

**Supplementary Information
Sustainable Energy & Fuels**

Micropores-in-Macroporous Gel Polymer Electrolytes for Alkali Metal Batteries

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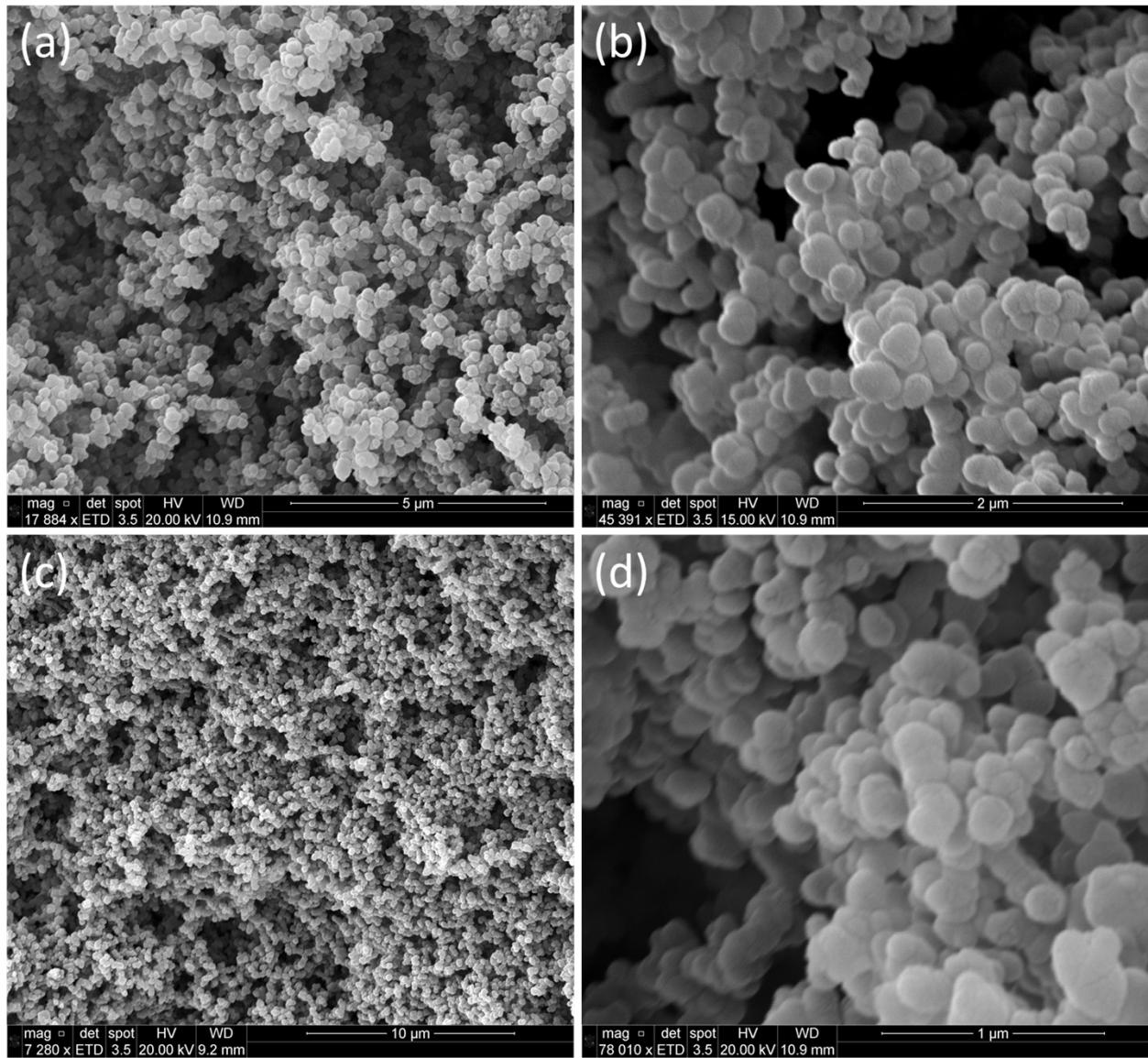


Figure S1. SEM images of (a, b) HCFu and (c, d) HCPy particles after being removed from the PH polymer host (see Experimental Section)

