

Supporting Information (SI)

Oxidative Steam Reforming of Ethanol over $M_x\text{La}_{2-x}\text{Ce}_{1.8}\text{Ru}_{0.2}\text{O}_{7-\delta}$ ($M = \text{Mg}, \text{Ca}$)

Catalysts: Effect of Alkaline Earth Metal Substitution and Support on Stability and Activity[†]

Ho-Chen Hsieh,^{a, c} Ping-Wen Tsai,^a Yuan-Chia Chang,^a Sheng-Feng Weng,^a Hwo-Shuenn Sheu,^d Yu-Chun Chuang,^d and Chi-Shen Lee^{a, b*}

^a*Department of Applied Chemistry, National Chiao-Tung University, 1001 University Rd., Hsinchu 30010, Taiwan*

^b*Center for Emergent Functional Matter Science, National Chiao-Tung University, Hsinchu 30010, Taiwan.*

^c*Graduate Degree Program of Science and Technology of Accelerator Light Source, National Chiao-Tung University, Hsinchu 30010, Taiwan*

^d*National Synchrotron Radiation Research Center, Hsinchu, 30010, Taiwan*

* Corresponding author: Chi-Shen Lee

Fax: +886 3 5723764

E-mail addresses: chishen@mail.nctu.edu.tw

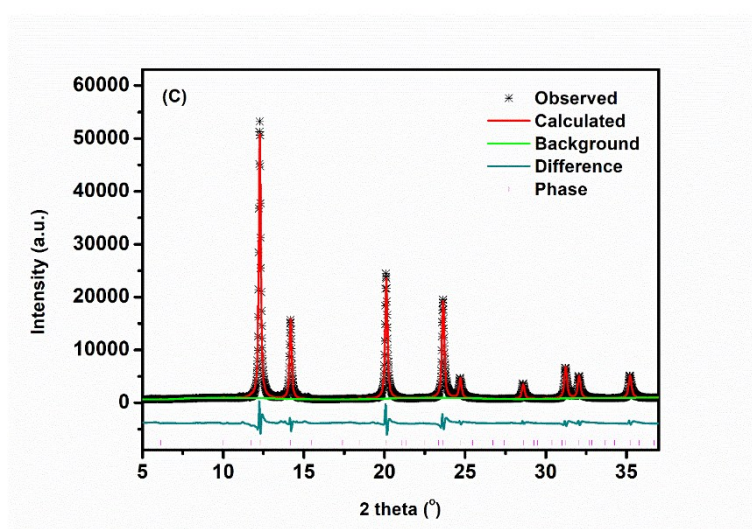
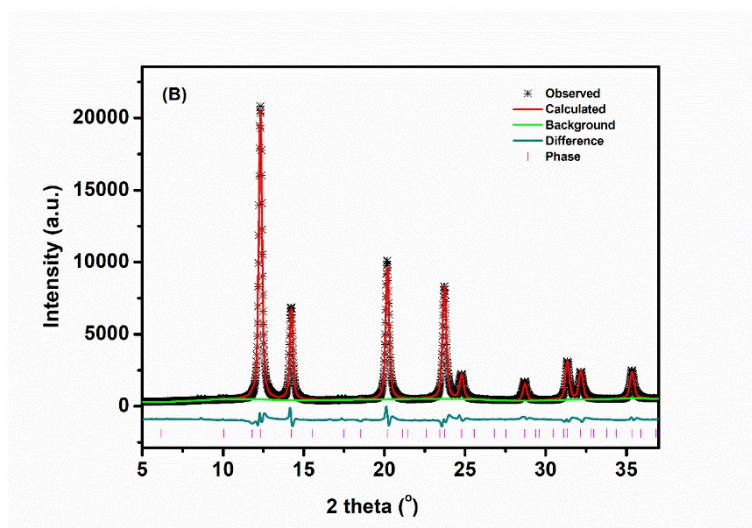
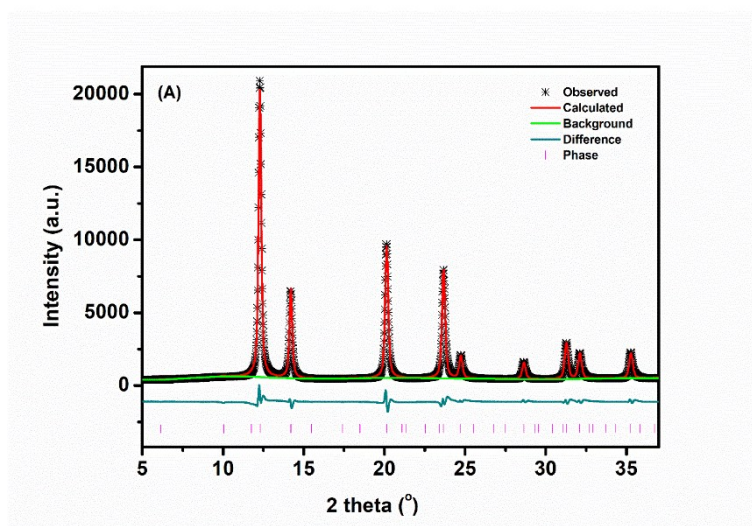
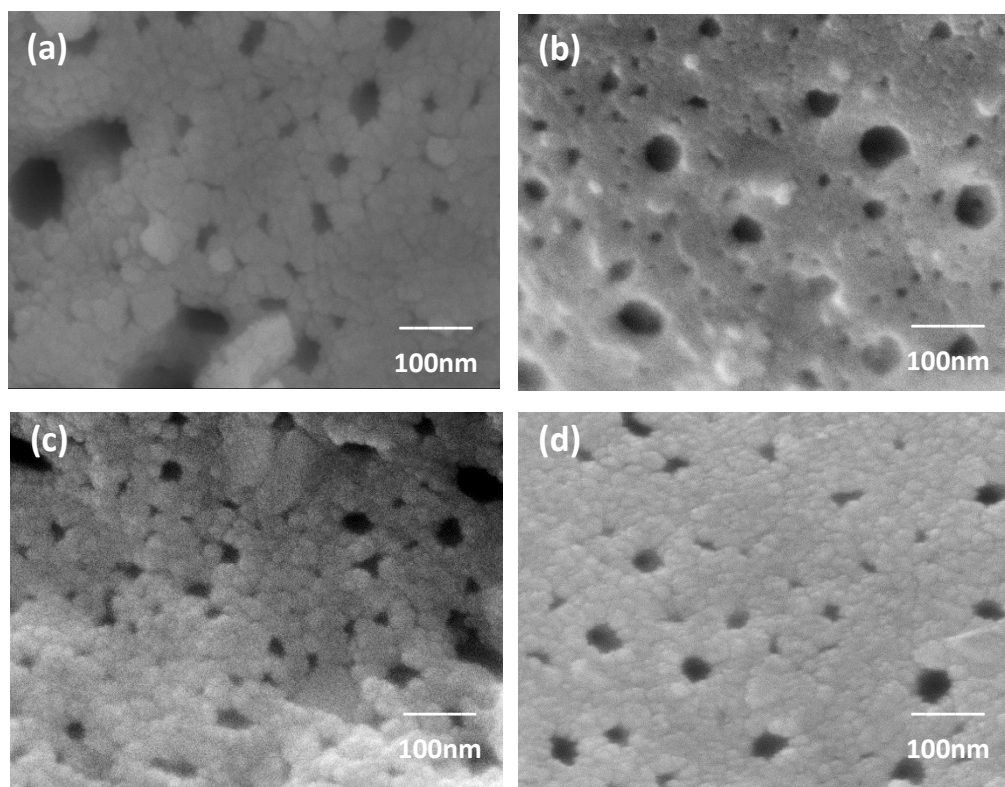
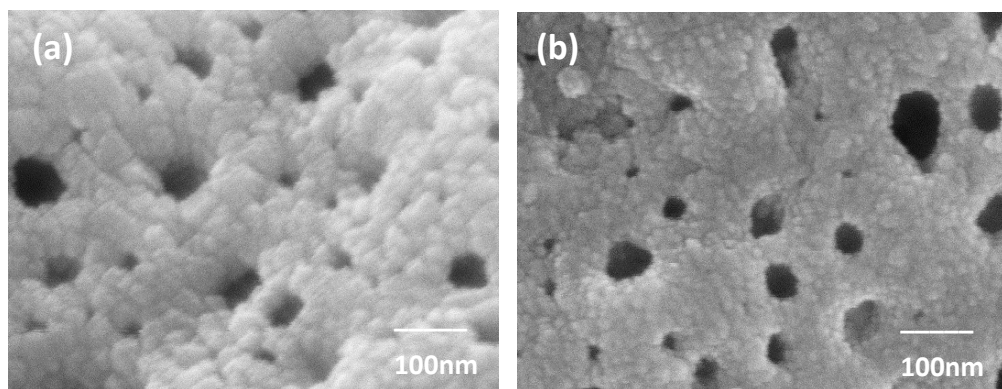


Figure S1. Rietveld Refinement of (A) LCRO, (B) MLCRO03, and (C) CLCRO02.



(A)



(B)

Figure S2. SEM image of solid solution (A) MLCRO (a) $x=0.0$, (b) $x=0.2$, (c) $x=0.3$ and (B) CLCRO (a) $x=0.1$ and (b) $x=0.2$.

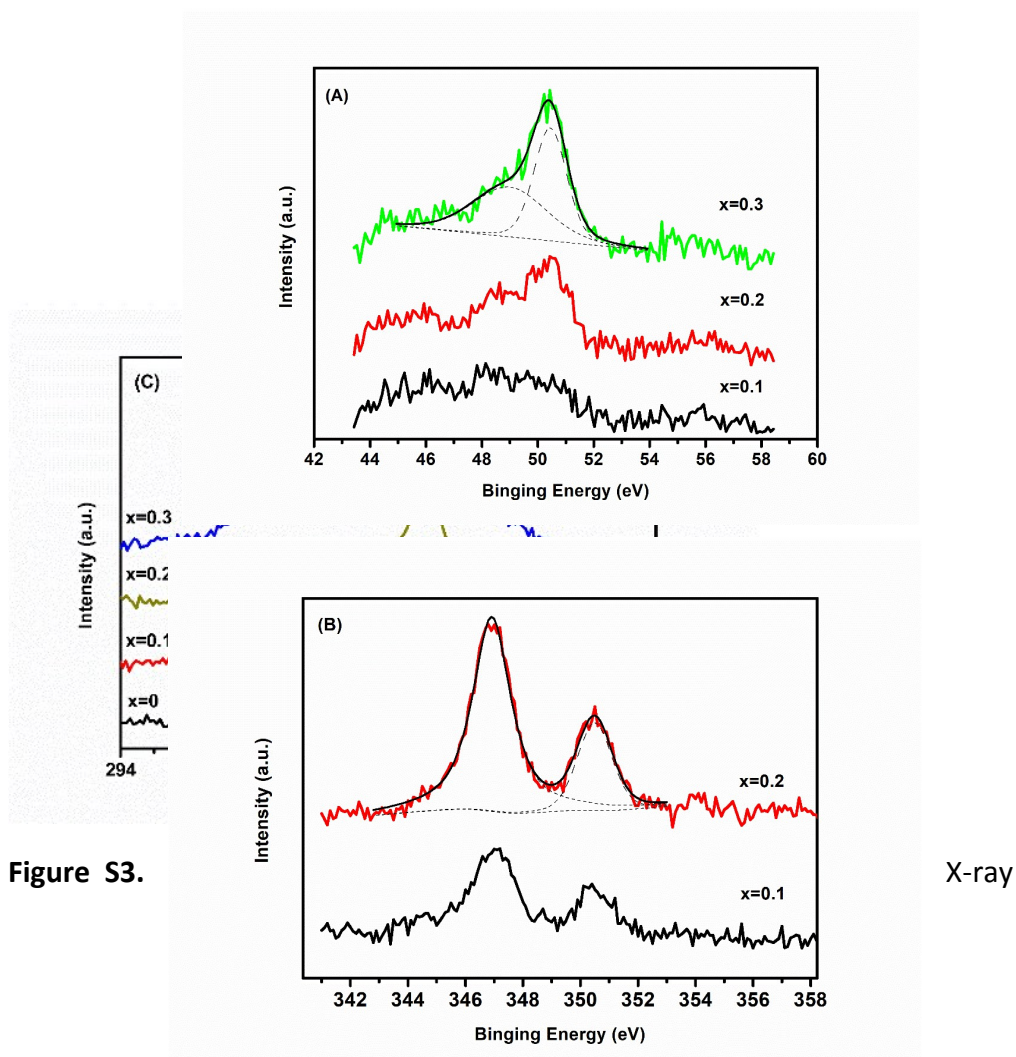


Figure S3.

