

Electronic Supporting Information

Circularly Polarised Luminescence (CPL) Control of Oligopeptide-Eu(III)

Hybridized Luminophores by interaction with peptide side chains

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Experimental Section

Reagents and chemicals

For spectroscopic measurements, spectroscopic-grade CHCl₃ (Dojindo [Kumamoto, Japan], Spectrosol) was used. Eu(III)(hfa)₃(H₂O)₂ was prepared according to a previously reported method.^[1]

Syntheses of chiral peptide-naphthalene ligands 1-4

Fmoc deprotection was carried out with 20% piperidine in DMF (for 7 min at room temperature). After six times washing with DMF, each amino acid was coupled using HBTU/NMM reagents with reaction time of 40 min per coupling at room temperature. No capping step was carried out. After the *N*-terminal Fmoc group was deprotected, the resin was washed with DCM and was treated with 95% TFA, 2.5% water, 2.5% TIS for 1.5 hours at room temperature. The crude peptides were analysed and purified by reversed phase high-pressure liquid chromatography (RP-HPLC) on C18 column with buffer A. 0.1% TFA in water; buffer B, acetonitrile and monitoring at 280 nm with a gradient of 0-100% B gradient over 20 min. The purified peptides were identified with MALDI-TOF mass spectroscopy (matrix, α -CHCA) (Figures S1-S4) and RP-HPLC.

The yields of peptides were 77% (9.8 mg; **L-1**), 67% (10.4 mg; **L-2**), 63% (11.7 mg; **L-3**), 59% (12.6 mg; **L-4**), 82% (10.4 mg; **D-1**), 69% (10.8 mg; **D-2**), 57% (10.5 mg; **D-3**), and 56% (12.0 mg; **D-4**).

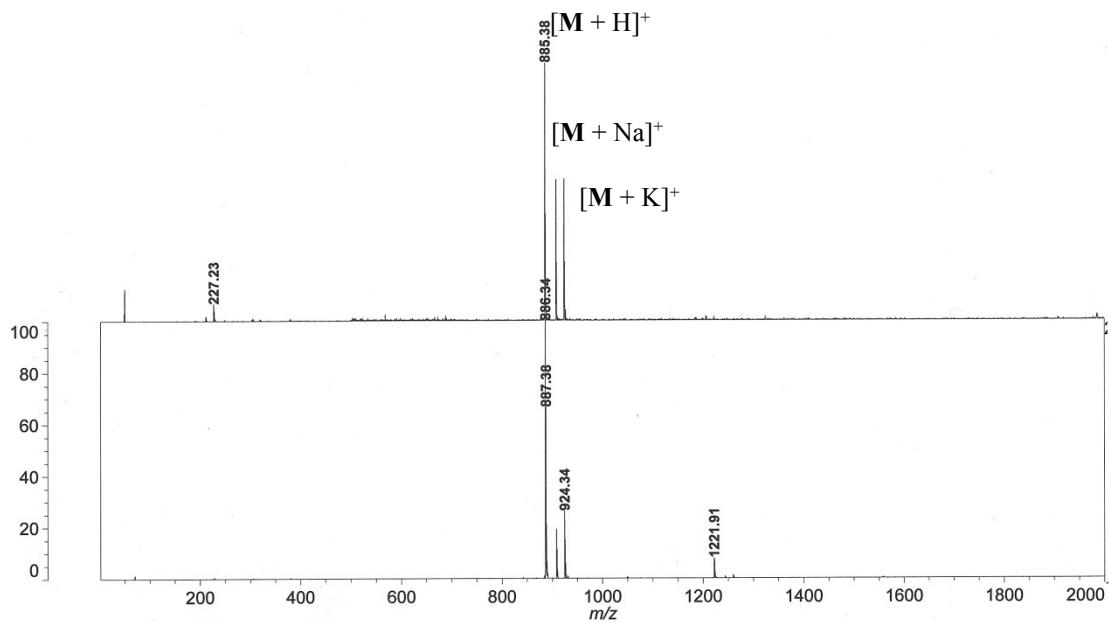


Fig. S1 MALDI-TOF mass spectra of **L-1** (lower) and **D-1** (upper). An α -CHCA was used as a matrix. **L-1**; calcd. *m/z* 885.50 for $[M+H]^+$, obsd. *m/z* 886.34, **D-1**; calcd. *m/z* 885.50 for $[M+H]^+$, obsd. *m/z* 885.38.

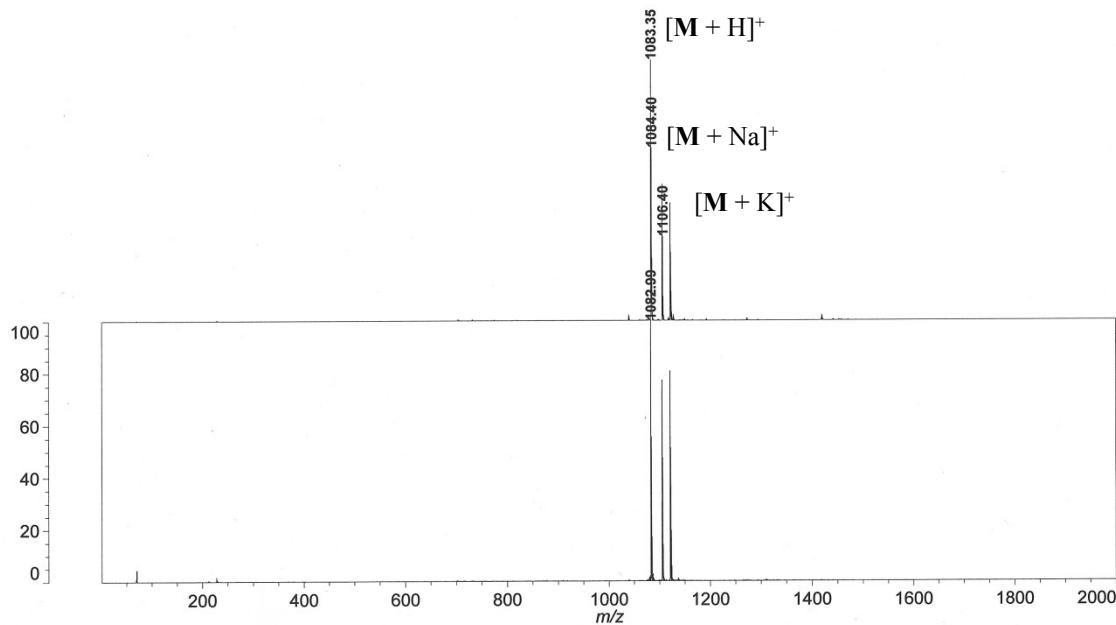


Fig. S2 MALDI-TOF mass spectra of **L-2** (lower) and **D-2** (upper). An α -CHCA was used as a matrix. **L-2**; calcd. *m/z* 1082.58 for $[M+H]^+$, obsd. *m/z* 1082.99, **D-2**; calcd. *m/z* 1082.58 for $[M+H]^+$, obsd. *m/z* 1083.55.

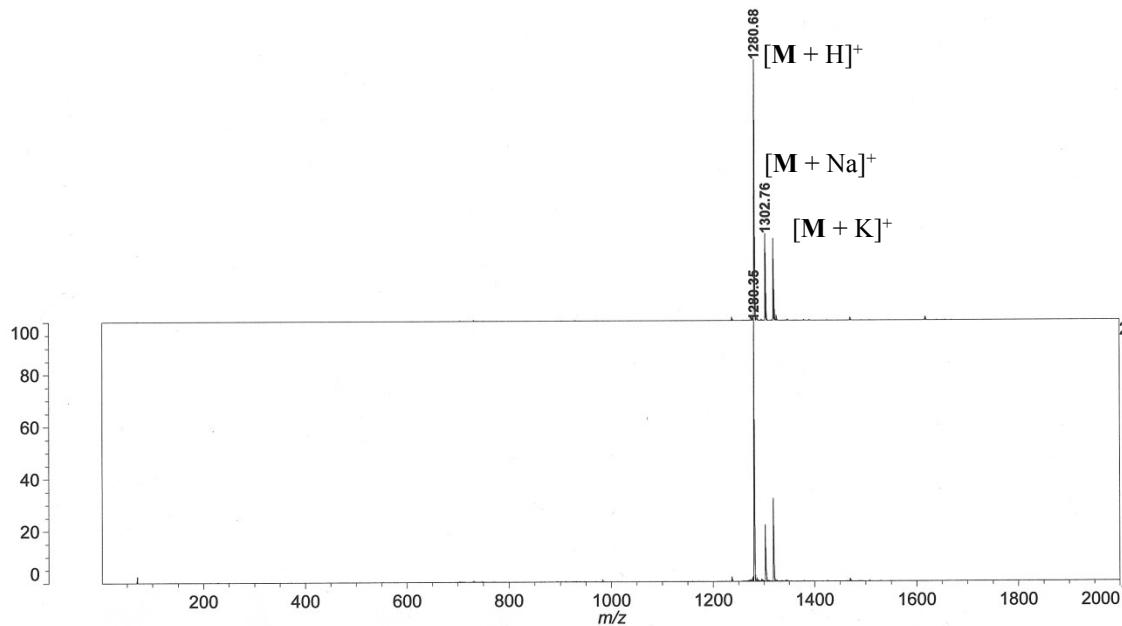


Fig. S3 MALDI-TOF mass spectra of **L-3** (lower) and **D-3** (upper). An α -CHCA was used as a matrix. **L-3**; calcd. m/z 1279.67 for $[\text{M} + \text{H}]^+$, obsd. m/z 1280.35, **D-3**; calcd. m/z 1279.67 for $[\text{M} + \text{H}]^+$, obsd. m/z 1280.68.

