# ELECTRONIC SUPPORTING INFORMATION: The Role of Urea on the Solubility of Cellulose in Aqueous Quaternary Ammonium Hydroxide

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#### 1 Two Phase Thermodynamic Method

The two-phase thermodynamic method (2PT)[1, 2] can be used to estimate entropy for MD simulations. This has been implemented successfully to estimate standard molar entropies and heat capacities of common liquid-phase solvents.[3] The following gives further details on how this method was implemented for this study. The velocity of each atom has translational, rotational, and vibrational modes:

$$v_i^{tot} = v_i^{trn} + v_i^{rot} + v_i^{vib} \tag{1}$$

Classical mechanics can be used to easily obtain  $v_i^{trn}$  and  $v_i^{rot}$ . The remainder,  $v_i^{vib}$ , can be found by subtracting the translational and rotational components from the total velocity. Each atom has a velocity autocorrelation function for each mode k:

$$c_i^k(t) = \lim_{t \to \infty} \frac{1}{2\tau} \int_{-\tau}^{\tau} v_i^k(t'+t) v_i^k(t) dt'$$
(2)

with correlation time t. The mass-weighted autocorrelation function is obtained for each mode. The DoS is found by applying a Fourier transform  $\mathcal{F}$  and weighting:

$$f_{tot}^k(\nu) = \frac{2}{k_B T} \mathcal{F}\left[\sum_{i=1}^N m_i c_i^k(t)\right]$$
(3)

with Boltzmann's constant  $k_B$ , temperature T, number of total atoms N, and mass of atom  $i m_i$ . The fluidicity factor  $g_k$  for each mode is found:

$$\delta_k^{-4.5} g_k^{7.5} - 3\delta_k^{-3} g_k^5 - \frac{1}{2} \delta_k^{-1.5} g_k^{2.5} + g_k - 1 = 0$$
(4)

$$\delta_k = \frac{2f_{tot}^k(0)}{9N} \left(\frac{\pi k_B T}{m}\right)^{1/2} \left(\frac{N}{V}\right)^{1/3} \left(\frac{6}{\pi}\right)^{2/3} \tag{5}$$

with m total mass and V partial volume of component. The gas and solid components of the DoS are as follows:

$$f_g^k(\nu) = \frac{f_{tot}^k(0)}{\left(1 + \frac{f_{tot}^k(0)\pi\nu}{6g_kN}\right)^{1/2}}$$
(6)

$$f_{s}^{k}(\nu) = f_{tot}^{k}(\nu) - f_{g}^{k}(\nu)$$
(7)

The gas component has translational mode weighting factor:

$$\hat{S}_{g}^{trans} = \frac{1}{3k_{B}} \left[ \frac{5}{2} + \ln \left[ \left( \frac{2\pi m k_{B} T}{h^{2}} \right)^{3/2} \frac{V c_{1}}{N g_{trans}} \right] + c_{2} \right]$$
(8)

$$c_1 = \frac{1+y+y^2-y^3}{(1-y)^3} \tag{9}$$

$$c_2 = \frac{y(3y-4)}{(1-y)^2} \tag{10}$$

$$y = g_{trans}^{2.5} \delta_{trans}^{2.5} \tag{11}$$

with  $c_1$  and  $c_2$  compressibility factors from the Carnahan-Starling equation of state. The rotational mode weighting factor (rotational entropy of a polyatomic ideal gas) is:

$$\hat{S}_{g}^{rot} = \frac{1}{3k_{B}} \left[ \ln \frac{\pi^{1/2} e^{3/2}}{\sigma} \left( \frac{T^{3}}{\Theta_{A} \Theta_{B} \Theta_{C}} \right)^{1/2} \right]$$
(12)

$$\Theta_j = \frac{h^2}{8\pi^2 I_j k_B} \tag{13}$$

$$I_j = \sum_{i=1}^N m_i d_i^2 \tag{14}$$

$$COM_j = \frac{1}{M} \sum_{i=1}^N m_i j_i \tag{15}$$

with  $\sigma$  rotational symmetry,  $\Theta_j$  rotational temperature in the *j* direction,  $I_j$  principle moment of inertia in the *j* direction,  $d_i$  distance of atom *i* from the center of mass (COM), and  $j_i$  the *j* coordinate of atom *i*. There is no fluidicity factor for the gas-phase vibrational component. The solid component has harmonic oscillator weighting factor for all modes:

