Pd nanoparticles on green support as dip-catalyst: A facile transfer hydrogenation of olefins and N-heteroarens in water

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Figure S1: FT-IR spectra of a) pure green support (GS) b) Pd@GS and c) Reused Pd@GS after 32 cycles



31.292

21,490,275

(a)

Figure S2: a) GC spectra of catalytic hydrogenations of trans- methylsitlbene b) Mass spectra of the peak at R_t =25.670 (hydrogenated product) and c) Mass spectra of the peak at R_t =31.295 (starting substrate).



Figure S3: a) GC spectra of pure methyl- acetamidoacrylate b) GC spectra of catalytic hydrogenations of methyl- acetamidoacrylate c) Mass spectra of hydrogenated product.





Figure S4: a) GC spectra of catalytic hydrogenations of cycloheptene b) Mass spectra of peak at $R_t = 2.360$ (starting substrate) and c) Mass spectra of peak at $R_t = 2.480$ (hydrogenated product).



Figure S5: a) GC spectra of catalytic hydrogenations of *p*-vinyl anisole b) Mass spectra of peak at R_t =4.715 (hydrogenated product) and c) Mass spectra of peak at R_t =5.473 (starting substrate).











Figure S6: a) GC spectra of catalytic hydrogenations of *o*-bromostyrene b) Mass spectra of peak at R_t =3.435 (de-halogenated but hydrogenated product) and c) Mass spectra of peak at R_t =3.9853 (de-halogenated product) d) Mass spectra of peak at R_t =9.875 (hydrogenated product) e) Mass spectra of peak at R_t =10.780 (unreacted substrate).



Figure S7: a) GC spectra of catalytic hydrogenations of *o*-bromostyrene in THF at 50 °C.



Figure S8: a) GC spectra of catalytic hydrogenations of *o*-bromostyrene in THF at 40 °C.



Figure S9: a) GC spectra of Pd@GS catalyzed hydrogenation of *m*-nitrostyrene at 45 °C in THF with 2.1 equiv. THDB b) Mass spectra of peak at $R_t = 11.506$ (where both functional group reduced) c) Mass spectra of peak at $R_t = 14.840$ (Selectively olefin group hydrogenated) d) *m*-nitrostyrene (unreacted).



Figure S10: (a) SEM image and (b) EDS spectrum of Re-used Pd@GS.