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Supporting Information

Hot Nitric Acid Diffuse in Fluoroelastomer Composite and its Degradation

Hailan Kang,^{1,2,*}Long Chen,¹ Hongyang Du,¹ Haoyu Wang, ¹ Donghan Li,^{1,2}

Qinghong Fang^{1,2,*}

¹ College of Materials Science and Engineering, Shenyang University of Chemical Technology, Shenyang, 110142, China

² Liaoning Provincial Key Laboratory of Rubber & Elastomer, Shenyang University of Chemical Technology, Shenyang 110142, China 1. NMR

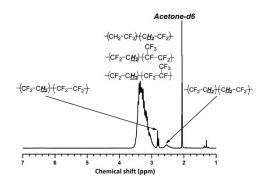


Figure S1 ¹H-NMR spectrum of fluoroelastomer in acetone-d6.

The ¹H-NMR spectrum of fluoroelastomer shows the presence of the multiplets at 3.55~2.70 ppm assigned to methylene groups of $-CF_2C\underline{H}_2$ - CF_2CH_2 -, $-CF_2C\underline{H}_2$ - CF_2-CF_2 - CF_3)-, $-CF_2C\underline{H}_2$ - $CF(CF_3)CF_2$ - and $-CF_2C\underline{H}_2$ - CF_2CF_2 - sequence resulting from VDF-VDF, VDF-HFP and VDF-TFE additions, respectively. Actually, tail-to-tail VDF addition ($-CF_2C\underline{H}_2$ - $C\underline{H}_2CF_2$ -) is also identified by the presence of the peak centered at 2.53 ppm assigned to both methylene groups in a role (as an inversed addition).

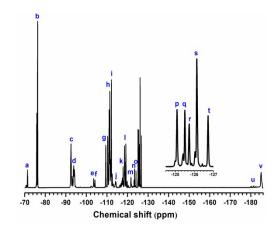


Figure S2¹⁹F-NMR spectrum of fluoroelastomer

As shown in Figure S2, the multiplets at $-70 \sim -76$ (a, b), $-117 \sim -120$ (l) and $-180 \sim -185$ ppm (u, v) assigned to $-CF_3$, $-CF_2-$ and -CF from HFP sequences, respectively;

the multiplets at -91~-117 ppm (c~k) assigned to $-CF_2$ - from VDF sequences; and the multiplets at -120~-126 ppm (m~t) assigned to $-CF_2$ - from TFE sequences.

Thus the main sequences and structure of fluoroelastomer can be showed as Figure S3.

 $\begin{array}{cccc} \mathsf{CF}_3 & \mathsf{CF}_3 & \mathsf{CF}_3 \\ \downarrow & \downarrow \\ \mathsf{CF}_2\mathsf{CH}_2\mathsf{CF}_2\mathsf{CF}_2\mathsf{CF}_2\mathsf{CH}_2\mathsf{CF}_2\mathsf{C$