

## Supplementary Material

# Novel Aqueous Amine Looping Approach for the Direct Capture, Conversion and Storage of CO<sub>2</sub> to Produce Magnesium Carbonate

*Meishen Liu,<sup>1</sup> Hassnain Asgar,<sup>1</sup> Soenke Seifert<sup>2</sup> and Greeshma Gadikota<sup>1,\*</sup>*

<sup>1</sup>School of Civil and Environmental Engineering, Cornell University

527 College Avenue, 117 Hollister Hall, Ithaca, NY 14850

<sup>2</sup>X-Ray Science Division, Building 433A Advanced Photon Source

Argonne National Laboratory, 9700 Cass Avenue, Lemont, IL 60439

The contents of the supporting information are listed below.

**Table S1.** The extents of carbon mineralization of MgO as a function of MEA concentrations and temperatures.

**Figure S1.** Identification of the functional groups present in the solid obtained after reacting MgO with water, 10 wt%, and 20 wt% MEA at 25 °C, 50 °C, 75 °C, and 90 °C with  $P_{CO_2} = 1$  atm for 3 hours and stirring rate of  $300 \text{ rpm} \pm 5 \text{ rpm}$ , using ATR-FTIR measurements.

**Figure S2.** Identification of the functional groups present in the fluid obtained from reacting MgO with water (a), 10 wt% (b), and 20 wt% (c) MEA at 25 °C, 50 °C, 75 °C, and 90 °C with  $P_{CO_2} = 1$  atm for 3 hours and stirring rate of  $300 \text{ rpm} \pm 5 \text{ rpm}$ , using ATR-FTIR measurements.

**Figure S3.** Experimental configuration of aqueous alkaline amine looping process for accelerated carbon mineralization.

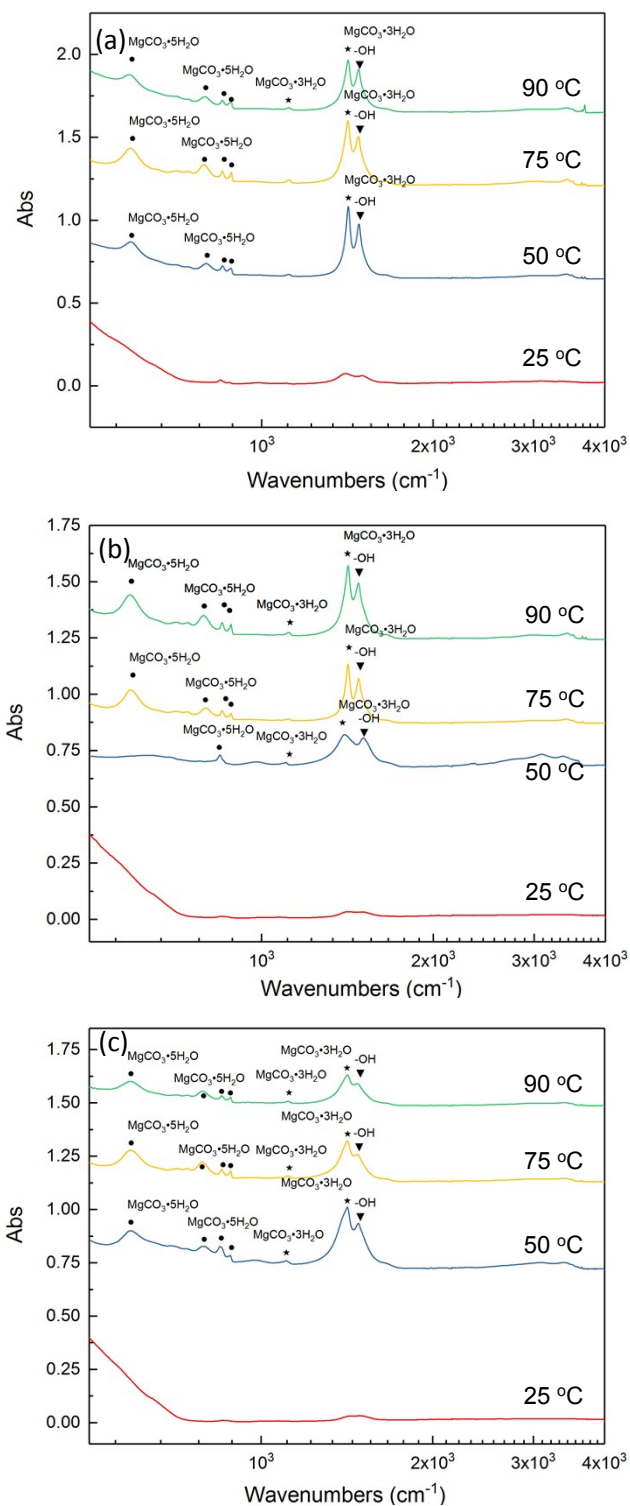
**Figure S4.** Experimental configuration of MgO reacting with CO<sub>2</sub>-loaded MEA in the USAXS/SAXS measurement.