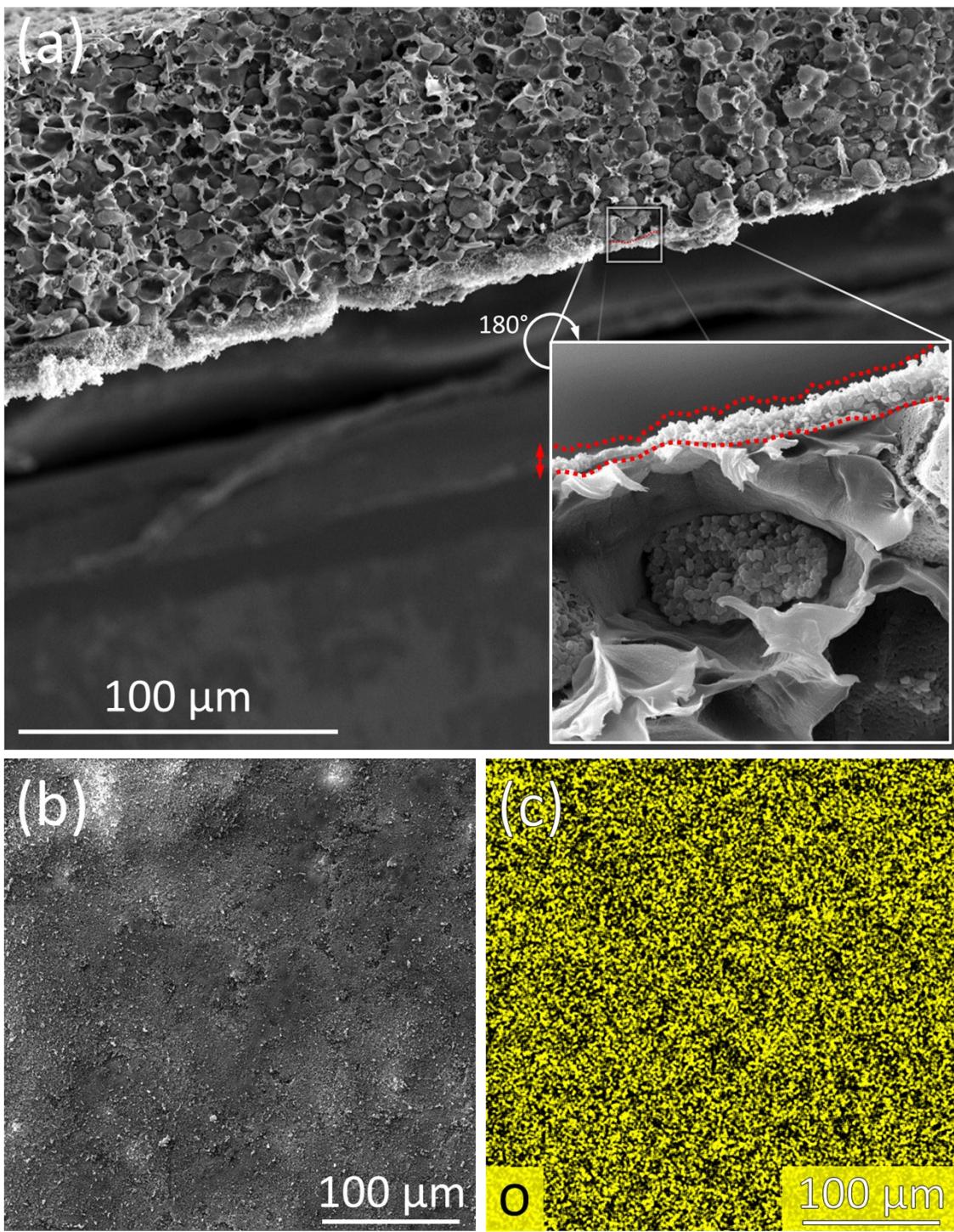


Figure S2. (a) Cross-sectional SEM image, (b) surface SEM image, and (c) oxygen-EDX image of HCFu-PH membrane

