

## Electronic Supporting Information

### Sr<sub>9</sub>In(VO<sub>4</sub>)<sub>7</sub> as a model ferroelectric in the structural family of β-Ca<sub>3</sub>(PO<sub>4</sub>)<sub>2</sub>-type phosphates and vanadates

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**Table S1.** Structure parameters of Sr<sub>9</sub>In(VO<sub>4</sub>)<sub>7</sub> at 293 K from laboratory X-ray powder diffraction data.

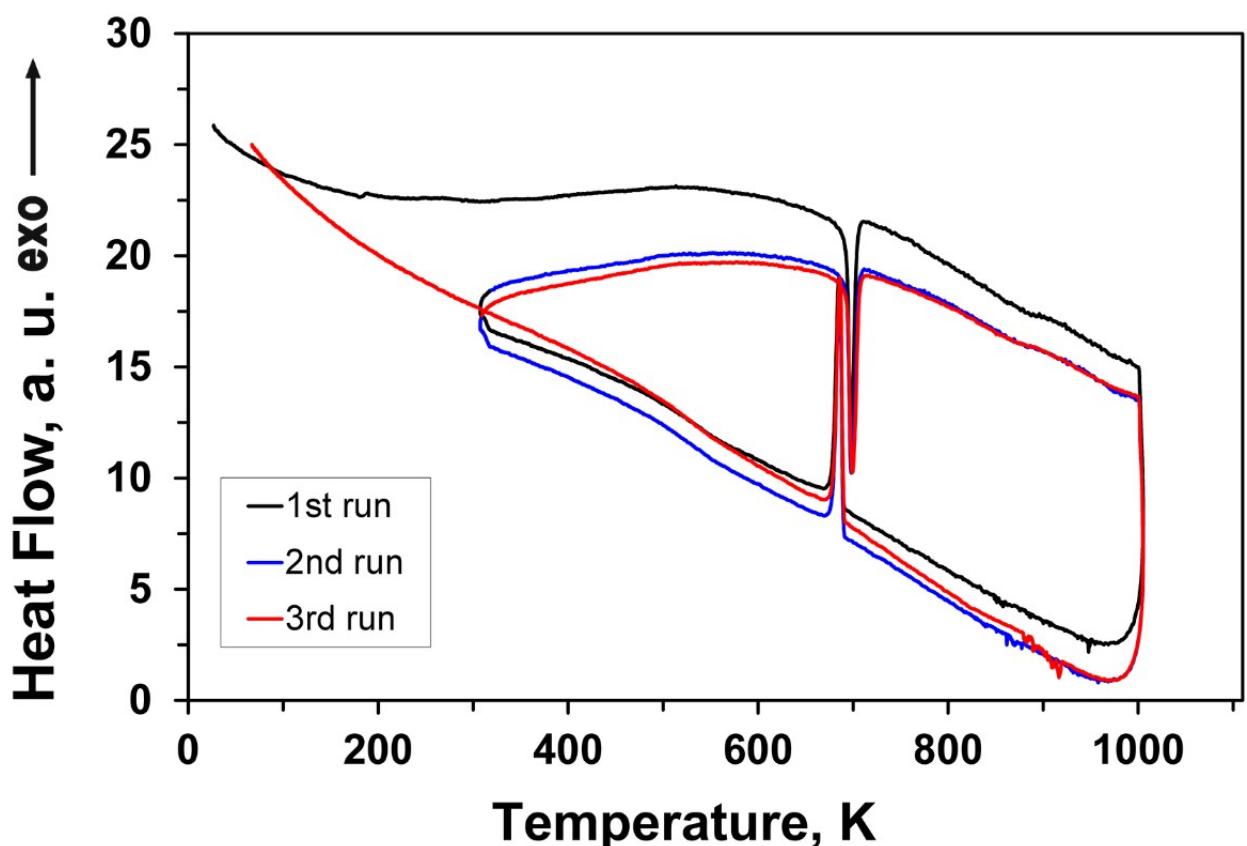
Site	Wyckoff position	x	y	z	B (Å <sup>2</sup> )
Sr1	18b	0.7233(2)	0.8571(2)	0.43262(8)	0.72(3)
Sr2	18b	0.6188(2)	0.8230(2)	0.23166(7)	0.60(4)
Sr3	18b	0.1281(2)	0.27378(14)	0.32492(7)	0.90(3)
In	6a	0.0	0.0	0.0	0.52(4)
V1	6a	0.0	0.0	0.27062(13)	1.04(10)
V2	18b	0.6845(3)	0.8567(4)	0.13523(10)	0.22(8)
V3	18b	0.6572(4)	0.8466(4)	0.03138(11)	0.80(9)
O11	6a	0.0	0.0	0.3136(4)	0.8(4)
O12	18b	0.0201(10)	0.8680(8)	0.2574(3)	0.6(3)
O21	18b	0.7423(13)	0.9231(11)	0.1737(3)	1.6(3)
O22	18b	0.7511(14)	0.7594(15)	0.1194(3)	0.6(3)
O23	18b	0.7162(13)	0.0034(10)	0.1108(3)	0.2(3)
O24	18b	0.5022(10)	0.7625(13)	0.1316(3)	0.2(2)
O31	18b	0.5968(11)	0.9446(11)	0.0444(3)	0.2(2)
O32	18b	0.5749(11)	0.6844(13)	0.0495(3)	0.5(3)
O33	18b	0.8316(11)	0.9199(16)	0.0375(3)	0.2(2)
O34	18b	0.6138(9)	0.8171(13)	0.9888(2)	0.1(2)

Space group *R*3*c* (No. 161, origin choice 1); *Z* = 6. *a* = 11.18016(9) Å, *c* = 39.6170(3) Å, and *V* = 4288.53(5) Å<sup>3</sup>; *R*<sub>wp</sub> = 5.86 %, *R*<sub>p</sub> = 4.52 %, *R*<sub>B</sub> = 1.35 %, and *R*<sub>F</sub> = 0.62 %. *ρ*<sub>cal</sub> = 3.968 g/cm<sup>3</sup>.

