

Supporting Information

Reduce graphene oxide/TiO₂(B) nanocomposite modified separator as efficient suppression of polysulfide shuttling in Li-S batteries

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Characterization

The crystal structure and phase identification of the samples were analyzed by Rigaku, D/max-2500 X-ray diffractometer (XRD). Morphology was detected by the FEI Quanta 250 field-emission scanning electron microscope (SEM) and Energy-dispersive spectroscopy (EDS). X-ray photoelectron spectroscopy (XPS) was tested on the ESCALABMKLL (Thermo Fisher Scientific company). The molar ratio of sulfur (S) and lithium sulfide (Li₂S) was 5:1 in electrolyte to synthesize Lithium polysulfide (Li₂S₆) solution and kept stirring at room temperature for 48 h in glovebox.

Cyclic voltammetry (CV) measurements were performed at a scan rate of 0.1 mV s^{-1} on the electrochemical station (CHI760D, Chenhua Instrument Company). Electrochemical impedance spectroscopy (EIS) tests were applied using the Biologic VMP3 electrochemical workstation in the frequency range from 0.01 Hz to 1 MHz. The cycling stability and rate capacity were tested using a Battery Testing System (LAND CT2001A) within a potential range of 1.7-2.8 V vs. Li/Li^+ .

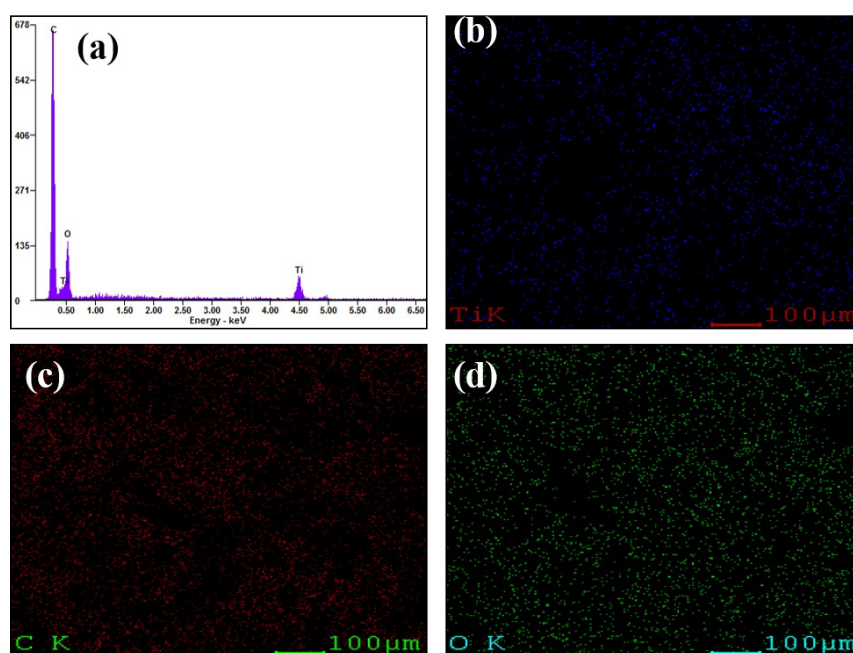


Fig. S1. EDX spectrum and mappings of the RGO/TiO₂(B) composite surface.



Fig. S2. The photograph of RGO/TiO₂(B) modified separator with thicker coating.

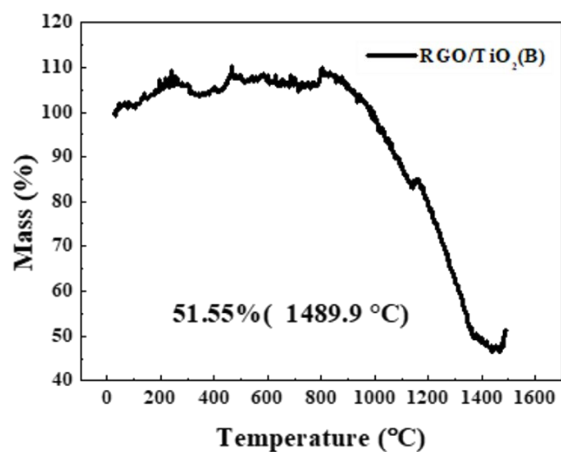


Fig. S3. Thermogravimetric analysis curve of RGO/TiO₂(B).

