RSC Advances

ELECTRONIC SUPPLEMENTARY INFORMATION

A one-pot synthesis of N^2 ,6-diaryl-5,6-dihydro-1,3,5-triazine-2,4-diamines and systematic evaluation of their ability to host ethanol in crystals

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Parameter		Parameter		Parameter		Parameter	
1b – molecule 1		1b – molecule 2		1f – molecule 1		1f – molecule 2	
C1-N1	1.447(6)	C17–N6	1.471(6)	C1-N1	1.448(6)	C16-N6	1.463(6)
C1-N3	1.470(6)	C17–N8	1.452(6)	C1-N3	1.474(6)	C16–N8	1.452(6)
C2-N1	1.305(6)	C18–N6	1.312(7	C2-N1	1.301(7)	C17–N6	1.297(7)
C2-N2	1.373(6)	C18–N7	1.366(6)	C2-N2	1.379(6)	C17-N7	1.388(6)
C3-N2	1.339(6)	C19–N7	1.340(6)	C3–N2	1.338(6)	C18–N7	1.318(6)
C3-N3	1.334(6)	C19–N8	1.342(7)	C3–N3	1.324(7)	C18–N8	1.336(7)
C1-N1-C2	115.1(4)	C17-N6-C18	113.3(4)	C1-N1-C2	115.6(4)	C16-N6-C18	114.6(4)
C2-N2-C3	114.0(4)	C18–N7–C19	115.1(4)	C2-N2-C3	113.6(4)	C17–N7–C18	114.9(4)
C1-N3-C3	117.6(4)	C17-N8-C19	117.8(4)	C1-N3-C3	118.4(4)	C16-N8-C18	118.2(4)
N1-C1-N3	110.5(4)	N6-C17-N8	110.0(4)	N1-C1-N3	110.5(4)	N6-C16-N8	111.0(4)
N1-C2-N2	127.8(4)	N6-C18-N7	127.4(4)	N1-C2-N2	128.1(4)	N6-C17-N7	126.9(4)
N2-C3-N3	122.5(4)	N7-C19-N8	121.0(4)	N2-C3-N3	123.3(4)	N7-C18-N8	122.6(4)

Table S1. Key geometric parameters (Å, °) for 1b.EtOH and 1f.EtOH

Table S2. Least-squares plane data for **1b**.EtOH and **1f**.EtOH.

1b	
N1-molecule	
r.m.s. deviation (N1-N3, C2 and C3)	0.042
deviation of C1	0.477(6)
Dihedral angles	
(N1-N3, C1-C3)/(C4-C9)	89.07(13)
(N1-N3, C1-C3)/(C11-C16)	19.2(3)
(C4-C9)/(C11-C16)	70.43(15)
N6-molecule	
r.m.s. deviation (N6-N8, C17 and C8)	0.043
deviation of C16	0.529(6)
Dihedral angles	
(N6-N8, C16-C18)/(C20-C25)	74.56(17)
(N6-N8, C16-C18)/(C27-C32)	13.6(3)
(C20-C25)/(C27-C32)	63.38(17)
1f	
N1-molecule	
r.m.s. deviation (N1-N3, C2 and C3)	0.030
deviation of C1	0.441(6)
Dihedral angles	
(N1-N3, C1-C3)/(C4-C9)	88.31(16)
(N1-N3, C1-C3)/(C10-C15)	18.3(2)
(C4-C9)/(C10-C15)	75.81(17)
N6-molecule	
r.m.s. deviation (N6-N8, C17 and C8)	0.032
deviation of C16	0.467(6)
Dihedral angles	
(N6-N8, C16-C18)/(C19-C24)	72.61(18)
(N6-N8, C16-C18)/(C25-C30)	19.3(3)
(C19-C24)/(C25-C30)	57.61(18)