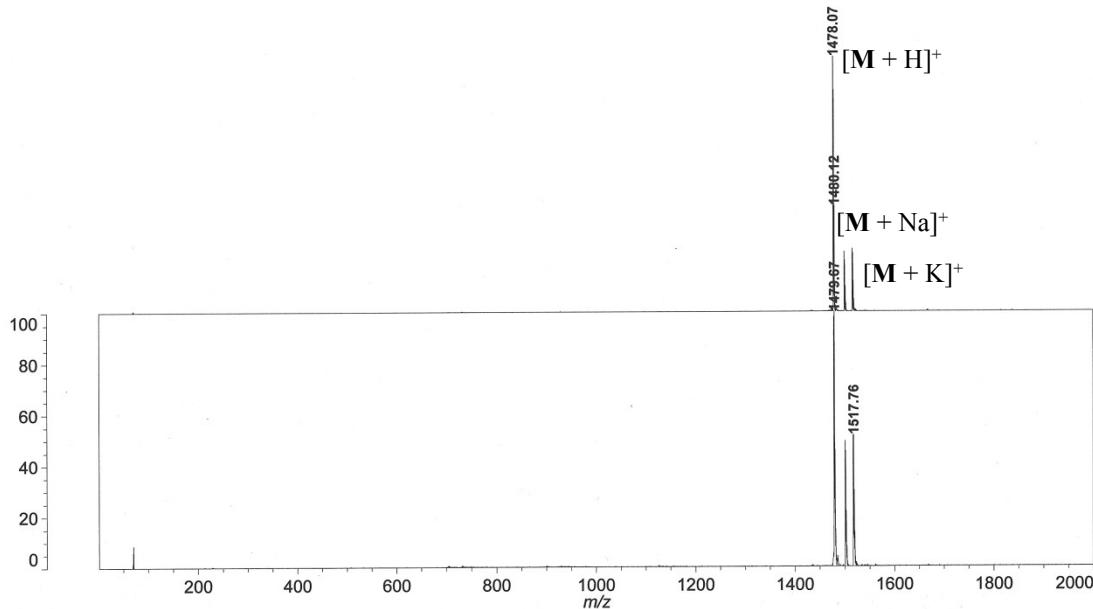


Fig. S4 MALDI-TOF mass spectra of **L-4** (lower) and **D-4** (upper). An α -CHCA was used as a matrix. **L-4**; calcd. m/z 1476.75 for $[\text{M} + \text{H}]^+$, obsd. m/z 1478.60, **D-4**; calcd. m/z 1476.75 for $[\text{M} + \text{H}]^+$, obsd. m/z 1478.07.

Measurement of absolute unpolarised photoluminescence (PL) quantum yield

The absolute unpolarised photoluminescence (PL) quantum yields were measured using an Absolute PL Quantum Yield Measurement System (C9920-02, Hamamatsu Photonics [Hamamatsu, Japan]) in air at room temperature. The excitation wavelength in CHCl₃ (path length 10 mm) was 300 nm. Spectroscopic-grade CHCl₃ (Dojindo [Kumamoto, Japan], Spectrosol) was used for spectroscopic measurements.

Measurement of photoluminescence and CPL spectra

The CPL spectra were obtained with a JASCO (Hachioji, Japan) CPL-300 spectrofluoropolarimeter at room temperature. A scattering angle of 0° was used for the excitation of non-polarised monochromatic incident light with a bandwidth of 10 nm and a bandwidth for emission of 10 nm. Scanning speed was 50 nm per min and the time constant of PMT was 8 s. The CPL and PL spectra were smoothed by 2 accumulation without any numerical smoothing. For spectroscopic measurements, spectroscopic-grade CHCl₃ (Dojindo [Kumamoto, Japan], Spectrosol) were used. The excitation wavelength in CHCl₃ solution was 300 nm (path length 10 mm).

Measurement of circular dichroism (CD) and UV-Vis absorption spectra

The circular dichroism (CD) and UV-Vis absorption spectra of the compounds in CHCl₃ solution states were measured using a JASCO J-820 spectropolarimeter at room temperature. The path length in solution measurement was 1 mm. Spectroscopic-grade CHCl₃ solutions (Dojindo [Kumamoto, Japan], Spectrosol) was used for spectroscopic measurements.

Measurement of MASS spectra

1. Mass spectrometry

Mass spectra were taken with a LCMS-IT-TOF mass spectrometer (Shimadzu Co. [Kyoto, Japan]) under the following conditions: ionization method, electrospray ionization (ESI); solvent, acetonitrile; mass range, *m/z* 100 – 4000; mode, positive; spray voltage, 4.5 kV (positive mode), or -3.5 kV (negative mode); nebulizer gas flow rate; 1.5 L min⁻¹; CDL temperature, 200 °C; heat block temperature, 200 °C; ion source pressure, 80 Pa; ion trap pressure, 1.8×10⁻² Pa; TOF pressure, 1.4×10⁻⁴ Pa; resolution, >10,000. Calibration and

tuning of the instrument were carried out using an available standard solution of trifluoroacetic acid (Shimadzu Co. [Kyoto, Japan]).

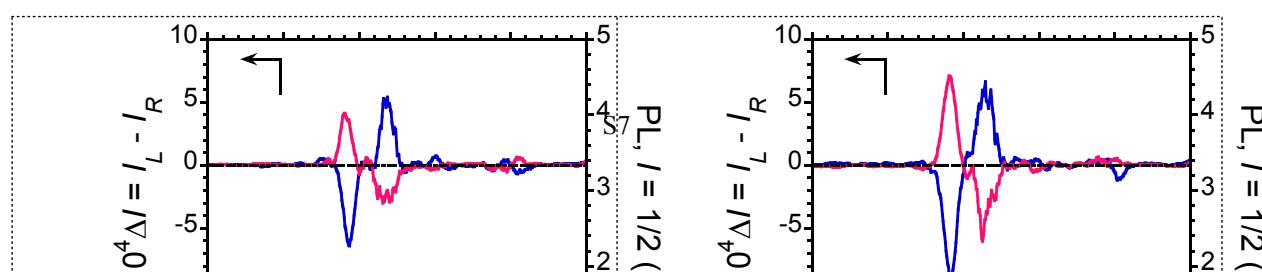
Measurement of NMR spectra

^{19}F -NMR spectra were taken with a JEOL ECA 600 spectrometer. Trifluoroacetic acid in chloroform-*d* was used as the external reference ($\delta = -76.5$ ppm).

NMR spectra were recorded on a JNM ECA-600 spectrometer (JEOL [Hachioji, Japan]). Samples were dissolved in chloroform-*d*. ^{19}F -NMR spectra were measured in the proton decoupling mode. Trifluoroacetic acid was used as an external standard ($\delta = -76.5$ ppm in chloroform-*d*) in ^{19}F -NMR experiments.

References:

- [1] N. Hara, M. Okazaki, M. Shizuma, S; Marumoto, N. Tajima, M. Fujiki and Y. Imai, *ChemistrySelect*, 2017, **2**, 10317.



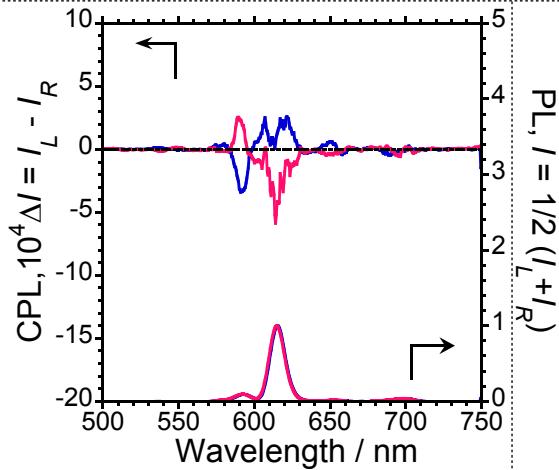


Fig. S5c L- and D-1/Eu(hfa)₃ = 1/2.

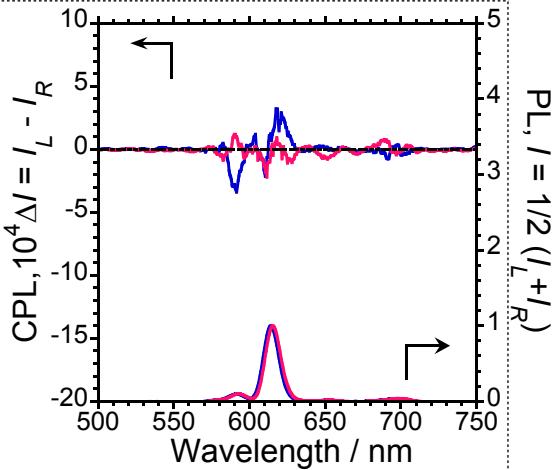


Fig. S5d L- and D-1/Eu(hfa)₃ = 1/3.

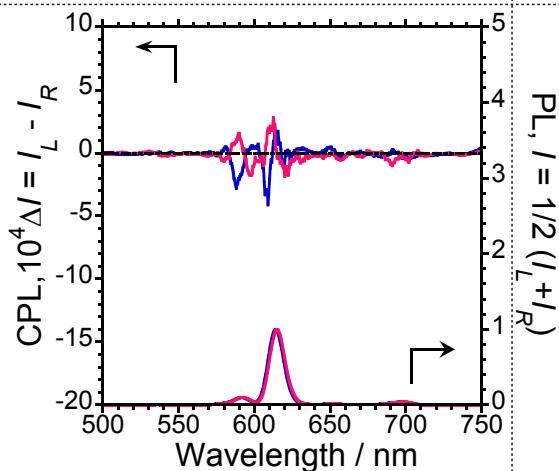


Fig. S5e L- and D-1/Eu(hfa)₃ = 1/4.

Fig. S5 CPL and PL spectra as a function of molar ratio of **L-1** (blue lines) (and **D-1** (red lines)) to Eu(hfa)₃ in EtOH-free chloroform. Path length 10 mm. $\lambda_{\text{ex}} = 300 \text{ nm}$. $[\text{Eu(hfa)}_3]_0 = 1.0 \times 10^{-4} \text{ M}$. Bandwidth for emission 10 nm, bandwidth for excitation 10 nm, response time of PMT 8 s, scanning rate 50 nm per min, and two-time scan.

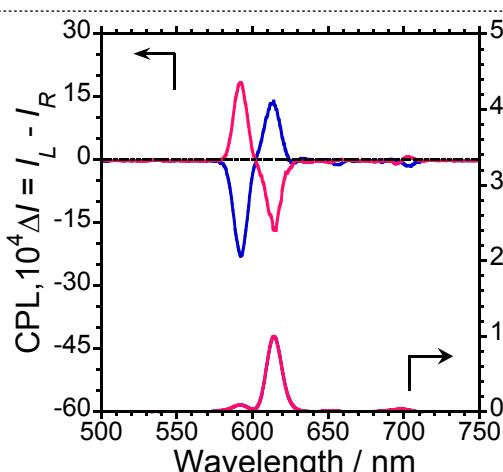


Fig. S6a L- and D-2/Eu(hfa)₃ = 1/0.5.

