

Controlling the Strength of Interaction between Carbon Dioxide and Nitrogen-Rich Carbon Materials by Molecular Design

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Electronic Supplementary Information (ESI†)

Supplementary Figures

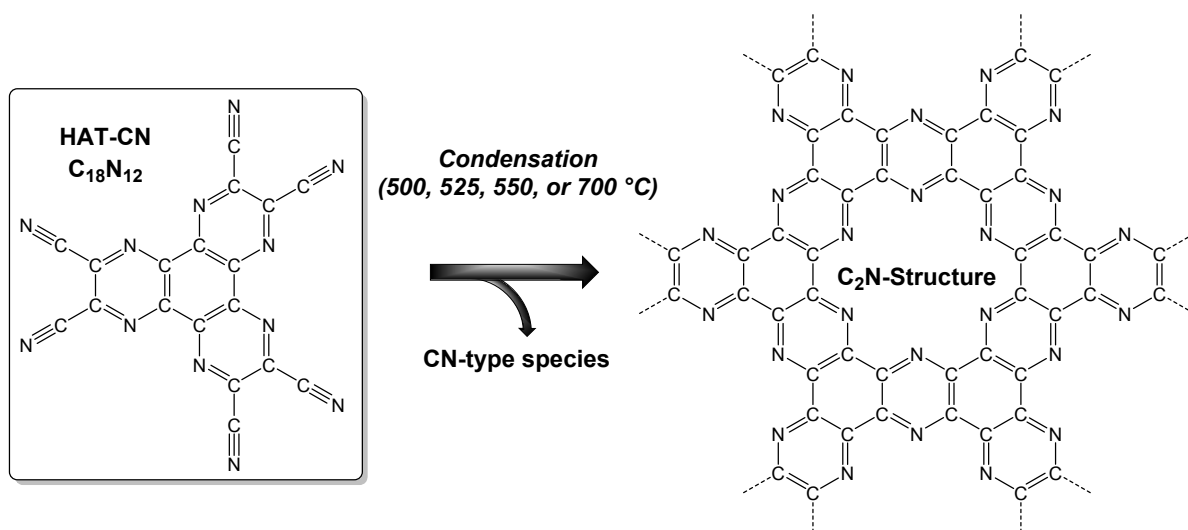


Figure S1. Thermal condensation of HAT-CN to C₂N-type products.

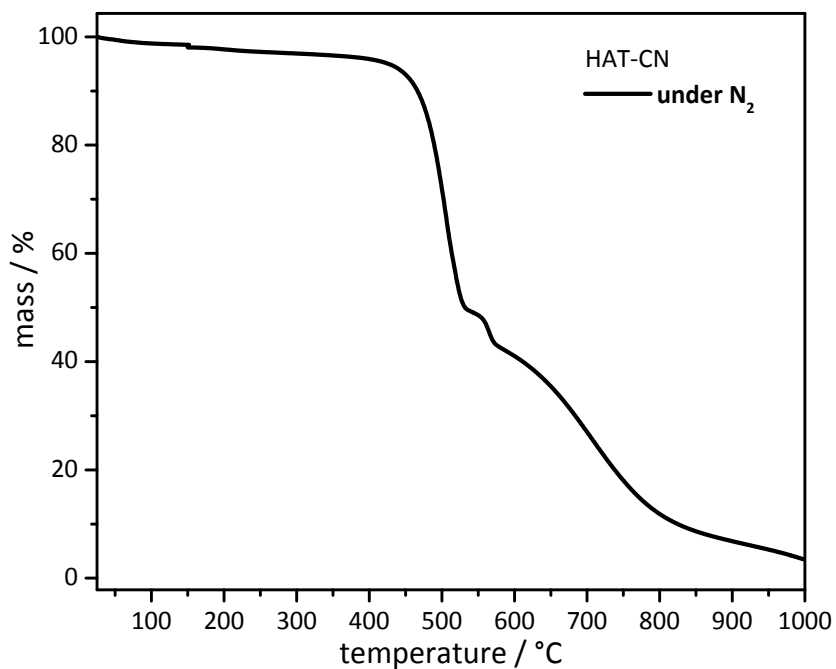


Figure S2. Thermal analysis of HAT-CN under nitrogen with a heating rate of 5 °C min⁻¹.

