

Electronic Supplementary Information

Preparation of Cyclic Imides from Alkene-tethered Amides: an Application of Homogeneous Cu(II) Catalytical System

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Abstract: A Cu-based homogeneous catalytic system was put forward for the preparation of imides from alkene-tethered amides. O₂ acted as a cheap and easily available oxygen source and a terminal oxidant. Cleavage of C=C bonds and formation of C-N bond were catalyzed by Cu(II) salts with proper nitrogen-containing ligand under 100 °C. The synthetic approach has potential of applying onto pharmaceutical synthesis. Moreover, scaled-up experiment confirmed the practical application ability.

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1. Reaction condition screening of catalytic system

Table S1 Ligand information

Entry	Code	Name
1	A	2,9-dimethyl-1,10-phenanthroline
2	B	2,2'-bipyridine
3	C	2,2'-bipyrimidine
4	D	5 <i>H</i> -cyclopenta[2,1- <i>b</i> :3,4- <i>b'</i>]dipyridin-5-one
5	E	4-methyl-2-(pyridin-2-yl)-4,5-dihydrooxazole
6	F	(<i>S</i>)-4-isopropyl-2-(pyridin-2-yl)-4,5-dihydrooxazole
7	G	2,2'-(propane-2,2-diyl)bis(4,5-dihydrooxazole)
8	H	2,2'-(propane-2,2-diyl)bis(4-isopropyl-4,5-dihydrooxazole)
9	I	<i>N</i> ¹ , <i>N</i> ¹ , <i>N</i> ² , <i>N</i> ² -tetramethylethane-1,2-diamine
10	J	<i>N</i> ¹ , <i>N</i> ¹ , <i>N</i> ³ , <i>N</i> ³ -tetramethylpropane-1,3-diamine
11	K	<i>N</i> ¹ , <i>N</i> ¹ , <i>N</i> ⁴ , <i>N</i> ⁴ -tetramethylbutane-1,4-diamine
12	L	<i>N</i> ¹ , <i>N</i> ¹ , <i>N</i> ⁵ , <i>N</i> ⁵ -tetramethylpentane-1,5-diamine
13	M	triethylamine
14	N	1 <i>H</i> -imidazole
15	O	(1 <i>E</i> ,2 <i>E</i>)- <i>N</i> ¹ , <i>N</i> ² -dicyclohexylethane-1,2-diimine
16	P	7-(tert-butyl)-2,5-dimethyl-3,4-dihydro-2 <i>H</i> -pyrano[2,3- <i>b</i>]quinoline
17	Q	1,2-bis(phenylsulfinyl)ethane
18	R	ethyl(4-methoxyphenyl)sulfane
19	S	triphenylphosphane
20	T	1,2-bis(diphenylphosphanyl)ethane
21	U	2,2'-bis(diphenylphosphanyl)-1,1'-binaphthalene
22	V	(oxybis(2,1-phenylene))bis(diphenylphosphane)

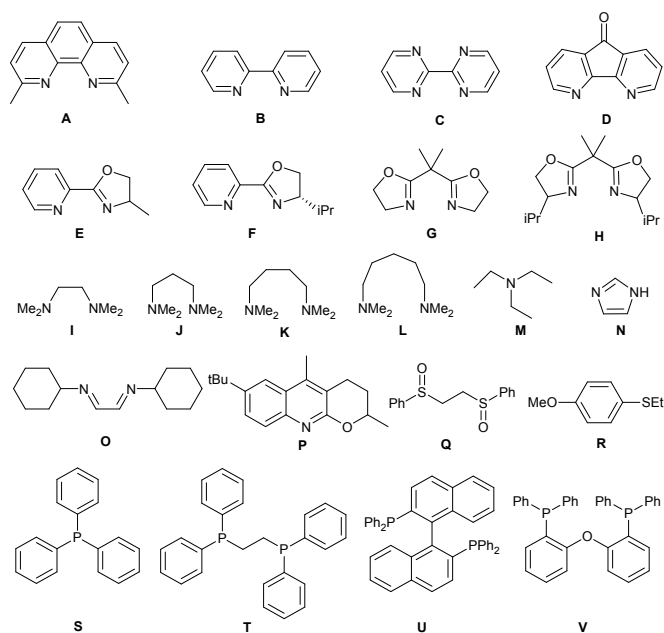
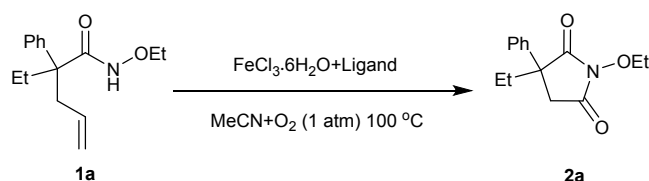
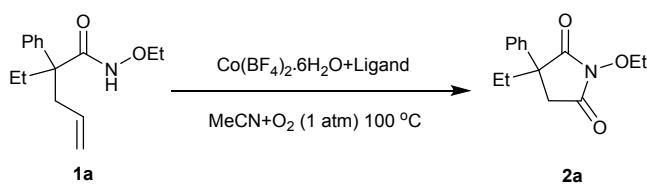


Table S2 Catalytic performance of $\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$: effect of ligands^a

Entry	Ligand Code	Yield/% ^b
1	A	0.5
2	B	Not detected
3	C	Not detected
4	D	0.2
5	E	0.5
6	F	Not detected
7	G	Not detected
8	H	Not detected
9	I	Not detected
10	J	Not detected
11	K	Not detected
12	L	Not detected
13	M	Not detected
14	N	Not detected
15	O	Not detected
16	P	0.8
17	Q	Not detected
18	R	Not detected
19	S	Not detected
20	T	Not detected
21	U	Not detected
22	V	Not detected

^a Reaction conditions: **1a**, 0.1 mmol; $\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$, 0.01 mmol (10 mol%); Ligand, 0.015 mmol (15 mol%); MeCN, 1.5 mL; O_2 (1 atm); $100\text{ }^\circ\text{C}$; 24 h.

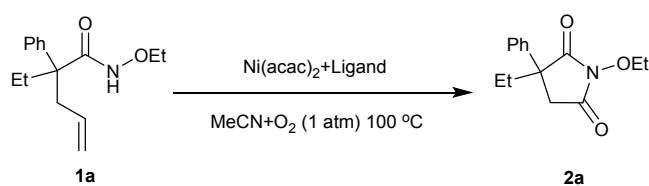
^b Determined by ^1H NMR analysis using 1,1,2,2-tetrachloroethane as an internal standard.

Table S3 Catalytic performance of $\text{Co}(\text{BF}_4)_2 \cdot 6\text{H}_2\text{O}$: effect of ligands^a

Entry	Ligand Code	Yield/% ^b
1	A	Not detected
2	B	Not detected
3	C	4.8
4	D	3.2
5	E	Not detected
6	F	Not detected
7	G	Not detected
8	H	Not detected
9	I	0.5
10	J	1
11	K	Not detected
12	L	Not detected
13	M	Not detected
14	N	Not detected
15	O	0.8
16	P	1.2
17	Q	Not detected
18	R	Not detected
19	S	Not detected
20	T	Not detected
21	U	Not detected
22	V	Not detected

^a Reaction conditions: **1a**, 0.1 mmol; $\text{Co}(\text{BF}_4)_2 \cdot 6\text{H}_2\text{O}$, 0.01 mmol (10 mol%); Ligand, 0.015 mmol (15 mol%); MeCN , 1.5 mL; O_2 (1 atm); $100\text{ }^\circ\text{C}$; 24 h.

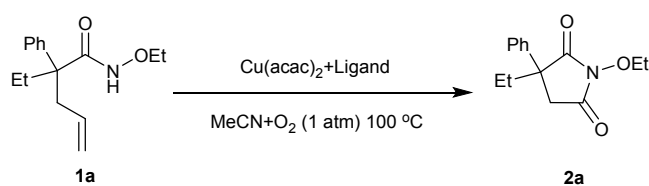
^b Determined by ^1H NMR analysis using 1,1,2,2-tetrachloroethane as an internal standard.

Table S4 Catalytic performance of Ni(acac)₂: effect of ligands^a

Entry	Ligand Code	Yield/% ^b
1	A	0.5
2	B	0.4
3	C	0.6
4	D	Not detected
5	E	Not detected
6	F	Not detected
7	G	Not detected
8	H	Not detected
9	I	Not detected
10	J	Not detected
11	K	Not detected
12	L	0.5
13	M	0.8
14	N	0.6
15	O	Not detected
16	P	0.5
17	Q	Not detected
18	R	Not detected
19	S	Not detected
20	T	Not detected
21	U	Not detected
22	V	Not detected

^a Reaction conditions: **1a**, 0.1 mmol; Ni(acac)₂, 0.01 mmol (10 mol%); Ligand, 0.015 mmol (15 mol%); MeCN, 1.5 mL; O₂ (1 atm); 100 °C; 24 h.

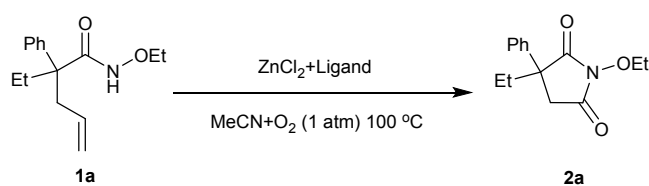
^b Determined by ¹H NMR analysis using 1,1,2,2-tetrachloroethane as an internal standard.

Table S5 Catalytic performance of Cu(acac)₂: effect of ligands^a

Entry	Ligand Code	Yield/% ^b
1	A	85
2	B	63
3	C	72
4	D	28
5	E	51
6	F	44
7	G	42
8	H	25
9	I	8
10	J	6
11	K	5
12	L	3
13	M	5
14	N	33
15	O	8
16	P	18
17	Q	2
18	R	1
19	S	Not detected
20	T	Not detected
21	U	Not detected
22	V	Not detected

^a Reaction conditions: **1a**, 0.1 mmol; Cu(acac)₂, 0.01 mmol (10 mol%); Ligand, 0.015 mmol (15 mol%); MeCN, 1.5 mL; O₂ (1 atm); 100 °C; 24 h.

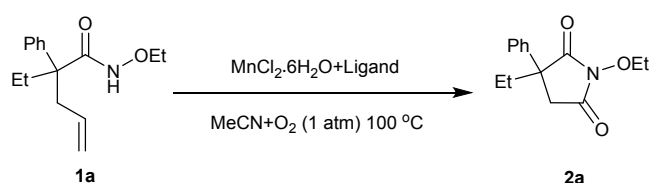
^b Determined by ¹H NMR analysis using 1,1,2,2-tetrachloroethane as an internal standard.

Table S6 Catalytic performance of ZnCl₂: effect of ligands^a

Entry	Ligand Code	Yield/% ^b
1	A	0.5
2	B	0.5
3	C	0.3
4	D	Not detected
5	E	Not detected
6	F	Not detected
7	G	Not detected
8	H	Not detected
9	I	Not detected
10	J	Not detected
11	K	Not detected
12	L	0.8
13	M	0.4
14	N	0.3
15	O	Not detected
16	P	0.5
17	Q	0.6
18	R	Not detected
19	S	Not detected
20	T	Not detected
21	U	Not detected
22	V	Not detected

^a Reaction conditions: **1a**, 0.1 mmol; ZnCl₂, 0.01 mmol (10 mol%); Ligand, 0.015 mmol (15 mol%); MeCN, 1.5 mL; O₂ (1 atm); 100 °C; 24 h.

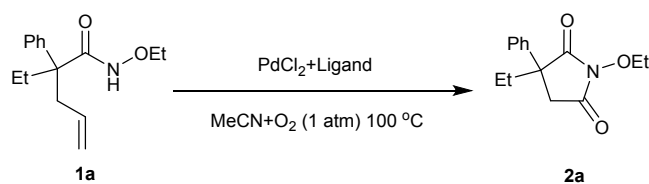
^b Determined by ¹H NMR analysis using 1,1,2,2-tetrachloroethane as an internal standard.

Table S7 Catalytic performance of $\text{MnCl}_2 \cdot 4\text{H}_2\text{O}$: effect of ligands^a

Entry	Ligand Code	Yield/% ^b
1	A	Not detected
2	B	Not detected
3	C	Not detected
4	D	Not detected
5	E	Not detected
6	F	0.4
7	G	0.5
8	H	0.6
9	I	0.3
10	J	Not detected
11	K	Not detected
12	L	Not detected
13	M	Not detected
14	N	Not detected
15	O	Not detected
16	P	Not detected
17	Q	0.3
18	R	0.5
19	S	Not detected
20	T	Not detected
21	U	Not detected
22	V	Not detected

^a Reaction conditions: **1a**, 0.1 mmol; $\text{MnCl}_2 \cdot 4\text{H}_2\text{O}$, 0.01 mmol (10 mol%); Ligand, 0.015 mmol (15 mol%); MeCN, 1.5 mL; O_2 (1 atm); $100\text{ }^\circ\text{C}$; 24 h.

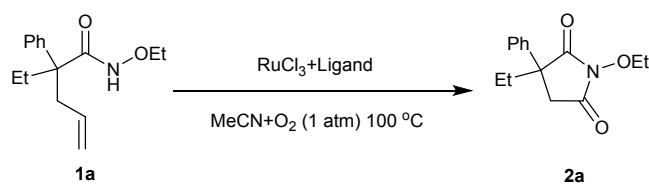
^b Determined by ^1H NMR analysis using 1,1,2,2-tetrachloroethane as an internal standard.

Table S8 Catalytic performance of PdCl₂: effect of ligands^a

Entry	Ligand Code	Yield/% ^b
1	A	0.5
2	B	0.6
3	C	5.6
4	D	3.2
5	E	Not detected
6	F	Not detected
7	G	Not detected
8	H	Not detected
9	I	Not detected
10	J	Not detected
11	K	Not detected
12	L	1.2
13	M	Not detected
14	N	1
15	O	Not detected
16	P	0.8
17	Q	0.5
18	R	Not detected
19	S	Not detected
20	T	Not detected
21	U	Not detected
22	V	Not detected

^a Reaction conditions: **1a**, 0.1 mmol; PdCl₂, 0.01 mmol (10 mol%); Ligand, 0.015 mmol (15 mol%); MeCN, 1.5 mL; O₂ (1 atm); 100 °C; 24 h.

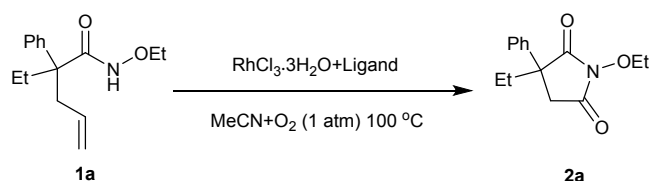
^b Determined by ¹H NMR analysis using 1,1,2,2-tetrachloroethane as an internal standard.

Table S9 Catalytic performance of RuCl₃: effect of ligands^a

Entry	Ligand Code	Yield/% ^b
1	A	Not detected
2	B	Not detected
3	C	Not detected
4	D	Not detected
5	E	Not detected
6	F	Not detected
7	G	Not detected
8	H	4.8
9	I	Not detected
10	J	1
11	K	Not detected
12	L	Not detected
13	M	Not detected
14	N	0.5
15	O	1.5
16	P	2
17	Q	0.5
18	R	Not detected
19	S	Not detected
20	T	Not detected
21	U	Not detected
22	V	Not detected

^a Reaction conditions: **1a**, 0.1 mmol; RuCl₃, 0.01 mmol (10 mol%); Ligand, 0.015 mmol (15 mol%); MeCN, 1.5 mL; O₂ (1 atm); 100 °C; 24 h.

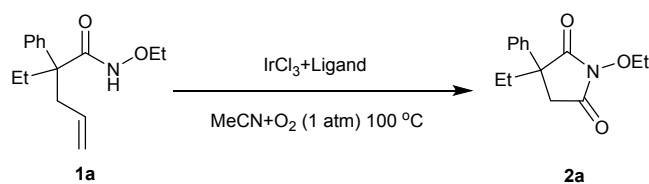
^b Determined by ¹H NMR analysis using 1,1,2,2-tetrachloroethane as an internal standard.

Table S10 Catalytic performance of $\text{RhCl}_3 \cdot 3\text{H}_2\text{O}$: effect of ligands^a

Entry	Ligand Code	Yield/% ^b
1	A	Not detected
2	B	Not detected
3	C	1
4	D	0.5
5	E	Not detected
6	F	0.6
7	G	5.5
8	H	Not detected
9	I	Not detected
10	J	Not detected
11	K	Not detected
12	L	Not detected
13	M	Not detected
14	N	Not detected
15	O	Not detected
16	P	0.5
17	Q	Not detected
18	R	0.2
19	S	Not detected
20	T	Not detected
21	U	Not detected
22	V	Not detected

^a Reaction conditions: **1a**, 0.1 mmol; $\text{RhCl}_3 \cdot 3\text{H}_2\text{O}$, 0.01 mmol (10 mol%); Ligand, 0.015 mmol (15 mol%); MeCN, 1.5 mL; O_2 (1 atm); 100 °C; 24 h.

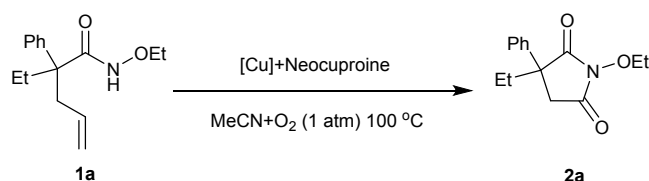
^b Determined by ^1H NMR analysis using 1,1,2,2-tetrachloroethane as an internal standard.

Table S11 Catalytic performance of IrCl₃: effect of ligands^a

Entry	Ligand Code	Yield/% ^b
1	A	3.2
2	B	0.5
3	C	0.2
4	D	Not detected
5	E	Not detected
6	F	Not detected
7	G	Not detected
8	H	Not detected
9	I	0.5
10	J	Not detected
11	K	1
12	L	Not detected
13	M	0.5
14	N	Not detected
15	O	Not detected
16	P	Not detected
17	Q	Not detected
18	R	Not detected
19	S	Not detected
20	T	Not detected
21	U	Not detected
22	V	Not detected

^a Reaction conditions: **1a**, 0.1 mmol; IrCl₃, 0.01 mmol (10 mol%); Ligand, 0.015 mmol (15 mol%); MeCN, 1.5 mL; O₂ (1 atm); 100 °C; 24 h.

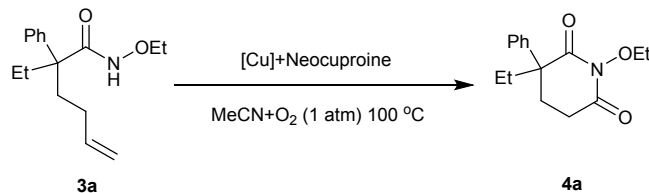
^b Determined by ¹H NMR analysis using 1,1,2,2-tetrachloroethane as an internal standard.

Table S12 Screening of Cu salts^a

Entry	Cu Salt	Yield/% ^b
1	CuSO ₄ ·5H ₂ O	75
2	Cu(NO ₃) ₂ ·2.5H ₂ O	78
3	Cu(BF ₄) ₂ ·2H ₂ O	82
4	CuF ₂	81
5	CuCl ₂ ·2H ₂ O	80
6	CuBr ₂	77
7	Cu(acac) ₂	85
8	CuAc ₂ ·H ₂ O	78

^a Reaction conditions: **1a**, 0.1 mmol; Cu salts, 0.01 mmol (10 mol%); Neocuproine, 0.015 mmol (15 mol%); MeCN, 1.5 mL; O₂ (1 atm); 100 °C; 24 h.

^b Determined by ¹H NMR analysis using 1,1,2,2-tetrachloroethane as an internal standard.

Table S13 Screening of Cu salts^a

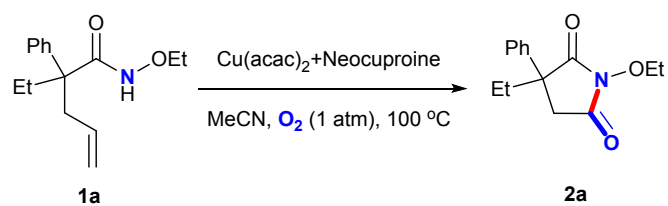
Entry	Cu Salt	Yield/% ^b
1	CuSO ₄ ·5H ₂ O	44
2	Cu(NO ₃) ₂ ·2.5H ₂ O	47
3	Cu(BF ₄) ₂ ·2H ₂ O	50
4	CuF ₂	58
5	CuCl ₂ ·2H ₂ O	49
6	CuBr ₂	45
7	Cu(acac) ₂	52
8	CuAc ₂ ·H ₂ O	48

^a Reaction conditions: **3a**, 0.1 mmol; Cu salts, 0.01 mmol (10 mol%); Neocuproine, 0.015 mmol (15 mol%); MeCN, 1.5 mL; O₂ (1 atm); 100 °C; 24 h.

^b Determined by ¹H NMR analysis using 1,1,2,2-tetrachloroethane as an internal standard.

2. Studies on the catalytic system

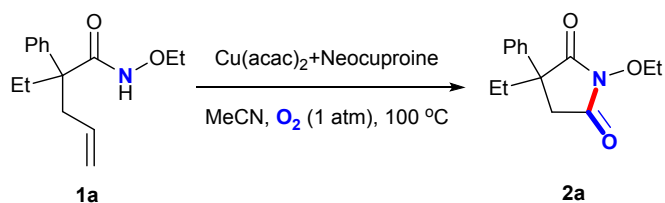
Table S14 Control experiments^a



Entry	Deviation from standard conditions	2a yield / % ^b
1	No Cu(acac) ₂	0
2	No neocuproine	0
3	No O ₂	0
4	No neocuproine or O ₂	0

^a Standard conditions: **1a**, 0.1 mmol; Cu(acac)₂, 0.01 mmol (10 mol%); ligand neocuproine, 0.015 mmol (15 mol%); MeCN, 1.5 mL; O₂ (1 atm); 100 °C; 24 h. ^b Determined by ¹H NMR analysis using 1,1,2,2-tetrachloroethane as an internal standard.

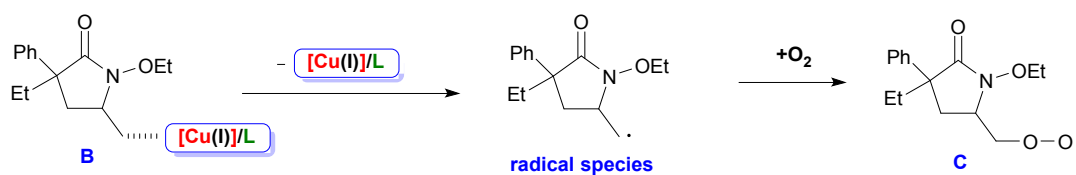
Table S15 Quantitative data for the kinetic studies^{a,b}



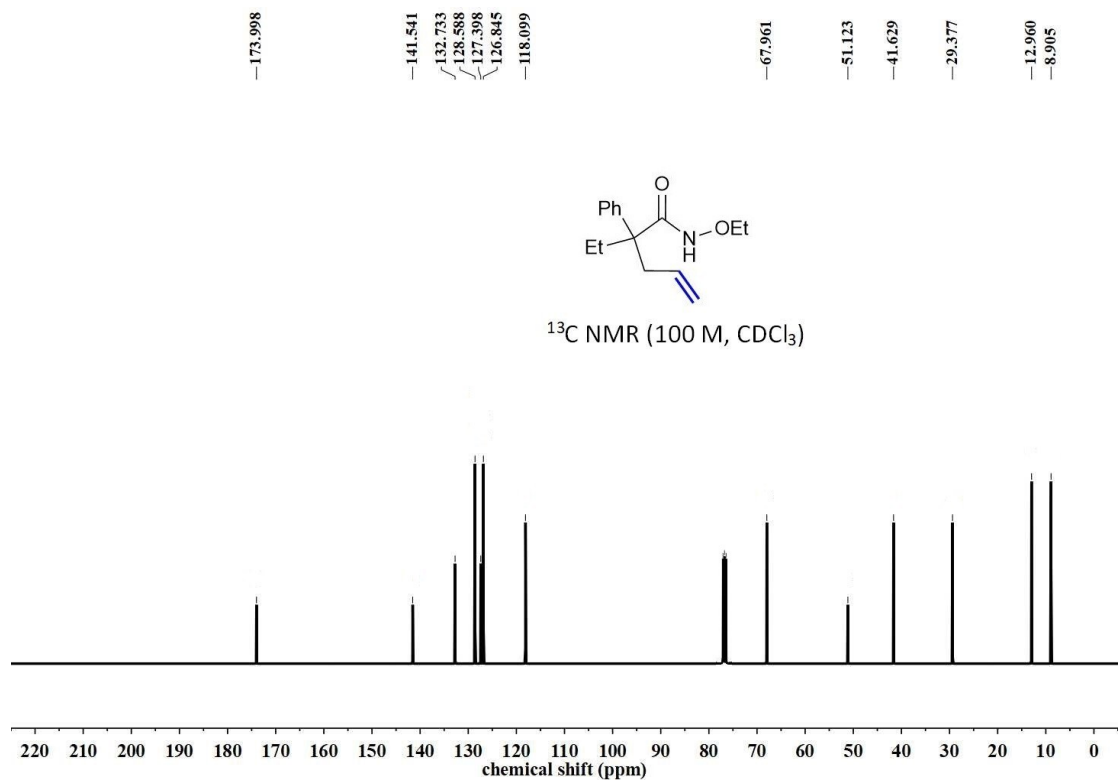
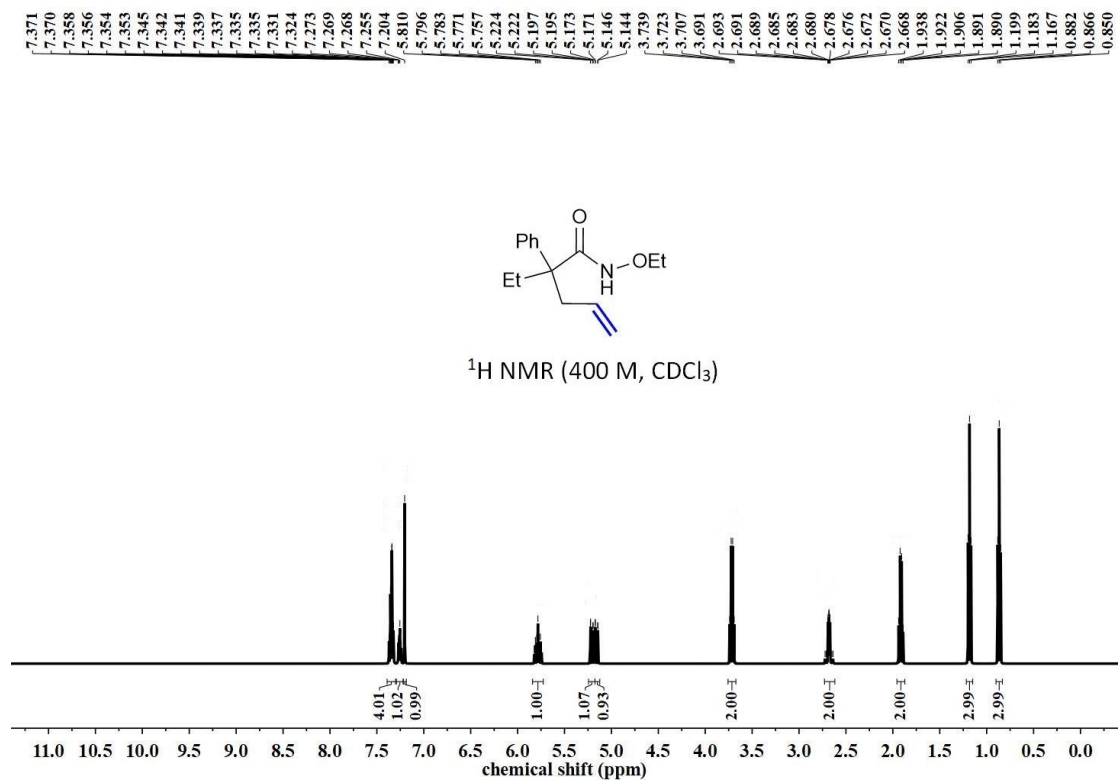
Entry	Time / h	1a / %	E / %	2a / %
1	0	100	0	0
2	0.5	60	30	8
3	1	40	27	18
4	1.5	35	21	25
5	2	25	18	38
6	3	19	16	49
7	6	17	13	58
8	9	16	10	63
9	12	15	8	68
10	15	12	5	72
11	18	10	3	79
12	21	9	2	82
13	24	8	2	85
14	27	8	2	85

