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Supporting Information - A general concept for the introduction of hydroxamic acids to polymers, T. Johann, J. Keth, M. Bros. H. Frey*

Supporting Information

A general concept for the introduction of hydroxamic acids into polymers

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Optimization of protected hydroxamic acids with aliphatic spacers o



Scheme S1. Overview over the synthesis attempts to yield protected hydroxamic acids with C4 and C3 aliphatic spacers.

SEC and LCST Polymer characterization



Figure S1. (a) SEC traces of PPO polymers. (b) SEC traces of PEEGE polymer. (c) benzylHAA-PEG polymers.



Figure S2. Cloud point measurement of HAA-PPO₂₁ in deionized water. Turbidity rapidly increased between 10 and 15 °C with 50 % of original transmittance at 12.9°C.

NMR and IR Characterization

Characterization of N-hydroxy-4-(2-hydroxyethoxy)benzamide (1b) "HA-OH"



Figure S3b. ¹³C NMR spectrum (75 MHz, DMSO-d₆) of N-hydroxy-4-(2-hydroxyethoxy)benzamide (1b).



Figure S3c. ¹H-¹H COSY NMR spectrum (300 MHz, DMSO-*d*₆) of *N*-hydroxy-4-(2-hydroxyethoxy)benzamide (1b).



Figure S3d. ¹H-¹³C HSQC NMR spectrum (300 MHz / 75 MHz, DMSO-*d*₆) of *N*-hydroxy-4-(2-hydroxyethoxy)benzamide (1b).



Figure S3e. ¹H-¹³C HMBC NMR spectrum (300 MHz / 75 MHz, DMSO-*d*₆) of *N*-hydroxy-4-(2-hydroxyethoxy)benzamide (1b).



Figure S3f. FT-ATR-IR spectrum of N-hydroxy-4-(2-hydroxyethoxy)benzamide (1b).

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Characterization of 2-(4-(5,5-dimethyl-1,4,2-dioxazol-3-yl)phenoxy)ethan-1-ol (1c) "HAA-OH"



8.0 7.8 7.6 7.4 7.2 7.0 6.8 6.6 6.4 6.2 6.0 5.8 5.6 5.4 5.2 5.0 4.8 4.6 4.4 4.2 4.0 3.8 3.6 3.4 3.2 3.0 2.8 2.6 2.4 2.2 2.0 1.8 1.6 Chemical shift (ppm)

Figure S4a. ¹H NMR spectrum (300 MHz, DMSO-*d*₆) of 2-(4-(5,5-dimethyl-1,4,2-dioxazol-3-yl)phenoxy)ethan-1-ol (1c). See synthesis procedure for detailed assignment of the signals.



Figure S4b. 13C NMR spetrcum (75 MHz, DMSO-d₆) of 2-(4-(5,5-dimethyl-1,4,2-dioxazol-3-yl)phenoxy)ethan-1-ol (1c) prior to purification via column chromatography.



Figure S4c. ¹H-¹H COSY NMR spetrum (300 MHz, DMSO-d₆) of 2-(4-(5,5-dimethyl-1,4,2-dioxazol-3-yl)phenoxy)ethan-1-ol (1c) prior to purification via column chromatography.



Figure S4d. ¹H-¹³C HSQC NMR spectrum (300 MHz / 75 MHz, DMSO-*d*₆) of 2-(4-(5,5-dimethyl-1,4,2-dioxazol-3-yl)phenoxy)ethan-1-ol (1c) prior to purification via column chromatography.



Figure S4e. ¹H-¹³C HMBC NMR spectrum (300 MHz / 75 MHz, DMSO-*d*₆) of 2-(4-(5,5-dimethyl-1,4,2-dioxazol-3-yl)phenoxy)ethan-1-ol (1c) prior to purification via column chromatography.



Figure S4f. FT-ATR-IR spectrum of 2-(4-(5,5-dimethyl-1,4,2-dioxazol-3-yl)phenoxy)ethan-1-ol (1c) prior to purification via column chromatography.





58 166 164 162 160 158 156 154 152 150 148 146 144 142 140 138 136 134 132 130 128 126 124 122 120 118 116 114 Chemical shift [ppm]

Figure S5b. 13 C NMR spectrum (75 MHz, DMSO- d_6) of N,4-dihydroxybenzamide (2a).





Figure S5d. ¹H-¹³C HSQC NMR spectrum (300 MHz / 75 MHz, DMSO-*d*₆) of *N*,4-dihydroxybenzamide (2a).



Figure S5e. ¹H-¹³C HMBC NMR spectrum (300 MHz / 75 MHz, DMSO-*d*₆) of *N*,4-dihydroxybenzamide (2a).



Figure S5f. FT-ATR-IR spectrum of N,4-dihydroxybenzamide (2a).