X-ray

photoelectron spectra of $M_x\text{La}_{2-x}\text{Ce}_{1.8}\text{Ru}_{0.2}\text{O}_{7-\delta}$ ($M = \text{Mg, Ca}$) (A) Mg2p and (B) Ca2p and (C) MLCRO0-03 of Ru3d.

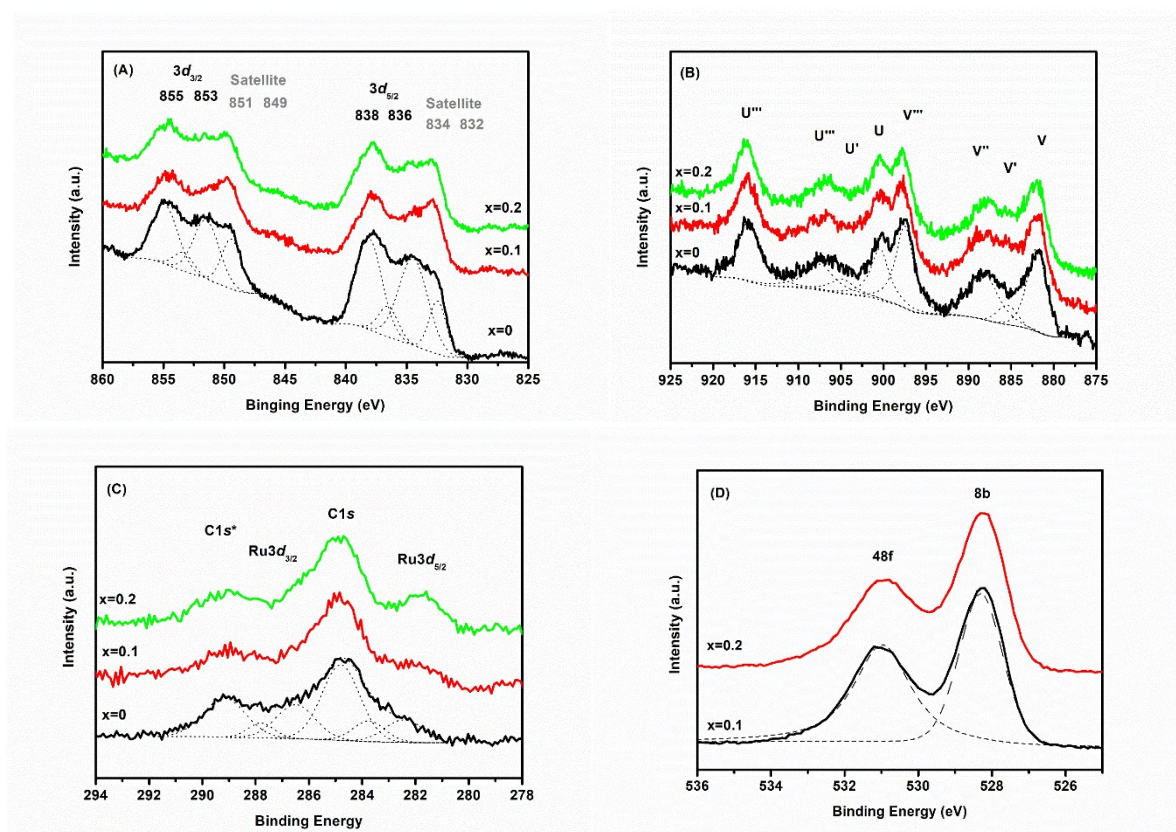


Figure S4. X-ray photoelectron spectra of $\text{Ca}_x\text{La}_{2-x}\text{Ce}_{1.8}\text{Ru}_{0.2}\text{O}_{7-\delta}$ (A) La 3d; (B) Ce 3d; (C) Ru 3d, and (D) O 1s.

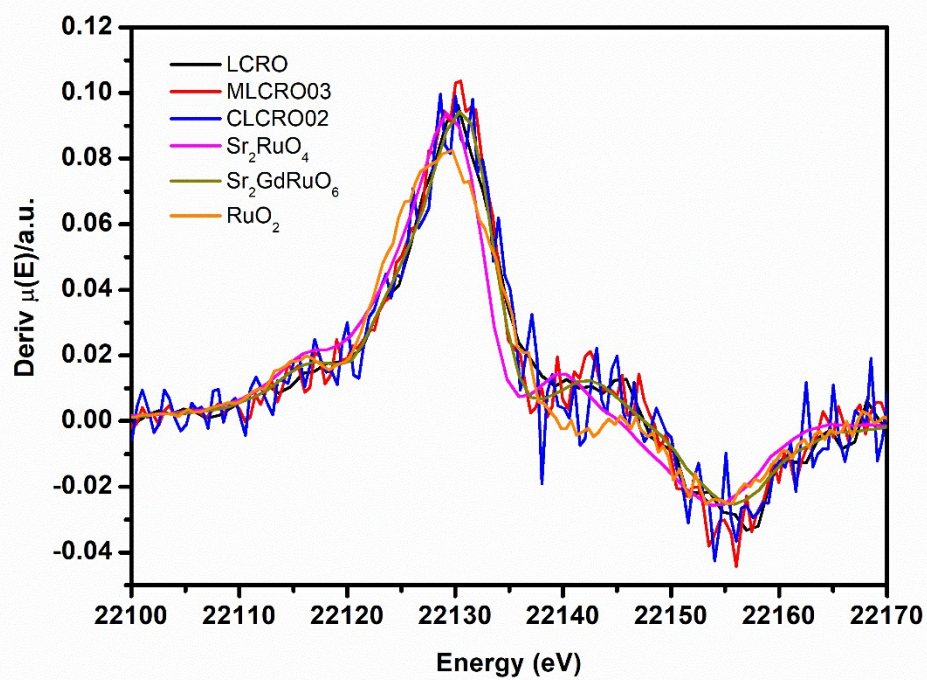


Figure S5. The first derivatives of the Ru K-edge XANES spectrum for LCRO, MLCRO03, and CLCRO02 catalysts, and the reference compounds of Sr_2RuO_4 , $\text{Sr}_2\text{GdRuO}_6$, and RuO_2 .

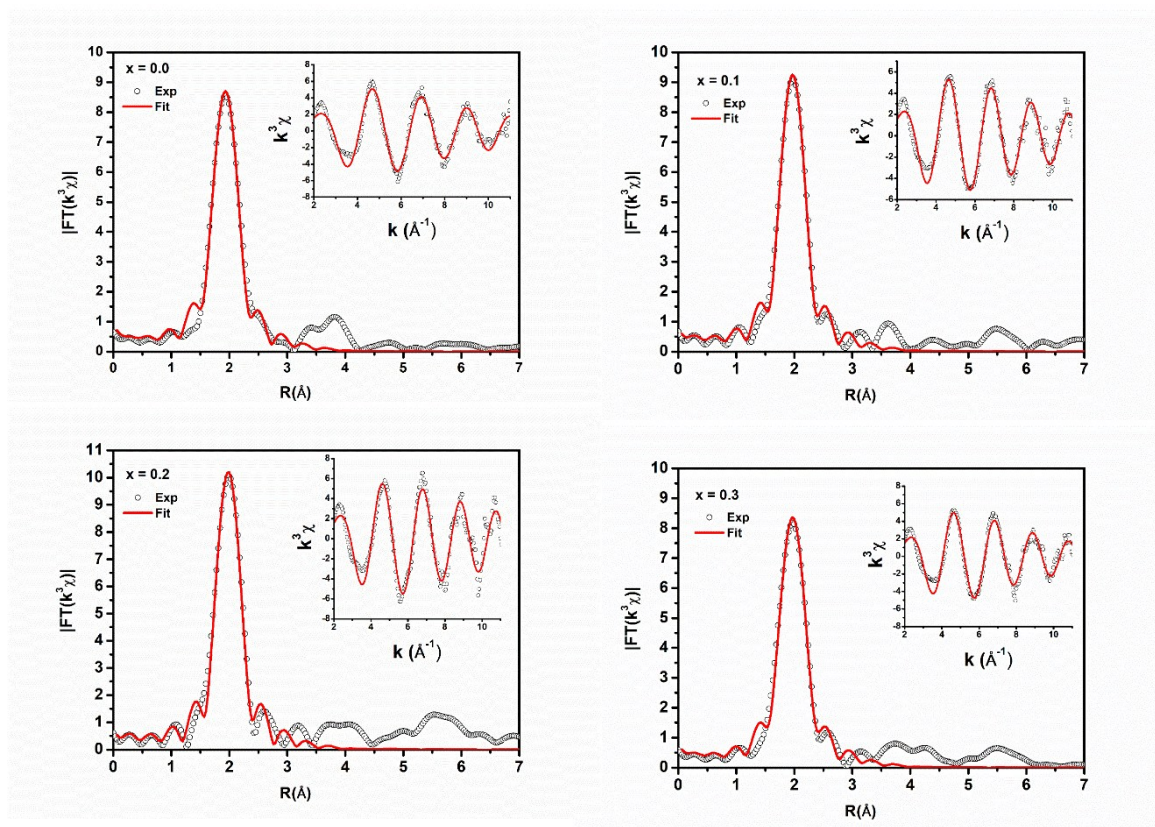


Figure S6. The Ru K-edge EXAFS spectra of MLCRO0-03.

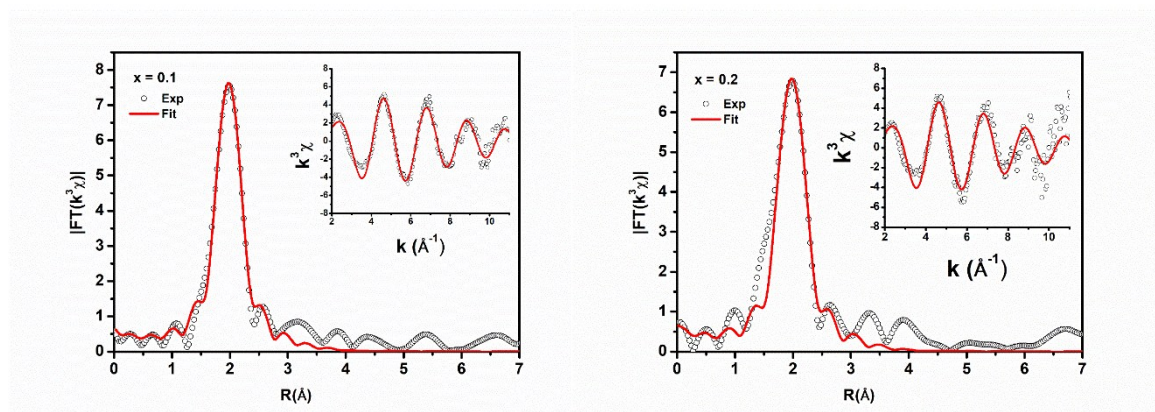


Figure S7. The Ru K-edge EXAFS spectra of CLCRO01-02.