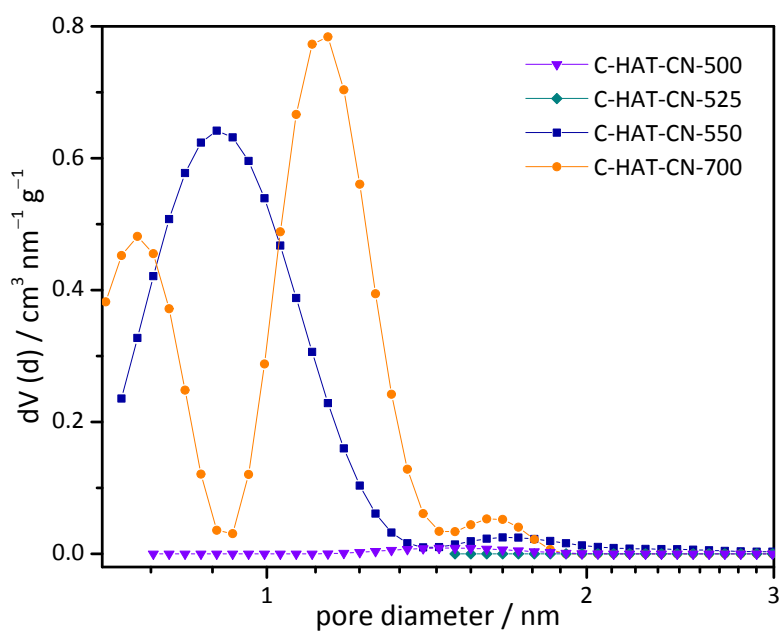


Figure S3. Differential pore size distribution plot calculated with QSDFT (Ar on carbons with cylindrical/sphere pores at -186 °C, adsorption branch kernel) of C-HAT-CN-500, of C-HAT-CN-525, of C-HAT-CN-550, and C-HAT-CN-700.

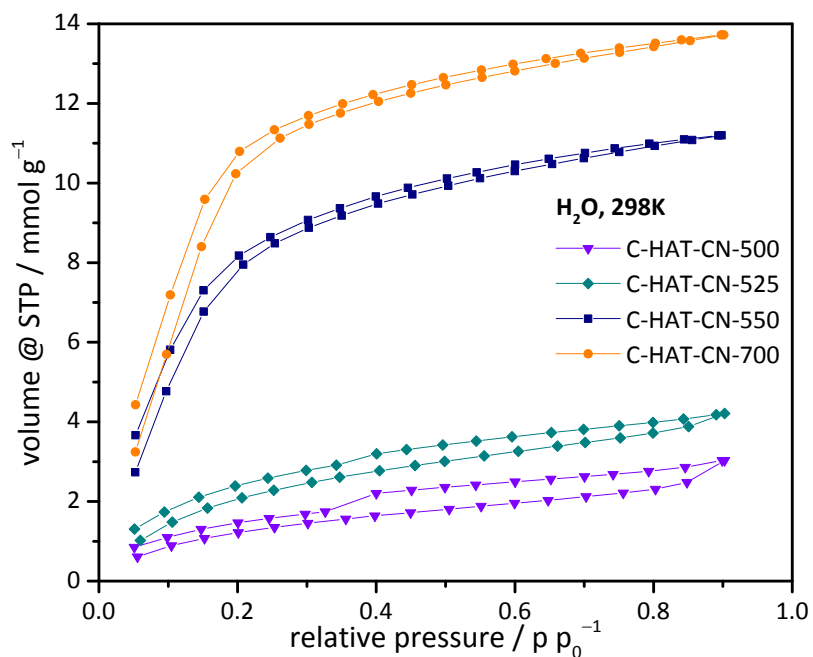


Figure S4. H₂O vapor physisorption isotherms (at 25 °C) of C-HAT-CN-500, C-HAT-CN-525, C-HAT-CN-550, and C-HAT-CN-700.