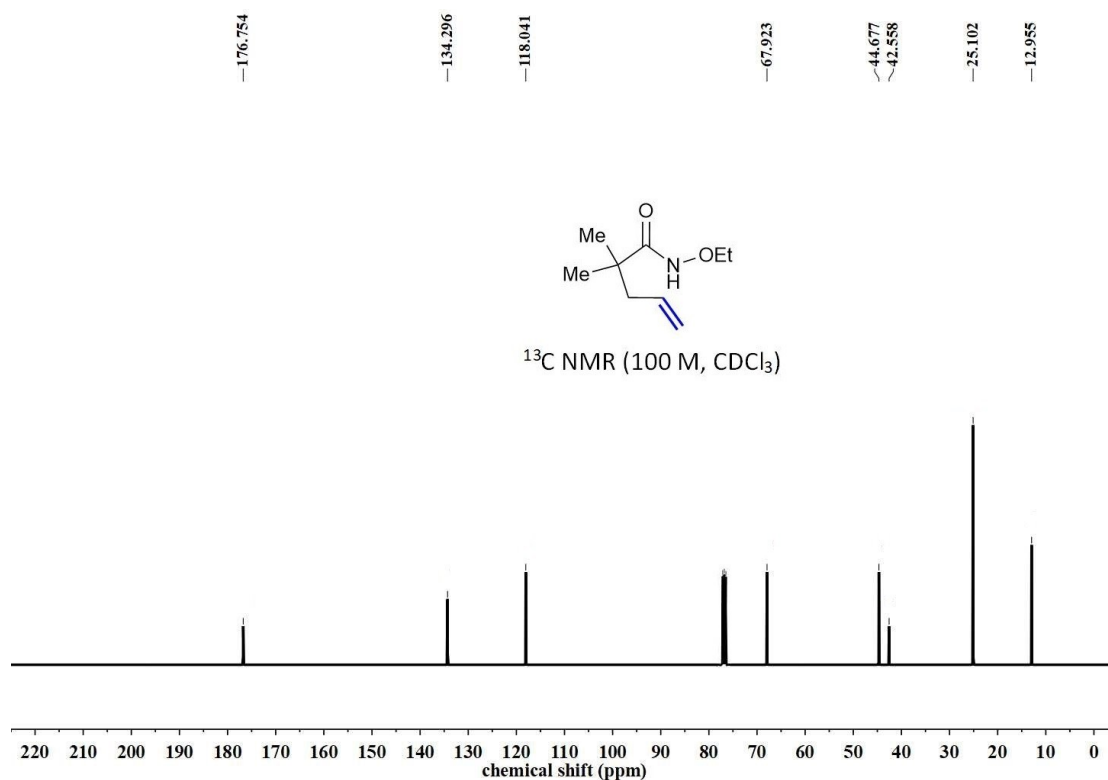
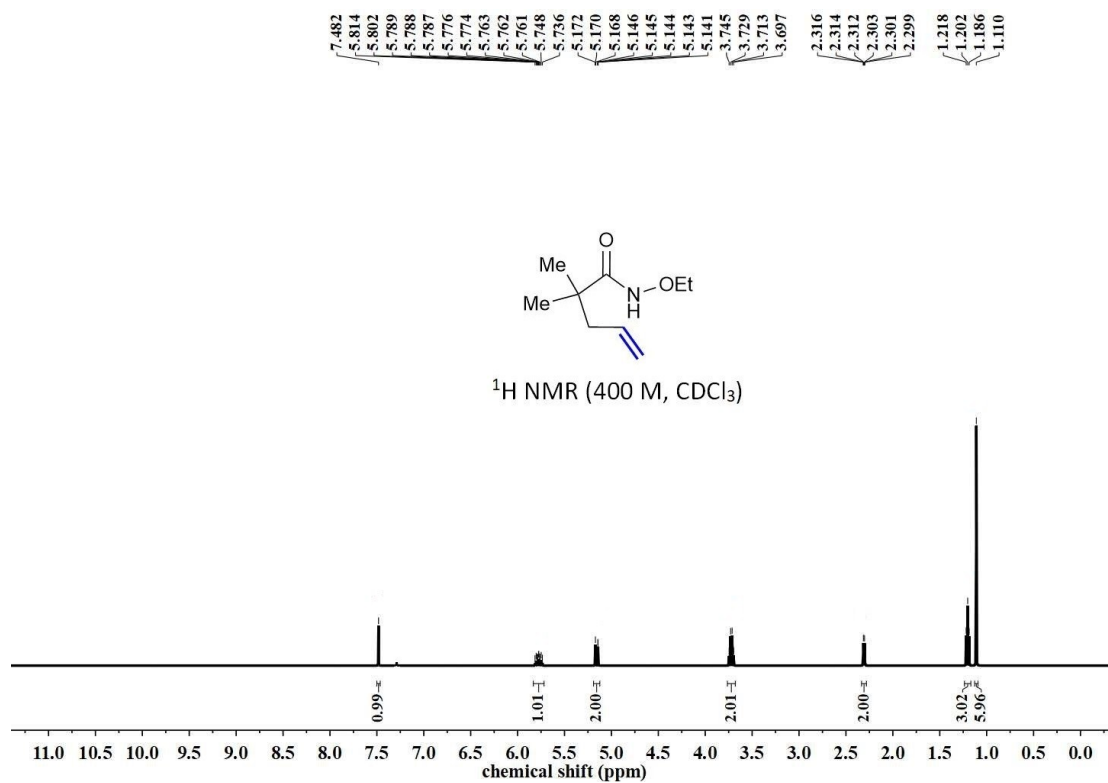
^a Reaction conditions: 1a, 0.1 mmol; Cu(acac)₂, 0.01 mmol (10 mol%); ligand neocuproine, 0.015 mmol (15 mol%); MeCN, 1.5 mL; O₂ (1 atm); 100 °C. ^b Yields were determined by ¹H NMR analysis using 1,1,2,2-tetrachloroethane as an internal standard.

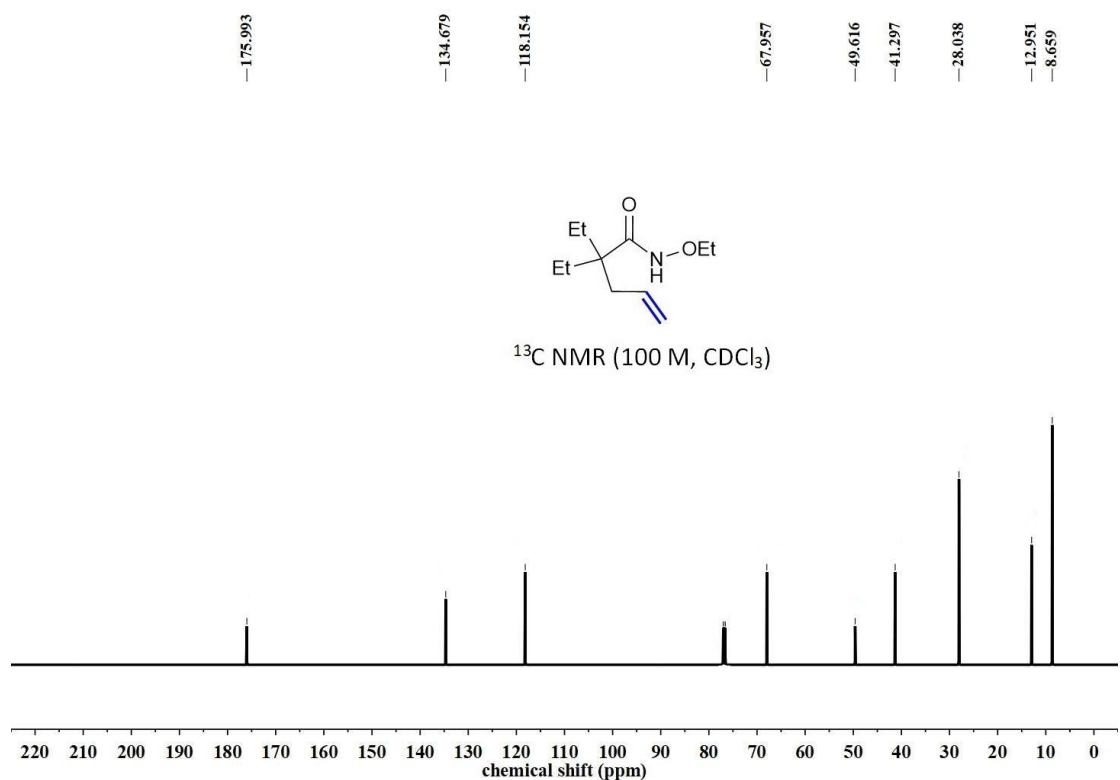
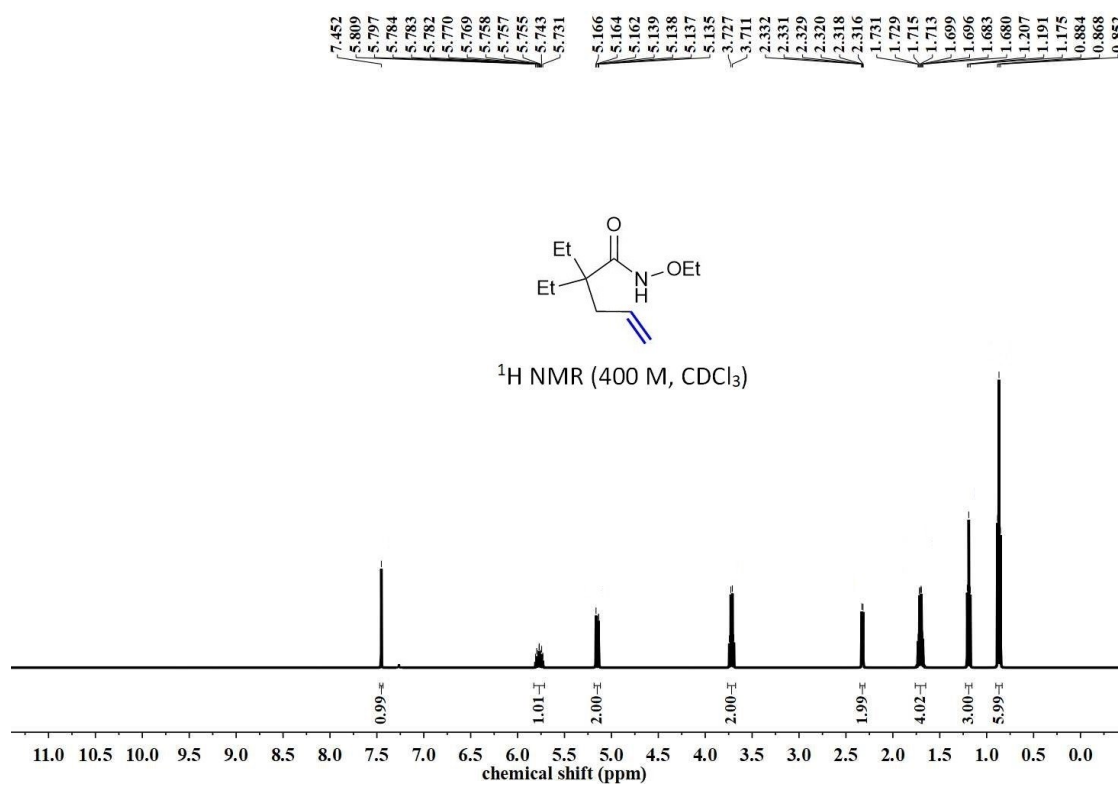
Scheme S1 Process between intermediate B and C

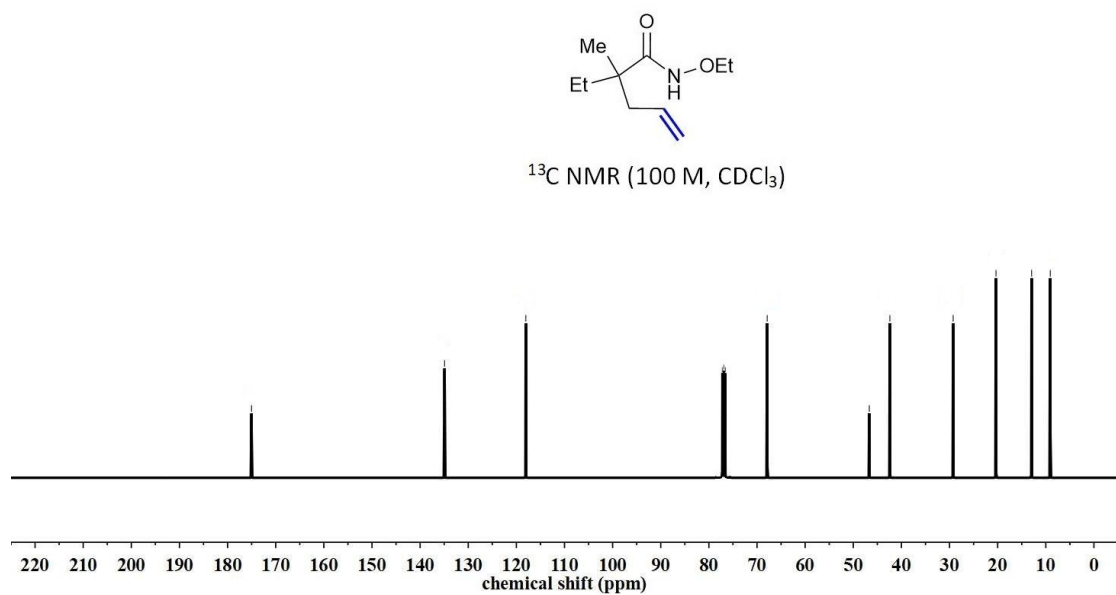
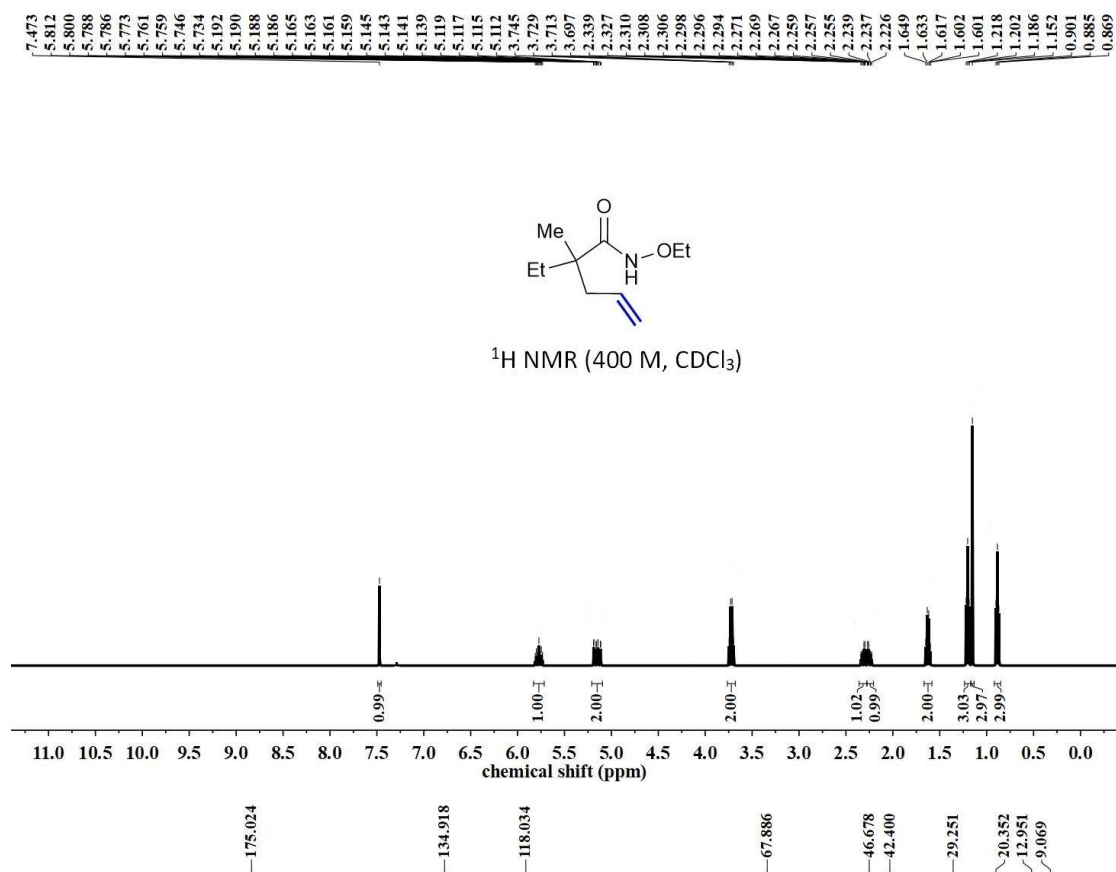


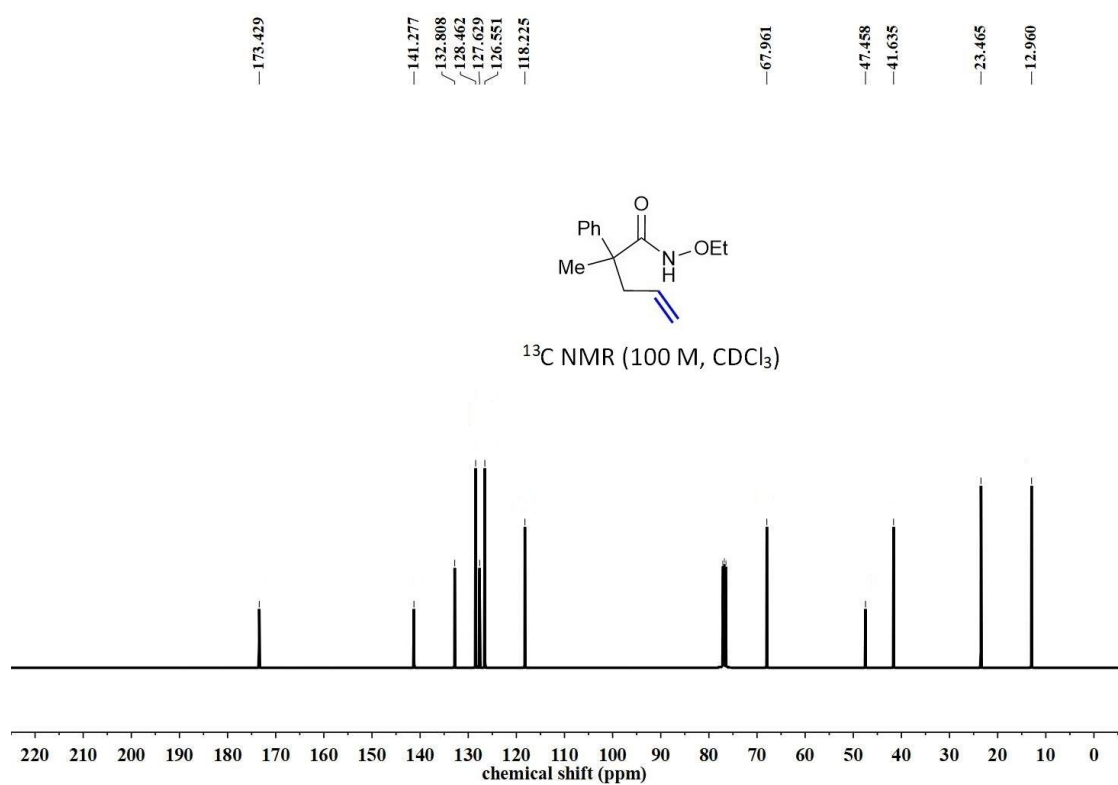
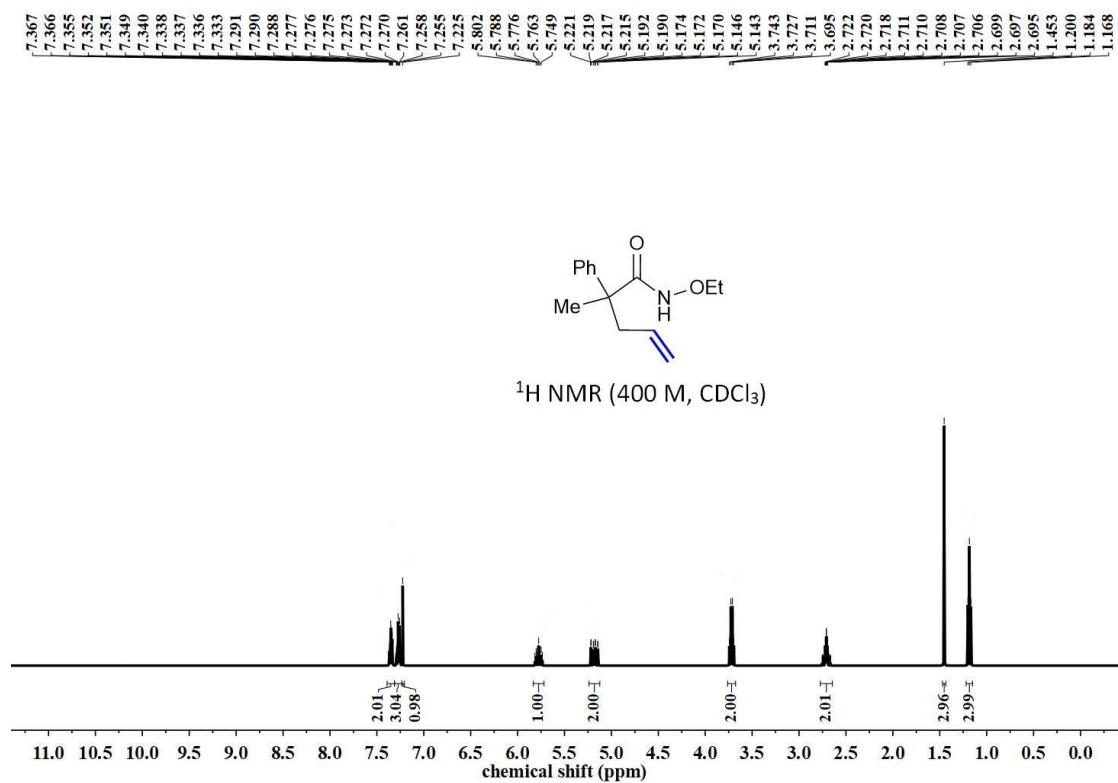
3. NMR spectra of substrates, products and intermediate



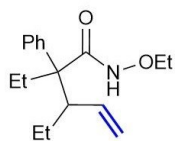




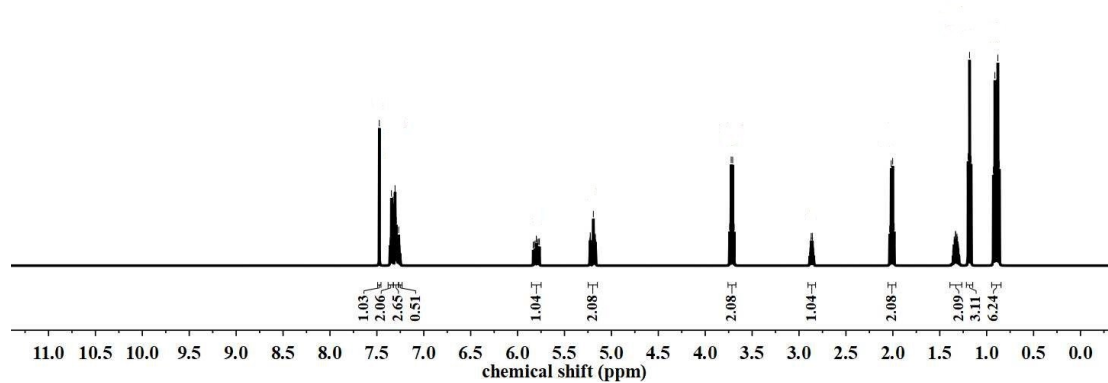




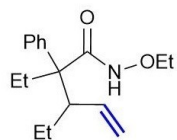
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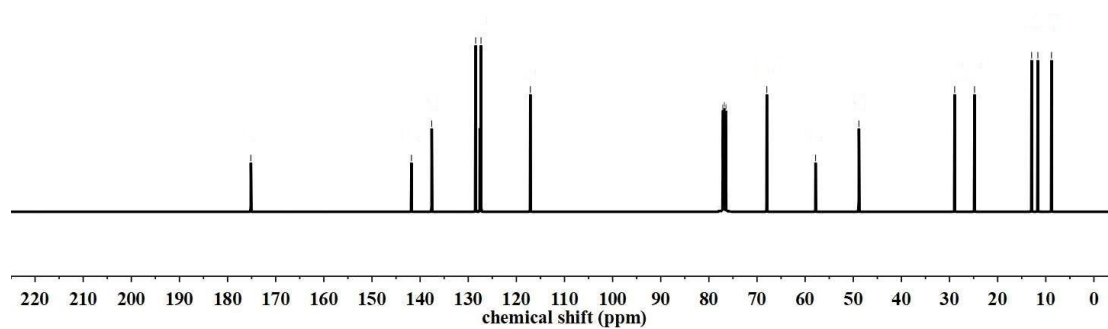
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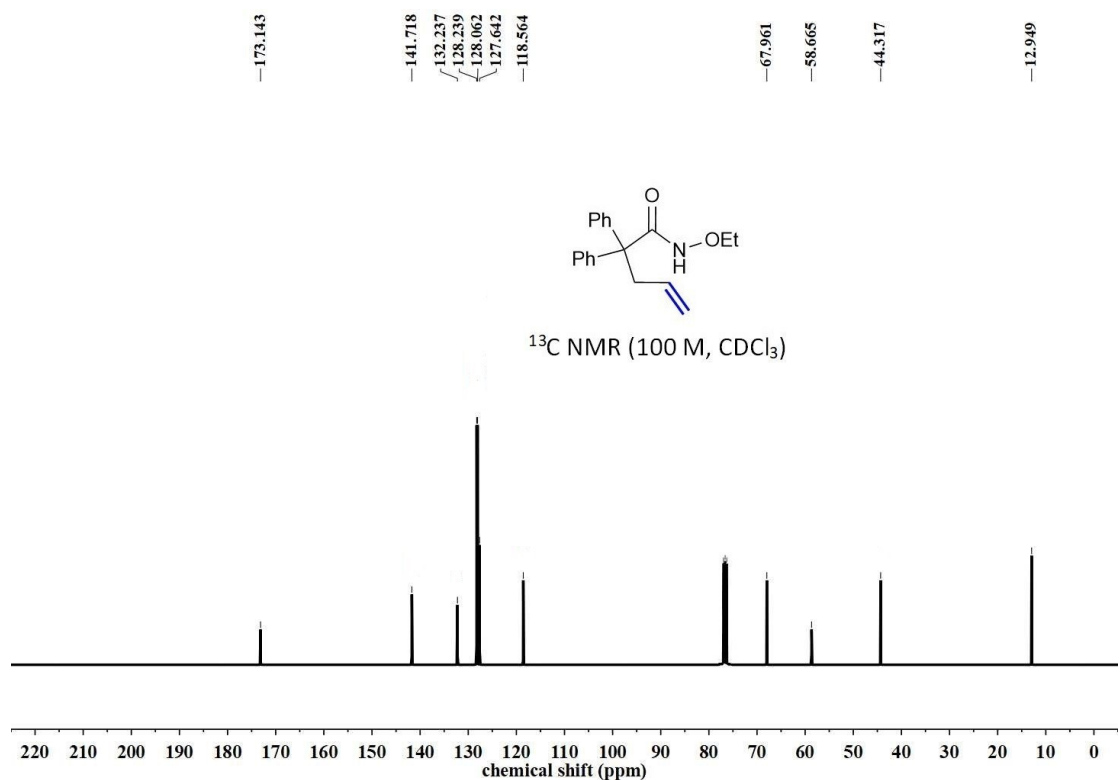
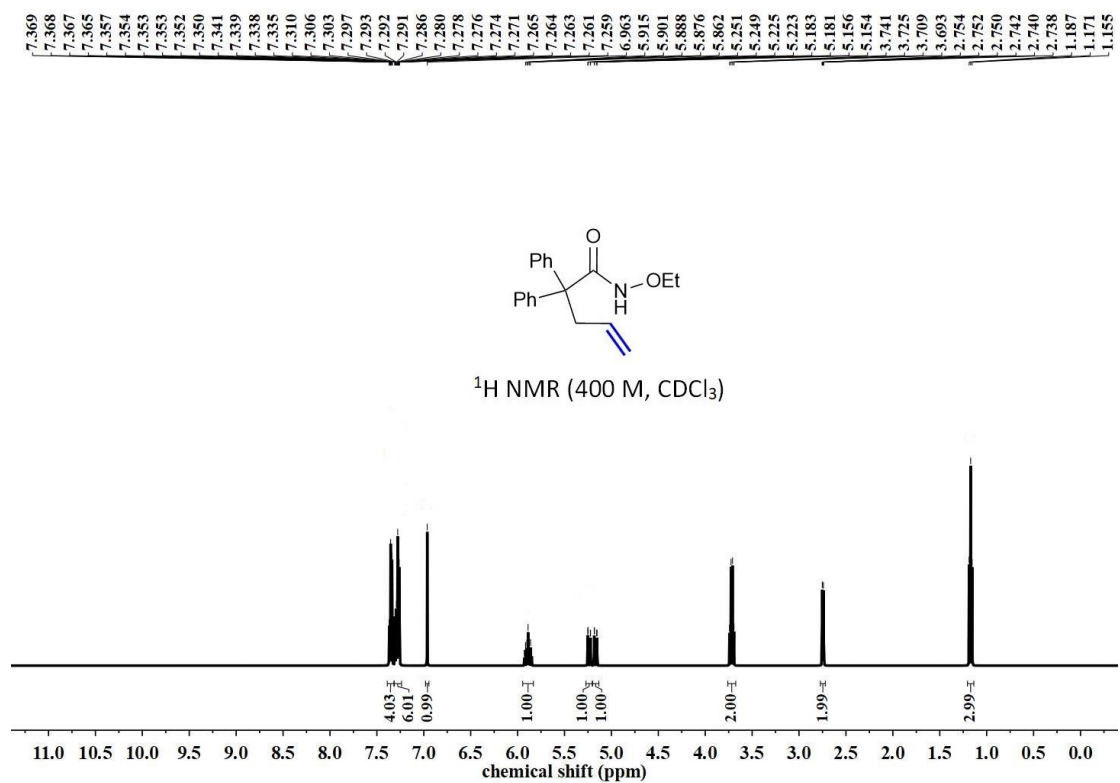