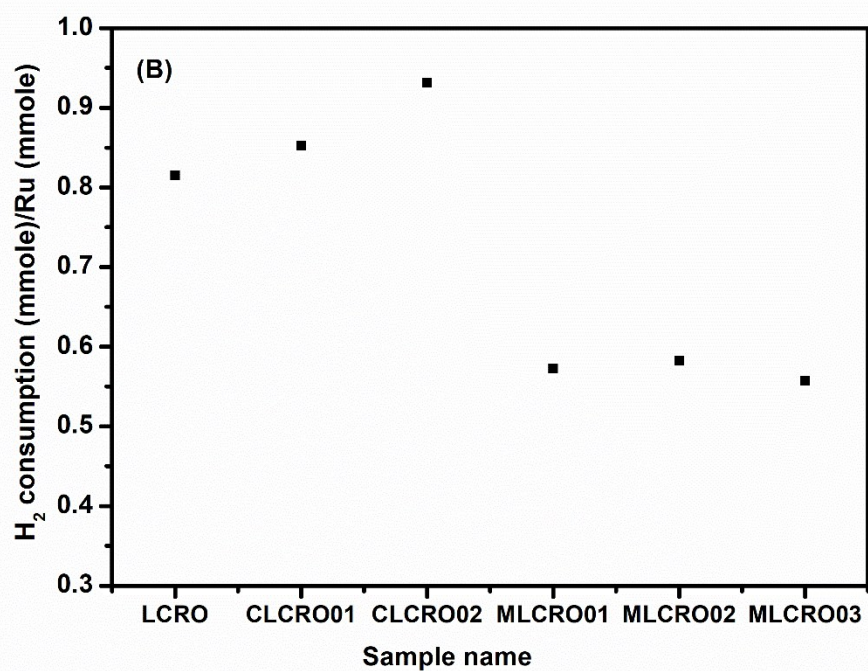
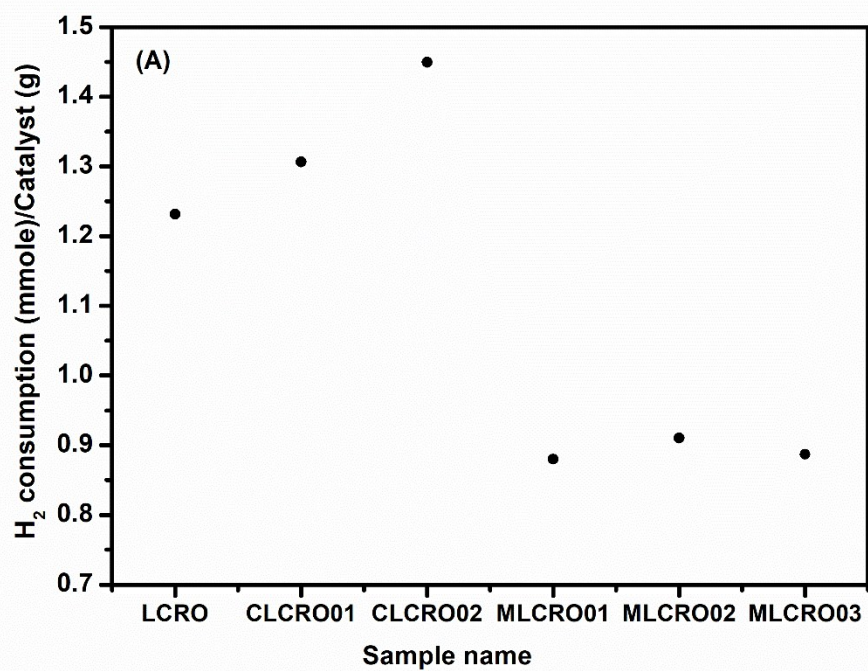


Figure S8. Temperature programmed reduction profile of (A) H₂ consumption per catalysts of $M_x\text{La}_{2-x}\text{Ce}_{1.8}\text{Ru}_{0.2}\text{O}_{7-\delta}$ ($M = \text{Mg}, \text{Ca}$) and (B) H₂ consumption per Ru atom of $M_x\text{La}_{2-x}\text{Ce}_{1.8}\text{Ru}_{0.2}\text{O}_{7-\delta}$ ($M = \text{Mg}, \text{Ca}$).

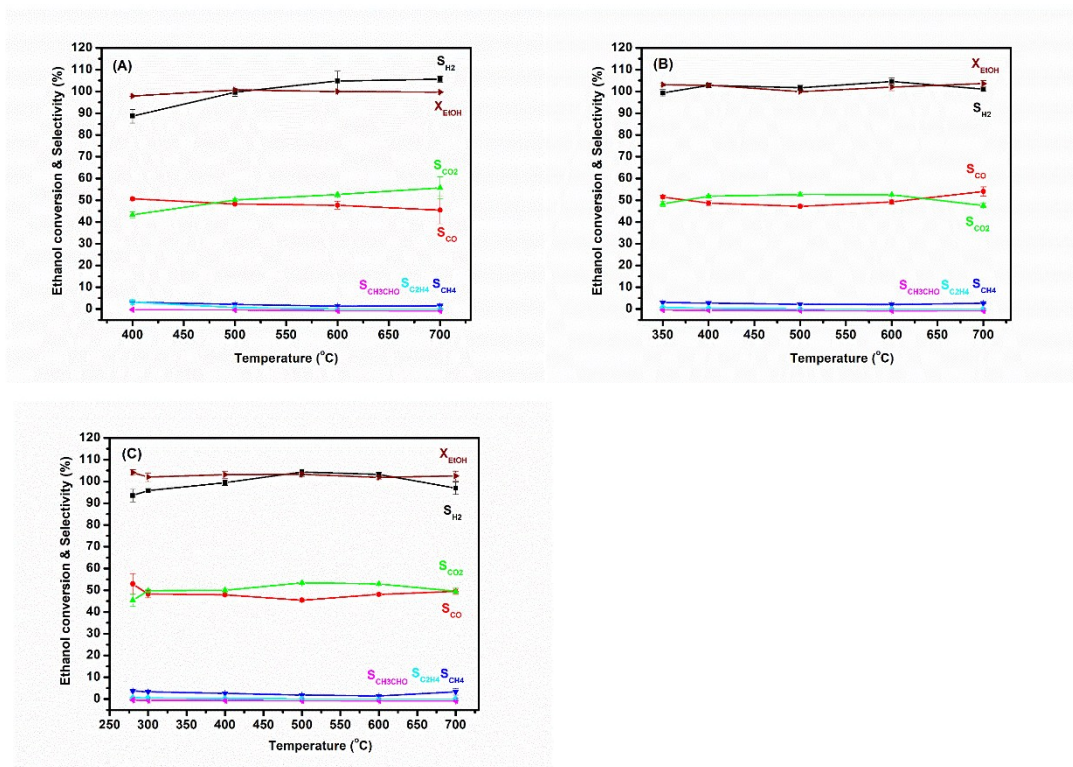


Figure S9. The activation temperature of catalysts for (A) LCRO/Al₂O₃, (B) MLCRO03/Al₂O₃, (C) CLCRO02/Al₂O₃ (H₂O/ethanol = 3 and GHSV = 1.6 × 10⁵ h⁻¹).

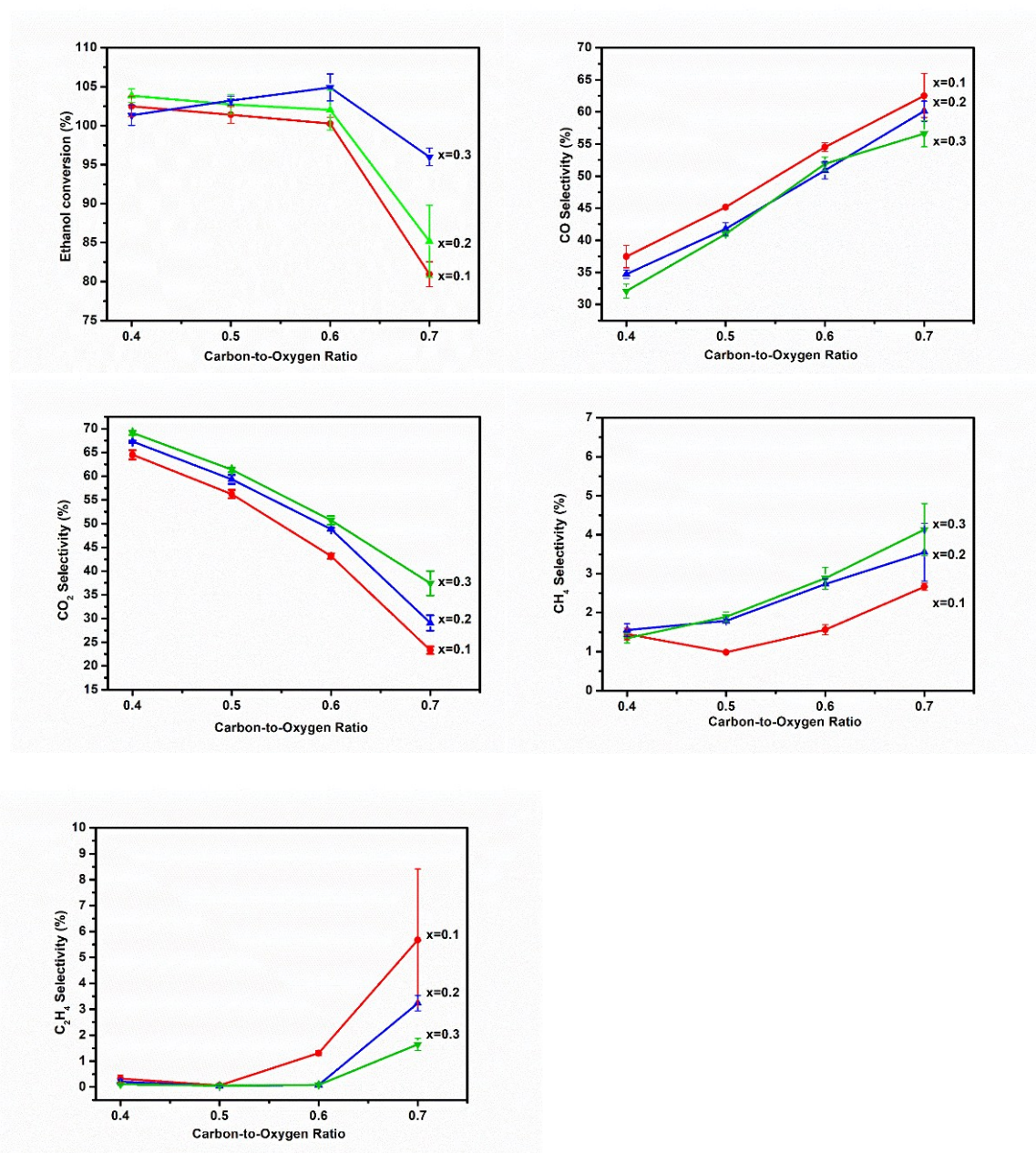


Figure S10 (A). Performance in OSRE of $\text{Mg}_x\text{La}_{2-x}\text{Ce}_{1.8}\text{Ru}_{0.2}\text{O}_{7-\delta}/\text{Al}_2\text{O}_3$ with Ethanol conversion, CO, CO₂, CH₄, and C₂H₄ selectivity ($\text{H}_2\text{O}/\text{ethanol} = 3$, GHSV = $1.6 \times 10^5 \text{ h}^{-1}$, $T = 350^\circ\text{C}$).