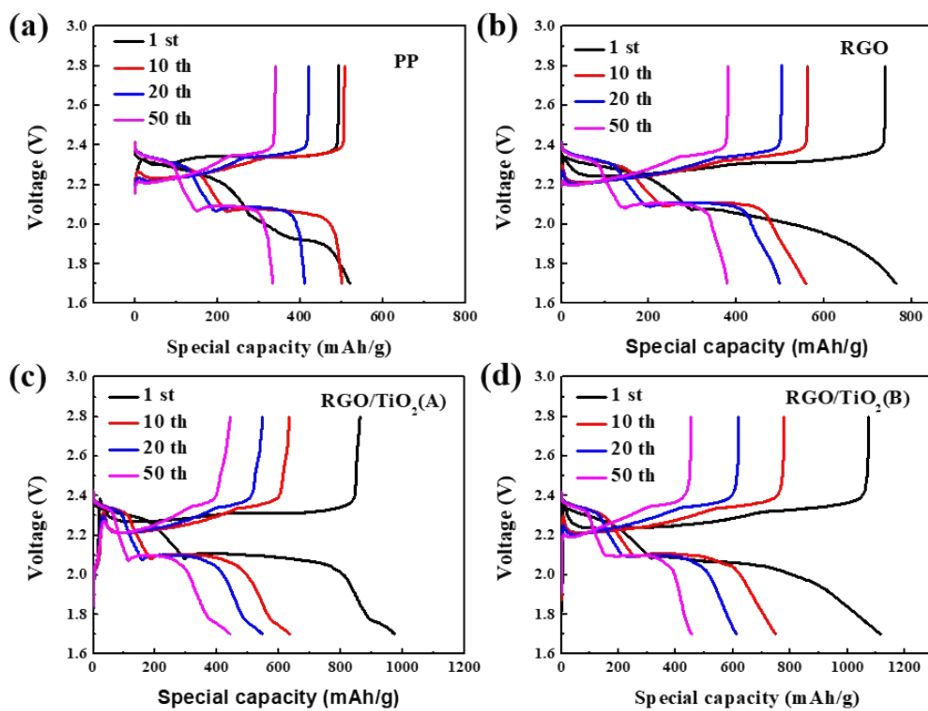


Fig. S4. The charge-discharge curves of different cycles at a rate of 0.2 C for batteries with the separators with different coating.

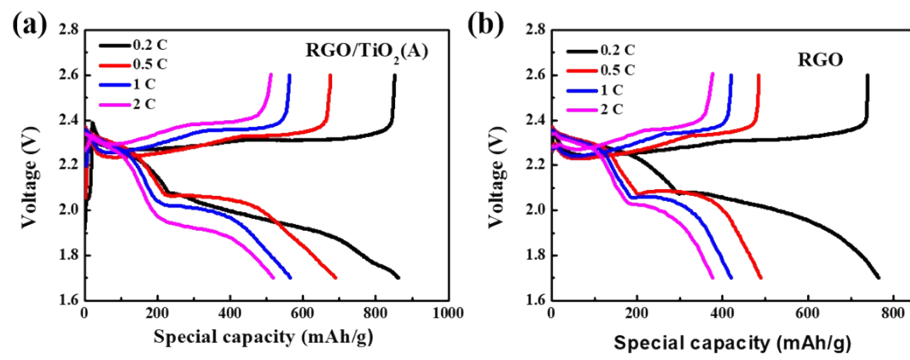


Fig. S5. The charge-discharge voltage curves of batteries with separators at various rates from 0.2 C to 2 C.