

Electronic Supplementary Information

Ni loaded on N-doped carbon encapsulated tungsten oxide nanowires as an alkaline-stable electrocatalyst for water reduction

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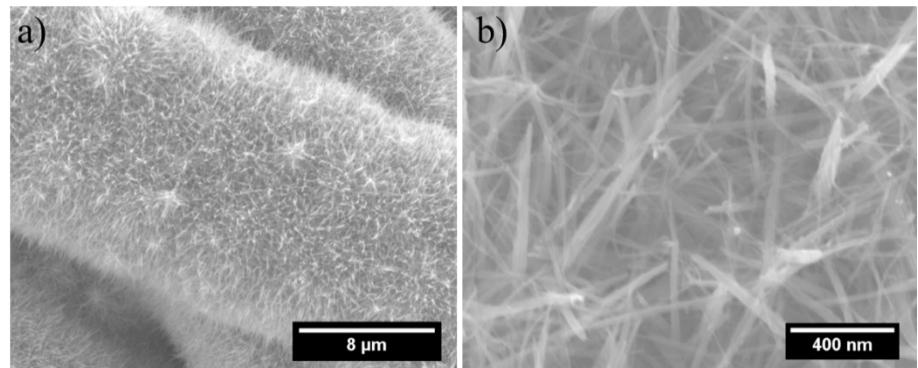


Figure S1. (a) Low- and (b) high-magnification SEM images of WO_x precursor loaded on CFP.

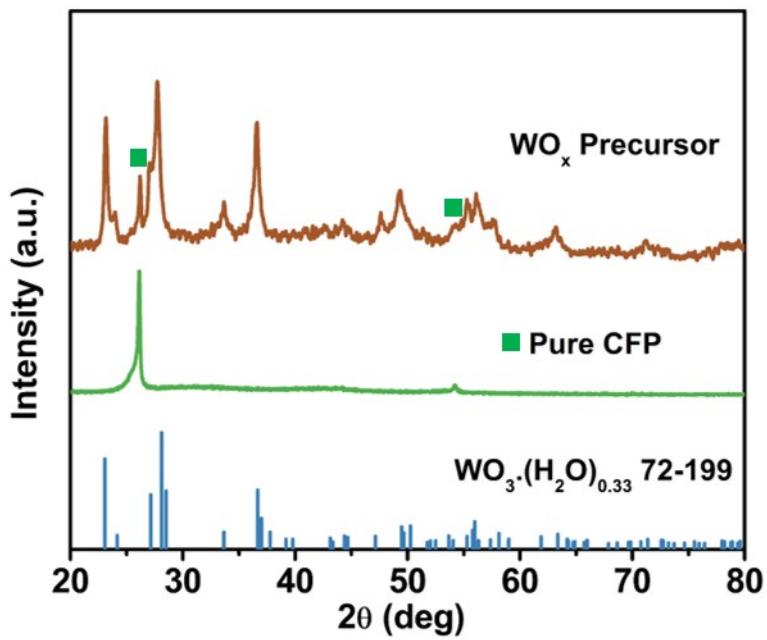


Figure S2. XRD pattern of WO_x precursor. The peaks of CFP are marked by blue square.

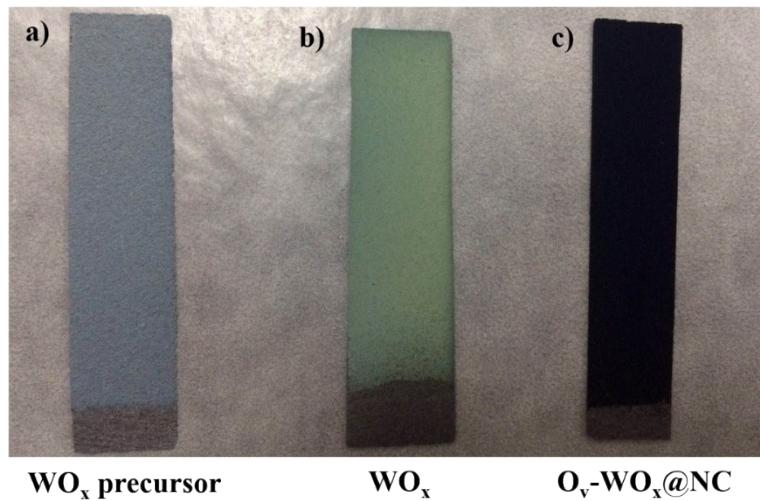


Figure S3. Optical photograph of the (a) WO_x precursor, (b) WO_x, (c) O_v-WO_x@NC.

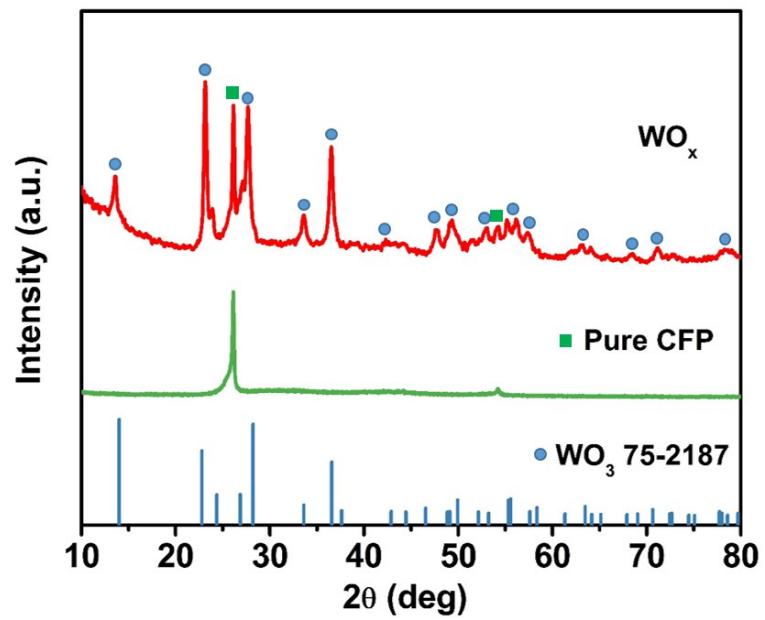


Figure S4. XRD pattern of WO_x .

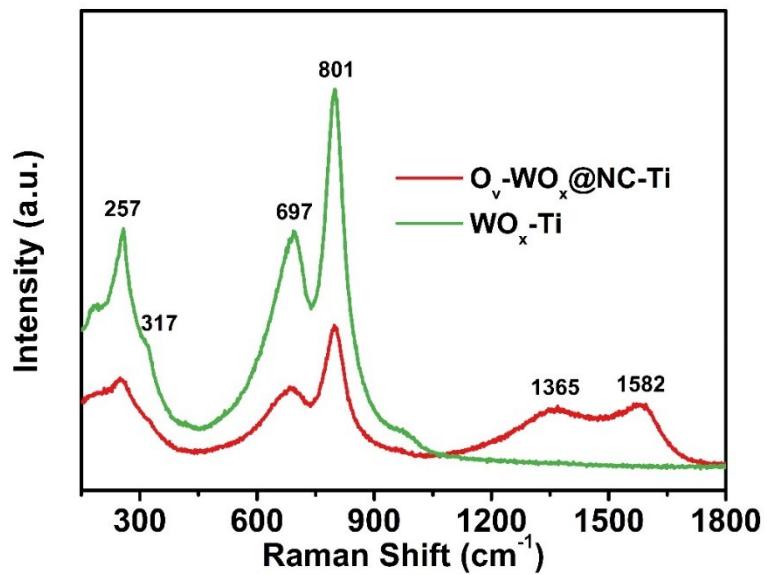


Figure S5. Raman pattern of $\text{O}_\nu\text{-WO}_x\text{@NC}$ and WO_x loaded on Ti foil.

