

# Site-specific hydrogen-atom elimination in photoexcited ethyl radical

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## Supporting Information

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# 1 H-atom action spectra

H-atom fragment action spectra from the photodissociation of the  $\text{CH}_3\text{CH}_2\text{I}$ ,  $\text{CH}_3\text{CD}_2\text{I}$  and  $\text{CD}_3\text{CH}_2\text{I}$  precursors and D-atom action spectrum from the photodissociation of the  $\text{CD}_3\text{CD}_2\text{I}$  precursor, as a function of the excitation wavelength, which varies in the 199.5-201.4 nm range, are depicted in Figure S1. We remark that the same nanosecond pump laser photodissociates the precursor molecule and excites the corresponding ethyl radical, while the H-atom (D-atom) fragment produced is detected in this case by a (2+1) resonance enhanced multiphoton ionization (REMPI) scheme at 243.1 nm (243.06 nm). See the main text for details about the experimental methodology employed.

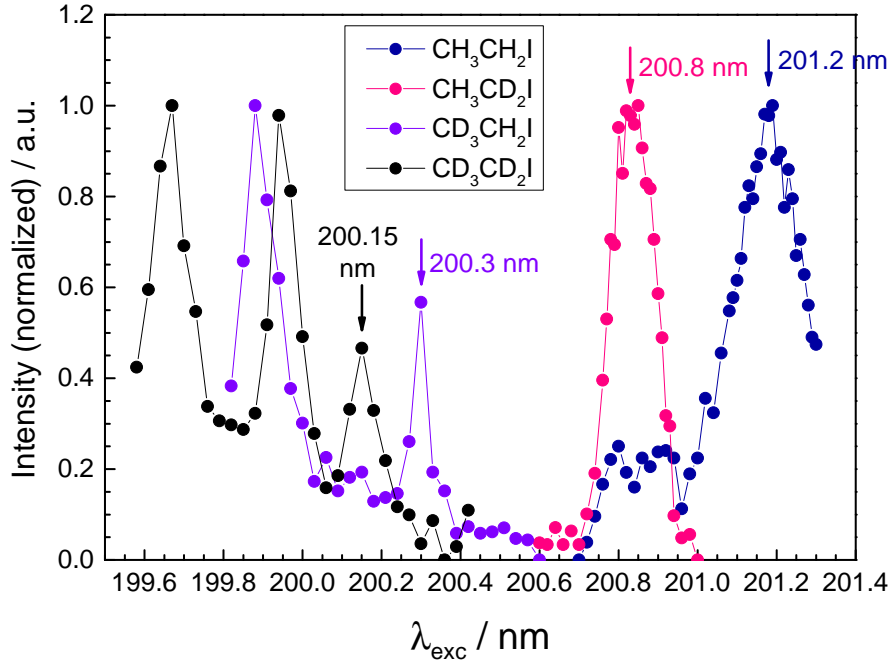


Figure S1: H-atom action spectra from the photodissociation of the  $\text{CH}_3\text{CH}_2\text{I}$ ,  $\text{CH}_3\text{CD}_2\text{I}$  and  $\text{CD}_3\text{CH}_2\text{I}$  precursors as a function of the excitation wavelength,  $\lambda_{\text{exc}}$ . A (2+1) REMPI scheme at 243.1 nm is employed to detect the H-atoms produced from the secondary dissociation of the corresponding ethyl radical. Vertical arrows highlight the selected excitation wavelength corresponding to the origin  $0_0^0$  band for each isotopologue occurring at 201.2 nm for  $\text{CH}_3\text{CH}_2\text{I}$ , 200.8 nm for  $\text{CH}_3\text{CD}_2\text{I}$  and 200.3 nm for  $\text{CD}_3\text{CH}_2\text{I}$ . The D-atom action spectrum from the photodissociation of the  $\text{CD}_3\text{CD}_2\text{I}$  with D-atom detection at 243.06 nm is also depicted. The  $0_0^0$  band for  $\text{CD}_3\text{CD}_2\text{I}$  occurs at 200.15 nm.

