Formation of toroidal Li₂O₂ in non-aqueous Li-O₂ batteries with Mo₂CT_x MXene/CNT composite

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CNT (m ² /g)	CNT/Mo ₂ CT _x (m ² /g)		
172.6	134.9		

Table S1. BET surface area of the CNT and CNT/Mo_2CT_x electrode



Figure S1. EDS mapping of the CNT/Mo_2CT_x electrode



Figure S2. Ex situ SEM images of the CNT electrode



Figure S3. (a) Discharge capacity (b) *Ex situ* SEM images of the CNT/Mo₂CT_x electrode

discharged under 1 M LiTFSI in TEGDME

Section 1. Vibrational Frequency Analysis

Phonon dispersion spectra were calculated using the PHONOPY code with 4 X 4 X 1 supercell. It provides the information for the dynamical stability of geometry optimized structures based on density functional perturbation theory.

There is the phonon dispersion spectrum of geometry optimized Mo_2C using PBE-D3 in Figure S1a. It indicates that there is no imaginary frequency in the high symmetry k-path. However, all models of $Mo_2C(OH)_2$ have some imaginary frequencies. These tendencies are also found in α -Mo₂C by B. I. Yakobson et al.² They claimed that the calculated phonon dispersion of the various models for α -Mo₂C^(OH)₂ have imaginary frequencies around the M points while α -Mo₂C don't have the imaginary frequencies in high symmetry k-path. So, α -Mo₂C^(OH)₂ is regarded as being structurally unstable. This would indicate that all the models in this paper are β -Mo₂C, but it has same tendency with α -Mo₂C. Although Mo₂C^(OH)₂ have some dynamical instability, the model 3 of Mo₂C^(OH)₂, which is the most stable model, was selected in order to calculate the interaction of O_2 adsorption. Because model 3 has only small imaginary frequencies around the Γ and M point as well as the lowest energy when comparing mode 1 and 2.



Figure S4. There are phonon dispersion spectrum calculated by the PHONOPY code with 4 X 4 X 1 supercell: (a)^{Mo_2C}, (b) the model 1 of $^{Mo_2C(OH)_2}$, (c) the model 2 of $^{Mo_2C(OH)_2}$, (d) the model 3 of $^{Mo_2C(OH)_2}$.

Section 2. Crystal structures

1. *Mo*₂*C*

data_O	pt_Mo2	C			
_symmetry_space_group_name_H-M 'P1'					
_symmetry_Int_Tables_number				1	
_symmetry_cell_setting			triclinic		
loop_					
_symm	netry_eq	uiv_pos_as_2	xyz		
x,y,z	Z				
_cell_l	ength_a			2.9649	
_cell_l	ength_b			2.9649	
_cell_l	ength_c			32.3383	
_cell_angle_alpha			90.0000		
_cell_angle_beta				90.0000	
_cell_angle_gamma			120.0000		
loop_					
_atom_site_label					
_atom_site_type_symbol					
_atom_site_fract_x					
atom	_site_fra	ct_y			
_atom_site_fract_z					
_atom_site_U_iso_or_equiv					
Mo1	Mo	0.66667	0.33333	0.46310	0.00000
Mo2	Mo	0.33333	0.66667	0.53690	0.00000
C3	С	0.00000	0.00000	0.50000	0.00000

2. Model 1 of $Mo_2C(OH)_2$

data_Opt_model1					
_symmetry_space_group_name_H-M 'P1'					
_symmetry_Int_Tables_number 1					
_symmetry_cell_setting				triclinic	
loop_					
_symm	etry_eq	uiv_pos_as_	xyz		
x,y,z					
_cell_le	ength_a			3.0901	
_cell_le	ength_b			3.0901	
_cell_le	ength_c			29.7716	
_cell_a	ngle_al	pha		90.0000	
_cell_a	ngle_be	eta		90.0000	
_cell_angle_gamma				120.0000	
loop_					
_atom_site_label					
_atom_site_type_symbol					
_atom_site_fract_x					
_atom_site_fract_y					
_atom_site_fract_z					
atom	site_U_	_iso_or_equiv	V		
Mo1	Mo	0.66667	0.33333	0.46155	0.00000
Mo2	Mo	0.33333	0.66667	0.53845	0.00000
H3	Н	0.33333	0.66667	0.63442	0.00000
O4	0	0.33333	0.66667	0.60213	0.00000
H5	Н	0.66667	0.33333	0.36558	0.00000
06	0	0.66667	0.33333	0.39787	0.00000
C7	С	0.00000	0.00000	0.50000	0.00000

3. Model 2 of $Mo_2C(OH)_2$

data_Opt_model2

_symm	etry_sp	ace_group_r	name_H-M	'P1'	
_symmetry_Int_Tables_number 1				1	
_symmetry_cell_setting			triclinic		
loop_					
_symm	etry_eq	uiv_pos_as_	xyz		
x,y,z					
_cell_le	ength_a			2.9726	
_cell_le	ength_b)		2.9726	
_cell_le	ength_c	;		32.1711	
_cell_a	ngle_al	pha		90.0000	
_cell_a	ngle_be	eta		90.0000	
_cell_a	ngle_ga	amma		120.0000	
loop_					
atom	site_lal	pel			
atom	site_ty	pe_symbol			
atom	site_fra	act_x			
_atom_site_fract_y					
_atom_site_fract_z					
atom	site_U	_iso_or_equi	V		
Mo1	Mo	0.66667	0.33333	0.46130	0.00000
Mo2	Mo	0.33333	0.66667	0.53870	0.00000
H3	Н	-0.00000	0.00000	0.38655	0.00000
O4	0	-0.00000	0.00000	0.58293	0.00000
Н5	Н	0.00000	-0.00000	0.61345	0.00000
06	0	0.00000	-0.00000	0.41707	0.00000
C7	С	0.00000	0.00000	0.50000	0.00000

4. Model 3 of $Mo_2C(OH)_2$

data_Opt_model3

_symm	etry_spa	ace_group_n	ame_H-M	'P1'	
_symmetry_Int_Tables_number				1	
_symmetry_cell_setting			triclinic		
loop_					
_symm	etry_eq	uiv_pos_as_	xyz		
x,y,z					
_cell_le	ength_a			3.2688	
_cell_le	ength_b			3.2688	
_cell_le	ength_c			26.6047	
_cell_a	ngle_alı	pha		90.0000	
_cell_a	ngle_be	eta		90.0000	
_cell_a	ngle_ga	ımma		120.0000	
loop_					
atom	site_lab	oel			
atom	site_typ	e_symbol			
atom	site_fra	ct_x			
_atom_site_fract_y					
_atom_site_fract_z					
atom	site_U_	iso_or_equiv	7		
Mo1	Mo	0.66667	0.33333	0.46424	0.00000
Mo2	Mo	0.33333	0.66667	0.53576	0.00000
03	0	0.66667	0.33333	0.58526	0.00000
04	0	0.33333	0.66667	0.41474	0.00000
H5	Н	0.33333	0.66667	0.37811	0.00000
H6	Н	0.66667	0.33333	0.62189	0.00000
C7	С	0.00000	0.00000	0.50000	0.00000