**Figure S5.** Changes in the combined slit-smearred USAXS/SAXS data as MgO is reacted with CO<sub>2</sub>-loaded MEA to produce magnesium carbonate (full data set).

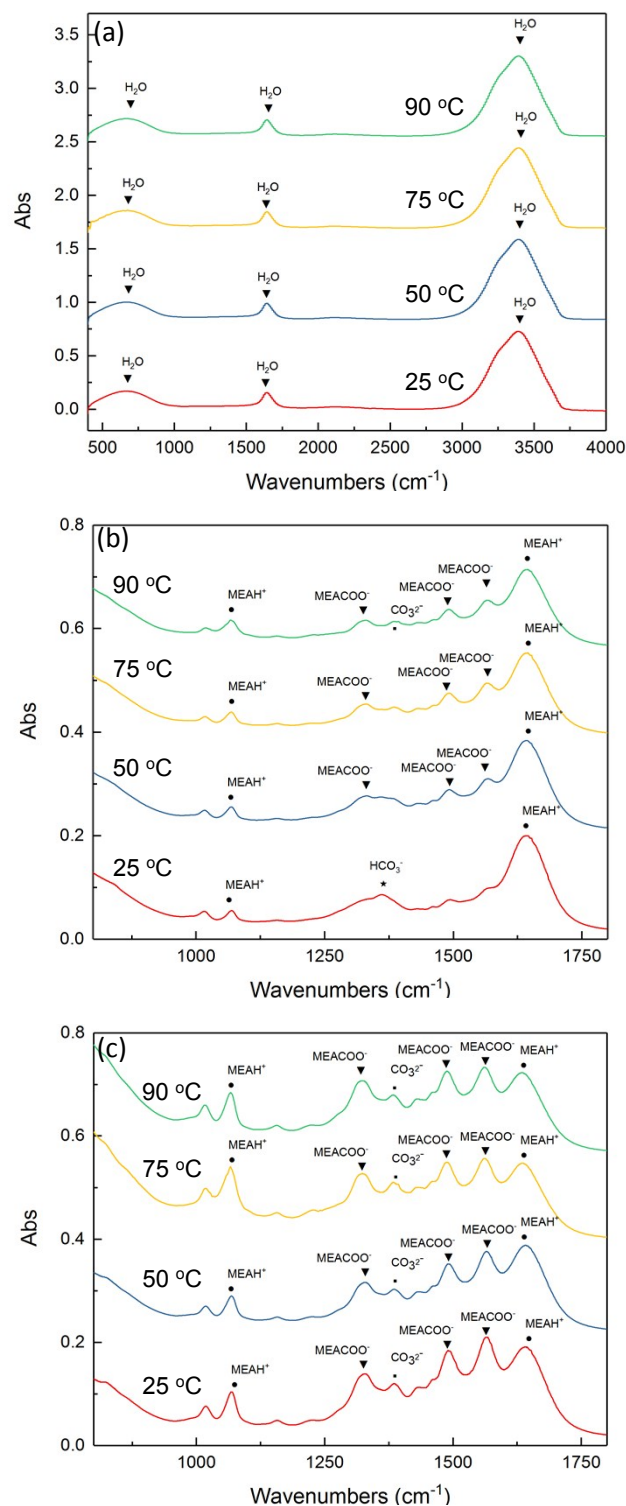
**Figure S6.** Experimental USAXS/SAXS scattering curves and fitted models for samples reacted for (a) 0 min, (b) 122 min, (c) 239 min, and (d) 385 min. The data were fitted using the Modelling-II tool in Irena package embedded in Igor Pro. The curve in (a) was modeled using two unified fit levels between the  $q$ -ranges of  $0.001 - 0.02 \text{ \AA}^{-1}$  and  $0.02 - 0.8 \text{ \AA}^{-1}$ . For curves in panels (b), (c) and (d) besides two unified fit levels in  $q$ -ranges of  $0.001 - 0.02 \text{ \AA}^{-1}$  and  $0.02 - 0.3 \text{ \AA}^{-1}$ , a Lorentzian diffraction peak was also fitted between  $q$  values of  $0.3 - 0.8 \text{ \AA}^{-1}$ .