## 2 D-atom images and translational energy distributions

The D-atom images measured are depicted in Fig. S2 along with the corresponding translational energy distribution. A weak ring is indeed observed in the image while a shoulder is recovered in the TED for the isotopologues  $\text{CD}_3\text{CD}_2$  and  $\text{CH}_3\text{CD}_2$ .

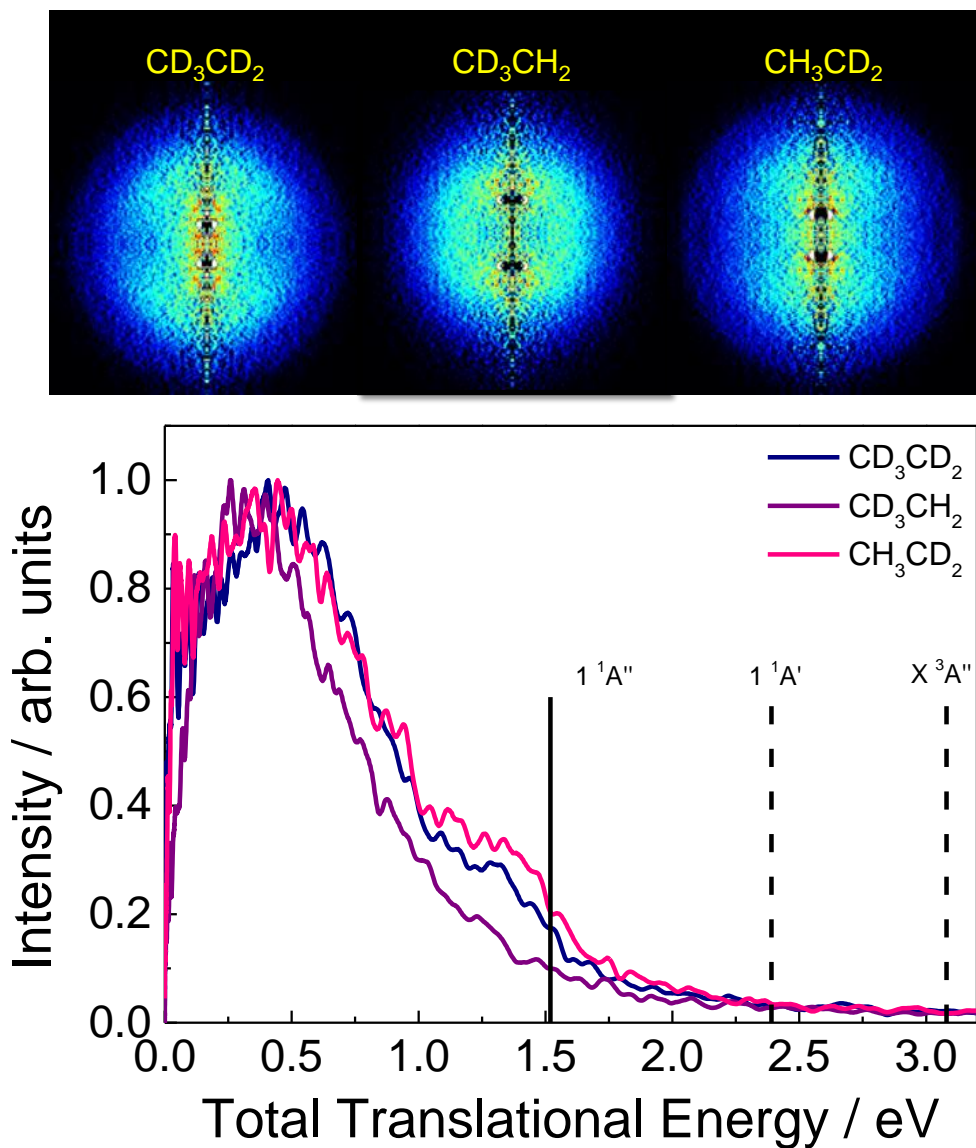


Figure S2: Top: D-atom Abel-inverted velocity map images for the three isotopologues  $\text{CD}_3\text{CD}_2$ ,  $\text{CH}_3\text{CD}_2$  and  $\text{CD}_3\text{CH}_2$ , excited at 200.15 , 200.8 and 200.3 nm, respectively (see Fig.S1). Bottom: Total translational energy distributions (TED) extracted from the images. Vertical bars specify the available energy for the three first  $\text{C}_\alpha$  dissociation limits (see the text for more details). The TEDs have been normalized independently with respect to their corresponding intensity maximum.

### 3 Computed vibrational frequencies, zero-point energies and dissociation energies

The computed vibrational harmonic and anharmonic corrected frequencies for  $\text{CH}_3\text{CH}_2$ ,  $\text{CH}_2\text{CH}_2$  and  $\text{CH}_3\text{CH}$  are summarized in Tables 1, 2 and 3, respectively. We note that the model employed to estimate anharmonic correction is not adapted for non-rigid doublet species. See the main text for details about the computational methodology employed. Table 4 shows the harmonic and anharmonic corrected zero point energies for those species. In Table 5, the harmonic zero point energies along with the dissociation energies  $D_0$  of all the isotopologues are summarized.