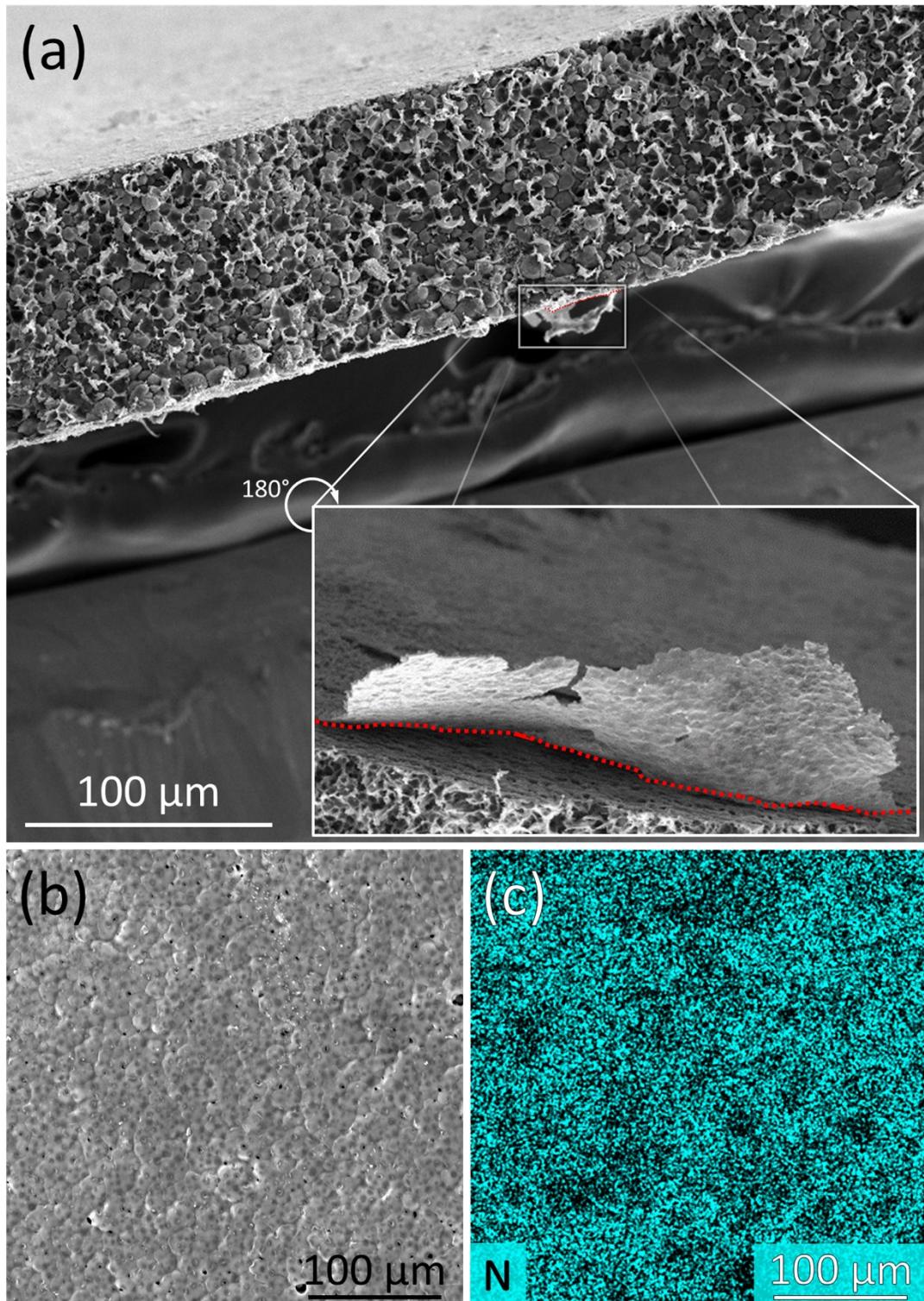


Figure S3. (a) Cross-sectional SEM image, (b) surface SEM image, and (c) nitrogen-EDX image of HCPy-PH membrane

