Supplementary Information

for

# Regio- and Diastereoselective Reaction of Chiral Secondary Alkylcopper Reagents with Propargylic Phosphates: Preparation of Chiral Allenes 

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## General

All reactions were carried out with magnetic stirring and under argon atmosphere in glassware dried with a heat gun. Syringes which were used to transfer anhydrous solvents or reagents were purged with argon three times prior to use. Unless otherwise indicated, yields as stated are isolated yields of compounds and are estimated to be $>95 \%$ pure as determined by ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(25^{\circ} \mathrm{C}\right)$ and capillary gas chromatography. The ratio of diastereoisomers was determined by ${ }^{1} \mathrm{H}-\mathrm{NMR},{ }^{13} \mathrm{C}-\mathrm{NMR}$ spectroscopy or capillary GC.

### 1.1 Solvents

All solvents were dried according to standard methods by distillation over drying agents as stated below and were stored under argon atmosphere. Solvents for column chromatography were distilled on a vacuum evaporator prior to use.
$\mathbf{C H}_{\mathbf{2}} \mathbf{C l}_{2}$ was predried over $\mathrm{CaCl}_{2}$ and distilled from $\mathrm{CaH}_{2}$.
EtOH was treated with phthalic anhydride ( $25 \mathrm{~g} / \mathrm{L}$ ) and sodium, heated to reflux for 6 h and distilled.
Diethyl ether was predried over calcium hydride and dried with the solvent purification system SPS-400-2 from INNOVATIVE TECHNOLOGIES INC.

MeOH was heated to reflux over magnesium methoxide and distilled.
THF was continuously refluxed and freshly distilled from sodium benzophenone ketyl under nitrogen.

### 1.2 Chromatography

Gas chromatography was performed with machines of Agilent Technologies 7890, using a column of type HP 5 (Agilent 5\% phenylmethylpolysiloxane; length: 15 m ; diameter: 0.25 mm ; film thickness: $0.25 \mu \mathrm{~m}$ ) or Hewlett-Packard 6890 or 5890 series II, using a column of type HP 5 (Hewlett-Packard, $5 \%$ phenylmethylpolysiloxane; length: 15 m ; diameter: 0.25 mm ; film thickness: $0.25 \mu \mathrm{~m}$ ).

Chiral gas chromatography (GC) was performed on the following column: Chirasil-Dex CB, Varian, CP7502 ( $25.0 \mathrm{mx} 250 \mu \mathrm{~m} \times 0.25 \mu \mathrm{~m}$ ), Average velocity $20, \mathrm{H}_{2}$-flux.

Flash column chromatography was performed using $\mathrm{SiO}_{2}(0.040-0.063 \mathrm{~mm}, 230-400$ mesh ASTM) from Merck if not specially indicated.
Thin layer chromatography (TLC) was performed using $\mathrm{SiO}_{2}$ pre-coated aluminium plates (Merck 60 , F-254). The chromatograms were examined by 254 nm UV irradiation.

### 1.3 Reagents

All reagents were obtained from commercial sources and used without further purification unless otherwise stated. Liquid reagents were distilled prior to use.
The concentration of $t$-BuLi (ca. 2M in $n$-pentane) was determined by titration with dry 2-propanol and 1,10-phenanthroline as indicator in THF.

### 1.4 Analytic data

${ }^{1} \mathbf{H}$-NMR and ${ }^{13}$ C-NMR spectra were recorded on VARIAN Mercury 200, BRUKER ARX 300, VARIAN VXR 300 S and Bruker AMX 600 instruments. Chemical shifts are reported as $\delta$-values in ppm relative to the solvent peak in $\mathrm{CDCl}_{3}$ (residual chloroform: $\delta 7.25 \mathrm{ppm}$ for ${ }^{1} \mathrm{H}-\mathrm{NMR}, \delta 77.0 \mathrm{ppm}$
for ${ }^{13} \mathrm{C}-\mathrm{NMR}$ ). Abbreviations for signal coupling are as follows: s (singlet), d (doublet), t (triplet), q (quartet), quint (quintet), m (multiplet) as well as br (broad).

Mass spectroscopy (MS): High resolution (HRMS) and low resolution (LRMS) spectra were recorded on a FINNIGAN MAT 95Q instrument. Electron impact ionization (EI) was conducted with an ionization energy of 70 eV .

Infrared spectra (IR) were recorded from $4500 \mathrm{~cm}^{-1}$ to $650 \mathrm{~cm}^{-1}$ on a PERKIN ELMER Spectrum BX59343 instrument. For detection a SMITHS DETECTION DuraSamplIR II Diamond ATR sensor was used and the absorption bands are reported in wavenumbers. The abbreviations for intensity are as follows: vs (very strong; maximum intensity), s (strong; above $75 \%$ of max. intensity), m (medium; from $50 \%$ to $75 \%$ of max. intensity), w (weak; below $50 \%$ of max. intensity) as well as br (broad).

Optical rotation values were recorded on a Perkin Elmer 241 or Anton Paar MCP 200 polarimeter. The specific rotation is calculated as follows:

$$
[\alpha]_{\lambda}^{\varphi}=\frac{[\alpha] \cdot 100}{c \cdot d}
$$

Thereby, the wavelength $\lambda$ is reported in nm and the measuring temperature $\phi$ in ${ }^{\circ} \mathrm{C} . \alpha$ represents the recorded optical rotation, $c$ the concentration of the analyte in $10 \mathrm{mg} / \mathrm{mL}$ and d the length of the cuvette in dm . Thus, the specific rotation is given in $10^{-1} \cdot \mathrm{deg} \cdot \mathrm{cm}^{2} \cdot \mathrm{~g}^{-1}$.Usage of the sodium D line $(\lambda=589 \mathrm{~nm})$ is indicated by D instead of the wavelength in nm . The respective concentration as well as the solvent is reported at the relevant section of the experimental section.

## 2 Typical procedures

### 2.1 Typical procedure for the phosphorylation of propargylic alcohols (TP1) ${ }^{[1]}$



According to literature, ${ }^{[1]}$ a dry and Ar-flushed Schlenk-flask was charged with a solution of propargylic alcohol ( 1.0 equiv), DMAP ( 0.1 equiv) and pyridine ( 1.1 equiv) in DCM ( 0.5 M ). After cooling to $0^{\circ} \mathrm{C}$, diethyl chlorophosphate ( 1.1 equiv) was added dropwise. The solution was allowed to warm to rt and stirred for 10 h . The reaction mixture was quenched with water and extracted with diethyl ether ( $3 \times 25 \mathrm{~mL}$ ) and purified by flash column chromatography on silica gel with $n$-pentane/diethyl ether (1/2) to afford the desired propargylic phosphate as a colourless oil.

### 2.2 Typical procedure for the preparation of allenes (TP2)



A dry and Ar-flushed Schlenk-tube was cooled to $-100{ }^{\circ} \mathrm{C}$ and charged with a solution of $t$-BuLi ( $0.22 \mathrm{mmol}, 2.2$ equiv) together with a mixture of diethyl ether $(1.00 \mathrm{~mL})$ and $n$-pentane $(1.50 \mathrm{~mL})$. A solution of iodide $4(0.10 \mathrm{mmol}, 1.0$ equiv) in diethyl ether $(0.40 \mathrm{~mL})$ was added dropwise for 1 min . After stirring for 10 sec , a solution of $\mathrm{CuBr} \cdot \mathrm{P}(\mathrm{OEt})_{3}(0.065 \mathrm{~mL}, 3 \mathrm{~m}$ in diethyl ether, 0.20 mmol , 2.0 equiv) was added and the reaction mixture was stirred for 1 min at $-100^{\circ} \mathrm{C}$ to observe the color change from yellow to green. The Schlenk-tube was transferred to a cooling bath $\left(-50^{\circ} \mathrm{C}\right)$ and the solvent was pumped away under high vacuum. After 10 min , precooled THF ( 2 mL ) and then the propargylic electrophile $6(0.30 \mathrm{mmol}, 3.0$ equiv) were added. The reaction mixture was stirred for 1 h at $-50^{\circ} \mathrm{C}$. After quenching the reaction mixture with aq. $\mathrm{NH}_{3}$ solution, the reaction mixture was extracted with diethyl ether ( $3 \times 10 \mathrm{~mL}$ ). The combined organic phases were dried over $\mathrm{MgSO}_{4}$ and the solvents were evaporated. The obtained crude product was purified by column chromatography on silica gel to afford alkenes of type 7.

## 3 Synthesis of starting materials and electrophiles

### 3.1 Alkyl Iodides

 $\left.(S)-4 \mathbf{e}^{[2 d]},(R)-4 \mathbf{e}^{[2 \mathrm{~d}]}\right)$ were prepared according to literature known procedures.

### 3.2 Electrophiles

The propargyl phosphates were prepared according to literature as described in TP1. ${ }^{[1]}$ Propargyl bromide $\mathbf{6 a}$ is commercially available as a solution in toluene (Sigma Aldrich). Propargyl acetate $\mathbf{6 b}$ is also commercially available (Sigma-Aldrich).


The propargyl pentafluorobenzoate $\mathbf{6 c}$ was prepared according to literature from propargyl alcohol ( $560 \mathrm{mg}, 0.59 \mathrm{~mL}, 10 \mathrm{mmol}$ ). ${ }^{[3]}$ The crude product was purified by flash column chromatography on silica gel with $n$-pentane/diethyl ether $=1 / 2$ to afford $\mathbf{6 c}(7.8 \mathrm{mmol}, 1.95 \mathrm{~g}, 78 \%$ yield) as a colourless oil.
${ }^{1} \mathbf{H}-\mathrm{NMR}\left(\mathbf{C D C l}_{3}, \mathbf{4 0 0} \mathbf{~ M H z}\right): \delta[\mathrm{ppm}]=4.97(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.58(\mathrm{t}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H})$.


6d
The propargyl diphenylphosphate $\mathbf{6 d}$ was prepared according to literature from propargyl alcohol ( $560 \mathrm{mg}, 0.59 \mathrm{~mL}, 10 \mathrm{mmol}$ ). ${ }^{[4]}$ The crude product was purified by flash column chromatography on silica gel with $n$-pentane/diethyl ether $=1 / 2$ to afford $\mathbf{6 d}(8.5 \mathrm{mmol}, 2.45 \mathrm{~g}, 85 \%$ yield) as a colourless oil.
${ }^{1} \mathbf{H}-\mathrm{NMR}\left(\mathbf{C D C l}_{3}, \mathbf{4 0 0} \mathbf{~ M H z}\right): \delta[\mathrm{ppm}]=7.40-7.29(\mathrm{~m}, 4 \mathrm{H}), 7.28-7.15(\mathrm{~m}, 6 \mathrm{H}), 4.85(\mathrm{dd}, J=10.7,2.5$ $\mathrm{Hz}, 2 \mathrm{H}), 2.60(\mathrm{t}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H})$.

$6 \mathbf{e}$
The propargyl diethylphosphate $\mathbf{6 e}$ was prepared according to TP1 from propargyl alcohol ( 560 mg , $0.59 \mathrm{~mL}, 10 \mathrm{mmol}) .{ }^{[5]}$ The crude product was purified by flash column chromatography on silica gel with $n$-pentane/diethyl ether $=1 / 2$ to afford $\mathbf{6 e}(8.2 \mathrm{mmol}, 1.57 \mathrm{~g}, 82 \%$ yield $)$ as a colourless oil.
${ }^{1} \mathbf{H}-\mathrm{NMR}\left(\mathbf{C D C l}_{3}, \mathbf{4 0 0} \mathbf{~ M H z}\right): \delta[\mathrm{ppm}]=\delta 4.66(\mathrm{dd}, J=10.1,2.5 \mathrm{~Hz}, 2 \mathrm{H}), 4.22-4.01(\mathrm{~m}, 2 \mathrm{H}), 2.56(\mathrm{t}$, $J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 1.35(\mathrm{td}, J=7.1,1.0 \mathrm{~Hz}, 6 \mathrm{H})$.


The propargyl phosphate $\mathbf{6 f}$ was prepared according to TP1 from but-2-yn-1-ol ( $700 \mathrm{mg}, 0.75 \mathrm{~mL}$, 10 mmol ). The crude product was purified by flash column chromatography on silica gel with $n$ pentane/diethyl ether $=1 / 2$ to afford $\mathbf{6 f}(8.9 \mathrm{mmol}, 1.83 \mathrm{~g}, 89 \%$ yield $)$ as a colourless oil.
${ }^{\mathbf{1}} \mathbf{H}-\mathbf{N M R}\left(\mathbf{C D C l}_{3}, 400 \mathrm{MHz}\right): \delta[\mathrm{ppm}]=4.64(\mathrm{q}, J=2.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.62(\mathrm{q}, J=2.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.19-4.07$ (m, 4H), $1.87(\mathrm{t}, J=2.4 \mathrm{~Hz}, 3 \mathrm{H}), 1.35(\mathrm{td}, J=7.1,0.9 \mathrm{~Hz}, 6 \mathrm{H})$.


The propargyl phosphate $\mathbf{6 g}$ was prepared according to TP1 from 3-(trimethylsilyl)prop-2-yn-1-ol $(1.28 \mathrm{~g}, 1.48 \mathrm{~mL}, 10 \mathrm{mmol})$. The crude product was purified by flash column chromatography on silica gel with $n$-pentane/diethyl ether $=1 / 2$ to afford $\mathbf{6 g}(6.2 \mathrm{mmol}, 1.54 \mathrm{~g}, 62 \%$ yield $)$ as a colourless oil.
${ }^{1} \mathbf{H}-\mathrm{NMR}\left(\mathbf{C D C l}_{3}, 400 \mathrm{MHz}\right): \delta[\mathrm{ppm}]=\delta 4.65(\mathrm{~d}, J=10.0 \mathrm{~Hz}, 2 \mathrm{H}), 4.20-4.06(\mathrm{~m}, 4 \mathrm{H}), 1.35(\mathrm{td}, J=$ $7.1,1.0 \mathrm{~Hz}, 6 \mathrm{H}), 0.17(\mathrm{~s}, 9 \mathrm{H})$.