А	Н	В	A–H	H […] B	A B	A–H […] B	Symmetry
							operation
N3	H3n	N7	0.88(4)	2.08(4)	2.932(6)	161(4)	$1-x, \frac{1}{2}+y, 1-z$
N8	H8n	N2	0.88(4)	2.08(3)	2.929(6)	161(5)	$1-x, \frac{1}{2}+y, 1-z$
N4	H4na	Cg(C27-C32)	0.88(4)	2.59(5)	137(4)	3.290(5)	<i>x</i> , <i>y</i> , 1+ <i>z</i>
N9	H9na	Cg(C11-C16)	0.88(5)	2.72(5)	134(4)	3.389(5)	<i>X</i> , <i>Y</i> , <i>Z</i>
C22	H22	Cg(C11-C16)	0.95	2.76	3.565(6)	143	1- <i>x</i> , 1/2+ <i>y</i> , 1- <i>z</i>
03	H3o	N1	0.88(5)	1.86(5)	2.712(5)	162(5)	<i>X</i> , <i>Y</i> , <i>Z</i>
O4	H4o	N6	0.87(5)	1.88(5)	2.732(5)	167(5)	<i>x</i> , <i>y</i> , 1+ <i>z</i>
N5	H5n	O3	0.88(4)	2.05(4)	2.916(6)	167(5)	$1-x, -\frac{1}{2}+y, 1-z$
N9	H9Nb	O3	0.88(5)	2.03(5)	2.861(6)	158(6)	<i>X</i> , <i>Y</i> , <i>Z</i>
N4	H4nb	O4	0.88(5)	2.02(5)	2.883(6)	170(5)	<i>X</i> , <i>Y</i> , <i>Z</i>
N10	H10n	O4	0.88(4)	2.07(4)	2.912(6)	163(5)	1- <i>x</i> , -½+ <i>y</i> , 1- <i>z</i>
C15	H15	O2	0.95	2.44	3.275(7)	147	-1+ <i>x</i> , <i>y</i> , <i>z</i>
C29	H29	O1	0.95	2.42	3.342(7)	165	$1-x, \frac{1}{2}+y, -z$
C33	H33a	Cl2	0.99	2.80	3.551(7)	133	$1-x, \frac{1}{2}+y, -z$

Table S3 Geometric parameters (Å, $^{\circ}$) characterizing the intermolecular interactionsoccurring in the crystal of 1b.EtOH

A	Н	В	A–H	H B	A B	A–H […] B	Symmetry
							operation
N3	H3n	N7	0.88(3)	2.04(3)	2.911(6)	168(4)	$1-x, -\frac{1}{2}+y, 1-z$
N8	H8n	N2	0.88(4)	2.10(4)	2.943(6)	161(5)	$1-x, -\frac{1}{2}+y, 1-z$
N4	H4na	Cg(C25-C30)	0.89(4)	2.58(6)	136(4)	3.279(5)	<i>x</i> , <i>y</i> , 1+ <i>z</i>
N9	H9na	Cg(C10-C15)	0.88(3)	2.65(6)	127(5)	3.257(6)	х, у, Z
C22	H22	Cg(C4-C9)	0.95	2.75	3.611(6)	152	<i>x</i> , <i>y</i> , 1+ <i>z</i>
01	H1o	N1	0.84(5)	1.89(5)	2.728(5)	175(5)	<i>X</i> , <i>Y</i> , <i>Z</i>
02	H2o	N6	0.83(5)	1.88(5)	2.708(5)	173(7)	х, у, Z
N5	H5n	O1	0.88(5)	2.02(5)	2.890(5)	172(5)	$1-x, \frac{1}{2}+y, 1-z$
N9	H9nb	O1	0.88(5)	2.00(5)	2.878(6)	177(6)	х, у, Z
N4	H4nb	O2	0.87(5)	2.02(5)	2.885(6)	173(4)	<i>x</i> , <i>y</i> , 1+ <i>z</i>
N10	H10n	O2	0.88(5)	2.03(5)	2.894(6)	170(5)	$1-x, \frac{1}{2}+y, z$
C7	H7	Br1	0.95	2.92	3.682(7)	138	- <i>x</i> , - ¹ / ₂ + <i>y</i> , 1- <i>z</i>

Table S4 Geometric parameters (Å, $^{\circ}$) characterizing the intermolecular interactionsoccurring in the crystal of 1f.EtOH



Figure S1. The molecular structures of the two independent molecules of **1b** comprising the asymmetric unit of **1b**.EtOH, showing atom labelling scheme and displacement ellipsoids at the 70% probability level.



