

## Supplementary material

### Coating-free TiO<sub>2</sub>@β-SiC alveolar foams as ready-to-use composite photocatalyst with tunable adsorption properties for water treatment

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**Figure S1.** (A) Dark adsorption experiments performed on the TiO<sub>2</sub>@β-SiC composite foams with different TiO<sub>2</sub> wt.% contents and calcined at 600°C ; (B) Diuron photodegradation kinetics obtained on those TiO<sub>2</sub>@β-SiC composite foams, with the corresponding apparent kinetic rate constant (pseudo first order reaction rate).

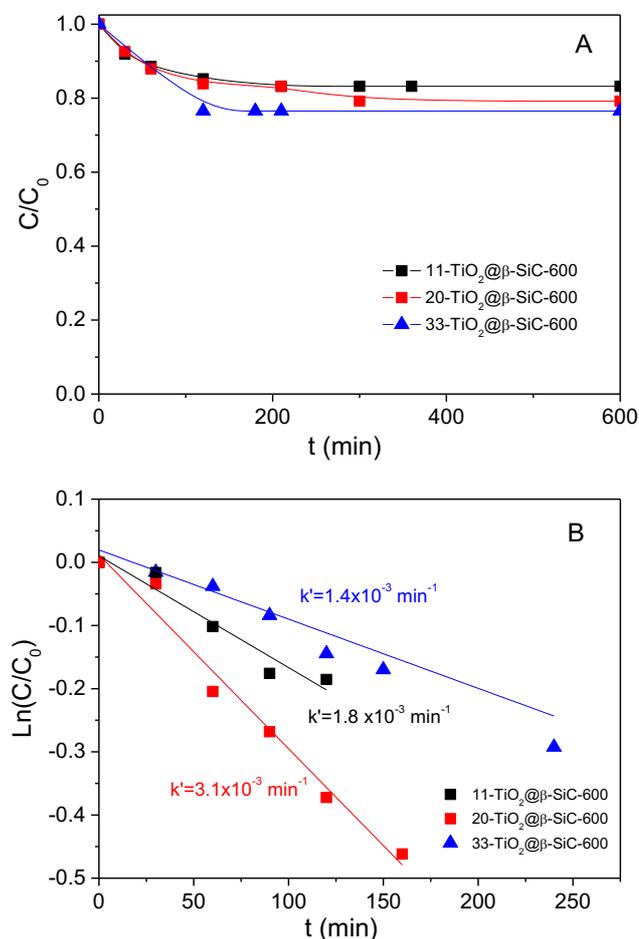
**Figure S2.** Spectral distribution of the simulated solar light.

**Figure S3.** Calibration curve expressing the influence of the TiO<sub>2</sub> concentration on the apparent rate constant for the chlorophenol degradation (pseudo first order reaction). [4-CP]<sub>0</sub> = 20 mg/L; T=25°C; UV-A irradiance of 60 W m<sup>-2</sup>.

**Figure S4.** A) Dark adsorption experiments performed on TiO<sub>2</sub>@β-SiC composite foams varying in terms of residual carbon content for different initial Diuron concentrations. B) Amount of diuron adsorbed as a function of the residual carbon content in wt.% for different initial Diuron concentrations.

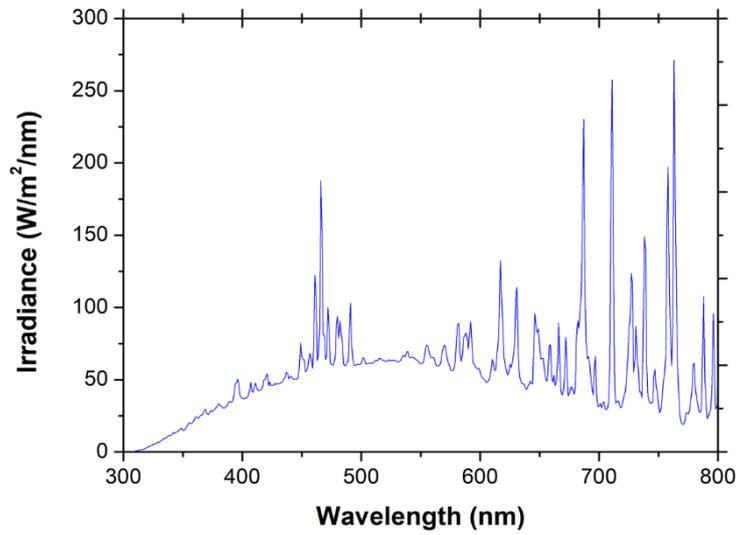
# Figure S1

For a preliminary study,  $\text{TiO}_2@ \beta\text{-SiC}$  composite foams were prepared with a  $\text{TiO}_2$  content of 11 wt.%, 20 wt.% and 33 wt.% on the  $\beta\text{-SiC}$  skeleton foam during the step 3 of the sequential multi-step carburization SMS method. The  $\text{TiO}_2@ \beta\text{-SiC}$  composite foams obtained after final calcination at  $600^\circ\text{C}$  displayed similar Diuron adsorption behavior in the dark independently of the  $\text{TiO}_2$  content, as depicted in Fig. S1A. By contrast, Fig. S1B evidenced the superiority of the composite foam with a 20 wt.%  $\text{TiO}_2$  content in terms of Diuron removal, as evidenced by the higher apparent kinetic rate constant for Diuron degradation, calculated at  $3.1 \cdot 10^{-3} \text{ min}^{-1}$ . Therefore, the main part of the work was further performed on composite foams obtained by impregnating the  $\beta\text{-SiC}$  skeleton foam with  $\text{TiO}_2$  at a 20 wt.% content.