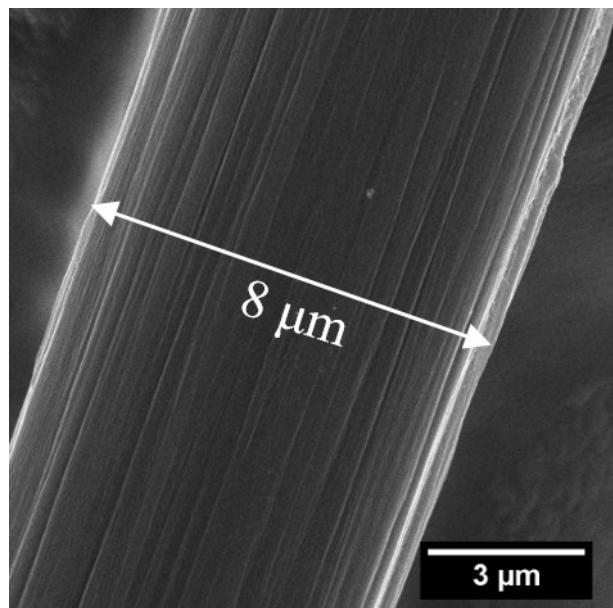


Figure S6. SEM image of the pristine CFP.

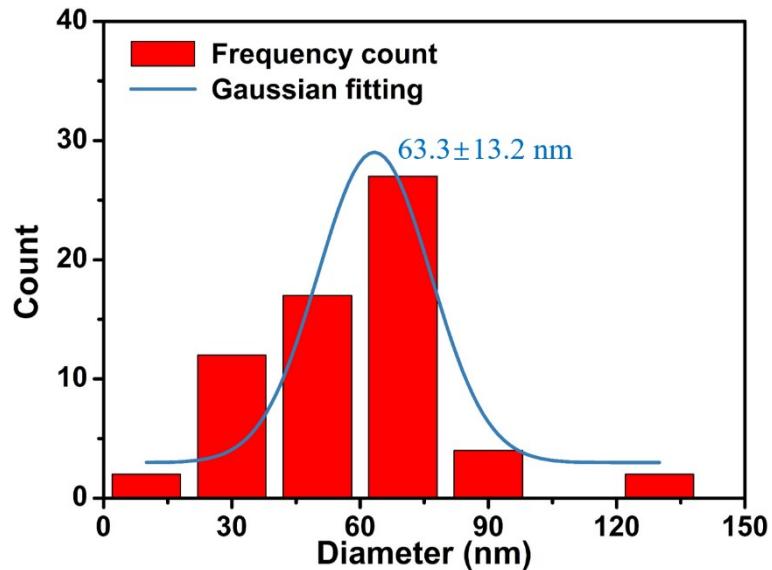


Figure S7. Diameter distribution of O_v - WO_x @NC. The dark cyan line shows the Gaussian fitting of data

Table S1. The amount of W, O, N, C and Ni in the composites detected by EDS analysis

Catalysts	Atomic (%)				
	W	N	C	O	Ni
WO _x	22.36	--	33.06	44.59	--
O _v -WO _x @NC	34.36	5.16	24.75	35.74	--
O _v -WO _x @NC-Ni	30.01	4.51	22.94	37.06	5.49

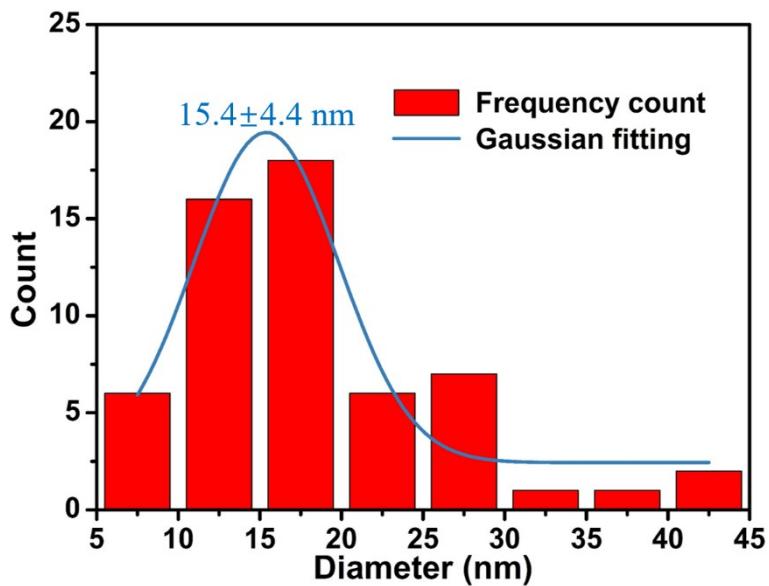


Figure S8. Diameter distribution of Ni nanoparticles. The dark cyan line shows the Gaussian fitting of data

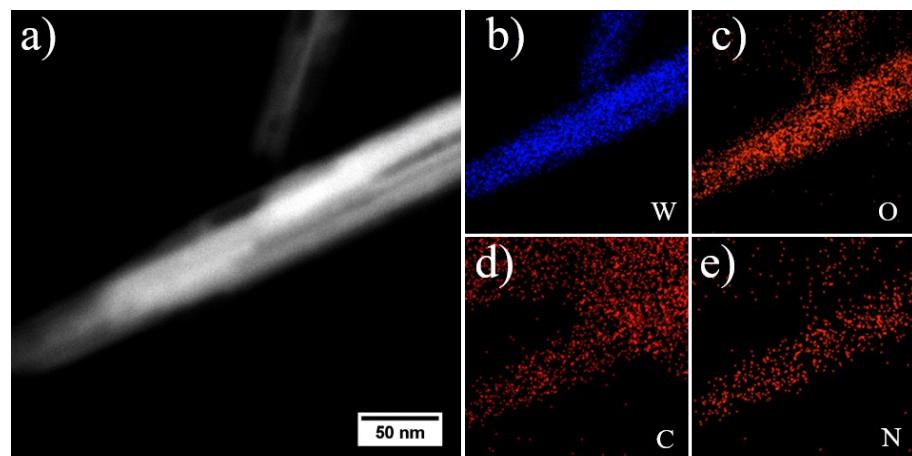


Figure S9. (a) Dark-field STEM image of O_v - WO_x @NC and corresponding elemental mapping images of (b) W, (c) O, (d) C and (e) N.