Table 1: Harmonic and anharmonic corrected vibrational frequencies of ethyl  $\text{CH}_3\text{CH}_2$  radical in  $\text{cm}^{-1}$ .

Vibrational mode	Harmonic frequency	Anharmonic frequency
$\nu_1$	3319.96	3186.11
$\nu_2$	3210.54	3098.66
$\nu_3$	3161.74	3009.03
$\nu_4$	3115.07	2980.35
$\nu_5$	3031.05	2942.54
$\nu_6$	1509.02	1465.53
$\nu_7$	1508.56	1460.03
$\nu_8$	1493.82	1456.08
$\nu_9$	1415.69	1377.71
$\nu_{10}$	1211.56	1177.93
$\nu_{11}$	1087.64	1064.21
$\nu_{12}$	995.42	935.05
$\nu_{13}$	817.53	821.46
$\nu_{14}$	471.15	477.30
$\nu_{15}$	134.40	-4.79*

Table 2: Harmonic and anharmonic corrected vibrational frequencies of the  $\text{CH}_3\text{CH}$  radical in  $\text{cm}^{-1}$ .

Vibrational mode	Harmonic frequency	Anharmonic frequency
$\nu_1$	3277.52	3159.86
$\nu_2$	3143.65	2996.24
$\nu_3$	3099.27	2958.97
$\nu_4$	3033.50	2928.73
$\nu_5$	1490.14	1439.97
$\nu_6$	1486.16	1441.37
$\nu_7$	1407.18	1368.88
$\nu_8$	1132.58	1081.39
$\nu_9$	1101.70	1043.48
$\nu_{10}$	1009.70	1007.46
$\nu_{11}$	790.67	776.88
$\nu_{12}$	194.41	125.07

Table 3: Harmonic and anharmonic corrected vibrational frequencies of ethylene  $\text{CH}_2\text{CH}_2$  in  $\text{cm}^{-1}$ .

Vibrational mode	Harmonic frequency	Anharmonic frequency
$\nu_1$	3288.60	3149.67
$\nu_2$	3260.94	3126.90
$\nu_3$	3191.86	3054.51
$\nu_4$	3172.84	3011.40
$\nu_5$	1677.76	1628.39
$\nu_6$	1478.99	1445.19
$\nu_7$	1379.59	1352.42
$\nu_8$	1244.88	1231.31
$\nu_9$	1047.13	1142.59
$\nu_{10}$	945.70	1005.77
$\nu_{11}$	975.49	967.71
$\nu_{12}$	823.29	826.67

Table 4: Harmonic and anharmonic corrected vibrational zero-point energies (ZPE) of the corresponding species in  $\text{cm}^{-1}$ .