**Figure S1 (A)** Dark adsorption experiments performed on the  $\text{TiO}_2@ \beta\text{-SiC}$  composite foams with different  $\text{TiO}_2$  wt.% contents and calcined at  $600^\circ\text{C}$ ; **(B)** Diuron photodegradation kinetics obtained on those  $\text{TiO}_2@ \beta\text{-SiC}$  composite foams, with the corresponding apparent kinetic rate constant (pseudo first order reaction rate).

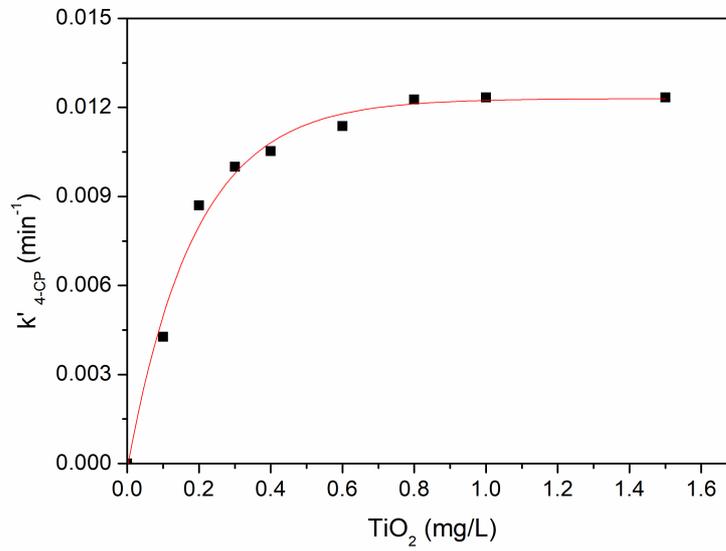
## Figure S2



**Figure S2.** Spectral distribution of the simulated solar light.

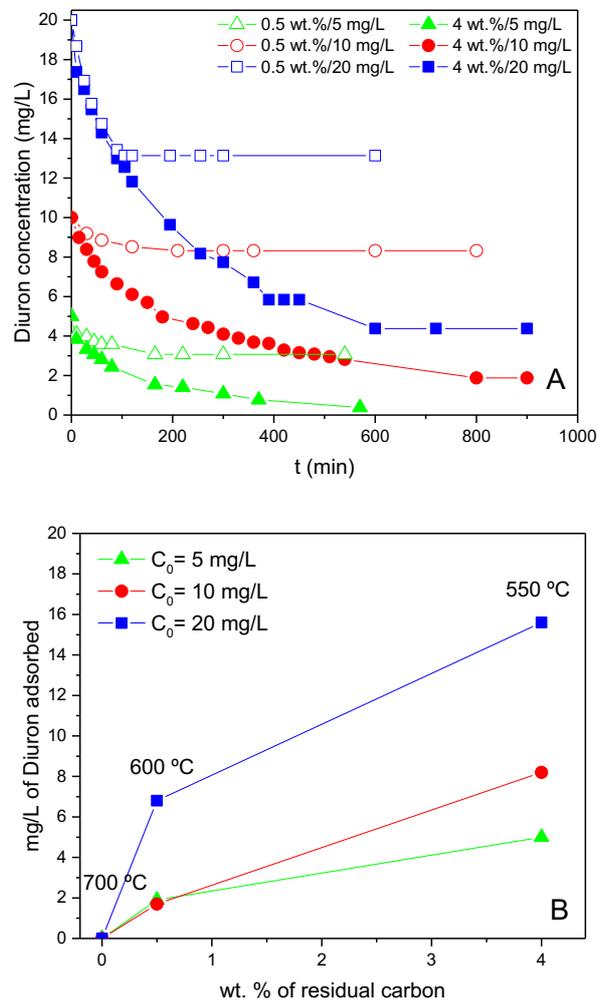
Measurements have been performed using a wideband RPS900-W rapid portable spectroradiometer from International Light Technology.

**Figure S3**



**Figure S3.** Calibration curve expressing the influence of the TiO<sub>2</sub> concentration on the apparent rate constant for the chlorophenol degradation (pseudo first order reaction). [4-CP]<sub>0</sub> = 20 mg/L; T = 25°C; UV-A irradiance of 60 W m<sup>-2</sup>.

# Figure S4



**Figure S4. A)** Dark adsorption experiments performed on  $\text{TiO}_2@ \beta\text{-SiC}$  composite foams varying in terms of residual carbon content for different initial Diuron concentrations. **B)** Amount of Diuron adsorbed as a function of the residual carbon content in wt.% for different initial Diuron concentrations.