$$\hat{S}_s^k(\nu) = \frac{\beta h\nu}{e^{\beta h\nu} - 1} - \ln(1 - e^{-\beta h\nu})$$
(16)

$$\beta = (k_B T)^{-1} \tag{17}$$

Integration is used to obtain the entropy for each mode in either of the phases:

$$\hat{S}_x^k = k_B \int_0^\infty \hat{S}_x^k(\nu) f_x^k(\nu) d\nu \tag{18}$$

with x the gas or solid component. The crystalline bundle and oligomer units were considered rigid rods with translation and rotation, and because the cellulose strands were tethered, these translational and rotational contributions were added to the entropy.

#### 2 Statistical Measurement of Thermodynamic Properties

#### 2.1 Statistical Measurements of a Given Composition and a Given State

The averages of the thermodynamic quantities and their uncertainties reported in the manuscript were calculated in the following manner. For a single simulation of a given composition j (e.g. TBAH with 60 wt% water and no urea) and a given state k (e.g. crystalline or dissociated) the thermodynamic quantity of interest, denoted generically as X, at a given moment in time i can be represented as  $X_i^{j,k}$ . These data were recorded once every 100 fs throughout the 10 ns production run equating with a total of 100 000 data points, denoted by N. The average value of that quantity is calculated as:

$$\langle X \rangle^{j,k} = \frac{1}{N} \sum_{i=1}^{N} X_i^{j,k}.$$
 (19)

The instantaneous enthalpy  $H_i^{j,k}$  is calculated as the sum of the instantaneous internal energy  $H_i^{j,k}$  and the product of the instantaneous pressure  $p_i^{j,k}$  and the instantaneous volume  $V_i^{j,k}$ 

$$H_i^{j,k} = U_i^{j,k} + p_i^{j,k} V_i^{j,k}.$$
(20)

Due to the deterministic nature of the computations each of the above quantities were considered as exact measurements of the instantaneous state of the system. The standard deviation of the thermodynamic quantity of interest at a given composition and given state is calculated as

$$\sigma^{j,k}(X^{j,k}) = \sqrt{\frac{1}{N} \sum_{i=1}^{N} (X_i^{j,k} - \langle X \rangle^{j,k})^2},$$
(21)

and the standard uncertainty of that thermodynamic quantity as

$$u_c^{j,k}(X^{j,k}) = \frac{\sigma^{j,k}(X^{j,k})}{\sqrt{N}},\tag{22}$$

and the *relative* standard uncertainty as

$$u_{c,r}^{j,k}(X^{j,k}) = \frac{u_c^{j,k}(X^{j,k})}{|X^{j,k}|}.$$
(23)

The standard uncertainty is a measure of the standard deviation of the mean. Unless otherwise noted, this is the measure represented as the error bars throughout the figures of the main manuscript.

For a single simulation of a given composition j and a given state k the average Gibbs free energy is calculated as the combination of the average enthalpy, average temperature, and average entropy of that respective simulation

$$\langle G \rangle^{j,k} = \langle H \rangle^{j,k} - \langle T \rangle^{j,k} \langle S \rangle^{j,k}.$$
(24)

The standard uncertainty  $(u_c)$  is obtained through error propagation:

$$u_c^{j,k}(\langle T \rangle^{j,k} \langle S \rangle^{j,k}) = |\langle T \rangle^{j,k} \langle S \rangle^{j,k} | \sqrt{u_{c,r}^{j,k}(T^{j,k})^2 + u_{c,r}^{j,k}(S^{j,k})^2}$$
(25)

$$u_{c}^{j,k}(G^{j,k}) = \sqrt{u_{c}^{j,k}(H^{j,k})^{2} + u_{c}^{j,k}(\langle T \rangle^{j,k} \langle S \rangle^{j,k})^{2}}.$$
(26)

#### 2.2 Statistical Measurement of Thermodynamic Changes

The average change in the Gibbs free energy, enthalpy, energy, pairwise interaction energy, temperature, and entropy, denoted here generally as  $\langle \Delta X \rangle$ , at a given composition *j* is calculated as the difference between the dissociated (diss) and crystalline (crys) states:

$$\langle \Delta X \rangle^j = \langle X \rangle^{j,diss} - \langle X \rangle^{j,crys},\tag{27}$$

and the standard uncertainty is estimated through the propagation of error

$$u_c^j((\Delta X)^j) = \sqrt{u_c^{j,diss}(\langle X \rangle^{j,diss})^2 + u_c^{j,crys}(\langle X \rangle^{j,crys})^2}.$$
(28)

