

Supplementary Information

An advanced and highly efficient Ce assisted NiFe-LDH electrocatalysts for overall water splitting

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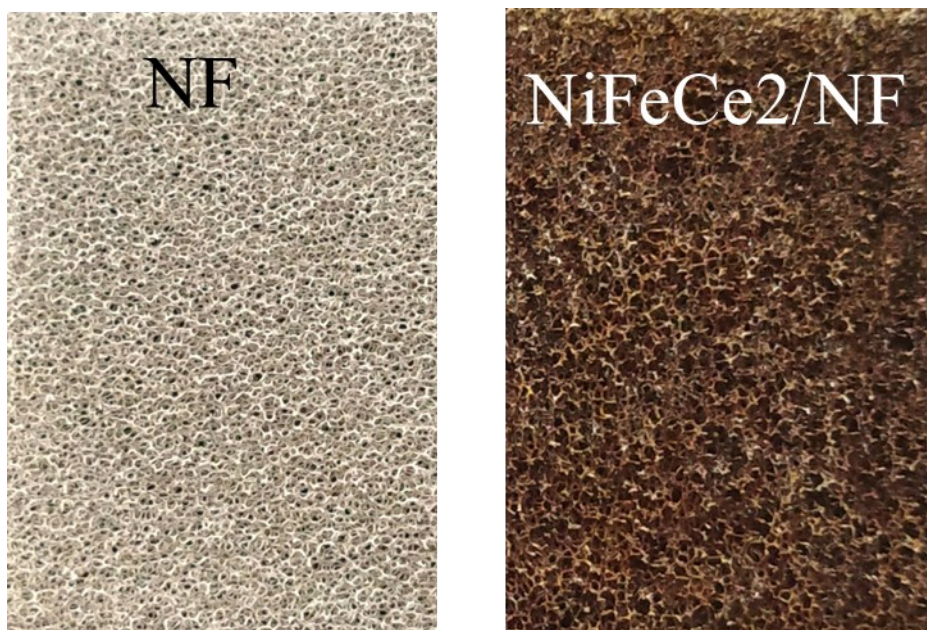


Figure S1. Photographs of (a) bare cleaned Nickel foam (NF) and (b) electrocatalysts deposited NF.

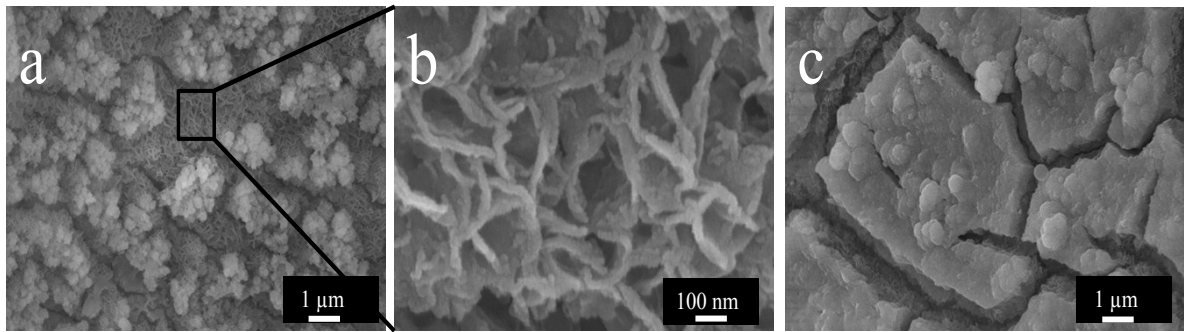


Figure S2. The FE-SEM images of (a) NiFe after 5 minute (b) its high magnified image (c) NiFe after 10 minute deposition time.

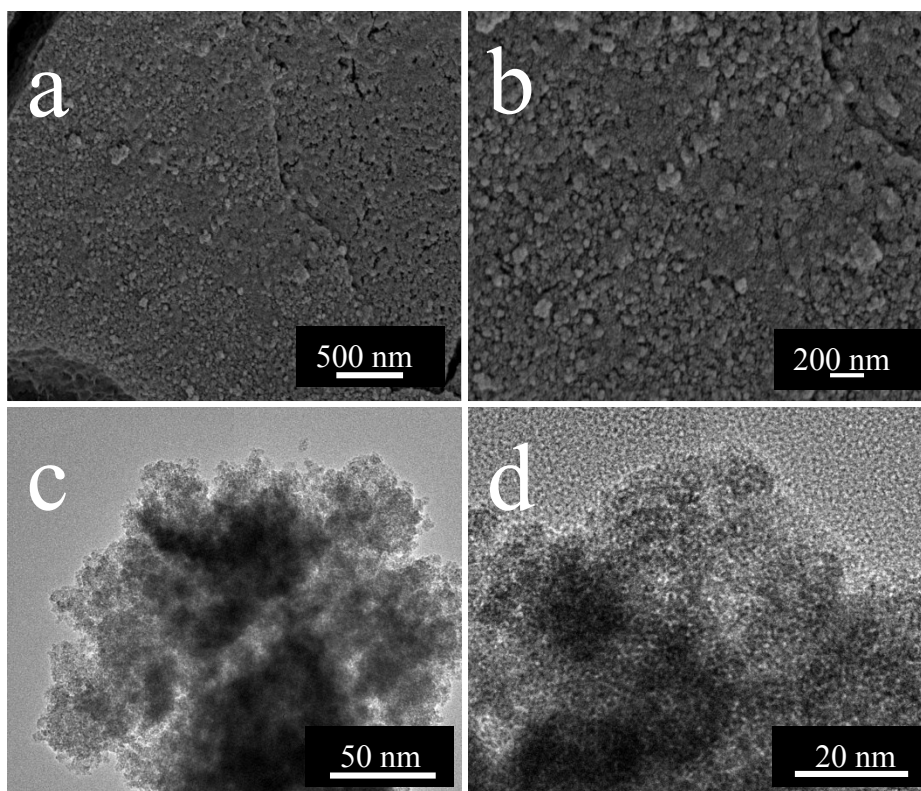


Figure S3 (a-b) FE-SEM images (c-d) TEM images of NiFe electrocatalysts at different magnification.