The chiral propargyl phosphate $(R)$ - 6 h was prepared according to TP1 from $(R)$-but-3-yn-2-ol (TCI, mind. er $=99: 1 ; 700 \mathrm{mg}, 0.73 \mathrm{~mL}, 10 \mathrm{mmol})$. The crude product was purified by flash column chromatography on silica gel with $n$-pentane/diethyl ether $=1 / 2$ to afford $(R)-6 h(6.4 \mathrm{mmol}, 1.31 \mathrm{~g}$, $64 \%$ yield) as a colourless oil.
${ }^{1} \mathbf{H}-\mathbf{N M R}\left(\mathbf{C D C l}_{3}, 400 \mathrm{MHz}\right): \delta[\mathrm{ppm}]=5.18-5.00(\mathrm{~m}, 1 \mathrm{H}), 4.20-4.03(\mathrm{~m}, 4 \mathrm{H}), 2.55(\mathrm{~d}, J=2.1 \mathrm{~Hz}$, $1 \mathrm{H}), 1.59(\mathrm{dd}, J=6.7,0.8 \mathrm{~Hz}, 3 \mathrm{H}), 1.35(\mathrm{tdd}, J=7.1,2.9,1.1 \mathrm{~Hz}, 6 \mathrm{H})$.


The chiral propargyl phosphate $(R)-\mathbf{6 i}$ was prepared according to TP1 from $(R)$-oct-3-yn-2-ol (1.26 g, $10 \mathrm{mmol})$. The crude product was purified by flash column chromatography on silica gel with $n$ pentane/diethyl ether $=1 / 2$ to afford $(R)-6 \mathbf{i}(5.8 \mathrm{mmol}, 1.52 \mathrm{~g}, 58 \%$ yield $)$ as a colourless oil.
${ }^{1} \mathbf{H}-\mathrm{NMR}\left(\mathbf{C D C l}_{3}, \mathbf{4 0 0} \mathbf{~ M H z}\right): \delta[\mathrm{ppm}]=5.15-5.03(\mathrm{~m}, 1 \mathrm{H}), 4.18-4.04(\mathrm{~m}, 4 \mathrm{H}), 2.20(\mathrm{td}, J=7.0,2.0$ $\mathrm{Hz}, 2 \mathrm{H}), 1.54(\mathrm{dd}, J=6.5,0.8 \mathrm{~Hz}, 3 \mathrm{H}), 1.50-1.37(\mathrm{~m}, 4 \mathrm{H}), 1.36-1.29(\mathrm{~m}, 6 \mathrm{H}), 0.90(\mathrm{t}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H})$.


The electrophile $\mathbf{6 j}$ was prepared according to TP1 from pent-1-en-4-yn-3-ol ${ }^{[6]}(820 \mathrm{mg}, 10 \mathrm{mmol})$. The crude product was purified by flash column chromatography on silica gel with $n$-pentane/diethyl ether $=1 / 2$ to afford $\mathbf{6 j}(3.5 \mathrm{mmol}, 0.76 \mathrm{~g}, 35 \%$ yield $)$ as a colourless oil.
${ }^{1} \mathbf{H}-\mathbf{N M R}\left(\mathbf{C D C l}_{3}, 400 \mathrm{MHz}\right): \delta[\mathrm{ppm}]=5.97(\mathrm{ddd}, J=17.0,10.2,5.7 \mathrm{~Hz}, 1 \mathrm{H}), 5.65-5.54(\mathrm{~m}, 1 \mathrm{H})$, $5.51-5.43(\mathrm{~m}, 1 \mathrm{H}), 5.35(\mathrm{dt}, J=10.2,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 4.25-3.94(\mathrm{~m}, 4 \mathrm{H}), 2.68(\mathrm{~d}, J=2.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.35$ (qd, $J=7.1,1.1 \mathrm{~Hz}, 6 \mathrm{H}$ ).

## 4 Characterization of allenes



The allene syn-7a was prepared according to TP2 from the alkyl iodide syn-(4-iodopentan-2-yl) benzene (syn-4a, $27.4 \mathrm{mg}, 0.1 \mathrm{mmol}$ ) and electrophile $\mathbf{6 e}(58 \mathrm{mg}, 0.3 \mathrm{mmol}$ ). The crude product was purified by flash column chromatography on silica gel with pentane to afford syn-7a $(0.046 \mathrm{mmol}$, $8.6 \mathrm{mg}, 46 \%$ yield, $\mathrm{dr}=6: 94)$ as a colorless oil.
${ }^{1} \mathbf{H}-\mathrm{NMR}\left(\mathbf{C D C l}_{3}, 400 \mathrm{MHz}\right): \delta[\mathrm{ppm}]=7.32-7.27(\mathrm{~m}, 2 \mathrm{H}), 7.21-7.15(\mathrm{~m}, 3 \mathrm{H}), 5.02(\mathrm{q}, J=6.7 \mathrm{~Hz}$, $1 \mathrm{H}), 4.69$ (dd, $J=6.6,2.7 \mathrm{~Hz}, 2 \mathrm{H}), 2.83(\mathrm{q}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.13-2.00(\mathrm{~m}, 1 \mathrm{H}), 1.75-1.61(\mathrm{~m}, 1 \mathrm{H})$, $1.53-1.42(\mathrm{~m}, 1 \mathrm{H}), 1.23(\mathrm{~d}, J=6.9 \mathrm{~Hz}, 3 \mathrm{H}), 1.01(\mathrm{~d}, J=6.7 \mathrm{~Hz}, 3 \mathrm{H})$.
${ }^{13} \mathbf{C}\left\{{ }^{1} \mathbf{H}\right\}$-NMR ( $\left.\mathbf{C D C l}_{3}, \mathbf{1 0 1} \mathbf{~ M H z}\right): \delta[\mathrm{ppm}]=207.6,147.8,128.5,127.1,126.0,96.2,75.7,45.8,37.5$, 30.7, 22.4, 20.6.

IR (ATR): $\mathbf{v}\left[\mathrm{cm}^{-1}\right]=2959$ (m), 2924 (vs), 2853 (m), 2360 (w), 2341 (w), 1955 (w), 1738 (w), 1493 (w), 1454 (m), 1377 (w), 1243 (w), 867 (w), 842 (m), 762 (m), 700 (s).

MS $\boldsymbol{m} / \boldsymbol{z}$ (\%): 171 (15), 157 (23), 144 (40), 143 (22), 129 (100), 128 (14), 118 (49), 117 (43), 115 (13), 105 (63), 103 (22), $91,(49), 79$ (24), 78 (10), 77 (21).

HRMS (EI) $\boldsymbol{m} / \boldsymbol{z}$ : calcd. for $\mathrm{C}_{13} \mathrm{H}_{15}$ [ $\left.\mathrm{M}^{+}-\mathrm{Me}\right]:$ 171.1174; found 171.1166.


The allene anti-7a was prepared according to TP2 from the alkyl iodide anti-(4-iodopentan-2-yl) benzene (anti-4a, $27.4 \mathrm{mg}, 0.1 \mathrm{mmol}$ ) and electrophile $\mathbf{6 e}(58 \mathrm{mg}, 0.3 \mathrm{mmol})$. The crude product was purified by flash column chromatography on silica gel with pentane to afford anti-7a ( 0.055 mmol , $10.2 \mathrm{mg}, 55 \%$ yield, $\mathrm{dr}=98: 2$ ) as a colorless oil.
${ }^{\mathbf{1}} \mathbf{H}-\mathbf{N M R}\left(\mathbf{C D C l}_{3}, \mathbf{4 0 0} \mathbf{~ M H z}\right): \delta[\mathrm{ppm}]=7.34-7.26(\mathrm{~m}, 2 \mathrm{H}), 7.20-7.17(\mathrm{~m}, 3 \mathrm{H}), 5.04(\mathrm{dt}, J=7.3,6.6$ $\mathrm{Hz}, 1 \mathrm{H}), 4.70-4.66(\mathrm{~m}, 2 \mathrm{H}), 2.92-2.75(\mathrm{~m}, 1 \mathrm{H}), 2.04-1.86(\mathrm{~m}, 1 \mathrm{H}), 1.64-1.53(\mathrm{~m}, 2 \mathrm{H}), 1.23(\mathrm{~d}, J=7.0$ $\mathrm{Hz}, 3 \mathrm{H}), 0.96(\mathrm{~d}, J=6.7 \mathrm{~Hz}, 3 \mathrm{H})$.
${ }^{13} \mathbf{C}\left\{{ }^{\mathbf{1}} \mathbf{H}\right\}-\mathbf{N M R}\left(\mathbf{C D C l}_{3}, \mathbf{1 0 1} \mathbf{~ M H z}\right): \delta[\mathrm{ppm}]=207.7,147.5,128.5,128.5,127.3,126.0,95.8,75.6$, 46.1, 37.8, 31.0, 22.9, 21.4.

IR (ATR): $\tilde{\mathbf{v}}\left[\mathbf{c m}^{-1}\right]=3062(\mathrm{vw}), 3028$ (w), 2960 (m), 2924 (m), 2870 (w), 2846 (w), 2360 (vw), 1956 (w), 1604 (w), 1494 (m), 1452 (m), 1376 (w), 1030 (w), 868 (w), 842 (m), 762 (m), 700 (vs).

MS $\boldsymbol{m} / \boldsymbol{z}$ (\%): 157 (17), 144 (41), 143 (20), 130 (11), 129 (100), 128 (12), 118 (32), 117 (32), 115 (12), 105 (59), 103 (23), 91 (46), 79 (26), 78 (11), 77 (23).

HRMS (EI) $\boldsymbol{m} / \boldsymbol{z}$ : calcd. for $\mathrm{C}_{14} \mathrm{H}_{17}\left[\mathrm{M}^{+}-\mathrm{H}\right]:$ 185.1330; found 185.1323.

anti-7b

The allene anti-7b was prepared according to TP2 from the alkyl iodide anti-(-4-iodopentan-2-yl) benzene (anti-4a, $27.4 \mathrm{mg}, 0.1 \mathrm{mmol}$ ) and electrophile $\mathbf{6 f}(62 \mathrm{mg}, 0.3 \mathrm{mmol})$. The crude product was purified by flash column chromatography on silica gel with pentane to afford anti-7b $(0.065 \mathrm{mmol}, 13.0$ $\mathrm{mg}, 65 \%$ yield, $\mathrm{dr}=97: 3$ ) as a colorless oil.
${ }^{\mathbf{1}} \mathbf{H}-\mathbf{N M R}\left(\mathbf{C D C l}_{3}, \mathbf{4 0 0} \mathbf{~ M H z}\right): \delta[\mathrm{ppm}]=7.32-7.26(\mathrm{~m}, 2 \mathrm{H}), 7.22-7.11(\mathrm{~m}, 3 \mathrm{H}), 4.59(\mathrm{qd}, J=3.2,1.6$ $\mathrm{Hz}, 2 \mathrm{H}), 2.83-2.70(\mathrm{~m}, 1 \mathrm{H}), 1.88-1.78(\mathrm{~m}, 1 \mathrm{H}), 1.74-1.65(\mathrm{~m}, 1 \mathrm{H}), 1.63(\mathrm{t}, J=3.1 \mathrm{~Hz}, 3 \mathrm{H}), 1.54-1.47$ $(\mathrm{m}, 1 \mathrm{H}), 1.22(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}), 0.96(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H})$.
${ }^{13} \mathbf{C}\left\{{ }^{\mathbf{1}} \mathbf{H}\right\}-\mathbf{N M R}\left(\mathbf{C D C l}_{3}, \mathbf{1 0 1} \mathbf{~ M H z}\right): \delta[\mathrm{ppm}]=205.8,147.8,128.4,127.3,126.0,102.5,74.4,44.1$, 37.8, 35.1, 22.8, 20.1, 16.2.

IR (ATR): $\tilde{\mathbf{v}}\left[\mathbf{c m}^{-1}\right]=3062(\mathrm{vw}), 3028$ (w), 2960 (m), 2924 (m), 2870 (w), 2846 (w), 2360 (vw), 1956 (w), 1604 (w), 1494 (m), 1452 (m), 1376 (w), 1054 (w), 1030 (w), 868 (w), 842 (m), 762 (m), 700 (vs).

MS $\boldsymbol{m} / \boldsymbol{z}$ (\%): 185 (13), 171 (18), 158 (52), 157 (26), 144 (11), 143 (100), 129 (14), 128 (15), 118 (68), 117 (52), 115 (14), 105 (59), 103 (22), 91 (40), 79 (35), 78 (11), 77 (26), 67 (24).

HRMS (EI) $\boldsymbol{m} / \boldsymbol{z}$ : calcd. for $\mathrm{C}_{14} \mathrm{H}_{17}\left[\mathrm{M}^{+}-\mathrm{Me}\right]: 185.1330$; found 185.1321.

anti-7d

The allene anti-7d was prepared according to TP2 from the alkyl iodide anti-(3-iodo-2methylbutyl)benzene (anti-4b, $27.4 \mathrm{mg}, 0.1 \mathrm{mmol}$ ) and electrophile $\mathbf{6 e}(58 \mathrm{mg}, 0.3 \mathrm{mmol})$. The crude product was purified by flash column chromatography on silica gel with pentane to afford anti-7d ( $0.058 \mathrm{mmol}, 10.8 \mathrm{mg}, 58 \%$ yield, $\mathrm{dr}=98: 2$ ) as a colorless oil.
${ }^{1} \mathbf{H}-\mathrm{NMR}\left(\mathbf{C D C l}_{3}, \mathbf{4 0 0} \mathbf{M H z}\right): \delta[\mathrm{ppm}]=7.31-7.27(\mathrm{~m}, 2 \mathrm{H}), 7.20-7.13(\mathrm{~m}, 3 \mathrm{H}), 5.12(\mathrm{q}, J=6.8 \mathrm{~Hz}$, $1 \mathrm{H}), 4.75-4.69(\mathrm{~m}, 2 \mathrm{H}), 2.73(\mathrm{dd}, J=13.4,5.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.33(\mathrm{dd}, J=13.3,8.9 \mathrm{~Hz}, 1 \mathrm{H}), 2.29-2.20(\mathrm{~m}$, $1 \mathrm{H}), 1.87-1.74(\mathrm{~m}, 1 \mathrm{H}), 1.03(\mathrm{~d}, J=6.9 \mathrm{~Hz}, 3 \mathrm{H}), 0.82(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H})$.
${ }^{13} \mathbf{C}\left\{{ }^{\mathbf{1}} \mathbf{H}\right\}-\mathrm{NMR}\left(\mathbf{C D C l}_{3}, \mathbf{1 0 1} \mathbf{M H z}\right): \delta[\mathrm{ppm}]=208.6,141.8,129.3,128.3,125.8,93.0,75.3,40.7,40.7$, 36.9, 17.6, 15.5.

