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> Supporting Information (Kato, Lin, Kuwayama, Nagase, Segawa, Scott, Itami) Two-step synthesis of a red-emissive warped nanographene derivative via a ten-fold C–H borylation

> > Supporting Information

# Two-step synthesis of a red-emissive warped nanographene derivative via a ten-fold C–H borylation

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#### **1. Experimental Section**

#### General

Unless otherwise noted, all materials including dry solvent were obtained from commercial suppliers and used without further purification. All reactions were performed with dry solvents under an atmosphere of nitrogen in dried glassware with standard vacuum-line techniques. Work-up and purification procedures were carried out with reagent-grade solvents under air. Warped nanographene  $(1)^{S1}$  was synthesized according to the reported procedure.

Analytical thin-layer chromatography (TLC) was performed using E. Merck silica gel 60 F254 precoated plates (0.25 mm). The developed chromatograms were analyzed by UV lamp (254 or 365 nm). Flash column chromatography was performed with KANTO Silica Gel 60N (spherical, neutral, 40-100  $\mu$ m). The high-resolution mass spectra (HRMS) were obtained from a JEOL JMS-S3000 SpiralTOF (MALDI-TOF MS). Melting points were measured on a MPA100 Optimelt automated melting point system. Preparative recycling gel permeation chromatography (GPC) was performed with a SHIMADZU Prominence high performance liquid chromatograph instrument equipped with JAIGEL-2H-40/JAIGEL-2H-40 columns using chloroform as an eluent. Nuclear magnetic resonance (NMR) spectra were recorded on a JEOL ECA600II spectrometer with UltraCool<sup>TM</sup> probe (<sup>1</sup>H 600 MHz, <sup>13</sup>C 150 MHz) or a JEOL ECA600 (<sup>19</sup>F 565 MHz). Chemical shifts for <sup>1</sup>H, <sup>13</sup>C, and <sup>19</sup>F NMR are expressed in parts per million (ppm) relative to C<sub>2</sub>HDCl<sub>4</sub> ( $\delta$  5.98 ppm), C<sub>2</sub>D<sub>2</sub>Cl<sub>4</sub> ( $\delta$  73.79 ppm), and C<sub>6</sub>F<sub>6</sub> ( $\delta$  –163 ppm), respectively. Data are reported as follows: chemical shift, multiplicity (s = singlet, m = multiplet, br = broad) and integration.





To a 20 mL J-Young Schlenk equipped with a magnetic stirring bar were added a solution of 1 (99.1 mg, 100  $\mu$ mol, 1.0 equiv) in dry cyclopentyl methyl ether (CPME; 1.0 mL), 4,4,5,5-tetramethyl-1,3,2-dioxaborolane (HBpin; 384 mg, 3.00 mmol, 30 equiv), [IrOMe(cod)]<sub>2</sub> (16.6 mg, 25.0  $\mu$ mol, 25 mol%), and 3,4,7,8-tetramethyl-1,10-phenanthroline (tmphen; 11.8 mg, 50.0  $\mu$ mol, 50 mol%) under nitrogen atmosphere. The Schlenk was sealed with a J-Young screw tap and the resultant mixture was stirred at 120 °C for 24 h. After cooling the mixture to room temperature, the reaction mixture was diluted in chloroform and passed short-path silica gel chromatography with chloroform. The solution was removed under reduce pressure. The crude material was purified by GPC (eluent: chloroform) to afford **2** (170 mg, 75% yield) as an orange solid.

<sup>1</sup>H NMR (600 MHz, C<sub>2</sub>D<sub>2</sub>Cl<sub>4</sub>, 140 °C)  $\delta$  9.08 (s, 10H), 7.85 (s, 10H), 1.46 (s, 120H); <sup>13</sup>C NMR (150 MHz, C<sub>2</sub>D<sub>2</sub>Cl<sub>4</sub>, 140 °C)  $\delta$  138.2 (4°), 136.8 (CH), 136.0 (4°), 133.0 (4°), 130.8 (4°), 129.2 (CH), 126.9 (4°), 83.83 (4°), 24.74 (CH<sub>3</sub>). HRMS (MALDI-TOF MS) *m*/*z* calcd for C<sub>140</sub>H<sub>140</sub>B<sub>10</sub>O<sub>20</sub> [M·]<sup>+</sup>: 2251.0970, found: 2251.0963; mp: >300 °C.

#### Synthesis of 3a



To a 3 mL screw-capped pressure vessel equipped with a magnetic stirring bar were added a solution of  $Pd_2(dba)_3 \cdot CHCl_3$ (51.8)mg, 50.0 µmol, 5.0 equiv), 2-dicyclohexylphosphino-2',6'-dimethoxybiphenyl (SPhos; 41.1 mg, 100 µmol, 5.0 equiv) in toluene (1.0 mL), 2 (22.5 mg, 10.0 µmol, 1.0 equiv), 1-iodo-4-(trifluoromethyl)benzene (4a; 68.7 mg, 300 µmol, 30 equiv), and a solution of Cs<sub>2</sub>CO<sub>3</sub> (19.6 mg, 60.0 µmol, 6.0 equiv) in H<sub>2</sub>O (0.50 mL) under nitrogen atmosphere. The vessel was sealed with a PTFE-coated screw cap and the resultant mixture was stirred at 100 °C for 3 h under nitrogen atmosphere. After cooling the mixture to room temperature, the reaction mixture was extracted with chloroform. The combined organic layer was dried over MgSO<sub>4</sub>, and the solvent was removed under reduce pressure. The crude material was purified by silica-gel column chromatography (eluent: hexane/EtOAc = 6:1) and GPC (eluent: chloroform) to afford **3a** (12.0 mg, 60%) as a reddish orange solid.