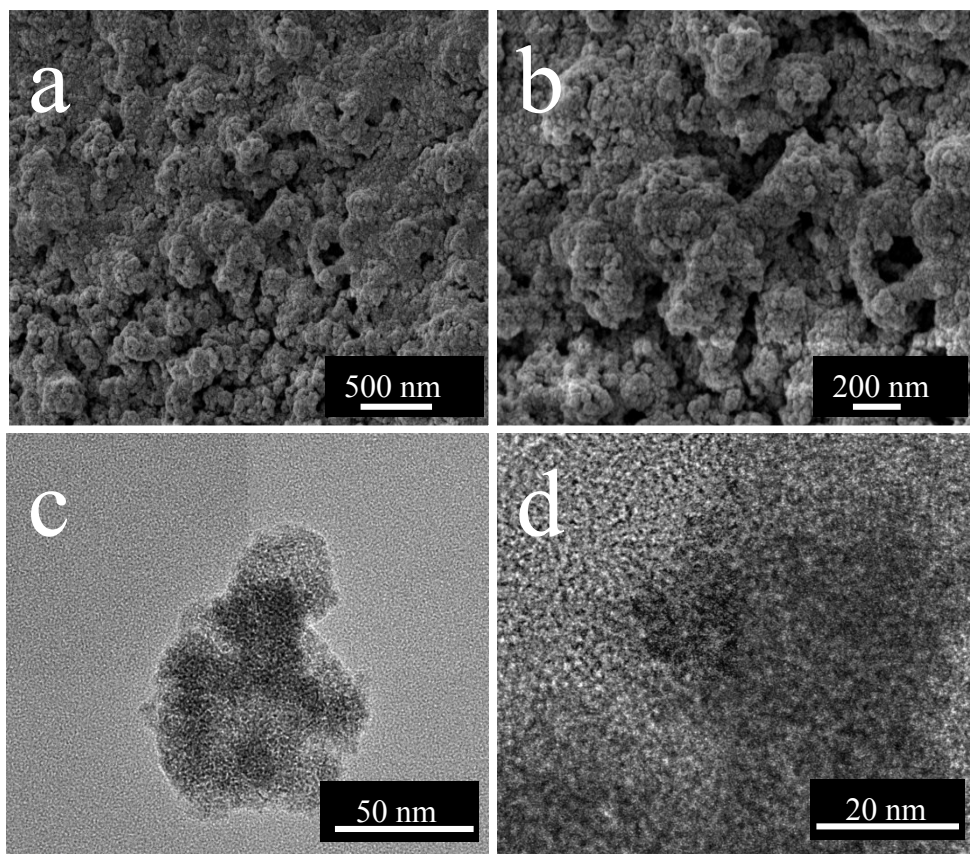


Figure S4: (a-b) FE-SEM images (c-d) TEM images of NiFe_{0.9}Ce_{0.1} electrocatalysts at different magnification.

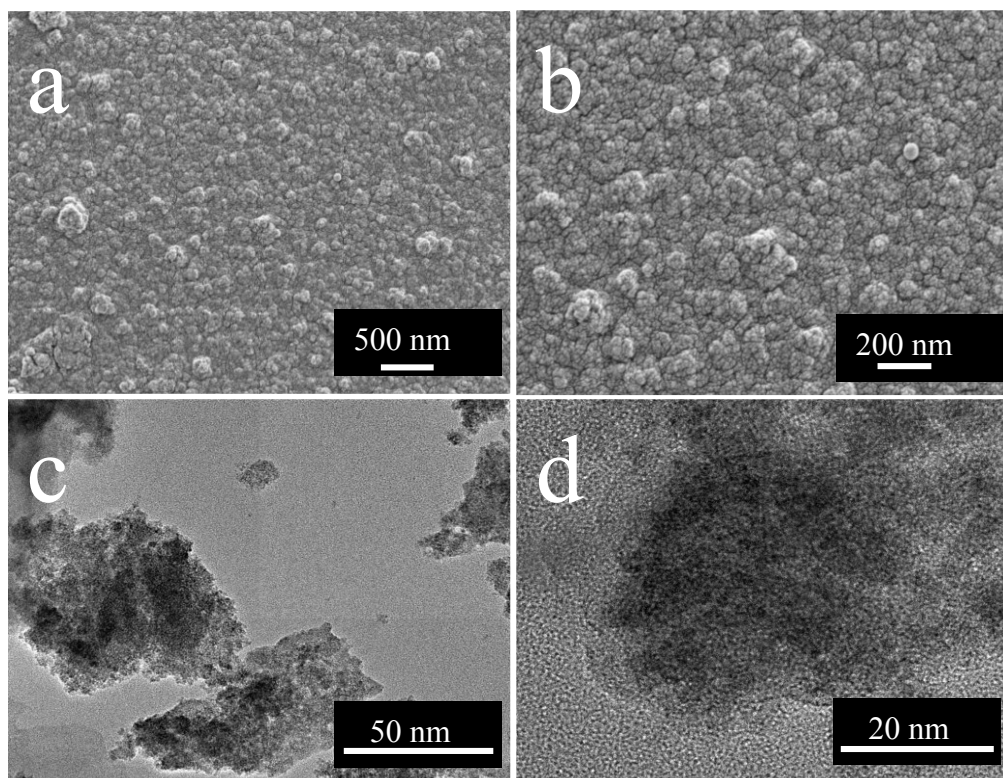


Figure S5: (a-b) FE-SEM images (c-d) TEM images of NiFe_{0.7}Ce_{0.3} electrocatalysts at different magnification.