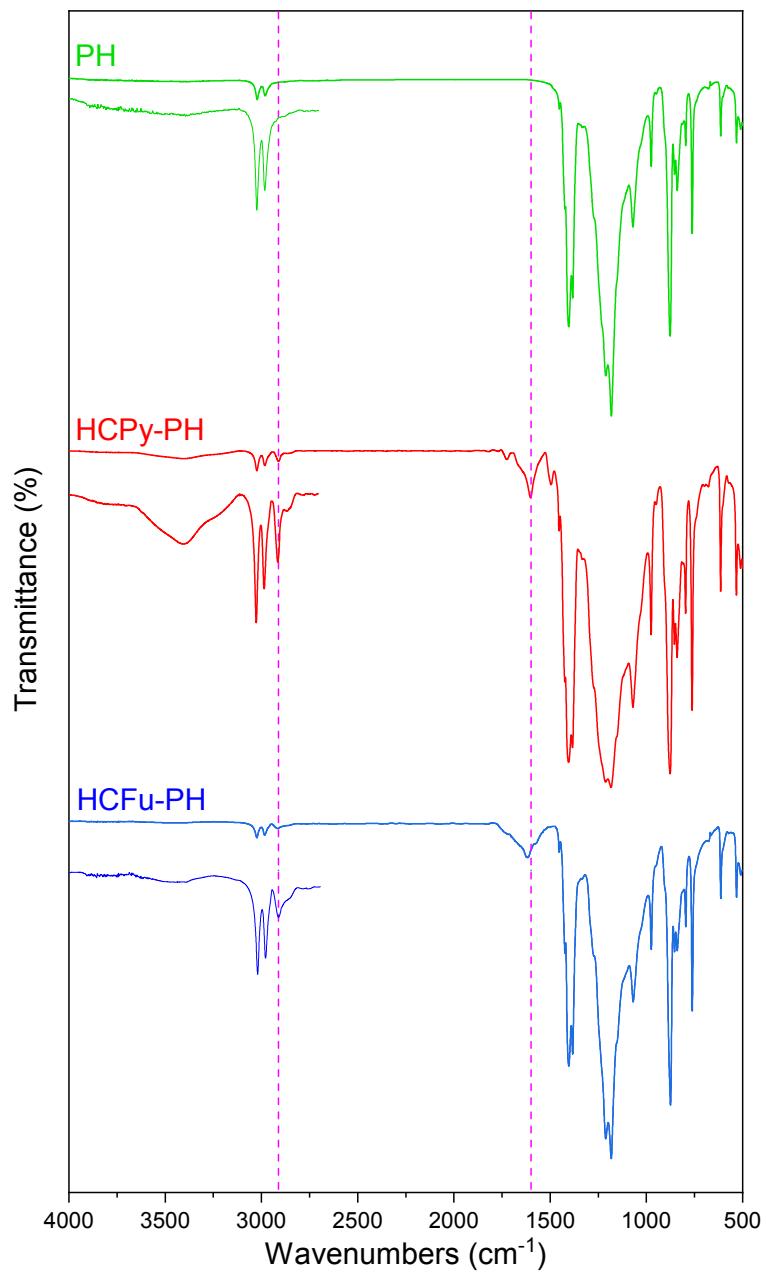


Figure S4. FTIR spectra of PH, HCPy-PH, and HCFu-PH membranes