<sup>1</sup>H NMR (600 MHz, C<sub>2</sub>D<sub>2</sub>Cl<sub>4</sub>, 140 °C)  $\delta$  7.80–8.00 (br m, 50H), 8.91 (br s, 10H); <sup>13</sup>C NMR (150 MHz, C<sub>2</sub>D<sub>2</sub>Cl<sub>4</sub>, 85 °C)  $\delta$  125.9 (CH), 127.6 (CH); <sup>19</sup>F NMR (565 MHz, C<sub>2</sub>D<sub>2</sub>Cl<sub>4</sub>, 140 °C)  $\delta$  –63.7 (s, 30F); HRMS (MALDI-TOF MS) *m/z* calcd for C<sub>150</sub>H<sub>60</sub>F<sub>30</sub> [M·]<sup>+</sup>: 2430.4216, found: 2430.4290; mp: >300 °C.

#### Synthesis of 3b



To a 3 mL screw-capped pressure vessel equipped with a magnetic stirring bar were added a solution of  $Pd_2(dba)_3 \cdot CHCl_3$  (51.8 mg, 50.0 µmol, 5.0 equiv), SPhos (41.1 mg, 100 µmol, 5.0 equiv) in toluene (1.0 mL), **2** (22.5 mg, 10.0 µmol, 1.0 equiv), 4-bromo-*N*,*N*-diphenylaniline (**4b**; 92.3 mg, 300 µmol, 30 equiv), and a solution of  $Cs_2CO_3$  (19.6 mg, 60.0 µmol, 6.0 equiv) in H<sub>2</sub>O (0.50 mL) under nitrogen atmosphere. The vessel was sealed with a PTFE-coated screw cap and the resultant mixture was stirred at 100 °C for 3 h under nitrogen. After cooling the mixture to room temperature, the reaction mixture was extracted with  $CH_2Cl_2$ . The combined organic layer was dried over MgSO<sub>4</sub>, and the solvent was removed under reduce pressure. The crude material was purified by silica-gel column chromatography (eluent: hexane/EtOAc = 6:1 and chloroform) and GPC (eluent: chloroform) to afford **3b** (19.9 mg, 58%) as a red solid.

<sup>1</sup>H NMR (600 MHz, C<sub>2</sub>D<sub>2</sub>Cl<sub>4</sub>, 140 °C)  $\delta$  7.08–7.30 (br m, 120H), 7.76–7.85 (br m, 30H), 8.87 (br s, 10H); <sup>13</sup>C NMR (150 MHz, C<sub>2</sub>D<sub>2</sub>Cl<sub>4</sub>, 85 °C)  $\delta$  123.1 (CH), 123.4 (CH), 124.6 (CH), 129.2 (CH), 145.1 (CH), 147.4 (CH); HRMS (MALDI-TOF MS) *m/z* calcd for C<sub>260</sub>H<sub>160</sub>N<sub>10</sub> [M·]<sup>+</sup>: 3423.2895, found: 3423.2880; mp: >300 °C.

#### 2. X-ray Crystallography

Details of the crystal data and a summary of the intensity data collection parameters for 2 are listed in Table S1. A suitable crystal was mounted with mineral oil on a MiTeGen MicroMeshes and transferred to the goniometer of a Rigaku PILATUS diffractometer. Graphite-monochromated Mo Ka radiation was used. The structures were solved by direct methods with (SIR-97)<sup>S2</sup> and refined by full-matrix least-squares techniques against  $F^2$  (SHELXL-2014/7)<sup>S3</sup> by using Yadokari-XG software package.<sup>84</sup> The intensities were corrected for Lorentz and polarization effects. The non-hydrogen atoms were refined anisotropically. Hydrogen atoms were placed using AFIX instructions. CCDC 1893066 contain the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data request/cif. Alert A (VERY LARGE Solvent Accessible VOID(S) in Structure) is caused by the disorder of solvent molecules. Suitable solvent molecules could not be placed because of the lack of observed electron density.