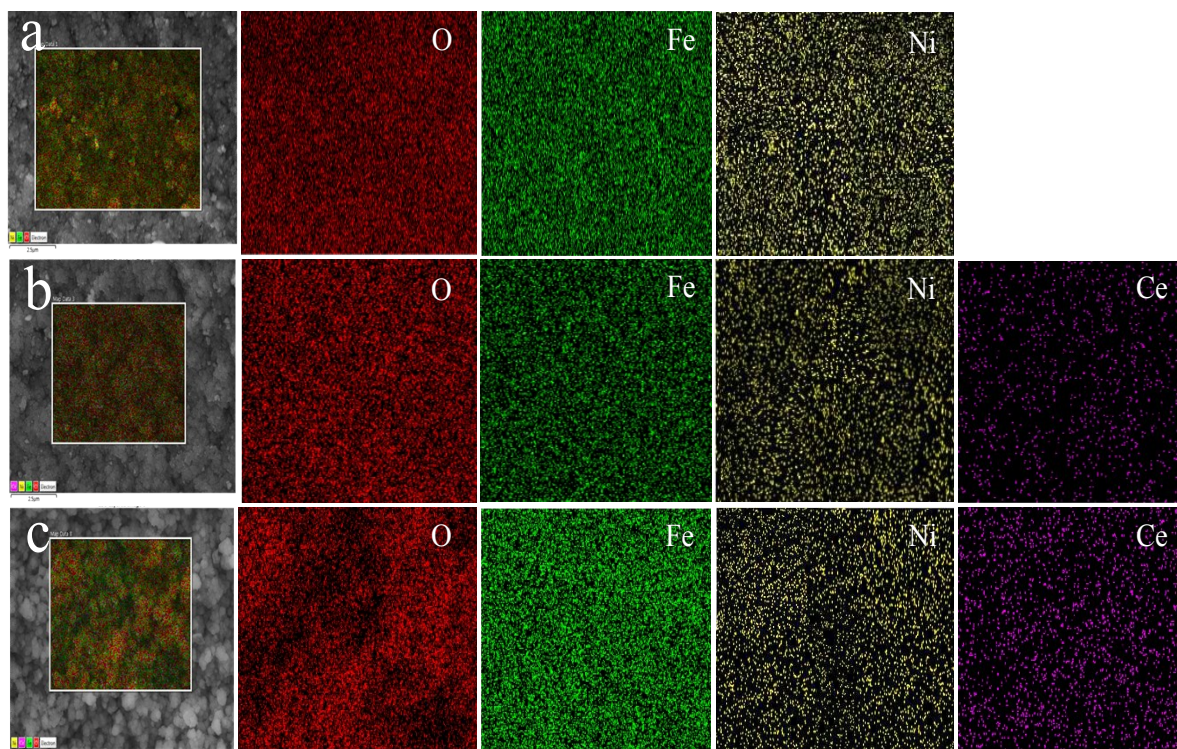


Figure S6: FE-SEM EDS elemental colour mapping for (a) NiFe (b) NiFe_{0.8}Ce_{0.2} (c) NiFe_{0.7}Ce_{0.3}.

Table S1: The At% of Ni, Fe, O and Ce elements in electrocatalysts investigated in present study.

Electrocatalysts	Ni(At%)	Fe(At%)	O(At%)	Ce(At%)
NiFe	19.10	22.60	58.30	-
NiFe _{0.9} Ce _{0.1}	18.80	21.72	58.70	0.78
NiFe _{0.8} Ce _{0.2}	18.60	20.95	59.00	1.45
NiFe _{0.7} Ce _{0.3}	18.40	19.90	59.30	2.40

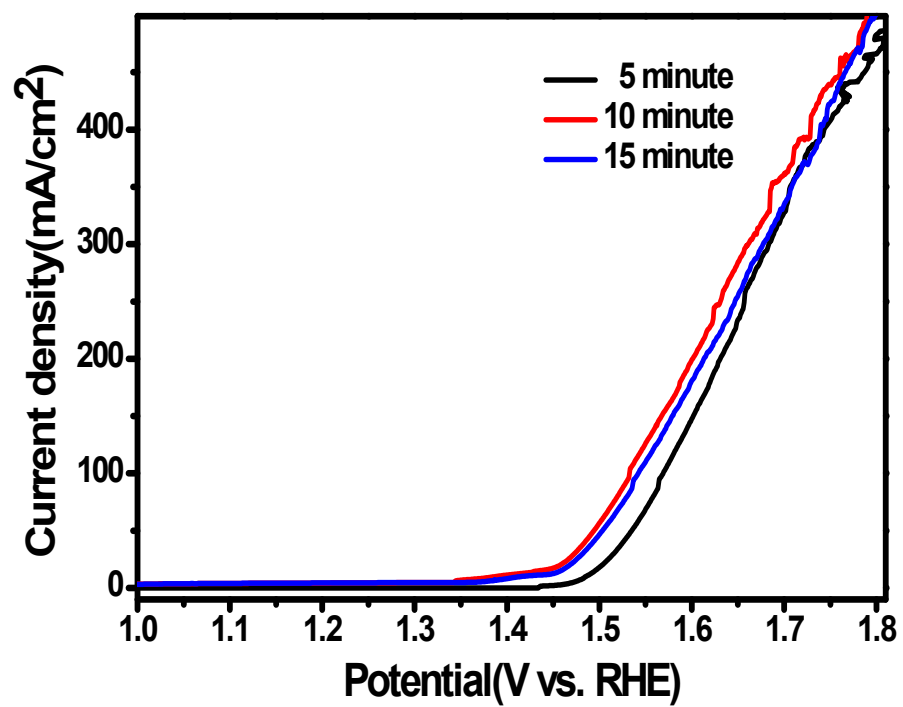


Figure S7: Polarization curves of NiFe electrocatalysts electrodeposited for different time.

Table S2: The OER overpotential required to achieve different current densities for various electrocatalysts tested in present study.

Electrocatalysts	OER activity (@10 mA/cm²)	OER activity (@50 mA/cm²)	OER activity (@100 mA/cm²)
NiFe	197	260	304
NiFe _{0.9} Ce _{0.1}	185	254	284
NiFe _{0.8} Ce _{0.2}	175	245	273
NiFe _{0.7} Ce _{0.3}	205	279	314
RuO ₂ /C	230	320	390

Table S3: The literature survey of various reported electrocatalysts for OER catalytic activity (@10 mA/cm²).

Electrocatalysts	Support*	OER activity (@10 mA/cm²) mV	Stability (Hour)	Reference
CoSe/NiFe	GF	250	10	10
Exfoliated NiFe-LDH	GC	290	12	31
NiFe-LDH nanosheets	NF	200	10	39
NiFe-LDH array	NF	224	50	44
NiFe-LDH/graphene	NF	205	1.5	50
Ni ₂ /3Fe ₁ /3-rGO	GC	210	10	51
Cu@NiFe-LDH	GF	199	48	52
Exfoliated NiFe-LDH/Defective graphene	GC	210	10	53
NiFe-OH/NiFeP	NF	199	24	54
NiFe-LDH	NF	197	50	Present work
NiFe_{0.8}Ce_{0.2}	NF	175	50	Present work

*Note: Nickel foam=NF

Glassy carbon disc electrode=GC

Exfoliated graphene foil=GF

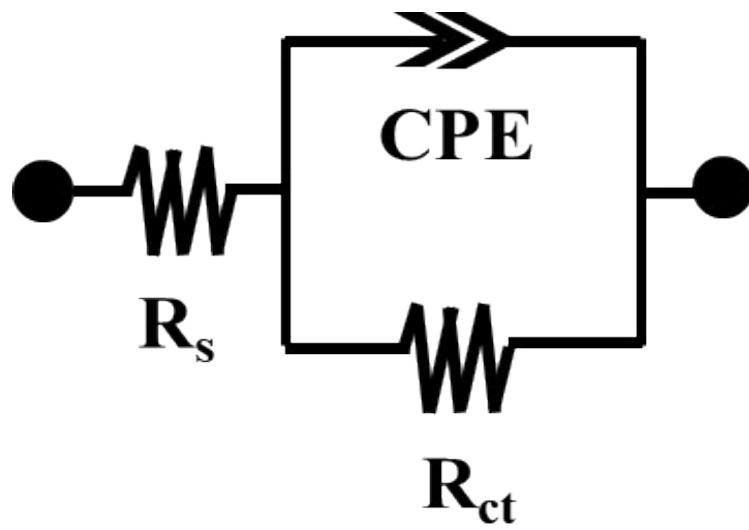


Figure S8: The equivalent circuit fitted for Nyquist plots.