**Table S1.** The extents of carbon mineralization of MgO as a function of MEA concentrations and temperatures.

Temperature	H <sub>2</sub> O	10 wt% MEA	20 wt% MEA	30 wt% MEA	50 wt% MEA
25 °C	8.1	2.1	1.6	1.5	4.2
50 °C	18.1	32.7	46.9	70.2	12.2
75 °C	42.0	46.3	52.4	62.2	52.7
90 °C	27.8	32.4	37.5	35.6	24.2

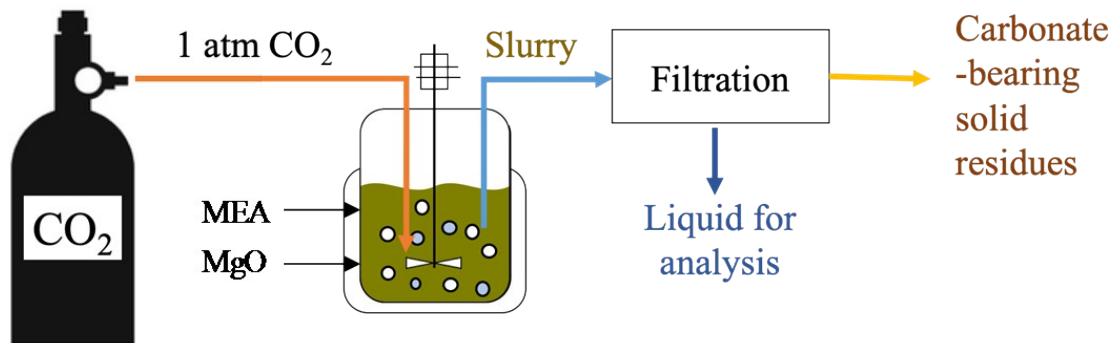


**Figure S1.** Identification of the functional groups present in the solid obtained after reacting MgO with water (a), 10 wt% (b), and 20 wt% MEA (c) at 25 °C, 50 °C, 75 °C, and 90 °C with  $P_{CO_2} = 1$  atm for 3 hours and stirring rate of  $300 \text{ rpm} \pm 5 \text{ rpm}$ , using ATR-FTIR measurements.