IR (ATR): $\boldsymbol{\tilde { v }}\left[\mathbf{c m}^{-1}\right]=3028(\mathrm{w}), 2960(\mathrm{~s}), 2924(\mathrm{~m}), 2870(\mathrm{w}), 2366(\mathrm{vw}), 1956(\mathrm{w}), 1604$ (w), 1494 (m), 1454 (m), 1376 (w), 1112 (w), 1048 (w), 1022 (w), 844 (m), 762 (m), 700 (vs).

MS $\boldsymbol{m} / \boldsymbol{z}$ (\%): 171 (19), 157 (21), 144 (11), 143 (24), 129 (13), 119 (11), 118 (43), 117 (15), 92 (15), 91 (100), 67 (14), 57 (10), 43 (35), 41 (25), 42 (21).

HRMS (EI) $m / z:$ calcd. for $\mathrm{C}_{14} \mathrm{H}_{18}: 186.1409$; found 186.1394 .

syn-7d

The allene syn-7d was prepared according to TP2 from the alkyl iodide syn-(3-iodo-2methylbutyl)benzene (syn-4b, $27.4 \mathrm{mg}, 0.1 \mathrm{mmol}$ ) and electrophile $\mathbf{6 e}(58 \mathrm{mg}, 0.3 \mathrm{mmol})$. The crude product was purified by flash column chromatography on silica gel with pentane to afford syn-7d ( $0.042 \mathrm{mmol}, 9.2 \mathrm{mg}, 42 \%$ yield, $\mathrm{dr}=6: 94$ ) as a colorless oil.
${ }^{1} \mathbf{H}-\mathbf{N M R}\left(\mathbf{C D C l}_{3}, 400 \mathrm{MHz}\right): \delta[\mathrm{ppm}]=7.30-7.25(\mathrm{~m}, 2 \mathrm{H}), 7.20-7.06(\mathrm{~m}, 3 \mathrm{H}), 5.11(\mathrm{q}, J=6.7 \mathrm{~Hz}$, $1 \mathrm{H}), 4.71(\mathrm{dd}, J=6.8,3.1 \mathrm{~Hz}, 2 \mathrm{H}), 2.79(\mathrm{dd}, J=13.4,5.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.38-2.28(\mathrm{~m}, 1 \mathrm{H}), 2.23-2.15(\mathrm{~m}$, $1 \mathrm{H}), 1.82-1.71(\mathrm{~m}, 1 \mathrm{H}), 1.03(\mathrm{dd}, J=6.9,3.8 \mathrm{~Hz}, 3 \mathrm{H}), 0.81(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H})$.
${ }^{13} \mathbf{C}\left\{{ }^{\mathbf{1}} \mathbf{H}\right\}-\mathrm{NMR}\left(\mathbf{C D C l}_{3}, \mathbf{1 0 1} \mathbf{~ M H z}\right): \delta[\mathrm{ppm}]=208.5,142.0,129.5,128.5,128.5,126.0,95.2,75.7$, 41.0, 40.5, 37.3, 15.8, 15.8 .

IR (ATR): $\boldsymbol{\tilde { v }}\left[\mathbf{c m}^{-1}\right]=3029$ (w), 2959 (s), 2921 (m), 2872 (w), 2354 (vw), 1948 (w), 1604 (w), 1494 (m), 1379 (w), 1115 (w), 1047 (w), 1022 (w), 843 (m), 772 (m), 702 (vs).

MS $\boldsymbol{m} / \boldsymbol{z}(\%): 171$ (30), 157 (39), 144 (13), 143 (53), 129 (40), 118 (57), 117 (57), 115 (12), 91 (100).
HRMS (EI) $\boldsymbol{m} / \boldsymbol{z}$ : calcd. for $\mathrm{C}_{14} \mathrm{H}_{17}\left[\mathrm{M}^{+}-\mathrm{H}\right]: 185.1330$; found 185.1322.

syn-7e

The allene syn-7e was prepared according to TP2 from the alkyl iodide syn-(tert-butyl((5-iodohexan-2-yl)oxy)diphenylsilane) benzene (syn- $\mathbf{4 c}, 46.6 \mathrm{mg}, 0.1 \mathrm{mmol}$ ) and electrophile $\mathbf{6 e}(58 \mathrm{mg}, 0.3 \mathrm{mmol})$. The crude product was purified by flash column chromatography on silica gel with diethyl ether/pentane $=1 / 300$ to afford syn-7e $(0.044 \mathrm{mmol}, 16.6 \mathrm{mg}, 44 \%$ yijeld, $\mathrm{dr}=4: 96)$ as a colorless oil.
${ }^{1} \mathbf{H}-\mathbf{N M R}\left(\mathbf{C D C l}_{3}, 400 \mathrm{MHz}\right): \delta[\mathrm{ppm}]=7.71-7.66(\mathrm{~m}, 4 \mathrm{H}), 7.45-7.34(\mathrm{~m}, 6 \mathrm{H}), 4.99(\mathrm{q}, J=6.7 \mathrm{~Hz}$, $1 \mathrm{H}), 4.69-4.61(\mathrm{~m}, 2 \mathrm{H}), 3.83(\mathrm{q}, J=6.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.09-1.95(\mathrm{~m}, 1 \mathrm{H}), 1.55-1.37(\mathrm{~m}, 2 \mathrm{H}), 1.35-1.24(\mathrm{~m}$, $2 \mathrm{H}), 1.08-1.03(\mathrm{~m}, 12 \mathrm{H}), 0.94(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H})$.
${ }^{13} \mathbf{C}\left\{{ }^{\mathbf{1}} \mathbf{H}\right\}$-NMR $\left(\mathbf{C D C l}_{3}, \mathbf{1 0 1} \mathbf{~ M H z}\right): \delta[\mathrm{ppm}]=207.6,136.0,135.0,134.7,129.6,129.5,127.6,127.5$, 96.1, 75.5, 69.9, 37.0, 33.0, 32.7, 27.2, 23.4, 20.4, 19.4.

IR (ATR): $\tilde{\mathbf{v}}\left[\mathbf{c m}^{-1}\right]=3071(\mathrm{w}), 3050(\mathrm{w}), 2963(\mathrm{~m}), 2931(\mathrm{~m}), 2858(\mathrm{~m}), 2362(\mathrm{w}), 2334(\mathrm{vw}), 1956$ (w), 1590 (vw), 1473 (w), 1462 (w), 1428 (m), 1390 (w), 1377 (w), 1362 (w), 1133 (m), 1111 (s), 1058 (m), 1030 (w), 1006 (w), 998 (m), 868 (w), 842 (w), 822 (m), 739 (m), 701 (vs), 687 (w).

MS m/z (\%): 321 (36), 199 (100), 183 (13), 181 (10), 135 (11), 77 (6).
HRMS (EI) $\boldsymbol{m} / \boldsymbol{z}$ : calcd for $\mathrm{C}_{21} \mathrm{H}_{25} \mathrm{OSi}\left[\mathrm{M}^{+}-t\right.$-Bu]: 321.1675; found 321.1661.

anti-7e

The allene anti-7e was prepared according to TP2 from the alkyl iodide anti-(tert-butyl((5-iodohexan-2-yl)oxy)diphenylsilane) benzene (anjti-4c, $46.6 \mathrm{mg}, 0.1 \mathrm{mmol}$ ) and electrophile $\mathbf{6 e}(58 \mathrm{mg}, 0.3 \mathrm{mmol})$. The crude product was purified by flash column chromatography on silica gel with diethyl ether/pentane $=1 / 300$ to afford anti-7e $(0.050 \mathrm{mmol}, 18.9 \mathrm{mg}, 50 \%$ yield, $\mathrm{dr}=95: 5)$ as a colorless oil.
${ }^{1} \mathbf{H}-\mathrm{NMR}\left(\mathbf{C D C l}_{3}, 400 \mathrm{MHz}\right): \delta[\mathrm{ppm}]=7.71-7.63(\mathrm{~m}, 4 \mathrm{H}), 7.45-7.32(\mathrm{~m}, 6 \mathrm{H}), 4.97(\mathrm{q}, J=6.5 \mathrm{~Hz}$, $1 \mathrm{H}), 4.69-4.59(\mathrm{~m}, 2 \mathrm{H}), 3.82(\mathrm{~h}, J=6.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.03(\mathrm{dt}, J=6.7,3.4 \mathrm{~Hz}, 1 \mathrm{H}), 1.56-1.29(\mathrm{~m}, 4 \mathrm{H})$, $1.08-1.02(\mathrm{~m}, 12 \mathrm{H}), 0.94(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H})$.
${ }^{13} \mathbf{C}\left\{{ }^{1} \mathbf{H}\right\}-N M R\left(\mathbf{C D C l}_{3}, \mathbf{1 0 1} \mathbf{~ M H z}\right): \delta[\mathrm{ppm}]=207.6,136.0,135.1,134.7,129.6,129.5,127.6,127.5$, $96.1,96.1,75.5,75.5,69.9,69.8,37.0,33.0,32.7,32.6,27.2,23.4,20.5,20.4,19.4$.

IR (ATR): $\boldsymbol{\tilde { v } [ \mathbf { c m } ^ { - 1 } ] = 3 0 7 1 ( \mathrm { w } ) , 3 0 5 0 ( \mathrm { vw } ) , 2 9 6 2 ( \mathrm { m } ) , 2 9 3 0 ( \mathrm { m } ) , 2 8 5 8 ( \mathrm { m } ) , 2 3 6 0 ( \mathrm { w } ) , 2 3 3 3 ( \mathrm { vw } ) , 1 9 5 5}$ (w), 1590 (vw), 1473 (w), 1462 (w), 1428 (m), 1390 (w), 1377 (w), 1362 (w), 1132 (m), 1110 (s), 1059 (m), 1029 (w), 1006 (w), 998 (m), 868 (w), 842 (w), 822 (m), 739 (m), 701 (vs), 687 (m)

MS $\boldsymbol{m} / \boldsymbol{z}$ (\%): 321 (10), 199 (100), 135 (8), 77 (5).
HRMS (EI) $\boldsymbol{m} / \boldsymbol{z}$ : calcd for $\mathrm{C}_{21} \mathrm{H}_{25} \mathrm{OSi}\left[\mathrm{M}^{+}-t\right.$-Bu]: 321.1675; found 321.1665.

(R)-7f

The chiral allene ( $R$ )-7f was prepared according to TP2 from the alkyl iodide ( $R$ )-(3-iodobutyl) benzene $((R)-\mathbf{4 d}, 26.0 \mathrm{mg}, 0.1 \mathrm{mmol})$ and electrophile $\mathbf{6 e}(58 \mathrm{mg}, 0.3 \mathrm{mmol})$. The crude product was purified by flash column chromatography on silica gel with pentane to afford $(R)-7 f(0.041 \mathrm{mmol}, 7.1 \mathrm{mg}, 41 \%$ yield, er =93:7) as a colorless oil.
$[\alpha]_{\mathrm{D}}{ }^{20}=-20.0\left(\mathrm{c}=0.5, \mathrm{CHCl}_{3}\right)$.
${ }^{1} \mathbf{H}-\mathrm{NMR}\left(\mathbf{C D C l}_{3}, 400 \mathrm{MHz}\right): \delta[\mathrm{ppm}]=7.30-7.26(\mathrm{~m}, 2 \mathrm{H}), 7.21-7.15(\mathrm{~m}, 3 \mathrm{H}), 5.10(\mathrm{q}, J=6.7 \mathrm{~Hz}$, $1 \mathrm{H}), 4.73(\mathrm{dd}, J=6.7,2.7 \mathrm{~Hz}, 2 \mathrm{H}), 2.75-2.56(\mathrm{~m}, 2 \mathrm{H}), 2.27-2.08(\mathrm{~m}, 1 \mathrm{H}), 1.80-1.59(\mathrm{~m}, 2 \mathrm{H}), 1.06(\mathrm{~d}$, $J=6.8 \mathrm{~Hz}, 3 \mathrm{H})$.
${ }^{13} \mathbf{C}\left\{{ }^{1} \mathbf{H}\right\}$-NMR ( $\left.\mathbf{C D C l}_{3}, \mathbf{1 0 1} \mathbf{~ M H z}\right): \delta[\mathrm{ppm}]=207.7,142.8,128.6,128.4,125.8,95.9,75.8,39.0,33.7$, 32.6, 20.7.

IR (ATR): $\mathbf{v}\left[\mathrm{cm}^{-1}\right]=3063(\mathrm{w}), 3027$ (w), 2961 (m), 2924 (m), 2856 (w), 1955 (m), 1604 (w), 1497 (m), 1454 (m), 1376 (w), 1031 (w), 868 (m), 842 (s), 746 (m), 722 (w), 698 (vs).

MS $\boldsymbol{m} / \boldsymbol{z}$ (\%): 157 (13), 144 (11), 143 (36), 129 (45), 128 (13), 115 (12), 104 (63), 103 (11), 91 (100), 79 (25), 78 (16), 77 (20), 67 (13), 65 (22).

HRMS (EI) $\boldsymbol{m} / \boldsymbol{z}$ : calcd for $\mathrm{C}_{13} \mathrm{H}_{16}$ : 172.1252; found 172.1242.

(S)-7f

The chiral allene ( $S$ )-7f was prepared according to TP2 from the alkyl iodide ( $S$ )-(3-iodobutyl) benzene $((S)-\mathbf{4 d}, 26.0 \mathrm{mg}, 0.1 \mathrm{mmol})$ and electrophile $\mathbf{6 e}(58 \mathrm{mg}, 0.3 \mathrm{mmol})$. The crude product was purified by flash column chromatography on silica gel with pentane to afford ( $S$ )-7f ( $0.048 \mathrm{mmol}, 8.3 \mathrm{mg}, 48 \%$ yield, er $=10: 90$ ) as a colorless oil.
$[\alpha]_{\mathrm{D}}{ }^{20}=+22.0\left(\mathrm{c}=0.3, \mathrm{CHCl}_{3}\right)$.
${ }^{1} \mathbf{H}-\mathbf{N M R}\left(\mathbf{C D C l}_{3}, 400 \mathrm{MHz}\right): \delta[\mathrm{ppm}]=7.33-7.24(\mathrm{~m}, 2 \mathrm{H}), 7.22-7.13(\mathrm{~m}, 3 \mathrm{H}), 5.10(\mathrm{q}, J=6.7 \mathrm{~Hz}$, $1 \mathrm{H}), 4.73$ (dd, $J=6.6,2.7 \mathrm{~Hz}, 2 \mathrm{H}), 2.76-2.55(\mathrm{~m}, 2 \mathrm{H}), 2.25-2.10(\mathrm{~m}, 1 \mathrm{H}), 1.75-1.59(\mathrm{~m}, 2 \mathrm{H}), 1.06(\mathrm{~d}$, $J=6.8 \mathrm{~Hz}, 3 \mathrm{H})$.
${ }^{13} \mathbf{C}\left\{{ }^{1} \mathbf{H}\right\}-\mathbf{N M R}\left(\mathbf{C D C l}_{3}, \mathbf{1 0 1} \mathbf{M H z}\right): \delta[\mathrm{ppm}]=207.7,142.8,128.6,128.4,125.8,95.9,75.8,39.0,33.7$, 32.6, 20.7.