## 3 Difference in the Change of the Pairwise Interaction Energy

The difference in the change of the pairwise interaction energy upon dissolution  $(\Delta\Delta U_{pair})$ , given in kcal/mol-AGU, between selected compositions. The interactions are grouped into five different resolutions. Interactions at the higher resolutions combine to form the merged cells at the lower resolutions. Abbreviations represent (CAT) cations, (HOX) hydroxide, (URE) urea, (HOH) water, (CEL) cellulose, (IL) ionic liquid, (MOL) molecular/IL mixture, (AQ) aqueous/molecular/IL mixture, and (TOTAL) all of the components. Colors indicate key changes that are (red) unfavorable, (blue) neutral, and (green) favorable.

Table S1: The difference in the change of the pairwise interaction energy upon dissolution  $(\Delta \Delta U_{pair})$ , given in kcal/mol-AGU, between TBAH with 40 wt% water and TBAH with 60 wt% water.

Ultr	a	Very-H	ligh	Hig	h	Med	ium	Lo	w
Interaction	$\Delta \Delta U_{pair}$								
CAT-CAT	16.3								
CAT-HOX	-34.5	IL-IL	1.9						
HOX-HOX	20.1			MOL-MOL	1.0				
CAT-URE	-			MIGH-MIGH	1.5				
HOX-URE	-	IL-ORE	_			10.10	0.4		
URE-URE	-	URE-URE	-			AQ-AQ	0.4		
CAT-HOH	1.1	п-нон	11.2			1			
HOX-HOH	-12.3	ill-illoii	-11.2	MOL-HOH	-11.2			TOTAL	1.7
URE-HOH	-	URE-HOH	-	1					
нон-нон	9.7	нон-нон	9.7	нон-нон	9.7	1			
CAT-CEL	2.2	II CEI	11.6					1	
HOX-CEL	9.4	IL-CEL	11.0	MOL-CEL	11.6	AO CEI	0.2		
URE-CEL	-	URE-CEL	-			AQ-OLL	2.3		
HOH-CEL	-9.3	HOH-CEL	-9.3	HOH-CEL	-9.3	1			
CEL-CEL	-1.0	CEL-CEL	-1.0	CEL-CEL	-1.0	CEL-CEL	-1.0		

Table S2: The difference in the change of the pairwise interaction energy upon dissolution  $(\Delta\Delta U_{pair})$ , given in kcal/mol-AGU, between TBAH with 60 wt% water/no urea and TBAH with 60 wt% water/33 wt% urea.

Ultr	a	Very-H	ligh	Hig	h	Medi	um	Low		
Interaction	$\Delta \Delta U_{pair}$	Interaction	$\Delta \Delta U_{pair}$	Interaction	$\Delta \Delta U_{pair}$	Interaction	$\Delta \Delta U_{pair}$	Interaction	$\Delta \Delta U_{pair}$	
CAT-CAT	10.8									
CAT-HOX	-21.6	IL-IL	1.9							
HOX-HOX	12.7	1		MOT MOT	0.2					
CAT-URE	1.7	IL-IIRE	5.8	MOL-MOL	0.5					
HOX-URE	4.1		0.0			10.40	0.4			
URE-URE	0.6	URE-URE	0.6			AQ-AQ	0.4			
CAT-HOH	-2.7	П-НОН	16.5							
HOX-HOH	-13.8		-10.5	MOL-HOH	-16.9			TOTAL	0.3	
URE-HOH	-0.4	URE-HOH	-0.4							
нон-нон	8.9	нон-нон	8.9	нон-нон	8.9					
CAT-CEL	2.6	II CEI	0.5					1		
HOX-CEL	6.9		9.0	MOL-CEL	2.3	AO CEI	0.2			
URE-CEL	-7.2	URE-CEL	-7.2			AQ-OLL	0.2			
HOH-CEL	-2.1	HOH-CEL -2.1		HOH-CEL	-2.1	1				
CEL-CEL	-0.2	CEL-CEL	-0.2	CEL-CEL	-0.2	CEL-CEL	-0.2			

Table S3: The difference in the change of the pairwise interaction energy upon dissolution  $(\Delta \Delta U_{pair})$ , given in kcal/mol-AGU, between TBAH with 40 wt% water/33 wt% urea and TBAH with 40 wt% water/40 wt% urea.