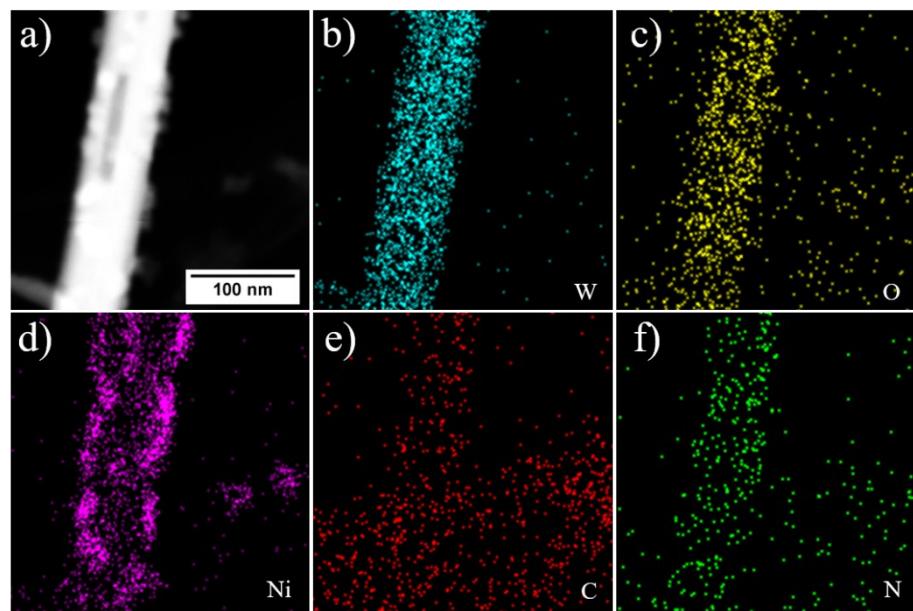


Figure S10. (a) Dark-field STEM image of O_v - WO_x @NC-Ni and corresponding elemental mapping images of (b) W, (c) O, (d) Ni, (e) C and (f) N.

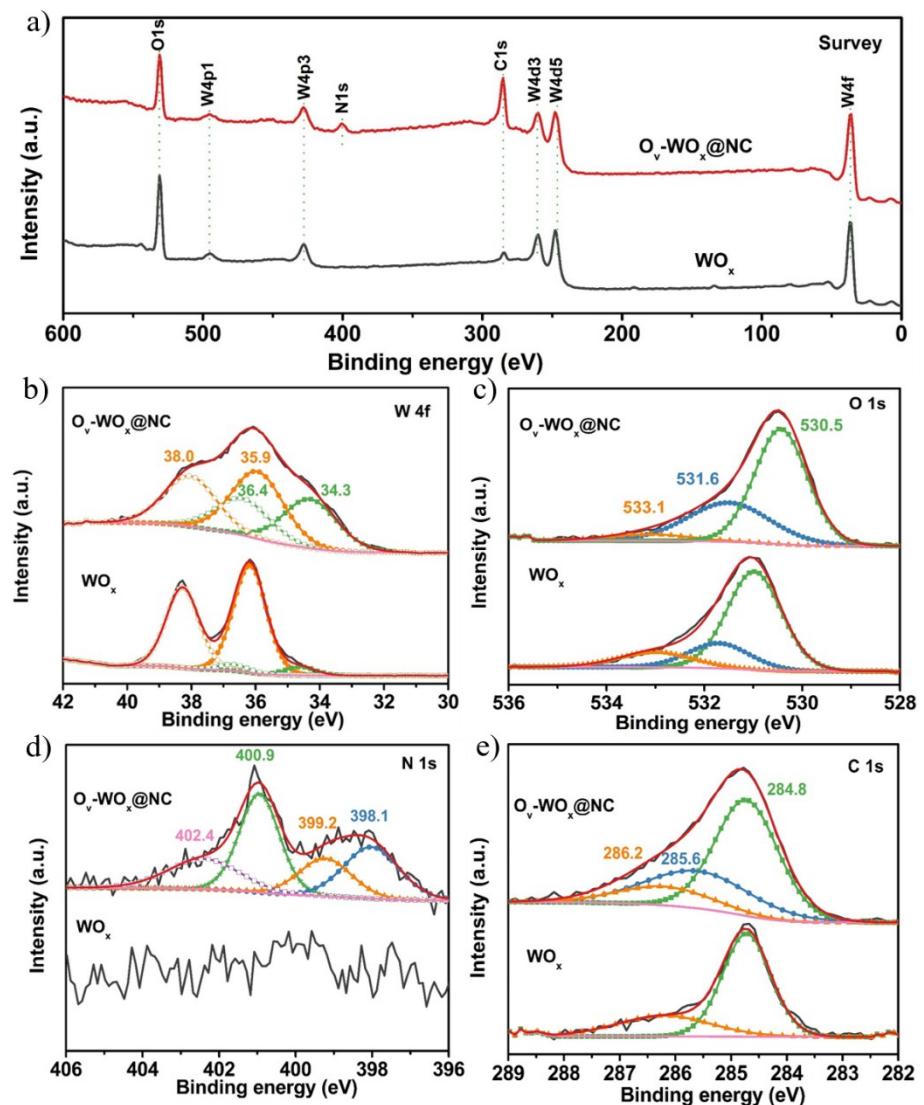


Figure S11. XPS spectra of WO_x and O_v - WO_x @NC.

Table S2. The amount of W, O, N, C and Ni in the composites detected by XPS analysis

Catalysts	Atomic (%)				
	W	N	C	O	Ni
WO _x	15.32	--	21.85	62.83	--
O _v -WO _x @NC	9.02	8.79	58.2	23.99	--
O _v -WO _x @NC-2Months-Air	7.66	9.5	58.91	23.93	--
O _v -WO _x @NC-5h-H ₂ O ₂	11.61	7.53	38.72	42.14	--
O _v -WO _x @NC-Ni	5.83	6.47	52.48	28.84	6.37
O _v -WO _x @NC-Ni-it	4.55	6.77	42.99	37.08	8.61

Table S3. Fitting parameters (peak position, peak area and species percentage) for both W 4f_{7/2} and W 4f_{5/2} spectra taken on the resultant samples.

Samples	species	Peaks		Area		W ⁵⁺ / W ⁶⁺
		4f _{7/2}	4f _{5/2}	4f _{7/2}	4f _{5/2}	
WO _x	W ⁵⁺	34.6	36.7	4713	3535	1:13.82
	W ⁶⁺	36.2	38.3	65119	48839	
O _v -WO _x @NC	W ⁵⁺	34.3	36.4	13000	9750	1:1.31
	W ⁶⁺	35.9	38.0	17242	12932	
O _v -WO _x @NC- 2Months-Air	W ⁵⁺	34.5	36.7	7522	5641	1:1.38
	W ⁶⁺	35.4	37.5	10405	7804	
O _v -WO _x @NC- 5h-H ₂ O ₂	W ⁵⁺	34.7	36.9	11852	8889	1:1.43
	W ⁶⁺	35.5	37.6	16999	12750	
O _v -WO _x @NC- Ni	W ⁵⁺	34.2	36.3	9001	6750	1:1.26
	W ⁶⁺	35.9	38.0	11295	8471	
O _v -WO _x @NC- Ni-it	W ⁵⁺	34.8	36.9	5500	4125	1:1.55
	W ⁶⁺	35.5	37.6	3540	2655	

Table S4. Fitting parameters (peak position, peak area and species percentage) for oxygen vacancies (O_v), W-O-W and O-C taken on the resultant samples.