IR (ATR): $\tilde{\mathbf{v}}\left[\mathbf{c m}^{-1}\right]=3028$ (w), 2961 (m), 2925 (m), 2855 (w), 1955 (m), 1604 (w), 1497 (w), 1454 (m), 1377 (w), 1031 (w), 907 (s), 868 (m), 843 (m), 731 (vs), 698 (vs).

MS $\boldsymbol{m} / \boldsymbol{z}$ (\% ): 157 (15), 143 (40), 129 (48), 128 (14), 115 (14), 104 (65), 103 (12), 91 (100), 79 (21), 78 (15), 77 (16), 65 (19).

HRMS (EI) $\boldsymbol{m} / \boldsymbol{z}$ : calcd for $\mathrm{C}_{13} \mathrm{H}_{16}$ : 172.1252; found 172.1243.


The chiral allene $(R, S)-7 \mathbf{g}$ was prepared according to $\mathbf{T P 2}$ from the alkyl iodide $(R)$-(3iodobutyl)benzene $((R)-\mathbf{4 d}, 26.0 \mathrm{mg}, 0.1 \mathrm{mmol})$ and electrophile $(R)-\mathbf{6 h}(62 \mathrm{mg}, 0.3 \mathrm{mmol})$. The crude product was purified by flash column chromatography on silica gel with pentane to afford $(R, S)-\mathbf{7 g}$ $(0.043 \mathrm{mmol}, 8.0 \mathrm{mg}, 43 \%$ yield, $\mathrm{dr}=92: 8)$ as a colorless oil.
$[\alpha]_{\mathrm{D}}{ }^{20}=+18.0\left(\mathrm{c}=0.8, \mathrm{CHCl}_{3}\right)$.
${ }^{1} \mathbf{H}-\mathrm{NMR}\left(\mathbf{C D C l}_{3}, 400 \mathrm{MHz}\right): \delta[\mathrm{ppm}]=7.31-7.26(\mathrm{~m}, 2 \mathrm{H}), 7.22-7.13(\mathrm{~m}, 3 \mathrm{H}), 5.17-5.08(\mathrm{~m}, 1 \mathrm{H})$, $5.07-5.01(\mathrm{~m}, 1 \mathrm{H}), 2.74-2.51(\mathrm{~m}, 2 \mathrm{H}), 2.16(\mathrm{dtd}, J=13.5,6.7,2.6 \mathrm{~Hz}, 1 \mathrm{H}), 1.67(\mathrm{dd}, J=6.9,3.3 \mathrm{~Hz}$, $3 \mathrm{H}), 1.65-1.58(\mathrm{~m}, 2 \mathrm{H}), 1.04(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H})$.
${ }^{13} \mathbf{C}\left\{{ }^{1} \mathbf{H}\right\}-\mathbf{N M R}\left(\mathbf{C D C l}_{3}, \mathbf{1 0 1} \mathbf{~ M H z}\right): \delta[\mathrm{ppm}]=203.9,143.0,128.6,128.6,128.4,125.7,96.2,86.6$, 39.0, 33.7, 33.1, 20.9, 14.9.

IR (ATR): $\tilde{\mathbf{v}}\left[\mathrm{cm}^{-1}\right]=3027$ (m), 2960 ( s$), 2925(\mathrm{~s}), 2856(\mathrm{~m}), 1497(\mathrm{~m}), 1455(\mathrm{~m}), 1373(\mathrm{w}), 872(\mathrm{~m})$, 745 (m), 720 (m), 698 (vs).

MS $\boldsymbol{m} / \boldsymbol{z}$ (\%): 171 (16), 158 (10), 157 (51), 143 (43), 129 (58), 115 (13), 104 (24), 91 (100), 82 (70), 79 (25), 77 (15), 67 (79), 65 (11).

HRMS (EI) $\boldsymbol{m} / \boldsymbol{z}$ : calcd for $\mathrm{C}_{14} \mathrm{H}_{19}$ : 186.1409; found 186.1400.

$(S, S)-7 \mathrm{~g}$

The allene ( $S, S$ )-7g was prepared according to TP2 from the alkyl iodide ( $S$ )-(3-iodobutyl)benzene ( $(S)$ $\mathbf{4 d}, 26.0 \mathrm{mg}, 0.1 \mathrm{mmol})$ and electrophile $(R)-\mathbf{6 h}(62 \mathrm{mg}, 0.3 \mathrm{mmol})$. The crude product was purified by flash column chromatography on silica gel with pentane to afford $(S, S)-\mathbf{7 g}(0.049 \mathrm{mmol}, 9.1 \mathrm{mg}, 49 \%$ yield, $\mathrm{dr}=12: 88$ ) as a colorless oil.
$[\alpha]_{\mathrm{D}}{ }^{20}=-16.0\left(\mathrm{c}=0.3, \mathrm{CHCl}_{3}\right)$.
${ }^{1} \mathbf{H}-\mathbf{N M R}\left(\mathbf{C D C l}_{3}, 400 \mathrm{MHz}\right): \delta[\mathrm{ppm}]=7.33-7.26(\mathrm{~m}, 2 \mathrm{H}), 7.22-7.15(\mathrm{~m}, 3 \mathrm{H}), 5.11(\mathrm{td}, J=6.7,2.5$ $\mathrm{Hz}, 1 \mathrm{H}), 5.02(\mathrm{dq}, J=6.4,3.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.75-2.55(\mathrm{~m}, 2 \mathrm{H}), 2.22-2.07(\mathrm{~m}, 1 \mathrm{H}), 1.68(\mathrm{dd}, J=6.9,3.2$ $\mathrm{Hz}, 3 \mathrm{H}), 1.66-1.58(\mathrm{~m}, 2 \mathrm{H}), 1.04(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H})$.
${ }^{13} \mathbf{C}\left\{{ }^{1} \mathbf{H}\right\}-N M R\left(\mathbf{C D C l}_{3}, \mathbf{1 0 1} \mathbf{~ M H z}\right): \delta[\mathrm{ppm}]=203.9,143.0,128.6,128.6,128.4,125.7,96.1,86.5$, 39.0, 33.7, 33.3, 20.8, 15.0.

IR (ATR): $\tilde{\mathbf{v}}\left[\mathbf{c m}^{-1}\right]=3027$ (w), 2959 (m), 2924 (m), 2856 (w), 1604 (w), 1496 (w), 1455 (m), 1372 (w), 1031 (w), 872 (m), 761 (w), 745 (m), 717 (m), 698 (vs).

MS $\boldsymbol{m} / \boldsymbol{z}$ (\%): 157 (26), 143 (23), 129 (35), 104 (20), 91 (100), 82 (70), 79 (28), 77 (21), 67 (83), 65 (17).

HRMS (EI) $\boldsymbol{m} / \boldsymbol{z}$ : calcd for $\mathrm{C}_{14} \mathrm{H}_{18}$ : 186.1409; found 186.1401.


The allene $(R, S)$-7h was prepared according to TP2 from the alkyl iodide ( $R$ )-(3-iodobutyl)benzene $((R)-\mathbf{4 d}, 26.0 \mathrm{mg}, 0.1 \mathrm{mmol})$ and electrophile $(R)-\mathbf{6 i}(79 \mathrm{mg}, 0.3 \mathrm{mmol})$. The crude product was purified by flash column chromatography on silica gel with pentane to afford $(R, S)-7 h(0.059 \mathrm{mmol}, 14.3 \mathrm{mg}$, $59 \%$ yield, $\mathrm{dr}=91: 9$ ) as a colorless oil.
$[\alpha]_{\mathrm{D}}{ }^{20}=+15.2\left(\mathrm{c}=0.9, \mathrm{CHCl}_{3}\right)$.
${ }^{1} \mathbf{H}-\mathrm{NMR}\left(\mathbf{C D C l}_{3}, 400 \mathrm{MHz}\right): \delta[\mathrm{ppm}]=7.31-7.26(\mathrm{~m}, 2 \mathrm{H}), 7.23-7.14(\mathrm{~m}, 3 \mathrm{H}), 5.16-5.02(\mathrm{~m}, 1 \mathrm{H})$, $2.73-2.50(\mathrm{~m}, 2 \mathrm{H}), 2.02-1.88(\mathrm{~m}, 2 \mathrm{H}), 1.80-1.70(\mathrm{~m}, 1 \mathrm{H}), 1.67(\mathrm{~d}, J=6.9 \mathrm{~Hz}, 3 \mathrm{H}), 1.62-1.53(\mathrm{~m}, 1 \mathrm{H})$, $1.47-1.19(\mathrm{~m}, 5 \mathrm{H}), 1.04(\mathrm{~d}, J=6.9 \mathrm{~Hz}, 3 \mathrm{H}), 0.95-0.85(\mathrm{~m}, 3 \mathrm{H})$.
${ }^{13} \mathbf{C}\left\{{ }^{\mathbf{1}} \mathbf{H}\right\}$-NMR ( $\left.\mathbf{C D C l}_{3}, \mathbf{1 0 1} \mathbf{~ M H z}\right): 201.2,143.3,128.5,128.4,125.7,108.4,87.4,37.8,36.6,34.3$, $33.8,30.8,30.2,22.6,22.5,20.3,15.3,14.2$.
 2334 (w), 1604 (w), 1496 (w), 1455 (m), 1372 (w), 1031 (w), 770 (w), 746 (m), 698 (s).

MS $\boldsymbol{m} / \boldsymbol{z}$ (\%): 242 (16), 138 (15), 123 (13), 105 (11), 96 (28), 91 (17), 74 (93), 59 (100), 45 (44), 32 (50).

HRMS (EI) $m / z$ : calcd for $\mathrm{C}_{18} \mathrm{H}_{26}$ : 242.2035; found 242.2030.

$(R, S)-7 \mathbf{i}$

The allene $(R, S)$ - $\mathbf{7 i}$ was prepared according to TP2 from the alkyl iodide ( $R$ )-1-(4-iodopentyl)-4methoxybenzene $((R)-4 \mathbf{e}, 30.4 \mathrm{mg}, 0.1 \mathrm{mmol})$ and electrophile $(R)-6 h(62 \mathrm{mg}, 0.3 \mathrm{mmol})$. The crude product was purified by flash column chromatography on silica gel with pentane/diethyl Ether $=300 / 1$ to afford $(R, S)-7 \mathbf{i}(0.052 \mathrm{mmol}, 11.2 \mathrm{mg}, 52 \%$ yield, $\mathrm{dr}=93: 7)$ as a colorless oil.
$[\alpha]_{\mathrm{D}}{ }^{20}=-6.8\left(\mathrm{c}=0.8, \mathrm{CHCl}_{3}\right)$.
${ }^{1} \mathbf{H}-\mathrm{NMR}\left(\mathbf{C D C l}_{3}, 400 \mathrm{MHz}\right): \delta[\mathrm{ppm}]=7.13-7.05(\mathrm{~m}, 2 \mathrm{H}), 6.89-6.73(\mathrm{~m}, 2 \mathrm{H}), 5.06(\mathrm{pd}, J=6.8,2.4$ $\mathrm{Hz}, 1 \mathrm{H}), 4.97(\mathrm{tt}, J=6.4,3.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.79(\mathrm{~s}, 3 \mathrm{H}), 2.54(\mathrm{t}, J=7.7 \mathrm{~Hz}, 2 \mathrm{H}), 2.13(\mathrm{pd}, J=6.8,2.5 \mathrm{~Hz}$, $1 \mathrm{H}), 1.64(\mathrm{dd}, J=7.0,3.2 \mathrm{~Hz}, 3 \mathrm{H}), 1.64-1.55(\mathrm{~m}, 2 \mathrm{H}), 1.40-1.20(\mathrm{~m}, 2 \mathrm{H}), 0.98(\mathrm{~d}, J=6.7 \mathrm{~Hz}, 3 \mathrm{H})$.
${ }^{13} \mathbf{C}\left\{{ }^{\mathbf{1}} \mathbf{H}\right\}-\mathrm{NMR}\left(\mathbf{C D C l}_{3}, \mathbf{1 0 1} \mathbf{~ M H z}\right): \delta[\mathrm{ppm}]=203.7,157.7,135.0,129.4,113.8,96.4,86.3,55.4,36.8$, 35.2, 33.6, 29.5, 20.7, 14.9.

IR (ATR): $\tilde{\mathbf{v}}\left[\mathrm{cm}^{-1}\right]=2956(\mathrm{~m}), 2922(\mathrm{vs}), 2891(\mathrm{~m}), 2875(\mathrm{~m}), 2851(\mathrm{~s}), 1613(\mathrm{~m}), 1513(\mathrm{vs}), 1465(\mathrm{~m})$, 1442 (m), 1303 (m), 1246 (m), 1178 (m), 1040 (m), 808 (w), 722 (w).

MS $\boldsymbol{m} / \boldsymbol{z}$ (\%): 148 (32), 147 (43), 134 (53), 122 (13), 121 (100), 91 (11), 74 (30), 59 (34), 45 (16), 43 (18).

HRMS (EI) $m / z$ : calcd for $\mathrm{C}_{16} \mathrm{H}_{22} \mathrm{O}: 230.1671$; found 230.1663.