Characterization of 4-(5,5-dimethyl-1,4,2-dioxazol-3-yl)phenol (2b)



Figure S6a. ¹H NMR spectrum (300 MHz, DMSO-*d*₆) of 4-(5,5-dimethyl-1,4,2-dioxazol-3-yl)phenol (2b). The ethyl ester residue originates from the side reaction during transketalization.



Figure S6b. ¹³C NMR spectrum (75 MHz, DMSO-d₆) of 4-(5,5-dimethyl-1,4,2-dioxazol-3-yl)phenol (2b).



Figure S6c. ¹H-¹H COSY NMR spectrum (300 MHz, DMSO-*d*₆) of 4-(5,5-dimethyl-1,4,2-dioxazol-3-yl)phenol (2b).



Figure S6d. ¹H-¹³C HSQC NMR spectrum (300 MHz / 75 MHz, DMSO-*d*₆) of 4-(5,5-dimethyl-1,4,2-dioxazol-3-yl)phenol (2b).



Figure S6e. ¹H-¹³C HMBC NMR spectrum (300 MHz / 75 MHz, DMSO-d₆) of 4-(5,5-dimethyl-1,4,2-dioxazol-3-yl)phenol (2b).

Characterization of N-hydroxy-4-(hydroxymethyl)benzamide (3a)



Figure S7b. ¹³C NMR spectrum (100 MHz, DMSO-*d*₆) of *N*-hydroxy-4-(hydroxymethyl)benzamide (3a).



Figure S7d. ¹H-¹³C HSQC NMR spectrum (400 MHz / 100 MHz, DMSO-d₆) of N-hydroxy-4-(hydroxymethyl)benzamide (3a).



Figure S7e. ¹H-¹³C HMBC NMR spectrum (400 MHz / 100 MHz, DMSO-d₆) of N-hydroxy-4-(hydroxymethyl)benzamide (3a).





7.8 7.6 7.4 7.2 7.0 6.8 6.6 6.4 6.2 6.0 5.8 5.6 5.4 5.2 5.0 4.8 4.6 4.4 4.2 4.0 3.8 3.6 3.4 3.2 3.0 2.8 2.6 2.4 2.2 2.0 1.8 1.6 1.4 1.2 Chemical shift [ppm]

Figure S8a. ¹H NMR spectrum (400 MHz, DMSO-d₆) of (4-(5,5-dimethyl-1,4,2-dioxazol-3-yl)phenyl)methanol (3b). See synthesis procedure for detailed assignment of the signals.



Figure S8b. ¹³C NMR spectrum (100 MHz, DMSO-*d*₆) of (4-(5,5-dimethyl-1,4,2-dioxazol-3-yl)phenyl)methanol (3b).



Figure S8d. ¹H-¹³C HSQC NMR spectrum (400 MHz / 100 MHz, DMSO-d₆) of (4-(5,5-dimethyl-1,4,2-dioxazol-3-yl)phenyl)methanol (3b).



Figure S8e. 1H-13C HMBC NMR spectrum (400 MHz / 100 MHz, DMSO-d₆) of (4-(5,5-dimethyl-1,4,2-dioxazol-3-yl)phenyl)methanol (3b).





.75 170 165 160 155 150 145 140 135 130 125 120 115 110 105 100 95 90 85 80 75 70 65 60 55 50 45 40 35 30 25 Chemical shift [ppm]

Figure S9b. ¹³C NMR spectrum (75 MHz, DMSO-d₆) of N,6-dihydroxyhexanamide (4a).



Figure S9c. ¹H-¹H COSY NMR spectrum (300 MHz, DMSO-*d*₆) of *N*,6-dihydroxyhexanamide (4a).



Figure S9d. ¹H-¹³C HSQC NMR spectrum (300 MHz / 100 MHz, DMSO-*d*₆) of *N*,6-dihydroxyhexanamide (4a).



Figure S9e. ¹H-¹³C HMBC NMR spectrum (300 MHz / 100 MHz, DMSO-*d*₆) of *N*,6-dihydroxyhexanamide (4a).



Figure S9f. FT-ATR-IR spectrum of N,6-dihydroxyhexanamide (4a).

Characterization of 5-(5,5-dimethyl-1,4,2-dioxazol-3-yl)pentan-1-ol (4b)



Figure S10. ¹H NMR spectrum (300 MHz, DMSO-*d*₆) of 5-(5,5-dimethyl-1,4,2-dioxazol-3-yl)pentan-1-ol (4b). The caprolactone signals originate from the side reaction during transketalization reaction.





7.8 7.6 7.4 7.2 7.0 6.8 6.6 6.4 6.2 6.0 5.8 5.6 5.4 5.2 5.0 4.8 4.6 4.4 4.2 4.0 3.8 3.6 3.4 3.2 3.0 2.8 2.6 2.4 2.2 2.0 1.8 1.6 1.4 1.4 Chemical shift [ppm]

Figure S11. ¹H NMR spectrum (400 MHz, DMSO-d₆) of 5,5-dimethyl-3-(4-(2-(oxiran-2-ylmethoxy)ethoxy)phonyl) -1,4,2-dioxazole (1d)

Characterization of 2-(4-(5,5-dimethyl-1,4,2-dioxazol-3-yl)phenoxy)ethyl methacrylate (1e)



Figure S12. ¹H NMR spectrum (300 MHz, DMSO-d₆) of 2-(4-(5,5-dimethyl-1,4,2-dioxazol-3-yl)phenoxy)ethyl methacrylate (1e).