Samples	species	Peaks	Area	O_v/O
WO _x	O_v	531.7	22001	
	W-O-W	530.9	76487	0.192
	O-C	533.0	16002	
O _v -WO _x @NC	O_v	531.6	12001	
	W-O-W	530.5	22497	0.329
	O-C	533.1	1999	
O _v -WO _x @NC- 2Months-Air	O_v	531.5	7675	
	W-O-W	530.5	14501	0.338
	O-C	533.1	499	
O _v -WO _x @NC- 5h-H ₂ O ₂	O_v	531.5	14121	
	W-O-W	530.4	26941	0.327
	O-C	533.1	2096	
O _v -WO _x @NC- Ni	O_v	531.6	15770	
	W-O-W	530.5	21001	
	O-C	533.2	8409	
	O-Ni	529.7	2001	
O _v -WO _x @NC- Ni-it	O_v	531.7	12646	
	W-O-W	530.7	15626	
	O-C	533.0	3532	
	O-Ni	529.8	1999	0.374

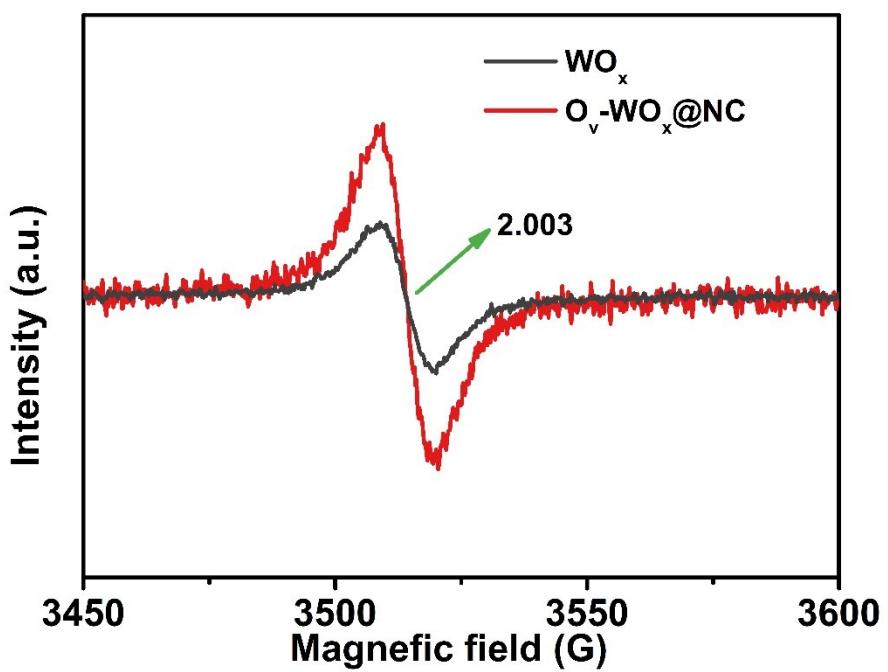


Figure S12. EPR spectra of WO_x and $\text{O}_v\text{-}\text{WO}_x@\text{NC}$ at room temperature.

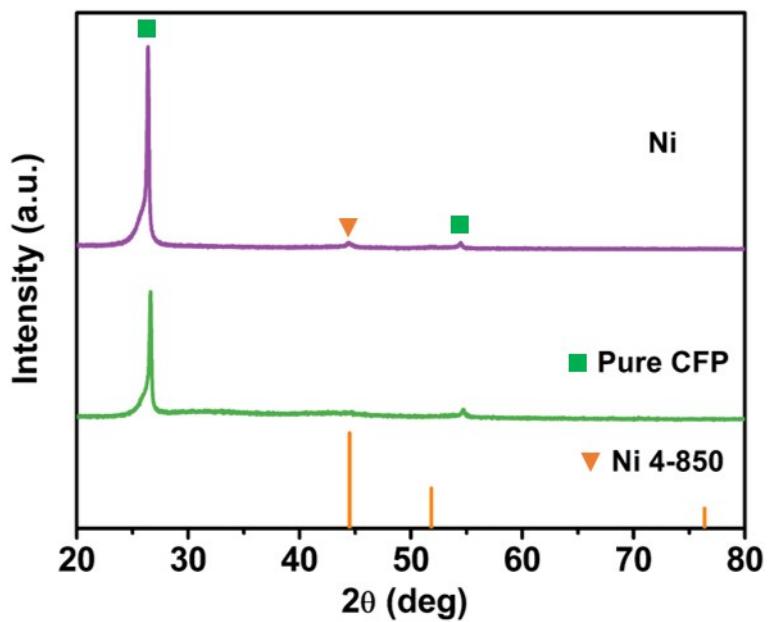


Figure S13. XRD pattern of Ni.

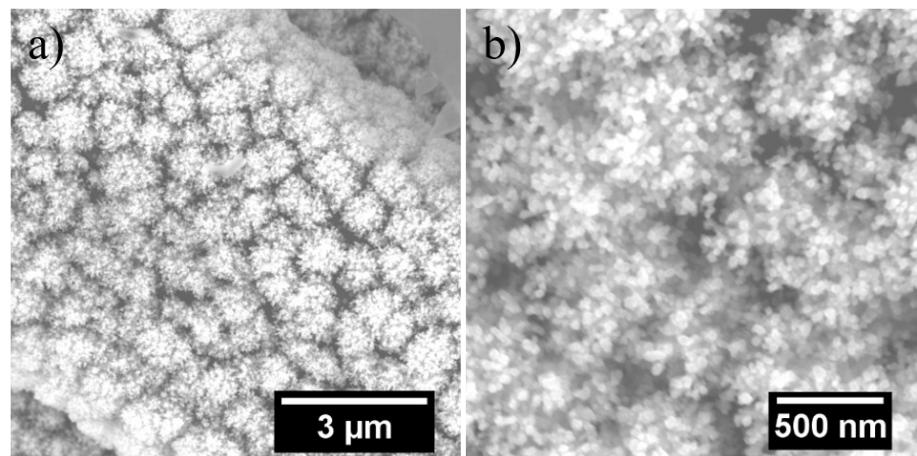


Figure S14. (a) Low- and (b) high-magnification SEM images of Ni.

Table S5. Performances of representative highly efficient TMOs-based HER catalysts in alkaline media.

Reference	Catalyst	Mass density (mg cm ⁻²)	η_{20} (mV)	η_{100} (mV)	Tafel slope (mV/dec)	Counter electrode	Electrolyte
This work	O _v -WO _x @NC-Ni	7.2	67	164	62	Graphite rod	1 M KOH
J. Mater. Chem. A, 2019, DOI: 10.1039/C9TA03652K	M-Co ₃ O ₄ layer	0.76	115	203	63	Graphite rod	1 M KOH
J. Mater. Chem. A, 2017, 5, 9655–9660	WO ₂ HN/NF	1.57	75	--	43	Graphite plate	1 M KOH
ACS Catal. 2018, 8, 5062-5069	Co/Co ₂ Mo ₃ O ₈	2.5	50	210	33	Graphite rod	1 M KOH
Inorg. Chem. Front., 2018, 5, 3042-3045	3D Ni ₃ N-CeO ₂ nanohybrid	--	105	200	122	Graphite plate	1 M KOH
Nano Energy, 2018, 43, 103–109	NiO NRs-m-O _v	0.2	150	280	100	Graphite rod	1 M KOH
Nat. Commun, 2015, 6, 7261	Lithium-induced NiFeO _x nanoparticles	1.6	133	--	--	Pt wire	1 M KOH
Adv. Energy Mater. 2016, 6, 1600528	mesoporous MoO _{3-x}	0.2	170	--	56	Graphite rod	0.1 M KOH
Nanoscale, 2017, 9, 4409–4418	NiO@NF	3.5	420	--	231	Graphite rod	1 M KOH
J. Am. Chem. Soc. 2015, 137, 2688-2694	Cobalt-cobalt oxide/N-doped carbon hybrids	0.42	280	--	115	Pt plate	1 M KOH
Energy Environ. Sci., 2017, 10, 2563	P-Co ₃ O ₄	0.4	135	165	52	Graphite rod	1 M KOH
Angew. Chem. Int. Ed. 2017, 56, 1324-1328	Hollow Co ₃ O ₄ microtube arrays	--	190	290	98	Pt wire	1 M KOH
Inorg. Chem. 2018, 57, 548-552	Ni ₂ P-CeO ₂ /Ti mesh	1.58	84	160	87	Platinum wire	1 M KOH
Nanoscale, 2018, 10, 2213-2217	CeO ₂ -Cu ₃ P/NF	--	148	275	132	Graphite sheet	1 M KOH