Table S4: The summary of series resistance (R_s), and charge transfer resistance (R_{ct}) obtained after fitted the Nyquist plots.

Electrocatalysts	R_s (Ω)	R_{ct} (Ω)
NiFe	1.15	3.25
NiFe _{0.9} Ce _{0.1}	1.10	2.25
NiFe _{0.8} Ce _{0.2}	1.05	2.05
NiFe _{0.7} Ce _{0.3}	1.15	4.00
NF	1.40	150

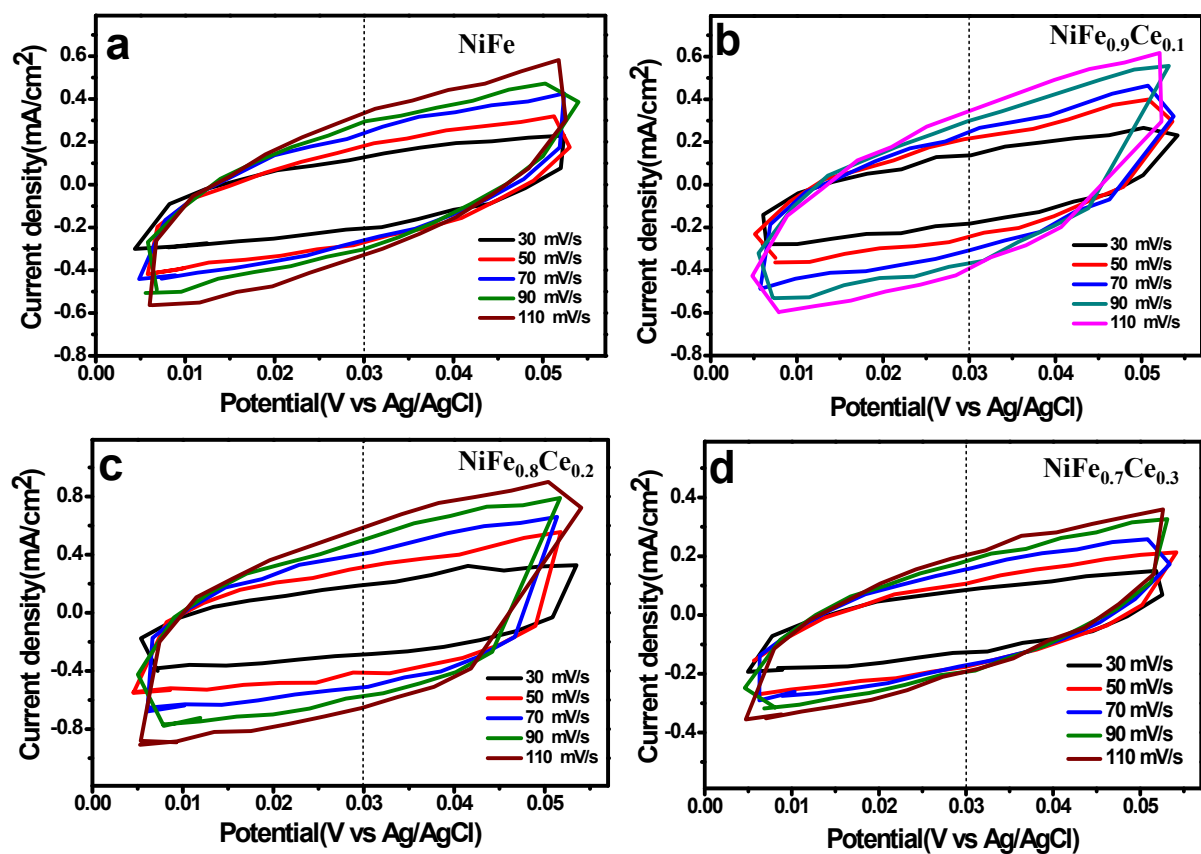


Figure S9: CV curves for different electrocatalysts carried out in non-faradic region at different scan rate in 1M KOH.

Table S5: The summarized double layer capacitance (DLC), electrochemical surface area (ECSA), roughness factor (RF) and active sites for various NiFe-based electrocatalysts tested in present study.

Electrocatalysts	DLC (mF/cm²)	ECSA (cm²)	RF (cm²)	Active sites Moles(10⁻⁵)
NiFe	2.35	58.75	235	7.66
NiFe _{0.9} Ce _{0.1}	2.60	65.00	260	10.13
NiFe _{0.8} Ce _{0.2}	4.00	100	400	14.04
NiFe _{0.7} Ce _{0.3}	1.60	40	160	4.43