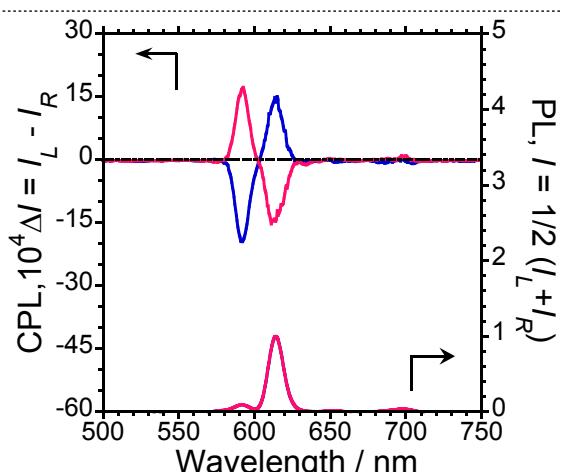


Fig. S6b L- and D-2/Eu(hfa)₃ = 1/1.

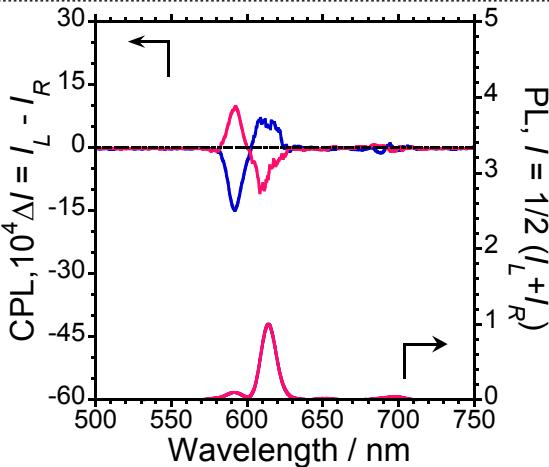


Fig. S6c L- and D-2/Eu(hfa)₃ = 1/2.

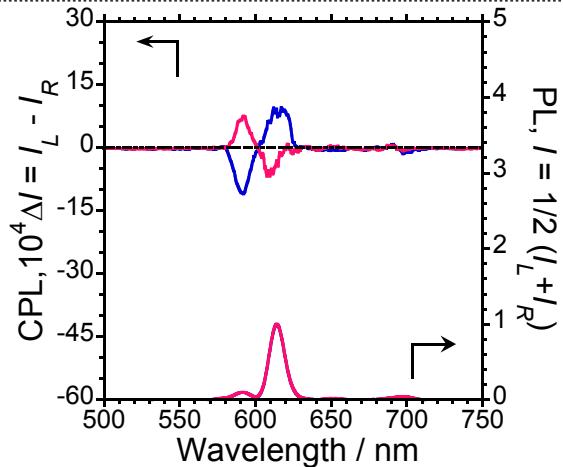


Fig. S6d L- and D-2/Eu(hfa)₃ = 1/3.

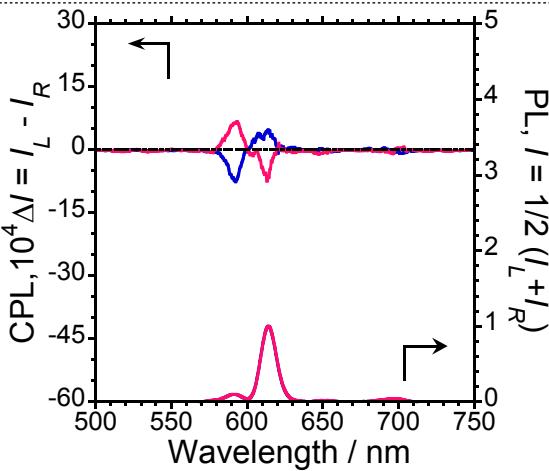


Fig. S6e L- and D-2/Eu(hfa)₃ = 1/4.

Fig. S6 CPL and PL spectra as a function of molar ratio of L-2 (blue lines) (and D-2 (red lines)) to Eu(hfa)₃ in EtOH-free chloroform. Path length 10 mm. $\lambda_{\text{ex}} = 300 \text{ nm}$. $[\text{Eu(hfa)}_3]_0 = 1.0 \times 10^{-4} \text{ M}$. Bandwidth for emission 10 nm, bandwidth for excitation 10 nm, response time of PMT 8 s, scanning rate 50 nm per min, and two-time scan.

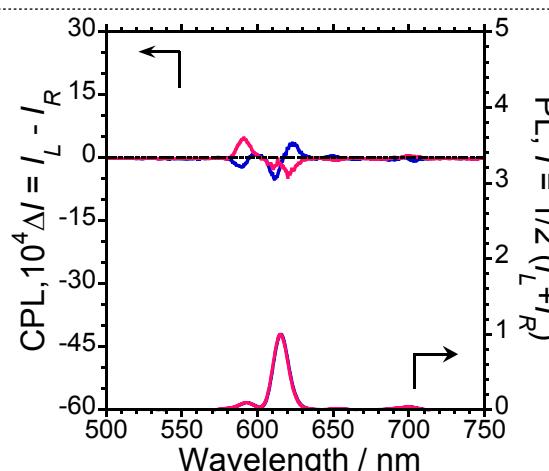


Fig. S7a L- and D-3/Eu(hfa)₃ = 1/0.5.

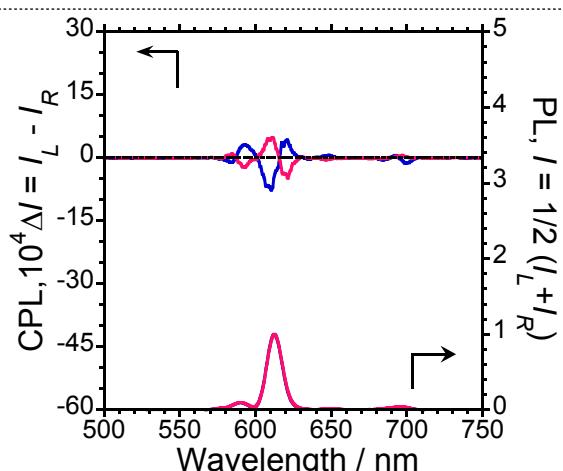


Fig. S7b L- and D-3/Eu(hfa)₃ = 1/1.

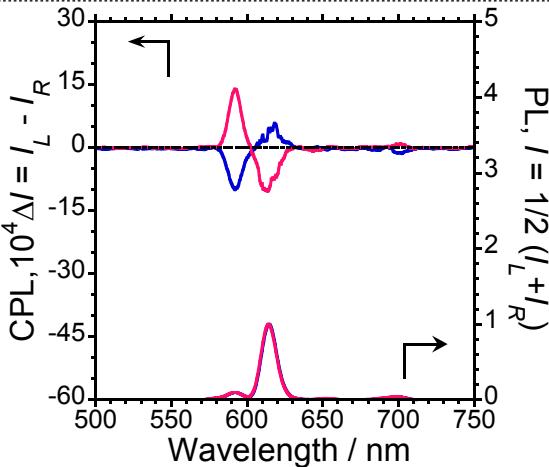


Fig. S7c L- and D-3/Eu(hfa)₃ = 1/2.

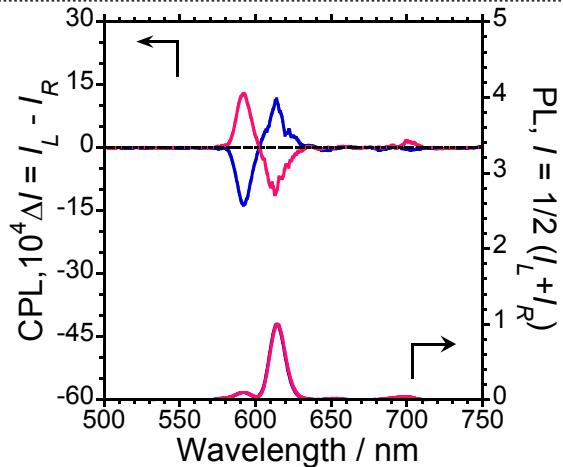


Fig. S7d L- and D-3/Eu(hfa)₃ = 1/3.

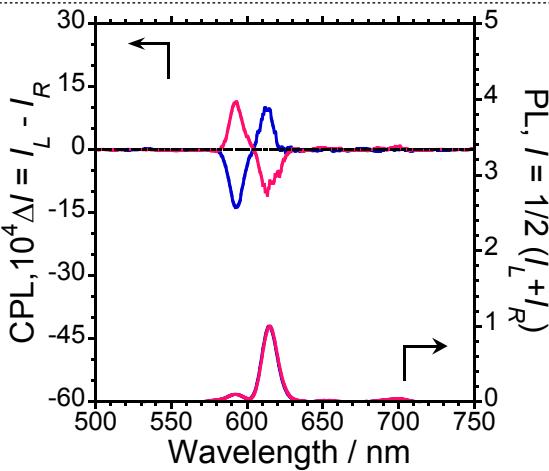


Fig. S7e L- and D-3/Eu(hfa)₃ = 1/4.

Fig. S7 CPL and PL spectra as a function of molar ratio of L-3 (blue lines) (and D-3 (red lines)) to Eu(hfa)₃ in EtOH-free chloroform. Path length 10 mm. $\lambda_{\text{ex}} = 300 \text{ nm}$. $[\text{Eu(hfa)}_3]_0 = 1.0 \times 10^{-4} \text{ M}$. Bandwidth for emission 10 nm, bandwidth for excitation 10 nm, response time of PMT 8 s, scanning rate 50 nm per min, and two-time scan.

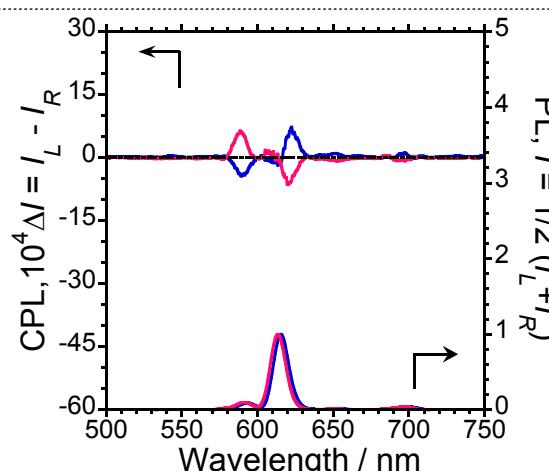


Fig. S8a L- and D-4/Eu(hfa)₃ = 1/0.5.