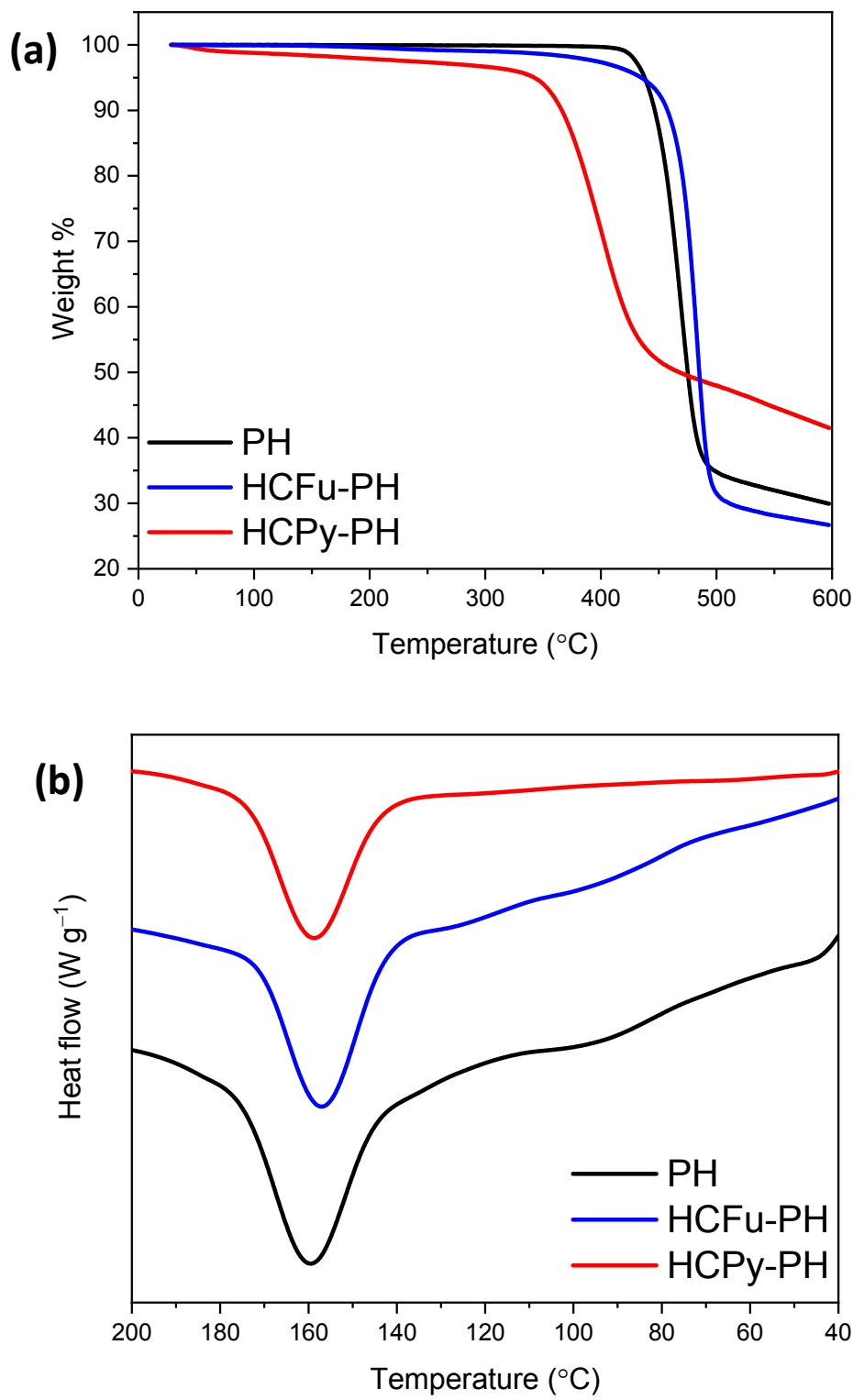


Figure S5. (a) TGA and (b) DSC curves of PH, HCPy-PH, and HCFu-PH membranes.