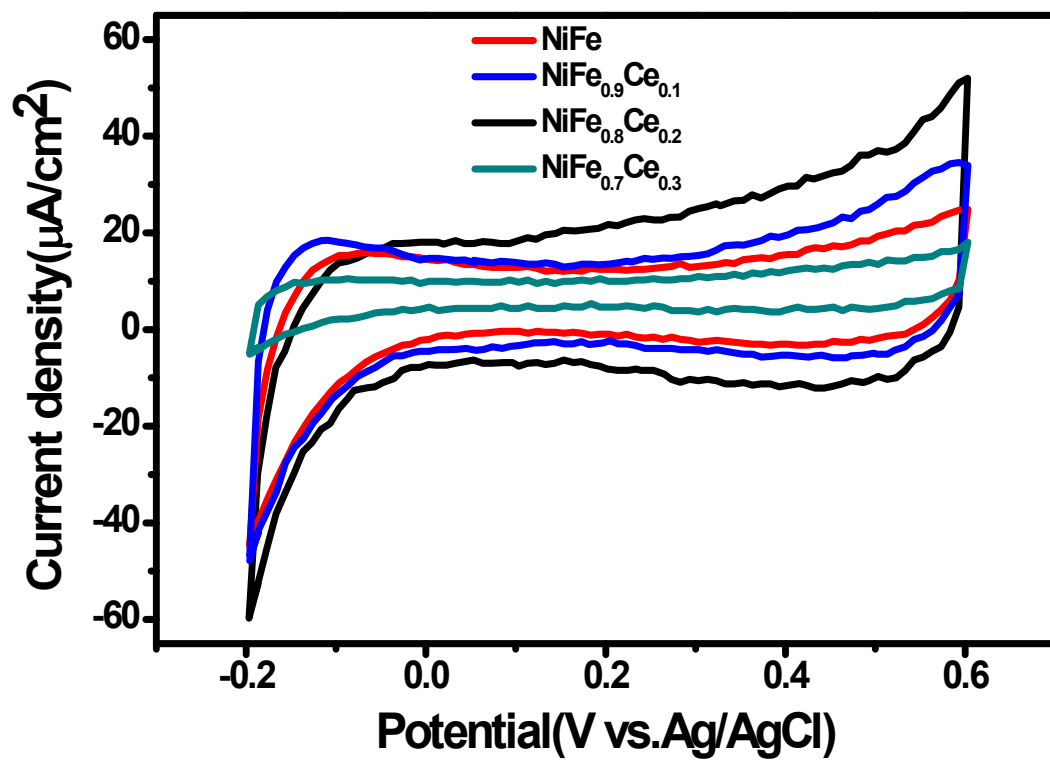


Figure S10: The CVs for various electrocatalysts were carried out in phosphate buffer solution (pH=7) at the scan rate of 50 mV.

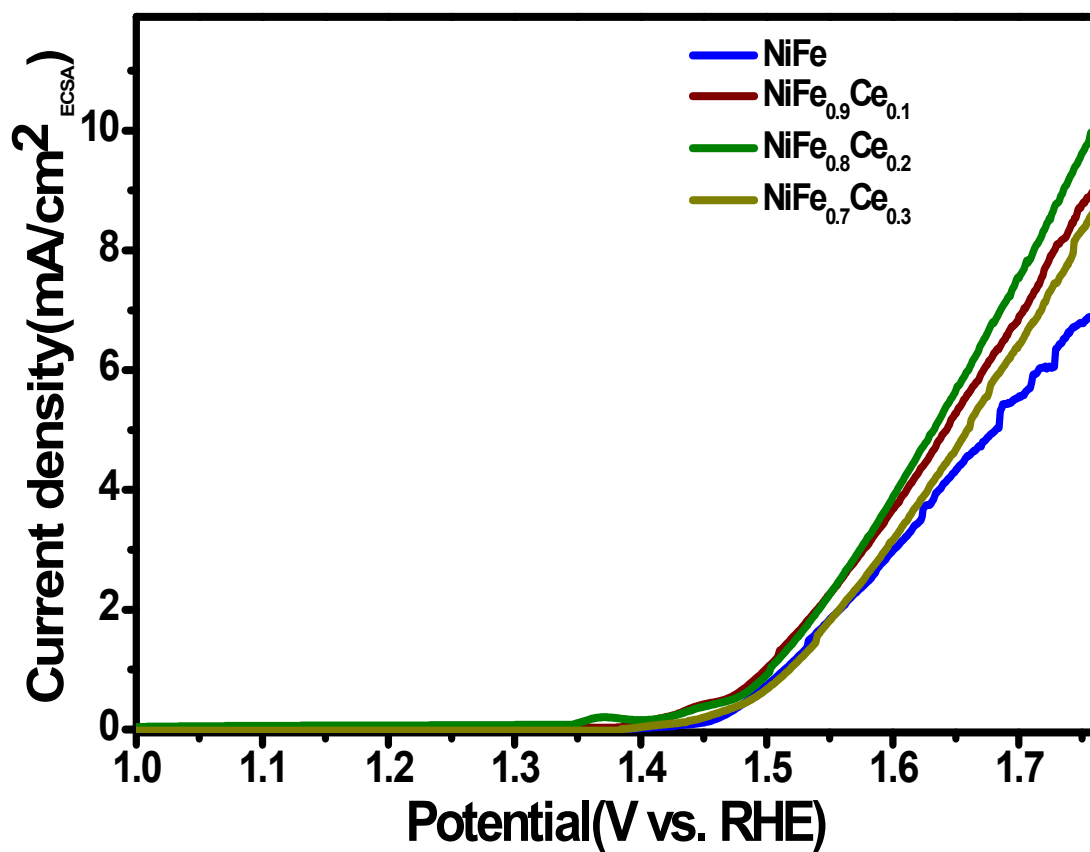


Figure S11: OER polarization curves with current density normalized by the calculated ECSA.

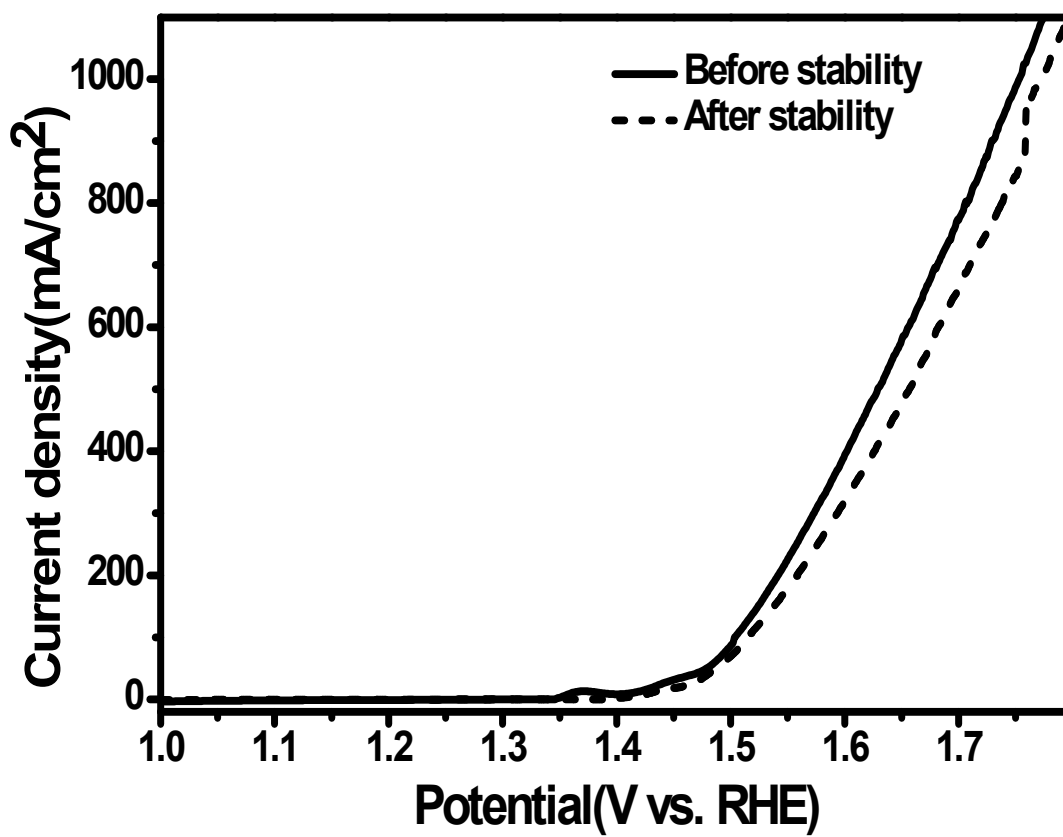


Figure S12: Polarization curves of NiFe_{0.8}Ce_{0.2} electrocatalysts before and after 50 hour OER test.