	$2 \cdot 1.5 C_4 H_8 O$
formula	$C_{146}H_{155}B_{10}O_{21.5}$
fw	2361.79
<i>T</i> (K)	123(2)
$\lambda$ (Å)	0.71073
cryst syst	Triclinic
space group	<i>P</i> -1
a (Å)	17.1883(7)
<i>b</i> (Å)	19.8237(8)
<i>c</i> (Å)	25.0041(10)
$\alpha$ (deg)	112.264(4)
$\beta$ (deg)	93.732(3)
$\gamma(\text{deg})$	109.808(4)
$V(\text{\AA}^3)$	7232.6(6)
Ζ	2
$D_{\text{calc}} \left( \mathbf{g} \cdot \mathbf{cm}^{-3} \right)$	1.084
$\mu$ (mm <sup>-1</sup> )	0.070
F(000)	2506
cryst size (mm)	$0.10\times0.10\times0.10$
$\theta$ range (deg)	1.790-25.000
reflns collected	105367
indep reflns/R <sub>int</sub>	25435/ 0.2479
params	1653
GOF on $F^2$	1.064
$R_1, wR_2 [I > 2\sigma(I)]$	0.1266, 0.3072
$R_1$ , $wR_2$ (all data)	0.3042, 0.4142

 Table S1. Crystallographic data and structure refinement details of 2



Figure S1. ORTEP drawing of 2 with 50% probability. All hydrogen atoms and solvent molecules  $(Et_2O)$  are omitted for clarity.

#### 3. Photophysical measurement

UV–Vis absorption spectra were recorded on a Shimadzu UV-3510 spectrometer with a resolution of 0.5 nm. Emission spectra were measured on Shimadzu RF-6000 spectrometer with a resolution of 0.4 nm. Absolute fluorescence quantum yields ( $\Phi_F$ ) were determined on a Shimadzu RF-6000 using a calibrated integrating sphere system upon excitation at 340 nm. For FL lifetime measurements, Hamamatsu Photonics Quantaurus-Tau® fluorescence lifetime spectrometer C11367-21 with LED as a light source was used.

**Table S2.** The fluorescence quantum yield ( $\Phi_F$ ), lifetime ( $\tau$ ), radiative ( $k_r$ ) and nonradiative ( $k_{nr}$ ) decay rate constants of 1, 2, 3a, and 3b.

	$arPsi_{ m F}$	$\tau(ns)$	$k_{\rm r}  ({ m s}^{-1})^a$	$k_{\rm nr}~({ m s}^{-1})^a$
1	0.26	6.86	$3.8  imes 10^7$	$1.1  imes 10^8$
2	0.20	6.34	$3.2 \times 10^7$	$1.3  imes 10^8$
<b>3</b> a	0.28	6.33	$4.4 \times 10^7$	$1.1  imes 10^8$
<b>3</b> b	0.47	6.61	$7.1 \times 10^7$	$8.0  imes 10^7$

<sup>*a*</sup>  $\boldsymbol{\Phi}_{\mathrm{F}} = k_{\mathrm{r}} \times \tau, k_{\mathrm{r}} + k_{\mathrm{nr}} = \tau^{-1}$ 

#### 4. Computational Study

The Gaussian16 program<sup>S5</sup> running on a SGI Altix4700 system was used for optimization (B3LYP/6-31G(d)).<sup>S6</sup> Structures were optimized without any symmetry assumptions. Zero-point energy, enthalpy, and Gibbs free energy at 298.15 K and 1 atm were estimated from the gas-phase studies. Harmonic vibration frequency calculation at the same level was performed to verify all stationary points as local minima (with no imaginary frequency). Visualization of the results was performed by use of GaussView 5.0.9 software.

Excitation [nm]	Oscillator strength (f)	Description
506	0.131	$H \rightarrow L, H-1 \rightarrow L+1$
492	0.004	H–2, H–1→L
		$H \rightarrow L+1$
479	0.005	H–3, H–2, H–1→L
467	0.048	H–3, $H$ –1, $H$ – $L$
		H→L

Table S3. TD-DFT vertical one-electron excitations of 2
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Figure S2. Distributions and energy levels of molecular orbitals of 2 (isovalue = 0.02).

Excitation [nm]	Oscillator strength (f)	Description
515	0.180	H→L
		$H-1 \rightarrow L+1$
500	0.004	H−2, H−1→L
		$H \rightarrow L+1$
486	0.008	H−3, H−2, H−1→L
474	0.093	H–3, H–2, H–1, H→L
		H–2, H–1→L+1

Table S4. TD-DFT vertical one-electron excitations of 3a

H: HOMO, L: LUMO.



Figure S3. Distributions and energy levels of molecular orbitals of 3a (isovalue = 0.02).

Excitation [nm]	Oscillator strength (f)	Description
548	0.321	H–10, H→L
		$H-1 \rightarrow L+1$
530	0.018	H–11, H–1→L
		$H \rightarrow L+1$
524	0.011	H–12, H–3, H–2, H–1→L
515	0.033	H–13, H–3, H–2→L
		$H-1 \rightarrow L+1$

 Table S5. TD-DFT vertical one-electron excitations of 3b.

H: HOMO, L: LUMO.



Figure S4. Distributions and energy levels of molecular orbitals of 3b (isovalue = 0.02).

