

Electronic Supplementary Information

Elucidating π - π Interaction-Induced Extension Effect in Sandwich Phthalocyaninato Compounds

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Caption of Content

- 1. Fig S1.** ^1H NMR and ^1H - ^1H COSY spectra of **3** in CDCl_3 .
- 2. Fig S2.** ^1H NMR and ^1H - ^1H COSY spectra of **4** in CDCl_3 .
- 3. Fig S3.** Cyclic voltammetry of quintuple-decker complexes $\{[(\text{Pc}^*)\text{Sm}][(\text{Pc}^*)\text{Cd}_n(\text{Pc}^*)_n][\text{Sm}(\text{Pc}^*)]\}$ ($n = 0-3$) (**3-6**) in CH_2Cl_2 containing 0.1 M $[\text{NBu}_4][\text{ClO}_4]$ at scan rate of 40 mV/s.
- 4. Fig S4.** The frontier molecular orbital energies for $\{[(\text{Pc})\text{La}][(\text{Pc})\text{Cd}_n(\text{Pc})_n][\text{La}(\text{Pc})]\}$ ($n = 0-2$) (**a-c**).
- 5. Fig S5.** The frontier molecular orbital map for $(\text{Pc})\text{La}(\text{Pc})\text{La}(\text{Pc})$.
- 6. Fig S6.** The frontier molecular orbital map for $(\text{Pc})\text{La}(\text{Pc})\text{Cd}(\text{Pc})\text{La}(\text{Pc})$.
- 7. Fig S7.** The frontier molecular orbital map for $(\text{Pc})\text{La}(\text{Pc})\text{Cd}(\text{Pc})\text{Cd}(\text{Pc})\text{La}(\text{Pc})$.
- 8. Table S1.** Half-wave redox potentials of the compounds **3-6** (V vs SCE) in CH_2Cl_2 containing 0.1 M TBAP.

