

S1 Figure 1: (a) Fourier Transform Infrared Spectra of the Plant, (b) UV-Vis Spectra of

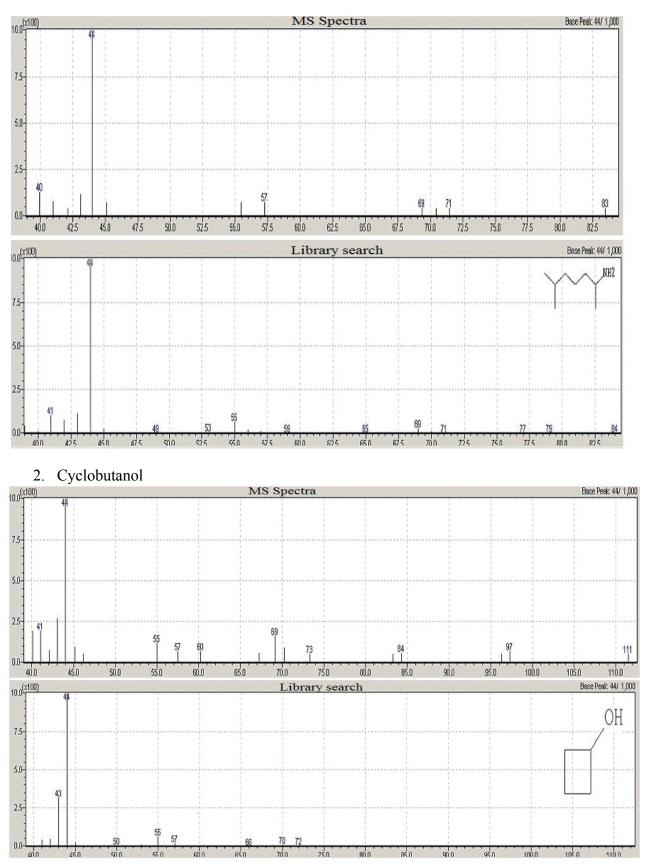
the Plant Leaf Extract, And (c) GC Chromatogram of the Methanolic Extract of *E. Cognata* Leaves

S2 .Table 1:	Phytochemicals	of methanolic leaf	extract of E.	Cognata Individu	alized by GC-MS.

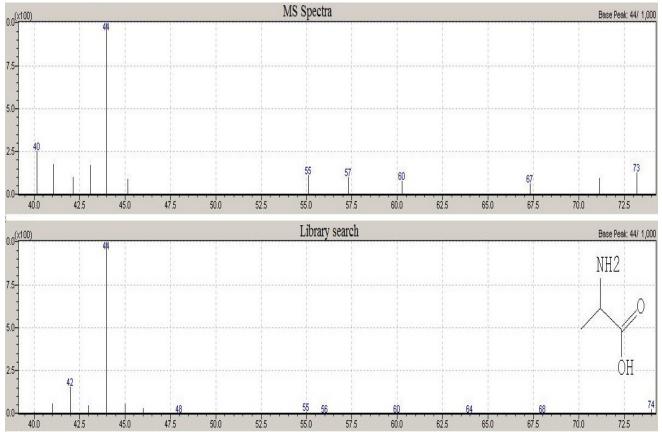
R. Time	Area (%)	Height (%)	Mol. weight	Chemical Formula	Name
12.64	3.70	7.49	129	C8H19N	Octodrine
17.28	0.88	1.41	72.11	C ₄ H ₈ O	cyclobutanol
21.4	0.79	0.98	171	C11H25N	1-Methyldecylamine
26.78	0.98	1.51	89	C3H7NO2	d-Alanine
28.38	12.92	24.95	172	C10H20O2	Decanoic acid
31.19	0.65	0.79	125	C8H15N	Azabicyclo[3.2.2]nonane
31.39	16.95	29.44	129	C8H19N	Octodrine
31.61	7.2	7.10	127	C8H17N	Cyclohexylethylamine

S3 MS Spectra of main identified Compounds

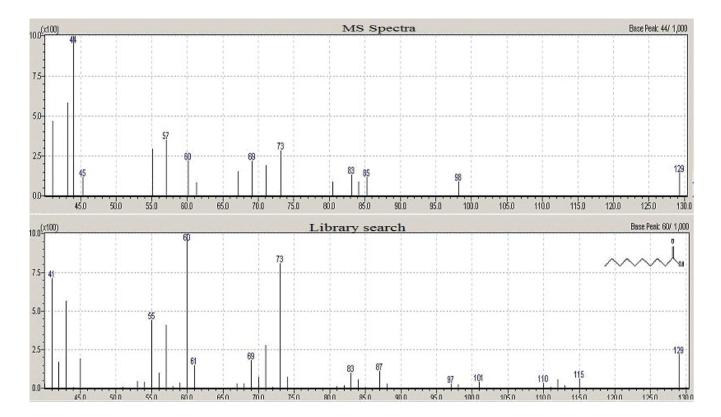
1. Octodrine



3. d-Alanine



4. Decanoic acid



Sol Gel synthesis of Nanomaterial

The materials used in sol–gel method consisted of Molybdenum(II) acetate $(Mo_2(O_2CCH_3)_4)$, Cobalt(II) acetate tetrahydrate $(C_4H_6CoO_4.4H_2O)$, and ethylene glycol $(C_2H_6O_2)$. Absolute ethanol (C_2H_5OH) was used as solvent and Citric acid $(C_6H_8O_7)$ as gelling agent. 20 mM solution of each precursor salt $(Mo_2(O_2CCH_3)_4, C_4H_6CoO_4.4H_2O)$ and 1 molar citric acid was dissolved separately in different beakers, each containing 50 ml of the solvent ethanol. To ensure solution homogeneity, the solutions were continuously stirred for 30 min using a magnetic stirrer at a speed of 300 rpm and slowly heated up to 45 °C to ensure complete material dissolution. After that, 0.5 ml of ethylene glycol was added to the solutions. Then, citric acid was added to each precursor salt solution under continuous stirring and heating at a temperature of 60 °C for 1 h to complete the reaction. The final solution was dried in oven at 95 °C for 10 h. After drying, the resulting product was treated at a temperature of 450 °C to remove all the remaining organic compounds and to get pure nanomaterial.