Table S6: The HER overpotential required to achieve different current densities for various electrocatalysts tested in present study.

Electrocatalysts	HER activity (@10 mA/cm²)	HER activity (@50 mA/cm²)	HER activity (@100 mA/cm²)
NiFe	175	298	369
NiFe _{0.9} Ce _{0.1}	167	257	321
NiFe _{0.8} Ce _{0.2}	147	254	313
NiFe _{0.7} Ce _{0.3}	195	296	374
Pt/C	55	155	209

Table S7: The literature survey of various reported NiFe-based electrocatalysts for HER catalytic activity (@10 mA/cm²).

Electrocatalysts	Support*	HER activity (@10 mA/cm²) mV	Stability (Hour)	Reference
EG/Co _{0.85} Se/NiFe-LDH	GF	265	10	10
CoSe/NiFe-LDH	GF	260	-	10
NiFe-LDH NS/DG10	GC	210	5.5	53
NiFe/NiCo ₂ O ₄	NF	105	10	56
Ni ₃ S ₂ /NF	NF	223	150	58
Cu@CoFe-LDH	CF	171	30	59
Ni _{0.75} Fe _{0.125} V _{0.125}	NF	125	15	60
NiCo ₂ O ₄ /NiFe-LDH	NF	192	10	61
NiFe-LDH	NF	175	50	Present work
NiFe_{0.8}Ce_{0.2}	NF	147	50	Present work

*Note: Nickel foam=Nf

Copper foam=CF

Glassy carbon disc electrode=GC

Exfoliated graphene=GF

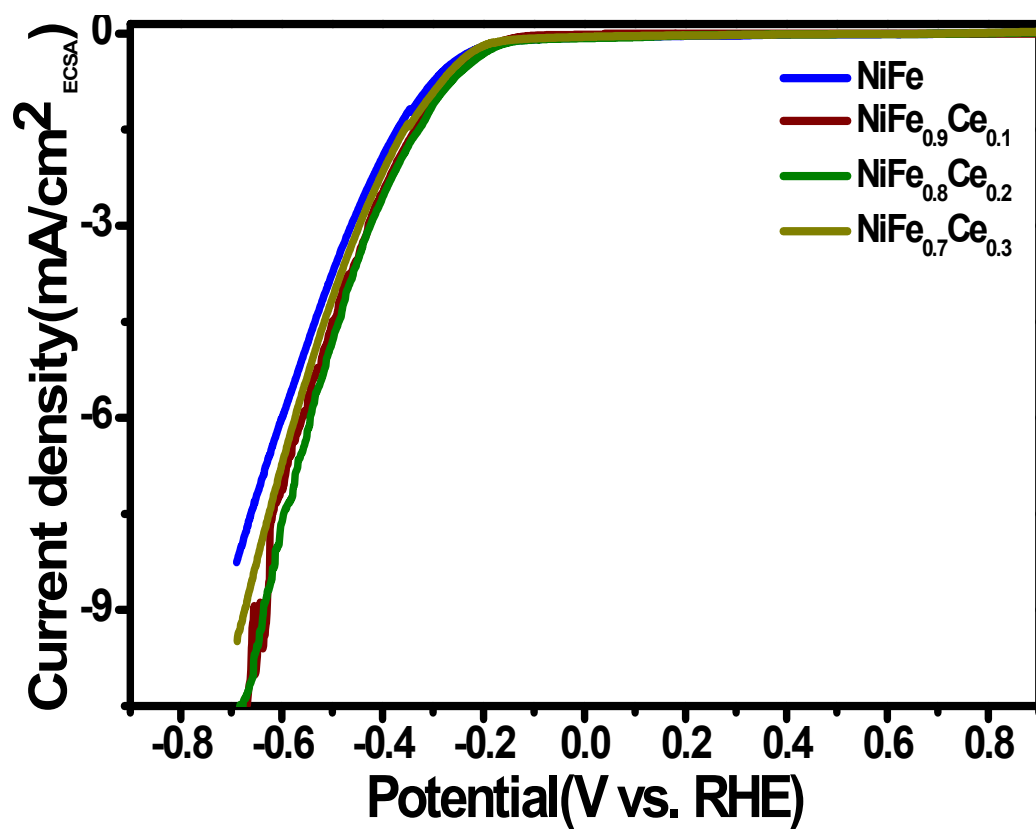


Figure S13: HER polarization curves with current density normalized by the calculated ECSA.