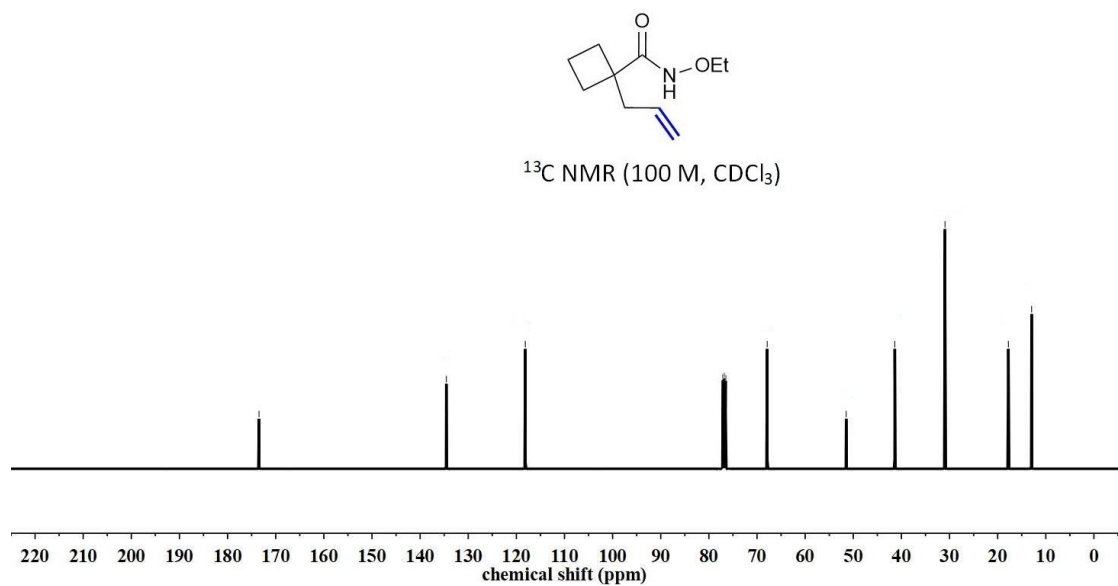
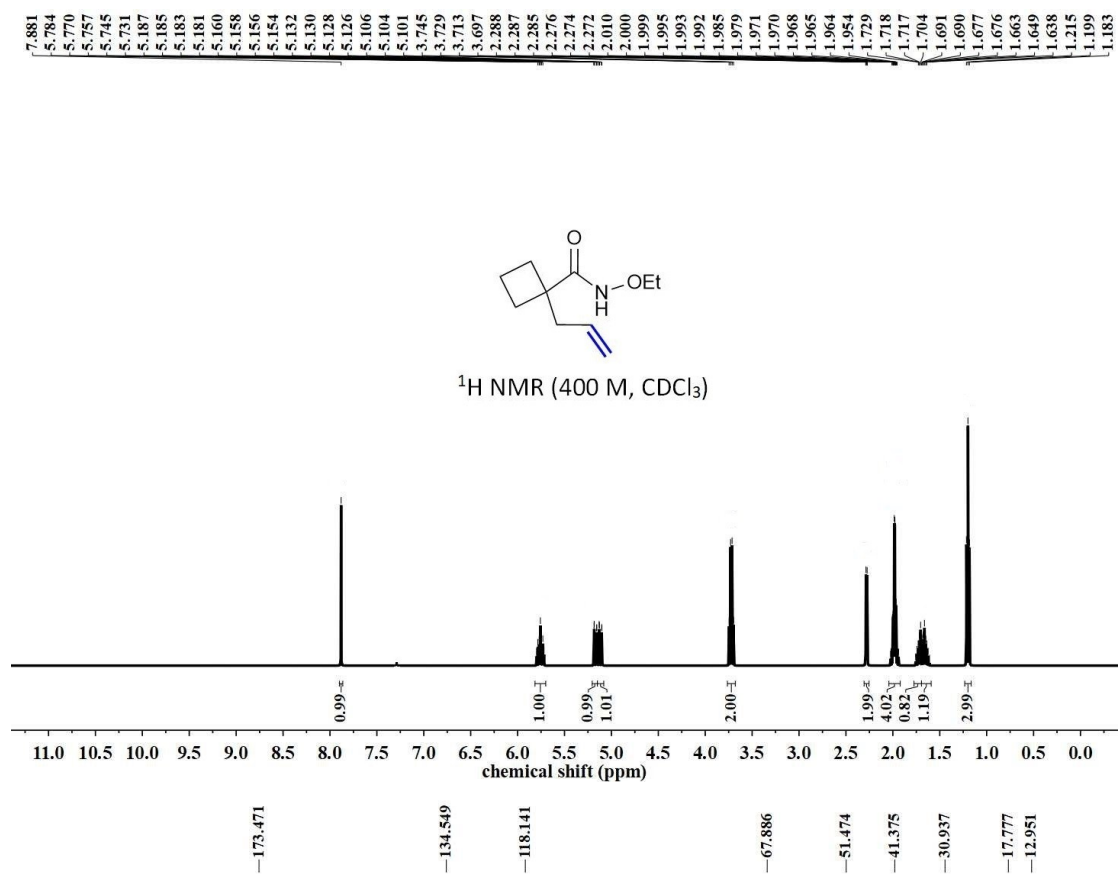
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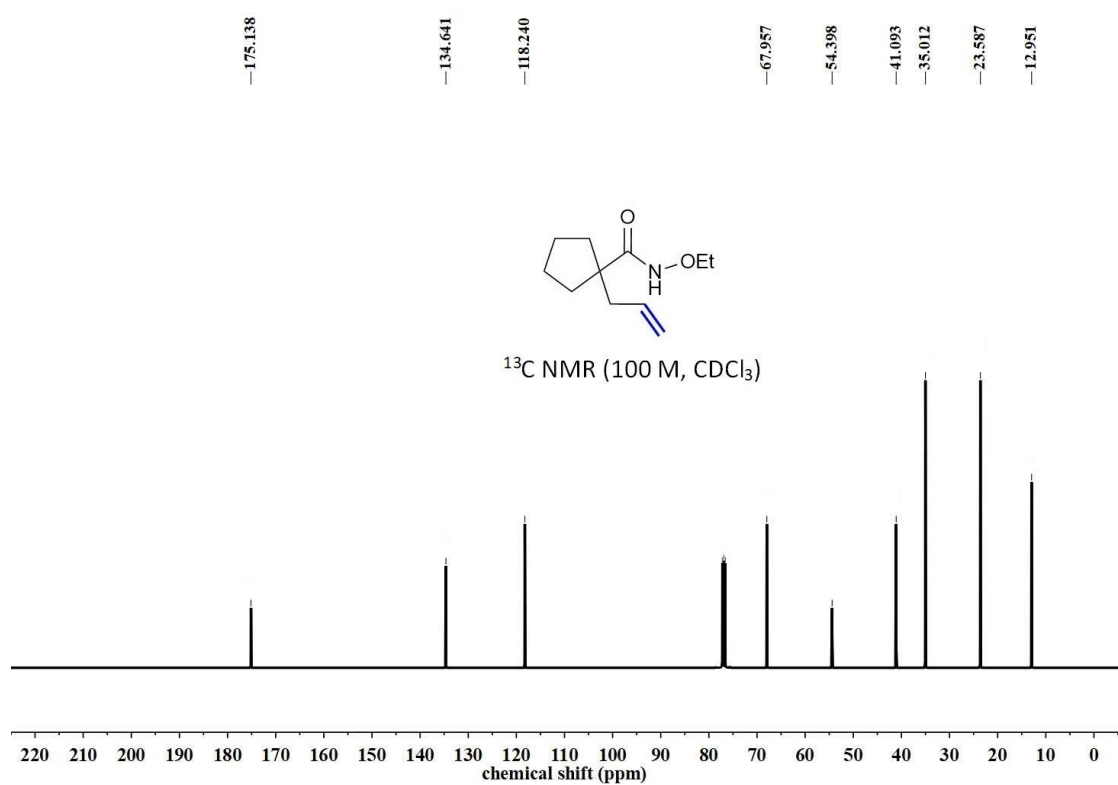
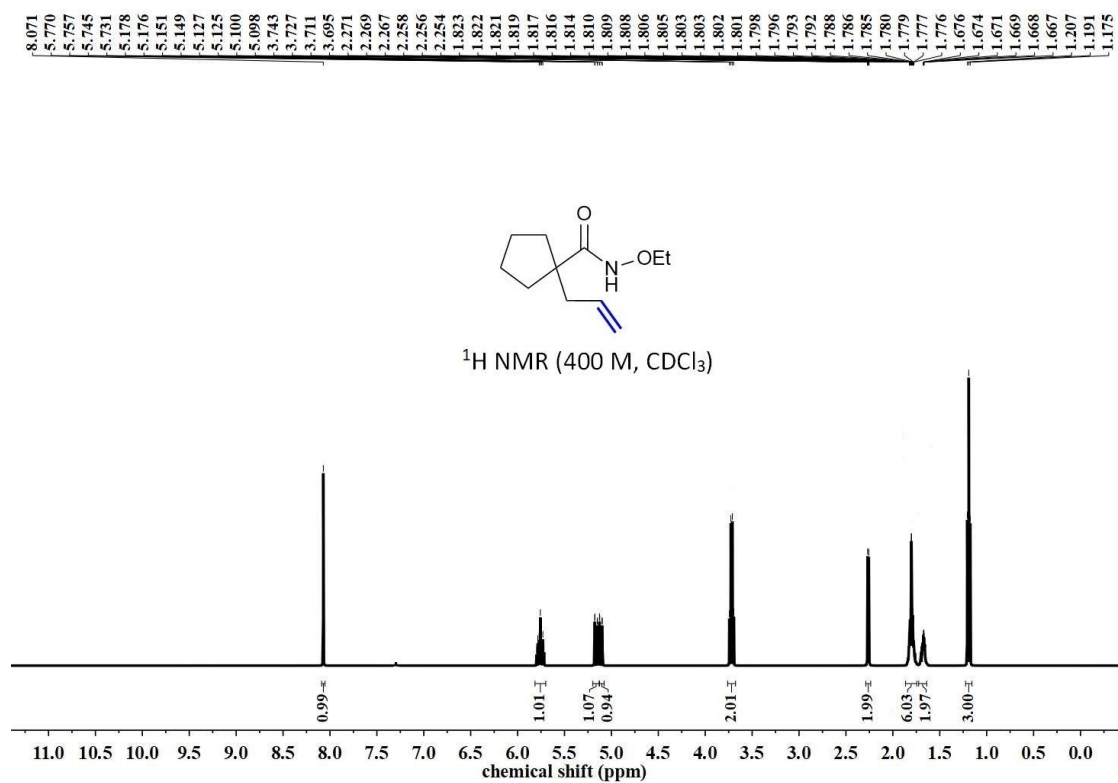


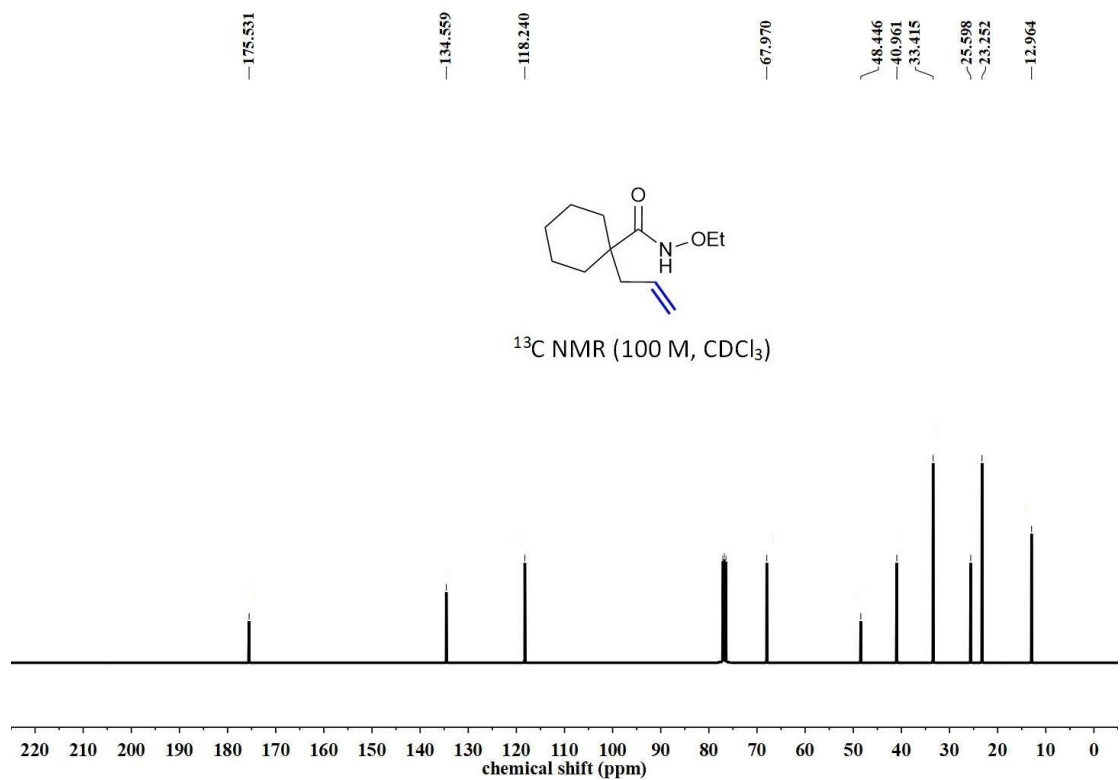
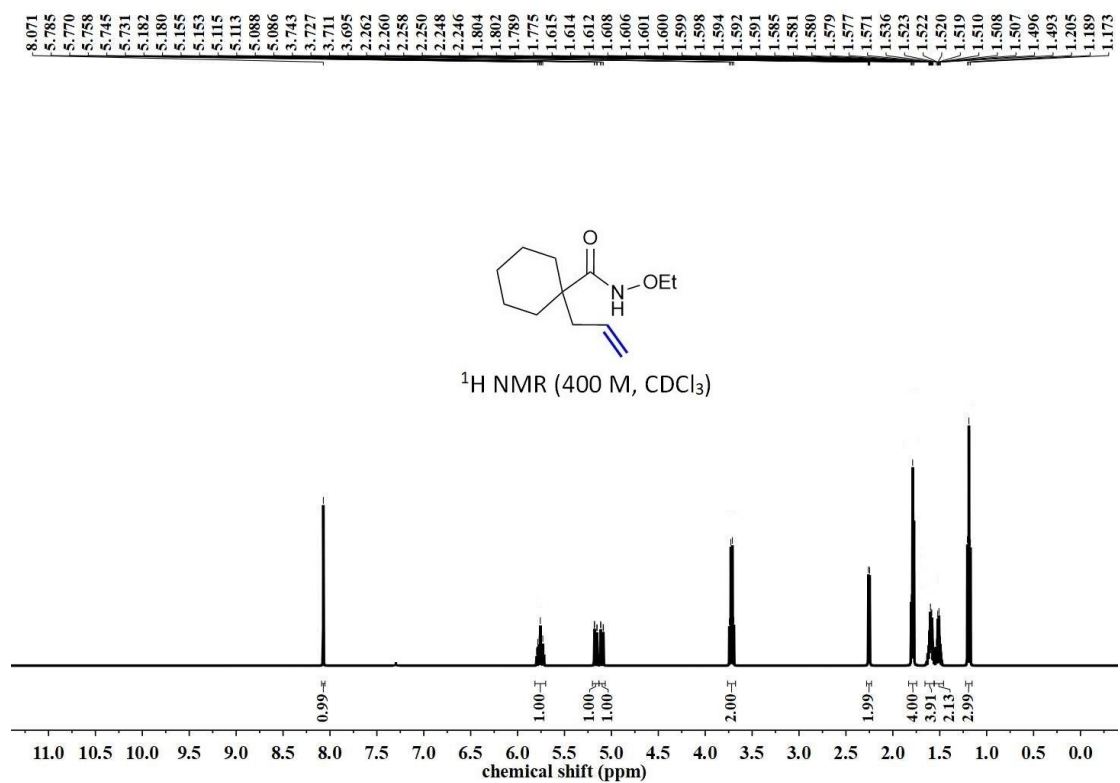
^{13}C NMR (100 M, CDCl_3)

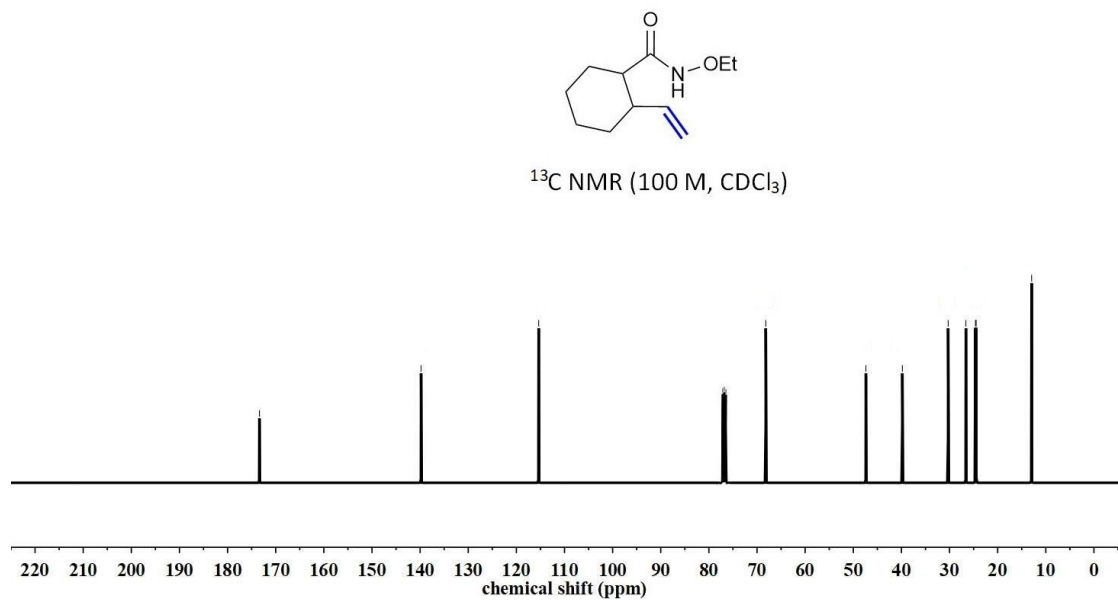
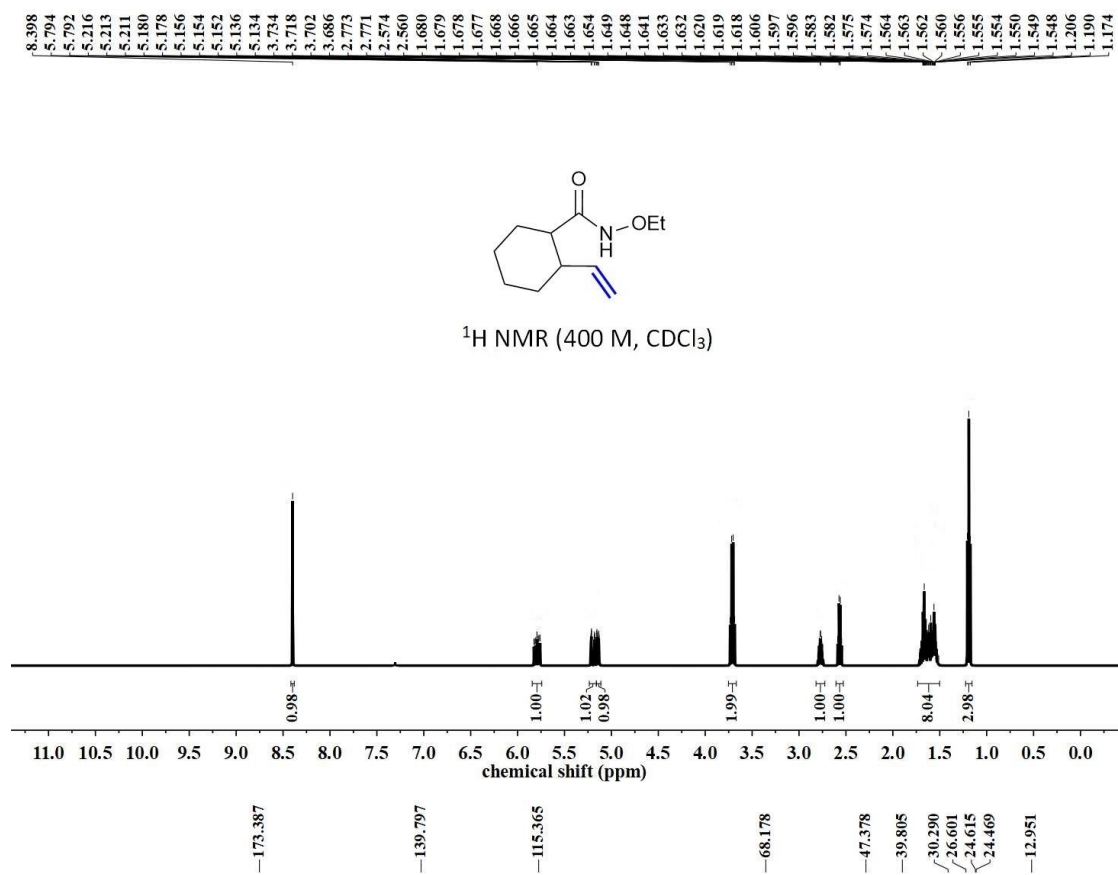


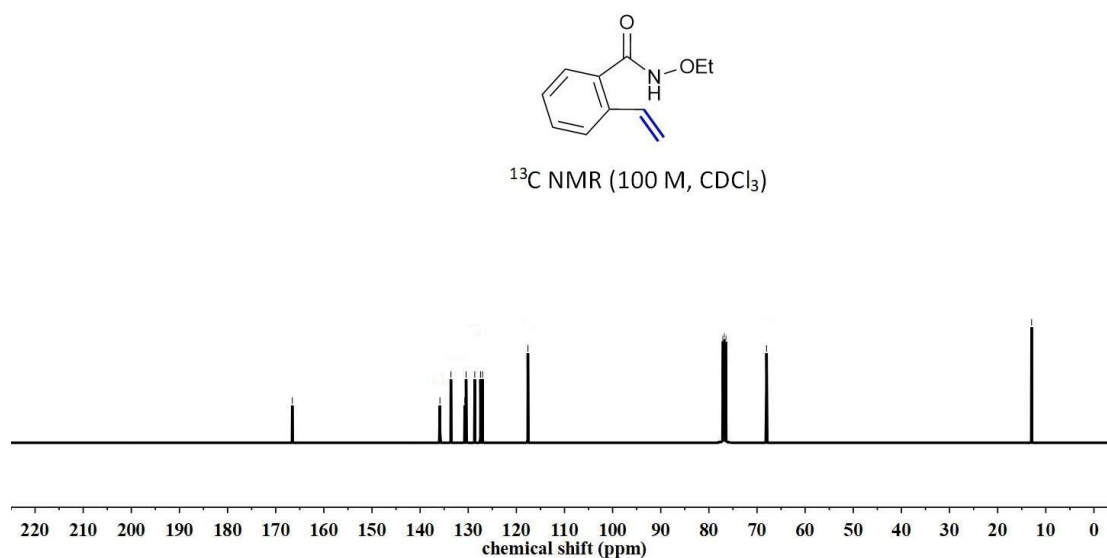
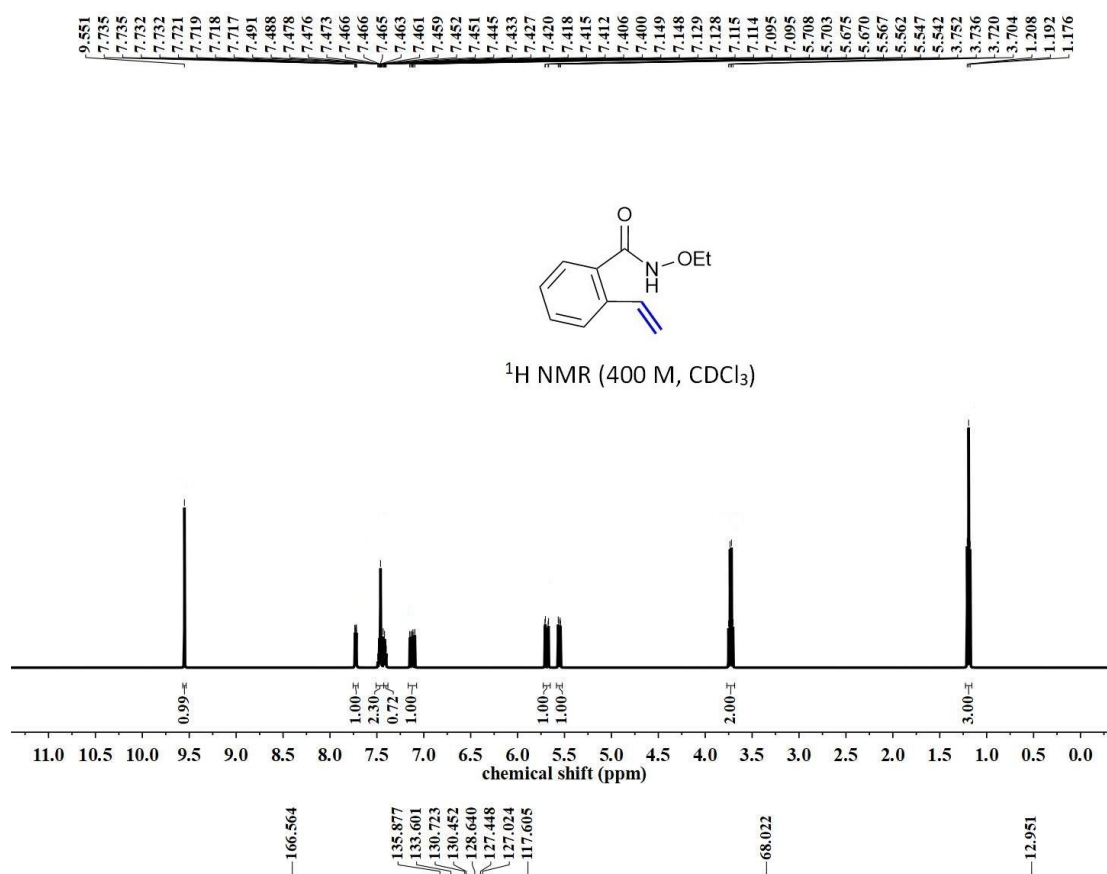