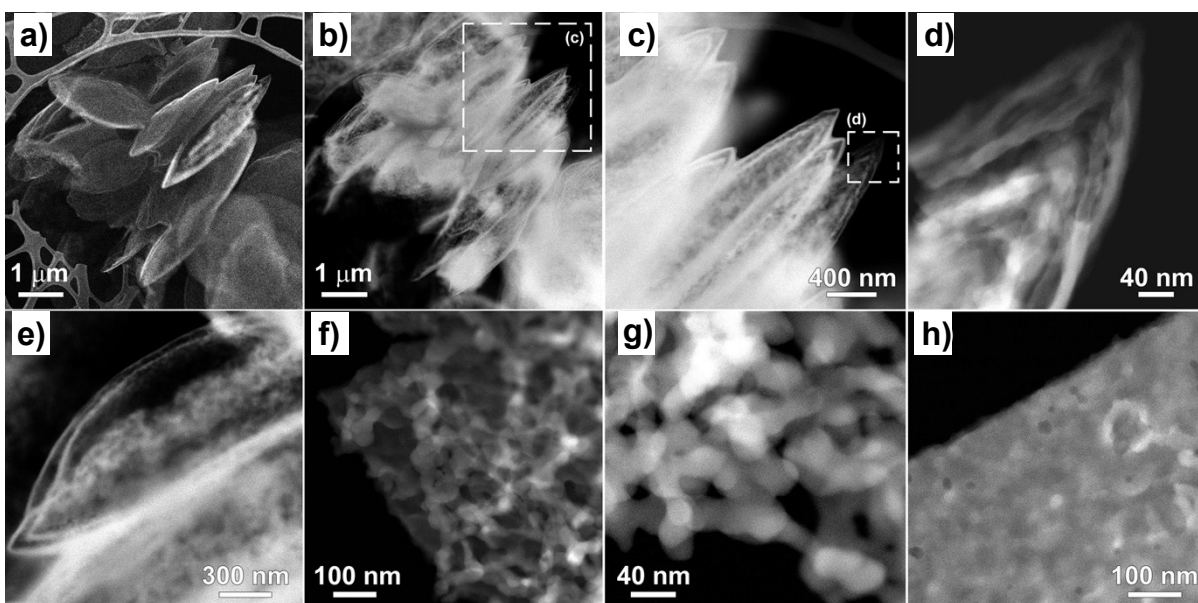


Figure S5. Backscattered electron image (a) and ADF-STEM images (b-h) of HAT-525. (b) shows the same areas as in (a), dotted lines in (b) and (c) indicate areas enlarged in (c) and (d), respectively. Images (f) and (g) show the inner morphology of the particles. (h) shows the morphology of the outer layer of the particles.

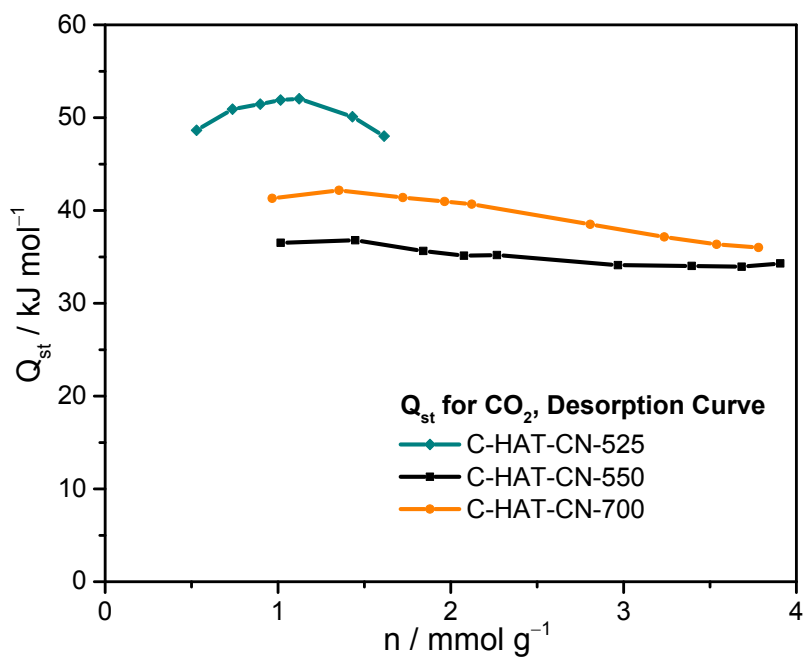


Figure S6. Isothermic heat (Q_{st}) of CO_2 adsorption of C-HAT-CN-525, C-HAT-CN-550, and C-HAT-CN-700.