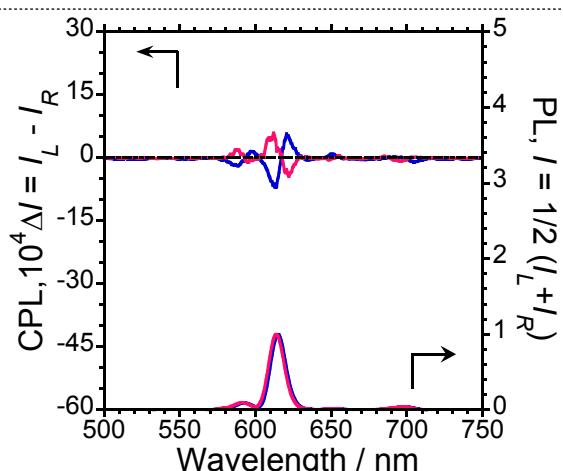


Fig. S8b L- and D-4/Eu(hfa)₃ = 1/1.

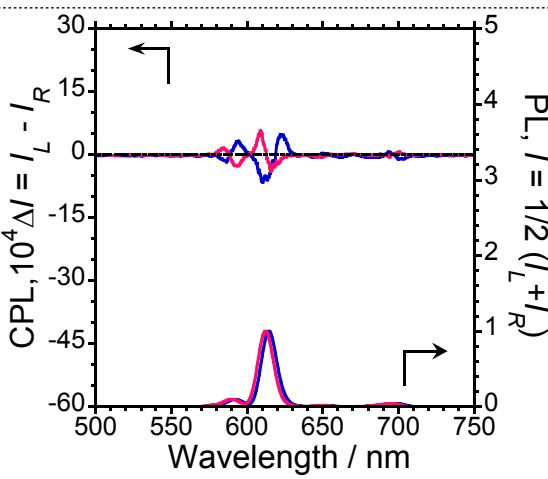


Fig. S8c L- and D-4/Eu(hfa)₃ = 1/2.

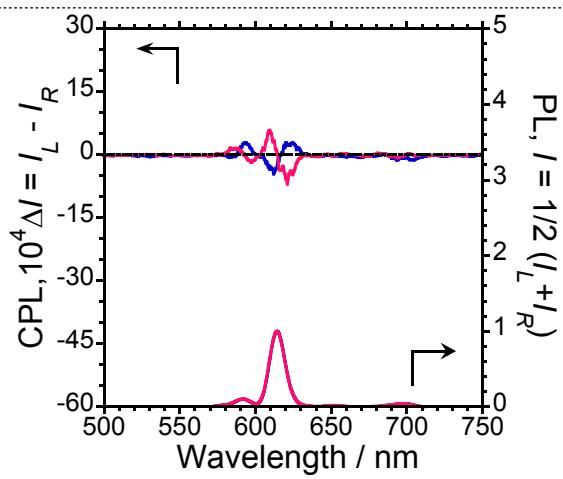


Fig. S8d L- and D-4/Eu(hfa)₃ = 1/3.

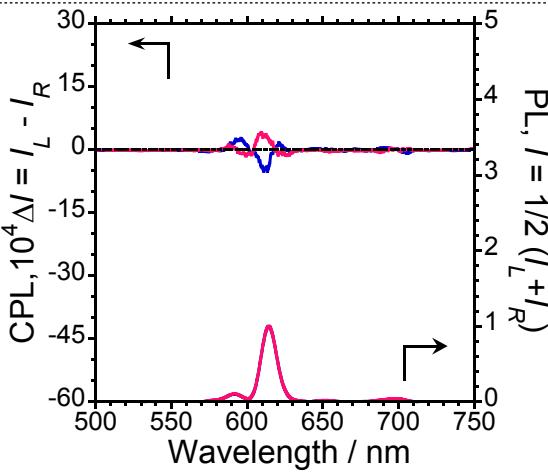


Fig. S8e L- and D-4/Eu(hfa)₃ = 1/4.

Fig. S8 CPL and PL spectra as a function of molar ratio of L-4 (blue lines) (and D-4 (red lines)) to Eu(hfa)₃ in EtOH-free chloroform. Path length 10 mm. $\lambda_{\text{ex}} = 300$ nm. $[\text{Eu(hfa)}_3]_0 = 1.0 \times 10^{-4}$ M. Bandwidth for emission 10 nm, bandwidth for excitation 10 nm, response time of PMT 8 s, scanning rate 50 nm per min, and two-time scan.

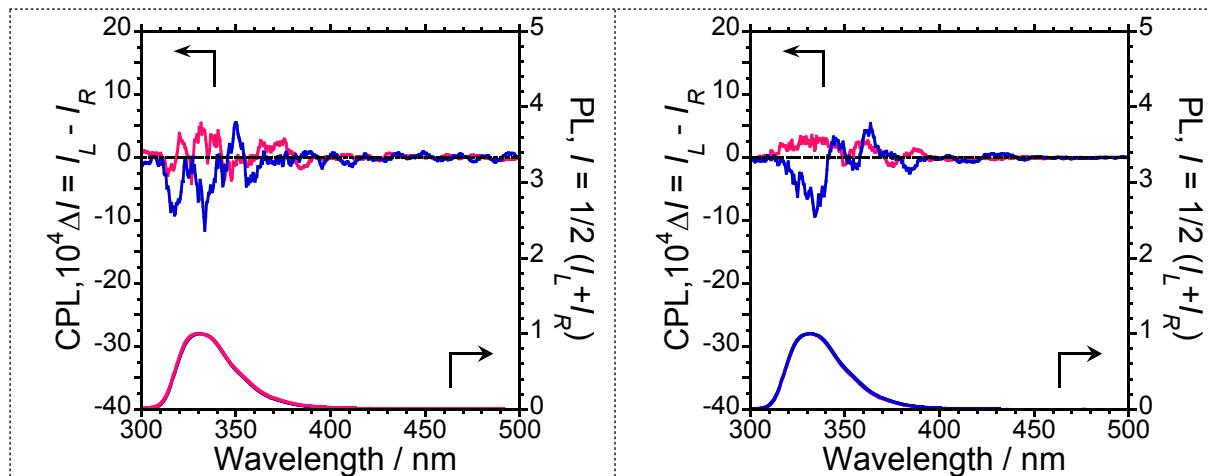


Fig. S9a L- and D-1.

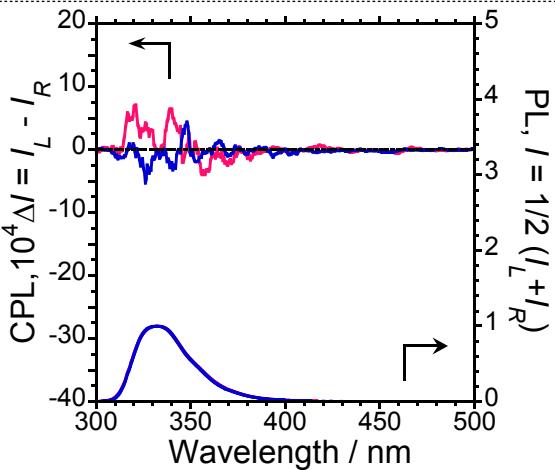


Fig. S9c L- and D-3.

Fig. S9b L- and D-2.

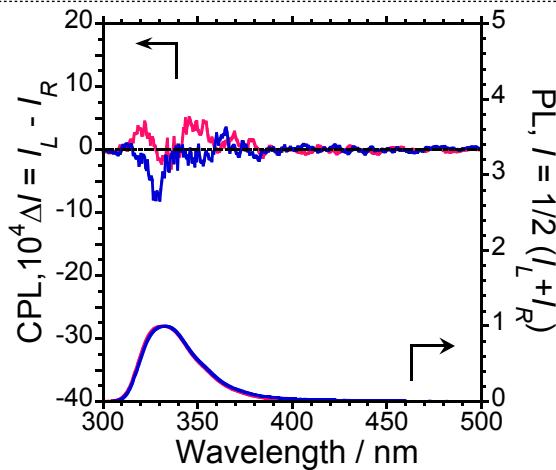


Fig. S9d L- and D-4.

Fig. S9 CPL and PL spectra of (a) **1**, (b) **2**, (c) **3**, and (d) **4** in EtOH-free chloroform. Blue and red lines are **L**- and **D**-isomers, respectively. Path length 10 mm. $\lambda_{\text{ex}} = 270 \text{ nm}$. Conc. = $1.0 \times 10^{-4} \text{ M}$. Bandwidth for emission 10 nm, bandwidth for excitation 10 nm, response time of PMT 8 s, scanning rate 50 nm per min, and two-time scan.

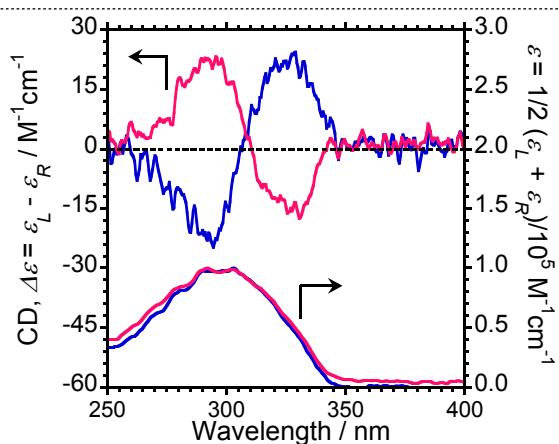
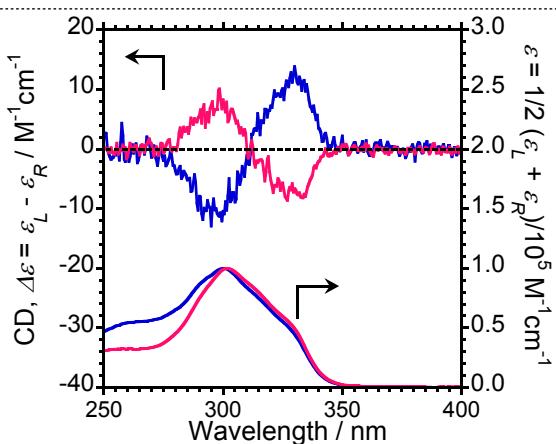


Fig. S10a L- and D-1/Eu(hfa)₃ = 1/1.

