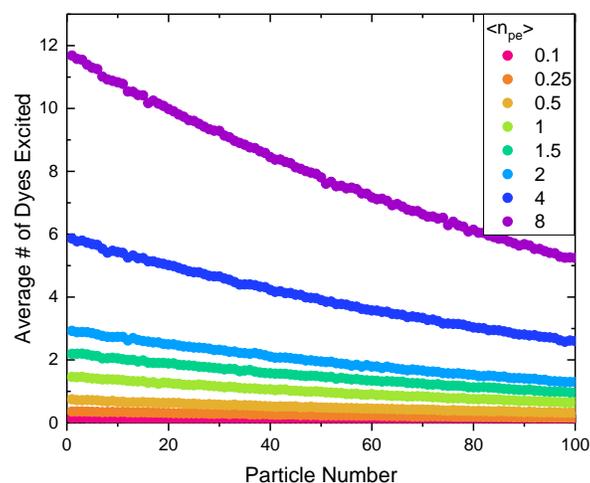


**Supporting Information for**  
**Numerical Monte Carlo Simulations of Charge Transport across the Surface of Dye and**  
**Cocatalyst Modified Spherical Nanoparticles under Conditions of Pulsed or Continuous**  
**Illumination**

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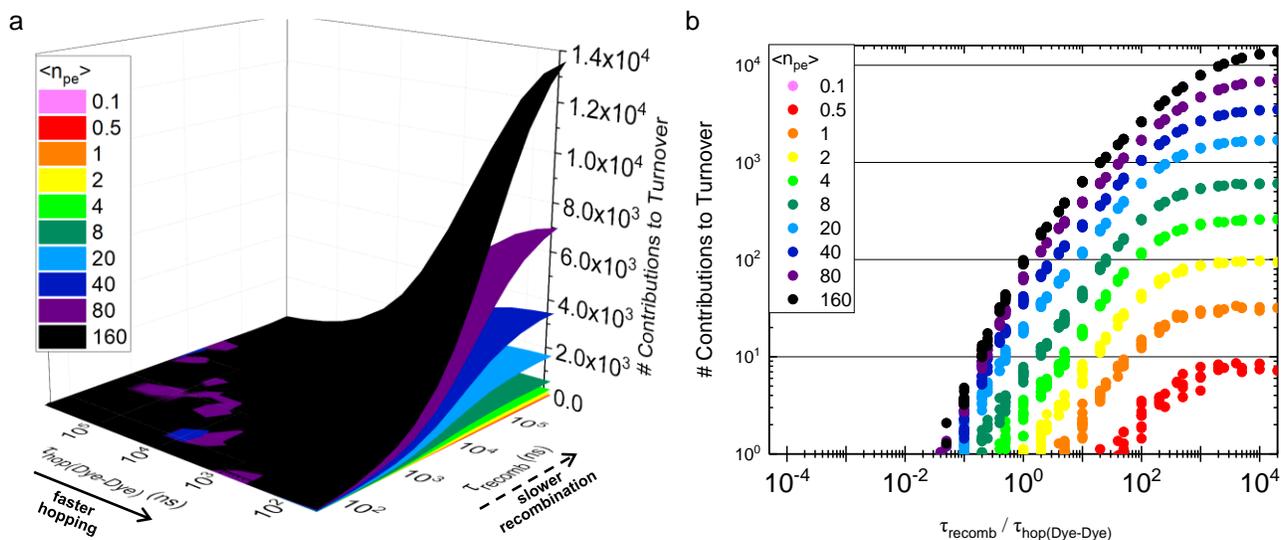
**Figure S1.** Simulated assignment of photoexcited dyes based on the Beer–Lambert law as a function of particle number/depth at the indicated excitation fluences and repeated a total of 50,000 times per condition.