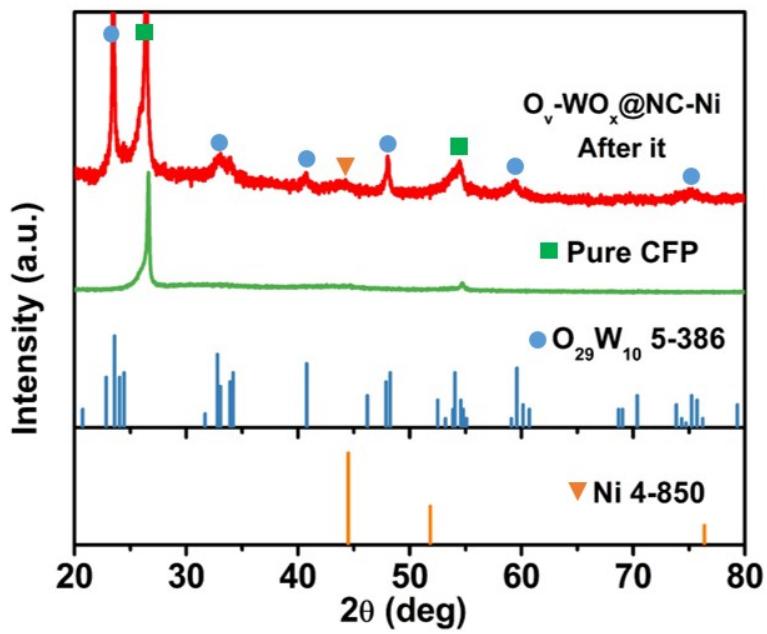


Figure S15. XRD pattern of $O_v\text{-}WO_x\text{@NC-Ni}$ after the electrochemical measurement.

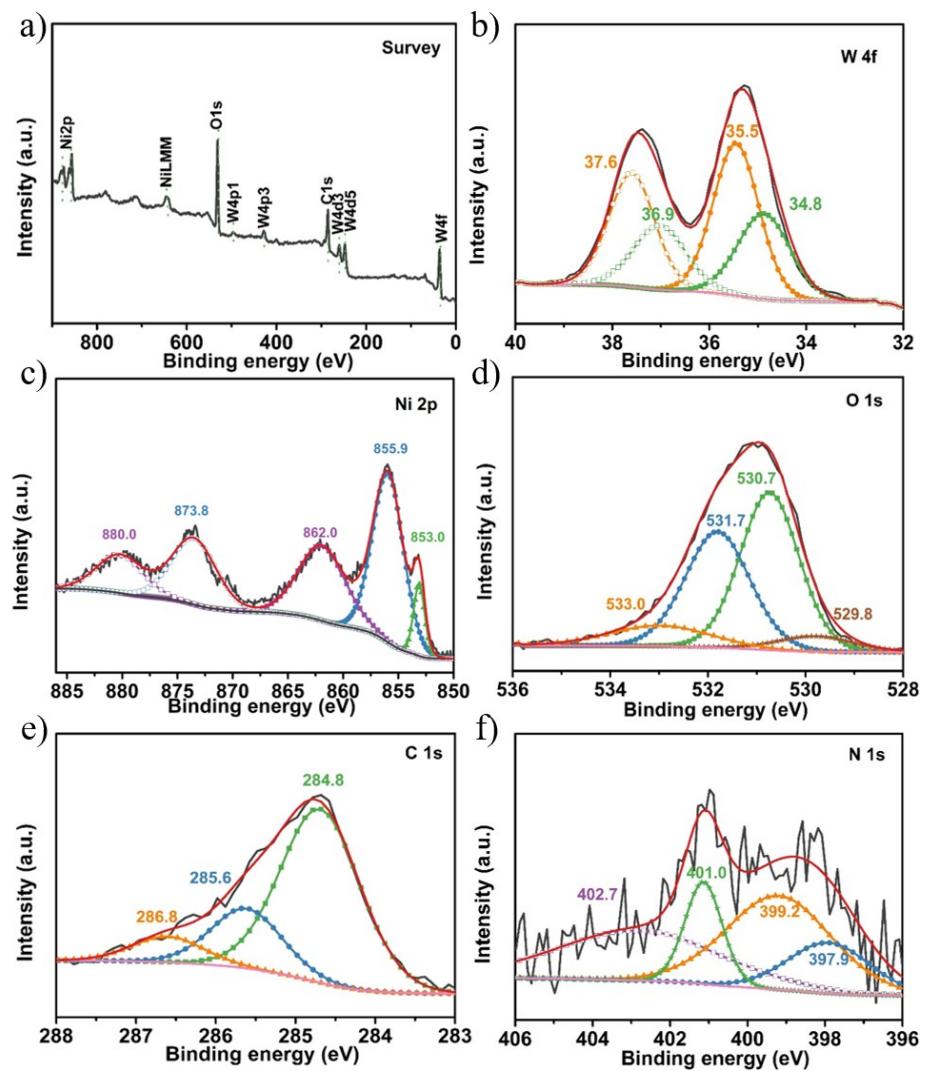


Figure S16. XPS pattern of $\text{O}_\nu\text{-WO}_x\text{@NC-Ni}$ after the electrochemical measurement.

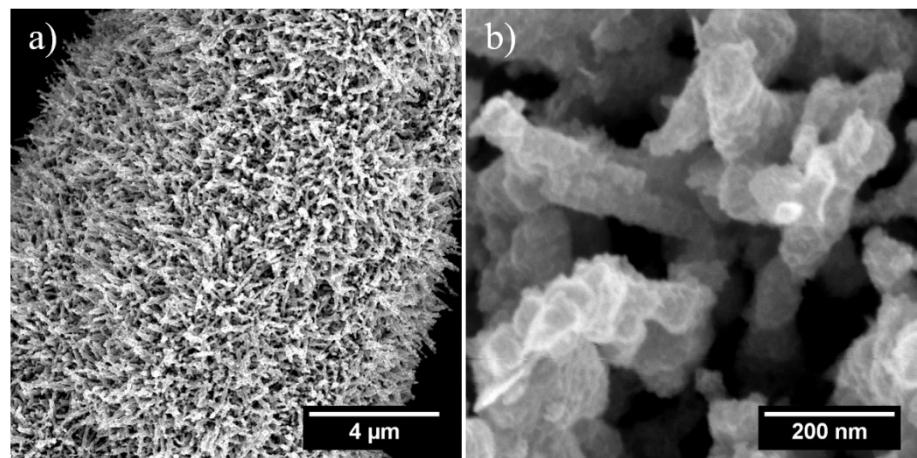


Figure S17. SEM pattern of O_v - WO_x @NC-Ni after the electrochemical measurement.