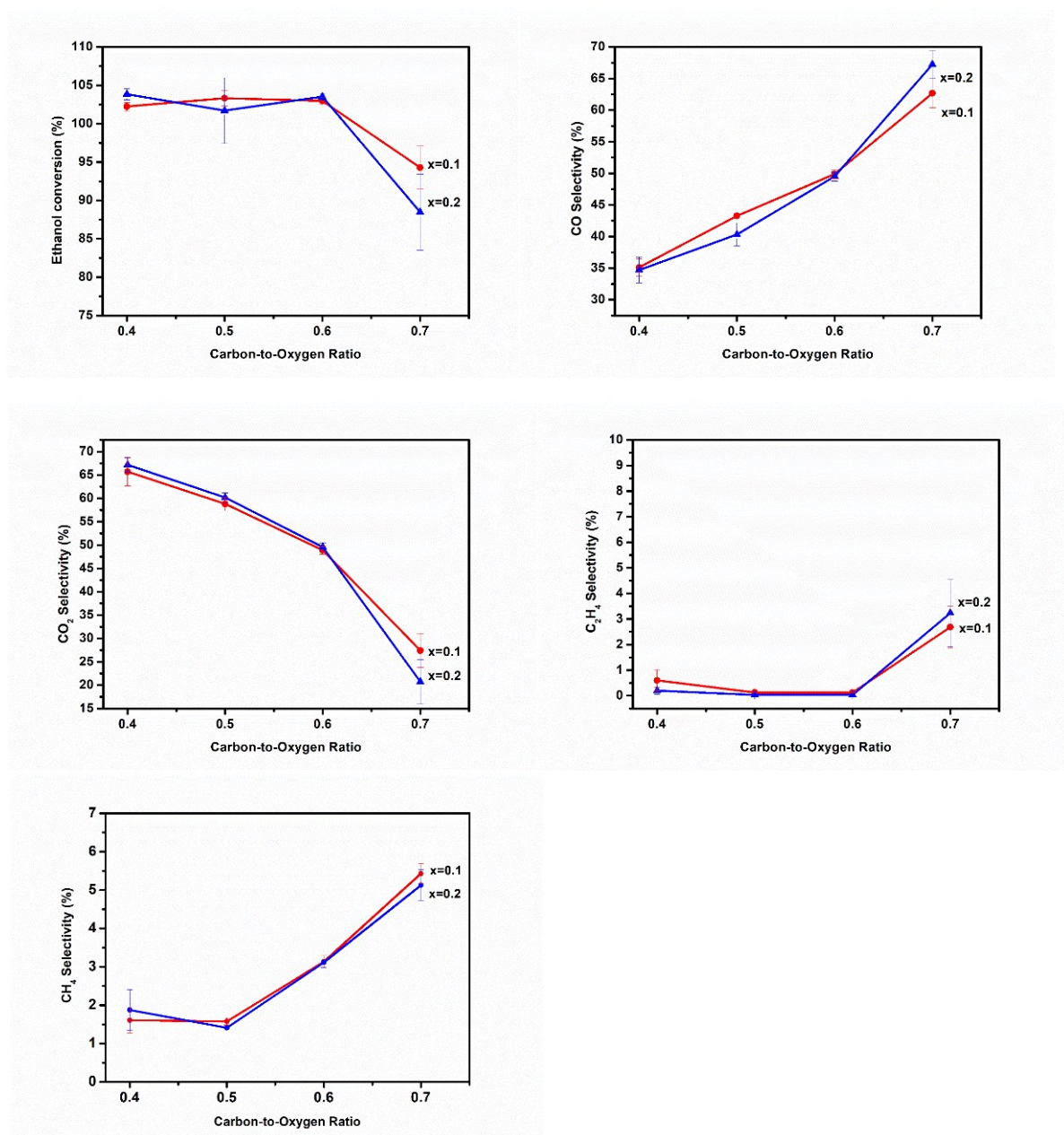


Figure S10 (B). Performance in OSRE of $\text{Ca}_x\text{La}_{2-x}\text{Ce}_{1.8}\text{Ru}_{0.2}\text{O}_{7-\delta}/\text{Al}_2\text{O}_3$ with Ethanol conversion, CO, CO₂, CH₄, and C₂H₄ selectivity (H₂O/ethanol = 3, GHSV = $1.6 \times 10^5 \text{ h}^{-1}$, T = 280°C).

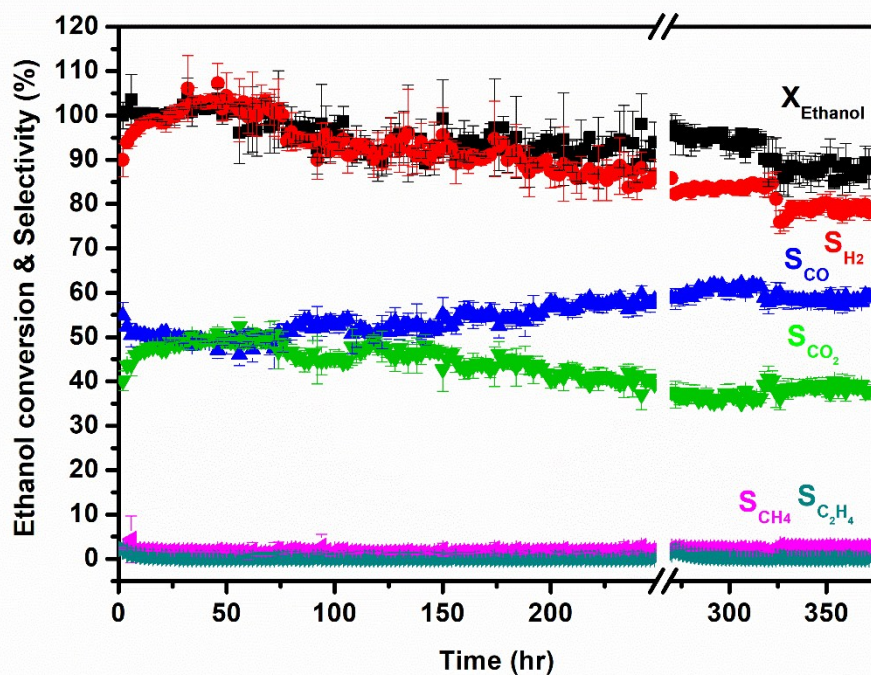


Figure S11. Stability test on catalyst MLCRO03/Al₂O₃ under T = 350°C, H₂O/ethanol = 3, GHSV = 1.6 x 10⁵ h⁻¹ condition.

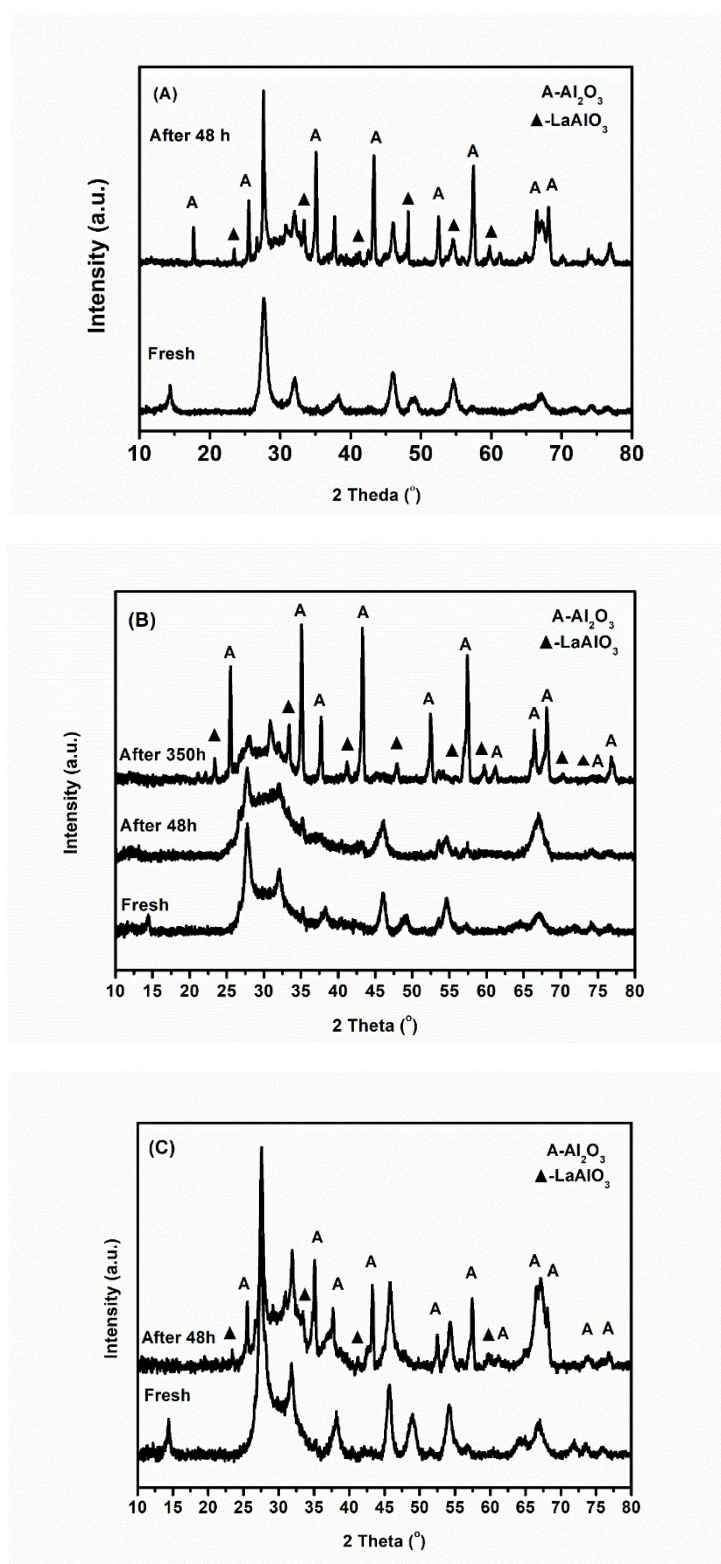


Figure S12. Powder X-ray diffraction of (A) LCRO/Al₂O₃, (B) MLCRO03/Al₂O₃, and (C) CLCRO02/Al₂O₃ before and after long-term stability test.

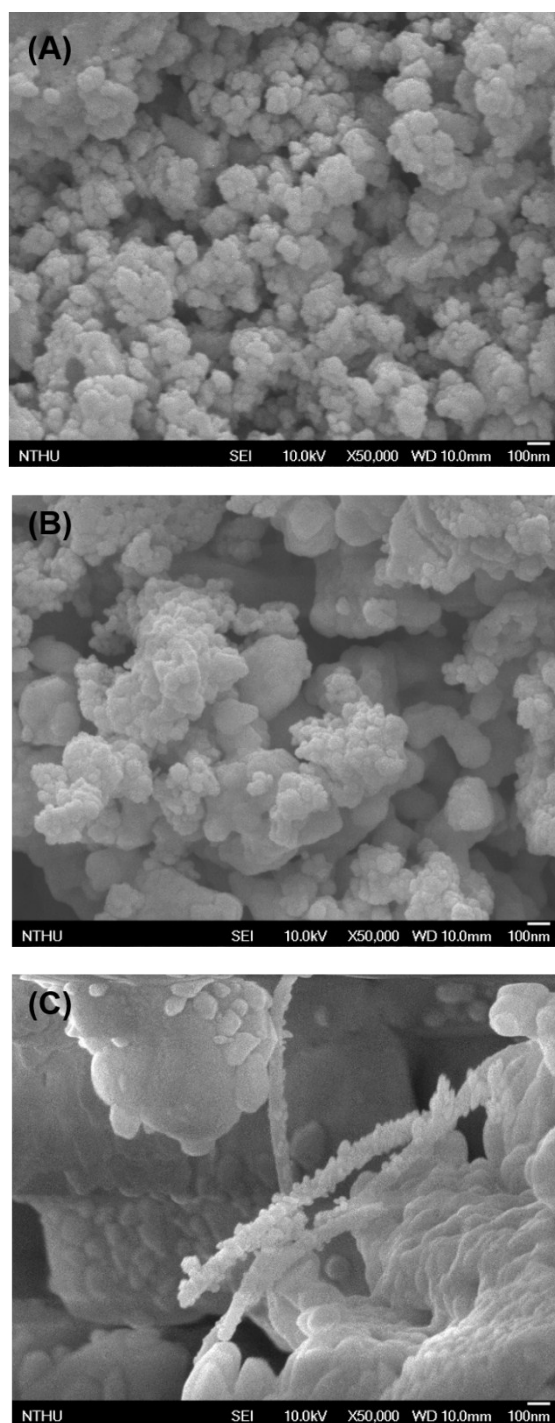


Figure S13. SEM for MLCRO03/Al₂O₃, T = 350°C, H₂O/ethanol = 3, GHSV = $1.6 \times 10^5 \text{ h}^{-1}$ condition (A) Fresh; (B) after 48 hours; (C) after 350 hours long-term stability test.