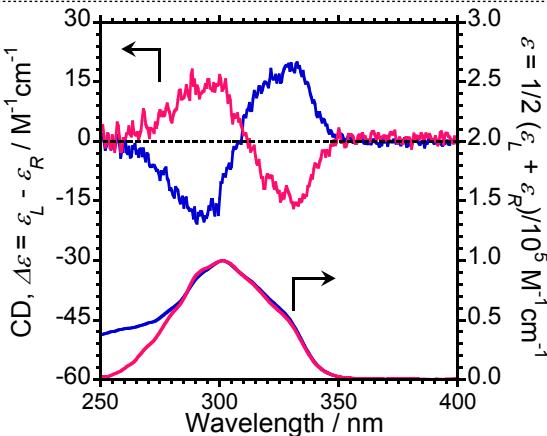


Fig. S10c L- and D-3/Eu(hfa)₃ = 1/1.

Fig. S10b L- and D-2/Eu(hfa)₃ = 1/1.

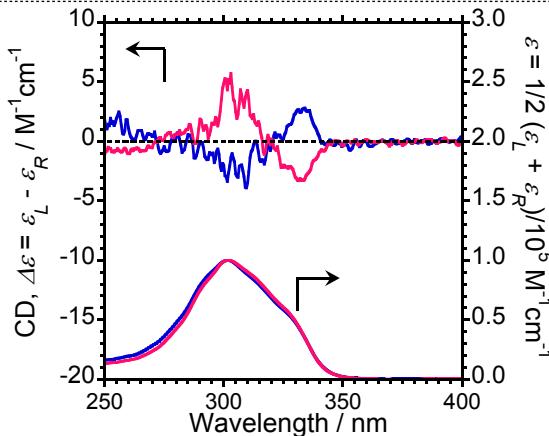


Fig. S10d L- and D-4/Eu(hfa)₃ = 1/1.

Fig. S10 Comparison of CD and UV spectra of (a) **1**/Eu(hfa)₃ = 1/1, (b) **2**/Eu(hfa)₃ = 1/1, (c) **3**/Eu(hfa)₃ = 1/1, and (d) **4**/Eu(hfa)₃ = 1/1 (in nominal molar ratio) in EtOH-free chloroform. Blue and red lines are L- and D-isomers, respectively. Path lengths were commonly 1 mm. [Eu]₀ = 1.0 × 10⁻⁴ M in EtOH-free chloroform.

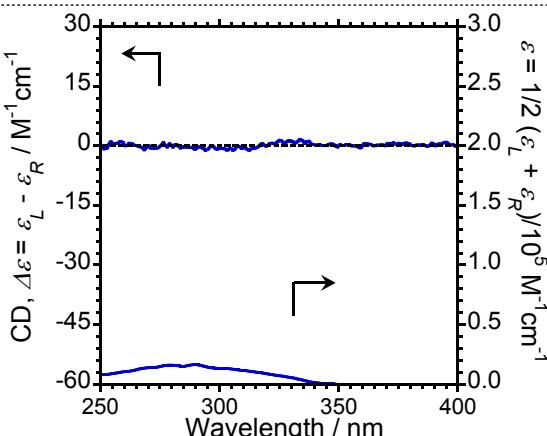


Fig. S11a L-3/Eu(hfa)₃ = 1/0.5.

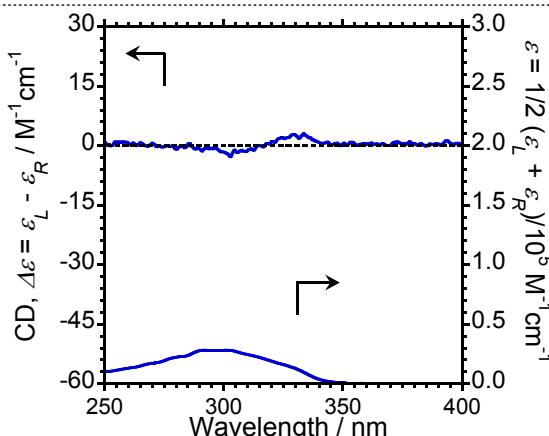
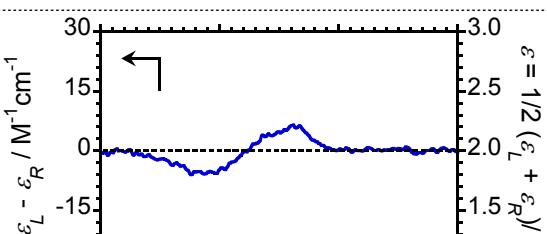


Fig. S11b L-3/Eu(hfa)₃ = 1/1.



S113

Fig. S11 Comparison of CD and UV spectra of (a) **L-3/Eu(hfa)₃** = 1/0.5, (b) **L-3/Eu(hfa)₃** = 1/1, (c) **L-3/Eu(hfa)₃** = 1/2, and **L-3/Eu(hfa)₃** = 1/3 (in nominal molar ratio) in EtOH-free chloroform. Blue lines are **L**-isomers, respectively. Path lengths were commonly 1 mm. [Eu]₀ = 1.0 × 10⁻⁴ M in EtOH-free chloroform.

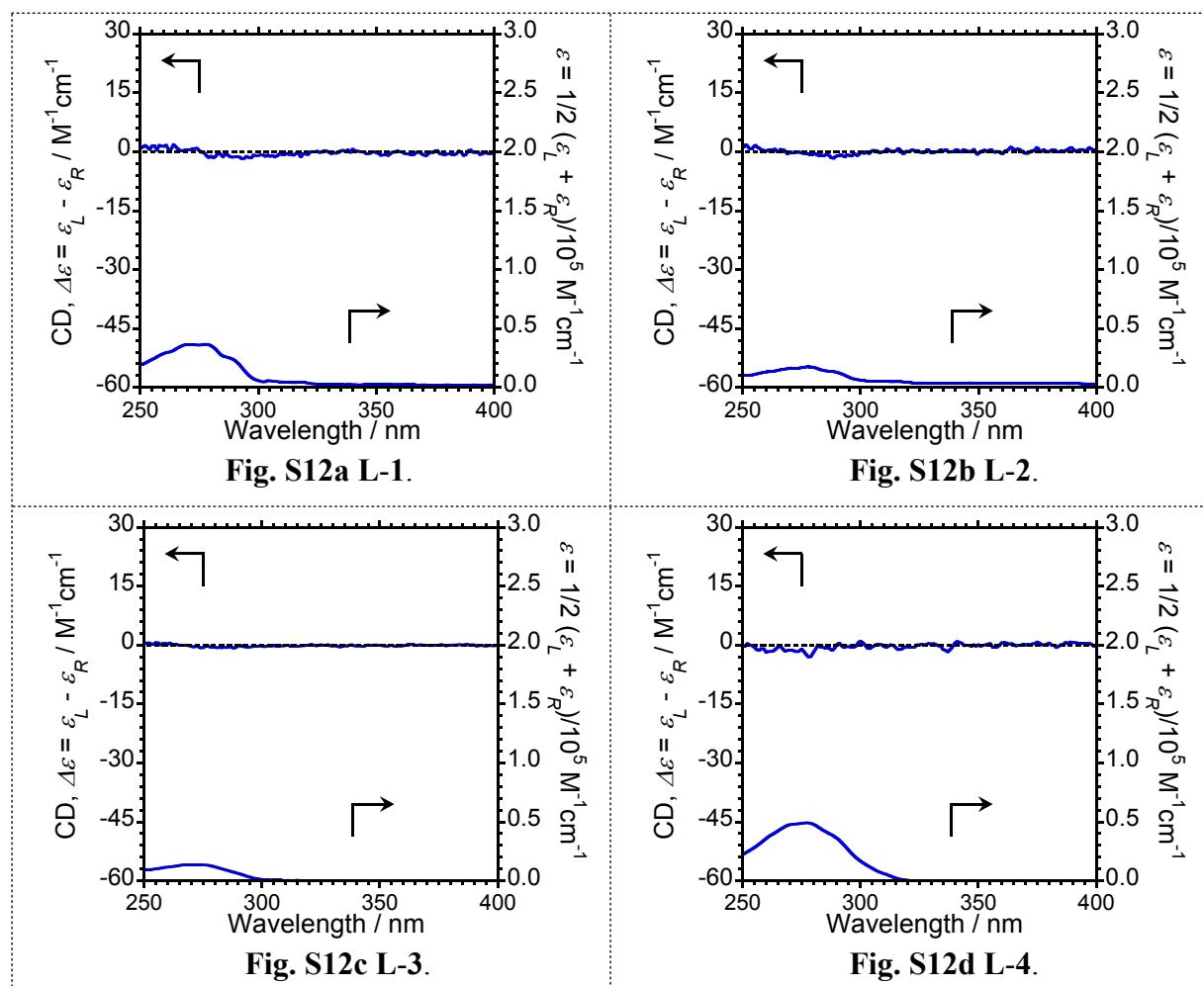
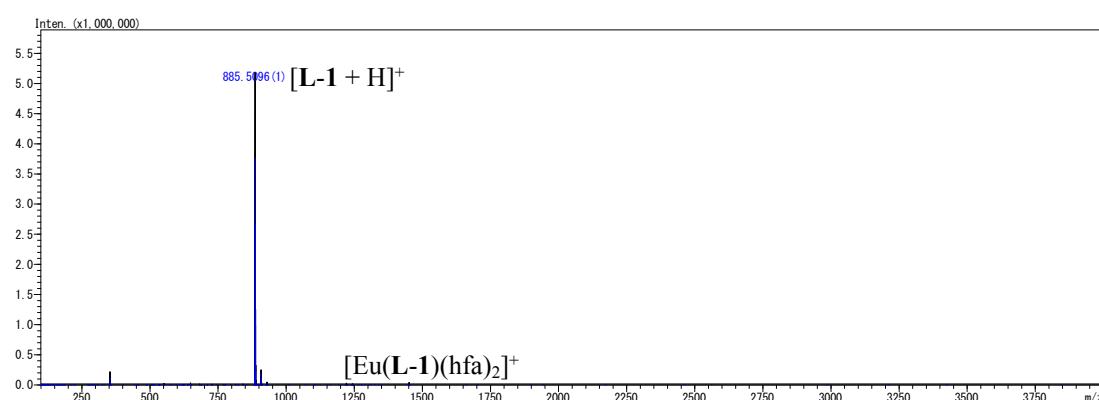


Fig. S12 Comparison of CD and UV spectra of (a) **L-1**, (b) **L-2**, (c) **L-3**, and (d) **L-4** in EtOH-free chloroform. Conc. = 1.0 × 10⁻⁴ M. Path lengths were 1 mm. (Path length of only 1

was 10 mm.)