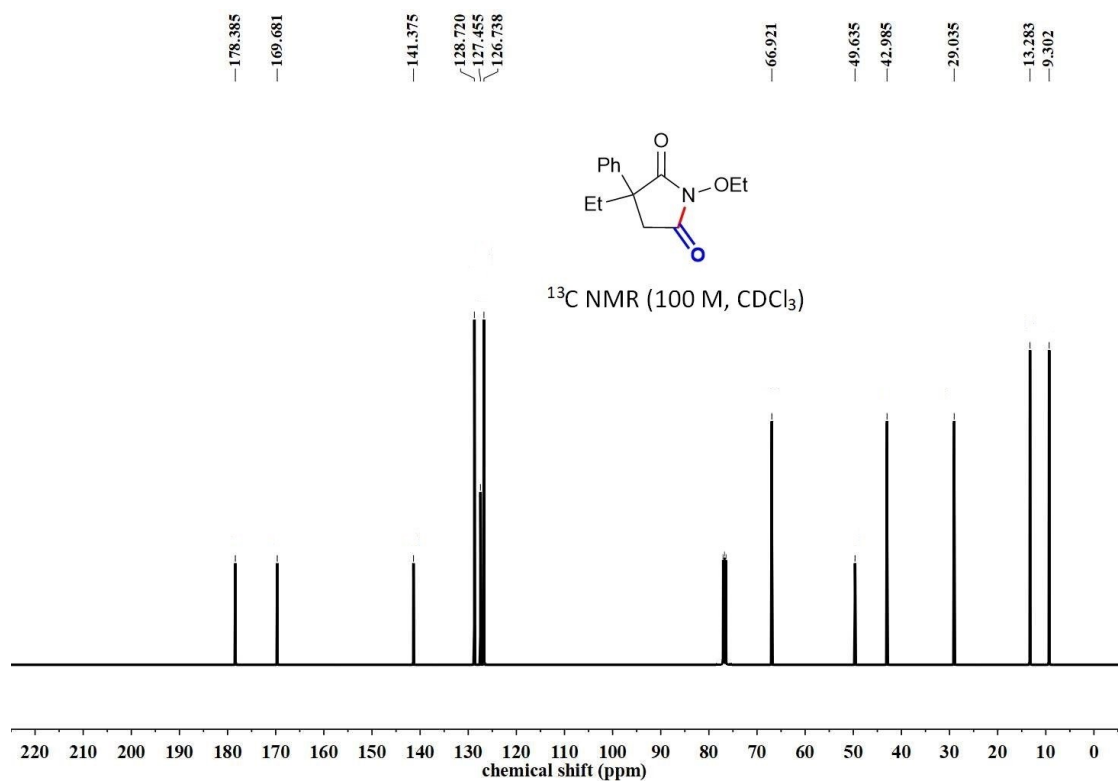
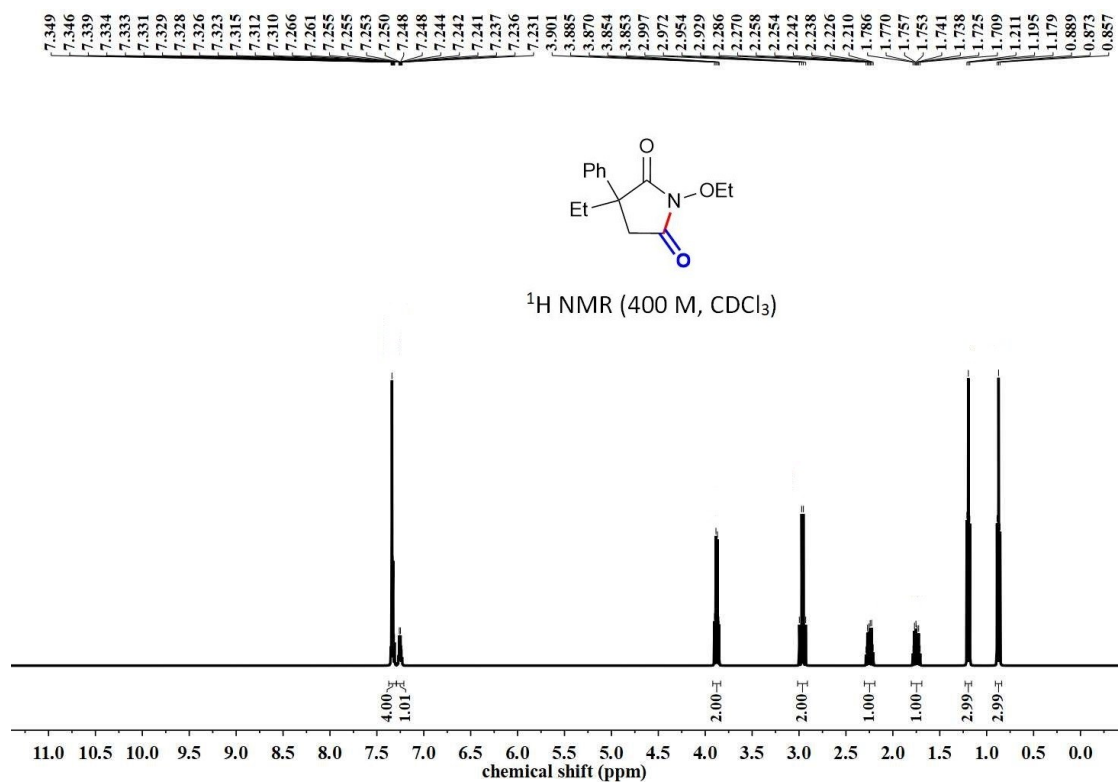


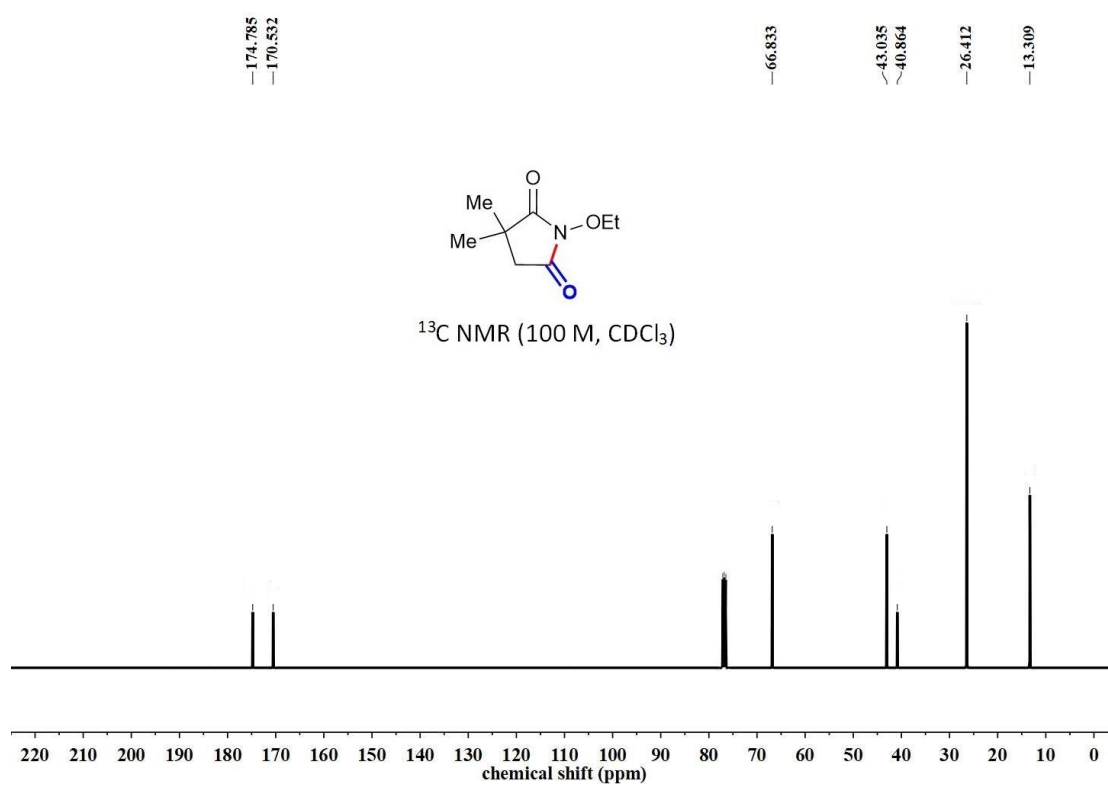
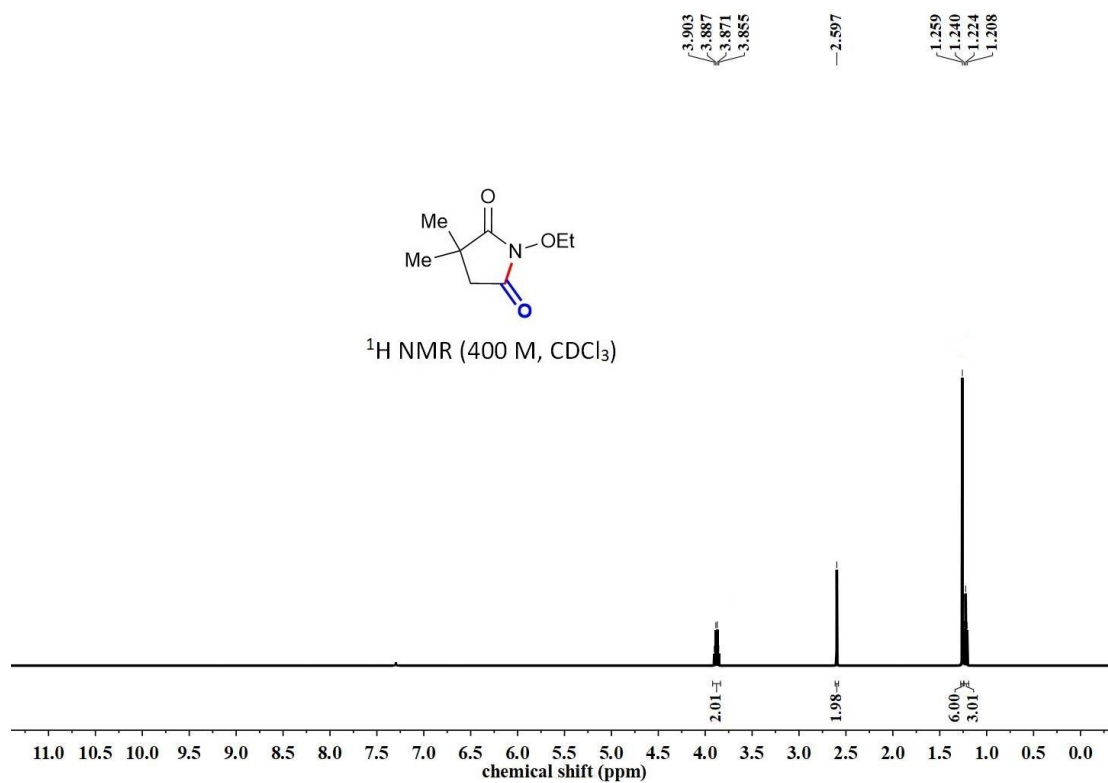


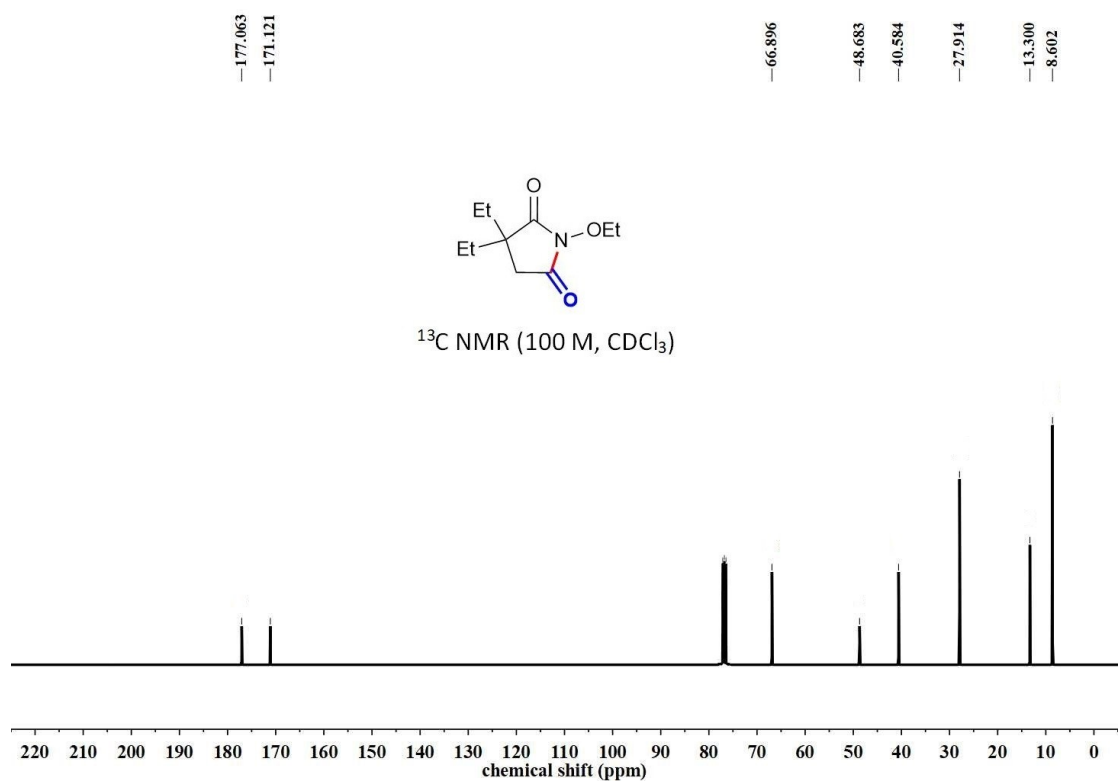
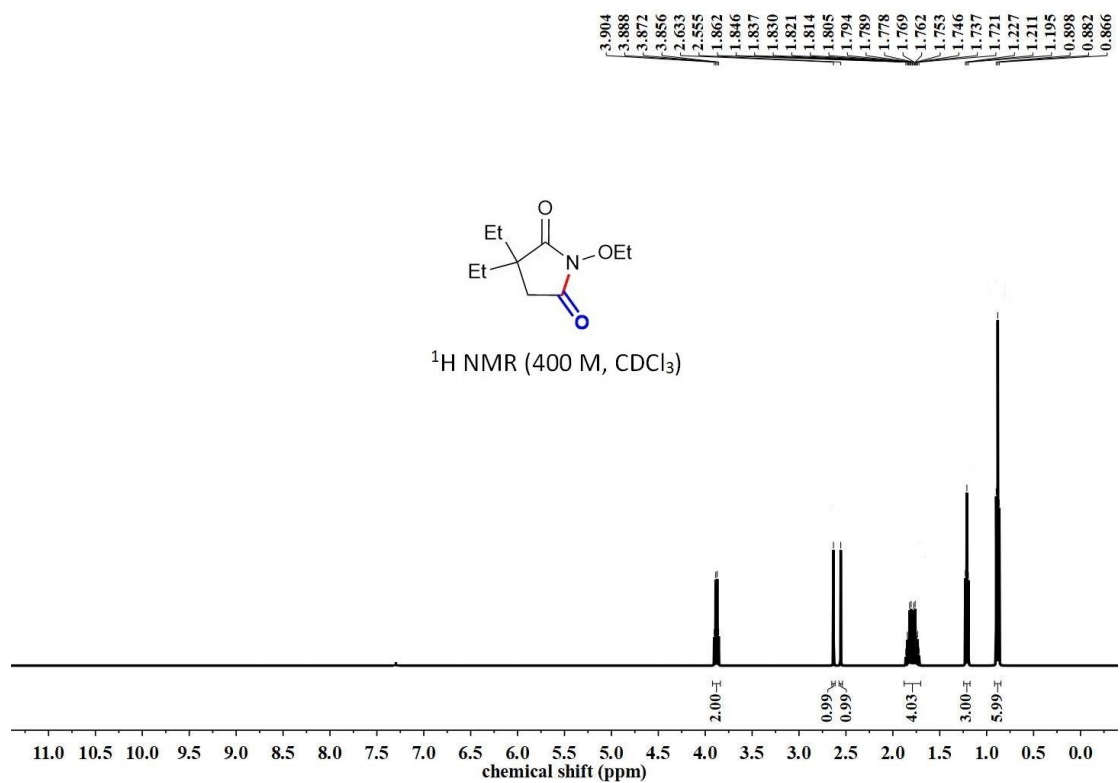


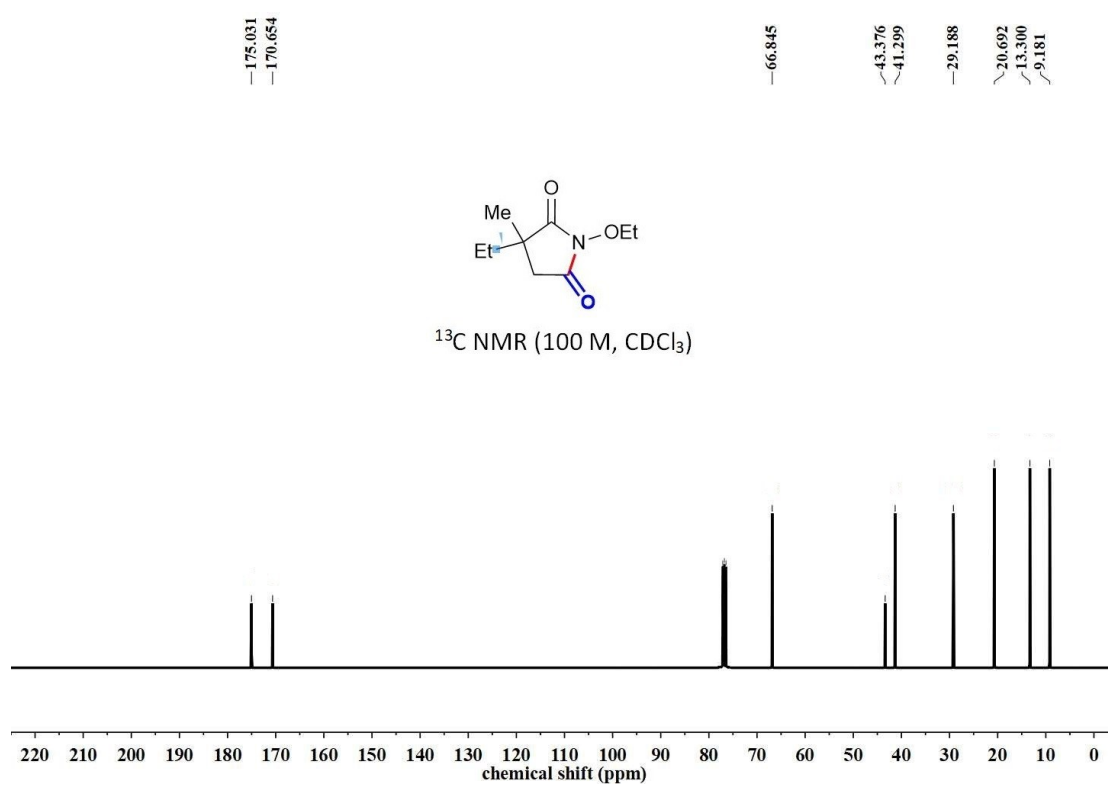
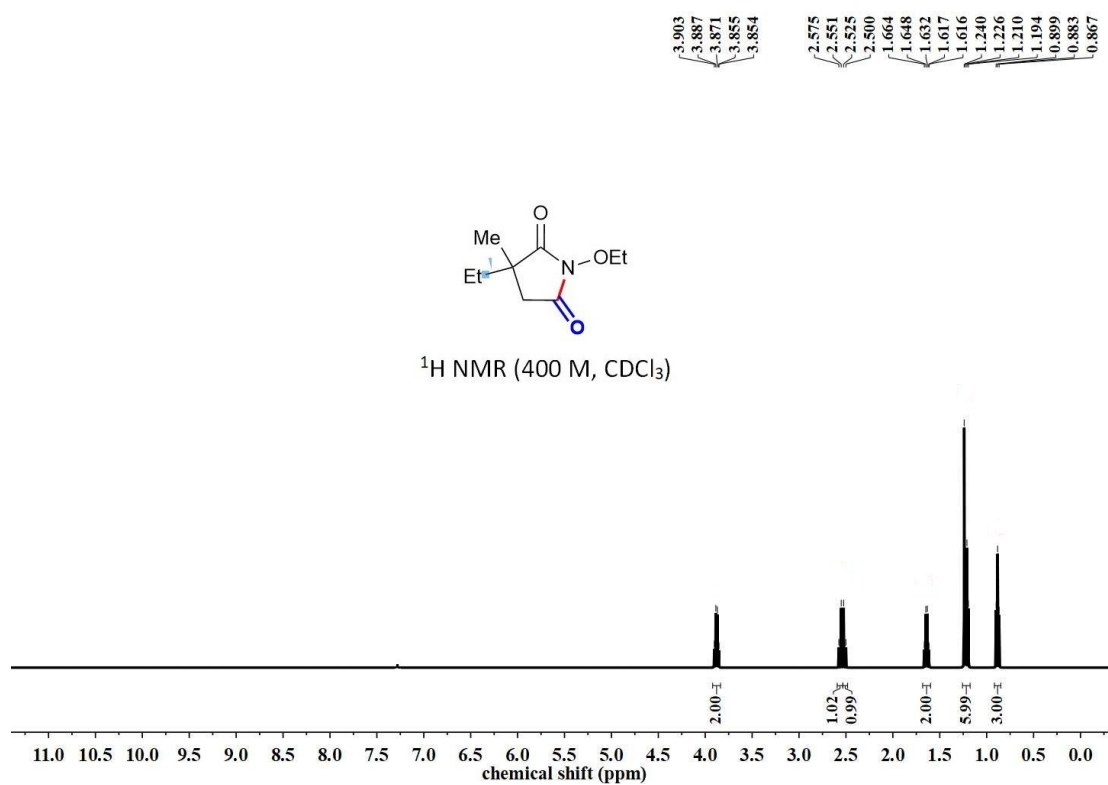


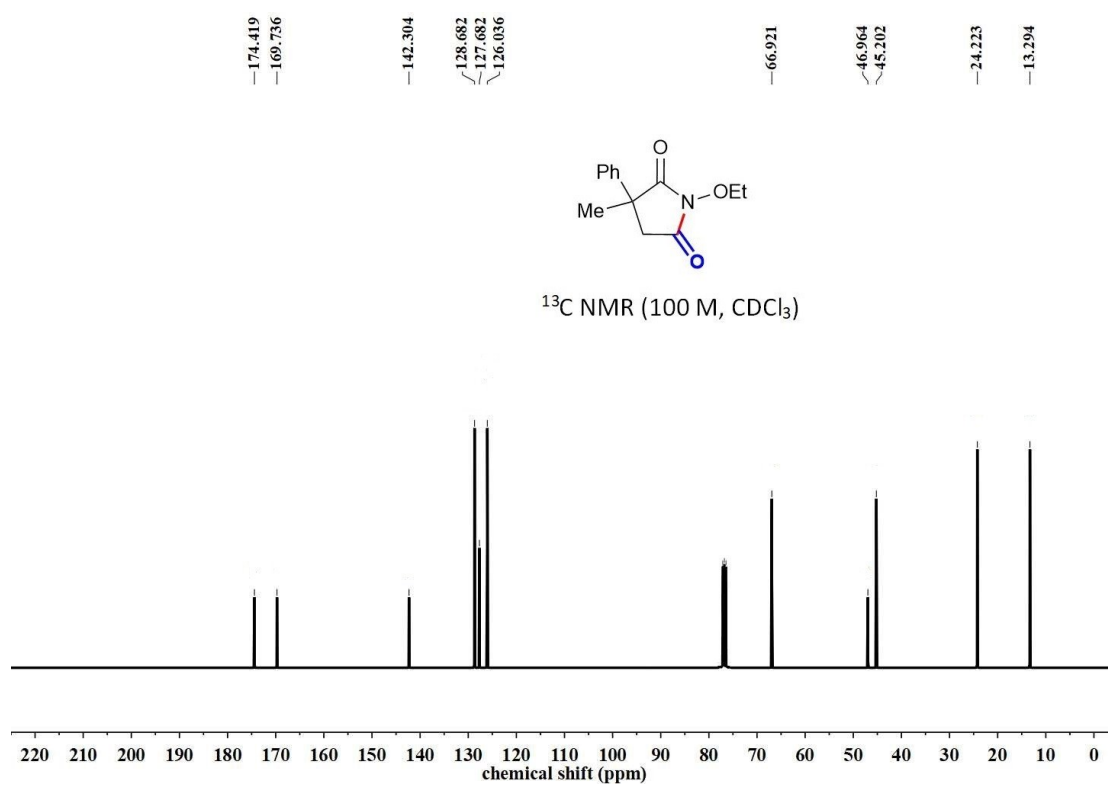
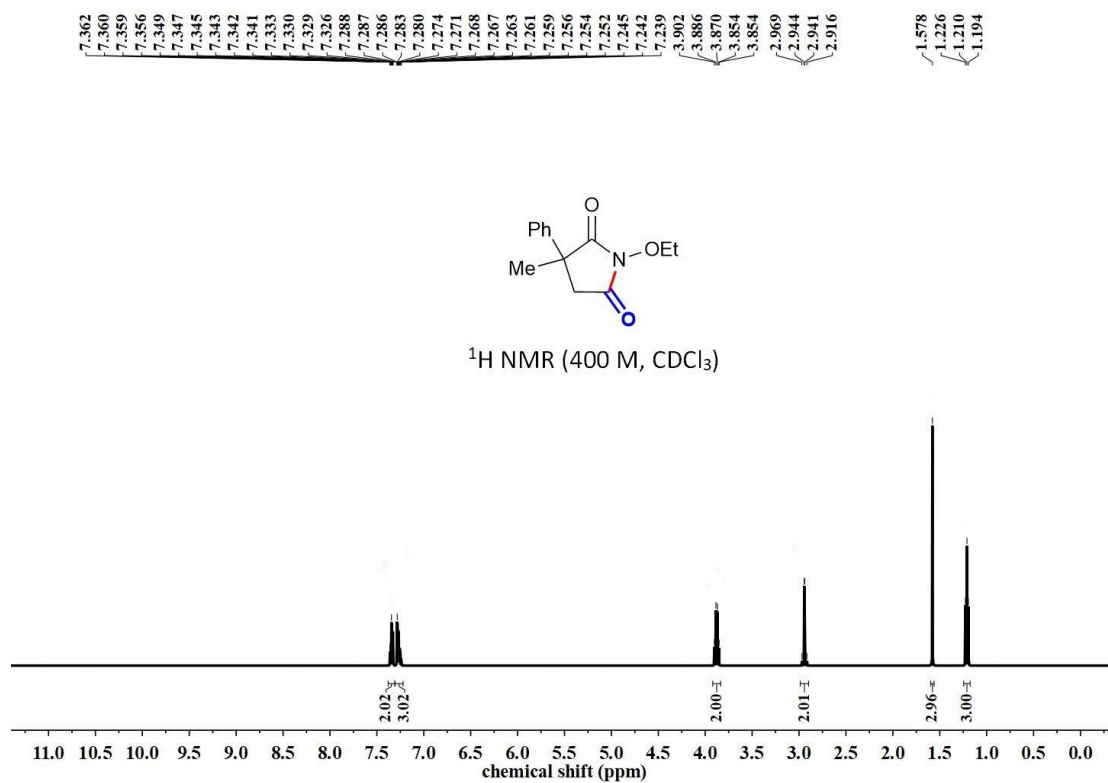


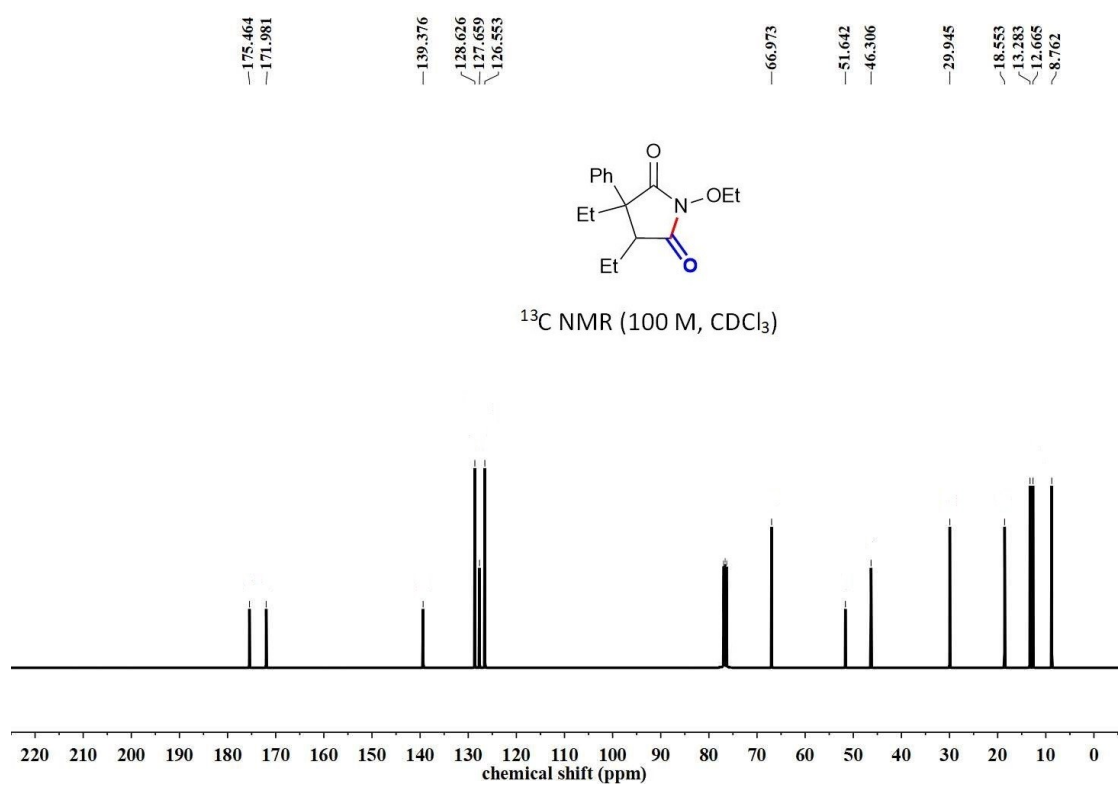
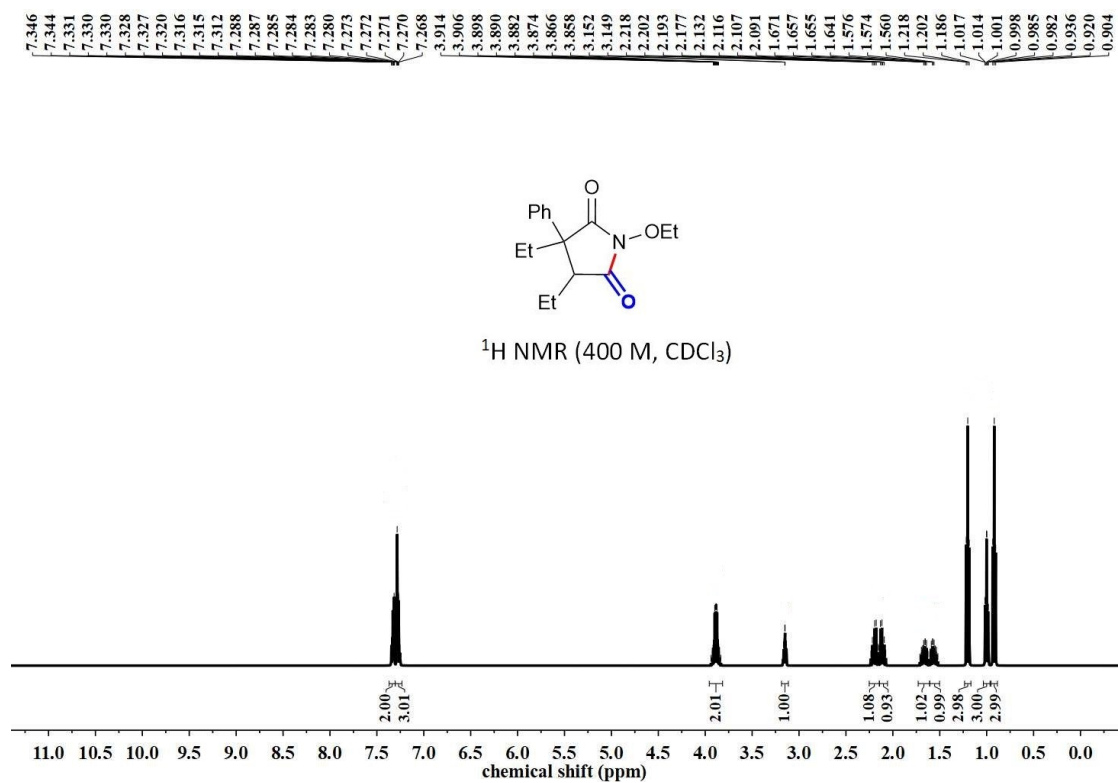