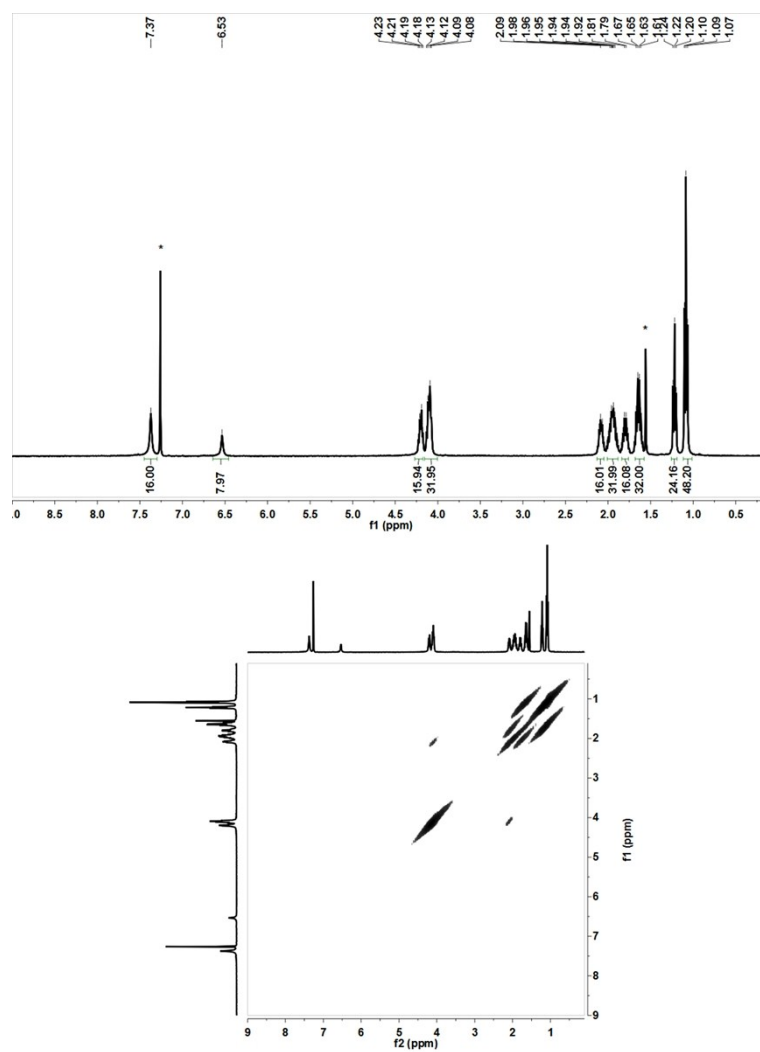


Fig S1. ¹H NMR and ¹H-¹H COSY spectra of **3** in CDCl₃

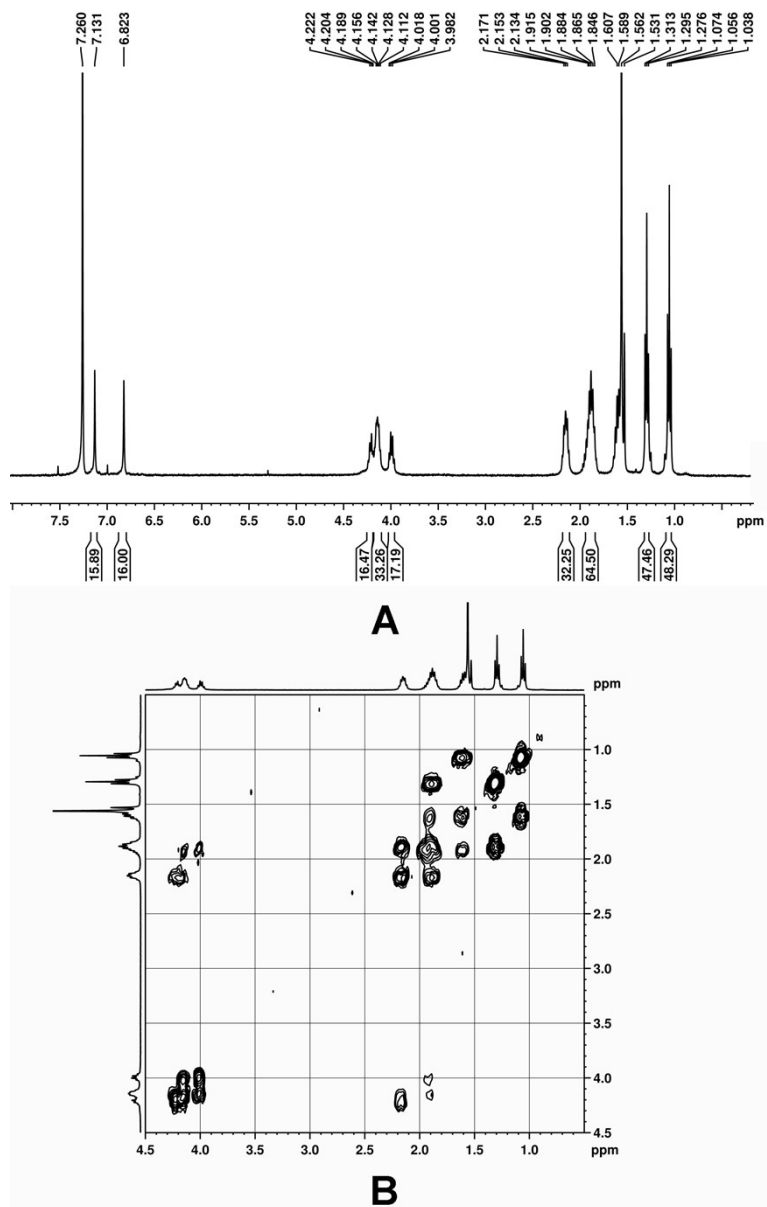


Fig S2. ^1H NMR and ^1H - ^1H COSY spectra of **4** in CDCl_3 .