(a)



(b)

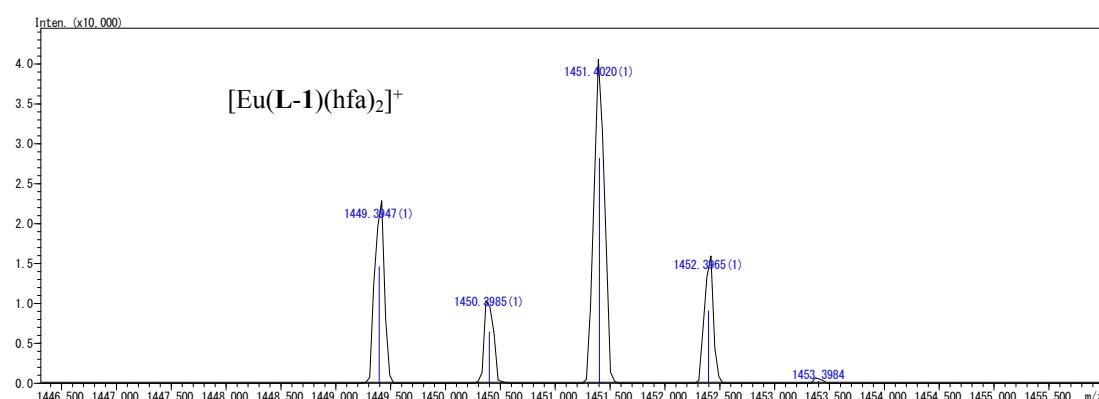


Fig. S13 Mass spectra of **L-1/Eu(hfa)₃** (1/1)

(a) All range, and (b) expanded range. Found m/z 1451.4020, calcd. m/z 1451.3970 as $[Eu(L-1)(hfa)_2]^+$.

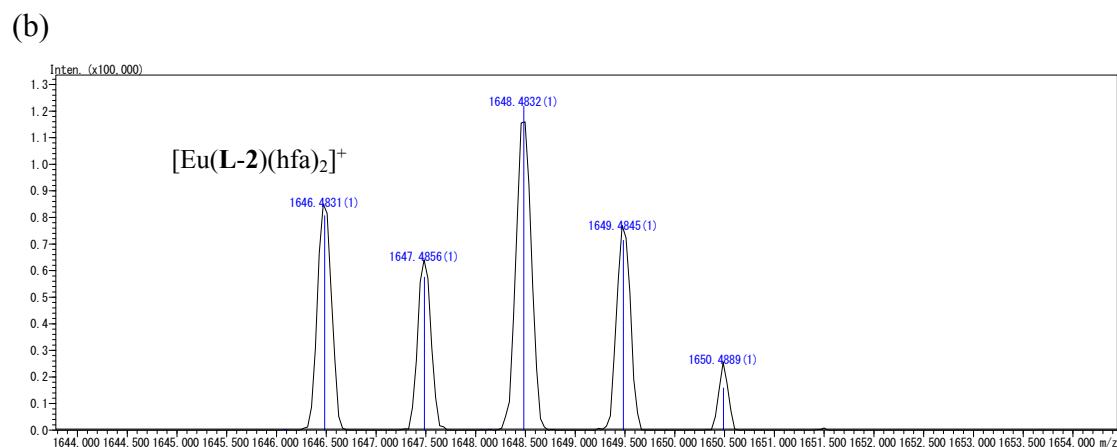
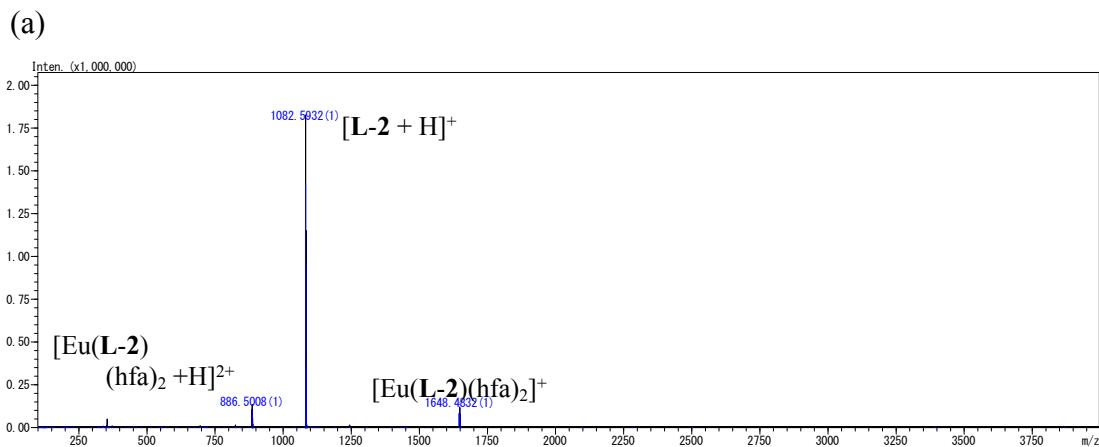
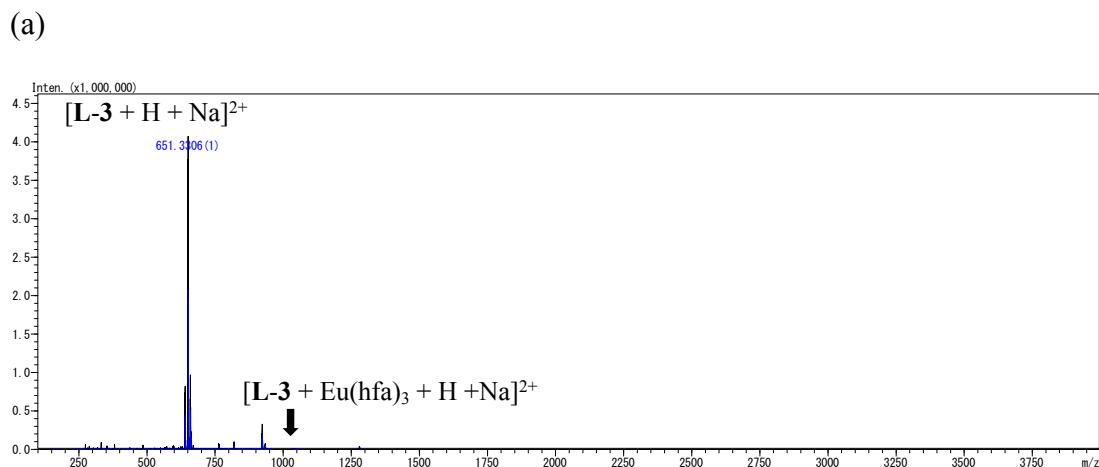


Fig. S14 Mass spectra of **L-2/Eu(hfa)₃** (1/1)

(a) All range, and (b) expanded range. Found m/z 1648.4832, calcd. m/z 1648.4814 as $[Eu(L-2)(hfa)_2]^{2+}$.



(b)

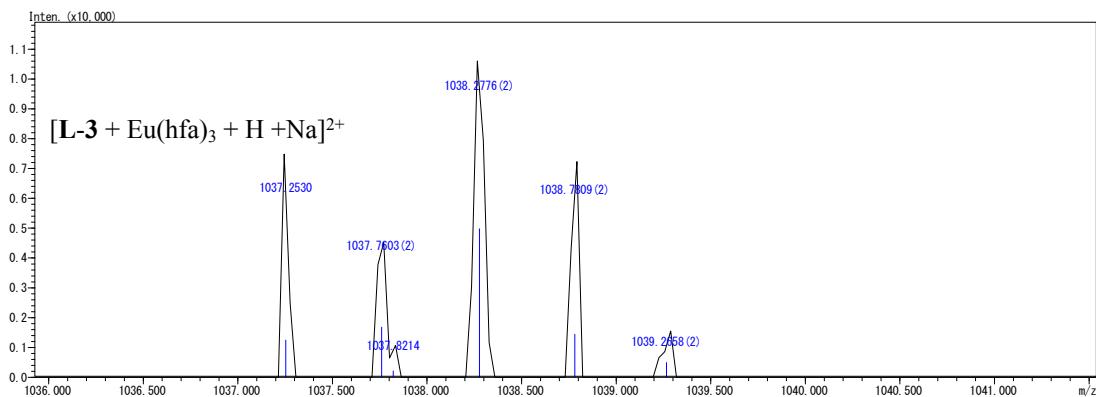
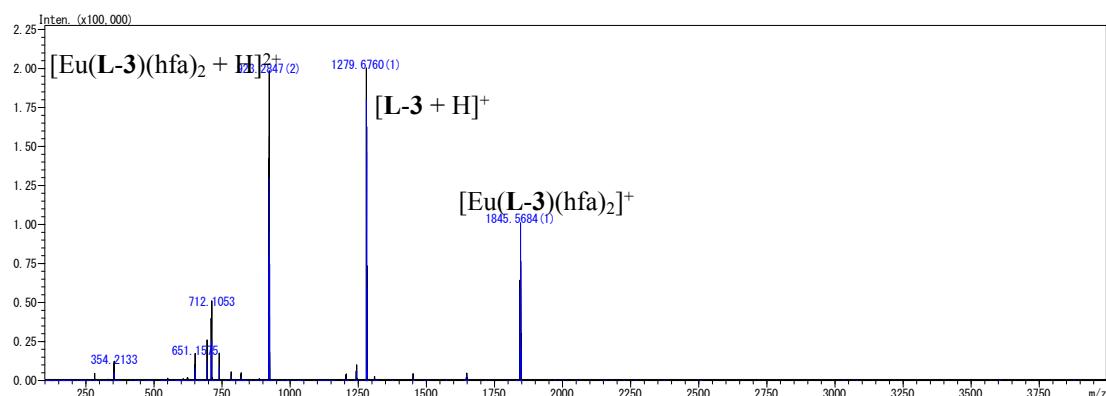


Fig. S15 Mass spectra of **L-3/Eu(hfa)₃** (1/1)

(a) All range, and (b) expanded range. Found m/z 1037.2530, calcd. m/z 1037.2743 as $[\text{Eu}(\text{L-3})(\text{hfa})_3 + \text{H} + \text{Na}]^{2+}$.