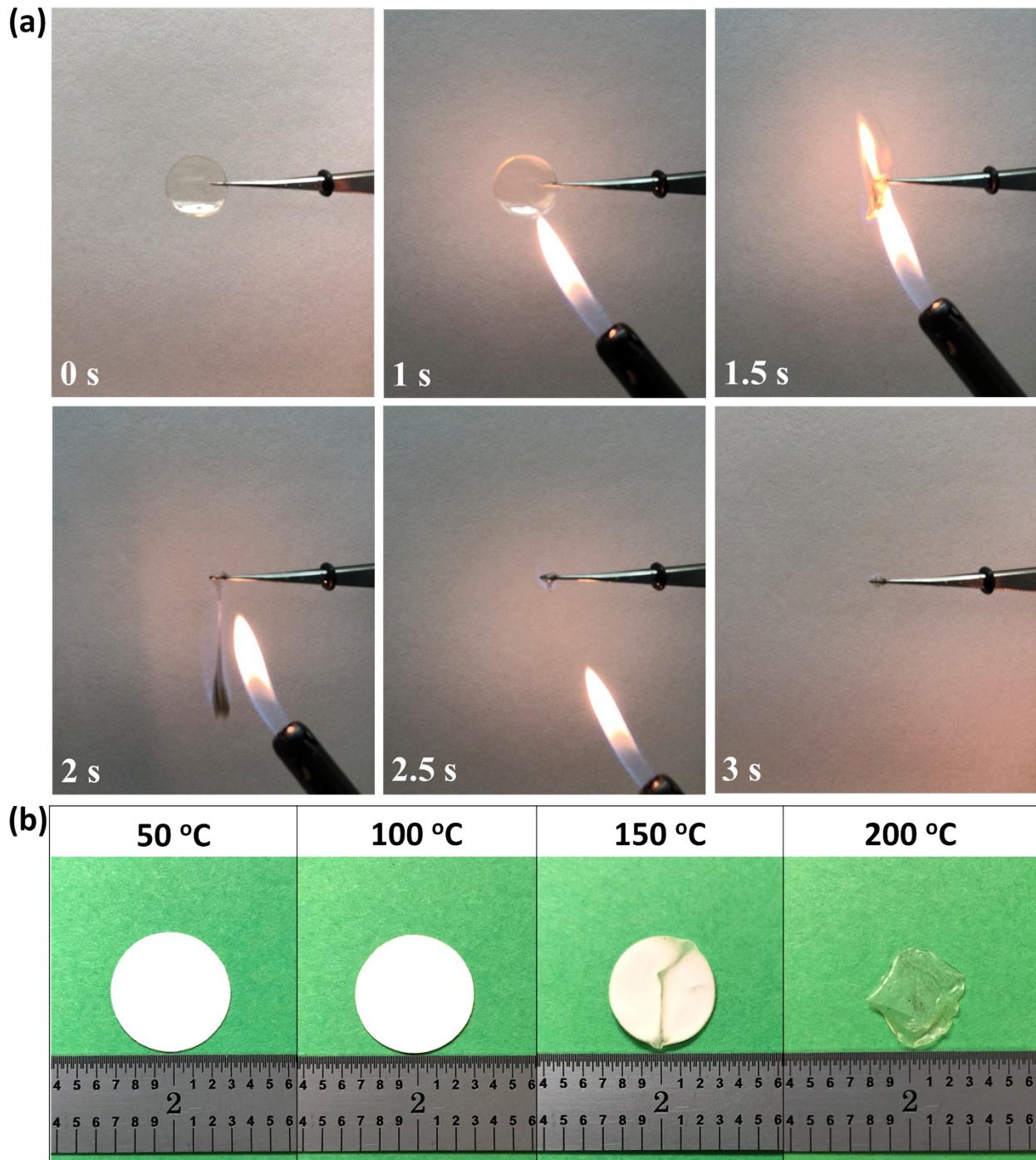


Figure S6. (a) Flame and (b) shrinkage tests for PH membrane.