	E	E + ZPE	Н	G
2	-7173.63841675	-7171.085028	-7170.933750	-7171.283292
<b>3</b> a	-8747.66122956	-8745.979531	-8745.845710	-8746.184210
3b	-10551.7776533	-10548.364020	-10548.159908	-10548.662104

Table S6. Uncorrected and thermal-corrected (298 K) energies of stationary points (Hartree).<sup>a</sup>

a) *E*: electronic energy; *ZPE*: zero-point energy;  $H (= E + ZPE + E_{vib} + E_{rot} + E_{trans} + RT)$ : sum of electronic and thermal enthalpies; G (= H - TS): sum of electronic and thermal free energies.

Table S7. Cartesian coordinates of optimized structures.

2											
С	-0.220175	-1.197262	-0.449534	С	7.799102	-5.087100	5.283841	Н	-10.948520	2.353070	3.669045
С	1.031916	-0.599548	-0.582521	0	6.523862	-5.111293	4.584329	Н	-10.430127	-1.314805	3.330503
С	0.837904	0.790039	-0.705389	0	8.062595	-4.390428	-3.099778	Н	-11.572331	-0.084836	3.909544
С	-0.537443	1.053270	-0.596057	С	8.870246	-4.891923	-4.200757	Н	-9.852452	0.315265	3.706284
С	-1.207595	-0.190553	-0.547502	С	9.325062	-3.559368	-4.917652	Н	-11.930851	-1.744940	1.439166
С	-0.434237	-2.473379	-0.007007	0	8.253889	-2.639240	-4.568394	н	-13.050155	-0.406832	1.753364
С	0.792838	-3.125094	0.474618	0	6.108919	5.301426	-5.078815	Н	-12.223109	-0.533353	0.184758
С	2.090034	-2.520818	0.280781	С	7.049857	5.666207	-6.126397	Н	-10.666863	-2.802638	-4.011039
С	2.230658	-1.205055	-0.331278	С	7.284118	4.288315	-6.862073	Н	-9.888137	-4.359105	-3.686783
С	3.353715	-0.318340	-0.691629	0	7.016464	3.324570	-5.805603	Н	-10.971078	-4.183291	-5.085639
С	3.161004	1.081274	-0.877488	0	5.210644	7.796057	2.483167	Н	-10.089621	-1.462762	-5.986042
С	1.838737	1.712466	-0.648381	С	6.497945	8.328156	2.903004	н	-8.658576	-1.813323	-6.963926
Ċ	1.362339	3.040204	-0.290533	Ċ	7.434335	7.940076	1.690820	н	-10.129736	-2.792585	-7.157573
С	-0.042578	3.302465	-0.032934	0	6.764354	6.770851	1.143624	н	-9.305097	-6.081842	-5.107033
Ċ	-1.094267	2.271495	-0.290425	0	-2.589298	7.516394	4.474958	Н	-9.572863	-5.672147	-6.816219
Ċ	-2.579599	2.177603	-0.195997	C	-2.513359	8.531261	5.513866	Н	-8.090410	-6.507122	-6.319721
С	-3.289423	0.899729	-0.345145	С	-1.050935	8.327810	6.077095	н	-6.527294	-4.848919	-7.228403
Ċ	-2.552413	-0.382545	-0.412427	0	-0.355065	7,758439	4.933325	н	-7.898905	-3.999044	-7.969283
Č	-2.869567	-1.847160	-0.409112	Õ	-6.923808	6.979463	-1.762893	Н	-6.744191	-3.115247	-6.947312
Č	-1.859152	-2.829111	-0.116180	Č	-7.374461	8.354254	-1.910482	Н	-1.753976	-9.908129	-3.568369
Ċ	0.673339	-4.210068	1.441149	C	-6.010990	9.145526	-2.018085	Н	-2.542368	-11.448893	-3.971577
Č	1.705445	-4.400526	2.402418	Õ	-5.093206	8.279454	-1.294229	Н	-3.282630	-9.910944	-4.460074
Č	3.249632	-3.094283	0.954511	Õ	-1.242802	-7.277926	5.282862	Н	-5.379440	-10.641080	-3.421192
Č	3.042824	-3.894067	2.113320	Č	-1.064752	-8.061654	6.495168	Н	-4.736246	-12.115479	-2.676993
Č	4.145556	-4.264828	2.904962	Č	0.097999	-7.284471	7.229055	Н	-5.525764	-10.876968	-1.674924
Č	5.447467	-3.909223	2.570572	Õ	0.802673	-6.662608	6.118321	Н	-2.033756	-12.744961	-1.683772
Ċ	5.642308	-3.211681	1.365732	C	-8.275288	8.436871	-3.141531	Н	-1.065648	-11.991910	-0.403910
Ċ	4.586009	-2.819837	0.543650	C	-8.177656	8.692457	-0.647075	Н	-0.881572	-11.449166	-2.076714
Ċ	4.563682	-0.939412	-1.203315	C	-5.477643	9.249487	-3.453470	Н	-2.960164	-11.279664	0.988366
Ċ	4.986439	-2.223317	-0.763691	C	-6.002367	10.520821	-1.353263	Н	-4.369299	-10.466376	0.290836
Ċ	5.960446	-2.898518	-1.494417	C -1	11.801513	2.077285	0.315956	н	-4.050417	-12.172873	-0.091738
Ċ	6.582670	-2.331975	-2.624257	C -1	11.565678	2.496598	2.779423	н	1.564752	-8.894307	7.341927
С	6.256495	-1.020507	-2.956401	C -1	10.688178	-0.252304	3.285164	н	1.869580	-7.535961	8.432936
Ċ	5.276457	-0.294966	-2.252667	C -1	12.109430	-0.679604	1.261171	Н	0.583606	-8.696155	8.810984
Ċ	4.140668	1.824396	-1.650143	C -1	10.209628	-3.678633	-4.481456	Н	0.449556	-5.534016	8.435469
Ċ	5.025314	1.118152	-2.511530	C	-9.497107	-2.268079	-6.432134	Н	-1.106115	-5.498188	7.591537
Ċ	5.675376	1.823946	-3.542277	C	-8.818016	-5.744213	-6.024594	Н	-0.899122	-6.527119	9.025898
Ċ	5.539732	3.201152	-3.692528	C	-7.274411	-4.064637	-7.071952	Н	-3.600292	7.264541	6.919208
Ċ	4.809455	3.899514	-2.709272	Ċ	-2.720553 -	10.412329	-3.666161	н	-3.585023	8.989521	7.349956
Ċ	4.141944	3.245191	-1.679100	C	-4.872334 -	11.037039	-2.535452	Н	-4.605620	8.409253	6.021690
Ċ	2.368799	4.041979	0.055258	C	-1.605076 -	11.797708	-1.336539	Н	-2.741661	10.709496	5.543125
Ċ	3.670387	4.062663	-0.526560	C	-3.581916 -	11.199908	0.091297	Н	-1.926627	10.084666	4.092316
С	4.626017	4.948131	-0.025725	С	1.085415	-8.162331	7.995610	н	-3.672604	9.868916	4.286118
Ċ	4.356403	5.854072	1.013346	C	-0.404097	-6.145245	8.126610	Н	-0.247411	10.311847	5.659978
Ċ	3.063174	5.879173	1.519073	Ċ	-3.637878	8.279290	6.516742	н	-0.849014	10.102218	7.319761
Ċ	2.063940	5.004982	1.051722	C	-2.718312	9.885896	4.821564	Н	0.685068	9.363949	6.826349
Ċ	-0.347863	4.486383	0.765867	C	-0.328171	9.608131	6.491345	Н	0.080883	7.058624	7.399937
Ċ	0.687246	5.142584	1.496480	C	-0.971018	7.287939	7.203223	Н	-1.422890	7.658171	8.129676
С	0.359404	5.984045	2.574894	С	6.342814	9.827317	3.152136	Н	-1.471603	6.356693	6.920349
С	-0.954995	6.309583	2.884991	С	6.863380	7.611969	4.210439	Н	7.305045	10.281041	3.416505