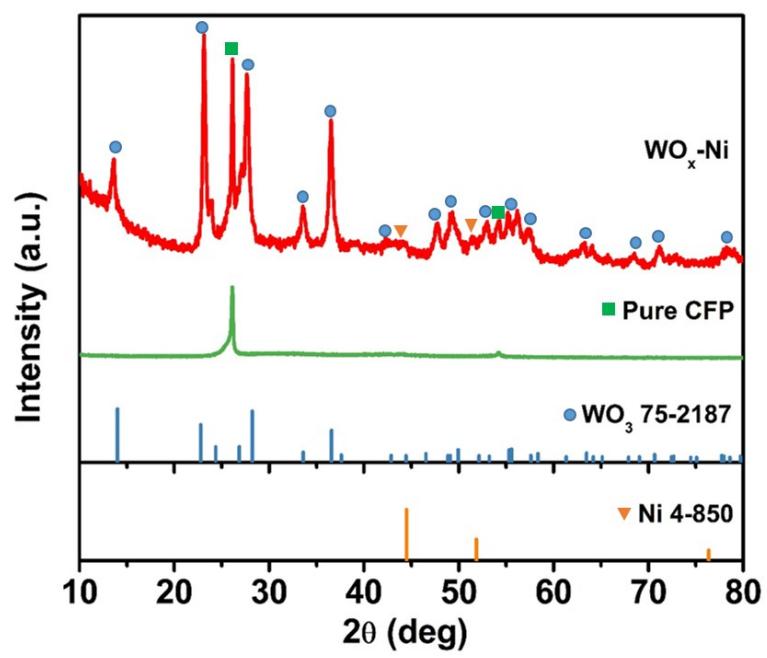


Figure S18. XRD pattern of WO_x -Ni.

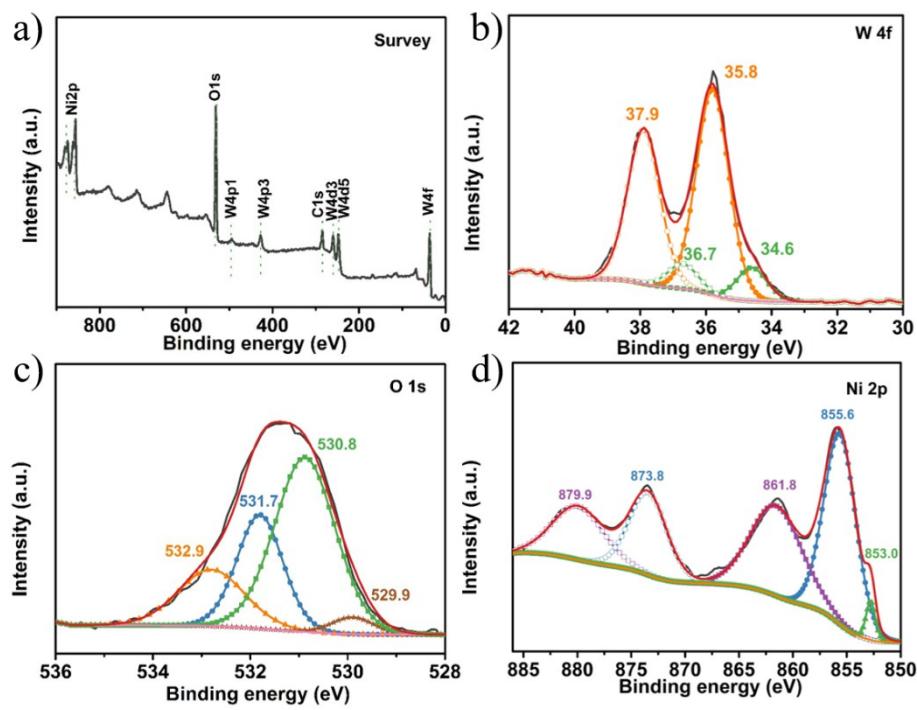


Figure S19. XPS pattern of $\text{WO}_x\text{-Ni}$.

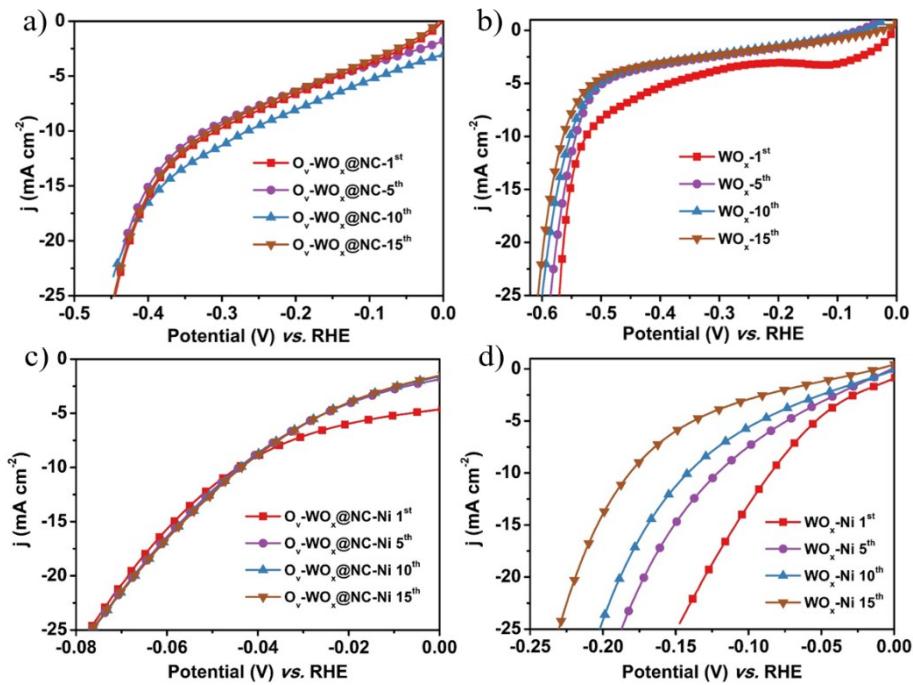


Figure S20. Polarization curves corresponding to different LSV activation scans of (a) $O_v\text{-WO}_x@\text{NC}$, (b) WO_x , (c) $O_v\text{-WO}_x@\text{NC-Ni}$ and (d) $\text{WO}_x\text{-Ni}$.