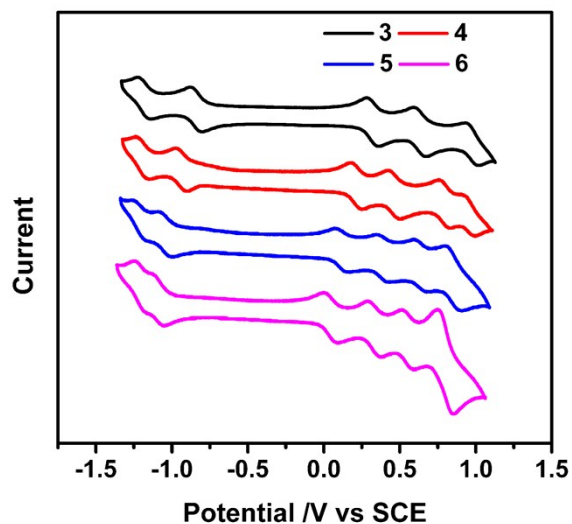


Fig S3. Cyclic voltammetry of quintuple-decker complexes $\{[(\text{Pc}^*)\text{Sm}][(\text{Pc}^*)\text{Cd}_n(\text{Pc}^*)_n][\text{Sm}(\text{Pc}^*)]\}$ ($n = 0-3$) (**3-6**) in CH_2Cl_2 containing 0.1 M $[\text{NBu}_4][\text{ClO}_4]$ at the scan rate of 40 mV/s.

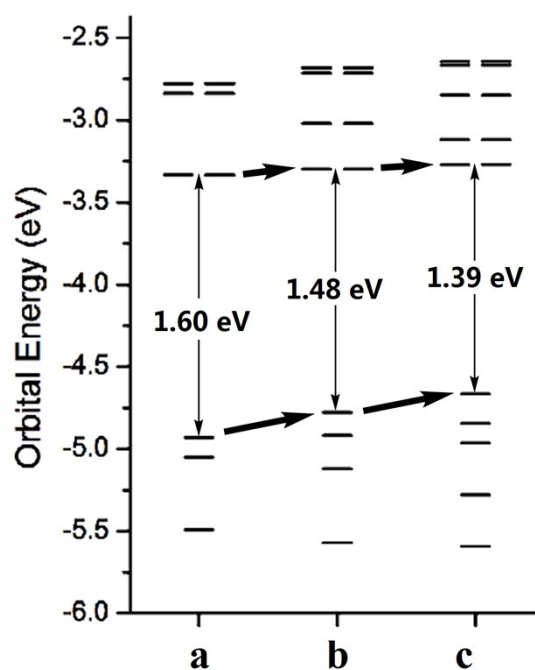


Fig S4. The frontier molecular orbital energies for $\{[(\text{Pc})\text{La}][(\text{Pc})\text{Cd}_n(\text{Pc})_n][\text{La}(\text{Pc})]\}$ ($n = 0-2$) (**a-c**).