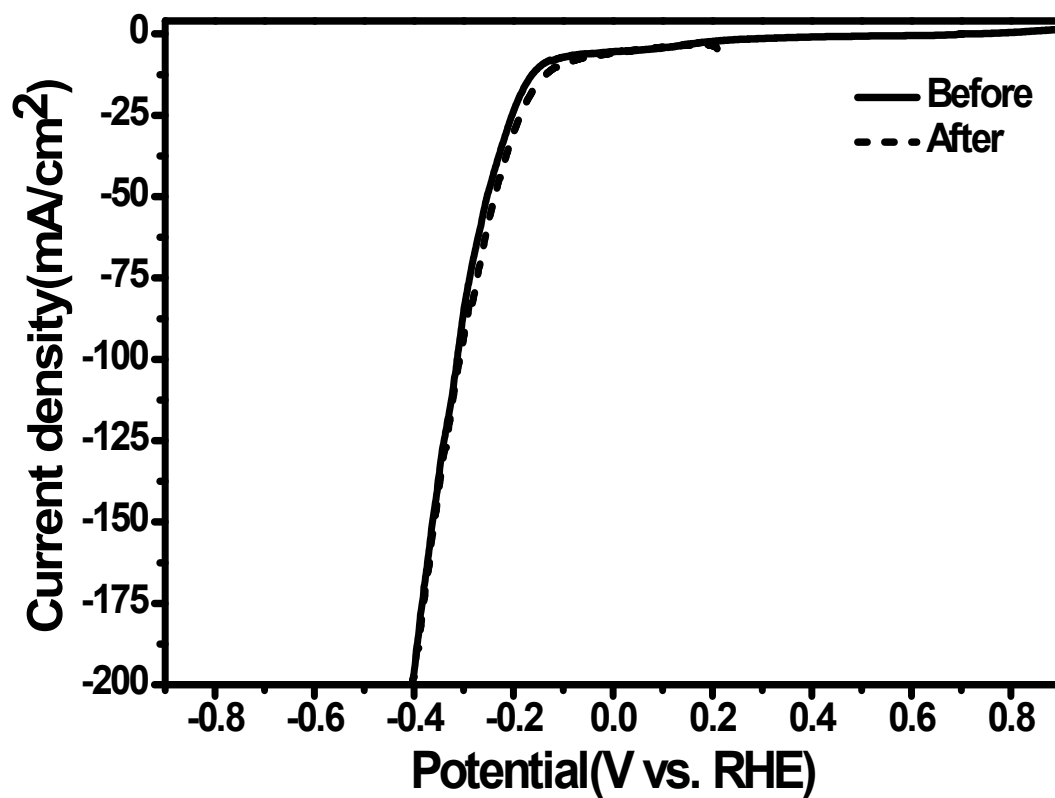


Figure S14: Polarization curves of NiFe_{0.8}Ce_{0.2} electrocatalysts before and after 50 hour HER test.

Table S8: The literature survey of various reported electrocatalysts for overall electrochemical water splitting (Overpotential reported @10 mA/cm²)

Electrocatalysts	Support*	Overpotential (@10 mA/cm²) mV	Stability (Hour)	Reference
EG/Co _{0.85} Se/NiFe-LDH	NF	1.65	10	10
Cu@NiFe-LDH	CF	1.54	48	52
NiFe/NiCo ₂ O ₄	NF	1.60	12	61
Ni _{0.75} Fe _{0.125} V _{0.125}	NF	1.59	15	62
NiSe nanowires	NF	1.63	20	63
NiFe LDH	NF	1.70	3	64
NiMo HRNs	NF	1.64	10	65
Ni ₃ Se ₂	CF	1.65	12	66
Fe doped CoP	TF	1.60	40	67
NiCoP	NF	1.58	24	68
NiFe_{0.8}Ce_{0.2}	NF	1.59	50	Present work

*Note: Nickel foam=Nf

Copper foam=CF

Ti foil=TF

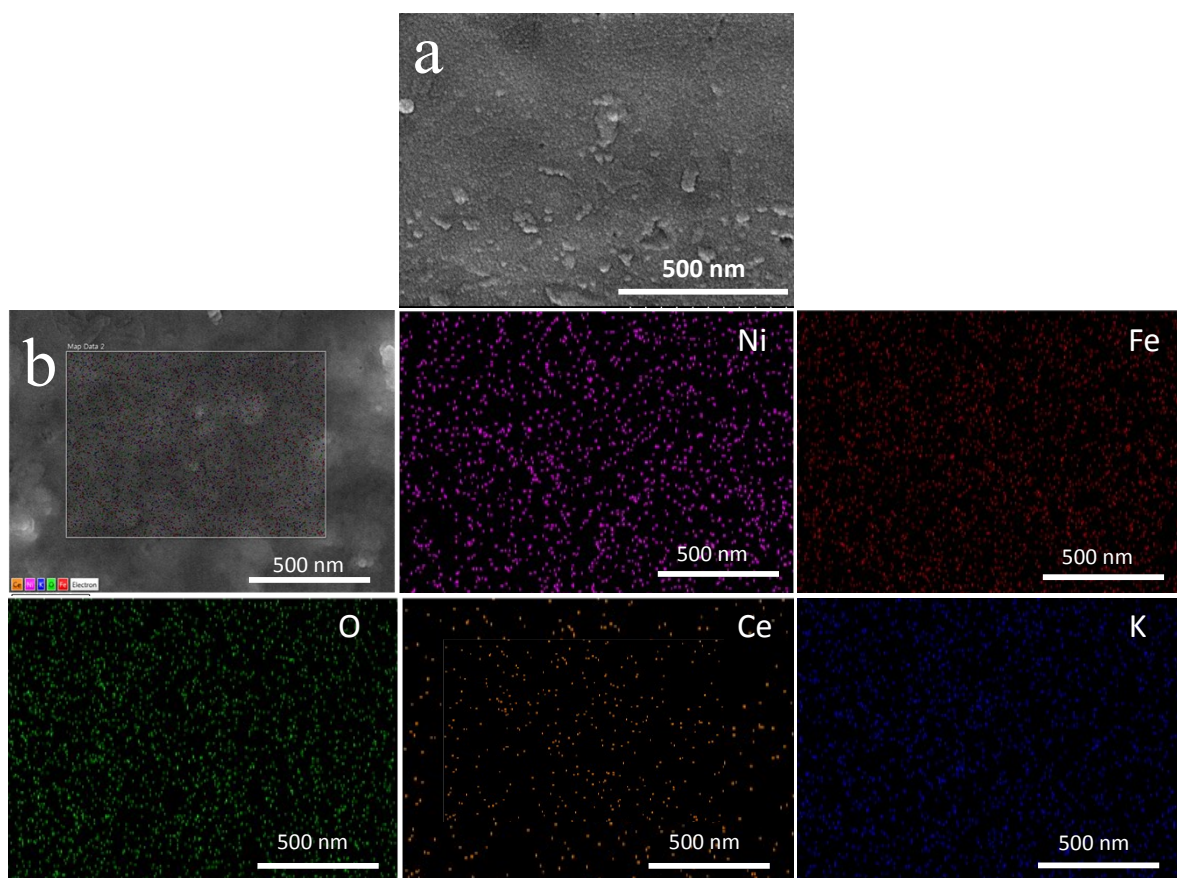


Figure S15: (a) FE-SEM image (b) EDS elemental colour mapping of anode carried out after 40 hour stability test (OER).