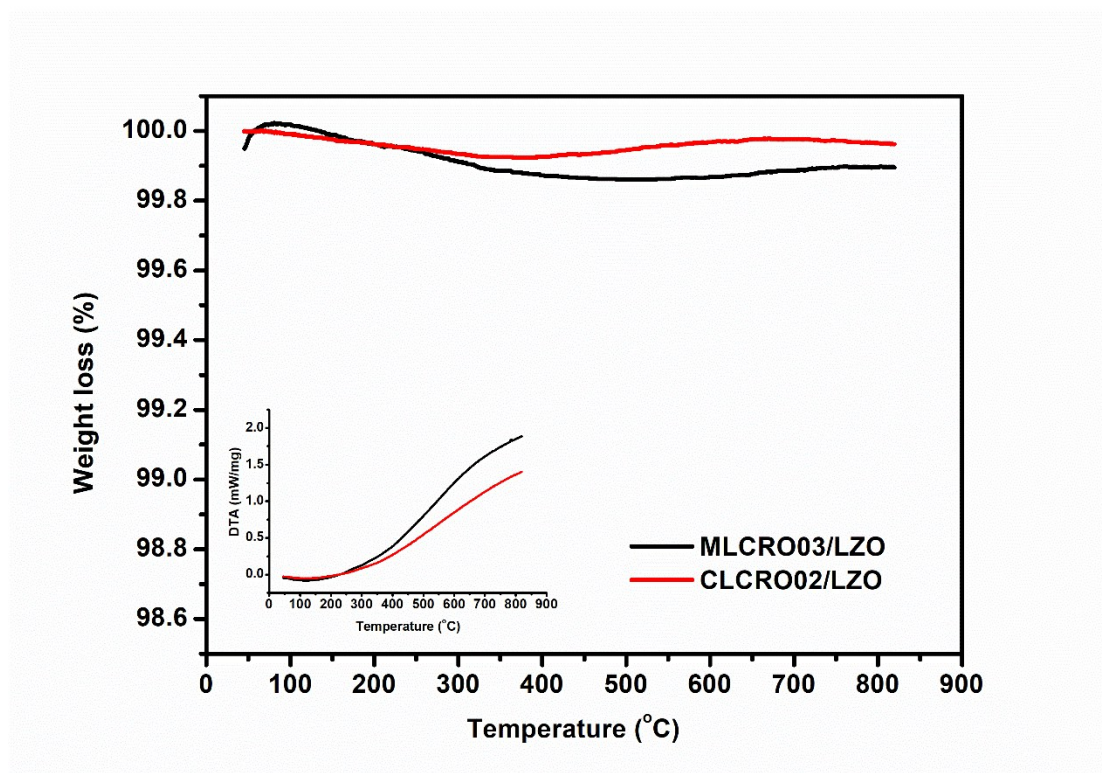


Figure S14. TGA and DSC profiles of ATR on LZO catalysts after long-term reaction.

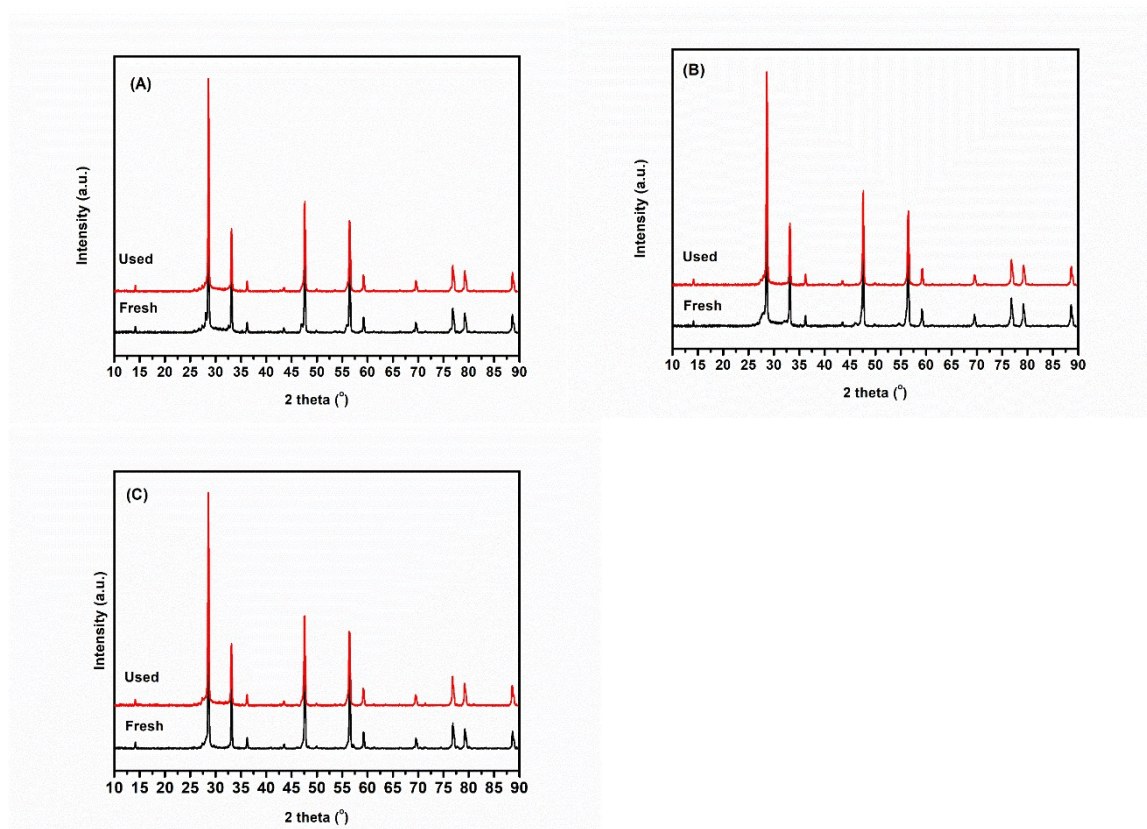


Figure S15. Powder X-ray diffraction of (a) LCRO/LZO, (b) MLCRO03/LZO, and (c) CLCRO02/LZO before and after long-term stability test (OSR, $\text{H}_2\text{O}/\text{ethanol} = 3$, GHSV = $1.6 \times 10^5 \text{ h}^{-1}$, $T = 400^\circ\text{C}$).

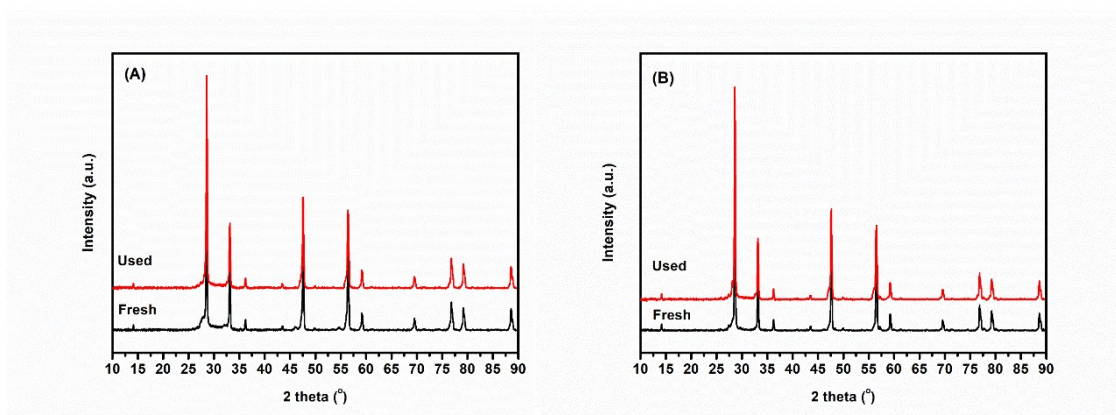


Figure S16. Powder X-ray diffraction of (A) MLCRO03/LZO and (B) CLCRO02/LZO before and after long-term stability test (ATR, H₂O/ethanol = 3, GHSV = 1.6 x 10⁵ h⁻¹).

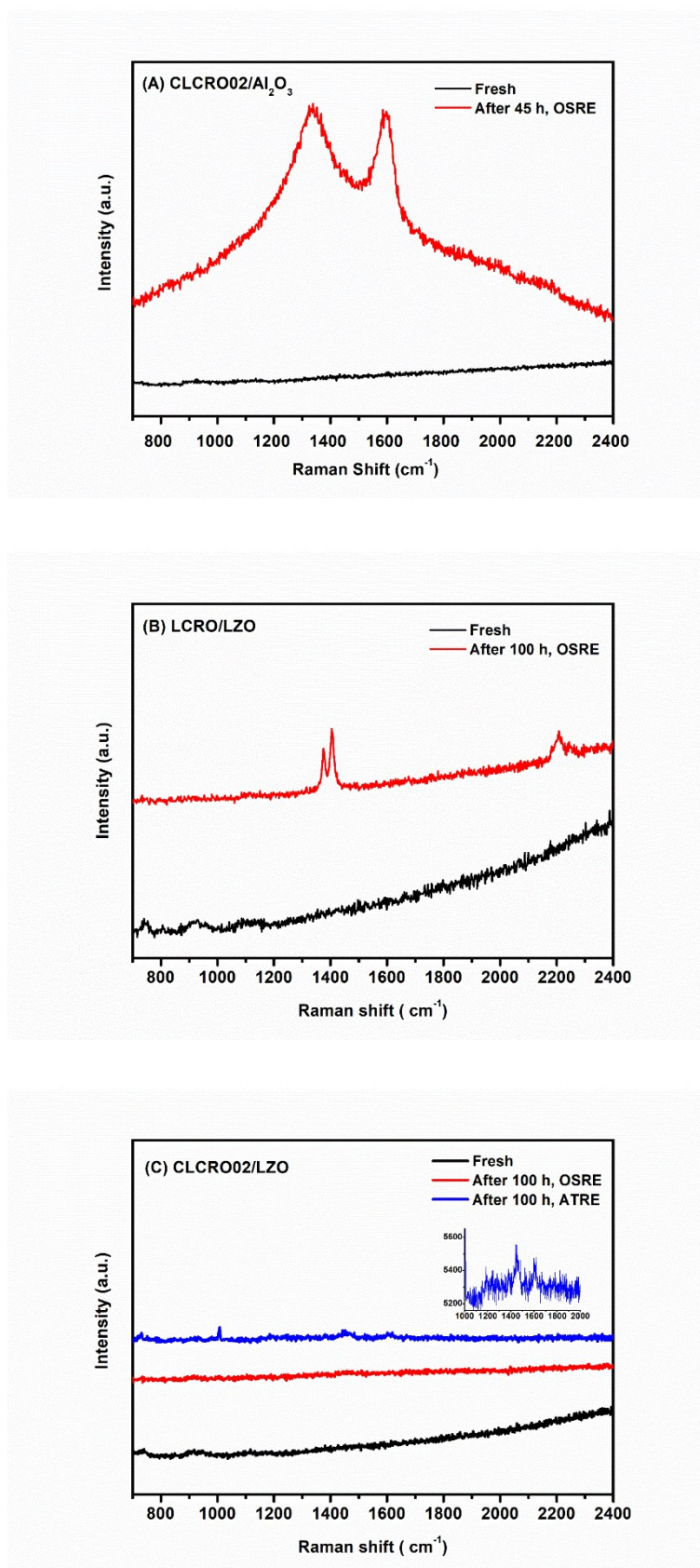
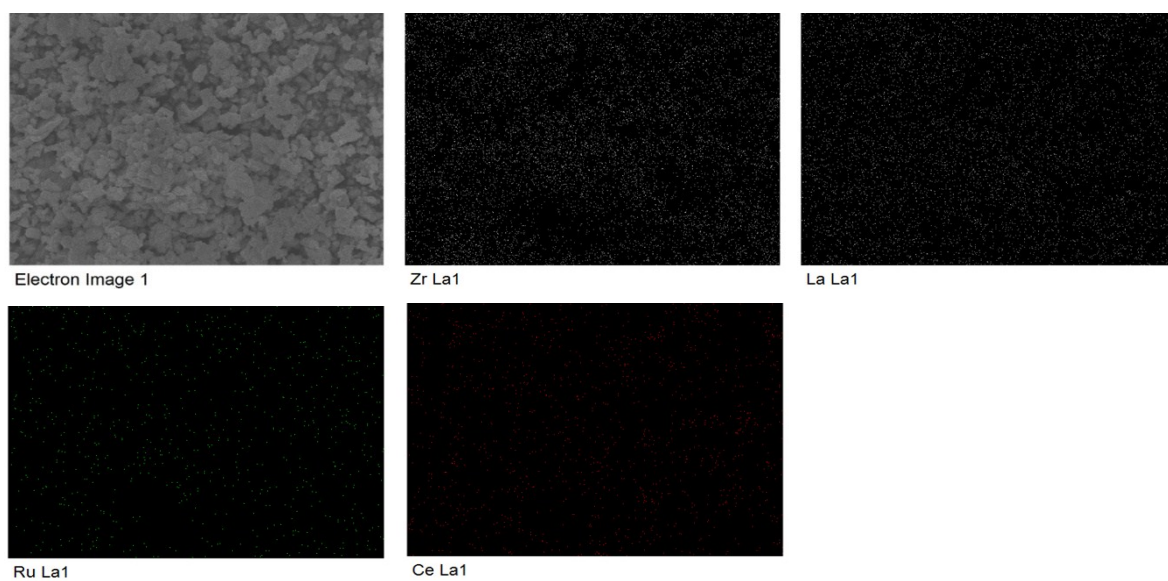