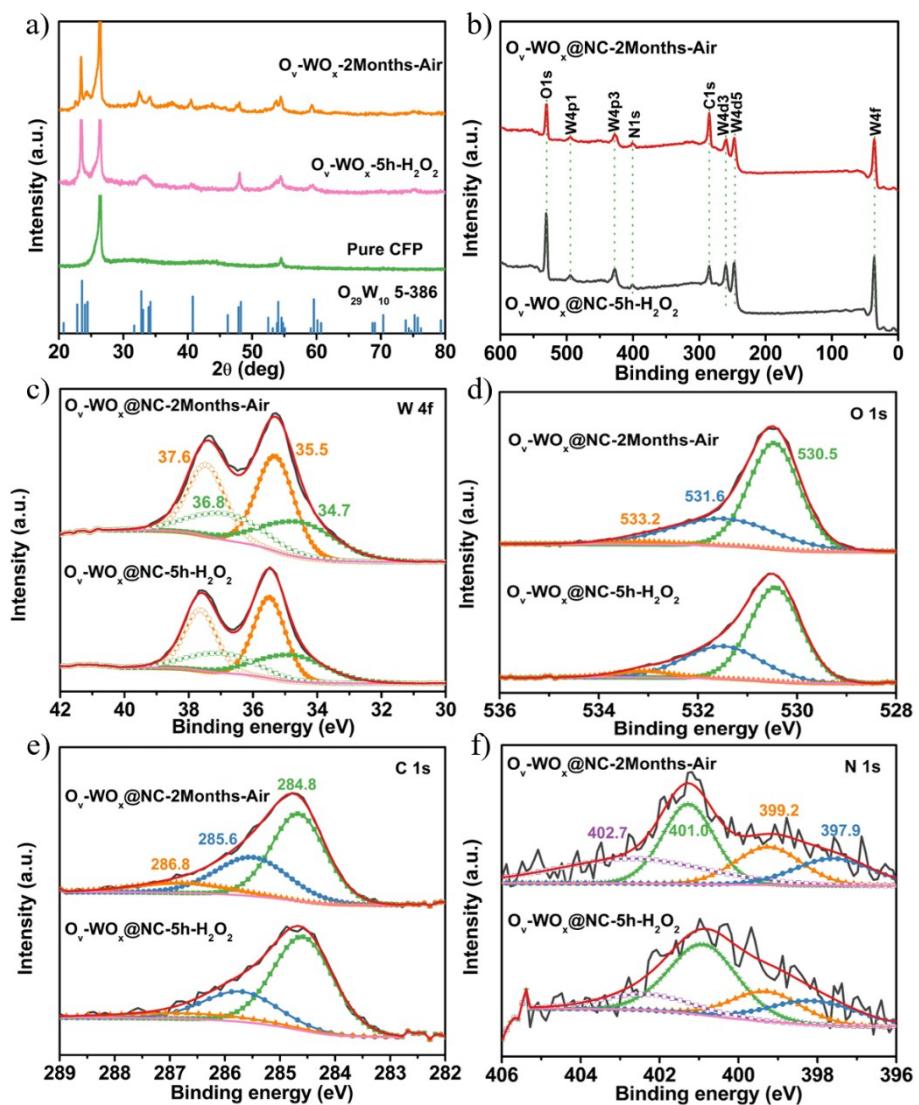


Figure S21. The XRD and XPS spectra of O_v - WO_x @NC-2Months-Air and O_v - WO_x @NC-5h- H_2O_2 .

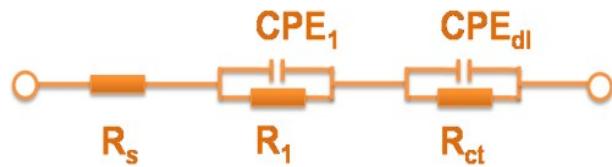


Figure S22. Equivalent circuit used to fit the EIS data. R_s is the overall series resistance, CPE_1 and R_1 are the constant phase element and resistance describing electron transport at GCE/electrocatalyst interface, respectively, CPE_{dl} is the constant phase element of the electrocatalyst/electrolyte interface, and R_{ct} is the charge transfer resistance at electrocatalyst /electrolyte interface.

Table S6. The fitting results of EIS spectra in basic solution.

Sample	R _s (Ω)	Q _l (F cm ⁻² S ⁿ⁻¹)	n _l	R _l (Ω)	Q _{ct} (F cm ⁻² S ⁿ⁻¹)	n _{ct}	R _{ct} (Ω)
WO _x	1.958	1.605e-3	0.7521	2.041	0.04355	0.2032	570.2
O _v -WO _x @NC	1.062	1.199e-6	1	2.054	0.01373	0.7535	295.4
Ni	1.598	7.566e-6	0.8635	1.866	0.001688	0.7888	30.35
O _v -WO _x @NC-Ni	1.069	9.471e-7	0.9286	2.52	0.046	0.832	12.09

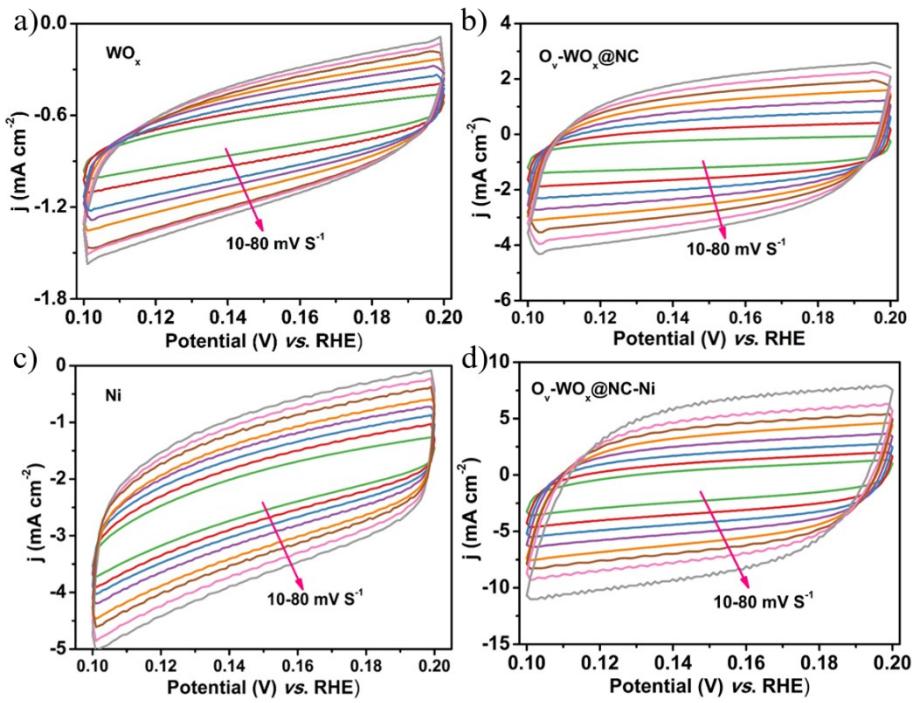


Figure S23. Cyclic voltammetry scans of (a) WO_x , (b) $\text{O}_v\text{-}\text{WO}_x@\text{NC}$, (c) Ni and (d) $\text{O}_v\text{-}\text{WO}_x@\text{NC-Ni}$.

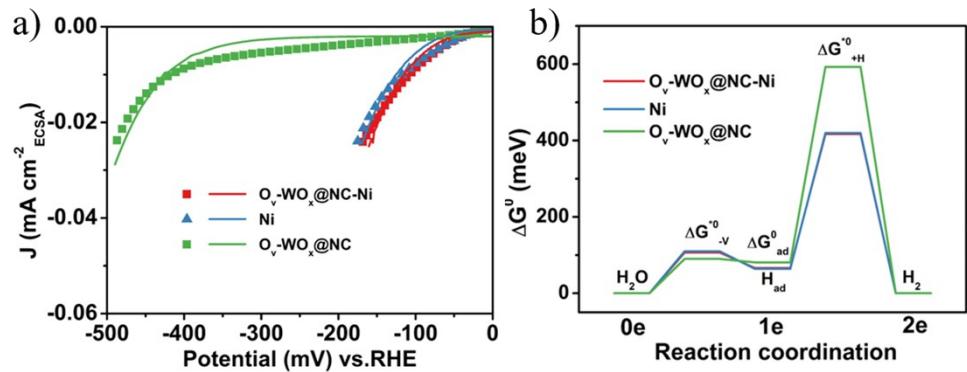


Figure S24. (a) Polarization curves normalized by ECSA: the scatters are experimental data, and lines are fitting curves using the dual-pathway kinetic model. (b) Free energy diagram in 1 M KOH fitted from J_{ECSA} .