Ultr	a	Very-F	ligh	Hig	h	Med	ium	Lo	w
Interaction	$\Delta \Delta U_{pair}$								
CAT-CAT	4.9								
CAT-HOX	-11.4	IL-IL	-4.7						
HOX-HOX	1.8	1		MOT MOT	2.0				
CAT-URE	0.8	IL-URE	0.8	MOL-MOL	-3.0				
HOX-URE	0.0	IL-OILL	0.0			10.40	1.4		
URE-URE	0.9	URE-URE	0.9			AQ-AQ	1.4		
CAT-HOH	0.4	п-нон	9.0						
нох-нон	8.6	ill-illoii	3.0	MOL-HOH	6.7			TOTAL	1.3
URE-HOH	-2.2	URE-HOH	-2.2						
нон-нон	-2.3	нон-нон	-2.3	нон-нон	-2.3				
CAT-CEL	1.0	II CEI	0.8					1	
HOX-CEL	-0.2	IL-CEL	0.8	MOL-CEL	0.1	AO CEI	0.1		
URE-CEL	-0.7	URE-CEL	-0.7			AQ-OLL	-0.1		
HOH-CEL	-0.2	HOH-CEL	-0.2	HOH-CEL	-0.2				
CEL-CEL	-0.1	CEL-CEL	-0.1	CEL-CEL	-0.1	CEL-CEL	-0.1		

## 4 Changes in Hydrogen Bonding upon Dissolution

S4: Breakdown in hydrogen bonds for (top) 40 wt% water and 0wt% urea, (middle) 40 wt% water and 33wt% urea, and (bottom)
ther and 55wt% urea. Columns in gray represent numbers for the dissolved state and columns in white represent numbers for the
state.

$u_c$	T	I	Ι	I	I	I	I	Ι	I	1	
Per URE	Ι	I	I	I	I	I	I	I	I	I	
$u_c$	1	I	I	I	I	I	I	I	I	I	
Per URE	1	Ι	I	I	Ι	I	I	I	I	I	
$n_c$	$\pm 0.00008$	$\pm 0.00009$	$\pm 0.00011$	I	$\pm 0.00018$	Ι	$\pm 0.0007$	Ι	Ι	$\pm 0.0007$	
Per HOH	0.009670732	0.042994194	0.06024	I	0.51694	I	0.8638	Ι	I	1.4936	
$u_c$	$\pm 0.00013$	$\pm 0.00003$	$\pm 0.00006$	I	$\pm 0.00018$	I	$\pm 0.0007$	I	I	$\pm 0.0007$	
Per HOH	0.03947	0.004962787	0.018320929	I	0.55845	I	0.8711	I	I	1.4923	
$u_c$	$\pm 0.0007$	$\pm 0.0009$	$\pm 0.0011$	I	$\pm 0.0018$	I	±0.006	I	I	$\pm 0.007$	
Per IL	0.0928	0.4128	0.5784	I	4.9631	I	8.293	Ι	1	14.340	
$u_c$	$\pm 0.0012$	$\pm 0.0003$	$\pm 0.0005$	I	$\pm 0.0017$	I	$\pm 0.006$	I	I	$\pm 0.007$	
Per IL	0.3789	0.0476	0.1759	I	5.3616	I	8.363	I	I	14.327	
$n_{c}$	$\pm 0.003$	$\pm 0.003$	$\pm 0.004$	I	±0.006	I	$\pm 0.02$	Ι	I	$\pm 0.02$	
Per AGU	0.313	1.393	1.952	I	16.750	I	27.99	Ι	I	48.40	
$n_c$	$\pm 0.004$	$\pm 0.0010$	$\pm 0.0018$	I	±0.006	I	$\pm 0.02$	I	I	$\pm 0.02$	
Per AGU	1.279	0.1608	0.5936	-	18.095		28.23	1		48.35	
Pair	CEL-CEL	CEL-HOX	CEL-HOH	CEL-URE	нон-хон	HOX-URE	нон-нон	HOH-URE	URE-URE	TOTAL	