С	-1.947487	5.836860	2.010890	С	7.458087	8.995071	0.576245	Н	5.941833	10.343453	2.276920
С	-1.666895	4.970587	0.952075	С	8.860486	7.544438	2.069495	н	5.651422	9,990497	3.985106
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c	2 821274	4 702316	0.064841	C	8 30/712	6 203300	5 423704	и Ц	6.060046	7 760817	4.036681
c	-2.021274	4.702310 5.929065	0.004841	C	8.304712	4.059309	-3.423794	11	6.071068	6.522646	4.950081
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С	-5.492451	4.521722	-0.687057	С	7.946801	-5.763566	-5.062436	Н	8.001106	9.895611	0.882392
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Ĉ	6 8 1 9 0 5 8	2 115887	0.253561	Č	9 355761	5 8/8108	3 3/0196	н	9 / 10 296	7 253727	1 168565
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C	-7.329040	0.955518	0.419/99	C	9.943033	-3.740040	4.381701	п	7.070913	7.054617	-/.81//04
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C	-3.113039	-3.927963	-2.379362	п	0.237898	-3.892022	-1.174926	п	9.436223	4.101/1/	-0.309930
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c	1 420862	5 256070	0.414521	и 11	2 076058	6 1 2 9 1 6 1	2 172274	и Ц	7 514275	6 5 4 0 1 2 6	4 435180
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р	6.007266	2.047940	1.549784	11	-4.194765	-0.505400	-2.101004	11	0.178898	-4.501455	-0.750257
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В	7.641260	-3.125206	-3.436341	Н	-1.618806	-6.351887	2.776673	Н	9.696298	-2.658236	-6.841489
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R	5 632997	7.014584	1 287737	н	8 610225	9 697247	0.699065	н	9 571365	2 821907	5 028212
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0	-9./41393	-0.213102	1.082105	п	-8.993800	7.970842	-0.342333	п	10.389939	-4.231070	5.506795
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C-	11.038722	1.693767	1.591566	Н	-4.448080	9.619210	-3.422667	Н	9.023634	-6.512537	6.378532
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Ĉ	-9 020424	-3 219038	-5 335895	н	-6 722651	11 192416	-1 834686	н	6 818475	-4 310859	7 037238
0	9.120424	2 487127	4 440128	и 11	6 241877	10.456422	0.280517	и 11	0.010475	7.008260	7.445601
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Ċ	8 802038	-4 660758	4 139925								
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Č	1.156837	2,232234	-0.318920	н	3,270317	6,700803	-0.305600	Ċ	0.337833	-7.909324	7.405498
$\tilde{c}$	1 700720	1 750122	0.662562	11 U	3 180554	6.061514	2 078601	с ц	7 367524	7 221 450	7 180720
c	-1./90/39	1./39132	-0.002303	H	0.001/02/	0.001340	2.078091	н	2.30/334	-1.221439	1.107/30
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Ĉ	-3 131815	1 168762	-0.887760	н	-6 355762	-3 716642	-1 271204	C	-9.081983	-4 865593	3 521775
c	2 637180	2 096299	-0 223722	н	-6 749101	-2 802931	0.952112	н	-8 085311	-4 564364	1 648947
c	0.130274	3 203210	0.050233	и Ц	4 182446	4 8 2 8 3 8 7	3 682508	C	8 068255	4 840461	4 014077
c	2 200586	0.709727	-0.039233	11	-4.182440	-4.828387	4 200442		-8.908255	4 459109	4.914077
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C	2.814522	-1.931380	-0.408625	н	1.415/6/	-0.370390	2.803890	н	-10.023287	-5.144240	3.039377
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C	0.480038	4.472202	0.727744	C	2.000183	8.072995	5.889909	C	-0.003217	3.430673	-4.346060
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С	4.766012	0.757012	-0.314189	Н	3.140269	8.193180	2.952248	Н	-6.633322	5.877860	-3.559658
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С	4.866041	3.217567	-0.359197	Н	0.658330	7.747579	7.220058	Н	-6.852992	3.851048	-8.077955