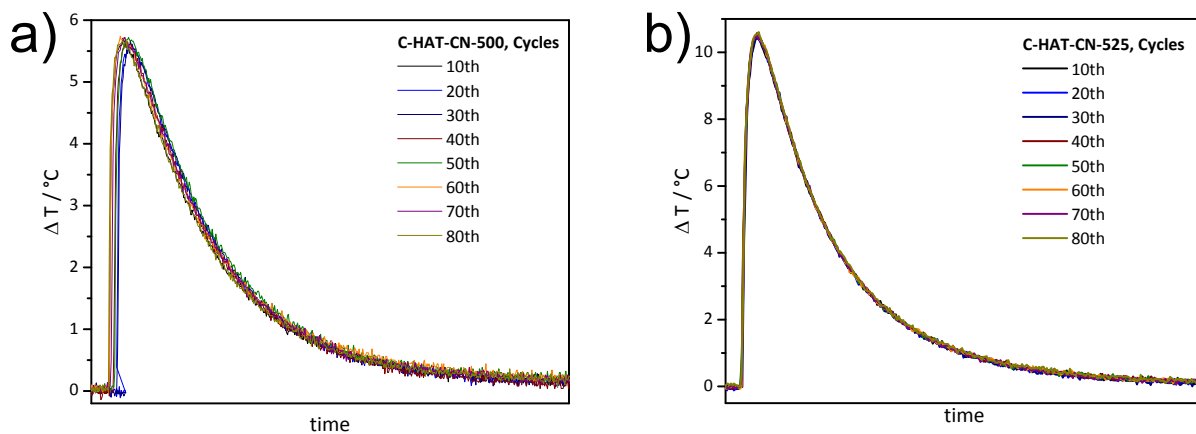


Figure S7. Thermal response curves during cycling experiments on C-HAT-CN-500 (a) and C-HAT-CN-525 (b) using CO_2 as test gas at 1 bar at 25 $^\circ\text{C}$ over 80 adsorption/desorption cycles.

Experimental and Computational Details

Electron Microscopy Investigations

For high-resolution transmission electron microscopy HRTEM and scanning transmission electron microscopy (STEM) observations, a suspension of the sample in ethanol was sonicated for 10 minutes and then drop-casted to the Cu grid with a lacey carbon support and dried for 5 minutes. The STEM study was performed using a double Cs corrected JEOL JEM-ARM200F (S)TEM operated at 80kV and equipped with a cold-field emission gun, a high-angle silicon drift Energy Dispersive X-ray (EDX) detector (solid angle up to 0.98 steradians with a detection area of 100 mm²) and a Gatan Quantum GIF spectroscopy system. Annular Dark Field Scanning Transmission Electron Microscopy (ADF - STEM) images were collected at a probe convergence semi-angle of 24 mrad. HRTEM was performed at the same instrument operating at 200 kV

Computational Details

Periodic density functional theory calculations were carried out using the hybrid Gaussian and plane wave approach,^[1] as implemented in the CP2K/Quickstep code.^[2] The Kohn-Sham orbitals were described by an accurate molecularly optimized double-zeta basis set with one additional set of polarization function, while the charge density was represented by plane waves with a density cutoff of 500 Ry.^[3] Separable norm-conserving pseudopotentials were used to mimic the interactions between the valence electrons and the ionic cores.^[4] The B97-D exchange and correlation functional, which is based on Becke's power-series *Ansatz*, plus a damped atom-pairwise dispersion correction to account for long-range van der Waals interactions was employed.^[5] The C₂N structure was modeled using a supercell with a=b=16.8, c=12.3 Å and $\alpha=\beta=90.0$ and $\gamma=16.0$, which consists of 4 C₂N-layers. Optimized structures were obtained by globally minimizing the potential energy, while varying the atomic positions by dynamical simulated annealing^[6] based on the second-generation Car-Parrinello approach of Kühne *et al.*^[7]

Atomic coordinates CO₂ in C₂N

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O	8.363932	9.699527	5.432231
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O	8.436074	9.699453	1.167809
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O	12.618105	2.456320	11.137695

O	12.618034	2.456198	8.801399
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O	12.618105	2.456320	4.537695
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Atomic coordinates CO₂ in C₃H

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C	7.701030	3.621620	9.899970
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