_												-	_
$u_c$	$\pm 0.0005$	$\pm 0.0007$	$\pm 0.0009$	$\pm 0.0005$	$\pm 0.0015$	$\pm 0.0011$	$\pm 0.0048$	$\pm 0.0018$	$\pm 0.0020$	$\pm 0.006$		$n_{c}$	$\pm 0.0002$
Per URE	0.0506	0.2296	0.4890	0.1971	2.9195	0.6799	5.2906	2.1657	0.5205	12.542		Per URE	0.0189
$n_c$	$\pm 0.0010$	$\pm 0.00010$	$\pm 0.0004$	$\pm 0.0002$	$\pm 0.0015$	$\pm 0.0015$	$\pm 0.0048$	$\pm 0.0023$	$\pm 0.0016$	$\pm 0.006$		<i>n</i> <sup>c</sup>	$\pm 0.0010$
Per URE	0.3083	0.03749	0.1471	0.0457	3.0779	0.7045	5.4198	2.1564	0.6236	12.521		Per URE	0.1351
$u_c$	$\pm 0.00007$	$\pm 0.00010$	$\pm 0.00014$	$\pm 0.00007$	$\pm 0.0002$	$\pm 0.00016$	$\pm 0.0007$	$\pm 0.0003$	$\pm 0.0003$	$\pm 0.0009$		100	$\pm 0.00007$
Per HOH	0.00758	0.03444	0.07332	0.02955	0.4378	0.10194	0.7933	0.3247	0.0780	1.8807		Per HOH	0.00684
$u_c$	$\pm 0.00015$	$\pm 0.00002$	$\pm 0.00006$	$\pm 0.00004$	$\pm 0.0002$	$\pm 0.0002$	$\pm 0.0007$	$\pm 0.0003$	$\pm 0.0002$	$\pm 0.0009$		<i>n</i>	$\pm 0.0004$
Per HOH	0.04623	0.00562	0.02205	0.00686	0.4615	0.1056	0.8127	0.3234	0.0935	1.8775		Per HOH	0.0488
$u_c$	$\pm 0.0007$	$\pm 0.0010$	$\pm 0.0013$	$\pm 0.0007$	$\pm 0.0022$	$\pm 0.0015$	$\pm 0.0069$	$\pm 0.0026$	$\pm 0.0028$	$\pm 0.009$		$n_{c}$	±0.0007
Per IL	0.0728	0.3306	0.7039	0.2837	4.2027	0.9787	7.6158	3.1175	0.7493	18.055		Per IL	0.0657
$n_c$	$\pm 0.0014$	$\pm 0.00015$	$\pm 0.0006$	$\pm 0.0003$	$\pm 0.0022$	$\pm 0.0021$	$\pm 0.0069$	$\pm 0.0032$	$\pm 0.0023$	$\pm 0.009$		n	$\pm 0.004$
Per IL	0.4438	0.05397	0.2117	0.0658	4.4306	1.0142	7.8018	3.1042	0.8977	18.024		Per IL	0.468
$u_c$	$\pm 0.002$	$\pm 0.003$	$\pm 0.004$	$\pm 0.002$	$\pm 0.006$	$\pm 0.004$	$\pm 0.019$	$\pm 0.007$	$\pm 0.008$	$\pm 0.02$		$n_{c}$	$\pm 0.002$
Per AGU	0.205	0.930	1.980	0.798	11.820	2.752	21.419	8.768	2.107	50.78		Per AGU	0.150
$u_c$	$\pm 0.004$	$\pm 0.0004$	$\pm 0.002$	$\pm 0.0010$	$\pm 0.006$	$\pm 0.006$	$\pm 0.020$	$\pm 0.009$	$\pm 0.007$	$\pm 0.05$		$n_{c}$	$\pm 0.008$
Per AGU	1.248	0.1518	0.595	0.1851	12.461	2.852	21.943	8.731	2.525	50.69		Per AGU	1.067
Pair	CEL-CEL	CEL-HOX	CEL-HOH	CEL-URE	нон-хон	HOX-URE	нон-нон	HOH-URE	URE-URE	TOTAL		Pair	CEL-CEL
												L	

