang, H. Zhang and H. J. Fan, *Adv. Mater.*, 2018, **30**, 1803181.
16. C. Xia, J. Guo, Y. J. Lei, H. F. Liang, C. Zhao and H. N. Alshareef, *Adv. Mater.*, 2018, **30**, 1705580.
17. Z. Peng, Q. L. Wei, S. S. Tan, P. He, W. Luo, Q. Y. An and L. Q. Mai, *Chem. Commun.*, 2018, **54**, 4041-4044.
18. F. Wan, Y. Zhang, L. Zhang, D. Liu, C. Wang, L. Song, Z. Niu and J. Chen, *Angew. Chem. Int. Edit.*, 2019, **58**, 7062-7067.