$(S, S)-\mathbf{7 i}$

The allene ( $S, S$ )-7i was prepared according to TP2 from the alkyl iodide ( $S$ )-1-(4-iodopentyl)-4methoxybenzene $((S)-\mathbf{4 e}, 30.4 \mathrm{mg}, 0.1 \mathrm{mmol})$ and electrophile $(R)-6 \mathbf{h}(62 \mathrm{mg}, 0.3 \mathrm{mmol})$. The crude product was purified by flash column chromatography on silica gel with pentane/diethyl ether $=300 / 1$ to afford $(S, S)-7 \mathbf{i}(0.054 \mathrm{mmol}, 12.4 \mathrm{mg}, 54 \%$ yield, $\mathrm{dr}=12: 88)$ as a colorless oil.
$[\alpha]_{D^{20}}=+9.2\left(\mathrm{c}=1.0, \mathrm{CHCl}_{3}\right)$.
${ }^{1} \mathbf{H}-\mathrm{NMR}\left(\mathbf{C D C l}_{3}, 400 \mathrm{MHz}\right): \delta[\mathrm{ppm}]=7.12-7.05(\mathrm{~m}, 2 \mathrm{H}), 6.88-6.77(\mathrm{~m}, 2 \mathrm{H}), 5.10-5.03(\mathrm{~m}, 1 \mathrm{H})$, $5.02-4.89(\mathrm{~m}, 1 \mathrm{H}), 3.79(\mathrm{~s}, 3 \mathrm{H}), 2.54(\mathrm{t}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H}), 2.24-1.96(\mathrm{~m}, 1 \mathrm{H}), 1.64(\mathrm{dd}, J=6.9,3.2 \mathrm{~Hz}$, $3 \mathrm{H}), 1.62-1.57(\mathrm{~m}, 2 \mathrm{H}) 1.43-1.28(\mathrm{~m}, 2 \mathrm{H}), 0.98(\mathrm{~d}, J=6.7 \mathrm{~Hz}, 3 \mathrm{H})$.
${ }^{13} \mathbf{C}\left\{{ }^{1} \mathbf{H}\right\}$-NMR ( $\left.\mathbf{C D C l}_{3}, \mathbf{1 0 1} \mathbf{~ M H z}\right): \delta[\mathrm{ppm}]=203.7,157.7,135.1,129.4,113.8,96.4,86.4,55.4,36.9$, 35.3, 33.4, 29.6, 20.8, 14.9.

IR (ATR): $\mathbf{v}\left[\mathrm{cm}^{-1}\right]=2955(\mathrm{~m}), 2927(\mathrm{~s}), 2892(\mathrm{~m}), 2856(\mathrm{~m}), 2837(\mathrm{w}), 1613(\mathrm{~m}), 1513(\mathrm{vs}), 1463$ (m), 1446 (m), 1442 (m), 1300 (m), 1245 (vs), 1177 (m), 1040 (m), $828(\mathrm{~m}), 807(\mathrm{w}), 724(\mathrm{w})$.

MS $\boldsymbol{m} / \boldsymbol{z}$ (\%): 173 (18), 159 (15), 148 (10), 147 (94), 134 (90), 121 (100), 119 (14), 117 (16), 91 (39), 77 (11).

HRMS (EI) $\boldsymbol{m} / z$ : calcd for $\mathrm{C}_{16} \mathrm{H}_{22} \mathrm{O}: 230.1671$; found 230.1664.


The allene $(R, S)-7 \mathbf{j}$ was prepared according to $\mathbf{T P 2}$ from the alkyl iodide ( $R$ )-1-(4-iodopentyl)-4methoxybenzene $((R)-4 \mathbf{e}, 30.4 \mathrm{mg}, 0.1 \mathrm{mmol})$ and electrophile $(R)-\mathbf{6 i}(79 \mathrm{mg}, 0.3 \mathrm{mmol})$. The crude product was purified by flash column chromatography on silica gel with pentane/diethyl ether $=200 / 1$ to afford $(R, S)-7 \mathbf{j}(0.051 \mathrm{mmol}, 13.9 \mathrm{mg}, 51 \%$ yield, $\mathrm{dr}=92: 8)$ as a colorless oil.
$[\alpha]_{\mathrm{D}}{ }^{20}=-5.0\left(\mathrm{c}=0.4, \mathrm{CHCl}_{3}\right)$.
${ }^{\mathbf{1}} \mathbf{H}-\mathbf{N M R}\left(\mathbf{C D C l}_{3}, \mathbf{4 0 0} \mathbf{~ M H z}\right): \delta[\mathrm{ppm}]=7.09(\mathrm{dd}, J=8.7,2.1 \mathrm{~Hz}, 2 \mathrm{H}), 6.88-6.74(\mathrm{~m}, 2 \mathrm{H}), 5.03(\mathrm{tdd}$, $J=6.8,5.2,3.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.79(\mathrm{~s}, 3 \mathrm{H}), 2.53(\mathrm{dt}, J=8.0,6.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.01-1.92(\mathrm{~m}, 1 \mathrm{H}), 1.90-1.82(\mathrm{~m}$, $2 \mathrm{H}), 1.61(\mathrm{dd}, J=6.8,1.3 \mathrm{~Hz}, 3 \mathrm{H}), 1.47-1.40(\mathrm{~m}, 1 \mathrm{H}), 1.36-1.21(\mathrm{~m}, 7 \mathrm{H}), 0.98(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H})$, 0.92-0.74 (m, 3H).
${ }^{13} \mathbf{C}\left\{{ }^{\mathbf{1}} \mathbf{H}\right\}$-NMR $\left(\mathbf{C D C l}_{3}, \mathbf{1 0 1} \mathbf{~ M H z}\right): \delta[\mathrm{ppm}]=201.3,157.7,135.2,129.4,113.8,108.7,87.1,55.4$, $36.8,35.3,31.6,30.3,30.2,29.5,22.7,22.6,20.2,15.3,14.2,14.2$.

IR (ATR): $\tilde{\mathbf{v}}\left[\mathbf{c m}^{-1}\right]=2956(\mathrm{~m}), 2929$ ( s$), 2870(\mathrm{w}), 2858(\mathrm{~m}), 2835(\mathrm{w}), 2359(\mathrm{~m}), 2342(\mathrm{w}), 1613(\mathrm{w})$, 1513 (vs), 1464 (w), 1442 (w), 1300 (w), 1246 (vs), 1176 (w), 1040 (m), 828 (w), 817 (w), 807 (w).

MS $\boldsymbol{m} / \boldsymbol{z}$ (\%): 148 (66), 147 (100), 134 (65), 121 (78), 91 (19), 81 (18).
HRMS (EI) $\boldsymbol{m} / \boldsymbol{z}$ : calcd for $\mathrm{C}_{20} \mathrm{H}_{30} \mathrm{O}: 286.2297$ : found 286.2292.

Regioselective addition of secondary alkylcopper reagent 1d to allylic and propargylic moiety containing phosphate $\mathbf{6 f}$ :
(3-methylocta-4,5,7-trien-1-yl)benzene $\mathbf{7 k}$ and (E)-(3-methyloct-5-en-7-yn-1-yl)benzene $\mathbf{7 m}$ was prepared according to $\mathbf{T P} 2$ from the alkyl iodide (3-iodobutyl)benzene ( $\mathbf{1 e}, 26.0 \mathrm{mg}, 0.1 \mathrm{mmol}$ ) and electrophile $\mathbf{6 j}$ ( $65 \mathrm{mg}, 0.3 \mathrm{mmol}$ ). The crude product was purified by flash column chromatography on silica gel with pentane to afford $7 \mathbf{k}(0.022 \mathrm{mmol}, 4.4 \mathrm{mg}, 22 \%$ yield $)$ and $7 \mathrm{~m}(0.036 \mathrm{mmol}, 7.1 \mathrm{mg}$, $36 \%$ yield) as colorless oils.


7k
${ }^{1} \mathbf{H}-\mathrm{NMR}\left(\mathbf{C D C l}_{3}, 400 \mathrm{MHz}\right): \delta[\mathrm{ppm}]=7.31-7.26(\mathrm{~m}, 2 \mathrm{H}), 7.21-7.13(\mathrm{~m}, 3 \mathrm{H}), 6.23-6.12(\mathrm{~m}, 1 \mathrm{H})$, $5.90-5.82(\mathrm{~m}, 1 \mathrm{H}), 5.35-5.25(\mathrm{~m}, 1 \mathrm{H}), 5.18(\mathrm{~d}, J=17.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.02-4.93(\mathrm{~m}, 1 \mathrm{H}), 2.78-2.48(\mathrm{~m}$, $2 \mathrm{H}), 2.22(\mathrm{dq}, J=11.2,4.6,3.3 \mathrm{~Hz}, 1 \mathrm{H}), 1.78-1.58(\mathrm{~m}, 2 \mathrm{H}), 1.11-1.02(\mathrm{~m}, 3 \mathrm{H})$.

IR (ATR): $\tilde{\mathbf{v}}\left[\mathbf{c m}^{-1}\right]=2952(\mathrm{~m}), 2922$ (vs), 2853 (s), 2360 (w), 2342 (w), 1457 (m), 1376 (w), 1261 (w), 1099 (w), 1019 (w), 800 (w).

HRMS (EI) $\boldsymbol{m} / \boldsymbol{z}$ : calcd for $\mathrm{C}_{15} \mathrm{H}_{17}\left[\mathrm{M}^{+}-\mathrm{H}\right]: 197.1330$; found 197.1324.

${ }^{1} \mathbf{H}-\mathrm{NMR}\left(\mathbf{C D C l}_{3}, 400 \mathrm{MHz}\right): \delta[\mathrm{ppm}]=7.31-7.26(\mathrm{~m}, 2 \mathrm{H}), 7.22-7.13(\mathrm{~m}, 3 \mathrm{H}), 6.08-5.93(\mathrm{~m}, 1 \mathrm{H})$, $5.57-5.46(\mathrm{~m}, 1 \mathrm{H}), 3.08(\mathrm{~s}, 1 \mathrm{H}), 2.73-2.53(\mathrm{~m}, 2 \mathrm{H}), 2.45-2.34(\mathrm{~m}, 1 \mathrm{H}), 2.32-2.15(\mathrm{~m}, 1 \mathrm{H}), 1.72-1.59$ (m, 2H), 1.53-1.40 (m, 1H), $0.98(\mathrm{~d}, J=6.6 \mathrm{~Hz}, 3 \mathrm{H})$.
${ }^{13} \mathbf{C}\left\{{ }^{\mathbf{1}} \mathbf{H}\right\}-\mathbf{N M R}\left(\mathbf{C D C l}_{3}, \mathbf{1 0 1} \mathbf{~ M H z}\right): \delta[\mathrm{ppm}]=144.9,143.1,128.7,128.7,128.6,126.0,109.4,81.7$, 81.0, 38.8, 37.6, 33.8, 33.0, 19.9.

IR (ATR): $\tilde{\mathbf{v}}\left[\mathrm{cm}^{-1}\right]=3300(\mathrm{~m}), 3026(\mathrm{~m}), 2955(\mathrm{~m}), 2925(\mathrm{~s}), 2871(\mathrm{~m}), 2859(\mathrm{~m}), 2360(\mathrm{vs}), 2341$ (s), 2333 ( s ), 1496 ( w ), 1455 (m), 1378 (w), 746 (m), 698 ( s$), 668$ (m), 656 (m).

HRMS (EI) $\boldsymbol{m} / \boldsymbol{z}$ : calcd for $\mathrm{C}_{15} \mathrm{H}_{17}\left[\mathrm{M}^{+}-\mathrm{H}\right]: 197.1330$; found 197.1325.

## 5 NMR Spectra


(s)



$\overbrace{6 \mathbf{r}}^{\mathrm{OP}(\mathrm{O})(\mathrm{OEt})_{2}}$
(
TMS
(s)




6j






anti-7b




syn-7d





(R)-7f


(S)-7f


$(R, S)-7 \mathbf{g}$


$(S, S)-7 \mathrm{~g}$


$(R, S)-7 \mathbf{h}$




$(R, S)-7 \mathbf{j}$



7k



$7 m$


## 6 Chiral chromatograms for determination of er

The er of $(R)-\mathbf{7 f}$ was determined by chiral HPLC analysis.
HPLC (column: OB-H; 100\% $n$-heptane, $0.2 \mathrm{~mL} / \mathrm{min}$ ): $\mathrm{t}_{\mathrm{R}}(\mathrm{min})=10.7(R), 10.9(S) . \mathrm{er}=93: 7$.

## Racemate:


( $\boldsymbol{R}$ )-Enantiomer:


## <Peak Table>

PDA Ch1 212nm

| Peak\# Ret. Time | Area | Height | Area\% |  |
| ---: | ---: | :--- | ---: | ---: |
| 1 | 9.039 | 5969550 | 387634 | 92.661 |
| 2 | 9.328 | 472824 | 36206 | 7.339 |

## (S)-Enantiomer:

The er of ( $S$ )-7f was determined by chiral HPLC analysis.
HPLC (column: OB-H; $100 \%$ n-heptane, $0.2 \mathrm{~mL} / \mathrm{min}): \mathrm{t}_{\mathrm{R}}(\mathrm{min})=10.7(R), 11.0(S)$. er $=10: 90$.


## <Peak Table>

PDACh2 200 nm

| PDACH2 200nm |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
| Peak\# Ret. Time | Area | Height | Conc. |  |
| 1 | 10.756 | 2720167 | 303673 | 10.136 |
| 2 | 11.055 | 24117079 | 1393325 | 89.864 |
| Total |  | 26837246 | 1696998 |  |

The er of $(R, S)-\mathbf{7} \mathbf{g}$ was determined by chiral GC analysis.
GC (Supelco $\beta$-Dex 120 ), $50^{\circ} \mathrm{C}(5 \mathrm{~min})$, ramp of $5^{\circ} \mathrm{C} / \mathrm{min}$ to $180^{\circ} \mathrm{C}$, hold 20 min ; $\mathrm{t}_{\mathrm{R}}(\mathrm{min})=31.84$ $((R, S)$ - and ( $S, S$ )-enantiomer; major), $31.93((R, R)$-enantiomer; minor), $31.98((S, R)$-enantiomer; minor). er $=99: 1$.

## Racemate:



| Peak \# | $\begin{aligned} & \text { RetTime } \\ & \text { [min] } \end{aligned}$ | Type | Width [min] | $\begin{gathered} \text { Area } \\ {[p A * s]} \end{gathered}$ | Height [pA] | Area \% |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 31.842 | BV | 0.0563 | 63.04992 | 18.25036 | 49.95659 |
| 2 | 31.925 | VV | 0.0364 | 31.25006 | 13.32911 | 24.76049 |
| 3 | 31.983 | VB | 0.0355 | 31.90942 | 13.53003 | 25.28292 |
| Total |  |  |  | 126.20940 | 45.10949 |  |

## ( $\boldsymbol{R}, S$ )-Enantiomer:



## (S,S)-Enantiomer:

The er of ( $S, S$ )-7g was determined by chiral GC analysis.
GC (Supelco $\beta$-Dex 120 ), $50^{\circ} \mathrm{C}(5 \mathrm{~min})$, ramp of $5^{\circ} \mathrm{C} / \mathrm{min}$ to $180^{\circ} \mathrm{C}$, hold 20 min ; $\mathrm{t}_{\mathrm{R}}(\mathrm{min})=31.60$ $((R, S)$ - and ( $S, S$ )-enantiomer; major), $31.98((R, R)$-enantiomer; minor), 31.98 ( $(S, R)$-enantiomer; minor). er = 99:1.


