Supporting Information

Elucidating the Unexpected Electrocatalytic Activity of Nanoscale PdO Layers on Pd Electrocatalyst Towards Ethanol Oxidation in Basic Solution

Atiweena Krittayavathananon^{†‡}, Salatan Duangdangchote^{†‡}, Narong Chanlek[#], Sethuraman Sathyamoorthi^{†‡}, Panvika Pannopard^{‡§}, Jumras Limtrakul^{†‡}, and Montree Sawangphruk^{*†‡}

[†]Department of Chemical and Biomolecular Engineering, School of Energy Science and Engineering, Vidyasirimedhi Institute of Science and Technology, Rayong 21210, Thailand

[‡]Center of Excellence for Energy Storage Technology (CEST), Vidyasirimedhi Institute of Science and Technology, Rayong 21210, Thailand

[§]Frontier Research Centre (FRC), Vidyasirimedhi Institute of Science and Technology, Rayong 21210, Thailand

Synchrotron Light Research Institute (Public Organization), Nakhon Ratchasima 30000, Thailand

*Corresponding author. E-mail address: <u>montree.s@vistec.ac.th</u> (M. Sawangphruk)



Figure S1. XRD pattern of FTO electrode.



Figure S2. Pd 3d XPS profiles of the as-electrodeposited palladium on FTO as a function of sputtering time (1 min/cycle) using Ar ion beam energy of 3 kV.



Figure S3. XRD patterns of the electrodeposited Pd before and after oxide reduction process for 3h.



Figure S4 XPS spectra of the as-prepared Pd after the oxide reduction process for 10min (a), and (b).



Figure S5. CVs of FTO with and without H_2 treatment at 100°C for 3 hours in 1M NaOH at a scan rate of 50 mV/s.



Figure S6. Top and side views of the atomic geometry of the most stable acetaldehyde (a, and b), and acetic acid (d and e) adsorption on the (a, d) p(2x2) Pd(111) unit cell, (b, e) PdO(101)/($\sqrt{5}x\sqrt{5}$)R27° Pd(100), and (c, f) (1x2) PdO(101).