Chemical looping hydrogen storage and production: Use of binary ferrite-spinel as oxygen carrier materials

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11 Figure S1. TEM images of the fresh materials (a) $Co_{0.25}Fe_{2.75}O_4$, (b) $Zn_{0.25}Fe_{2.75}O_4$, (c)

12 $Cu_{0.25}Fe_{2.75}O_4$, (d) $Ni_{0.25}Fe_{2.75}O_4$ and (e) $Mn_{0.25}Fe_{2.75}O_4$.





$$Cu_{0.25}Fe_{2.75}O_4$$
, (d) $Mn_{0.25}Fe_{2.75}O_4$ and (e) $Ni_{0.25}Fe_{2.75}O_4$.





18 Figure S3. XPS results of the re-oxidized materials. (a) $Co_{0.25}Fe_{2.75}O_4$, (b) $Zn_{0.25}Fe_{2.75}O_4$, (c) $Cu_{0.25}Fe_{2.75}O_4$,

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(d) $Mn_{0.25}Fe_{2.75}O_4$ and (e) $Ni_{0.25}Fe_{2.75}O_4$.



Figure S4. TEM images of the materials over 10 cycles (a) $Co_{0.25}Fe_{2.75}O_4$, (b) $Zn_{0.25}Fe_{2.75}O_4$, (c)

 $Cu_{0.25}Fe_{2.75}O_4$, (d) $Ni_{0.25}Fe_{2.75}O_4$ and (e) $Mn_{0.25}Fe_{2.75}O_4$.