Table S7. The fitting results of free energies corresponding to different steps during the HER.

Sample	ΔG_{-V}^{*0} (meV)	ΔG_{+H}^{*0} (meV)	ΔG_{+T}^{*0} (meV)	ΔG_{ad}^0 (meV)	$\Delta G_{+H}^{*0} - \Delta G_{ad}^0$ (meV)
O _v -WO _x @NC	90	593	962	81	512
Ni	110	420	1043	64	359
O _v -WO _x @NC-Ni	107	417	1041	66	351

ΔG_{-V}^{*0} : the standard activation free energy for Volmer step.

ΔG_{+H}^{*0} : the standard activation free energy for Heyrovsky step.

ΔG_{+T}^{*0} : the standard activation free energy for Tafel step.

ΔG_{ad}^0 : the standard free energy of adsorption for the reaction intermediate.

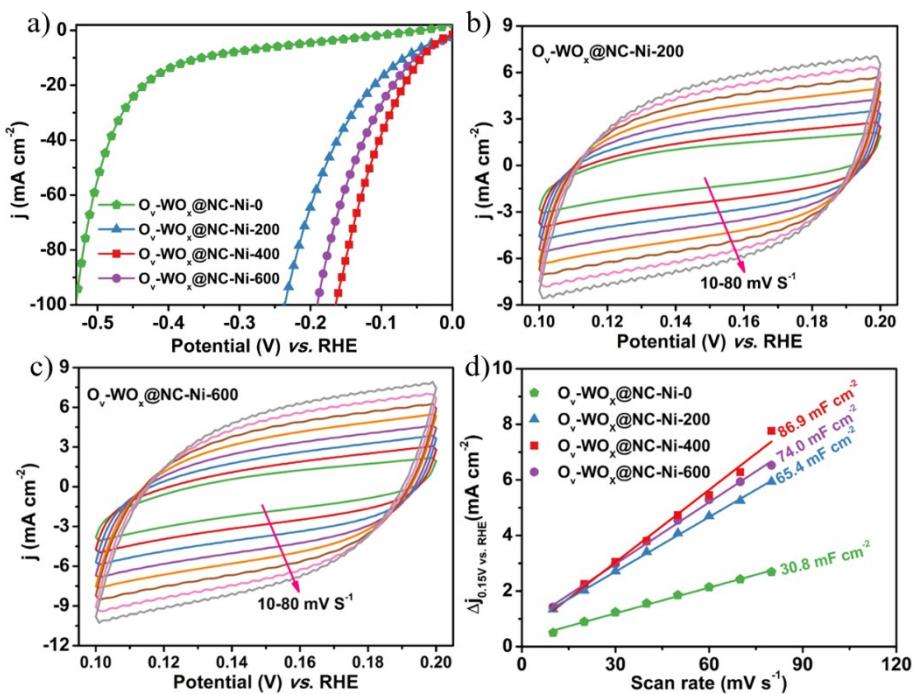


Figure S25. (a) Polarization curves of $\text{O}_\nu\text{-WO}_x\text{@NC-Ni-XX}$. XX is the electrodeposition time of Ni. $\text{O}_\nu\text{-WO}_x\text{@NC-Ni-0}$ is $\text{O}_\nu\text{-WO}_x\text{@NC}$ mentioned above, and $\text{O}_\nu\text{-WO}_x\text{@NC-Ni-400}$ is also named as $\text{O}_\nu\text{-WO}_x\text{@NC-Ni}$. Cyclic voltammetry scans of (b) $\text{O}_\nu\text{-WO}_x\text{@NC-Ni-200}$ and (c) $\text{O}_\nu\text{-WO}_x\text{@NC-Ni-600}$. (d) Estimation of the C_{dl} through plotting the current density difference ($\Delta J = 1/2(J_a - J_c)$) at 0.15 V vs. RHE obtained from the CV against scan rate to fit a linear regression.

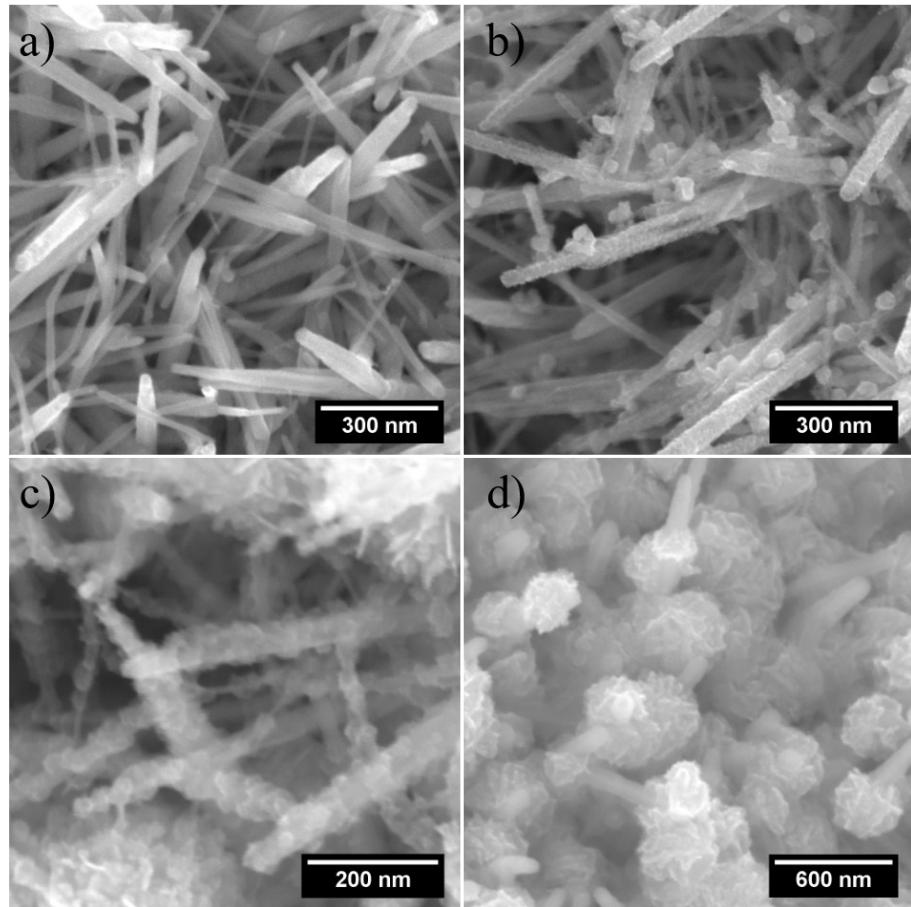


Figure S26. SEM iamges of (a) O_v - WO_x @NC-Ni-0 (i.e. O_v - WO_x @NC), (b) O_v - WO_x @NC-Ni-200, (c) O_v - WO_x @NC-Ni-400 (i.e. O_v - WO_x @NC-Ni) and (d) O_v - WO_x @NC-Ni-600.