$u_c$	$\pm 0.0002$	$\pm 0.0003$	$\pm 0.0004$	$\pm 0.0004$	$\pm 0.0007$	$\pm 0.0009$	$\pm 0.002$	$\pm 0.0014$	$\pm 0.0015$	$\pm 0.0033$
Per URE	0.0189	0.0815	0.2267	0.1781	1.0061	0.5253	1.934	1.8953	0.8346	6.7006
$u_c$	$\pm 0.0010$	$\pm 0.00007$	$\pm 0.0002$	$\pm 0.0002$	$\pm 0.0007$	$\pm 0.0007$	$\pm 0.002$	$\pm 0.002$	$\pm 0.0014$	$\pm 0.003$
Per URE	0.1351	0.02145	0.0670	0.0341	1.0963	0.5305	1.985	1.964	0.9192	6.752
$n_c$	$\pm 0.00007$	$\pm 0.00011$	$\pm 0.00013$	$\pm 0.00014$	$\pm 0.0003$	$\pm 0.0003$	$\pm 0.0008$	$\pm 0.0005$	$\pm 0.0005$	$\pm 0.0012$
Per HOH	0.00684	0.02944	0.08186	0.06432	0.3633	0.1897	0.6983	0.6843	0.3014	2.4194
$u_c$	$\pm 0.0004$	$\pm 0.00003$	$\pm 0.00008$	$\pm 0.00006$	$\pm 0.0002$	$\pm 0.0003$	$\pm 0.0007$	$\pm 0.0006$	$\pm 0.0005$	$\pm 0.0012$
Per HOH	0.0488	0.00775	0.02419	0.01232	0.3958	0.1915	0.7166	0.7091	0.3319	2.4380
$u_c$	$\pm 0.0007$	$\pm 0.0010$	$\pm 0.0013$	$\pm 0.0013$	$\pm 0.002$	$\pm 0.003$	$\pm 0.008$	$\pm 0.005$	$\pm 0.005$	$\pm 0.011$
Per IL	0.0657	0.2826	0.7859	0.6175	3.488	1.821	6.704	6.570	2.893	23.227
$u_c$	$\pm 0.004$	$\pm 0.0003$	$\pm 0.0007$	$\pm 0.0006$	$\pm 0.002$	$\pm 0.002$	$\pm 0.007$	$\pm 0.006$	$\pm 0.005$	$\pm 0.012$
Per IL	0.468	0.0744	0.2322	0.1183	3.800	1.839	6.880	6.807	3.186	23.406
$u_c$	$\pm 0.002$	$\pm 0.002$	$\pm 0.003$	$\pm 0.003$	$\pm 0.006$	$\pm 0.007$	$\pm 0.018$	$\pm 0.011$	$\pm 0.012$	$\pm 0.0259$
Per AGU	0.150	0.644	1.790	1.407	7.944	4.148	15.270	14.965	6.590	52.9069
$u_c$	$\pm 0.008$	$\pm 0.0006$	$\pm 0.002$	$\pm 0.0013$	$\pm 0.005$	$\pm 0.005$	$\pm 0.016$	$\pm 0.014$	$\pm 0.011$	$\pm 0.06$
Per AGU	1.067	0.1694	0.529	0.2695	8.656	4.189	15.671	15.506	7.258	53.31
Pair	CEL-CEL	CEL-HOX	CEL-HOH	CEL-URE	нон-хон	HOX-URE	нон-нон	HOH-URE	URE-URE	TOTAL

able S5: Breakdown in hydrogen bonds for (top) 60 wt% water and 0wt% urea, (middle) 60 wt% water and 33wt% urea, and (bottom)
0 wt% water and 55wt% urea. Columns in gray represent numbers for the dissolved state and columns in white represent numbers for the
rystalline state.