Figure S2. The molecular structures of the two independent molecules of **1e** comprising the asymmetric unit of **1e**.EtOH, showing atom labelling scheme and displacement ellipsoids at the 70% probability level.



(b)

Figure S3. Molecular packing in **1b**.EtOH: (a) view of the unit cell contents in projection down the *a*-axis highlighting the stacking of layers and (b) a plan view of the layer (non-participating hydrogen atoms are omitted for reasons of clarity). The N–H^{...}N, N–H^{...} π , C–H^{...} π and C–H^{...}O interactions are shown as blue, purple, pink and plum dashed lines, respectively. The hydrogen-bonding involving the ethanol molecules are shown as orange dashed lines.



Figure S4. Molecular packing in **1f**.EtOH: view of the unit cell contents in projection down the *b*-axis highlighting the stacking of layers. The N–H^{...}N, N–H^{...} π , C–H^{...} π and C–H^{...}Br interactions are shown as blue, purple, pink and green dashed lines, respectively. Hydrogenbonding involving the ethanol molecules are shown as orange dashed lines.



Figure S5. Traces for the STA conducted on (a) 1b, (b) 1e and (c) 1f.

Temp. range Expt'l calcd $\Delta H_{desolvation}$ (°C) wght loss (%) wght loss (%) (kJ/mol) 90 - 162 1b.EtOH 88.8 86.6 46.7 94 - 146 87.2 86.7 72.6 1e.EtOH 100 - 156 88.2 57.5 1f.EtOH 88.8

Data for the first weight loss (first exothermic process):

Data for the second exothermic process:

	Temp. range	Peak temp.	$\Delta H_{melting}$
	(°C)	(°C)	(kJ/mol)
1e.EtOH	204 - 227	219.4	29.2
1f.EtOH	211 - 232	225.9	27.6



(b)

Figure S6. Experimental powder X-ray diffraction patterns for (a) **1e** and (b) **1f**. The red trace is of the original materials, *i.e.* **1e**.EtOH and **1f**.EtOH. The blue traces are of the desolvated materials as per the STA analyses, showing that the crystal structures are different. The green traces are measurements taken after exposing the desolvated materials to EtOH vapour overnight, showing no reversion to the original structures of **1e**.EtOH and **1f**.EtOH had occurred.

Copies of NMR spectra of prepared products **1a-k** · **EtOH** and **1a**



N²,6-Diphenyl-5,6-dihydro-1,3,5-triazine-2,4-diamine ethanolate (1a · EtOH)



N^{2} , 6-Diphenyl-5, 6-dihydro-1, 3, 5-triazine-2, 4-diamine ethanolate (1a · EtOH)



N²-(4-Chlorophenyl)-6-(4-methoxyphenyl)-5,6-dihydro-1,3,5-triazine-2,4-diamine ethanolate (1b · EtOH)



N²-(4-Chlorophenyl)-6-(4-methoxyphenyl)-5,6-dihydro-1,3,5-triazine-2,4-diamine ethanolate (1b · EtOH)



N²-(4-Fluorophenyl)-6-phenyl-5,6-dihydro-1,3,5-triazine-2,4-diamine ethanolate (1c · EtOH)



N^2 -(4-Fluorophenyl)-6-phenyl-5,6-dihydro-1,3,5-triazine-2,4-diamine ethanolate (1c · EtOH)



N²-(3-Chlorophenyl)-6-phenyl-5,6-dihydro-1,3,5-triazine-2,4-diamine ethanolate (1d · EtOH)



N²-(3-Chlorophenyl)-6-phenyl-5,6-dihydro-1,3,5-triazine-2,4-diamine ethanolate (1d · EtOH)



N²-(4-Chlorophenyl)-6-phenyl-5,6-dihydro-1,3,5-triazine-2,4-diamine ethanolate (1e · EtOH)