**Table S1.** Values and expressions used for parameters in the Monte Carlo simulations.

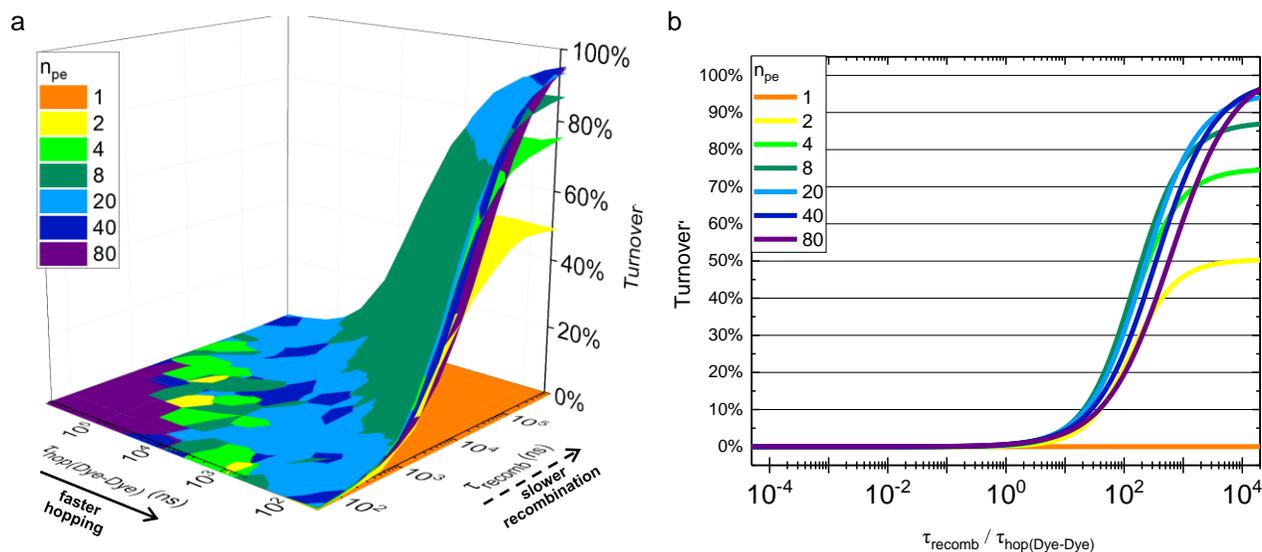
Name	Value(s)	Unit
$\tau_{\text{hop}}(\text{Dye–Dye})$	40, 80, 160, 400, 800, 1600, 4000, 8000, 16000, 40000, 80000, 160000, 400000, 800000	ns
$\tau_{\text{hop}}(\text{Cat–Cat})$	$\tau_{\text{hop–DyetoDye}}$	ns
$\tau_{\text{hop}}(\text{Dye–Cat})$	$\tau_{\text{hop–DyetoDye}} / 27$	ns
$\tau_{\text{hop}}(\text{Cat–Dye})$	$\tau_{\text{hop–DyetoDye}} \times 10^{13}$	ns
$\tau_{\text{recomb}}(\text{SC–Dye})$ per particle	40, 80, 160, 400, 800, 1600, 4000, 8000, 16000, 40000, 80000, 160000, 400000, 800000	ns
$\tau_{\text{recomb}}(\text{SC–Cat})$ per particle	$\tau_{\text{recomb–SCtoDye}}$	ns
time step, $t_{\text{step}}$	Minimum[ $3.75 \times \tau_{\text{hop–DyetoDye}}$ , $\tau_{\text{recomb–SCtoDye}}$ ] / 350	ns
number of trials per data point	25	–
percent of incident light transmitted through the thin film	43.4	%
number of initially excited dyes per stack	10, 50, 100, 200, 400, 800, 2000, 4000, 8000, 16000	–
number of particles in the stack	100	–
number of molecular positions (points) per particle	252	–
percent surface coverage of molecules	100	%
maximum number of points adjacent to each molecule	$6^{\dagger}$	–
maximum redox state of electrocatalysts	1, 2, 4	–
number of electrocatalysts per stack	252	–
number of electrocatalysts per particle <sup>††</sup>	2	–
number of initial photoexcitation events per particle ( $n_{pe}$ ) <sup>††</sup>	1, 2, 4, 8, 20	–