$u_c$	Ι	I	I	I	I	I	I	I	I	I	
Per URE	1	1	Ι	1	Ι	I	Ι	Ι	1	1	
$n_c$	I	I	I	I	Ι	I	Ι	I	I	ļ	
Per URE	1	I	Ι	I	Ι	I	I	Ι	I	1	
$u_c$	$\pm 0.00004$	$\pm 0.00008$	$\pm 0.00009$	1	$\pm 0.00014$	I	$\pm 0.0006$	Ι	1	$\pm 0.0007$	
Per HOH	0.00396	0.02103	0.06235	I	0.23679	Ι	1.1235	Ι	Ι	1.4477	
$n_c$	$\pm 0.00010$	$\pm 0.000014$	$\pm 0.00005$	I	$\pm 0.00011$	I	$\pm 0.0007$	I	I	$\pm 0.0007$	
Per HOH	0.01701	0.003046	0.01930	I	0.25602	I	1.1489	I	I	1.4443	
$n_c$	$\pm 0.0008$	$\pm 0.0018$	$\pm 0.0019$	I	$\pm 0.003$	1	$\pm 0.014$	Ι	I	$\pm 0.015$	
Per IL	0.0856	0.4544	1.3470	I	5.115	Ι	24.271	Ι	T	31.273	
$u_c$	$\pm 0.002$	$\pm 0.0003$	$\pm 0.0011$	I	$\pm 0.002$	I	$\pm 0.015$	I	I	$\pm 0.02$	
Per IL	0.367	0.0658	0.4170	Ι	5.531	I	24.819	Ι	I	31.20	
$u_c$	$\pm 0.0018$	$\pm 0.004$	$\pm 0.004$	I	$\pm 0.007$	I	$\pm 0.03$	Ι	1	$\pm 0.03$	
Per AGU	0.1926	1.022	3.031	1	11.509	1	54.61	Ι	1	70.36	
$n_c$	$\pm 0.005$	$\pm 0.0007$	$\pm 0.002$	Ι	$\pm 0.006$	I	$\pm 0.03$	Ι	I	$\pm 0.03$	
Per AGU	0.827	0.1481	0.938	1	12.444		55.84	1		70.20	
Pair	CEL-CEL	CEL-HOX	CEL-HOH	CEL-URE	нон-хон	HOX-URE	нон-нон	HOH-URE	URE-URE	TOTAL	

		~ 10			_	~	~	_	10			~	
$n^{2}$	$\pm 0.000$	±0.000	$\pm 0.0007$	$\pm 0.0006$	3000.0土	$\pm 0.0007$	$\pm 0.0048$	$\pm 0.0020$	$\pm 0.0015$	$\pm 0.006$		u <sub>c</sub>	$\pm 0.00014$
Per URE	0.0251	0.1234	0.4587	0.1800	1.3879	0.2287	6.4229	2.7364	0.6230	12.186		Per URE	0.00959
$u_c$	$\pm 0.0011$	$\pm 0.00012$	$\pm 0.0004$	$\pm 0.0003$	$\pm 0.0011$	$\pm 0.0008$	$\pm 0.0045$	$\pm 0.0020$	$\pm 0.0018$	$\pm 0.006$		$n_c$	$\pm 0.0004$
Per URE	0.1515	0.01384	0.1265	0.0735	1.4884	0.2453	6.6574	2.8183	0.6549	12.230		Per URE	0.1102
$u_c$	$\pm 0.00004$	$\pm 0.00007$	$\pm 0.00011$	$\pm 0.00009$	$\pm 0.00014$	$\pm 0.00011$	$\pm 0.0007$	$\pm 0.0003$	$\pm 0.0002$	$\pm 0.0008$		$u_c$	$\pm 0.00005$
Per HOH	0.00376	0.01850	0.06875	0.02698	0.20803	0.03427	0.9627	0.4102	0.0934	1.8265		Per HOH	0.00347
$u_c$	$\pm 0.0002$	$\pm 0.00002$	$\pm 0.00006$	$\pm 0.00005$	$\pm 0.0002$	$\pm 0.00012$	$\pm 0.0007$	$\pm 0.0003$	$\pm 0.0003$	$\pm 0.0008$		$u_c$	$\pm 0.00013$
Per HOH	0.0227	0.00207	0.01896	0.01102	0.2231	0.03677	0.9979	0.4224	0.0982	1.8331		Per HOH	0.03980
$u_c$	$\pm 0.0010$	$\pm 0.0015$	$\pm 0.002$	$\pm 0.0019$	$\pm 0.003$	$\pm 0.002$	$\pm 0.016$	$\pm 0.006$	$\pm 0.005$	$\pm 0.018$		$n_c$	$\pm 0.0011$
Per IL	0.0813	0.3999	1.486	0.5834	4.498	0.741	20.817	8.869	2.019	39.494		Per IL	0.0749
$u_c$	$\pm 0.004$	$\pm 0.0004$	$\pm 0.0013$	$\pm 0.0010$	$\pm 0.004$	$\pm 0.003$	$\pm 0.015$	$\pm 0.006$	$\pm 0.006$	$\pm 0.018$		$u_c$	$\pm 0.003$
Per IL	0.491	0.0448	0.4101	0.2384	4.824	0.795	21.576	9.134	2.122	39.636		Per IL	0.860
$u_c$	$\pm 0.0017$	$\pm 0.003$	$\pm 0.004$	$\pm 0.003$	$\pm 0.005$	$\pm 0.004$	$\pm 0.03$	$\pm 0.011$	$\pm 0.009$	$\pm 0.03$		$u_c$	$\pm 0.0014$
Per AGU	0.1406	0.692	2.570	1.009	7.778	1.281	36.00	15.335	3.491	68.29		Per AGU	0.0978
$u_c$	$\pm 0.006$	$\pm 0.0007$	$\pm 0.002$	$\pm 0.0018$	$\pm 0.006$	$\pm 0.005$	$\pm 0.03$	$\pm 0.011$	$\pm 0.010$	$\pm 0.07$		$n_c$	$\pm 0.004$
Per AGU	0.849	0.0776	0.709	0.4122	8.341	1.375	37.31	15.794	3.670	68.54		Per AGU	1.123
Pair	CEL-CEL	CEL-HOX	CEL-HOH	CEL-URE	НОН-ХОН	HOX-URE	нон-нон	HOH-URE	URE-URE	TOTAL		Pair	CEL-CEL