Figure S17. The Raman spectrum of fresh and long-termed used catalysts (A) CLCRO02/ Al_2O_3 , (B) LCRO/LZO, (C) CLCRO02/LZO.

(A) LCRO/LZO Fresh



(B) LCRO/LZO Used

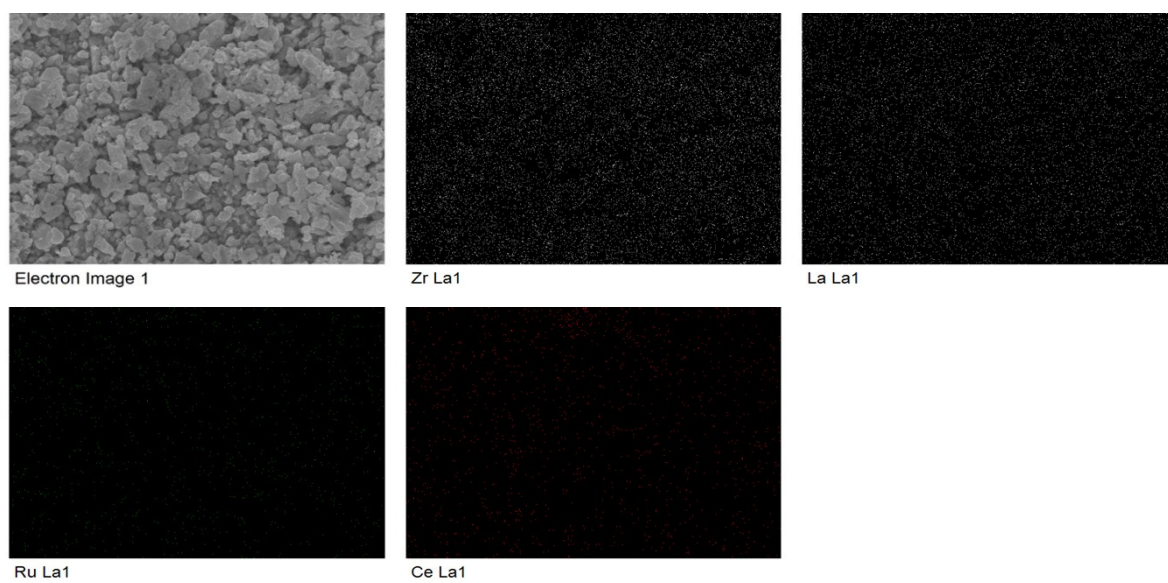
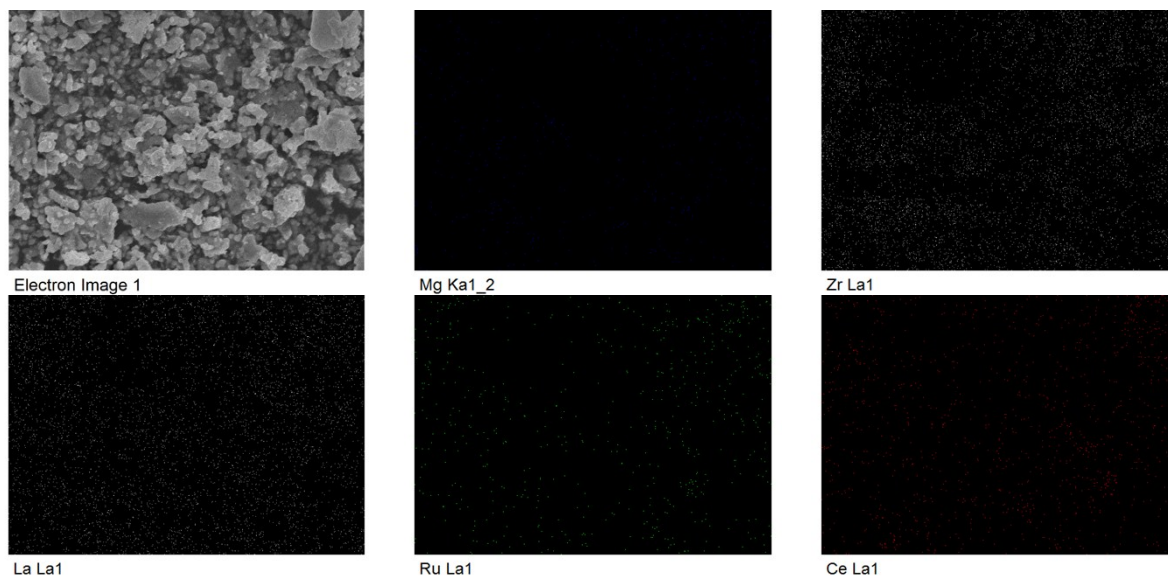


Figure S18. SEM and EDX of (A) fresh and (B) used catalyst LCRO/LZO for long-term used in OSRE.

(A) MLCRO03/LZO Fresh



(B) MLCRO03/LZO Used

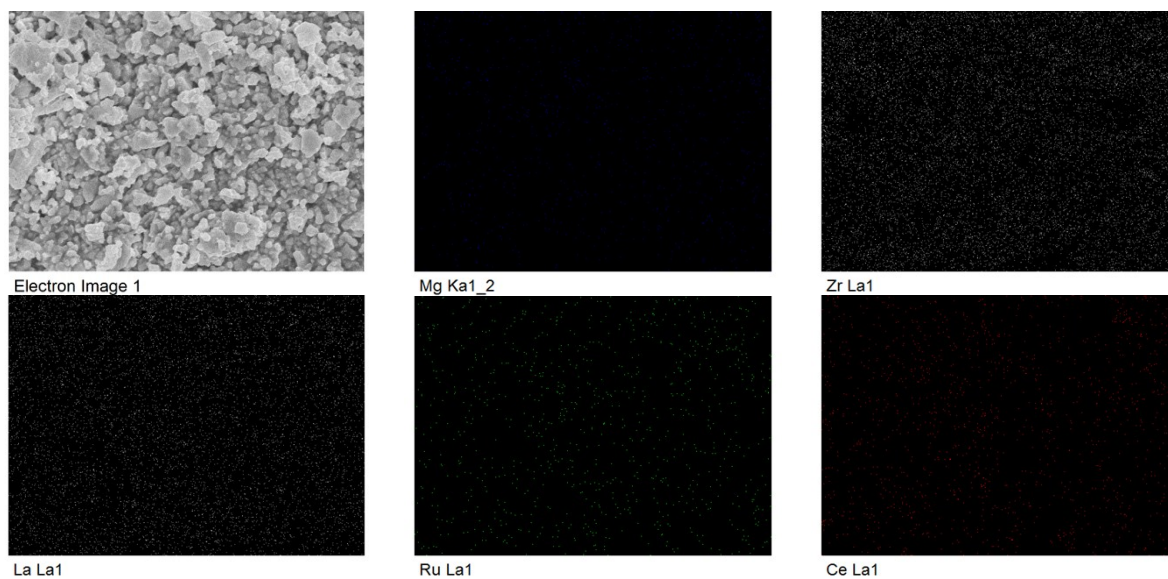
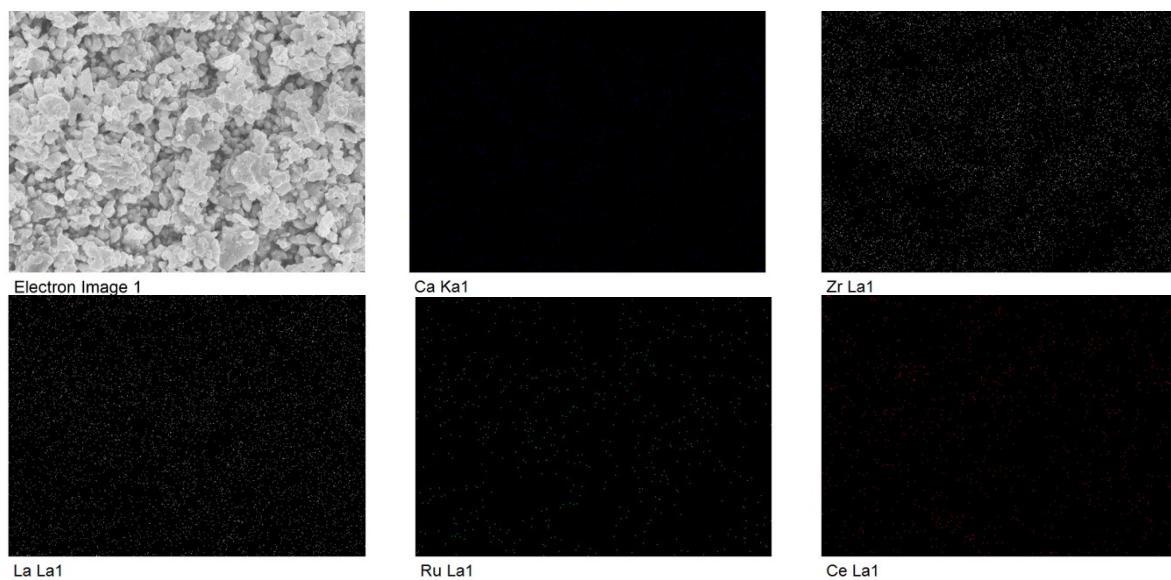


Figure S19. SEM and EDX of (A) fresh and (B) used catalyst MLCRO03/LZO for long-term used in OSRE.