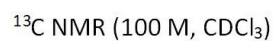


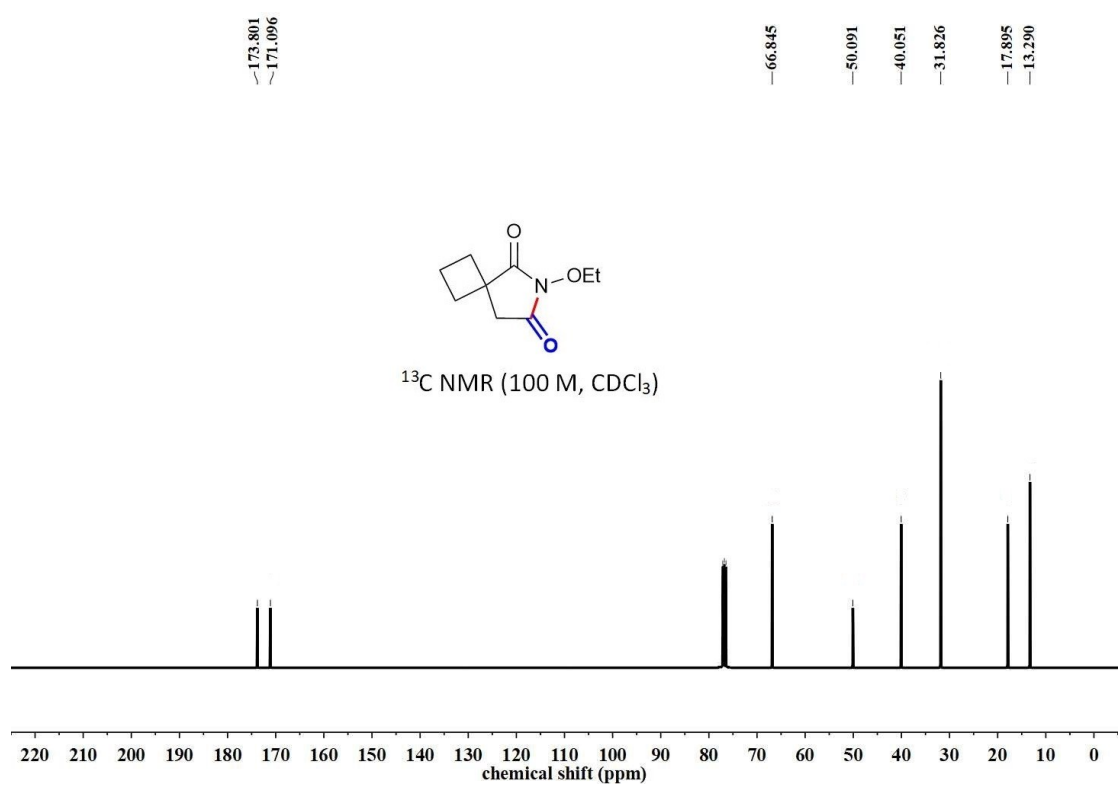
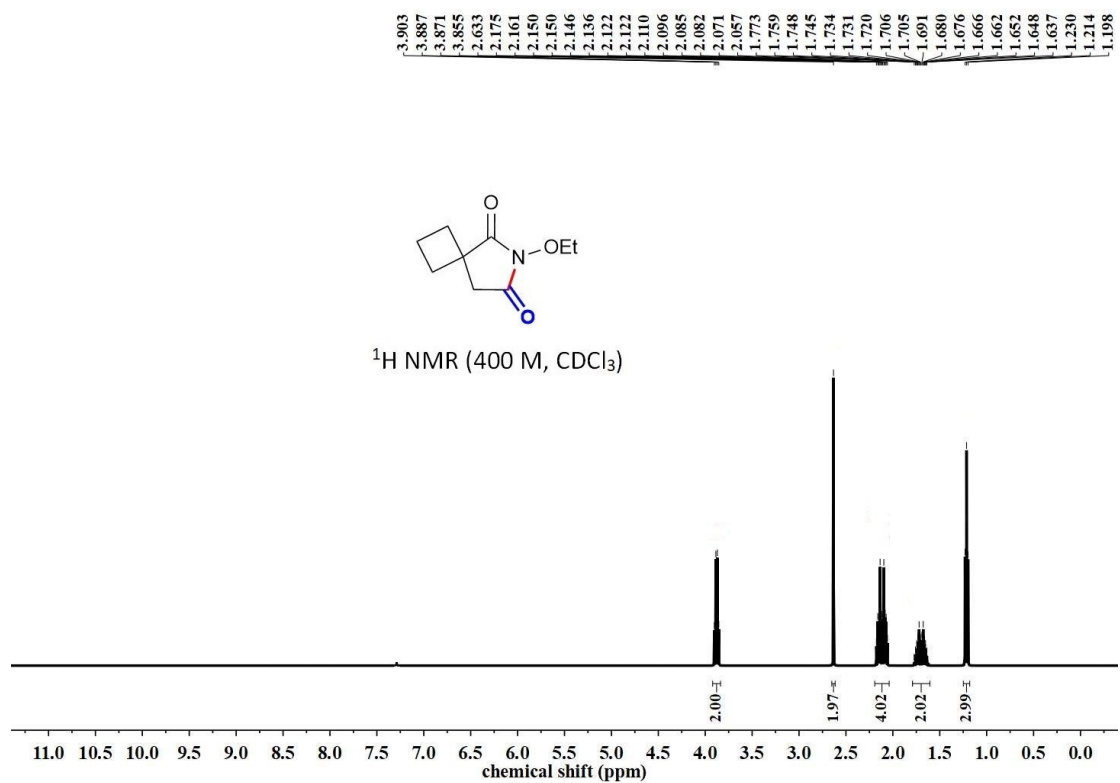


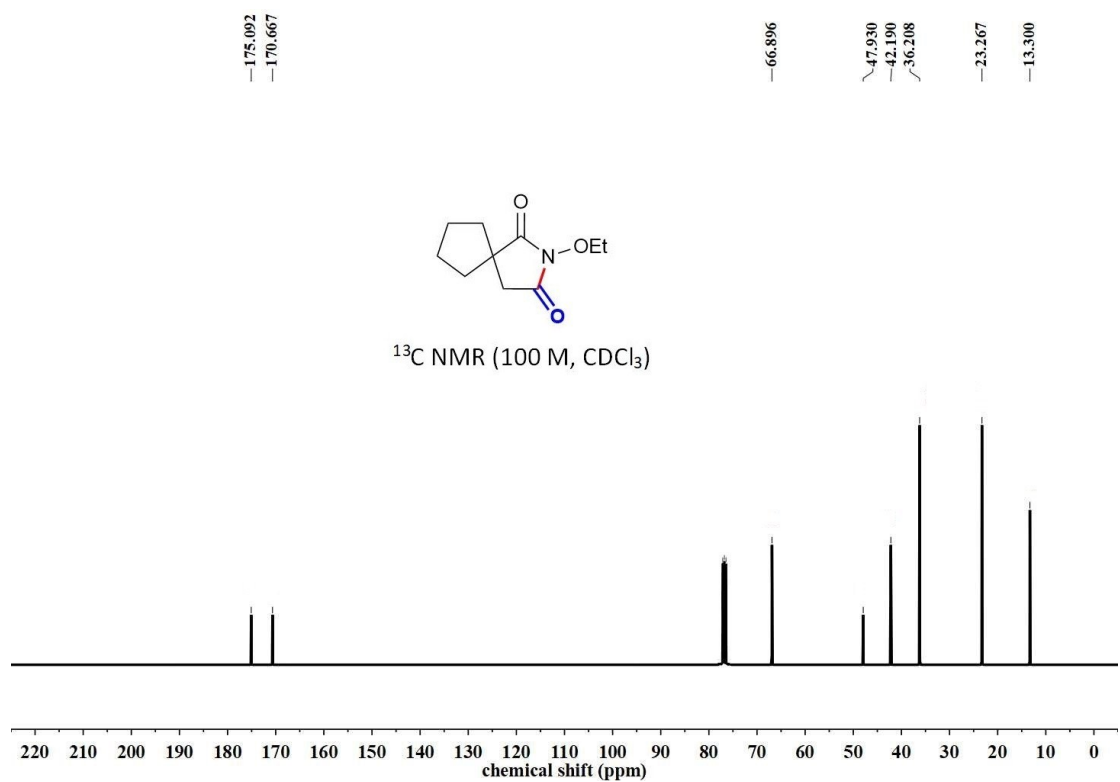
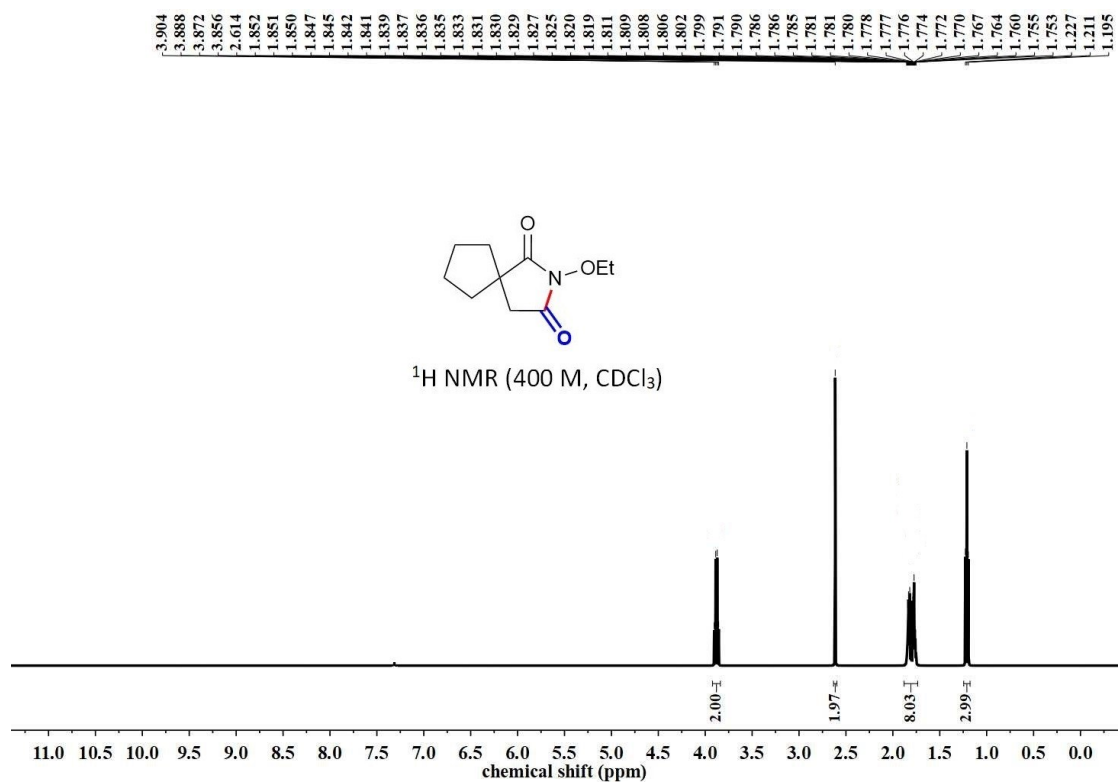


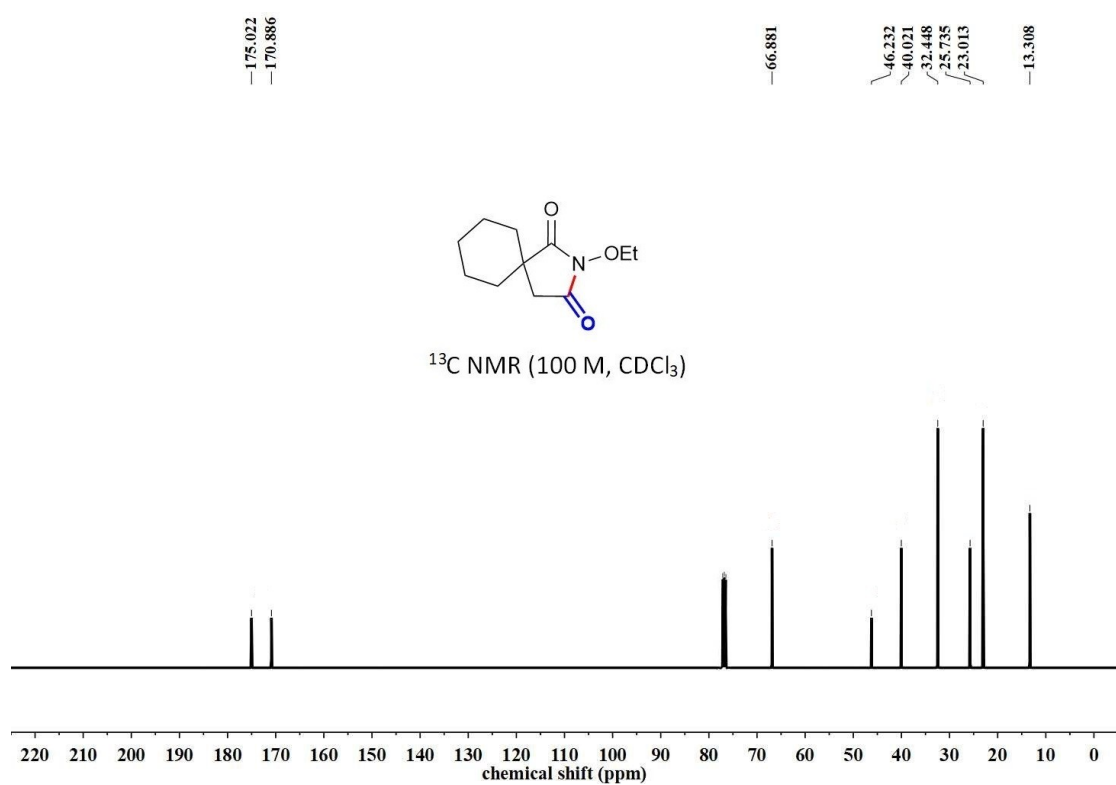
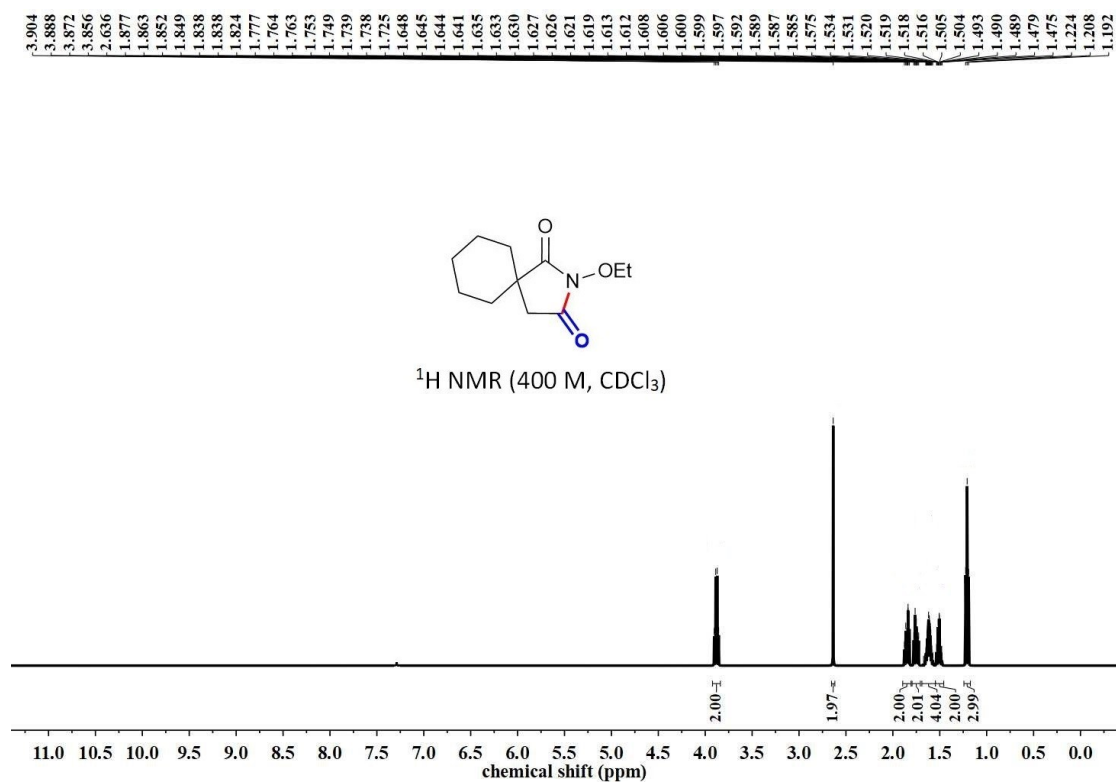


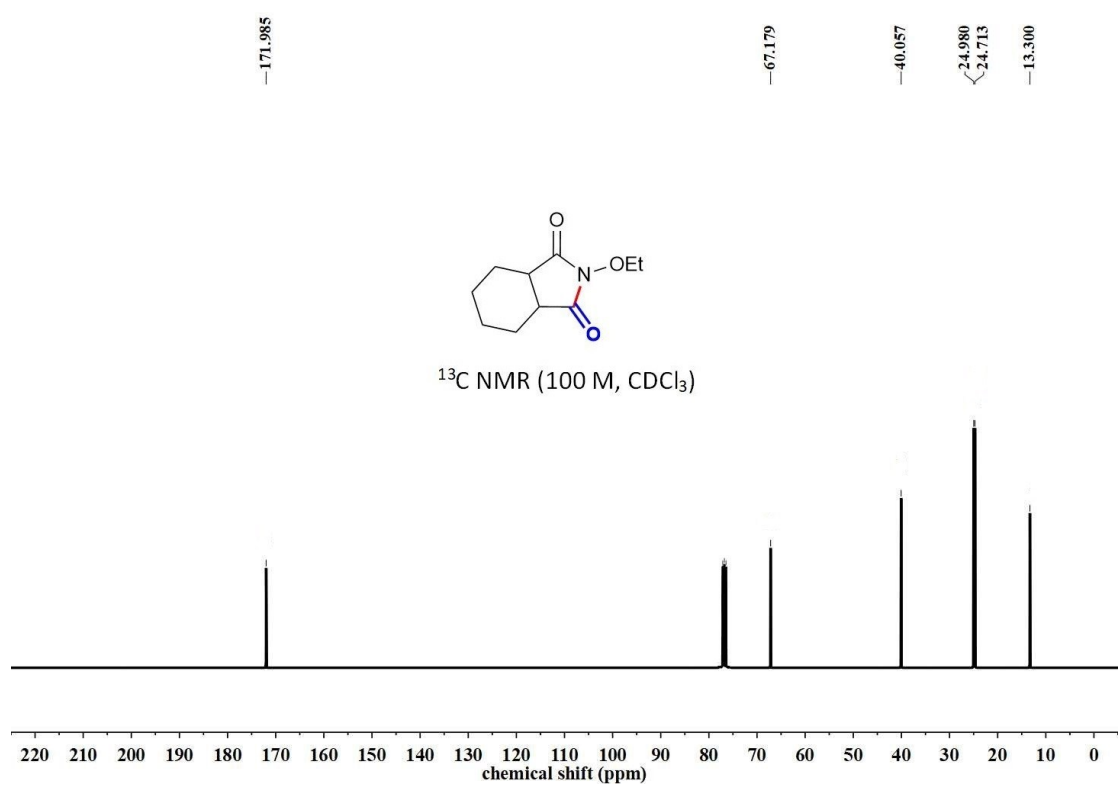
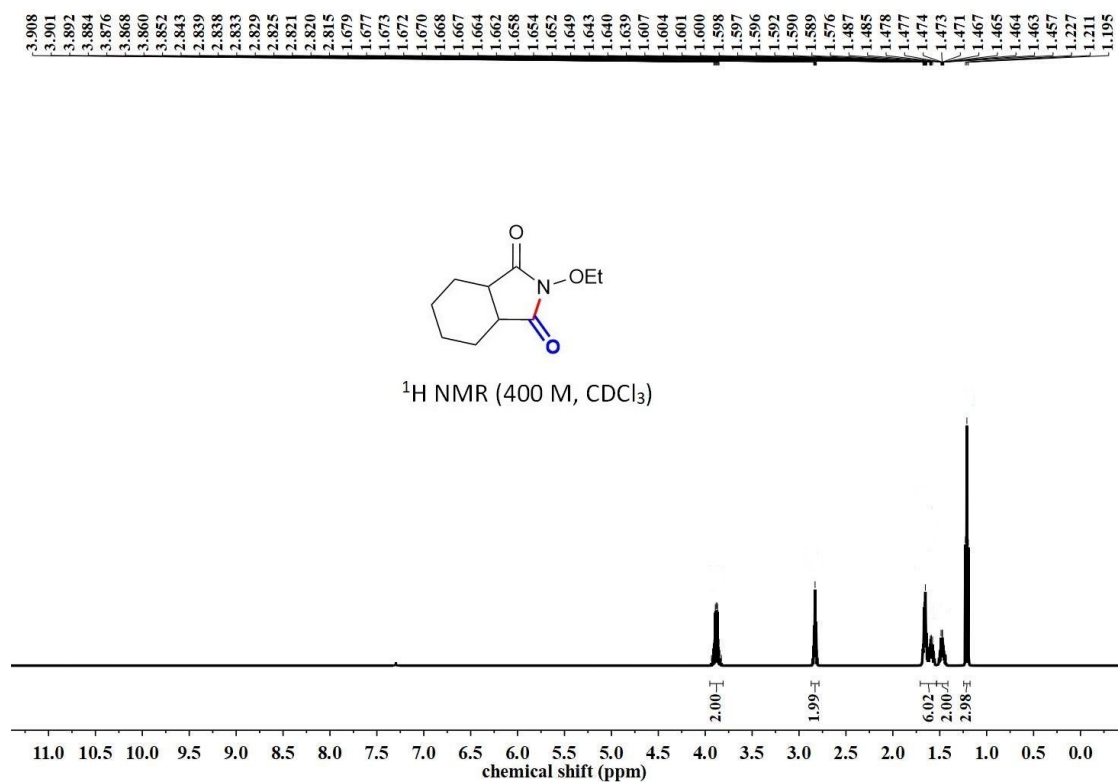








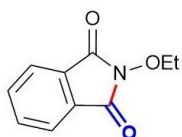




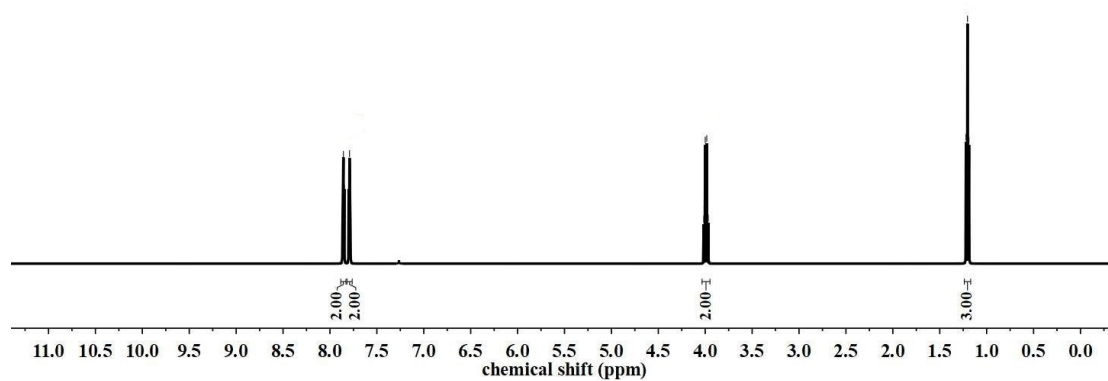
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7.791
7.790
7.783
7.781

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1.204
1.188



^1H NMR (400 M, CDCl_3)