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c	2 438654	-7 074361	-0 790596	č	7 149622	6 931027	-1 164865	C	-10 129773	-5 272655	5 771278
c	3 265333	6 111547	1 355303	C	5 787438	8 861565	2 641810	C	0 504481	8 604217	8 71/7/6
c	5.005104	4 102275	2 520604	с ц	4 046005	7 644200	2 251001	C	2 120220	-0.004217	2 071258
c	6.020194	-4.103273	-2.550094	п	7.926140	2.044209	-2.331091	C	0.700229	-12.091420	-2.071338
C	6.032783	-3.220182	-2.903693	U U	7.836140	8.033462	-1.003443	C	9.799232	-4.536775	-7.119318
C	6.116681	-2.013236	-2.201359	Н	7.684023	6.201795	-0.563567	C	12.967384	0.603533	2.479047
С	5.176420	-1.632054	-1.243447	С	7.156750	9.003978	-2.404590	С	7.907254	10.16/64/	-2.992256
С	5.552228	-0.429303	-0.458823	Н	5.257204	9.613783	-3.216618	С	2.668854	9.568709	7.375136
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С	7.548756	0.722968	0.410023	С	8.929825	0.674288	0.945623	F	-9.369814	5.989355	-8.009933
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c	1 816674	4 932605	0 886884	н	9 601126	-0.919400	-0 351129	F	-11 315259	-5.001817	5 181674
c	2 138109	5 805128	1 928962	C	11 550954	0.602305	1 972562	F	10 120/18	4 652711	6 972251
c	1 174717	6 201806	2 810726	с ц	10.012507	2 102016	2 278112	E E	10 100515	6 604067	6.017604
c	0.151166	5 000784	2.519720	и П	11 887228	0.002222	0.564217	Г Б	0.286616	-0.004007	0.017004
c	-0.131100	5.999704	2.340447	п	7.009410	-0.992332	0.304317	Г	0.380010	-7.778713	9.707711
C	-2.8/023/	5.955138	1.536819	C	7.008410	-3.562722	-3.970203	F	-0.219330	-9.669633	8.886500
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С	-4.728517	4.059312	-2.657233	С	8.394243	-5.189002	-5.138259	F	9.130799	-4.880197	-8.245647
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С	-6.283588	-0.815707	-2.998295	Н	8.759788	-2.101060	-6.513381	F	13.752056	1.445614	1.765535
Ĉ	-6 632694	-2 127618	-2 701090	н	8 761383	-6 204749	-5 242346	F	13 532212	-0.621483	2 396724
č	-6 023787	-2 732201	-1 584652	C .	2 598474	-8 512868	-1 116516	F	13 039881	1 000322	3 769013
č	-5 041886	-2 090076	-0.8366/13	r C	2 91801/	-8 920906	_2 422821	F	3 606807	8 9300022	8 114722
c	1 672204	2.020070	0.465551		2.710714	0.520500	0 135265	г Б	1 626754	0.230203	8 107007
C	-4.0/2304	-2./10220	1.262010		2.431013	-7.004893	-0.133203	г г	1.020/04	7.020981	0.19/99/
C	-3./321//	-3.093076	1.203819		3.0/3898	-10.20/523	-2.13/219	F F	3.199162	10./5/919	1.012461
C	-3.392307	-3.803458	2.40393/	Н	3.01/498	-8.1/5818	-3.206/11	F	-8.21401/	9.409/64	4.554988
C	-4.304898	-4.188193	2.818251	С	2.584072	-10.853161	-0.444431	F	-7.891890	10.879652	2.933726
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С	0.521826	-5.762030	2.714205	Н	3.306053	-10.566152	-3.754239	F	8.378564	9.885082	-4.229615
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С	1.186988	-0.276072	-0.635198	Н	7.139717	-1.781868	-4.769393	C 3.335530 -13.639386 -1.031506
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Ĉ	0.800954	0.8158/19	0.800111	н	6 937176	5 808013	3 5/18163	C 4 480836 13 269314 0 308022
c	1 071401	0.560572	0.671971	C	0.557170 9.744051	4 400025	6.062964	C = 2.224029 = 15.772926 = 0.102056
C	-1.0/1401	-0.300372	-0.0/18/1	C	6.744931	-4.490053	-0.003804	0.122830
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c	-1.748051	1.795525	-0.131393	c	2.157591	-9.303239	-0.110344	$\begin{array}{cccccccccccccccccccccccccccccccccccc$
C	-2.302295	-1.098566	-0.422375	C	2.880399	-10.396131	-2.008845	H 5.919202 -13.837546 1.179789