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Characterization of HAA-PEG (P1a), HA-PEG (P1b) and benzylHAA-PEG (P2a) polymers



7.8 7.6 7.4 7.2 7.0 6.8 6.6 6.4 6.2 6.0 5.8 5.6 5.4 5.2 5.0 4.8 4.6 4.4 4.2 4.0 3.8 3.6 3.4 3.2 3.0 2.8 2.6 2.4 2.2 2.0 1.8 1.6 1. Chemical shift (ppm)

Figure S13a. Exemplary ¹H NMR spectrum (400 MHz, DMSO-d₆) of HAA-PEG (P1a). See synthesis procedure for detailed assignment of the signals.



Figure S13b. Exemplary ¹H NMR spectrum (400 MHz, DMSO-d₆) of HA-PEG (P1b). See synthesis procedure for detailed assignment of the signals.



Figure S14. Exemplary ¹H NMR spectrum (300 MHz, CDCl₃) of ^{benzyl}HAA-PEG (P2a).

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Characterization of HAA-PPO (P3a), benzylHAA-PPO (P4a), HAA-PEEGE (P5a) and benzylHAA-PEEGE (P6a) polymers



Figure S15. Exemplary ¹H NMR spectrum (300 MHz, CDCl₃) of HAA-PPO (P3a).



Figure S16. Exemplary ¹H NMR spectrum (300 MHz, CDCl₃) of ^{benzyl}HAA-PPO (P4a).



Figure S17. Exemplary ¹H NMR spectrum (300 MHz, CDCl₃) of HAA-PEEGE (P5a).

-7.26 CDCl3



Figure S18. Exemplary ¹H NMR spectrum (300 MHz, CDCl₃) of ^{benzyl}HAA-PEEGE (P6a).



Polymer mass spectroscopy characterization

Figure S19. ESI-MS spectrum of HAA-PEG₃₁. The circles denote the HAA-PEG distribution with sodium as a counter ion.



Figure S20. ESI-MS spectrum of HA-PEG₃₁. The circles denote the HA-PEG distribution with sodium as a counter ion.

Complexation properties



Figure S21. UV-Vis spectra of tris(HA-PEG)iron(III) (red) and mono(HA-PEG)iron(III) (blue) complexes in methanol.



Figure S22. Contact angle measurements by the sessile drop method. PEG₄₄ was used as a reference. In all cases, increased hydrophilicity of the corresponding surface was detected when HA-PEG₁₈₈ was employed as a coating material.

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Nanoparticle characterization



Figure S23. Proof of principle demonstration of the altered solubilisation behaviour of HA-PEG coated nanoparticles. Left image shows the non-modified nanoparticles solely soluble in *n*-hexane, while after coating with HA-PEG₁₈₈ the nanoparticles become freely soluble in methanol.

Sample	Z _{avg} diameter [nm]	PDI	Peak diameter [nm]	Peak diameter standard deviation [nm]
NP-Oleat	20.9	0.10	21.3	4.5
NP-Oleat + PEG ₄₄	22.2	0.10	23.7	6.7
NP-HA-PEG ₂₁	25.8	0.08	27.1	6.5
NP-HA-PEG ₈₀	35.8	0.07	38.9	9.7
NP-HA-PEG ₁₈₈	49.4	0.09	53.4	12.3



T-Cell proliferation assay and MTT metabolic activity analysis

Figure S24. MTT metabolic activity assay. In all cases no significant decrease in metabolic activity, translating to cell toxicity can be observed. Only in case of low MW HA "HA-OH" (1b) a subtle decrease in metabolic activity was detected, suggesting increased toxicity compared to HA-PEG₁₈₈. (Data denote the mean±SEM of values obtained for PBMC prepared from 4 different healthy donors and tested in triplicates.)



Figure S25. T-Cell proliferation assay. Human PBMC were stimulated with PHA to induce proliferation of T-cells in the presence of the different test compounds. In case of Desferal, proliferation was completely inhibited. Low MW HA compound "HA-OH" (1b) showed dose-dependent inhibition of proliferation. In contrast, no influence of HAA-PEG or HA-PEG was found regarding T-cell proliferation. (Data denote the mean±SEM of values obtained for PBMC prepared from 4 different healthy donors and tested in triplicates.)



Figure S26. T-Cell proliferation assay. Unstimulated human PBMC were incubated with different test compounds. No significant effect towards background proliferation were found. (Data denote the mean±SEM of values obtained for PBMC prepared from 4 different healthy donors and tested in triplicates.)