164.752

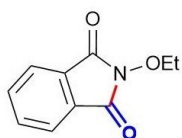
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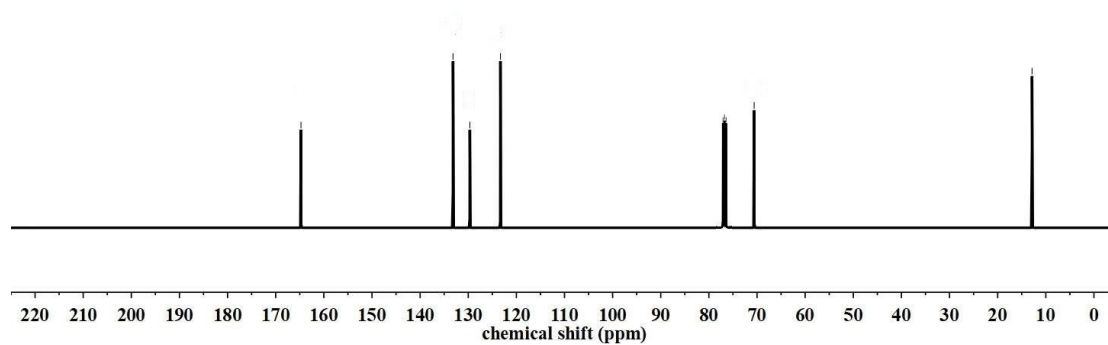
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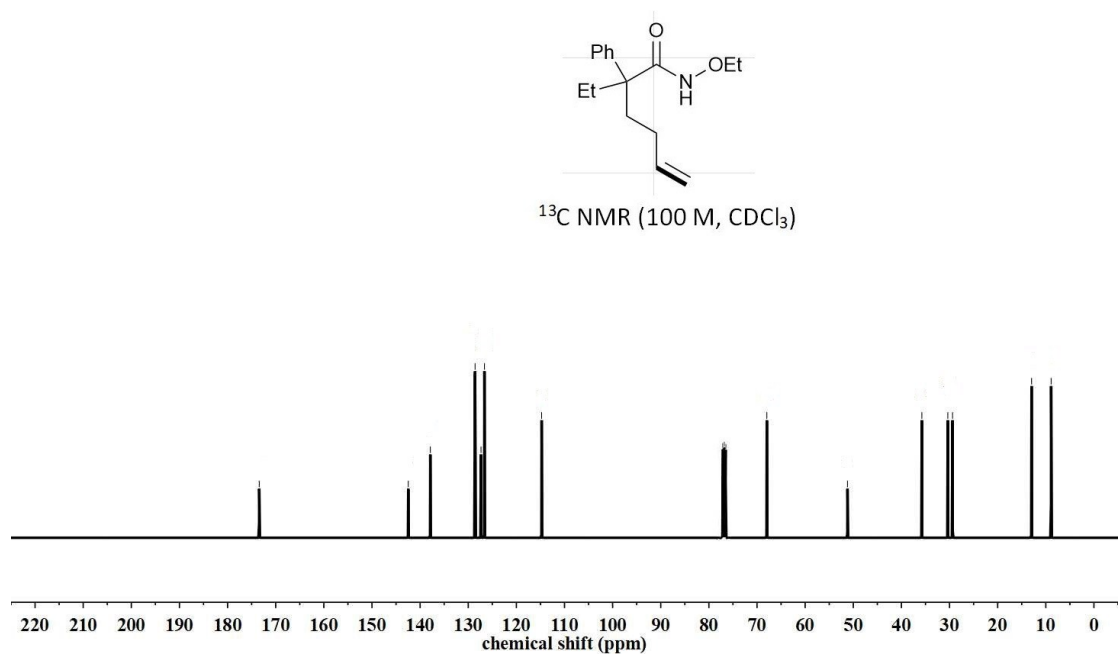
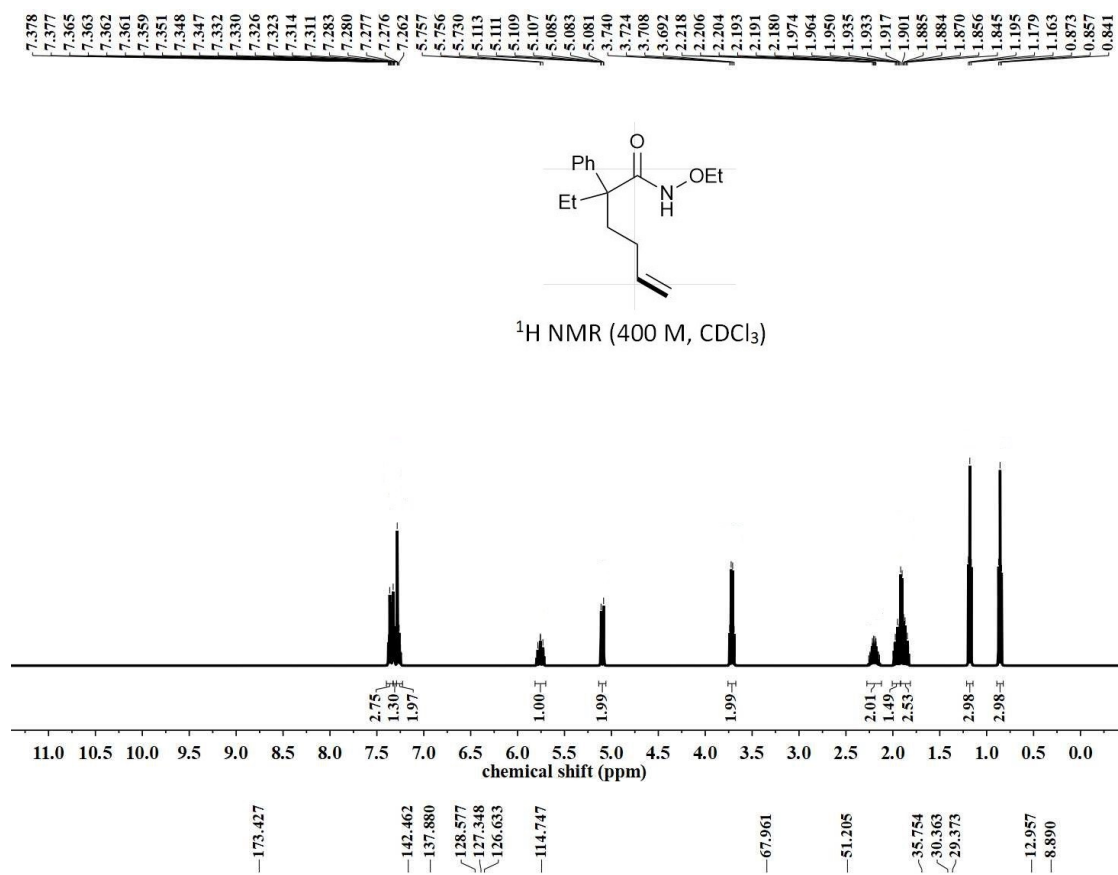
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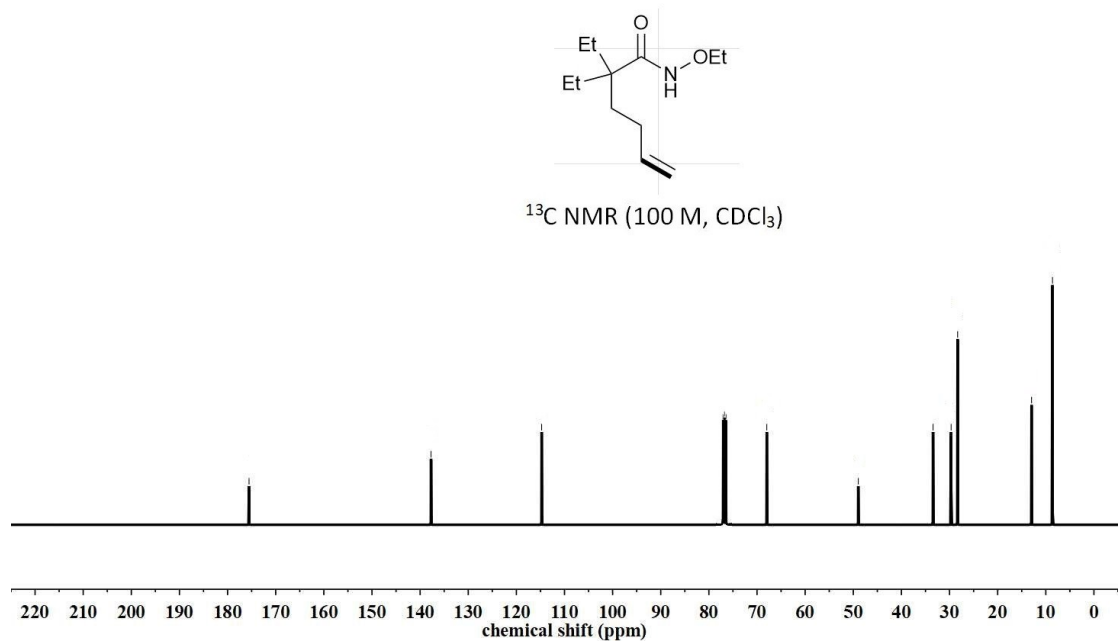
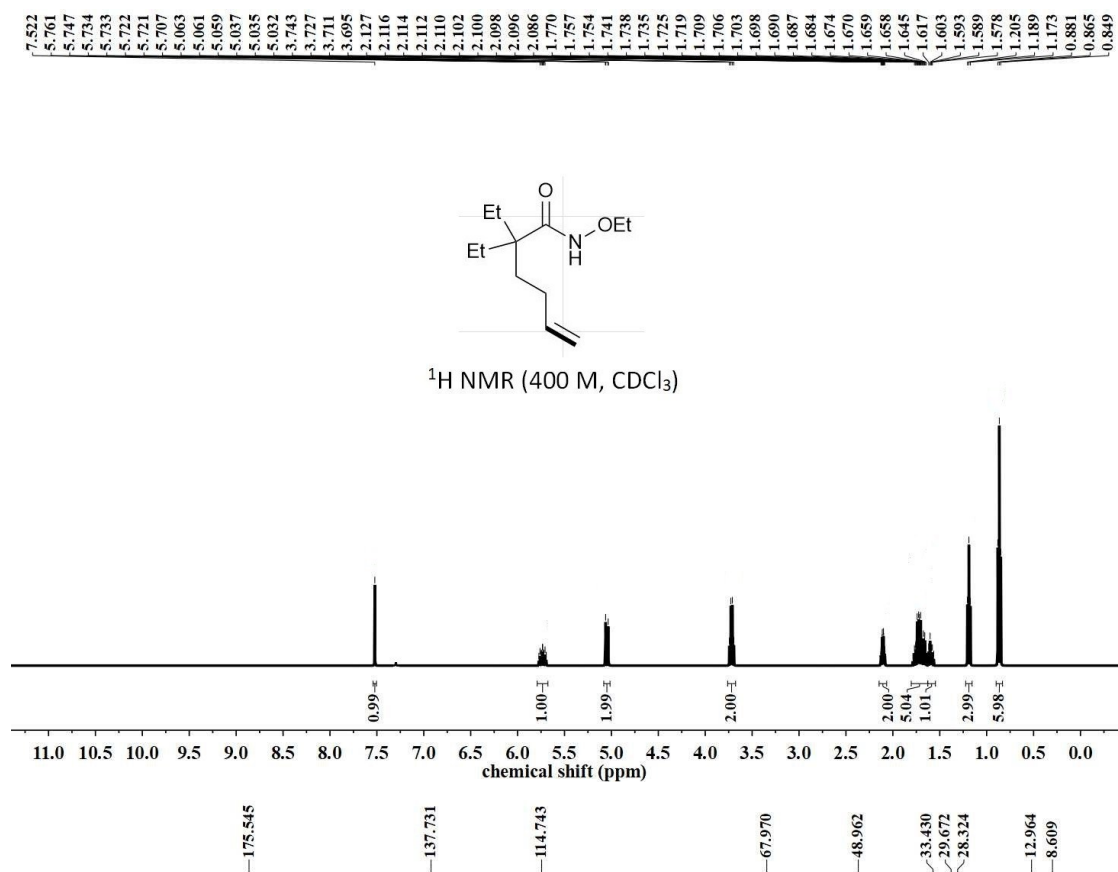
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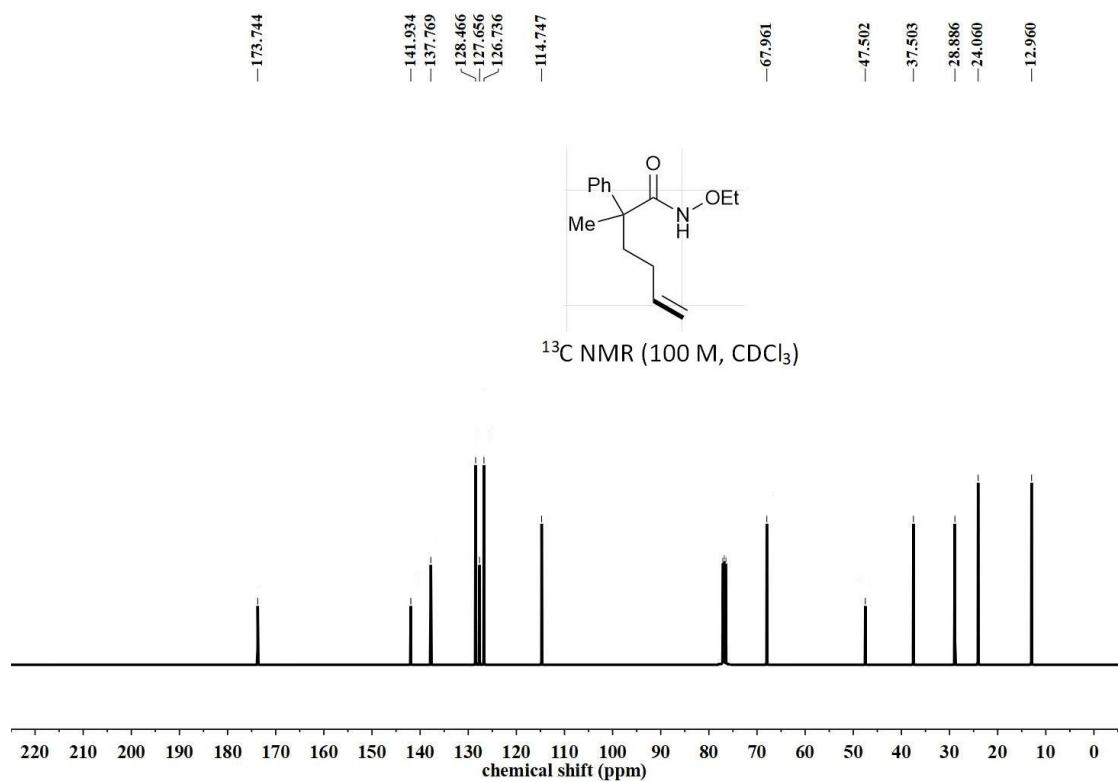
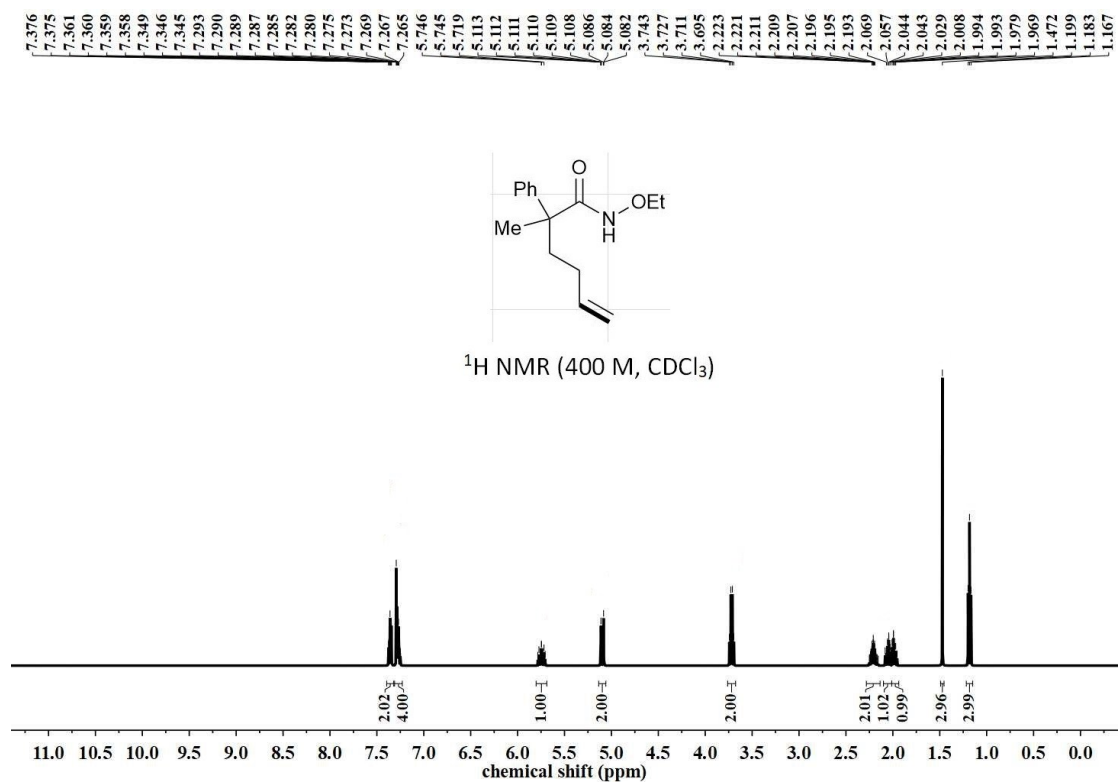


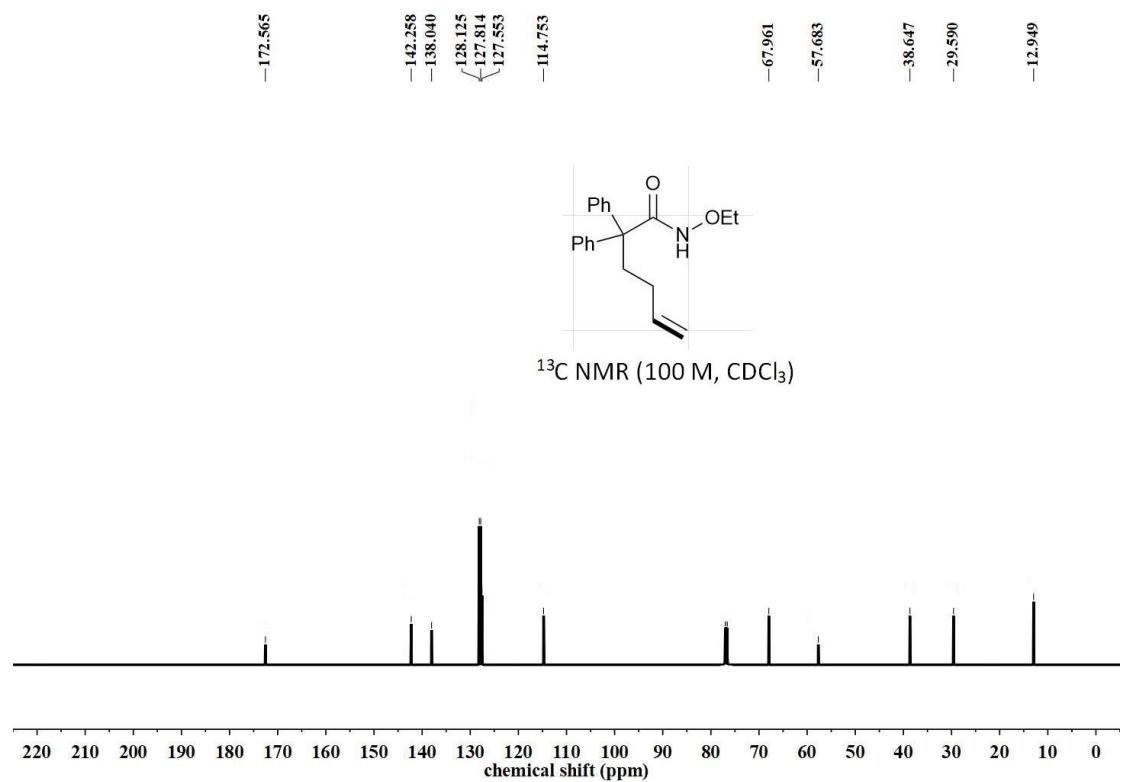
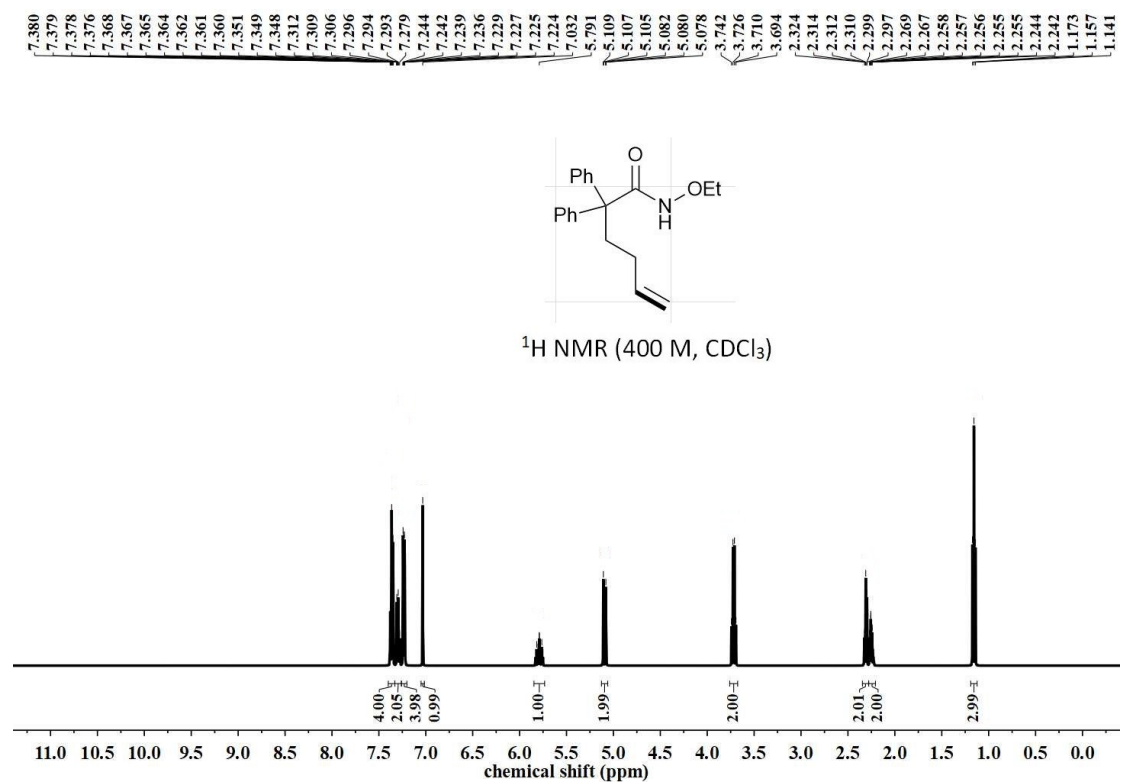
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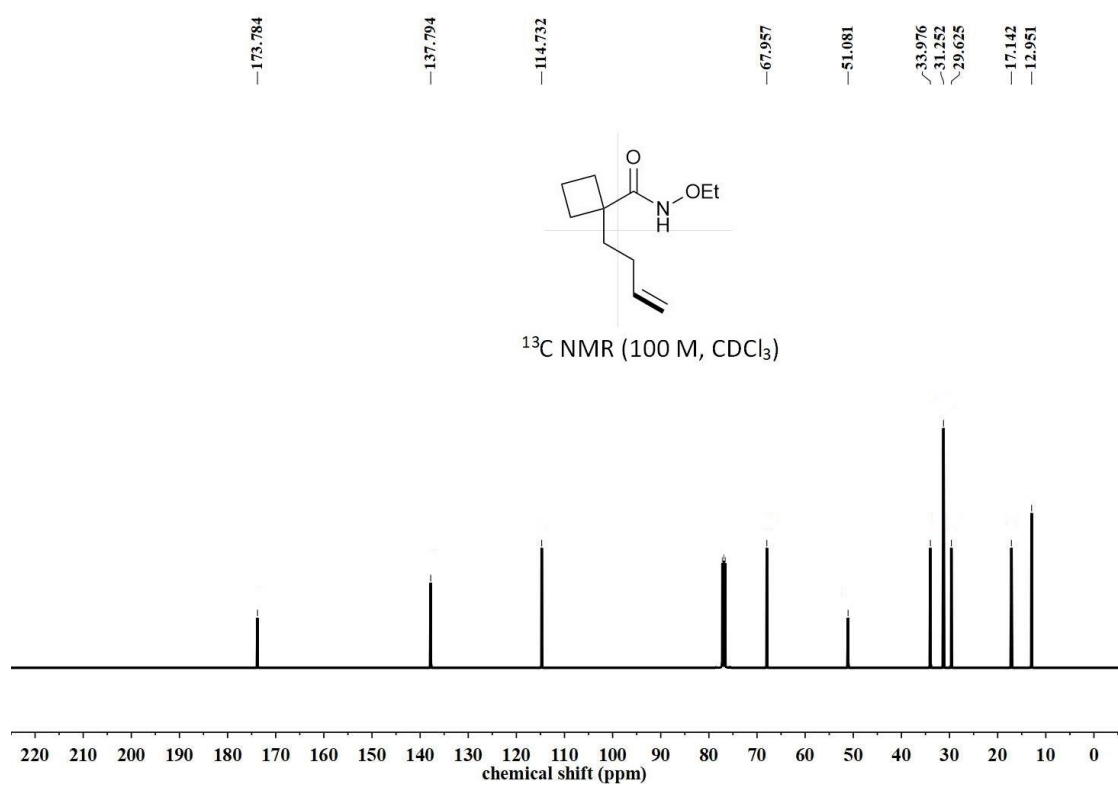
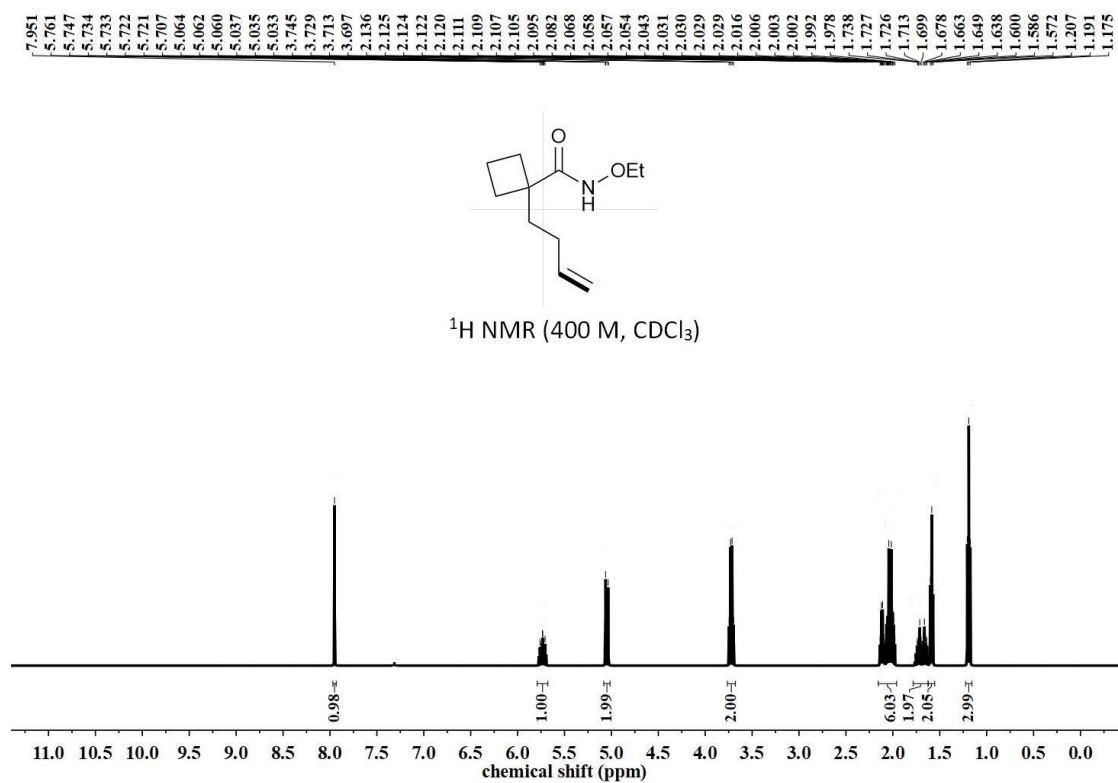


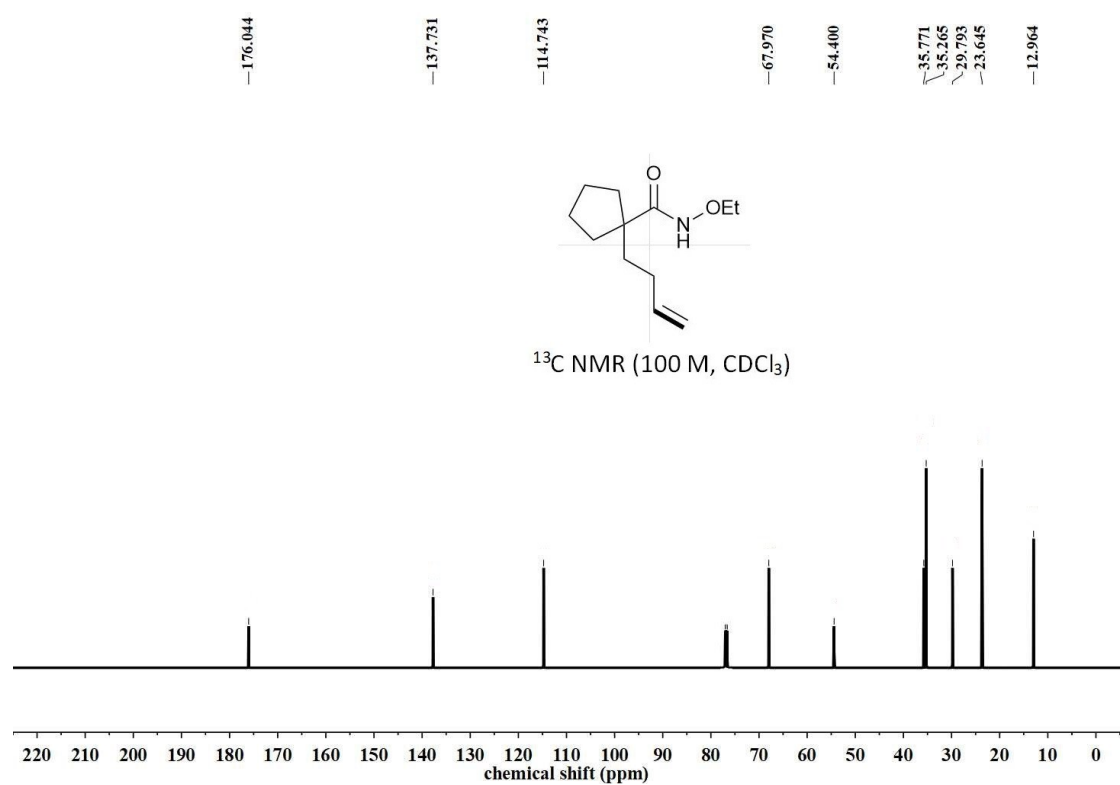
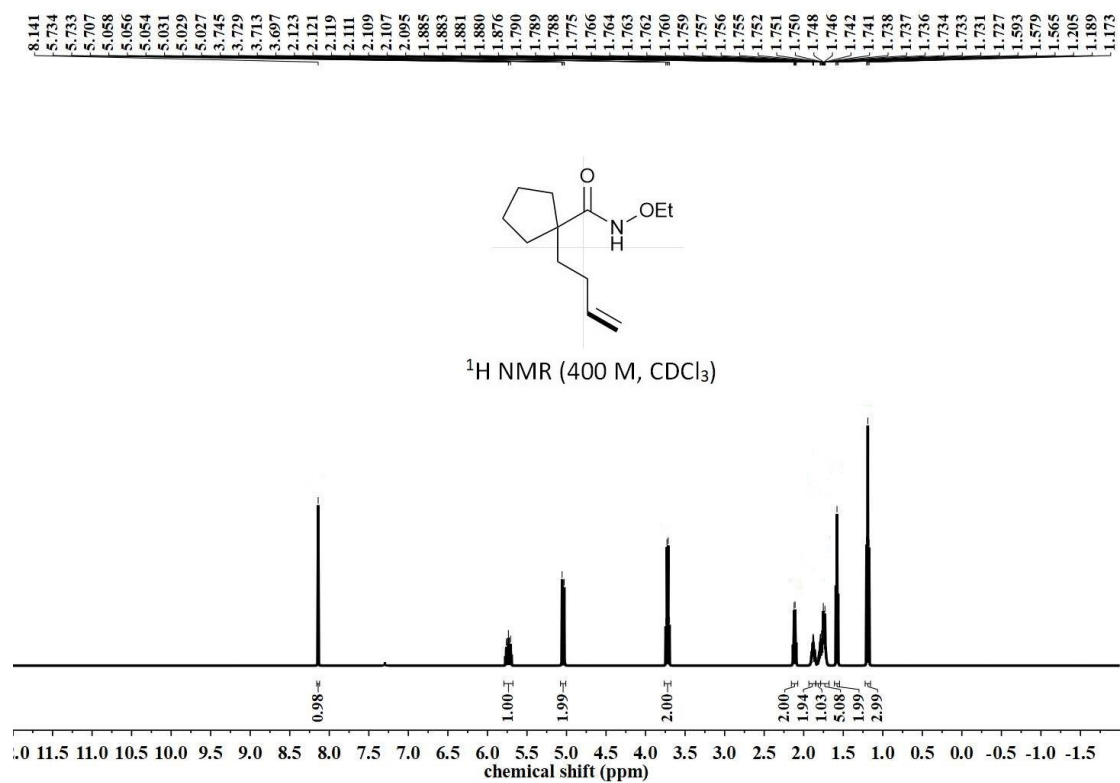