(A) CLCRO02/LZO Fresh



(B) CLCRO02/LZO Used

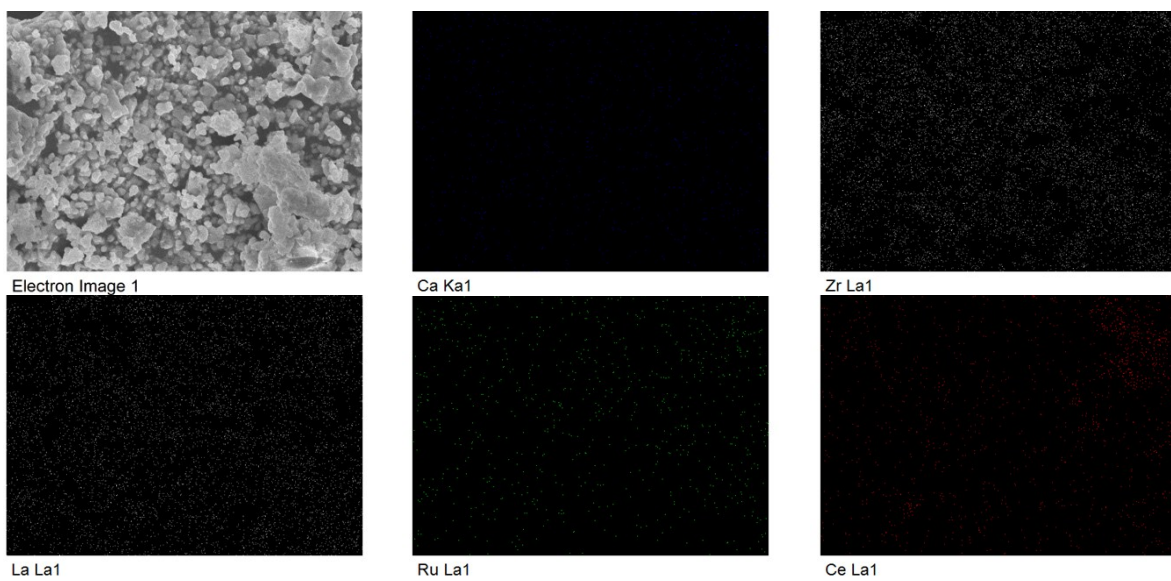


Figure S20. SEM and EDX of (A) fresh and (B) used catalyst CLCRO02/LZO for long-term used in OSRE.

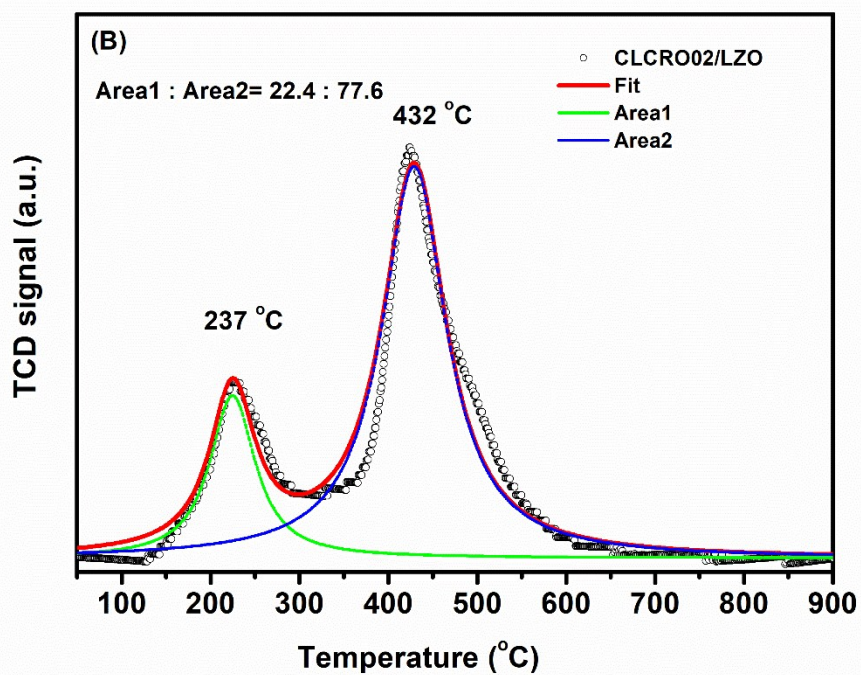
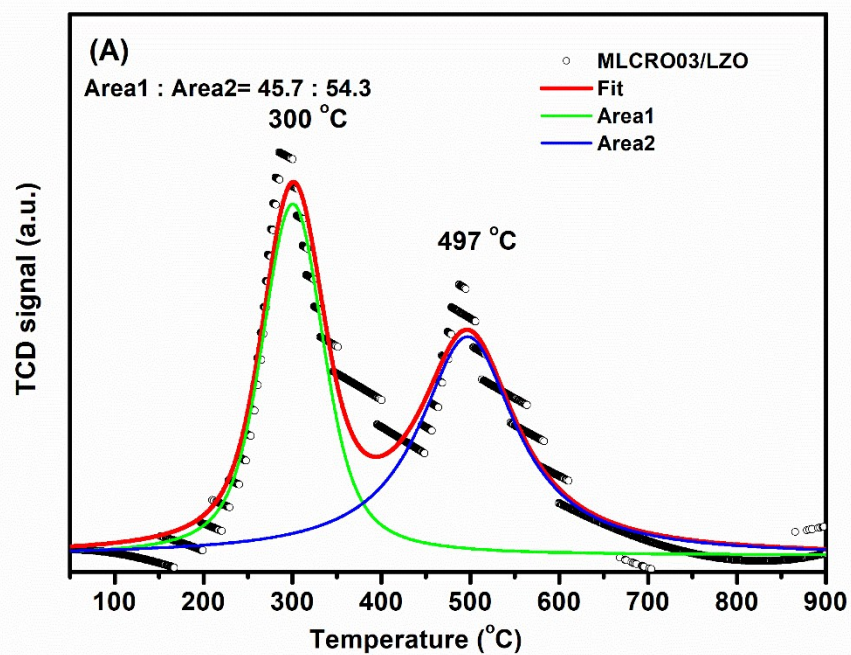


Figure S21. Temperature programmed reduction profile of (A) MLCRO03/LZO and (B) CLCRO02/LZO

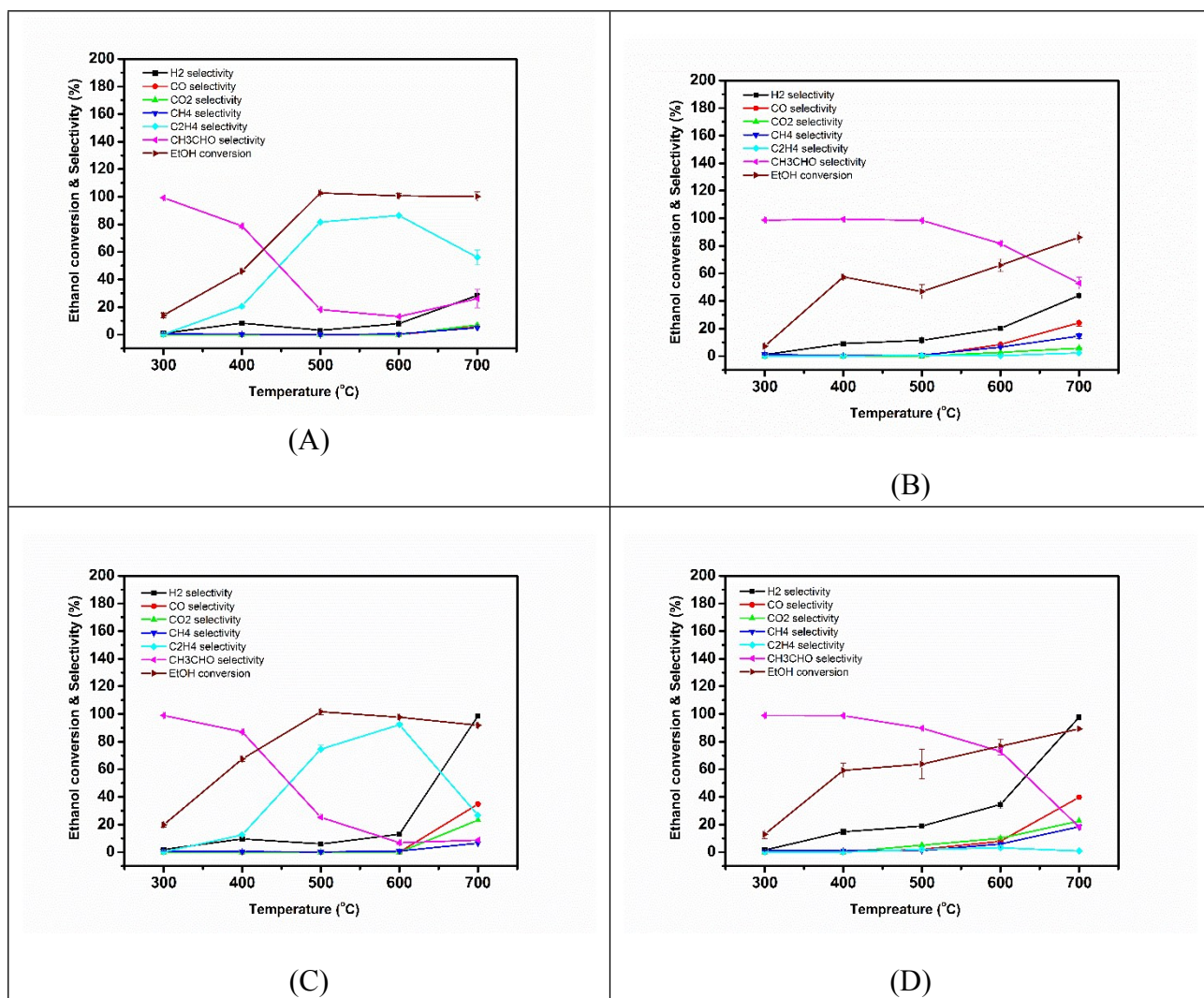


Figure S22. Ethanol conversion on non-oxidative condition ($P(O_2) = 0$, $S/E=3$, $GHSV=16,000 \text{ h}^{-1}$) with catalysts of (A) MLCRO03/Al₂O₃, (B) MLCRO03/LZO, (C) CLCRO02/Al₂O₃ and (D) CLCRO02/LZO.

Table S1. Powder diffraction of $M_x\text{La}_{2-x}\text{Ce}_{1.8}\text{Ru}_{0.2}\text{O}_{7-\delta}$ ($M = \text{Mg}, \text{Ca}$) as obtained from Rietveld Refinement of synchrotron diffraction data.

x	LCRO	MLCRO03	CLCRO02
R_{wp} (%)	5.96	6.94	9.78
R_p (%)	4.89	5.55	8.42
a (Å)	11.14(1)	11.11(2)	11.15(1)
x(O48f)	0.367(2)	0.392(2)	0.384(3)
B-O48f (Å)	2.36(1)	2.52(2)	2.47(2)
A-O8b (Å)	2.412(1)	2.407(3)	2.414(2)
A-O48f (Å)	2.46(1)	2.30(1)	2.36(2)
B-O48f-B (°)	113.0(12)	102.3(10)	105.7(13)
A-O48f-A (°)	111.1(5)	117.4(12)	113.4(14)
O48f-B-O48f (°)	107.6(7)	112.8(4)	111.3(6)
B-O48f-A (°)	109.3(1)	109.0(1)	109.3(1)

continue from the previous page

(a) LCRO

Atom	Site	x	y	z	Occupancy	100* U_{iso} (Å ²)
La	16 <i>d</i>	0.5	0.5	0.5	0.9	2.53(2)
Ce	16 <i>c</i>	0.0	0.0	0.0	0.9	2.53(2)
Ru	16 <i>c</i>	0.0	0.0	0.0	0.1	2.53(2)
O2	8 <i>b</i>	0.375	0.375	0.375	1.01(2)	5.3(4)
O3	48 <i>f</i>	0.367(2)	0.125	0.125	1.01(2)	5.3(4)

(b) MLCRO03

Atom	Site	x	y	z	Occupancy	100* U_{iso} (Å ²)
La	16 <i>d</i>	0.5	0.5	0.5	0.9	2.19(3)
Mg	16 <i>d</i>	0.5	0.5	0.5	0.1	2.19(3)
Ce	16 <i>c</i>	0.0	0.0	0.0	0.9	2.19(3)
Ru	16 <i>c</i>	0.0	0.0	0.0	0.1	2.19(3)
O2	8 <i>b</i>	0.375	0.375	0.375	0.96(2)	5.5(6)
O3	48 <i>f</i>	0.392(2)	0.125	0.125	0.96(2)	5.5(6)