## 7 Computational Calculations

All calculations were conducted with the program package GAUSSIAN16. ${ }^{[7]}$

## Calculation of Gibbs Free Energies:

Optimizations of minimum geometries were performed using the B3LYP functional ${ }^{[8-11]}$ with the 6 $311+\mathrm{G}(\mathrm{d}, \mathrm{p})$ basis set ${ }^{[12]}$ for atoms $\mathrm{C}, \mathrm{H}, \mathrm{O}$, and P , and the LANL2DZ ${ }^{[13]}$ effective core potential for Cu . Solvent effects were accounted for through the Polarizable Continuum Model (PCM) ${ }^{[14-16]}$, using the adequate dielectric constant according to the solution used in the experiment (THF for anti-9 and tetralin for anti-10; anti-8 was calculated in both solutions). Tetralin was used because its dielectric constant corresponds to a $3: 2$ mixture of ether and pentane. The minimum structures were verified by a frequency analysis on the same level of theory by the absence of negative modes. For each geometry, a thermochemical analysis was performed at $-100^{\circ} \mathrm{C},-50^{\circ} \mathrm{C}$ and $-10^{\circ} \mathrm{C}$, according to experiments. The Gibbs free energy $\mathrm{G}_{\text {solv }}$ in solution was extracted from the thermochemistry output for each structure ("Sum of electronic and thermal Free Energies") and is indicated below. To obtain the $\Delta \mathrm{G}_{\text {solv }}$ values used in the main manuscript, they are compared to the Gibbs free energy of anti-8, which was the most stable structure, according to

$$
\begin{equation*}
\Delta \mathrm{G}_{\text {solv }}=\mathrm{G}_{\text {solv }}-\mathrm{G}_{\text {solv }}^{\mathrm{ref}} . \tag{1}
\end{equation*}
$$

## Calculation of transition state barriers:

The transition state $t s-\mathbf{8}$ for the epimerization between $\operatorname{syn} \mathbf{- 8}$ and anti-8 was optimized on the same level of electronic structure theory as above. The PCM was not included in this calculation, since the transition state optimization did not converge using the PCM. Energies for the epimerization are therefore given using the gas phase optimized geometries with single point calculations considering the PCM. The relative energy between syn- $\mathbf{8}$ and anti-8 is changed by $0.5 \mathrm{kcal} / \mathrm{mol}$ when switching off the PCM and the transition state barrier is way higher than this difference (above $50 \mathrm{kcal} / \mathrm{mol}$ ). The statement of the barrier being too high for epimerization at the investigated temperatures holds. The transition state was further verified by displacing the $t s-\mathbf{8}$ structure along this mode's displacement vector in positive and negative direction, and subsequent geometry optimizations leading to syn-8 and anti-8, respectively. For comparison $t s-\mathbf{9}$ and $t s-10$, the transition states for the epimerization of anti-9 and anti-10 to syn-9 and syn-10, respectively, were optimized, as well. In these calculations the PCM was included in the optimization itself. Both barriers are slightly higher (around $4 \mathrm{kcal} / \mathrm{mol}$ ) than the barrier for the epimerization of structure anti-8.

## Calculation of bond energies:

Bond dissociation energies (BDEs) of the $\mathrm{Cu}-\mathrm{C}$ bond at the stereocenter were calculated from the enthalpies of formation of the participating molecular moieties according to

$$
\begin{equation*}
\Delta H^{\circ}(298 \mathrm{~K})=\sum_{\text {products }} \Delta \mathrm{H}_{\mathrm{prod}}^{\circ}(298 \mathrm{~K})-\Delta \mathrm{H}_{\text {reactant }}^{\circ}(298 \mathrm{~K}) . \tag{2}
\end{equation*}
$$

In all species investigated here, $\Delta H_{\text {reactant }}^{\circ}$ is the enthalpy of formation of the complete molecule (AB), and $\Delta H_{p r o d}^{\circ}(298 K)$ are the enthalpies of formation of the two radicals ( $\mathrm{A}+\mathrm{B}$ ) after homolytic bond cleavage. These values can be extracted from the GAUSSIAN thermochemistry output ("Sum of electronic and thermal enthalpies).

To calculate BDEs, all molecules (AB) were optimized at the UB3LYP/6-311+G* level with subsequent frequency analysis. Afterwards, both radicals A and B were optimized separately using the same method to account for relaxation effects and extracting $\Delta H_{p r o d}^{\circ}(298 K)$ from the thermochemistry
output. BDEs were calculated according to Equation 2 and values are given in Table 2, together with the bond length of the bond before cleavage. All BDEs are well above thermally accessible energies at room temperature.

Table 1: Gibbs free energies ( $\Delta \mathrm{G}_{\text {solv }}$ ) of investigated molecular species. $\Delta \mathrm{G}_{\text {solv }}$ is given according to Equation 1, with anti-8 as the reference. $\Delta \mathrm{G}_{\text {solv }}$ of the transition state barriers $t s-\mathbf{8}, t s-\mathbf{9}$ and $t s-\mathbf{1 0}$ is given with the respective anti-8, anti-9 and anti-10 structures as reference.

| Molecule | $\Delta \mathrm{G}_{\text {solv }}\left(+25^{\circ} \mathrm{C}\right)$ <br> $[\mathrm{kcal} / \mathrm{mol}]$ | $\Delta \mathrm{G}_{\text {solv }}\left(-100^{\circ} \mathrm{C}\right)$ <br> $[\mathrm{kcal} / \mathrm{mol}]$ | $\Delta \mathrm{G}_{\text {solv }}\left(-50^{\circ} \mathrm{C}\right)$ <br> $[\mathrm{kcal} / \mathrm{mol}]$ | $\Delta \mathrm{G}_{\text {solv }}\left(-10^{\circ} \mathrm{C}\right)$ <br> $[\mathrm{kcal} / \mathrm{mol}]$ |
| :---: | :---: | :---: | :---: | :---: |
| anti-8 | 0.0 | 0.0 | 0.0 | 0.0 |
| $t s-\mathbf{8}$ | 51.9 | 52.4 | 52.2 | 52.1 |
| syn-8 | 2.9 | 2.7 | 2.8 | 2.8 |
| anti-9 | 4.6 | 5.0 | 4.7 | 4.6 |
| $t s-\mathbf{9}$ | 55.7 | 55.6 | 55.7 | 55.7 |
| anti-10 | 6.8 | 6.8 | 6.8 | 6.8 |
| $t s-\mathbf{1 0}$ | 57.4 | 57.1 | 57.2 | 57.3 |
| no coordination | 11.9 | 16.2 | 14.4 | 13.0 |

Table 2: Bond dissociation energies (BDEs) of the $\mathrm{Cu}-\mathrm{C}$ bond of investigated molecular species. BDE values are calculated according to Equation 2.

| Molecule | Bond type | Bond length <br> $[\mathrm{pm}]$ | $\mathrm{BDE}\left(25^{\circ} \mathrm{C}\right)$ <br> $[\mathrm{kcal} / \mathrm{mol}]$ | $\mathrm{BDE}\left(-50{ }^{\circ} \mathrm{C}\right)$ <br> $[\mathrm{kcal} / \mathrm{mol}]$ |
| :---: | :---: | :---: | :---: | :---: |
| anti-8 | Cu-C | 198.5 | -53.9 | -54.0 |
| anti-9 | Cu-C | 195.8 | -51.3 | -51.3 |
| anti-10 | Cu-C | 195.9 | -50.6 | -50.7 |

anti-8
B3LYP // 6-311+G(d,p) (C,H,P,O) // LANL2DZ (Cu) // PCM (Tetrahydrofuran) // $298.15{ }^{\circ} \mathrm{C}$ Sum of electronic and thermal Free Energies $=-1429.0303$ a.u.

| $\mathbf{C}$ | -3.694868 | -1.176985 | -0.566547 |
| :--- | ---: | ---: | ---: |
| $\mathbf{C}$ | -2.241846 | -0.677807 | -0.284884 |
| $\mathbf{H}$ | -2.219448 | -0.326846 | 0.755344 |
| $\mathbf{H}$ | -2.079298 | 0.215041 | -0.902323 |
| $\mathbf{C}$ | -1.096588 | -1.683810 | -0.512617 |
| $\mathbf{H}$ | -1.115531 | -1.997783 | -1.566222 |
| $\mathbf{C}$ | -1.237517 | -2.946976 | 0.357122 |
| $\mathbf{H}$ | -1.232401 | -2.702230 | 1.426519 |
| $\mathbf{H}$ | -2.166395 | -3.508101 | 0.166994 |
| $\mathbf{H}$ | -0.412278 | -3.646251 | 0.187119 |
| $\mathbf{C}$ | -4.705994 | -0.125020 | -0.142358 |
| $\mathbf{C}$ | -4.875364 | 1.068136 | -0.860232 |
| $\mathbf{C}$ | -5.480498 | -0.305965 | 1.010634 |
| $\mathbf{C}$ | -5.784710 | 2.038806 | -0.443332 |
| $\mathbf{H}$ | -4.292635 | 1.243210 | -1.758215 |
| $\mathbf{C}$ | -6.392061 | 0.661780 | 1.433977 |
| $\mathbf{H}$ | -5.368448 | -1.221348 | 1.583924 |
| $\mathbf{C}$ | -6.548768 | 1.840738 | 0.707307 |
| $\mathbf{H}$ | -5.898280 | 2.951561 | -1.018826 |
| $\mathbf{H}$ | -6.980947 | 0.492331 | 2.329277 |
| $\mathbf{H}$ | -7.257948 | 2.594631 | 1.030714 |
| $\mathbf{C u}$ | 0.674723 | -0.816241 | -0.291439 |
| $\mathbf{C}$ | -3.886492 | -1.614924 | -2.028025 |
| $\mathbf{H}$ | -4.914239 | -1.942492 | -2.209460 |
| $\mathbf{H}$ | -3.666055 | -0.799948 | -2.724736 |
| $\mathbf{H}$ | -3.219018 | -2.444277 | -2.273023 |
| $\mathbf{H}$ | -3.874380 | -2.053251 | 0.065183 |
| $\mathbf{P}$ | 2.735723 | 0.167099 | -0.051706 |
| $\mathbf{O}$ | 2.872084 | 1.380850 | 1.039697 |
| $\mathbf{O}$ | 4.000454 | -0.737531 | 0.423439 |
| $\mathbf{O}$ | 3.291092 | 0.745821 | -1.449521 |
| $\mathbf{C}$ | 4.604782 | 1.358056 | -1.629397 |
| $\mathbf{H}$ | 4.738796 | 2.126430 | -0.864336 |
| $\mathbf{H}$ | 5.360499 | 0.584075 | -1.485996 |
| $\mathbf{C}$ | 4.657000 | 1.942233 | -3.025000 |
| $\mathbf{H}$ | 3.892200 | 2.710635 | -3.157078 |
| $\mathbf{H}$ | 5.636748 | 2.397724 | -3.193868 |
| $\mathbf{H}$ | 4.504739 | 1.165282 | -3.777113 |
| $\mathbf{C}$ | 3.966349 | -1.475951 | 1.674242 |
| $\mathbf{H}$ | 4.027282 | -0.759944 | 2.497384 |
| $\mathbf{H}$ | 3.014741 | -2.012931 | 1.747515 |
| $\mathbf{C}$ | 5.37545 | -2.435666 | 1.686429 |
| $\mathbf{H}$ | 6.082470 | -1.894138 | 1.603729 |
| $\mathbf{H}$ | 5.142780 | -2.995409 | 2.625596 |
| $\mathbf{H}$ | 5.066987 | -3.146552 | 0.860559 |
| $\mathbf{C}$ | 1.819015 | 2.371759 | 1.187539 |
| $\mathbf{H}$ | 1.972981 | 3.148201 | 0.433202 |
| $\mathbf{H}$ | 0.848534 | 1.899508 | 1.002141 |
| $\mathbf{C}$ | 1.890801 | 2.944106 | 2.587989 |
|  |  |  |  |


| $\mathbf{H}$ | 2.867247 | 3.397946 | 2.771701 |
| :--- | :--- | :--- | :--- |
| $\mathbf{H}$ | 1.124671 | 3.714574 | 2.709825 |
| $\mathbf{H}$ | 1.720054 | 2.166689 | 3.335802 |