(a)



(b)

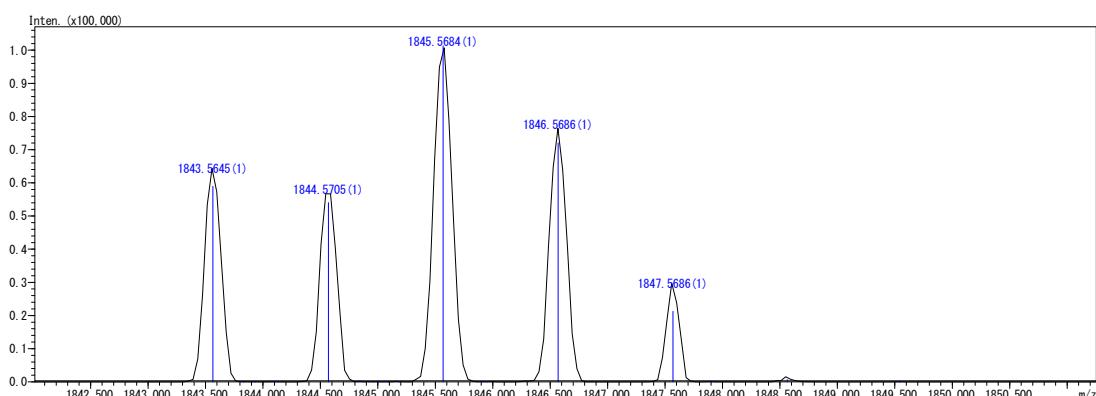
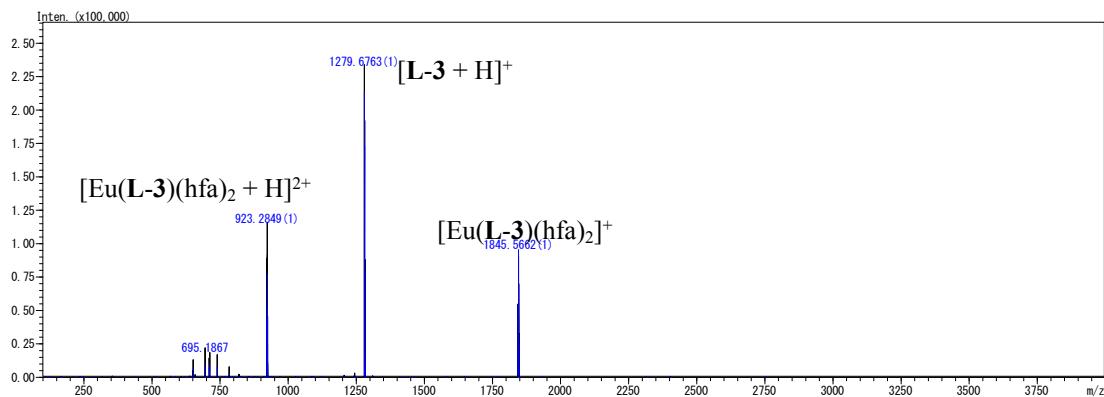


Fig. S16 Mass spectra of **L-3/Eu(hfa)₃** (1/2)

(a) All range, and (b) expanded range. Found m/z 1845.5684, calcd. m/z 1845.5657 as $[\text{Eu}(\text{L-3})(\text{hfa})_2]^{+}$.

(a)



(b)

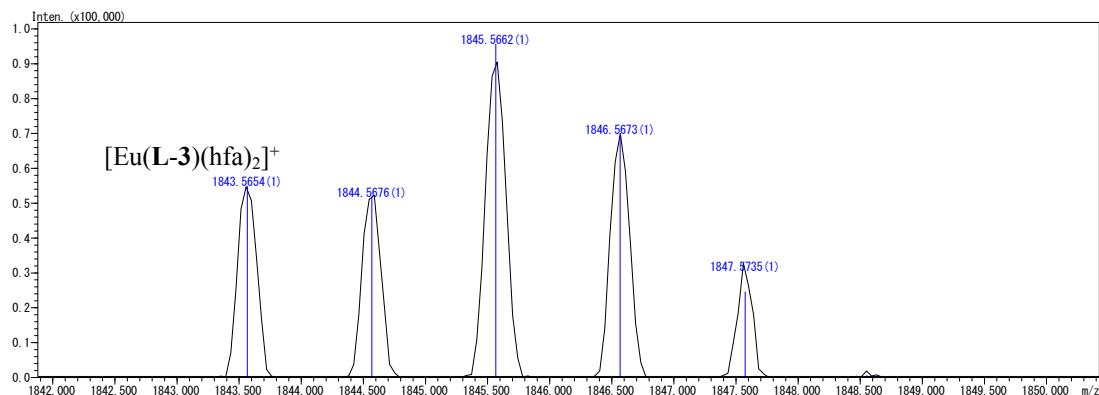
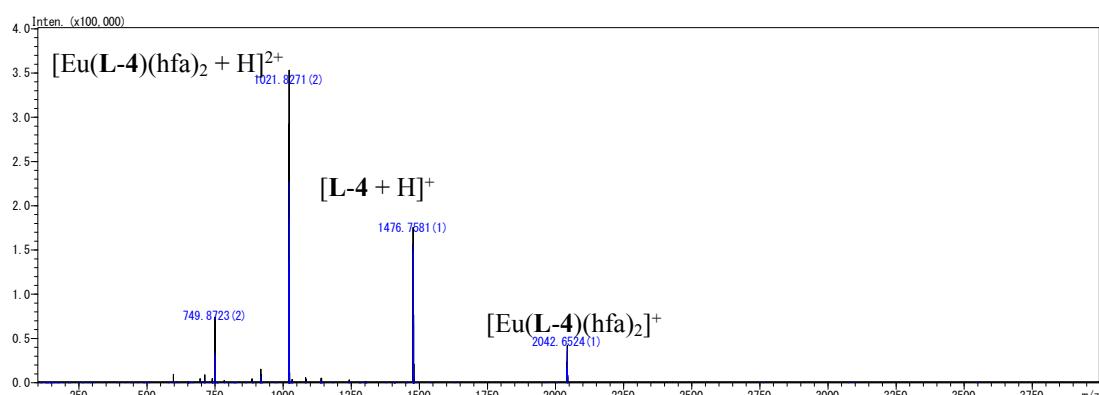


Fig. S17 Mass spectra of **L-3/Eu(hfa)₃** (1/3)

(a) All range, and (b) expanded range. Found m/z 1845.5662, calcd. m/z 1845.5657 as $[\text{Eu}(\text{L-3})(\text{hfa})_2]^+$.

(a)



(b)

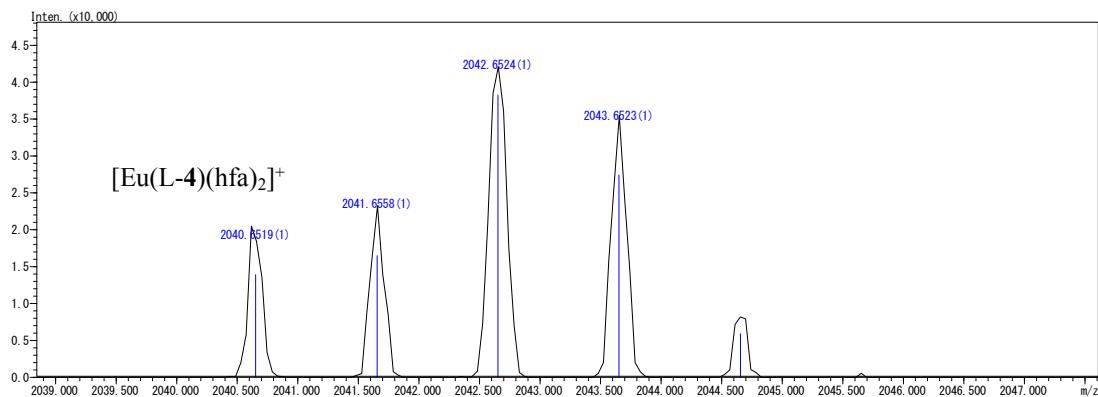


Fig. S18 Mass spectra of **L-4/Eu(hfa)₃** (1/1)

(a) All range, and (b) expanded range. Found m/z 2042.6524, calcd. m/z 2042.6501 as $[\text{Eu}(\text{L-4})(\text{hfa})_2]^+$.