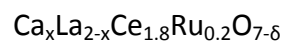
(c) CLCRO02

Atom	Site	x	y	z	Occupancy	100* U_{iso} (Å ²)
La	16 <i>d</i>	0.5	0.5	0.5	0.9	2.18(4)
Ca	16 <i>d</i>	0.5	0.5	0.5	0.1	2.18(4)
Ce	16 <i>c</i>	0.0	0.0	0.0	0.9	2.18(4)
Ru	16 <i>c</i>	0.0	0.0	0.0	0.1	2.18(4)
O2	8 <i>b</i>	0.375	0.375	0.375	1.00(2)	4.6(5)
O3	48 <i>f</i>	0.384(3)	0.125	0.125	1.00(2)	4.6(5)

Table S2 Assignments of Ce3d XPS lines and calculations of Ce⁴⁺/Ce³⁺ of M_xLa_{2-x}_xCe_{1.8}Ru_{0.2}O_{7-δ} (M = Mg, Ca)Mg_xLa_{2-x}Ce_{1.8}Ru_{0.2}O_{7-δ}

Ce ⁴⁺ /Ce ³⁺ contribution	Peak	x = 0.0		x = 0.1		x = 0.2		x = 0.3	
		Position (eV)	Area	Position (eV)	Area	Position (eV)	Area	Position (eV)	Area
IV	V	882.0	8516	881.8	7822	881.9	8603	881.8	8176
III	V'	885.0	2150	885.0	2139	884.9	2352	885.0	2368
IV	V''	888.2	6826	888.0	6941	888.0	6329	888.0	6149
IV	V'''	897.5	8393	897.4	7656	897.5	7241	897.5	7749
IV	U	900.5	5489	900.3	5377	900.5	5173	900.4	4694
III	U'	904.8	2406	904.6	1862	903.7	1845	903.9	1993
IV	U''	907.9	3286	907.3	3441	907.4	3124	907.2	3045
IV	U'''	915.8	6539	915.9	5758	915.9	5422	915.9	5785
[Ce ³⁺]		10.45%		9.76%		10.47%		10.91%	
[Ce ⁴⁺]		89.55%		90.24%		89.53%		89.09%	

continue from the previous page



Cerium contribution	Peak	x = 0.1		x = 0.2	
		Position (eV)	Area	Position (eV)	Area
IV	V	881.9	8994	882.0	7898
III	V'	885.4	4167	885.0	2785
IV	V''	888.5	5714	888.1	5862
IV	V'''	897.6	7140	897.6	7128
IV	U	900.4	5775	900.3	4862
III	U'	903.4	2045	902.6	2162
IV	U''	907.0	3564	906.6	4145
IV	U'''	916.0	5898	916.1	6078
[Ce ³⁺]		14.35%		12.09%	
[Ce ⁴⁺]		85.65%		87.91%	

Table S3. Assignments of Ru3d XPS lines of $M_xLa_{2-x}Ce_{1.8}Ru_{0.2}O_{7-\delta}$ (M = Mg, Ca) $Mg_xLa_{2-x}Ce_{1.8}Ru_{0.2}O_{7-\delta}$

Mg^{2+} dopant (x)	Ru3d3/2		C1s		Ru3d5/2	
	BE (eV)	Area (a.u.)	BE (eV)	Area (a.u.)	BE (eV)	Area (a.u.)
0.0	286.5	1316	284.8	2845	282.4	856
	287.8	384			283.7	656
0.1	286.2	967	284.8	3139	281.9	951
	287.9	556			283.3	480
0.2	286.3	1066	284.8	3243	281.7	1247
	288.1	689			283.4	663
0.3	286.1	1250	284.8	2793	281.8	1052
	287.4	833			283.6	711

 $Ca_xLa_{2-x}Ce_{1.8}Ru_{0.2}O_{7-\delta}$

Ca^{2+} dopant (x)	Ru3d3/2		C1s		Ru3d5/2	
	BE (eV)	Area (a.u.)	BE (eV)	Area (a.u.)	BE (eV)	Area (a.u.)
0.1	286.2	967	284.8	3139	281.9	951
	287.9	556			283.3	480
0.2	286.3	1066	284.8	3243	281.7	1247
	288.1	689			283.4	663

Table S4. The first derivatives of the Ru K-edge XANES spectrum of LCRO, MLCRO03, and CLCRO02 catalysts with reference compounds RuO₂.

Sample	LCRO	MLCRO03	CLCRO02	RuO ₂
Deriv $\mu(E)/e.V$	22129.73	22129.86	22129.73	22128.57

Table S5. Review of catalysts tested for hydrogen production from OSRE/SRE.

Catalyst	T (°C)	C ₂ H ₅ OH/O ₂ /H ₂ O	X _{EtOH} (%)	H ₂ yield (%)	Ref
Ir/CeO ₂	550	1/0.6/1.8	100	55 ^b	1
Ir/CeO ₂	650	1/0.6/1.8	100	65 ^b	2
Rh/CeO ₂ /Al ₂ O ₃	700 ^a (ATR)	1/1.17/3	100	68	3
Ru/Ce _{1-x} M _x O ₂	700 ^a (ATR)	1/1.17/3	100	65	4
Rh/CeO ₂	800 ^a (ATR)	1/1.17/3	100	69	5
Pt/CeZrO ₂	500	1/0.5/2	90	55 ^b	6
CeNiH ₂ O _y	280 ^a (ATR)	1/1.6/3	100	45 ^b	7
NiAl ₂ O ₄ -FeAl ₂ O ₄	600	1/0.5/3	100	65	8
LaNiO ₃	500	1/0.5/3	60	60 ^b	9
Ni/La ₂ O ₃	500	1/0.5/3	100	55 ^b	9
Ni/CeO ₂	650	1/0.8/8.8	75	80 ^c	10
Ni/MgO	650	1/0.8/8.8	75	85 ^c	10
Co/CeO ₂	500	1/0.5/3	100	58 ^b	11
La _{0.9} Ce _{0.1} NiO ₃	500	1/0.5/3	100	50 ^b	12
La _{1-x} Ca _x Fe _{1-x} Co _x O ₃	650	1/0.5/0.33	100	68 ^b	13
La _{0.6} Sr _{0.4} CoO _{3-δ}	600	1/0.5/3	100	70 ^b	14
La ₂ Ce _{1.55} Ni _{0.45} O ₇ /Al ₂ O ₃	500	1/1.17/3	100	53	15
5%Ru/CeO ₂ /Al ₂ O ₃	400	1/1.17/3	96	57	16
5%Ru/La ₂ Ce ₂ O ₇ /Al ₂ O ₃	400	1/1.17/3	97	56	16
La ₂ Ce _{1.8} Ru _{0.2} O ₇ /Al ₂ O ₃	400	1/1.17/3	100	60	16
Li _{0.6} La _{1.4} Ce _{1.8} Ru _{0.2} O ₇ /Al ₂ O ₃	350	1/1.17/3	100	62	17
Li _{0.6} La _{1.4} Ce _{1.8} Ru _{0.2} O ₇ /LZO	350	1/1.17/3	100	58	17
Li _{0.6} La _{1.4} Ce _{1.8} Ru _{0.2} O ₇ /Al ₂ O ₃	372* (ATR)	1/1.17/3	91	41	17
Li _{0.6} La _{1.4} Ce _{1.8} Ru _{0.2} O ₇ /LZO	380* (ATR)	1/1.17/3	90	43	17
Mg _{0.3} La _{1.7} Ce _{1.8} Ru _{0.2} O ₇ /Al ₂ O ₃	350	1/1.17/3	100	61	d
Ca _{0.2} La _{1.8} Ce _{1.8} Ru _{0.2} O ₇ /Al ₂ O ₃	280	1/1.17/3	90	55	d
La ₂ Ce _{1.8} Ru _{0.2} O ₇ /LZO	400	1/1.17/3	100	54	d
Mg _{0.3} La _{1.7} Ce _{1.8} Ru _{0.2} O ₇ /LZO	400	1/1.17/3	100	54	d
Ca _{0.2} La _{1.8} Ce _{1.8} Ru _{0.2} O ₇ /LZO	400	1/1.17/3	100	54	d
Mg _{0.3} La _{1.7} Ce _{1.8} Ru _{0.2} O ₇ /LZO	425* (ATR)	1/1.17/3	94	45	d
Ca _{0.2} La _{1.8} Ce _{1.8} Ru _{0.2} O ₇ /LZO	420* (ATR)	1/1.17/3	100	49	d

^a denote ATR experiments were monitored by a K-type thermocouple on the reaction tube;

^b mole% ; ^c Selectivity(%); ^d this study

Reference

1. F. Wang, W. Cai, H. Provendier, Y. Schuurman, C. Descorme, C. Mirodatos and W. Shen, *Int. J. Hydrogen Energy*, 2011, **36**, 13566-13574.
2. W. Cai, F. Wang, E. Zhan, A. C. Van Veen, C. Mirodatos and W. Shen, *J. Catal.*, 2008, **257**, 96-107.
3. G. A. Deluga, J. R. Salge, L. D. Schmidt and X. E. Verykios, *Science*, 2004, **303**, 993-997.
4. W.-T. Chen, K.-B. Chen, M.-F. Wang, S.-F. Weng, C.-S. Lee and M. C. Lin, *Chem. Commun.*, 2010, **46**, 3286-3288.

5. W.-I. Hsiao, Y.-S. Lin, Y.-C. Chen and C.-S. Lee, *Chem. Phys. Lett.*, 2007, **441**, 294-299.
6. S. M. de Lima, I. O. da Cruz, G. Jacobs, B. H. Davis, L. V. Mattos and F. B. Noronha, *J. Catal.*, 2008, **257**, 356-368.
7. C. Pirez, M. Capron, H. Jobic, F. Dumeignil and L. Jallowiecki-Duhamel, *Angew. Chem. Int. Ed.*, 2011, **50**, 10193-10197.
8. L. Huang, J. Xie, W. Chu, R. Chen, D. Chu and A. T. Hsu, *Catal. Commun.*, 2009, **10**, 502-508.
9. S. M. de Lima, A. M. da Silva, L. O. O. da Costa, J. M. Assaf, G. Jacobs, B. H. Davis, L. V. Mattos and F. B. Noronha, *Appl. Catal., A*, 2010, **377**, 181-190.
10. F. Frusteri, S. Freni, V. Chiodo, S. Donato, G. Bonura and S. Cavallaro, *Int. J. Hydrogen Energy*, 2006, **31**, 2193-2199.
11. S. M. de Lima, A. M. da Silva, L. O. O. da Costa, U. M. Graham, G. Jacobs, B. H. Davis, L. V. Mattos and F. B. Noronha, *J. Catal.*, 2009, **268**, 268-281.
12. S. M. de Lima, A. M. da Silva, L. O. O. da Costa, J. M. Assaf, L. V. Mattos, R. Sarkari, A. Venugopal and F. B. Noronha, *Appl. Catal., B*, 2012, **121-122**, 1-9.
13. Z. Wang, H. Wang and Y. Liu, *RSC Adv.*, 2013, **3**, 10027-10036.
14. M. Morales and M. Segarra, *Appl. Catal., A*, 2015, **502**, 305-311.
15. S.-F. Weng, H.-C. Hsieh and C.-S. Lee, *Int. J. Hydrogen Energy*, 2017, **42**, 2849-2860.
16. S.-F. Weng, Y.-H. Wang and C.-S. Lee, *Appl. Catal., B*, 2013, **134-135**, 359-366.
17. H.-C. Hsieh, Y.-C. Chang, P.-W. Tsai, Y.-Y. Lin, Y.-C. Chuang, H.-S. Sheu and C.-S. Lee, *Catal. Sci. Technol.*, 2019, **9**, 1406-1419.