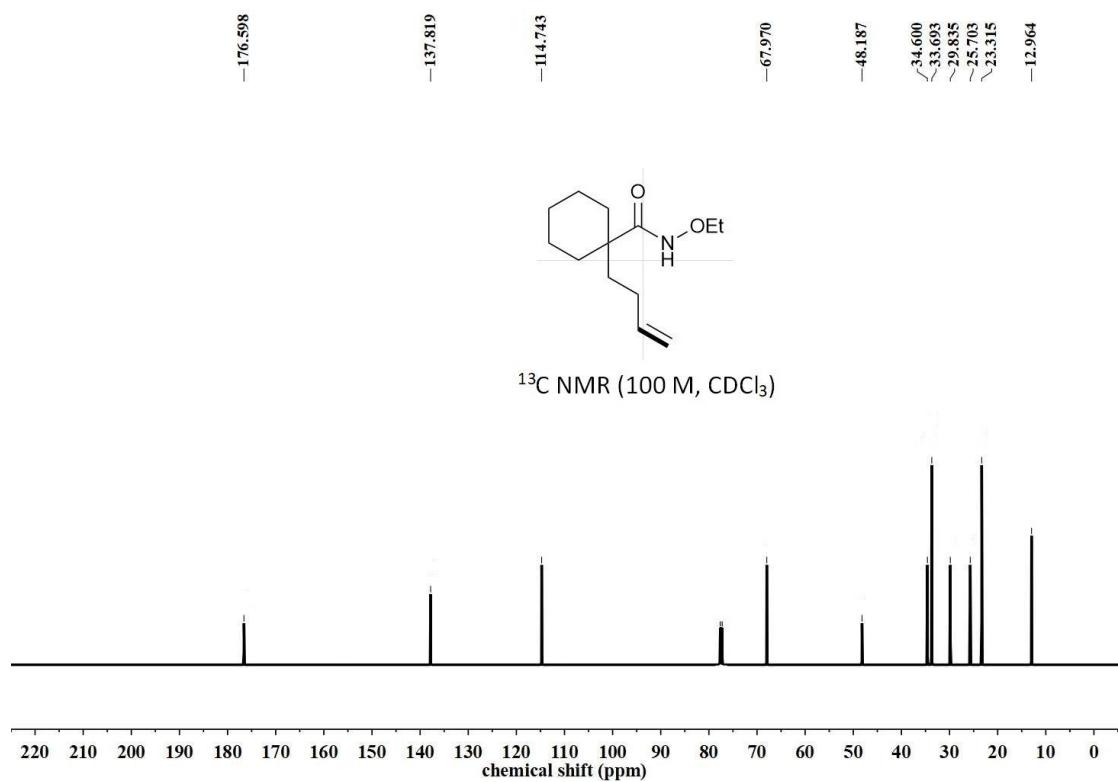
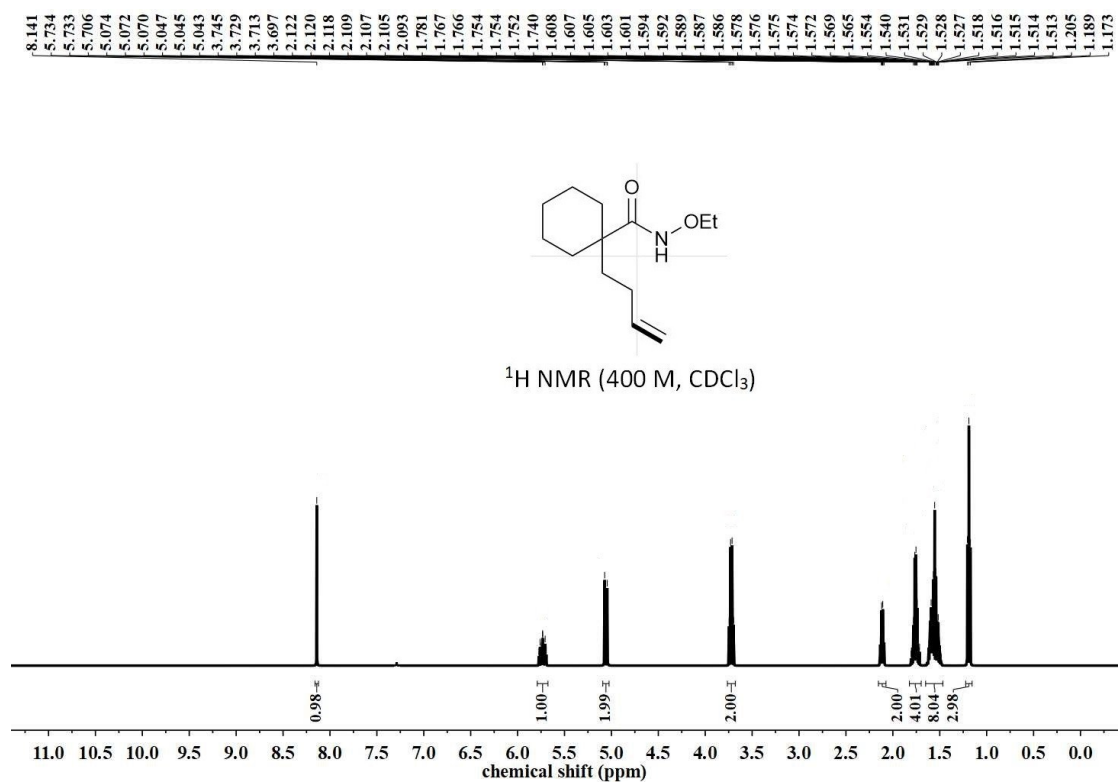


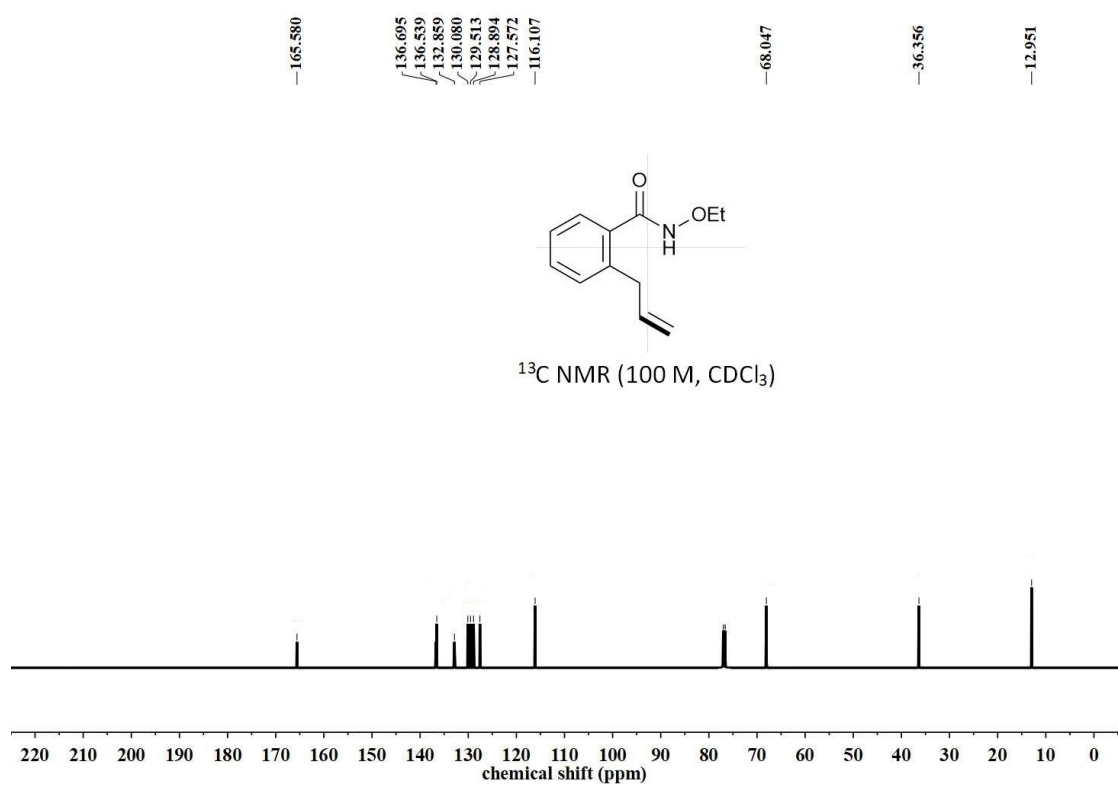
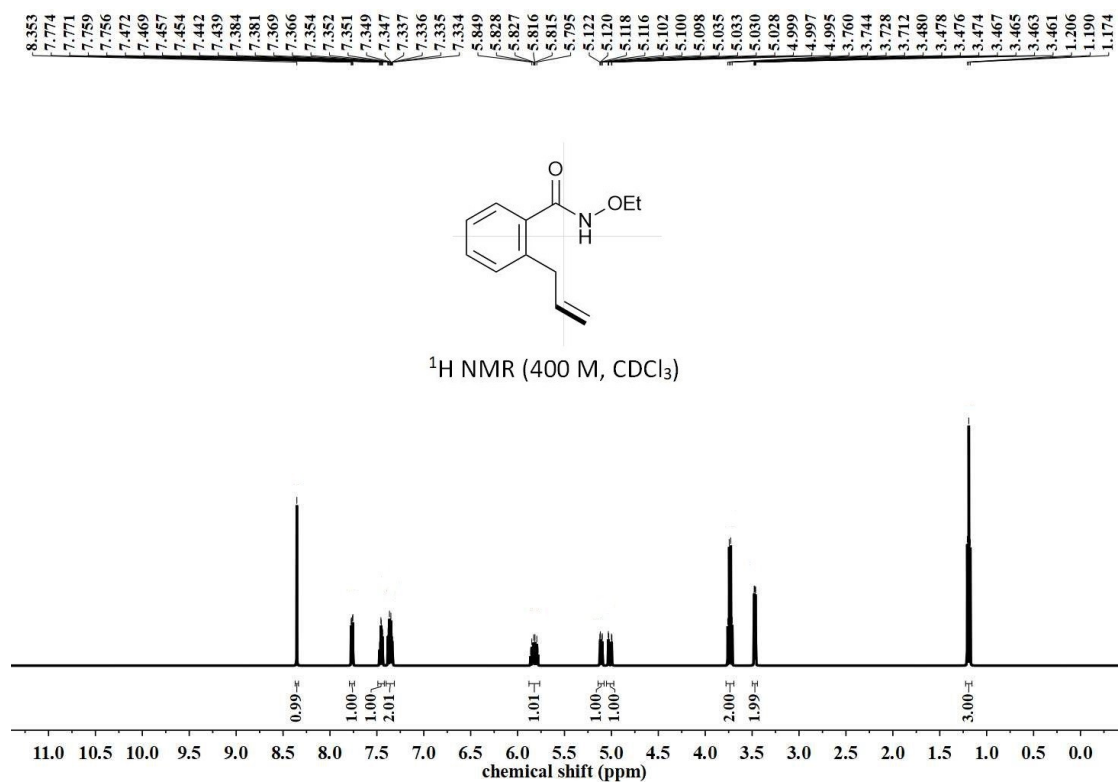


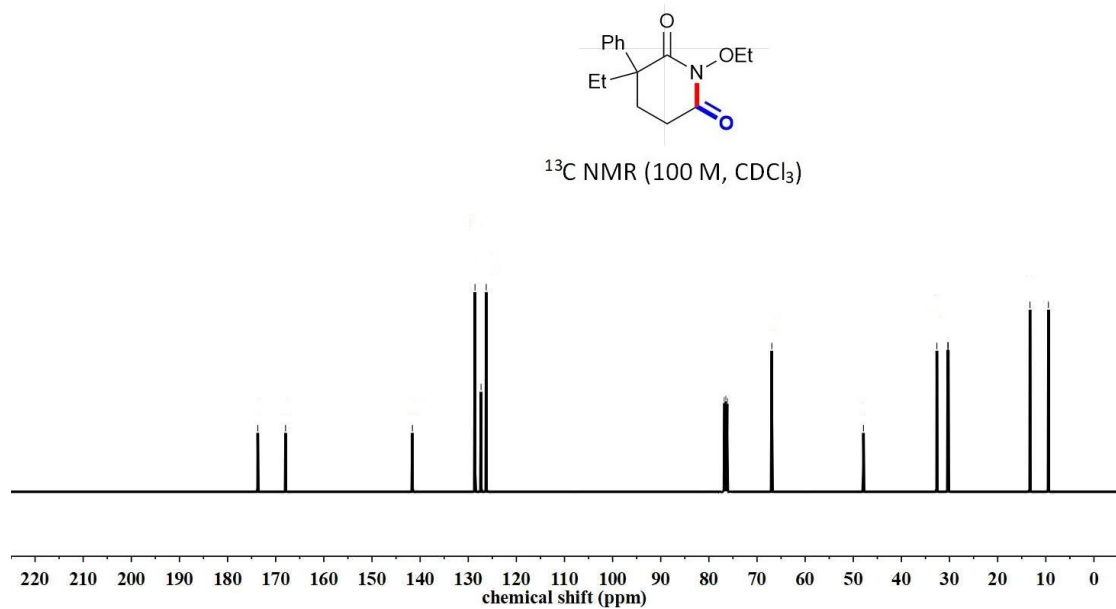
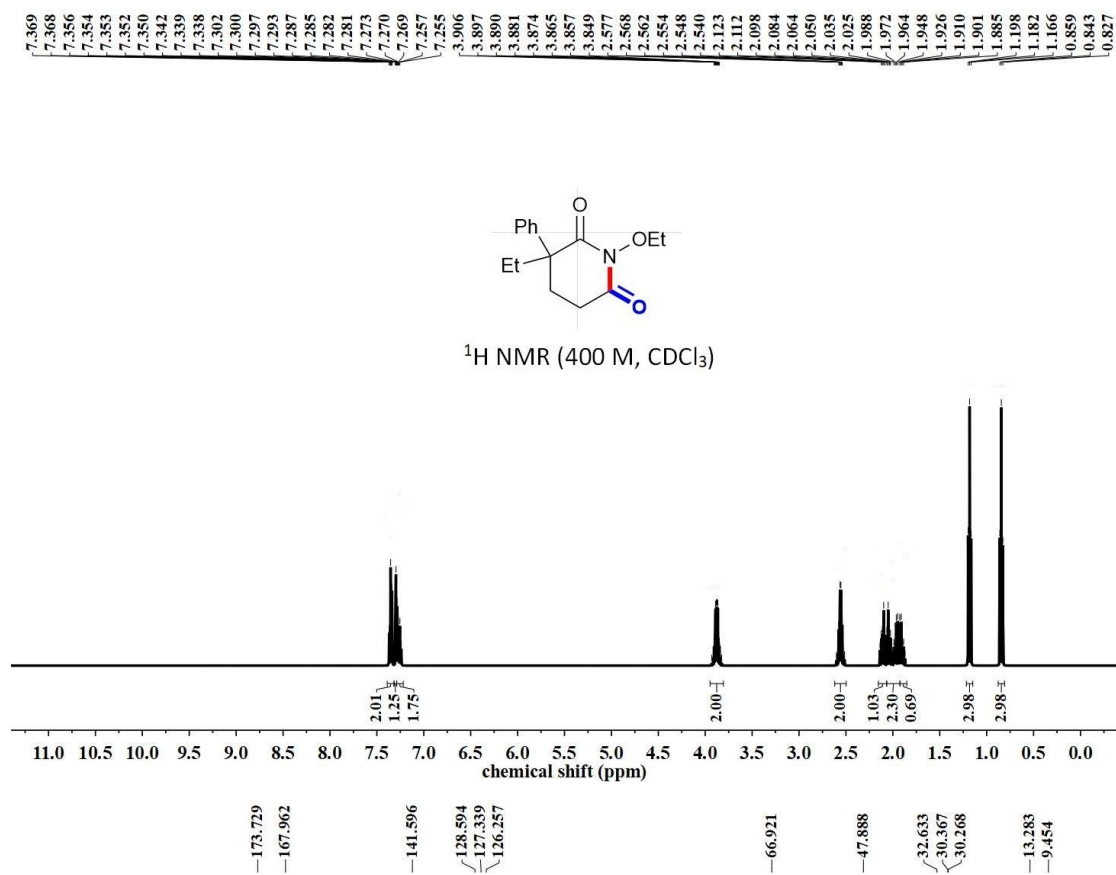


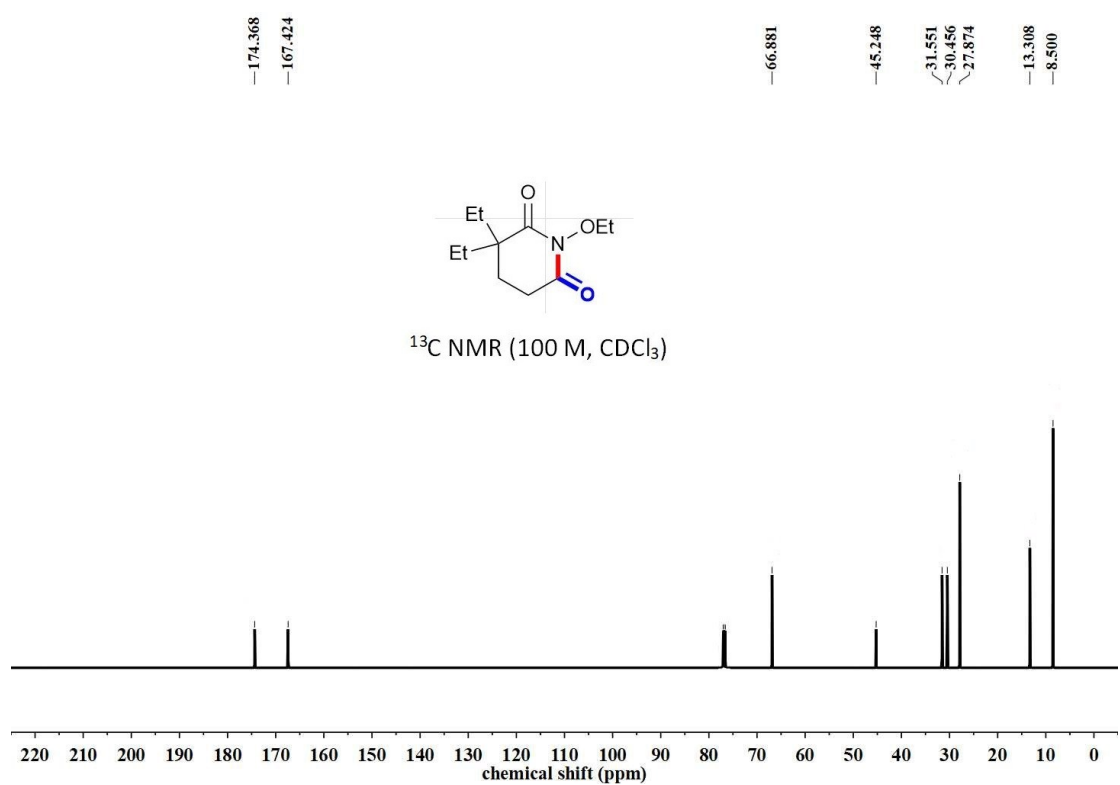
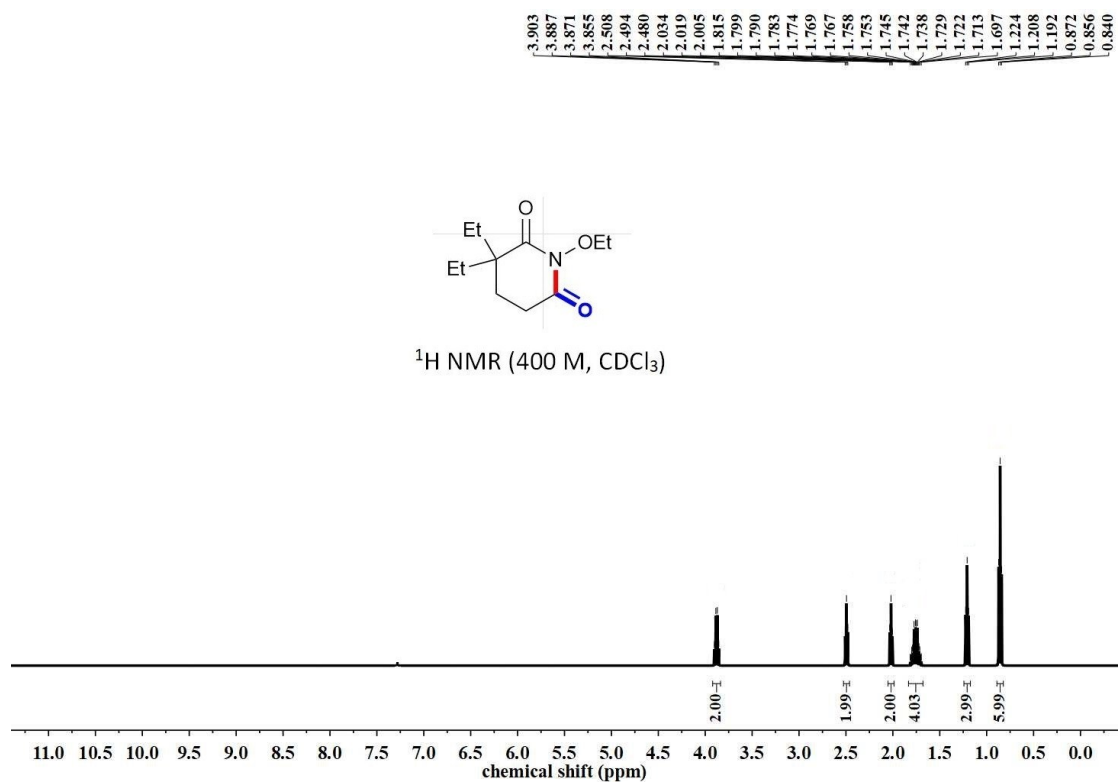


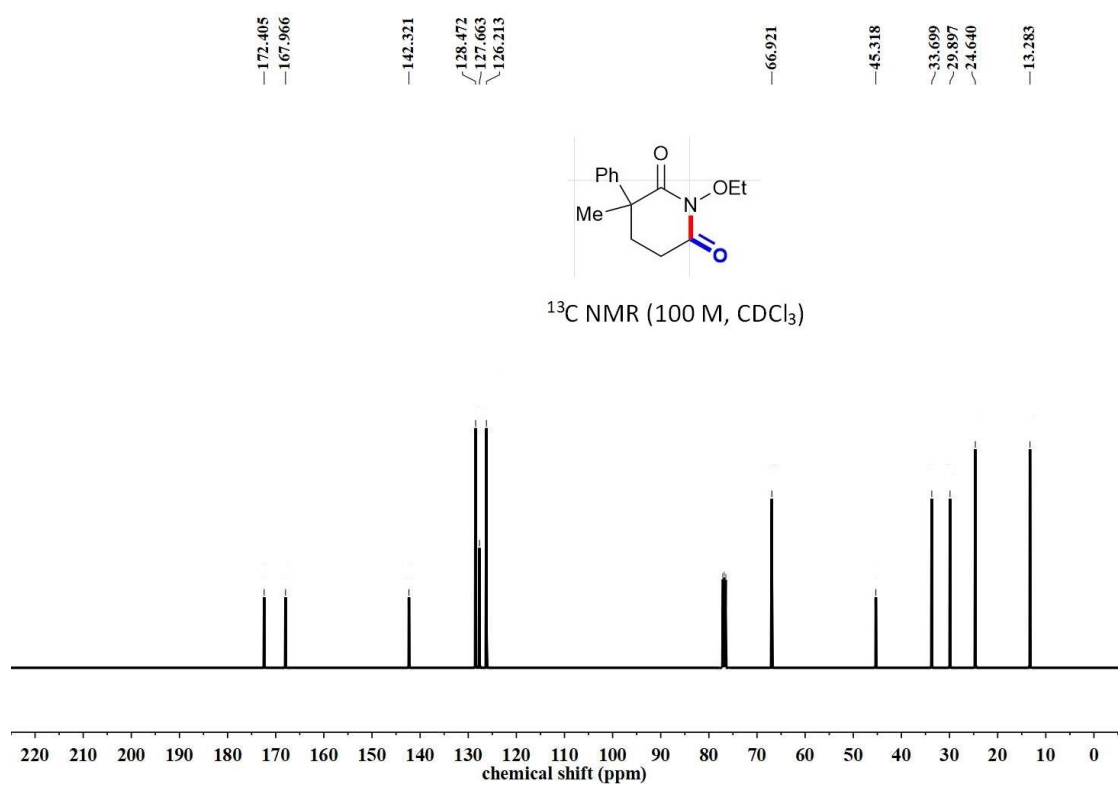
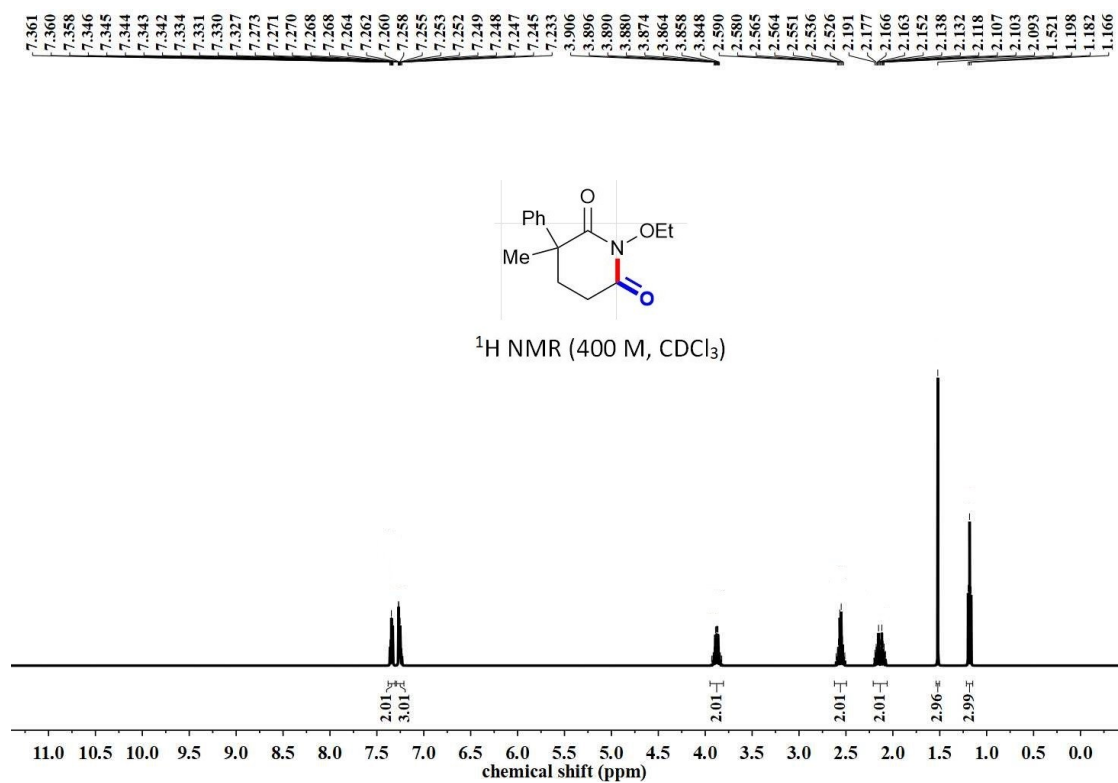