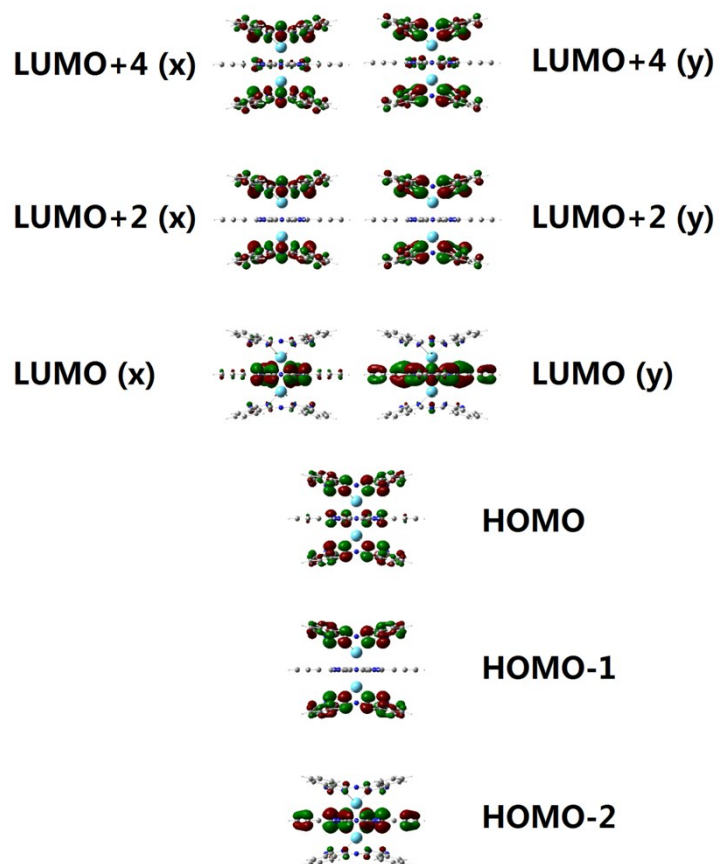


Fig S5. Frontier molecular orbital map for (Pc)La(Pc)La(Pc).

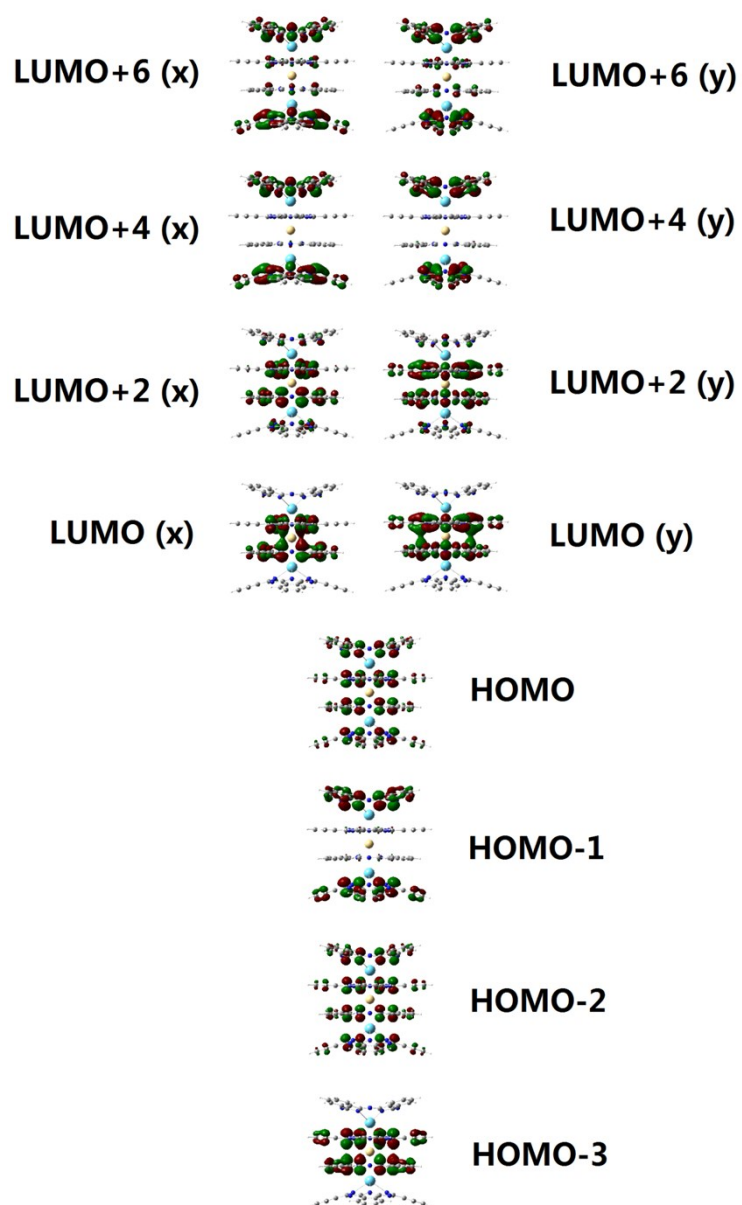


Fig S6. Frontier molecular orbital map for (Pc)La(Pc)Cd(Pc)La(Pc).