## anti-8

B3LYP // 6-311+G(d,p) (C,H,P,O) // LANL2DZ (Cu) // $298.15^{\circ} \mathrm{C}$
Sum of electronic and thermal Free Energies $=-1429.0209$ a.u.
$\begin{array}{llll}\text { C } & 3.700825 & -1.169149 & 0.574885\end{array}$
$\begin{array}{llll}\text { C } & 2.249134 & -0.672176 & 0.281392\end{array}$
$\begin{array}{llll}\mathbf{H} & 2.231634 & -0.330893 & -0.762298\end{array}$
$\begin{array}{llll}\mathbf{H} & 2.083736 & 0.226433 & 0.889657\end{array}$
$\begin{array}{lllll}\text { C } & 1.101981 & -1.675696 & 0.511660\end{array}$
$\begin{array}{lllll}\mathbf{H} & 1.119307 & -1.983098 & 1.567518\end{array}$
$\begin{array}{lllll}\text { C } & 1.247637 & -2.945546 & -0.348162\end{array}$
$\begin{array}{lllll}\mathbf{H} & 1.245894 & -2.708513 & -1.419362\end{array}$
H $\quad \begin{array}{llll}1.176511 & -3.504610 & -0.151011\end{array}$
H $\quad 0.421859 \quad-3.643984-0.176907$
$\begin{array}{lllll}\text { C } & 4.714992 & -0.123596 & 0.142045\end{array}$
$\begin{array}{lllll}\text { C } & 4.880536 & 1.079907 & 0.844119\end{array}$
$\begin{array}{lllll}\text { C } & 5.496489 & -0.321432 & -1.003897\end{array}$
$\begin{array}{llll}\text { C } & & 5.793057 & 2.044424 \\ & 0.418661\end{array}$
$\begin{array}{llll}\mathbf{H} & 4.292354 & 1.268729 & 1.735744\end{array}$
$\begin{array}{lllll}\text { C } & 6.411208 & 0.640135 & -1.435679\end{array}$
$\begin{array}{llll}\text { H } & 5.387656 & -1.244687 & -1.565048\end{array}$
$\begin{array}{lllll}\mathbf{C} & 6.564116 & 1.829579 & -0.724740\end{array}$
$\begin{array}{llll}\mathbf{H} & 5.903169 & 2.965447 & 0.981427\end{array}$
$\begin{array}{llll}\mathbf{H} & 7.005083 & 0.457853 & -2.325105\end{array}$
$\begin{array}{lllll}\mathbf{H} & 7.275221 & 2.578702 & -1.054841\end{array}$
$\begin{array}{lllll}\mathbf{C u} & -0.670985 & -0.811290 & 0.272943\end{array}$
$\begin{array}{lllll}\text { C } & 3.886056 & -1.588603 & 2.042682\end{array}$
$\begin{array}{lllll}\mathbf{H} & 4.911638 & -1.919009 & 2.231726\end{array}$
$\begin{array}{llll}\text { H } & 3.667259 & -0.763436 & 2.727792\end{array}$
$\begin{array}{lllll}\mathbf{H} & 3.214363 & -2.411982 & 2.296325\end{array}$
$\begin{array}{lllll}\text { H } & 3.882320 & -2.053284 & -0.045094\end{array}$
$\begin{array}{llll}\mathbf{P} & -2.738642 & 0.160495 & 0.044493\end{array}$
O $\quad-2.899415 \quad 1.383330-1.031135$
O $\quad-4.000161-0.752040-0.424439$
$\begin{array}{llll}\text { O } & -3.281693 & 0.718583 & 1.456513\end{array}$
$\begin{array}{llll}\text { C } & -4.589332 & 1.342632 & 1.653223\end{array}$
$\begin{array}{lllll}\mathbf{H} & -4.721518 & 2.118609 & 0.896111\end{array}$
H $\quad-5.353438 \quad 0.576927 \quad 1.510648$
$\begin{array}{lllll}\text { C } & -4.624529 & 1.916156 & 3.053495\end{array}$
$\begin{array}{llll}\mathbf{H} & -3.851891 & 2.676849 & 3.184924\end{array}$
$\begin{array}{llll}\mathbf{H} & -5.598560 & 2.379792 & 3.232093\end{array}$
$\begin{array}{llll}\text { H } & -4.475678 & 1.132234 & 3.799228\end{array}$
C $\quad-3.970986-1.484087-1.681276$
H $\quad-4.042150 \quad-0.763835 \quad-2.499544$
$\begin{array}{lllll}\text { H } & -3.016947 & -2.014560 & -1.763764\end{array}$
C $\quad-5.136716-2.450057-1.690578$
H $\quad-6.084445-1.914820 \quad-1.598783$
$\begin{array}{lllll}\text { H } & -5.144482 & -3.003422 & -2.633377\end{array}$
$\begin{array}{lllll}\text { H } & -5.056090 & -3.166413 & -0.870246\end{array}$

| $\mathbf{C}$ | -1.860530 | 2.391713 | -1.177732 |
| :--- | :--- | :--- | :--- |
| $\mathbf{H}$ | -2.009443 | 3.148930 | -0.403653 |
| $\mathbf{H}$ | -0.881800 | 1.926517 | -1.022313 |
| $\mathbf{C}$ | -1.967947 | 2.992773 | -2.563511 |
| $\mathbf{H}$ | -2.952149 | 3.441293 | -2.716669 |
| $\mathbf{H}$ | -1.212251 | 3.773694 | -2.683310 |
| $\mathbf{H}$ | -1.804740 | 2.233678 | -3.331624 |

anti-9
B3LYP // 6-311+G(d,p) (C,H,O) // LANL2DZ (Cu) // PCM (Tetrahydrofuran) // $298.15^{\circ} \mathrm{C}$ Sum of electronic and thermal Free Energies $=-856.7173$ a.u.

| $\mathbf{C}$ | 2.533305 | 1.098613 | -0.286072 |
| :--- | ---: | ---: | ---: |
| $\mathbf{C}$ | 1.093242 | 0.530337 | -0.079506 |
| $\mathbf{H}$ | 1.076444 | 0.043066 | 0.904011 |
| $\mathbf{H}$ | 0.951093 | -0.274067 | -0.811898 |
| $\mathbf{C}$ | -0.074674 | 1.532664 | -0.175773 |
| $\mathbf{H}$ | -0.055910 | 1.987991 | -1.176250 |
| $\mathbf{C}$ | 0.034508 | 2.668900 | 0.858148 |
| $\mathbf{H}$ | 0.035270 | 2.282377 | 1.884461 |
| $\mathbf{H}$ | 0.951031 | 3.269586 | 0.745518 |
| $\mathbf{H}$ | -0.807118 | 3.364362 | 0.781157 |
| $\mathbf{C}$ | 3.569691 | 0.023026 | -0.004667 |
| $\mathbf{C}$ | 3.766901 | -1.057988 | -0.876978 |
| $\mathbf{C}$ | 4.341272 | 0.066051 | 1.163909 |
| $\mathbf{C}$ | 4.700464 | -2.053849 | -0.593035 |
| $\mathbf{H}$ | 3.186794 | -1.125454 | -1.791010 |
| $\mathbf{C}$ | 5.276758 | -0.928005 | 1.454767 |
| $\mathbf{H}$ | 4.207939 | 0.892456 | 1.855558 |
| $\mathbf{C}$ | 5.461189 | -1.994127 | 0.575581 |
| $\mathbf{H}$ | 4.834787 | -2.878003 | -1.285710 |
| $\mathbf{H}$ | 5.862096 | -0.866930 | 2.366144 |
| $\mathbf{H}$ | 6.188411 | -2.767650 | 0.796105 |
| $\mathbf{C u}$ | -1.802477 | 0.615031 | -0.087659 |
| $\mathbf{C}$ | 2.715910 | 1.733290 | -1.675004 |
| $\mathbf{H}$ | 3.735852 | 2.106377 | -1.806169 |
| $\mathbf{H}$ | 2.515860 | 1.014172 | -2.475598 |
| $\mathbf{H}$ | 2.030309 | 2.573077 | -1.809777 |
| $\mathbf{H}$ | 2.692494 | 1.885574 | 0.458413 |
| $\mathbf{O}$ | -3.562638 | -0.326899 | -0.017554 |
| $\mathbf{C}$ | -4.553753 | -0.318513 | -1.089100 |
| $\mathbf{C}$ | -4.042071 | -1.105901 | 1.120203 |
| $\mathbf{C}$ | -5.823854 | -0.875326 | -0.454617 |
| $\mathbf{H}$ | -4.648358 | 0.705763 | -1.449385 |
| $\mathbf{H}$ | -4.183626 | -0.955209 | -1.897106 |
| $\mathbf{C}$ | -5.271612 | -1.844862 | 0.601891 |
| $\mathbf{H}$ | -3.232514 | -1.760776 | 1.442336 |
| $\mathbf{H}$ | -4.288873 | -0.409879 | 1.926282 |
| $\mathbf{H}$ | -6.464381 | -1.364667 | -1.189459 |
| $\mathbf{H}$ | -6.395813 | -0.074731 | 0.021777 |
| $\mathbf{H}$ | -4.981707 | -2.791556 | 0.138502 |
| $\mathbf{H}$ | -5.984074 | -2.057448 | 1.399968 |
|  |  |  |  |
| $\mathbf{H}$ |  | 0 |  |

anti-10
B3LYP // 6-311+G(d,p) (C,H,O) // LANL2DZ (Cu) // PCM (Tetralin) // $298.15{ }^{\circ} \mathrm{C}$ Sum of electronic and thermal Free Energies $=-857.9087$ a.u.

| $\mathbf{C}$ | -2.078435 | 0.882238 | 1.248324 |
| :--- | ---: | ---: | ---: |
| $\mathbf{C}$ | -0.534165 | 1.112083 | 1.277487 |
| $\mathbf{H}$ | -0.281744 | 1.719568 | 0.399020 |
| $\mathbf{H}$ | -0.055053 | 0.138968 | 1.113138 |
| $\mathbf{C}$ | 0.049821 | 1.770670 | 2.543054 |
| $\mathbf{H}$ | -0.180244 | 1.129420 | 3.405563 |
| $\mathbf{C}$ | -0.548592 | 3.162520 | 2.817904 |
| $\mathbf{H}$ | -0.366827 | 3.851870 | 1.984711 |
| $\mathbf{H}$ | -1.638032 | 3.139953 | 2.978414 |
| $\mathbf{H}$ | -0.110144 | 3.619148 | 3.710574 |
| $\mathbf{C}$ | -2.509726 | 0.372285 | -0.116949 |
| $\mathbf{C}$ | -2.211592 | -0.929240 | -0.546538 |
| $\mathbf{C}$ | -3.195276 | 1.209959 | -1.005792 |
| $\mathbf{C}$ | -2.587029 | -1.375439 | -1.812541 |
| $\mathbf{H}$ | -1.681017 | -1.605999 | 0.114564 |
| $\mathbf{C}$ | -3.573400 | 0.770031 | -2.274737 |
| $\mathbf{H}$ | -3.438139 | 2.222283 | -0.697067 |
| $\mathbf{C}$ | -3.270662 | -0.527280 | -2.684417 |
| $\mathbf{H}$ | -2.346187 | -2.387971 | -2.119045 |
| $\mathbf{H}$ | -4.106391 | 1.440352 | -2.940766 |
| $\mathbf{H}$ | -3.564551 | -0.874569 | -3.668736 |
| $\mathbf{C u}$ | 2.007033 | 1.813022 | 2.482695 |
| $\mathbf{C}$ | -2.553658 | -0.033359 | 2.388882 |
| $\mathbf{H}$ | -3.633500 | -0.199410 | 2.335973 |
| $\mathbf{H}$ | -2.061845 | -1.010561 | 2.352089 |
| $\mathbf{H}$ | -2.327568 | 0.410676 | 3.361059 |
| $\mathbf{H}$ | -2.563307 | 1.854568 | 1.384179 |
| $\mathbf{O}$ | 4.037921 | 1.895314 | 2.382938 |
| $\mathbf{C}$ | 4.889631 | 0.998956 | 3.134363 |
| $\mathbf{H}$ | 4.914505 | 0.023762 | 2.638500 |
| $\mathbf{H}$ | 5.901454 | 1.418101 | 3.125410 |
| $\mathbf{C}$ | 4.368958 | 0.875869 | 4.552215 |
| $\mathbf{H}$ | 3.354472 | 0.469876 | 4.565129 |
| $\mathbf{H}$ | 5.013706 | 0.197685 | 5.117745 |
| $\mathbf{H}$ | 4.362198 | 1.846561 | 5.052330 |
| $\mathbf{C}$ | 4.625780 | 2.423113 | 1.169242 |
| $\mathbf{H}$ | 3.970210 | 3.244113 | 0.876790 |
| $\mathbf{H}$ | 5.603672 | 2.842876 | 1.428466 |
| $\mathbf{C}$ | 4.740957 | 1.397500 | 0.052417 |
| $\mathbf{H}$ | 3.763589 | 0.969971 | -0.183702 |
| $\mathbf{H}$ | 5.125090 | 1.887684 | -0.846642 |
| $\mathbf{H}$ | 5.426788 | 0.586442 | 0.306995 |
|  |  |  |  |
| $\mathbf{H}$ | -1 |  |  |

## syn-8

B3LYP // 6-311+G(d,p) (C,H,P,O) // LANL2DZ (Cu) // $298.15{ }^{\circ} \mathrm{C}$
Sum of electronic and thermal Free Energies $=-1429.0264$ a.u.

| C | 3.551191 | $-1.134768$ | $-0.455345$ |
| :---: | :---: | :---: | :---: |
| C | 2.260659 | -0.897398 | 0.382029 |
| H | 1.971399 | 0.142613 | 0.192193 |
| H | 2.519691 | -0.930621 | 1.449733 |
| C | 1.079550 | -1.866986 | 0.113759 |
| C | 4.558599 | -0.018187 | -0.224258 |
| C | 5.218545 | 0.132455 | 1.004430 |
| C | 4.838597 | 0.911600 | -1.233393 |
| C | 6.124927 | 1.170690 | 1.213929 |
| H | 5.024177 | -0.568592 | 1.809170 |
| C | 5.745036 | 1.953717 | -1.030389 |
| H | 4.341091 | 0.816242 | -2.193885 |
| C | 6.393198 | 2.087799 | 0.196305 |
| H | 6.623120 | 1.264283 | 2.173122 |
| H | 5.944642 | 2.658090 | $-1.831005$ |
| H | 7.099047 | 2.894823 | 0.358523 |
| Cu | -0.667131 | -0.923035 | 0.092982 |
| C | 4.182480 | $-2.522273$ | $-0.246014$ |
| H | 5.077199 | -2.638661 | $-0.865105$ |
| H | 4.473113 | -2.680940 | 0.796608 |
| H | 3.479220 | -3.311191 | -0.520339 |
| H | 3.251295 | -1.073917 | $-1.509232$ |
| P | -2.691677 | 0.160168 | 0.052654 |
| 0 | -2.752245 | 1.653153 | -0.614725 |
| 0 | -3.938551 | -0.517841 | $-0.741016$ |
| 0 | -3.316782 | 0.294074 | 1.533072 |
| C | -4.620364 | 0.885486 | 1.831696 |
| H | -4.677125 | 1.863008 | 1.347939 |
| H | -5.389950 | 0.237629 | 1.409226 |
| C | -4.745395 | 0.998047 | 3.335852 |
| H | -3.965408 | 1.642855 | 3.746627 |
| H | -5.717794 | 1.429477 | 3.588632 |
| H | -4.672770 | 0.015516 | 3.807204 |
| C | -3.841281 | -0.840038 | $-2.156451$ |
| H | -3.853425 | 0.094396 | $-2.722220$ |
| H | -2.890368 | -1.349519 | $-2.342644$ |
| C | -5.015508 | -1.723762 | -2.520370 |
| H | -5.960884 | -1.213564 | $-2.322693$ |
| H | -4.972630 | -1.967140 | -3.585314 |
| H | -4.992951 | -2.655935 | -1.951904 |
| C | -1.686990 | 2.618678 | -0.390724 |
| H | -1.872376 | 3.114987 | 0.565322 |
| H | -0.729448 | 2.091926 | -0.326343 |
| C | -1.690976 | 3.610171 | $-1.535373$ |
| H | -2.654127 | 4.121099 | $-1.602706$ |
| H | -0.914268 | 4.361842 | $-1.371057$ |
| H | -1.491794 | 3.109732 | -2.485521 |
| C | 1.051570 | -3.056354 | 1.096470 |
| H | 1.991259 | -3.629258 | 1.121884 |
| H | 0.861371 | $-2.723357$ | 2.123359 |


| $\mathbf{H}$ | 0.254456 | -3.764534 | 0.844239 |
| :--- | ---: | ---: | ---: |
| $\mathbf{H}$ | 1.203788 | -2.274137 | -0.901659 |