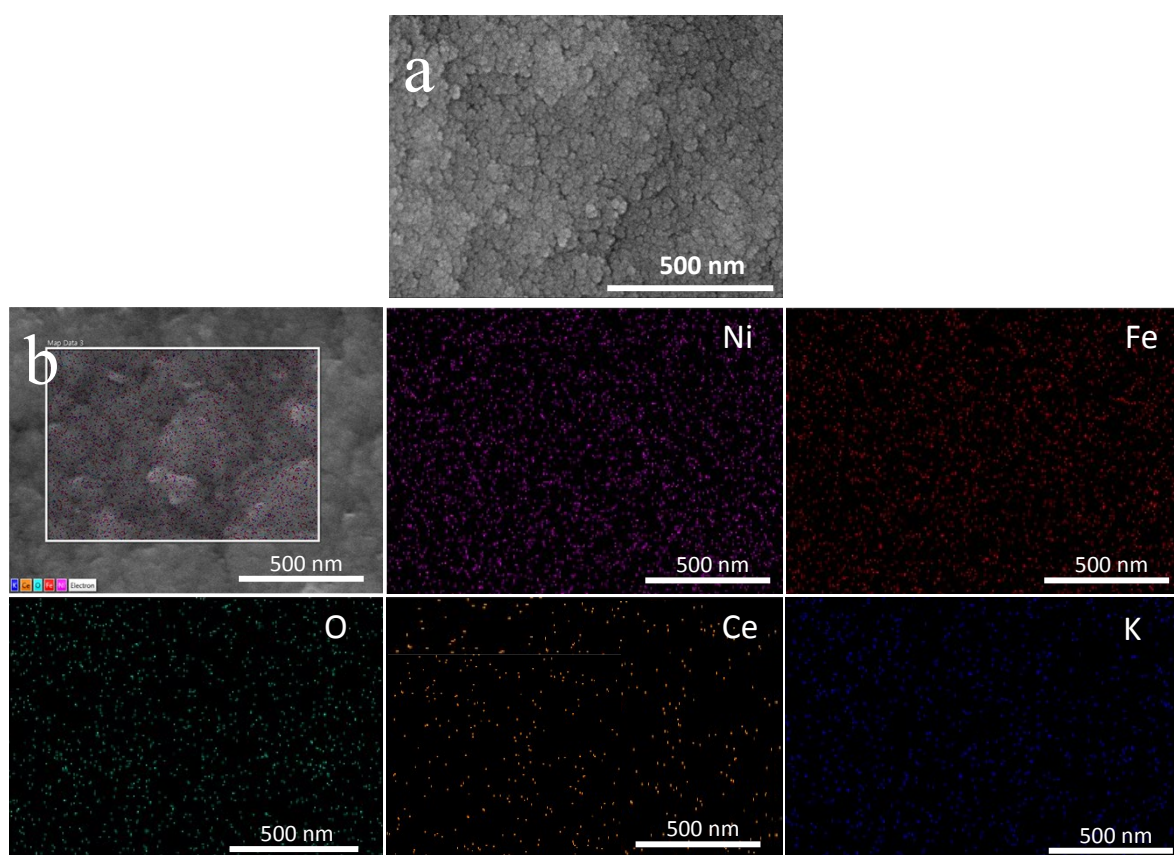


Figure S16: (a) FE-SEM image (b) EDS elemental colour mapping of cathode carried out after 40 hour stability test (HER).

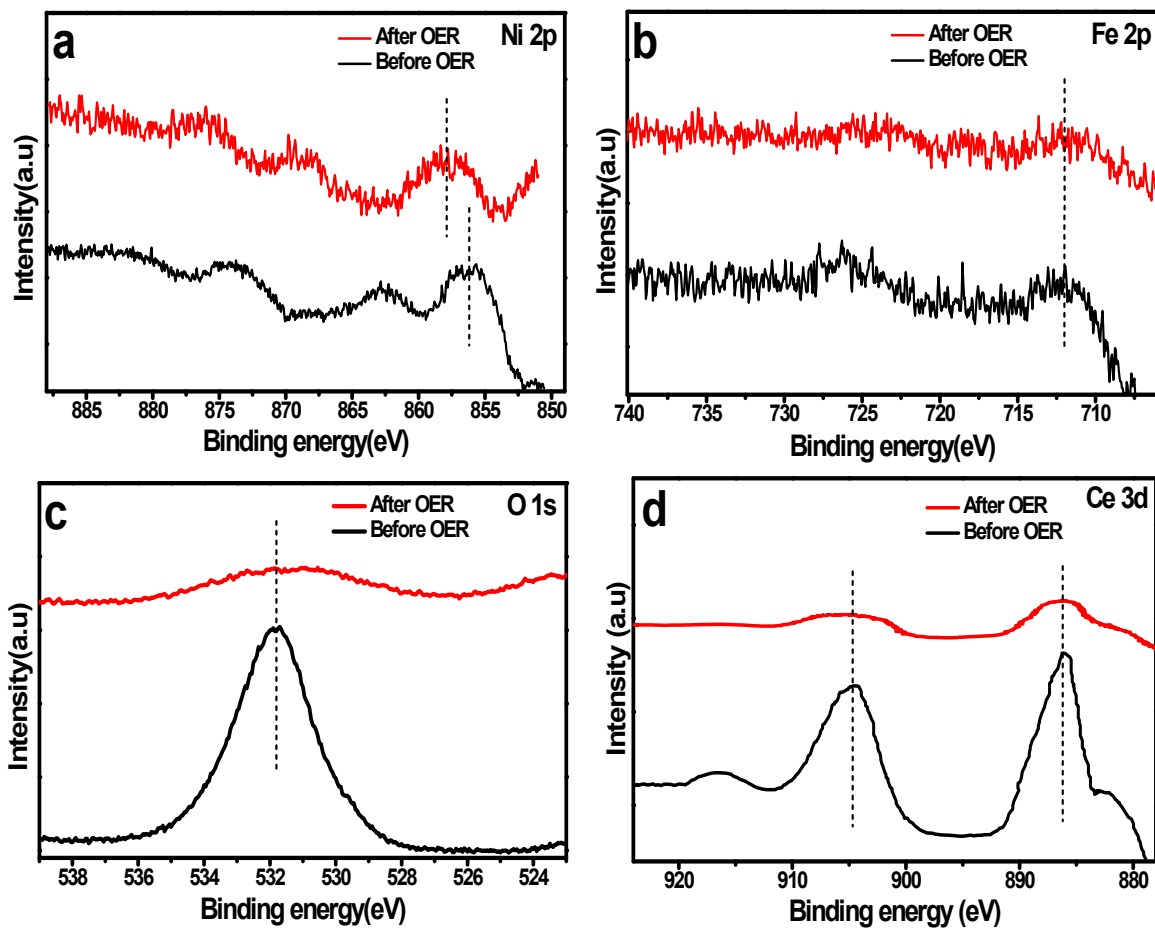


Figure S17: XPS analysis of $\text{NiFe}_{0.8}\text{Ce}_{0.2}$ electrocatalysts after 50 hour OER stability test.