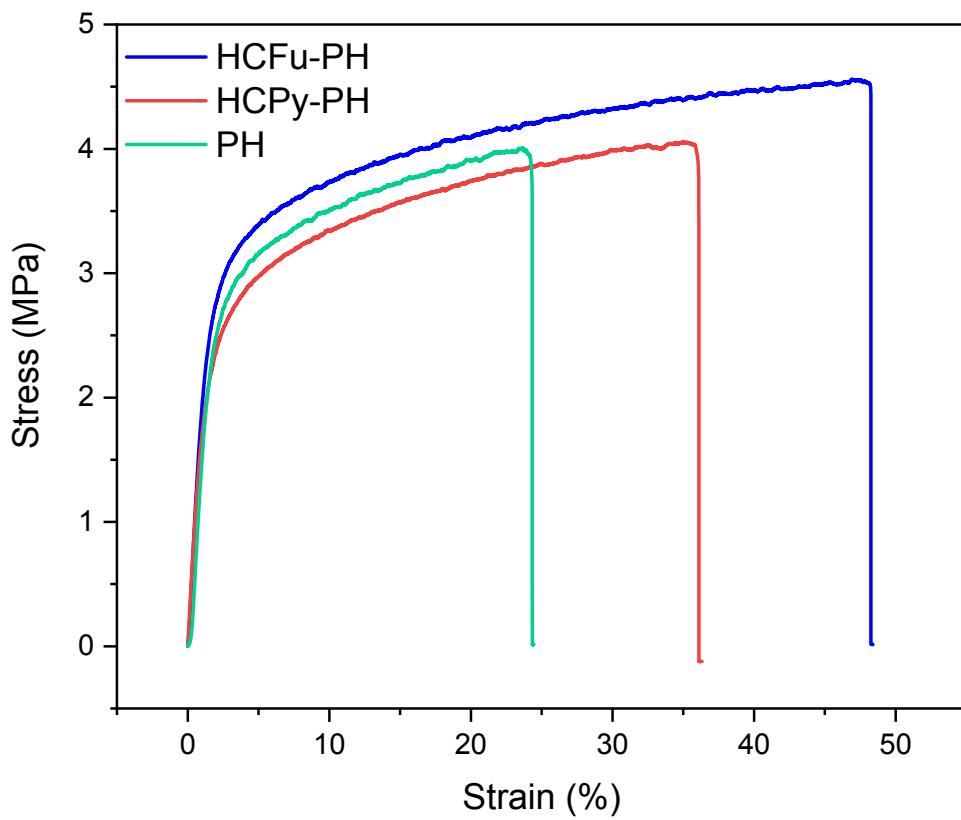


Figure S7. Stress-strain curves of PH, HCPy-PH, and HCFu-PH membranes.

Table S1. The performance comparison of recently reported GPEs for sodium and lithium batteries.

Cathode	Liquid Electrolyte	Anode	Polymer Electrolyte/Separator	Conductivity (S/cm)	ΔV	Reported full-cell performance	Ref.
LiFePO ₄	LiPF ₆ in EC/DMC/EMC	Li	PVDF-LiPVAOB ^a	2.60×10^{-4}	4.8	25 cycles at 0.2 C	1
LiNi _{0.5} Co _{0.2} Mn _{0.3} O ₂	LiTFSI in Dimethyl Sulfoxide	Li	Cellulose Membrane	6.34×10^{-3}	4.6	50 cycles at 0.2 C (90% C.R.) ^h	2
LiFePO ₄	LiPF ₆ in EC/DMC	Li	P(EGDA-co-VC)/PVDF-HFP ^b	1.49×10^{-3}	4.2	100 cycles at 0.3 C	3
Li _{1.18} Co _{0.15} Ni _{0.15} Mn _{0.52} O ₂	LiPF ₆ in EC/DMC/EMC	Li	BN ^c -(PVDF-HFP)	4.10×10^{-4}	—	300 cycles at 0.5 C (79% C.R.)	4
LiFePO ₄	LiPF ₆ in EC/DMC	Li	(PEGDE+DEBA+DPPO) ^d @PVDF-HFP	2.36×10^{-3}	4.2	200 cycles at 0.3 C (99.3% C.R.)	5
LiFePO ₄	LiPF ₆ in EC/DMC/EMC/DEC/2 wt% VC	Li	Al-doped Li _{6.75} La ₃ Zr _{1.75} Ta _{0.25} O ₁₂ @PVDF-HFP	0.74×10^{-3}	—	500 cycles at 2 C	6
LiFePO ₄	LiPF ₆ in EC/DEC	Li	PVDF-HFP- Li ₇ La ₃ Zr ₂ O ₁₂	3.71×10^{-4}	4.65	200 cycles at 0.2 C (83.8% C.R.)	7
LiCoO ₂	LiPF ₆ in EC/DMC	Li	PEO/PMMA/P(VDF-HFP)/SiO ₂	2.02×10^{-3}	5.3	100 cycles at 0.1 mA/cm ² (99% C.R.)	8
Na ₃ V ₂ (PO ₄) ₂ F ₃	NaPF ₆ in PC	Na	PEO/PMMA/P(VDF-HFP)/SiO ₂	8.80×10^{-4}	4.9	100 cycles at 0.1 mA/cm ² (93% C.R.)	8
Na ₃ V ₂ (PO ₄) ₃	NaClO ₄ in EC/PC/ 5 wt% FEC	Na	Cross-linked MATEMP ^e	5.13×10^{-3}	5	1000 cycles at 1 C (81.6% C.R.)	9
Na _{0.44} MnO ₂	NaClO ₄ in PC/ 2 wt% FEC	Na	PVDF-HFP	1.91×10^{-3}	4	20 cycles at 0.1 C	10
Na ₃ V ₂ (PO ₄) ₃	NaPF ₆ in EC/DMC/ 2 wt% FEC	Na	(cross-linked PEGDE-DPPO) @GF ^f	2.18×10^{-3}	4.8	2000 cycles at 1 C (95% C.R.)	11
Na ₃ V ₂ (PO ₄) ₃	NaClO ₄ in EC/PC/ 5 wt% FEC	Na	Cross-linked MATEPP ^g	6.29×10^{-3}	4.9	10000 cycles at 5 C (69.2% C.R.)	12
LiFePO ₄	LiPF ₆ in EC/DEC	Li	HCFu-PH	6.4×10^{-3}	4.7	1000 cycles at 1 C (78% C.R.)	This work
Na ₃ V ₂ (PO ₄) ₃	NaClO ₄ in PC/ 5 wt% FEC	Na	HCPy-PH	4.3×10^{-3}	4.5	1000 cycles at 1 C (94% C.R.)	This work