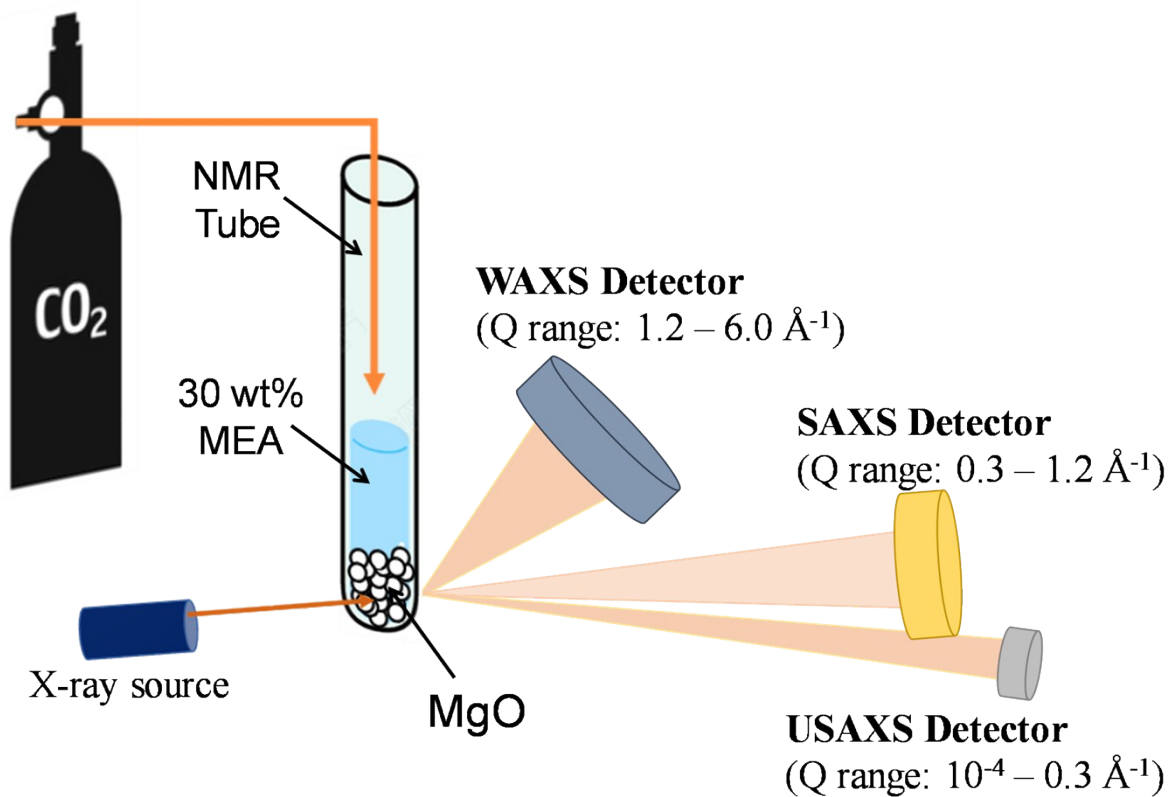


**Figure S2.** Identification of the functional groups present in the fluid obtained from reacting MgO with water (a), 10 wt% (b), and 20 wt% at 25 °C, 50 °C, 75 °C, and 90 °C with  $P_{CO_2} = 1$  atm for 3 hours and stirring rate of  $300 \text{ rpm} \pm 5 \text{ rpm}$ , using ATR-FTIR measurements.



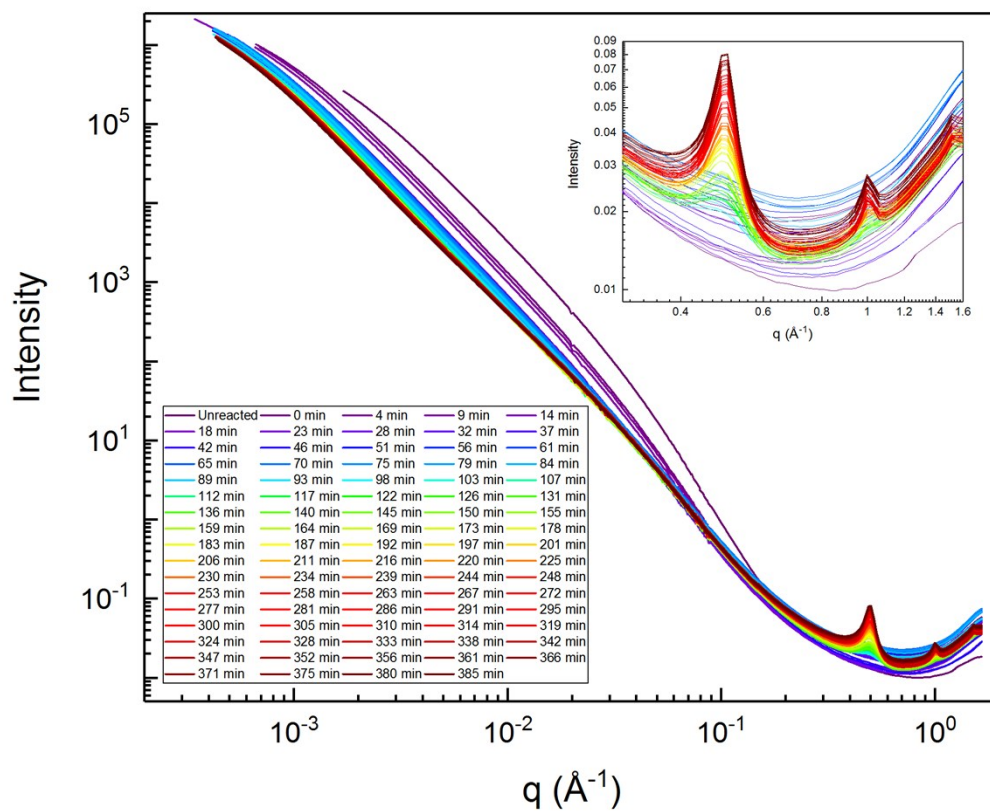


**Figure S3.** Experimental configuration of aqueous alkaline amine looping process for accelerated carbon mineralization.