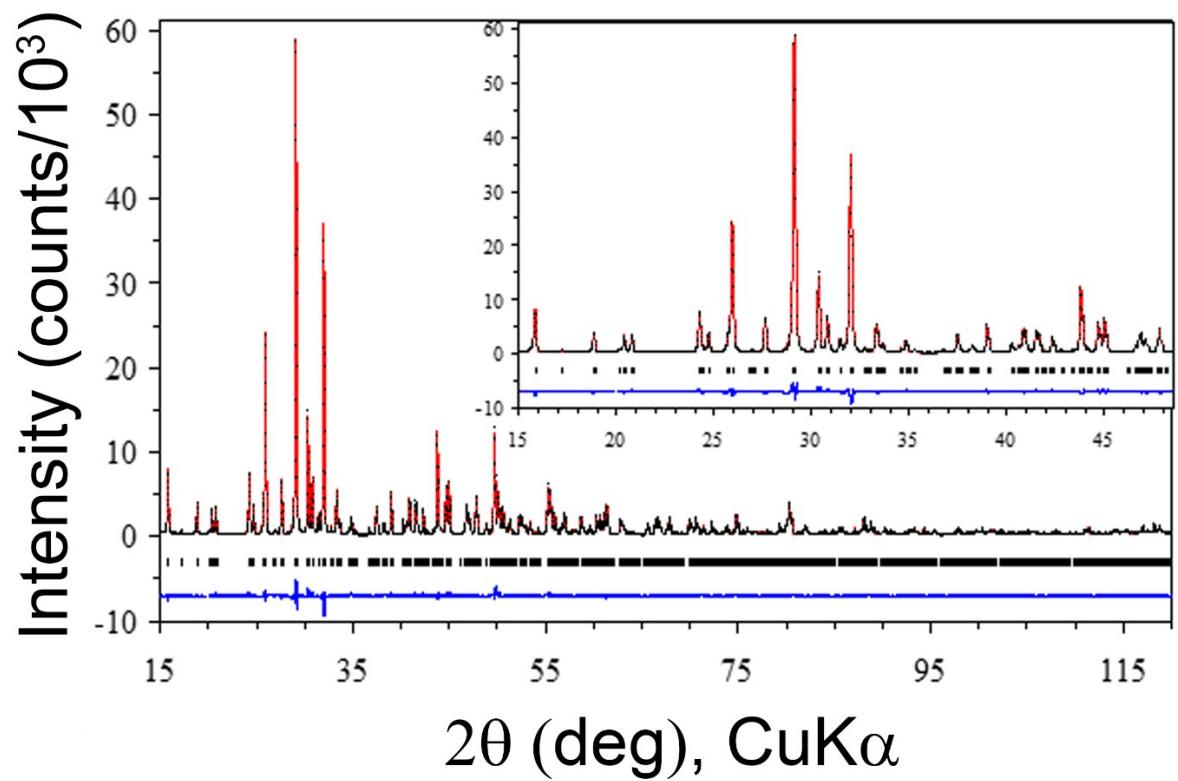
The occupation factor of all the sites is unity (*g* = 1).

**Table S2.** Selected Interatomic Distances ( $\text{\AA}$ ) and Bond Valence Sum (BVS) in  $\text{Sr}_9\text{In}(\text{VO}_4)_7$ .

Sr1–O32	2.471(2)	Sr2–O12	2.488(9)
Sr1–O34	2.467(9)	Sr2–O23	2.499(1)
Sr1–O31	2.611(2)	Sr2–O21	2.626(1)
Sr1–O12	2.680(9)	Sr2–O22	2.548(2)
Sr1–O24	2.584(2)	Sr2–O33	2.605(2)
Sr1–O23	2.558(3)	Sr2–O'33	2.684(2)
Sr1–O'24	2.751(3)	Sr2–O32	2.917(2)
Sr1–O22	2.951(4)	Sr2–O31	2.899(1)
$\langle \text{Sr1–O} \rangle$	<b>2.634</b>	$\langle \text{Sr2–O} \rangle$	<b>2.658</b>
BVS( $\text{Sr}^{2+}$ )	2.13	BVS( $\text{Sr}^{2+}$ )	2.01
Sr3–O22	2.628(2)	In–O34 ( $\times 3$ )	2.150(1)
Sr3–O23	2.597(1)	In–O43 ( $\times 3$ )	2.207(1)
Sr3–O31	2.567(1)	$\langle \text{In–O} \rangle$	<b>2.178</b>
Sr3–O34	2.677(2)	BVS( $\text{In}^{3+}$ )	2.85
Sr3–O11	2.690(3)		
Sr3–O'34	2.699(3)	V1–O1	1.703(2)
Sr3–O32	2.707(2)	V1–O12 ( $\times 3$ )	1.683(8)
Sr3–O21	2.781(1)	$\langle \text{V1–O} \rangle$	<b>1.688</b>
Sr3–O12	3.010(1)		
$\langle \text{Sr3–O} \rangle$	2.707		
BVS( $\text{Sr}^{2+}$ )	1.92		



**Fig. S1.** The heating/cooling DTA curves of  $\text{Sr}_9\text{In}(\text{VO}_4)_7$  (the heating/cooling rate was  $10 \text{ K min}^{-1}$ ). Three heating/cooling cycles were run to verify the reproducibility.



**Fig. S2.** The XRD data for  $\text{Sr}_9\text{In}(\text{VO}_4)_7$  in whole region of data collection.