<sup>†</sup> in 12/252 cases, tessellation resulted in points that were pentagonally packed with only 5 adjacent points

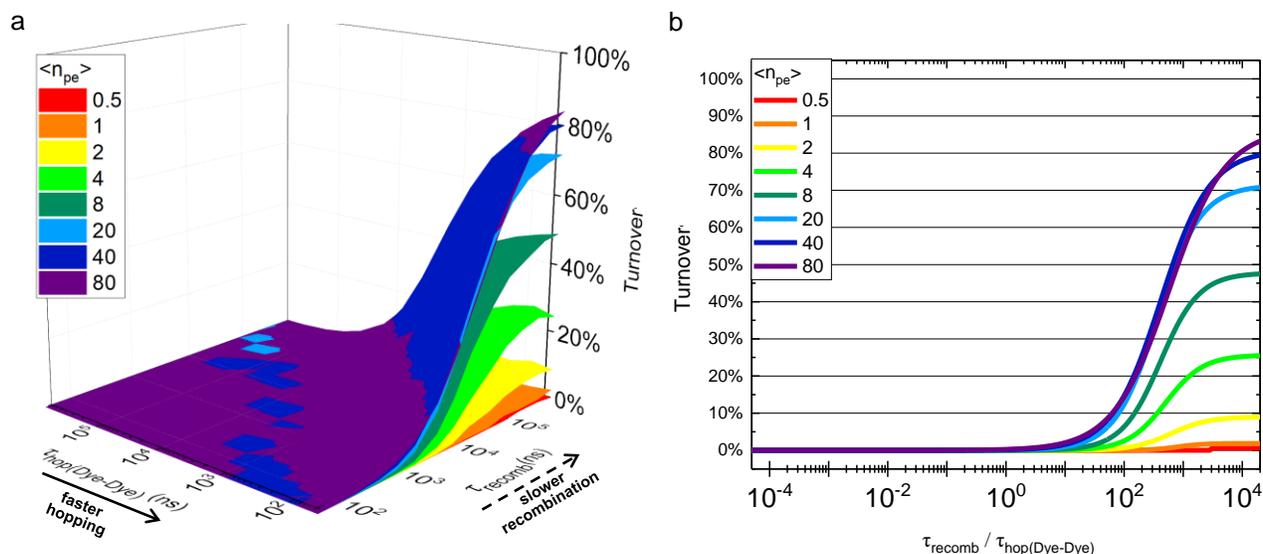
<sup>††</sup> only used when absorption was homogeneous across the stack and did not follow the Beer–Lambert law



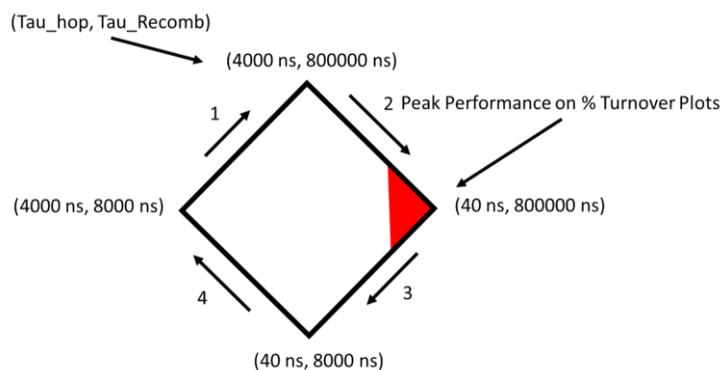
**Figure S2.** (a) Sheet plot representing the number of photoexcited dyes that ultimately *contribute to double oxidation/reduction of an electrocatalyst and turnover* when electrocatalysts are present at 1% surface coverage at the indicated initial pulsed-light excitation fluences. (b) Representation of the data in panel a as a function of the ratio of the recombination time constant to the hopping time constant using base-10 logarithmic scaling of the y-axis values so that lower fluence data can be seen more clearly.