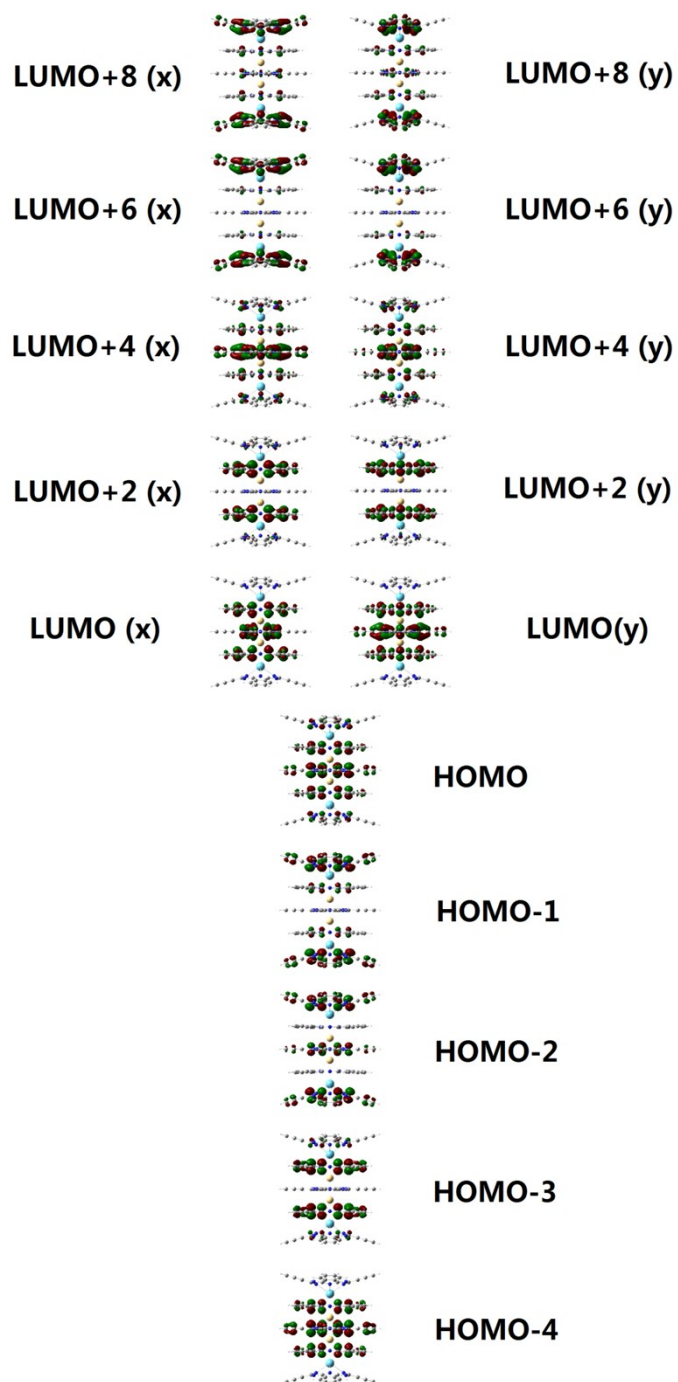


Fig S7. Frontier molecular orbital map for (Pc)La(Pc)Cd(Pc)Cd(Pc)La(Pc).

Table S1. Half-wave redox potentials of the compounds **3-6**.^a

Compound	Oxd ₄	Oxd ₃	Oxd ₂	Oxd ₁	Red ₁	Red ₂	ΔE ^o _{1/2}
3	--	0.98	0.63	0.33	-0.85	-1.18	1.18
4	0.94	0.79	0.47	0.22	-0.95	-1.20	1.17
5	0.85	0.64	0.39	0.12	-1.05	-1.22	1.17
6	0.81	0.55	0.34	0.06	-1.09	-1.22	1.15

^a Recorded with [Bu₄N][ClO₄] as electrolyte in CH₂Cl₂ (0.1 mol dm⁻³) at ambient temperature. Potentials were obtained by cyclic voltammetry with a scan rate of 40 mV s⁻¹, and are expressed as half-wave potentials ($E_{1/2}$) in V relative to SCE unless otherwise stated.