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c	1 10(75)	2 002596	0.207624	ц П	2.057710	10.705690	2 670150	C 1 505297 11 245422 10 4490(7
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c	2 758057	2 016720	0.460764	c	1 175861	7 246572	6742405	H = 1.001700 - 12.202575 - 5.445050 H = 1.008425 - 11.426803 - 11.441007
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С	-0.376991	5,175661	1.371264	С	-9.136748	-4.656036	4.848206	C -10.488327 -4.288075 6.872947
Č	1 768762	5 1056/3	0.952423	й	7 702000	1 387871	6 51/28/	C 10.893010 / 987/35 8.021339
c	2 120169	4 1 40 497	0.007902	11 11 1	-7.792990	4.926490	0.514204	C 10.214221 2 200765 ( 040227
C	-2.139108	4.149487	-0.027803	н-і	10.190060	-4.830480	2.974528	C-10.314221 -2.890705 0.949337
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C	3 544744	3 191388	-0 292538	Ċ	-8 589947	4 749095	-4 536448	H -10 008767 -2 348284 6 063936
c	1 1 4 9 0 4 1	5 220515	0.422408	п	6 796075	4 607965	2 202505	C 10.041070 2.022152 0.280076
C	1.146041	-3.330313	0.422498	п	-0.780073	-4.027803	-3.393393	C-10.941979 -2.923133 9.289970
C	1.302056	-6.684543	0.129897	C	-9.774744	-2.675721	-4.875650	H -11.438114 -4.862992 10.093354
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c	5 030517	3 4 20004	2 052670	и 1	0.610327	2 124514	5 204760	C 12 551701 5 811816 5 366506
c	5.959517	-3.420094	-2.952070	п-1 С	6.040006	-2.124514	-5.294700	C 10 (07420 7 1005(0 4 (01405
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С	7 567763	0 521252	0 321775	н	-5 813492	2 717790	-6 292046	H -9.617240 -7.351016 4.594062
c	6.015015	1 722058	0.150062	C	7.006112	6 3 1 0 0 8 0	5 678024	C = 12,056224, 8,005867, 4,420055
č	0.913013	1.733936	0.150002	C H	-7.090112	0.310980	-3.078024	C -12.930324 -8.003807 4.420033
C	5./2/8/0	4.182679	-0.854831	Н	-6.434511	6.0618/9	-3.662121	H -14.50418/ -6.6549/9 5.0/8626
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С	3.073949	4.528875	-0.115191	Н	-7.447054	7.323192	-5.505185	C -10.265009 -5.773668 -6.763465
Ĉ	1 0/65/0	1 882/33	0.778682	N	2 802143	0 572071	7 152307	C 11.005216 6.063853 6.848545
c	1.240249	T.002733	1.0106002	T N	0 110017	0.012(70	2 0050/1	C = 0.125226 = 5.14757 = 5.00010
C	2.294230	5./54286	1.812689	IN	8.11991/	9.912679	-3.085861	C -9.135336 -5.614767 -7.582312
С	1.348527	6.289801	2.701071	Ν	12.913653	0.240244	2.398366	C-10.625466 -7.965645 -7.739778
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Ć	-4.017942	6.065346	0.969333	N	0.358199	-8.566351	8.644883	Н -8.562415 -4.694532 -7.527660
$\tilde{c}$	1 252160	5 167402	0.057200	N 1	0.361074	4 070260	5 650254	C = 0.405008 = 7.004050 = 7.004009
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C	-3.452169	4.23/431	-0.5/5548	N -1	0.656207	-4./43971	-5.863059	н -11.210145 -8.880442 -7.792214
С	-3.986730	3.455592	-1.726113	Ν	-7.845800	6.459425	-8.011564	Н -7.875393 -6.489840 -9.083032
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С	-5.411528	3.519229	-3.719663	С	9.179792	10.488726	-2.331660	C -12.031357 -4.403041 -5.735441
Ċ	-5.614859	2.149072	-3.582827	Ċ	9,038670	10 703766	-0.951172	C -12.593441 -4.164575 -4.470848
$\tilde{c}$	6 202600	0.676042	2 06/255	č	10 202102	10 954004	2 057050	C = 12.000 + 11 - 1.000 + 0.0000 + 0.0000 + 0.0000 + 0.0000 + 0.0000 + 0.000000 + 0.000000 + 0.000000 + 0.000000 + 0.000000 + 0.000000 + 0.000000 + 0.000000 + 0.0000000 + 0.0000000 + 0.0000000 + 0.0000000 + 0.0000000 + 0.0000000 + 0.0000000 + 0.0000000 + 0.0000000 + 0.0000000 + 0.0000000 + 0.00000000
C	-0.303009	-0.070943	-3.004233	U	10.302180	10.034294	-2.931939	C-12.04044/ -4.3U2//4 -0.8/3U31