- a. Lithium polyvinyl alcohol oxalate borate (LiPVAOB).
- b. Poly (ethylene glycol) diacrylate-co-poly (vinylene carbonate) and PVDF-HFP.
- c. Boron nitride (BN).
- d. Cross-linked GPE based on poly (ethylene glycol) diglycidyl ether (PEGDE), diglycidyl ether of bisphenol-A (DEBA), and diamino-poly (propylene oxide) (DPPO).
- e. Di(2-methylacryloyltriethoxyethyl) methyl phosphonate (MATEMP)
- f. Cross-linked poly(ethylene glycol) diglycidyl ether (PEGDE) and diamino-poly (propylene oxide) (DPPO) in the network of glass fiber (GF) membrane
- g. Di(2-methacryloyltriethoxyethyl)phenylphosphonate (MATEPP)
- h. C.R.: Capacity retention

References:

1. Y. Zhu, S. Xiao, Y. Shi, Y. Yang, Y. Hou and Y. Wu, 2014, **4**, 1300647.
2. Z. Du, Y. Su, Y. Qu, L. Zhao, X. Jia, Y. Mo, F. Yu, J. Du and Y. Chen, *Electrochimica Acta*, 2019, **299**, 19-26.
3. Q. Lu, J. Yang, W. Lu, J. Wang and Y. Nuli, *Electrochimica Acta*, 2015, **152**, 489-495.
4. X. Bian, J. Liang, X. Tang, R. Li, L. Kang, A. Su, X. Su and Y. Wei, *Journal of Alloys and Compounds*, 2019, **803**, 1075-1081.
5. Q. Lu, Y.-B. He, Q. Yu, B. Li, Y. V. Kaneti, Y. Yao, F. Kang and Q.-H. Yang, 2017, **29**, 1604460.
6. X. Shi, Q. Sun, B. Boateng, Y. Niu, Y. Han, W. Lv and W. He, *Journal of Power Sources*, 2019, **414**, 225-232.
7. Y. F. Liang, S. J. Deng, Y. Xia, X. L. Wang, X. H. Xia, J. B. Wu, C. D. Gu and J. P. Tu, *Materials Research Bulletin*, 2018, **102**, 412-417.
8. J. Shi, H. Xiong, Y. Yang and H. Shao, *Solid State Ionics*, 2018, **326**, 136-144.
9. J. Zheng, X. Liu, Y. Duan, L. Chen, X. Zhang, X. Feng, W. Chen and Y. Zhao, *Journal of Membrane Science*, 2019, **583**, 163-170.
10. D. T. Vo, H. N. Do, T. T. Nguyen, T. T. H. Nguyen, V. M. Tran, S. Okada and M. L. P. Le, *Materials Science and Engineering: B*, 2019, **241**, 27-35.
11. Q. Yu, Q. Lu, X. Qi, S. Zhao, Y.-B. He, L. Liu, J. Li, D. Zhou, Y.-S. Hu, Q.-H. Yang, F. Kang and B. Li, *Energy Storage Materials*, 2019, DOI: <https://doi.org/10.1016/j.ensm.2019.03.011>.
12. J. Zheng, Y. Zhao, X. Feng, W. Chen and Y. Zhao, *Journal of Materials Chemistry A*, 2018, **6**, 6559-6564.