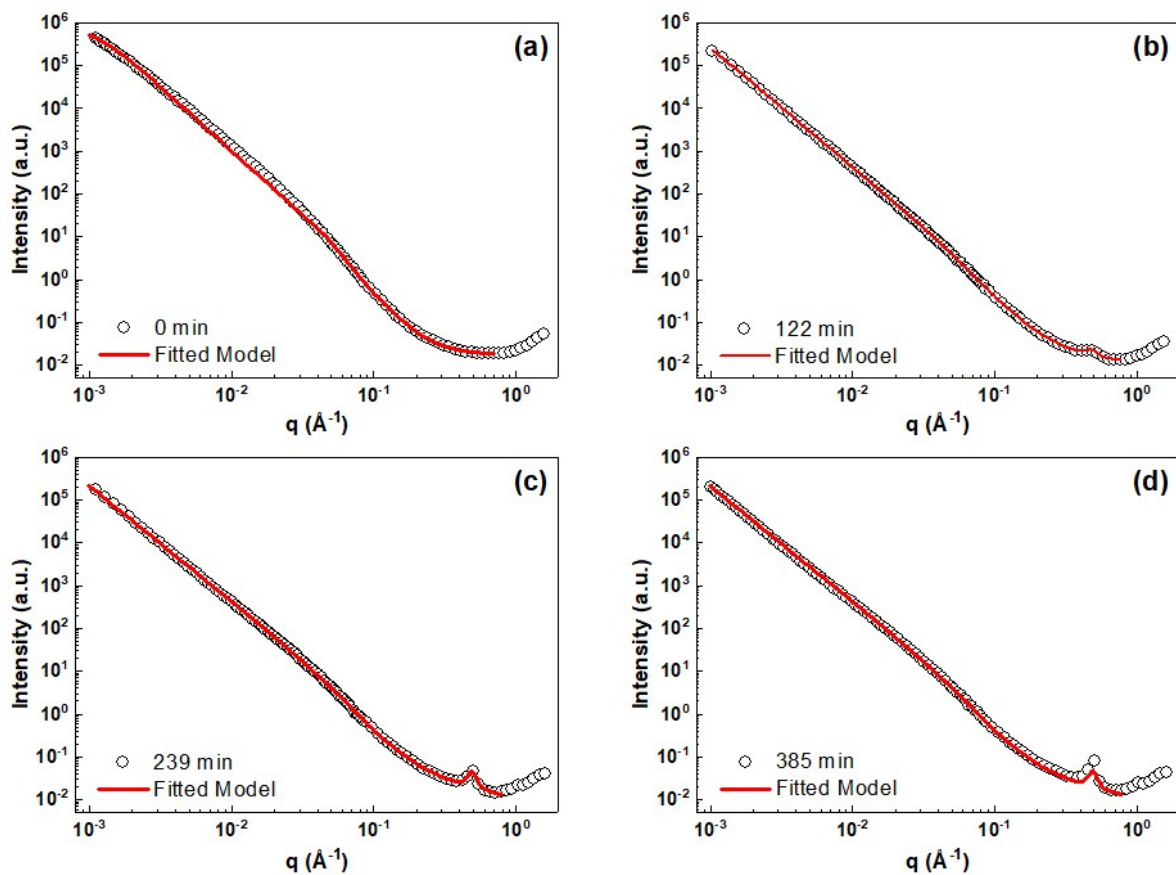


**Figure S4.** Experimental configuration of MgO reacting with CO<sub>2</sub>-loaded MEA in the USAXS/SAXS measurement.





**Figure S5.** Changes in the combined slit-smearred USAXS/SAXS data as MgO is reacted with CO<sub>2</sub>-loaded MEA to produce magnesium carbonate (full set of data).



**Figure S6.** Experimental USAXS/SAXS scattering curves and fitted models for samples reacted for (a) 0 min, (b) 122 min, (c) 239 min, and (d) 385 min. The data were fitted using the Modelling -II tool in Irena package embedded in Igor Pro. The curve in (a) was modeled using two unified fit levels between the  $q$ -ranges of  $0.001 - 0.02 \text{ \AA}^{-1}$  and  $0.02 - 0.8 \text{ \AA}^{-1}$ . For curves in panels (b), (c) and (d) besides two unified fit levels in  $q$ -ranges of  $0.001 - 0.02 \text{ \AA}^{-1}$  and  $0.02 - 0.3 \text{ \AA}^{-1}$ , a Lorentzian diffraction peak was also fitted between  $q$  values of  $0.3 - 0.8 \text{ \AA}^{-1}$ .