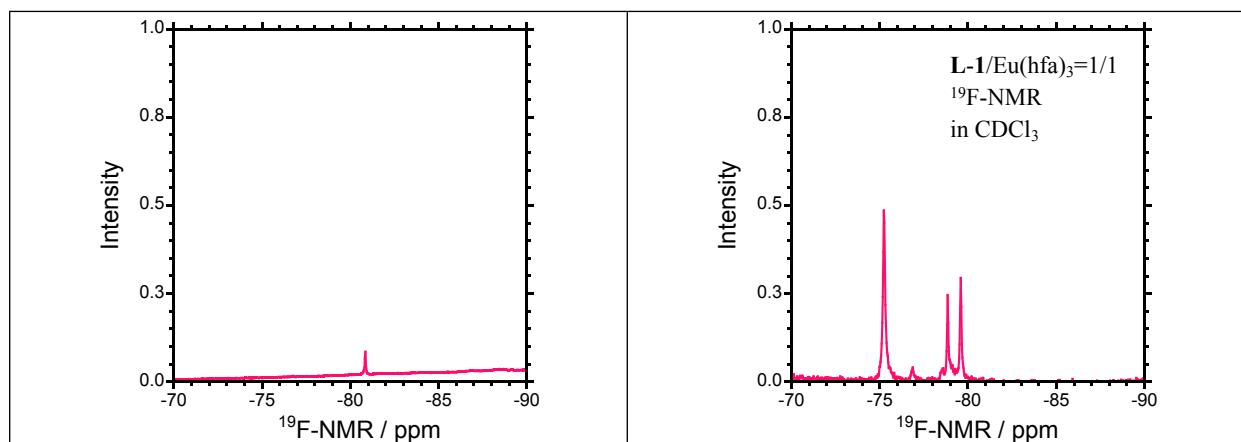


Fig. S19 ^{19}F -NMR spectrum of $\text{Eu}(\text{hfa})_3$ and **L-1/Eu(hfa)₃** (1/1) in CDCl_3

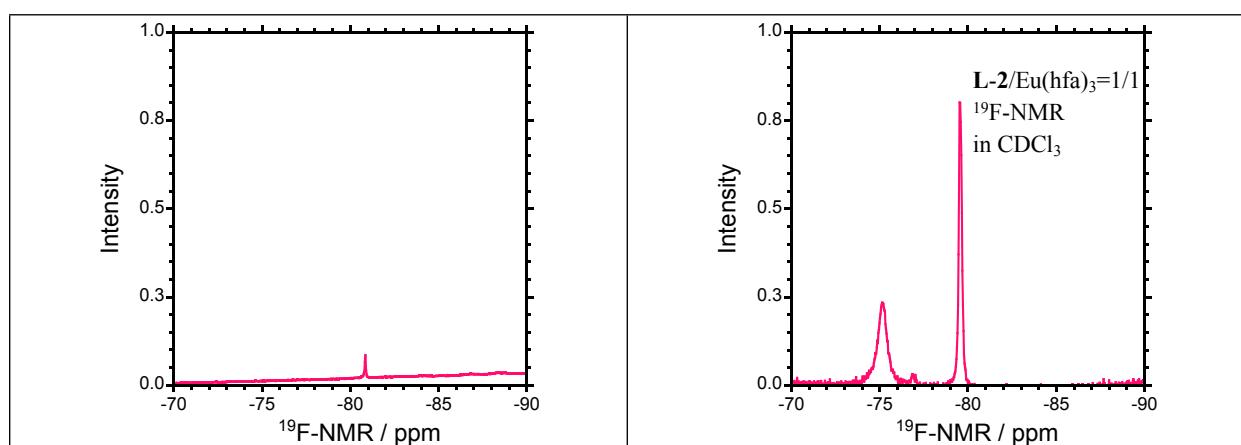


Fig. S20 ^{19}F -NMR spectrum of $\text{Eu}(\text{hfa})_3$ and **L-2/Eu(hfa)₃** (1/1) in CDCl_3

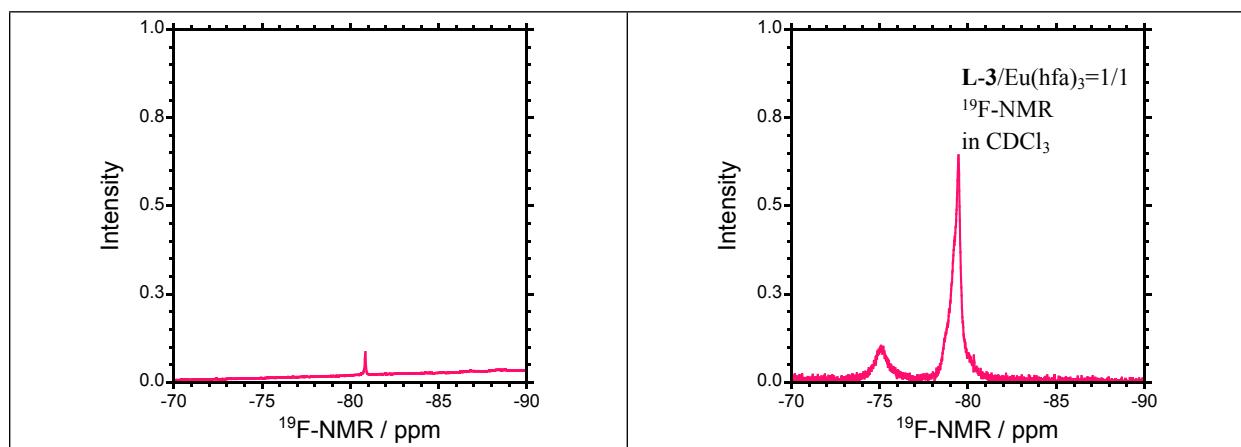


Fig. S21 ¹⁹F-NMR spectrum of Eu(hfa)₃ and **L-3/Eu(hfa)₃** (1/1) in CDCl₃

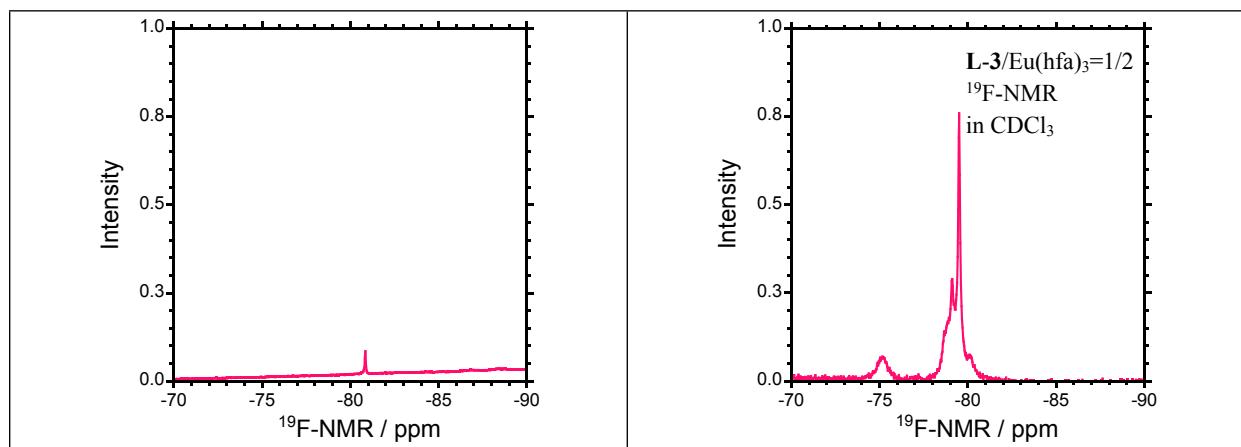


Fig. S22 ¹⁹F-NMR spectrum of Eu(hfa)₃ and **L-3/Eu(hfa)₃** (1/2) in CDCl₃

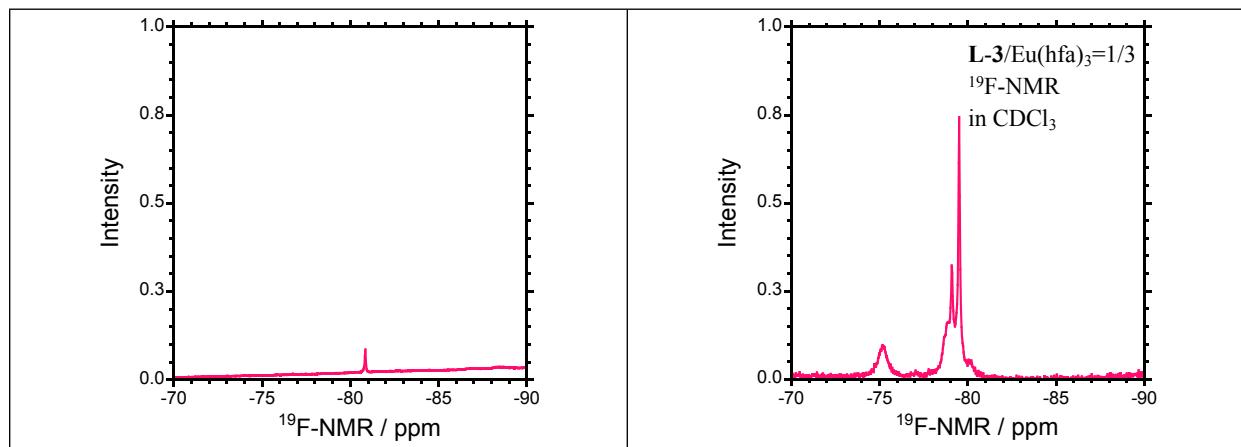


Fig. S23 ¹⁹F-NMR spectrum of Eu(hfa)₃ and **L-3/Eu(hfa)₃** (1/3) in CDCl₃

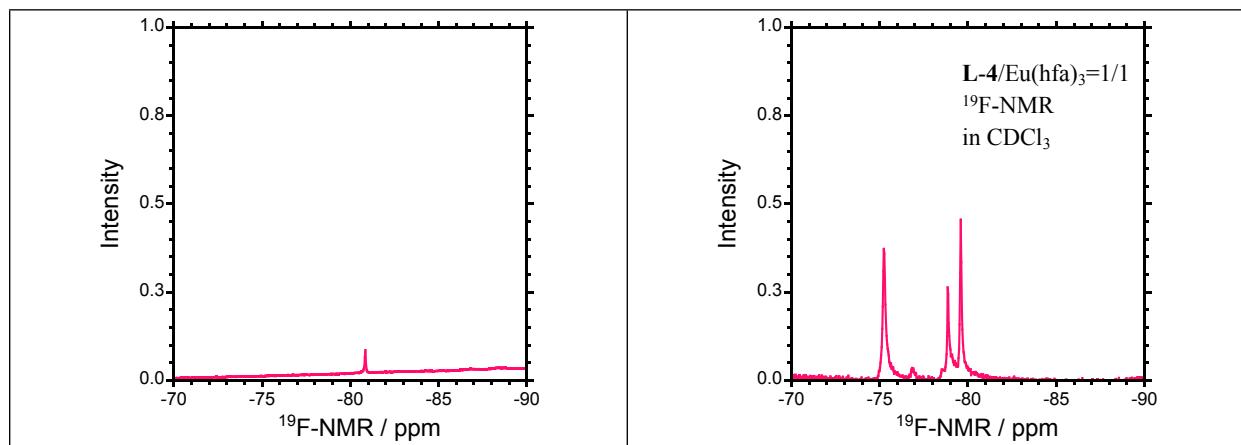


Fig. S24 ^{19}F -NMR spectrum of $\text{Eu}(\text{hfa})_3$ and **L-4**/ $\text{Eu}(\text{hfa})_3$ (1/1) in CDCl_3