B3LYP // 6-311+G(d,p) (C,H,P,O) // LANL2DZ (Cu) // $298.15{ }^{\circ} \mathrm{C}$ Sum of electronic and thermal Free Energies $=-1428.9325$ a.u.

| C | -3.883979 | -0.306324 | -0.683391 |
| :---: | :---: | :---: | :---: |
| C | -2.811056 | 0.210288 | 0.335418 |
| H | -3.143236 | -0.115351 | 1.341815 |
| H | -2.859659 | 1.308511 | 0.342777 |
| C | -1.401821 | -0.251919 | 0.043226 |
| H | -0.746045 | 0.657234 | 0.034884 |
| C | -1.110021 | -1.680357 | 0.511591 |
| H | -1.451174 | -1.894728 | 1.544368 |
| H | -1.627882 | -2.402990 | -0.140160 |
| H | -0.054719 | -2.035573 | 0.462338 |
| C | -5.295116 | -0.068122 | -0.170225 |
| C | -5.801345 | 1.230652 | 0.013998 |
| C | -6.131045 | -1.145689 | 0.161824 |
| C | -7.091994 | 1.442621 | 0.504662 |
| H | -5.180218 | 2.089438 | -0.229781 |
| C | -7.425079 | -0.941840 | 0.652523 |
| H | -5.761341 | -2.161412 | 0.032199 |
| C | -7.912500 | 0.355970 | 0.826190 |
| H | -7.458253 | 2.458426 | 0.636915 |
| H | -8.050409 | -1.797336 | 0.898435 |
| H | -8.917626 | 0.520189 | 1.207101 |
| C | -3.660632 | 0.293948 | $-2.081520$ |
| H | -4.349509 | -0.138038 | -2.818097 |
| H | -3.814101 | 1.381501 | -2.078181 |
| H | -2.629074 | 0.100657 | -2.396498 |
| H | -3.752216 | -1.394278 | -0.767177 |
| Cu | 0.710184 | -0.138902 | 0.410883 |
| P | 2.889231 | 0.045093 | 0.204974 |
| 0 | 3.599942 | -1.377158 | 0.547939 |
| C | 5.032344 | -1.589622 | 0.621014 |
| H | 5.502674 | -1.181037 | -0.281177 |
| H | 5.414049 | -1.049847 | 1.495253 |
| C | 5.282862 | -3.081778 | 0.748234 |
| H | 4.896658 | -3.618248 | $-0.124781$ |
| H | 6.360067 | -3.272393 | 0.824710 |
| H | 4.795342 | -3.480577 | 1.643732 |
| O | 3.577784 | 1.161904 | 1.150746 |
| C | 4.745021 | 1.977880 | 0.854816 |
| H | 5.630554 | 1.332896 | 0.822420 |
| H | 4.612555 | 2.433759 | -0.129722 |
| C | 4.862686 | 3.021990 | 1.950046 |
| H | 5.738476 | 3.654874 | 1.762116 |
| H | 3.972752 | 3.659152 | 1.975351 |
| H | 4.978982 | 2.550698 | 2.931581 |
| 0 | 3.509625 | 0.501956 | -1.243682 |
| C | 2.993057 | -0.066337 | -2.47598 |


| $\mathbf{H}$ | 3.416732 | -1.070386 | -2.597158 |
| :--- | ---: | ---: | ---: |
| $\mathbf{H}$ | 1.901786 | -0.158473 | -2.399937 |
| $\mathbf{C}$ | 3.389506 | 0.844064 | -3.623770 |
| $\mathbf{H}$ | 4.478533 | 0.944032 | -3.688230 |
| $\mathbf{H}$ | 3.024813 | 0.424160 | -4.568660 |
| $\mathbf{H}$ | 2.953548 | 1.840720 | -3.498618 |

Sum of electronic and thermal Free Energies $=-856.6285$ a.u.

| $\mathbf{C}$ | -2.680544 | 0.799771 | -0.689453 |
| :--- | ---: | ---: | ---: |
| $\mathbf{C}$ | -1.436370 | 0.437883 | 0.180210 |
| $\mathbf{H}$ | -1.566580 | -0.615325 | 0.502525 |
| $\mathbf{H}$ | -1.484779 | 1.039904 | 1.096312 |
| $\mathbf{C}$ | -0.100475 | 0.665919 | -0.537962 |
| $\mathbf{H}$ | 0.574586 | 0.901893 | 0.370934 |
| $\mathbf{C}$ | 0.221942 | -0.539823 | -1.444073 |
| $\mathbf{H}$ | 0.123756 | -1.543591 | -0.980131 |
| $\mathbf{H}$ | -0.488463 | -0.537171 | -2.280259 |
| $\mathbf{H}$ | 1.206692 | -0.502079 | -1.937217 |
| $\mathbf{C}$ | -3.947381 | 0.156271 | -0.148216 |
| $\mathbf{C}$ | -4.452634 | 0.484311 | 1.118965 |
| $\mathbf{C}$ | -4.643077 | -0.799269 | -0.899210 |
| $\mathbf{C}$ | -5.608417 | -0.117986 | 1.613626 |
| $\mathbf{H}$ | -3.938420 | 1.219503 | 1.728837 |
| $\mathbf{C}$ | -5.801906 | -1.405621 | -0.410759 |
| $\mathbf{H}$ | -4.271571 | -1.071913 | -1.882392 |
| $\mathbf{C}$ | -6.290437 | -1.066923 | 0.849981 |
| $\mathbf{H}$ | -5.978097 | 0.153662 | 2.596831 |
| $\mathbf{H}$ | -6.321372 | -2.141136 | -1.015920 |
| $\mathbf{H}$ | -7.190307 | -1.534815 | 1.233664 |
| $\mathbf{C}$ | -2.843096 | 2.320280 | -0.849182 |
| $\mathbf{H}$ | -3.660316 | 2.563461 | -1.535144 |
| $\mathbf{H}$ | -3.059511 | 2.797491 | 0.112439 |
| $\mathbf{H}$ | -1.918300 | 2.750584 | -1.238148 |
| $\mathbf{H}$ | -2.515288 | 0.379017 | -1.687395 |
| $\mathbf{C u}$ | 2.055856 | 0.313312 | 0.059551 |
| $\mathbf{O}$ | 3.984370 | 0.037648 | 0.107601 |
| $\mathbf{C}$ | 4.986437 | 1.100365 | -0.015964 |
| $\mathbf{C}$ | 4.629379 | -1.261229 | 0.313922 |
| $\mathbf{C}$ | 6.293970 | 0.451066 | 0.420111 |
| $\mathbf{H}$ | 5.006659 | 1.416953 | -1.061304 |
| $\mathbf{H}$ | 4.667995 | 1.930987 | 0.612945 |
| $\mathbf{C}$ | 6.100310 | -1.015476 | 0.002549 |
| $\mathbf{H}$ | 4.467620 | -1.552996 | 1.354301 |
| $\mathbf{H}$ | 4.146374 | -1.979522 | -0.348024 |
| $\mathbf{H}$ | 6.417037 | 0.528434 | 1.503323 |
| $\mathbf{H}$ | 7.155567 | 0.918045 | -0.058195 |
| $\mathbf{H}$ | 6.748428 | -1.701398 | 0.549086 |
| $\mathbf{H}$ | 6.290929 | -1.138845 | -1.066483 |
|  | - |  |  |
| $\mathbf{H}$ |  |  |  |

## $t s-10$

B3LYP // 6-311+G(d,p) (C,H,O) // LANL2DZ (Cu) // PCM (Tetralin) // $298.15^{\circ} \mathrm{C}$ Sum of electronic and thermal Free Energies $=-857.8173$ a.u.

| $\mathbf{C}$ | 2.493225 | -0.821068 | -0.410167 |
| :--- | ---: | ---: | ---: |
| $\mathbf{C}$ | 1.271576 | -0.153399 | 0.296045 |
| $\mathbf{H}$ | 1.427218 | 0.942848 | 0.239187 |
| $\mathbf{H}$ | 1.324633 | -0.409694 | 1.361417 |
| $\mathbf{C}$ | -0.072843 | -0.589850 | -0.288887 |
| $\mathbf{H}$ | -0.737381 | -0.557745 | 0.655013 |
| $\mathbf{C}$ | -0.411709 | 0.241532 | -1.540165 |
| $\mathbf{H}$ | -0.299062 | 1.342691 | -1.448726 |
| $\mathbf{H}$ | 0.277192 | -0.055151 | -2.341296 |
| $\mathbf{H}$ | -1.407845 | 0.051821 | -1.975836 |
| $\mathbf{C}$ | 3.790934 | -0.104226 | -0.075370 |
| $\mathbf{C}$ | 4.291124 | -0.067120 | 1.234663 |
| $\mathbf{C}$ | 4.521268 | 0.558751 | -1.069071 |
| $\mathbf{C}$ | 5.474826 | 0.603178 | 1.537321 |
| $\mathbf{H}$ | 3.750181 | -0.568509 | 2.030121 |
| $\mathbf{C}$ | 5.708267 | 1.230942 | -0.773744 |
| $\mathbf{H}$ | 4.154045 | 0.547045 | -2.090868 |
| $\mathbf{C}$ | 6.191257 | 1.255956 | 0.533129 |
| $\mathbf{H}$ | 5.839712 | 0.616086 | 2.559081 |
| $\mathbf{H}$ | 6.254198 | 1.733807 | -1.565118 |
| $\mathbf{H}$ | 7.113206 | 1.776529 | 0.767718 |
| $\mathbf{C}$ | 2.569646 | -2.324541 | -0.100523 |
| $\mathbf{H}$ | 3.361659 | -2.811378 | -0.677747 |
| $\mathbf{H}$ | 2.772856 | -2.499869 | 0.961308 |
| $\mathbf{H}$ | 1.611875 | -2.789958 | -0.342336 |
| $\mathbf{H}$ | 2.343135 | -0.714599 | -1.490452 |
| $\mathbf{C u}$ | -2.222963 | -0.142150 | 0.147045 |
| $\mathbf{O}$ | -4.206405 | -0.016547 | 0.055826 |
| $\mathbf{C}$ | -4.846663 | 1.235901 | -0.301629 |
| $\mathbf{H}$ | -5.855879 | 1.218054 | 0.120494 |
| $\mathbf{H}$ | -4.925438 | 1.284383 | -1.393084 |
| $\mathbf{C}$ | -4.990076 | -1.185834 | -0.305000 |
| $\mathbf{H}$ | -5.993190 | -1.044044 | 0.107852 |
| $\mathbf{H}$ | -5.063301 | -1.228004 | -1.397082 |
| $\mathbf{C}$ | -4.338950 | -2.429806 | 0.258913 |
| $\mathbf{H}$ | -4.929607 | -3.304945 | -0.023677 |
| $\mathbf{H}$ | -3.329085 | -2.569951 | -0.138710 |
| $\mathbf{H}$ | -4.288236 | -2.388270 | 1.348957 |
| $\mathbf{C}$ | -4.04603 | 2.395505 | 0.251065 |
| $\mathbf{H}$ | -3.033395 | 2.415617 | -0.161645 |
| $\mathbf{H}$ | -4.535369 | 3.333837 | -0.023846 |
| $\mathbf{H}$ | -3.983416 | 2.347395 | 1.339988 |
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## 8 <br> References

[1] a) H. Ohmiya, U. Yokobori, Y. Makida and M. Sawamura, J. Am. Chem. Soc. 2010, 132, 28952897. b) J. Skotnitzki, L. Spessert and P. Knochel, Angew. Chem., Int. Ed. 2019, 58, 15091514.
[2] a) K. Moriya, D. Didier, M. Simon, J. M. Hammann, G. Berionni, K. Karaghiosoff, H. Zipse, H. Mayr and P. Knochel, Angew. Chem., Int. Ed. 2015, 54, 2754-2757. b) V. Morozova, K. Moriya, P. Mayer and P. Knochel, Chem. Eur. J. 2016, 22, 9962-9965. c) G. Dagousset, K. Moriya, R. Mose, G. Berionni, K. Karaghiosoff and P. Knochel, Angew. Chem., Int. Ed. 2014, 53, 1425-1429. d) V. Morozova, J. Skotnitzki, K. Moriya, K. Karaghiosoff and P. Knochel, Angew. Chem., Int. Ed. 2018, 57, 5516-5519.
[3] N. N. Solodukhin, N. E. Borisova, A. V. Churakov and K. V. Zaitsev, J. Fluor. Chem., 2016, 187, 15-23.
[4] J. Eisenblaetter, M. Bruns, U. Fehrenbacher, L. Barner and C. Barner-Kowollik, Polym. Chem., 2013, 4, 2406-2413.
[5] M. Hojo, R. Sakuragi, S. Okabe and A. Hosomi, Chem. Comm., 2001, 357-358.
[6] A. Czepa and T. Hofmann, J. Agric. Food Chem., 2004, 52, 4508-4514.
[7] M. J. Frisch, et al. Gaussian16 Revision B.01. (2016).
[8] P. J. Stephens, F. J. Devlin, C. F. Chabalowski and M. J. Frisch, J. Phys. Chem., 1994, 98, 11623-11627.
[9] S. H. Vosko, L. Wilk and M. Nusair, Can. J. Phys., 1980, 58, 1200-1211.
[10] C. Lee, W. Yang and R. G. Parr, Phys. Rev. B, 1988, 37, 785-789.
[11] A. D. Becke, J. Chem. Phys., 1993, 98, 5648-5652.
[12] A. D. McLean and G. S. Chandler, J. Chem. Phys., 1980, 72, 5639-5648.
[13] P. J. Hay and W. R. Wadt, J. Chem. Phys., 1985, 82, 270-283.
[14] S. Miertuš, E. Scrocco and J. Tomasi, Chem. Phys., 1981, 55, 117-129.
[15] Continuum Solvation Models in Chemical Physics. (John Wiley \& Sons, Ltd, 2007). doi:10.1002/9780470515235
[16] J. Tomasi, B. Mennucci and R. Cammi, Chem. Rev., 2005, 105, 2999-3094.