N²-(4-Chlorophenyl)-6-phenyl-5,6-dihydro-1,3,5-triazine-2,4-diamine ethanolate (1e · EtOH)



N²-(4-Bromophenyl)-6-phenyl-5,6-dihydro-1,3,5-triazine-2,4-diamine ethanolate (1f · EtOH)



N²-(4-Bromophenyl)-6-phenyl-5,6-dihydro-1,3,5-triazine-2,4-diamine ethanolate (1f · EtOH)



N²-(4-Methoxyphenyl)-6-phenyl-5,6-dihydro-1,3,5-triazine-2,4-diamine ethanolate (1g · EtOH)



N²-(4-Methoxyphenyl)-6-phenyl-5,6-dihydro-1,3,5-triazine-2,4-diamine ethanolate (1g · EtOH)

000 00777 4 4,0,0,0,0,0,0,0,0,0,0,0,0,0,4,4,4 4 M N N 4 4 4 H O O O . BRUKER 0 てててママシシシシシシシシシシシシシシシシシシシシシントレイトレイトレート Current Data Parameters JX0007 NAME EXPNO 1 PROCNO 1 1.06 (t, J=6.99 Hz, 3 H) F2 - Acquisition Parameters 3.45 (q, J=6.99 Hz, 2 H) Date 20180513 Me Time 14.36 6.56 (d, J=7.35 Hz, 1 H) INSTRUM FOURIER300 6.98 (dd, J=7.79, 7.79 Hz, 1 H) PROBHD 5 mm DUL 13C-1 HN 7.27 (t, J=7.05 Hz, 1 H) PULPROG zg30 TD 65536 7.36 (dd, J=7.29, 7.29 Hz, 2 H) SOLVENT DMSO H_2N N N H 7.42 (dd, J=1.16, 8.06 Hz, 2 H) 16 NS DS 2 7.60 (d, J=8.13 Hz, 1 H) SWH 6103.516 Hz FIDRES 0.093132 Hz AQ 5.3687091 sec 31.623 RG DW 81.920 usec DE 6.50 usec TE 300.0 K D1 1.00000000 sec TD0 1 EtOH EtOH ====== CHANNEL f1 ======= 300.1618536 MHz SF01 NUC1 1H P1 13.50 usec 7.6 7.2 7.0 6.8 6.6 ppm 7.4 9.30000019 W PLW1 F2 - Processing parameters 0.98 1.03 1.99 1.99 8 <u>[5</u>] SI 65536 300.1600025 MHz SF WDW EM SSB 0 LB 0.30 Hz GB 0 PC 1.00 ******* ****** 11 10 9 7 6 5 3 2 8 1 0 ppm <u>;</u> 2.90 <u>1.99</u> **5**.03 88885 0.64 86

N²-(3-Methylphenyl)-6-phenyl-5,6-dihydro-1,3,5-triazine-2,4-diamine ethanolate (1h · EtOH)

N²-(3-Methylphenyl)-6-phenyl-5,6-dihydro-1,3,5-triazine-2,4-diamine ethanolate (1h · EtOH)





N²-(4-Methylphenyl)-6-phenyl-5,6-dihydro-1,3,5-triazine-2,4-diamine ethanolate (1i · EtOH)



N²-(4-Methylphenyl)-6-phenyl-5,6-dihydro-1,3,5-triazine-2,4-diamine ethanolate (1i · EtOH)



6-Phenyl-N²-(4-(trifluoromethoxy)phenyl)-5,6-dihydro-1,3,5-triazine-2,4-diamine ethanolate (1j · EtOH)



6-Phenyl-N²-(4-(trifluoromethoxy)phenyl)-5,6-dihydro-1,3,5-triazine-2,4-diamine ethanolate (1j · EtOH)



6-(4-Chlorophenyl)-N²-phenyl-5,6-dihydro-1,3,5-triazine-2,4-diamine ethanolate (1k · EtOH)



6-(4-Chlorophenyl)-N²-phenyl-5,6-dihydro-1,3,5-triazine-2,4-diamine ethanolate (1k · EtOH)