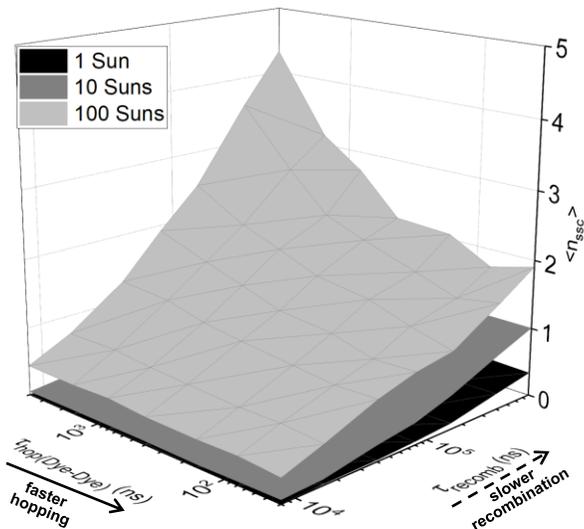
**Figure S3.** (a) Sheet plot representing the percentage of photoexcited dyes that ultimately contribute to double oxidation/reduction of an electrocatalyst and turnover when *electrocatalysts are present at exactly 2 per particle* at the indicated initial pulsed-light excitation fluences as a uniform distribution over the stack. (b) Non-linear least squares sigmoidal best-fits of the data in panel a as a function of the ratio of the recombination time constant to the hopping time constant.



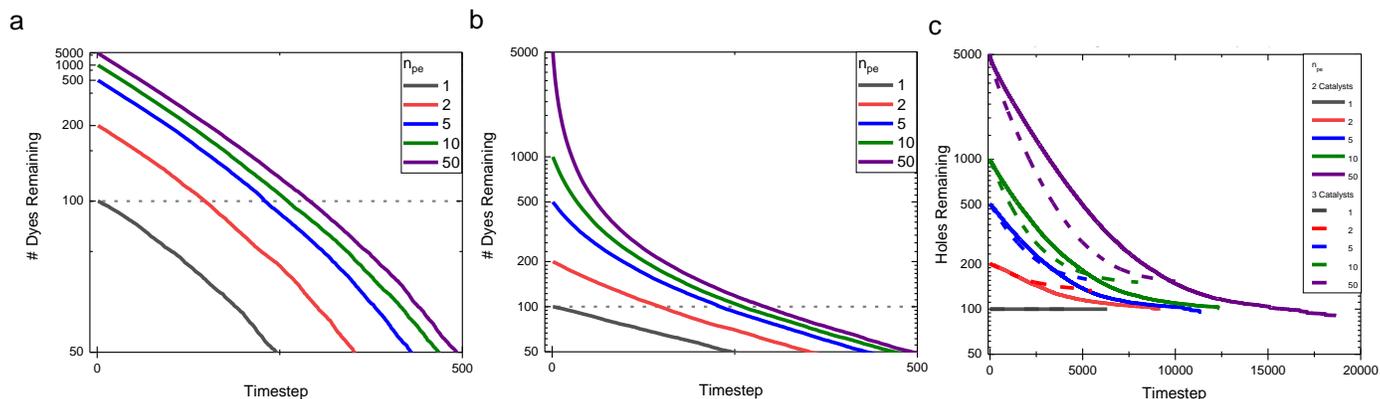
**Figure S4.** (a) Sheet plots representing the percentage of photoexcited dyes that ultimately contribute to quadruple oxidation/reduction of an electrocatalyst and turnover when electrocatalysts are present at 1% surface coverage at the indicated initial pulsed-light excitation fluences. (b) Non-linear least squares sigmoidal best-fits of the data in panel a as a function of the ratio of the recombination time constant to the hopping time constant.