~	6 600 450	1 0 - 0 1 1 1		a	10.000011		0.01/001	~	12 020 5 52	2	4 9 5 9 5 9 4
C	-6.688459	-1.9/9141	-2./65112	C	10.082814	11.262195	-0.216321	C -	13.939553	-3.823850	-4.352704
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c	4 740140	2 605002	0.091052	11	10 500426	10 695604	4.022508	П	12 410545	4 470722	7 95 4007
C	-4./49140	-2.003902	0.409898	н	10.500420	10.085004	-4.025598	н -	12.419545	-4.4/9/33	-7.834227
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С	-5 705380	-3 671816	$2\ 407044$	н	9 956020	11 421506	0.851366	Н-	14 357605	-3 642752	-3 365861
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11	2 022202	6 572026	1 062459	U U	7.055018	12 501740	2 7 2 9 5 5 6	л П	0.265997	6 222710	10 262210
н	3.933303	-0.3/3830	-1.903438	н	7.955018	12.301740	-3.728330	н	-9.203887	0.222/19	-10.203219
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н	6 761 574	4 010159	-1 132118	н	6 967640	9 452783	-7 514220	н	-4 362205	6 264872	-11 007759
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н	-4 517990	5 239527	-2 768038	н	11 462145	-0 680435	4 448545	н	-8 251456	8 910237	-8 997819
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Н	-4.320888	-4.735543	3.621203	Н	15.335312	2.086926	5.842824	Н -	11.893907	7.344904	-6.017123
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C	6 365404	6 675193	1 636760	С	1/ 089390	0 510033	0.260126	C	8 61/1137	8 426157	1 935222
c	4 6 4 4 9 5 9	0.075175	1.000024	c	14.000050	1.245067	1.100(20	c	11 221 221	0.420157	4.055222
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	0.024011	0.007047	2.517700	11	15.005011	-1.554270	1.447020	11	0.2011(4	7.751405	4.495020
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С	1 401518	7 905362	6 137619	С	10 689183	-3 186879	-10 267555	C	-7 960330	13 505454	3 263154
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ц	0 120342	7 8/1709	1 70/655	ц	12 882/126	6 808001	1 158004	ц	1 870072	0 07/37/	11 6160/5
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č	10 (04000	1 200 ( 00	0.407420		1.001255	10.200077	2 770000		1 1540005	11 200750	7 644505
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ŭ	0 557264	1 2/11/00	0 250075		0.400540	12 102721	2 222075	11	1 045 414	10 755797	0 251227
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С	6.903029	-3.790482	-4.014610	Н	-0.289480 -12.967791	-5.416313	Н	5.749803	13.073681	7.995598
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С	7.313708	-5.121135	-4.206619	Н	1.074886 -14.637033	-6.664745	Н	4.598643	14.548477	6.350511
С	8.350954	-3.154994	-5.874291							

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# 6. <sup>1</sup>H NMR and <sup>13</sup>C NMR Spectra

<sup>1</sup>H NMR spectrum of **2** (600 MHz, C<sub>2</sub>D<sub>2</sub>Cl<sub>4</sub>, 140 °C)







<sup>1</sup>H NMR spectrum of **3a** (600 MHz,  $C_2D_2Cl_4$ , 140 °C)



<sup>13</sup>C NMR spectrum of **3a** (150 MHz, C<sub>2</sub>D<sub>2</sub>Cl<sub>4</sub>, 85 °C)



<sup>19</sup>F NMR spectrum of **3a** (565 MHz, C<sub>2</sub>D<sub>2</sub>Cl<sub>4</sub>, 140 °C)



# MALDI-TOF mass spectrum of 3a



<sup>1</sup>H NMR spectrum of **3b** (600 MHz, C<sub>2</sub>D<sub>2</sub>Cl<sub>4</sub>, 140 °C)



 $^{13}\text{C}$  NMR spectrum of **3b** (150 MHz, C<sub>2</sub>D<sub>2</sub>Cl<sub>4</sub>, 85 °C)



## MALDI-TOF mass spectrum of **3b**

