## **Supporting Information**

Control loading Au nanoparticles on the surface of hydroxyl pillar[5]arene functionalized single-walled carbon nanotube and its application for catalysis and sensing

Timur Borjigin,<sup>a</sup> Genfu Zhao,<sup>a</sup> Yuhao Zhang,<sup>a</sup> Mengfang Liang,<sup>a</sup> Beihong Liu,<sup>a</sup> Hui Liu,<sup>a</sup> Xiaofei Yang<sup>\*b</sup> and Hong Guo<sup>\*a</sup>

<sup>a</sup>School of Materials Science and Engineering, Green Energy Key Laboratory of All-Solid Ion Battery in Yunnan Province University, Yunnan University, No. 2, Green Lake North Road, Kunming 650091, China.

<sup>b</sup>Department of Mechanical and Materials Engineering, University of Western Ontario, London, Ontario, N6A 5B9, Canada.

## \*Corresponding Author

E-mail: guohongcom@126.com (H. Guo)

**S1. Reagents and methods**: 1,4-Bis(2-hydroxyethoxy)benzene, dichloroethane, boron trifluoride diethyl etherate, carbon tetrabromide, triphenylphosphine, acetonitrile, paraformaldehyde, were reagent grade and used as received. Solvents were either employed as purchased or driedaccording to procedures described in the literature. <sup>1</sup>H NMR and <sup>13</sup>C NMR spectra were recorded on a Bruker Avance DMX-400 spectrometer at 400 MHz and 600 MHz. HP5<sup>S1,2</sup> was synthesized according to the previous papers procedures.





Scheme S1. Synthetic route of HP5.

The <sup>1</sup>H NMR spectrum of **2** is shown in Figure S1. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>, rt)  $\delta$  (ppm): 6.857 (s, 4H), 4.239 (t, J = 4.0 Hz, 4H), 3.608 (t, J = 4 Hz, 4H). The <sup>13</sup>C NMR spectrum of **2** is shown in Figure S2. <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>, rt)  $\delta$  (ppm): 151.93, 114.79, 67.61, 28.21. The <sup>1</sup>H NMR spectrum of **3** is shown in Figure S3. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>, rt)  $\delta$  (ppm): 6.914 (s, 10H), 4.226 (t, J = 5.4 Hz, 20H), 3.844 (s, 10H), 3.632 (t, J = 5.4 Hz, 20H). The <sup>13</sup>C NMR spectrum of **3** is shown in Figure S4. <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>, rt)  $\delta$  (ppm): 149.71, 129.21, 116.26, 69.13, 30.70, 29.64. The <sup>1</sup>H NMR spectrum of **HP5** is shown in Figure S5. <sup>1</sup>H NMR (600 MHz, D<sub>2</sub>O, rt)  $\delta$  (ppm): 6.948 (s, 10H), 4.494 (s, 20H), 3.982-3.894 (m, 50H), 3.584 (s, 20H), 3.246 (s, 60H). The <sup>13</sup>C NMR spectrum of **HP5** is shown in Figure S6. <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>, rt)  $\delta$  (ppm): 149.65, 128.79 115.96, 66.48, 63.66, 62.99, 55.32, 52.26, 29.83.



Figure S2. <sup>13</sup>C NMR spectrum (100 MHz, CDCl<sub>3</sub>, 298 K) of 2.



S4



Figure S6. <sup>13</sup>C NMR spectrum (125 MHz, D<sub>2</sub>O, 298 K) of HP5.



Figure S7. The photographs of SWCNT and HP5@SWCNT complexes in water.



Figure S8. The TEM of Au@HP5 (A) and Au@SWCNT (B), respectively.



Figure S9. High resolution XPS spectra of N 1s for Au@HP5@SWCNT.



Figure S10. Chemical structures of *p*-DNB, *m*-DNB (a), *o*-DNB (b), *p*-nitrophenol (c), *m*-nitrophenol (d), *o*-nitrophenol (e), nitrobenzene (f), *p*-nitrotoluene (g),
hydroquinone (h), *p*-nitroaniline (i), *m*-nitroaniline (j), *o*-nitroaniline (k), respectively.

## Table S1

Sample	Added (µM)	Found (µM)	RSD (%)	Recovery (%)
Tap water	0	-	-	-
	1	$0.98\pm0.01$	1.0	98
	2	$1.97\pm0.11$	5.5	98.5
Waste water	4	$4.01 \pm 0.21$	5.2	100.2
	0	-	-	-
	1	$1.01 \pm 0.04$	3.9	101
	2	$2.12 \pm 0.12$	5.6	106
	10	$9.78\pm0.26$	2.6	97.8

Determination of *p*-DNB in tap water and waste water samples.

## References

S1. R. Joseph, A. Naugolny, M. Feldman, I. M. Herzog, M. Fridman and Y. Cohen, J. Am. Chem. Soc., 2016, **138**, 754–757.

S2. Y. J. Ma, X. F. Ji, F. Xiang, X. D. Chi, C. Y. Han, J. M. He, Z. Abliz, W. X. Chen and F. H. Huang, *Chem. Commun.*, 2011, **47**, 12340–12342.