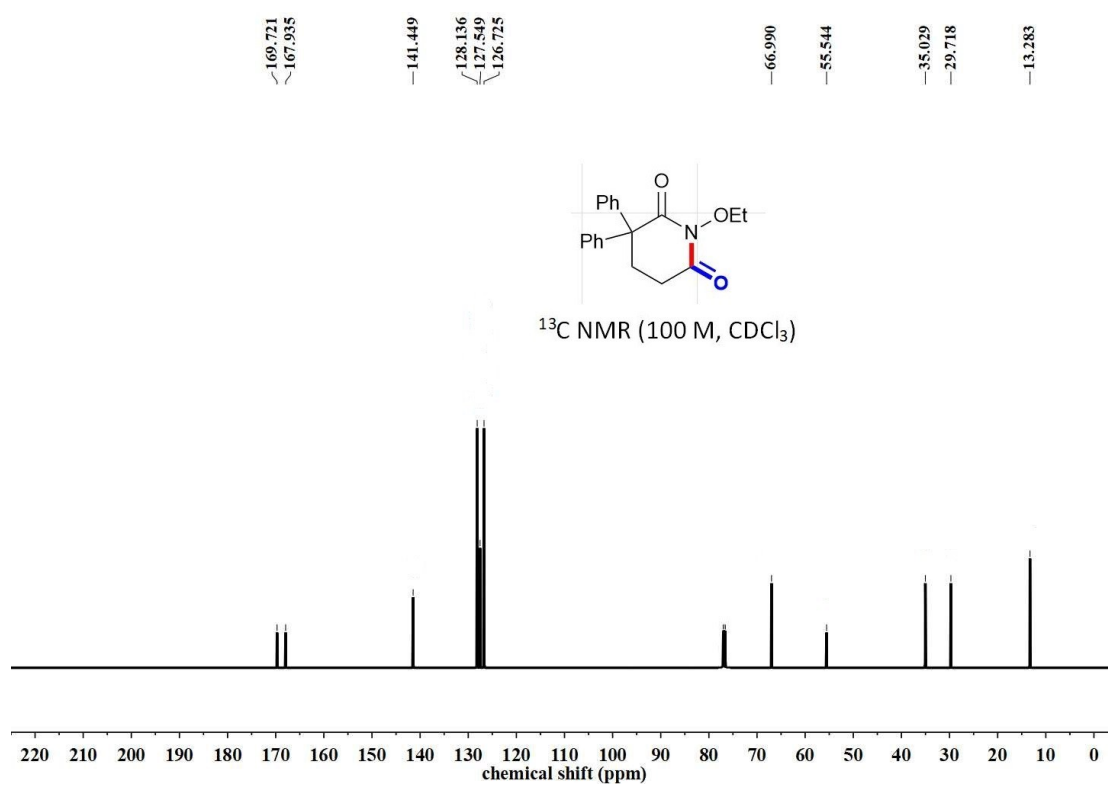
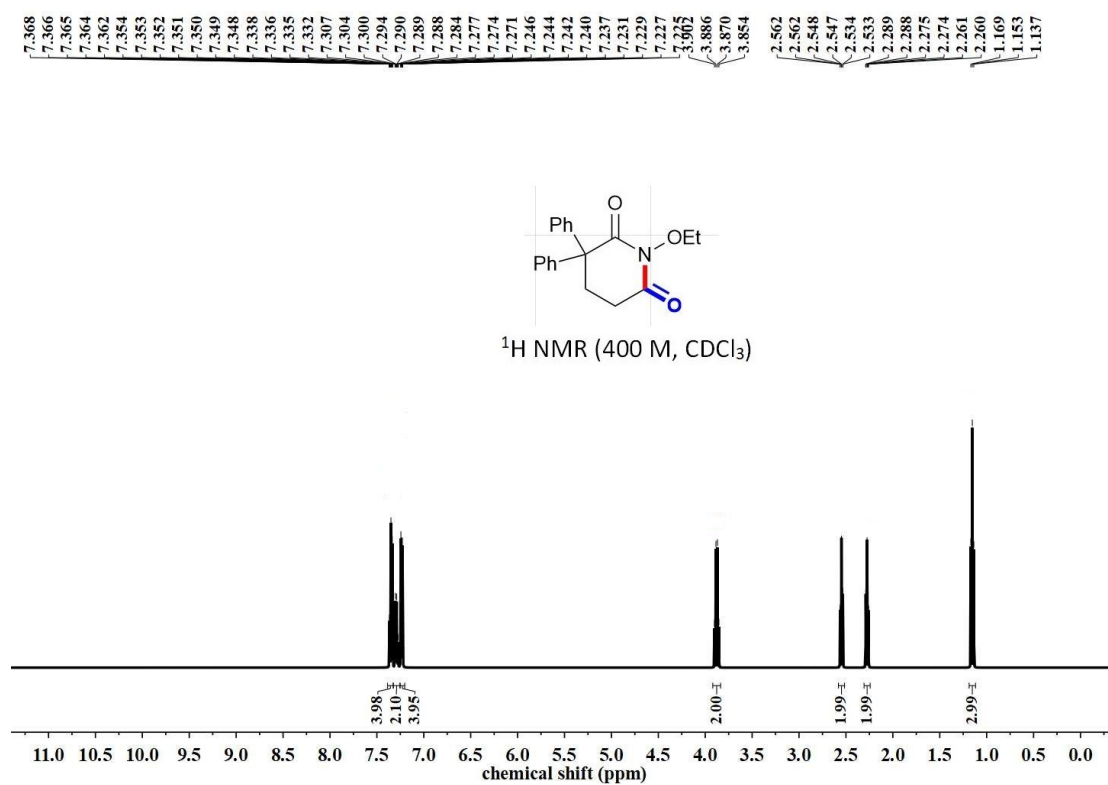


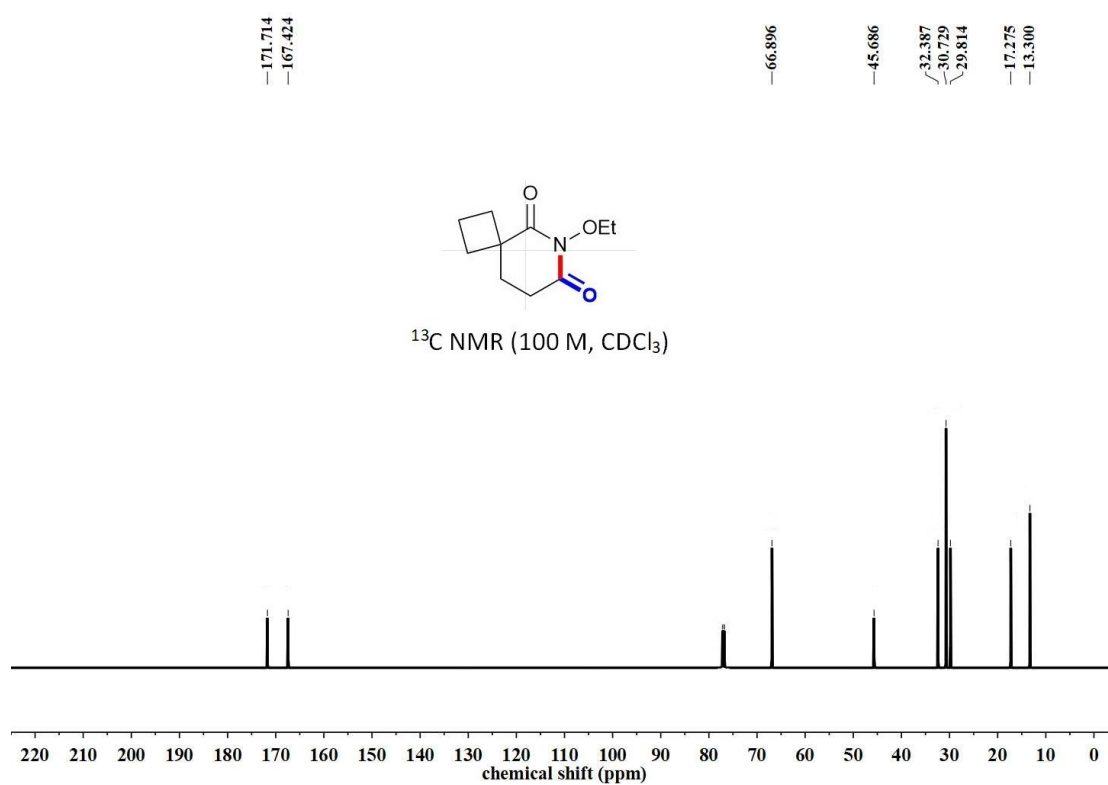
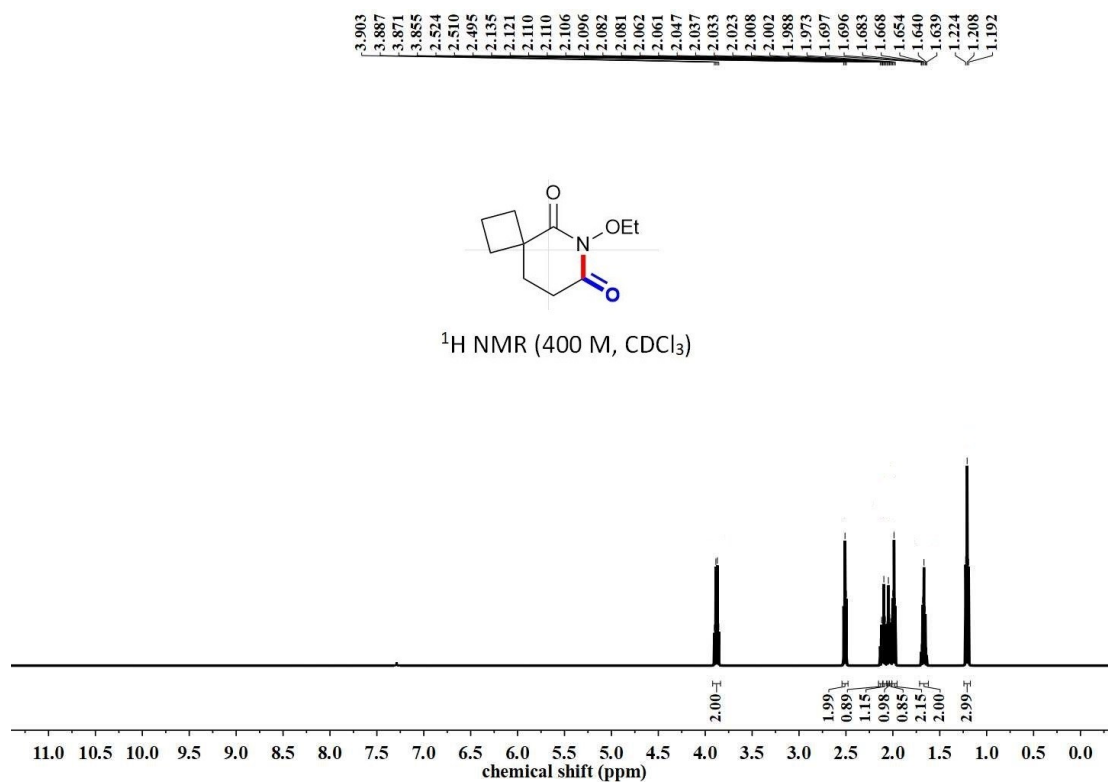


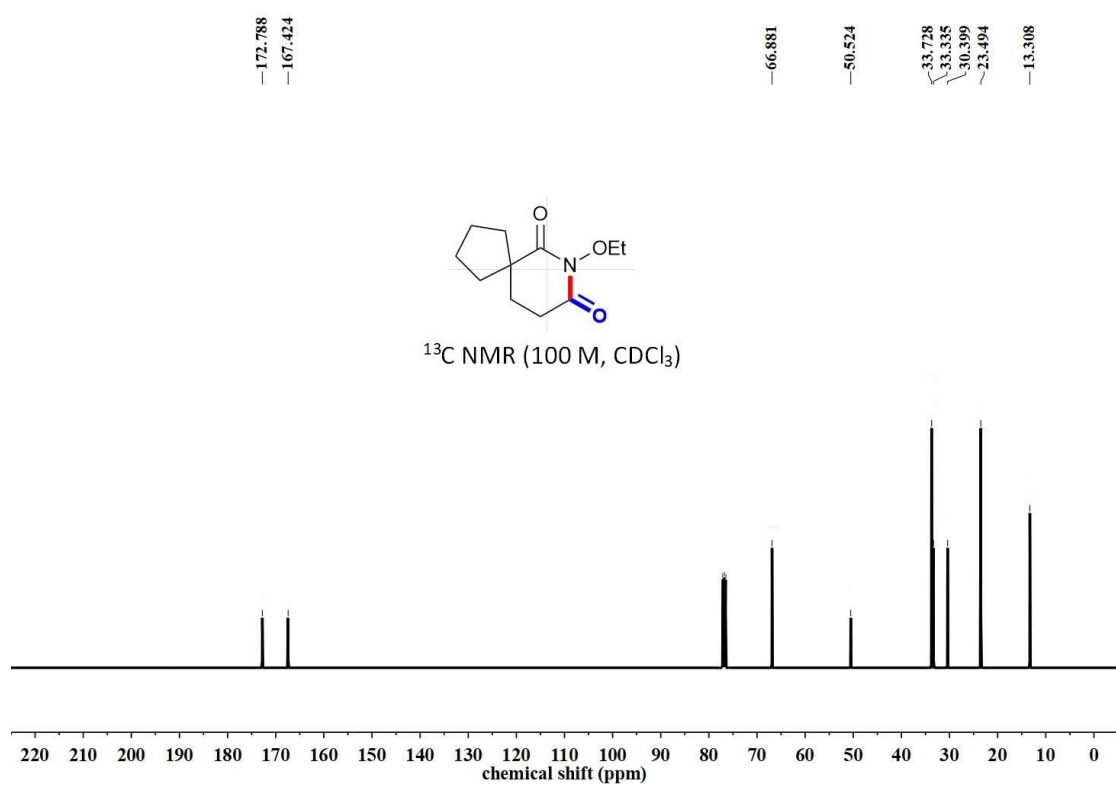
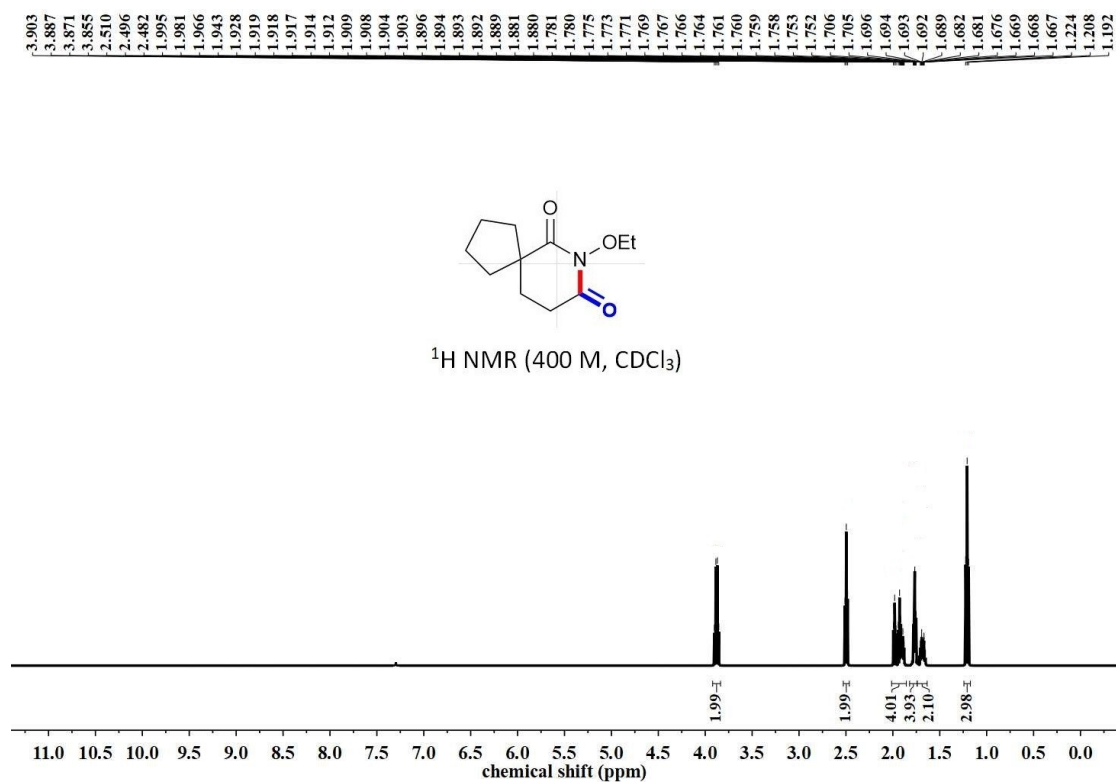


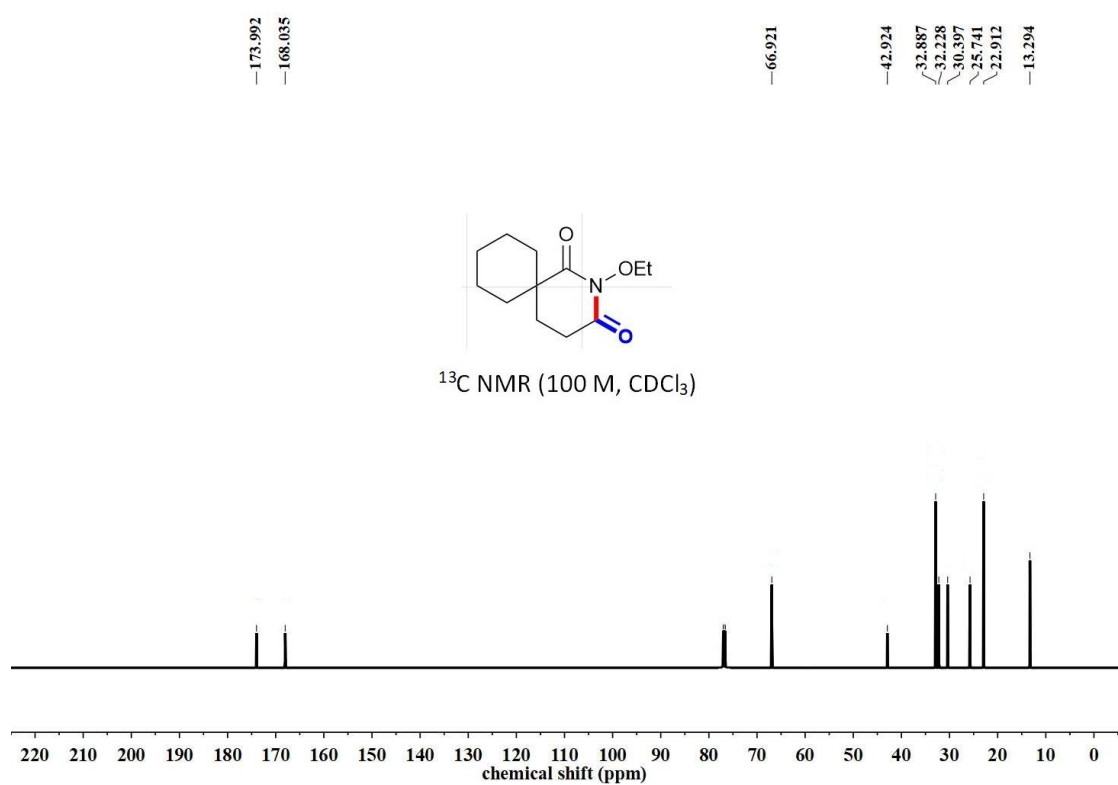
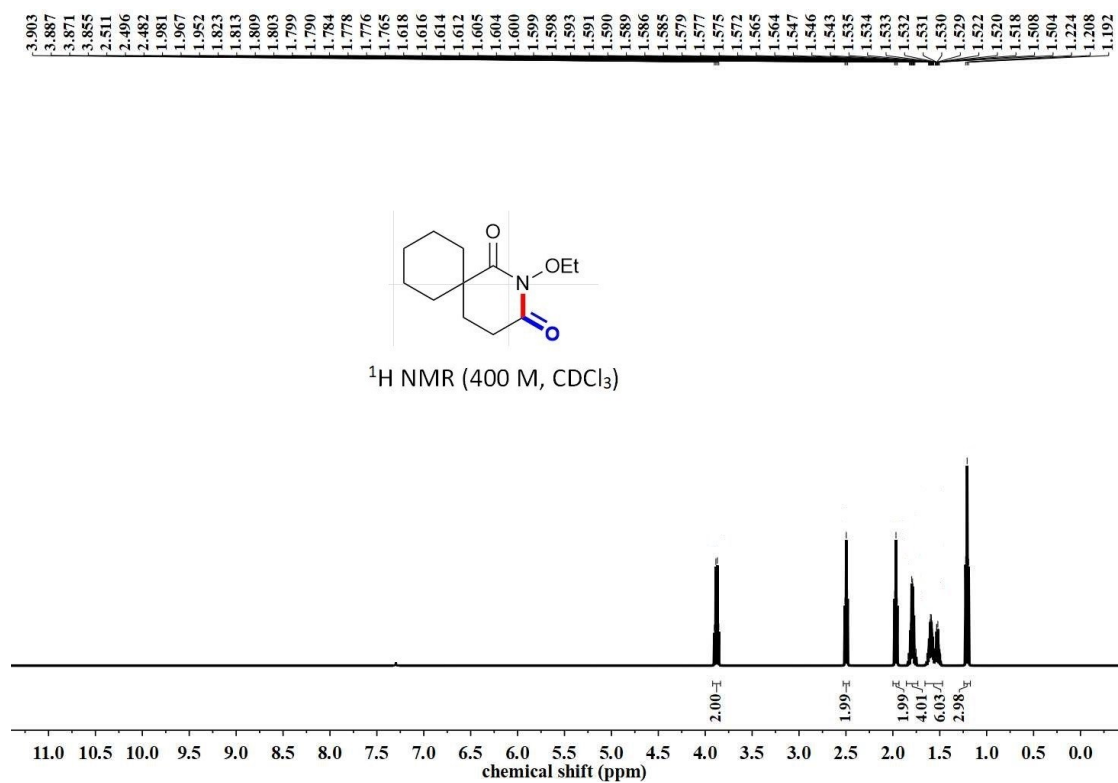


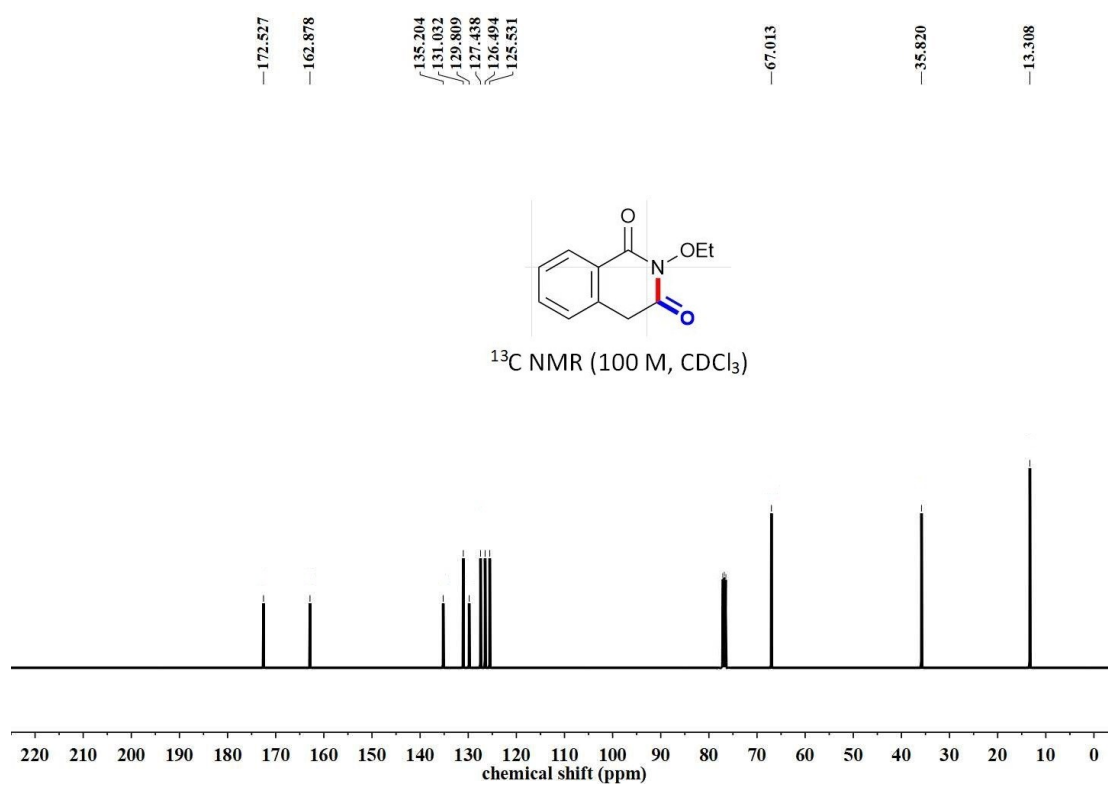
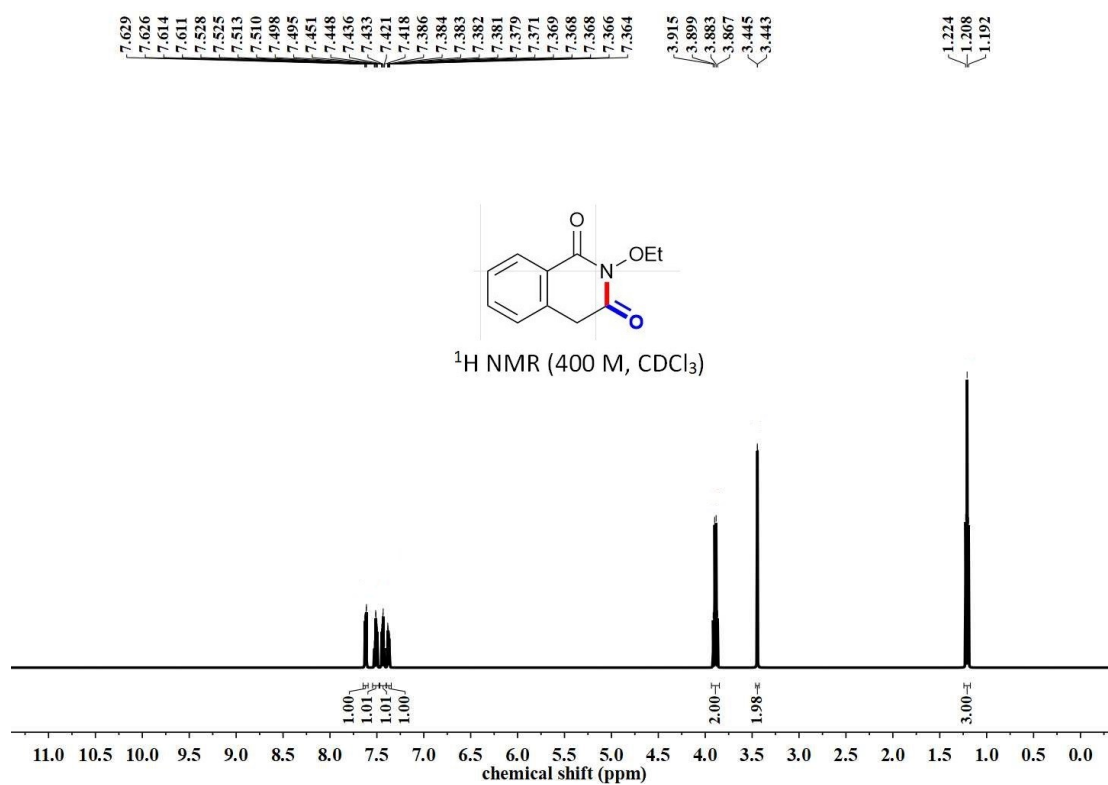












4. ESI-MS measurement: instrument parameters and results

Acquisition Mass Control	
Detection Mode	Broad band
TD (Acquisition)	1 M
Low Mass	57.75 m/z
High Mass	2000.00 m/z
Estimated R.P.	38000
Transient Length	0.2097 sec
Data transfer mode	standard

API Source		Source Gas Tune		Source Gas Control	
Source Type	ESI	Dry Gas	4.0 L/min	Drying Gas Flow	Yes
Capillary -	2600 V	Dry Temp	100 °C	Drying Gas Heater	Yes
End Plate Offset	-500 V	Nebulizer	1.0 bar	Nebulizer Gas Flow	Yes
Corona Needle	0.0 nA	Vaporizar Temp	10.0 °C	Nebulizer Gas Heater	No

Figure S1 ESI-MS spectra of mechanism research