$n_c$	$\pm 0.00014$	$\pm 0.0002$	$\pm 0.0004$	$\pm 0.0004$	$\pm 0.0005$	$\pm 0.0005$	$\pm 0.002$	$\pm 0.0014$	$\pm 0.0013$	$\pm 0.0029$
Per URE	0.00959	0.0449	0.2000	0.1896	0.5008	0.1858	2.198	2.1928	0.9704	6.4922
$u_c$	$\pm 0.0004$	$\pm 0.00004$	$\pm 0.0003$	$\pm 0.0002$	$\pm 0.0005$	$\pm 0.0005$	$\pm 0.002$	$\pm 0.0014$	$\pm 0.0014$	$\pm 0.003$
Per URE	0.1102	0.00223	0.0579	0.0558	0.5337	0.2128	2.283	2.2822	1.0175	6.555
$u_c$	$\pm 0.00005$	$\pm 0.00008$	$\pm 0.0002$	$\pm 0.00015$	$\pm 0.0002$	$\pm 0.0002$	$\pm 0.0007$	$\pm 0.0005$	$\pm 0.0005$	$\pm 0.0011$
Per HOH	0.00347	0.01622	0.0723	0.06849	0.1809	0.0671	0.7943	0.7923	0.3506	2.3457
$u_c$	$\pm 0.00013$	$\pm 0.000013$	$\pm 0.00010$	$\pm 0.00009$	$\pm 0.0002$	$\pm 0.0002$	$\pm 0.0007$	$\pm 0.0005$	$\pm 0.0005$	$\pm 0.0011$
Per HOH	0.03980	0.000805	0.02090	0.02017	0.1928	0.0769	0.8247	0.8246	0.3676	2.3684
$n_c$	$\pm 0.0011$	$\pm 0.002$	$\pm 0.003$	$\pm 0.003$	$\pm 0.004$	$\pm 0.004$	$\pm 0.016$	$\pm 0.011$	$\pm 0.010$	$\pm 0.023$
Per IL	0.0749	0.351	1.562	1.480	3.910	1.451	17.165	17.122	7.578	50.694
$u_c$	$\pm 0.003$	$\pm 0.0003$	$\pm 0.002$	$\pm 0.002$	$\pm 0.004$	$\pm 0.004$	$\pm 0.016$	$\pm 0.011$	$\pm 0.011$	$\pm 0.023$
Per IL	0.860	0.0174	0.452	0.436	4.167	1.662	17.824	17.821	7.945	51.184
$u_c$	$\pm 0.0014$	$\pm 0.002$	$\pm 0.005$	$\pm 0.004$	$\pm 0.005$	$\pm 0.005$	$\pm 0.020$	$\pm 0.014$	$\pm 0.014$	$\pm 0.0299$
Per AGU	0.0978	0.458	2.039	1.932	5.105	1.895	22.410	22.354	9.893	66.1839
$u_c$	$\pm 0.004$	$\pm 0.0004$	$\pm 0.003$	$\pm 0.002$	$\pm 0.005$	$\pm 0.005$	$\pm 0.021$	$\pm 0.014$	$\pm 0.014$	±0.07
Per AGU	1.123	0.0227	0.590	0.569	5.441	2.170	23.270	23.266	10.373	66.82
Pair	CEL-CEL	CEL-HOX	CEL-HOH	CEL-URE	нон-хон	HOX-URE	НОН-НОН	HOH-URE	URE-URE	TOTAL

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