**Figure S5.** Schematic detailing the process used to create a panoramic plot by tracing the perimeter of the parameter space covered by the sheet plot as 1, 2, 3, and 4, to allow for facile two-dimensional viewing for a wide range of parameters.



**Figure S6.** (a) Sheet plots – oriented like all other sheet plots – representing the steady-state number of oxidized/reduced species when electrocatalysts require double oxidation/reduction for turnover and are present at 1% surface coverage at the indicated continuous illumination solar-simulated fluences.



**Figure S7.** (a,b) Number of oxidized/reduced dyes remaining over time on the 100 particle stack after the indicated initial uniform pulsed-light excitation fluences, in the absence of electrocatalysts. (c) Number of oxidized/reduced species remaining over time on the 100 particle stack after the indicated initial uniform pulsed-light excitation fluences at the indicated uniform number of electrocatalysts per particle, in the absence of recombination. The y-axis in panel a is reciprocally scaled so that linear behavior indicates equal-concentration 2<sup>nd</sup>-order kinetic processes, while the y-axes in panels b and c are logarithmically scaled so that linear behavior indicates 1<sup>st</sup>-order kinetic processes. Kinetic parameters from best-fits of these data are shown in Table S2.

**Table S2.** Best-fit rate constants from the linear regions of the data in Figure S7.

	<b>Recombination, # excitations remaining &gt; 100 (Figure S7a)</b>	<b>Recombination, # excitations remaining &lt; 100 (Figure S7b)</b>	<b>Turnover, <i>initial</i> (2 electrocatalysts per particle) (Figure S7c)</b>	<b>Turnover, <i>initial</i> (3 electrocatalysts per particle) (Figure S7c)</b>
<b>kinetics</b>	<b>equal-concentration 2<sup>nd</sup>-order</b>	<b>1<sup>st</sup>-order</b>	<b>1<sup>st</sup>-order</b>	<b>1<sup>st</sup>-order</b>
<b><math>n_{pe} = 1</math></b>	–	$1.23 \times 10^{-3} \text{ timestep}^{-1}$	$0 \text{ timestep}^{-1}$	$0 \text{ timestep}^{-1}$
<b><math>n_{pe} = 2</math></b>	$3.32 \times 10^{-5} \text{ timestep}^{-1}$	$1.43 \times 10^{-3} \text{ timestep}^{-1}$	$5.46 \times 10^{-5} \text{ timestep}^{-1}$	$5.92 \times 10^{-5} \text{ timestep}^{-1}$
<b><math>n_{pe} = 5</math></b>	$3.15 \times 10^{-5} \text{ timestep}^{-1}$	$1.34 \times 10^{-3} \text{ timestep}^{-1}$	$1.52 \times 10^{-4} \text{ timestep}^{-1}$	$1.86 \times 10^{-4} \text{ timestep}^{-1}$
<b><math>n_{pe} = 10</math></b>	$3.17 \times 10^{-5} \text{ timestep}^{-1}$	$1.38 \times 10^{-3} \text{ timestep}^{-1}$	$2.01 \times 10^{-4} \text{ timestep}^{-1}$	$2.80 \times 10^{-4} \text{ timestep}^{-1}$
<b><math>n_{pe} = 50</math></b>	$3.11 \times 10^{-5} \text{ timestep}^{-1}$	$1.47 \times 10^{-3} \text{ timestep}^{-1}$	$2.44 \times 10^{-4} \text{ timestep}^{-1}$	$3.55 \times 10^{-4} \text{ timestep}^{-1}$
<b>mean</b>	$(319 \pm 9) \times 10^{-7} \text{ timestep}^{-1}$	$(137 \pm 9) \times 10^{-5} \text{ timestep}^{-1}$	–	–