Species	Harmonic ZPE	Anharmonic ZPE
$\text{CH}_3\text{CH}_2$	13241.58	13025.94
$\text{CH}_3\text{CH}$	10583.24	10408.38
$\text{CH}_2\text{CH}_2$	11243.54	11155.92

Table 5: Harmonic zero-point energies (ZPE) and dissociation energies,  $D_0$ , of all the isotopologues in  $\text{cm}^{-1}$  and eV, respectively.

Species	Harmonic ZPE	$D_0$
$\text{CH}_3\text{CH}_2$	13241.58	4.35
$\text{CH}_3\text{CH}$	10583.24	–
$\text{CH}_2\text{CH}_2$	11243.54	–
$\text{CD}_3\text{CD}_2$	9799.4	4.45
$\text{CD}_3\text{CD}$	7752.69	–
$\text{CD}_2\text{CD}_2$	8495.65	–
$\text{CD}_3\text{CH}_2$	11003.72	4.36
$\text{CD}_3\text{CH}$	8290.85	–
$\text{CD}_2\text{CH}_2$	9878.8	–
$\text{CH}_3\text{CD}_2$	11893.74	4.44
$\text{CH}_3\text{CD}$	9787.22	–

## 4 Conical Intersections

The resulting geometries from the optimization of the three conical intersections found are shown in Fig. S3.

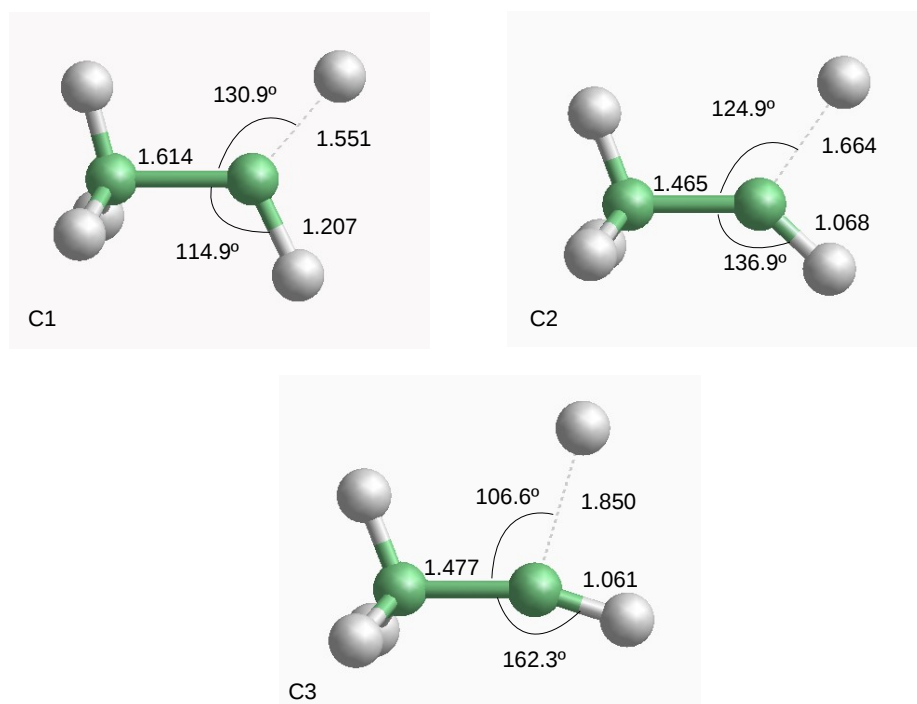


Figure S3: Geometries of the three conical intersections found and optimized, labelled C1, C2 and C3.