

## Electronic supplementary material

# Hydrothermal *in-situ* construction of AgVO<sub>3</sub>/LaVO<sub>4</sub> phase junctions for the efficient visible-light-driven pollutants disposal and photoelectrocatalytic methanol oxidation

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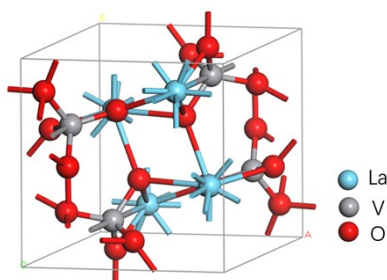
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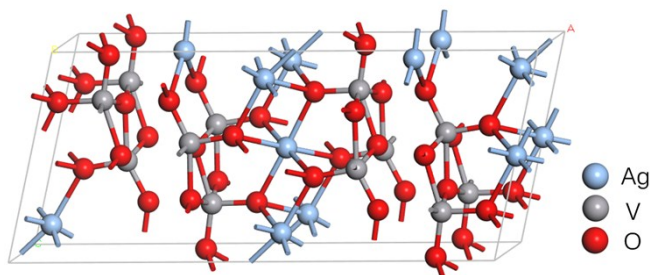
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As known,  $\text{LaVO}_4$  crystalline has two polymorphs, that is, tetragonal phase (t) with zircon structure and monoclinic phase (m) with monazite structure.<sup>1</sup> m- $\text{LaVO}_4$  has drawn considerable interest because of its surface catalytic properties, and the absorption of visible light. Herein, m- $\text{LaVO}_4$  was prepared and applied. m- $\text{LaVO}_4$  belongs to the monoclinic structure with group of  $P21/n$ (NO. 14). The lattice parameters of m- $\text{LaVO}_4$  are  $a=7.139$ ,  $b=7.353$ ,  $c=6.804$ ,  $\beta= 105.306^\circ$ . Every V atom lies in the center of tetrahedron composed by 4 O atoms, every La atom connects with O atoms in 9 tetrahedral  $\text{VO}_4$ , as shown as Fig. S1.



**Fig. S1** The structure of m- $\text{LaVO}_4$

$\text{AgVO}_3$  can be crystallized in four different structures.  $\alpha$ - $\text{AgVO}_3$ ,  $\gamma$ - $\text{AgVO}_3$  and  $\delta$ - $\text{AgVO}_3$  are metastable and  $\beta$ - $\text{AgVO}_3$  is the only one of thermodynamically stable.  $\alpha$ - $\text{AgVO}_3$  and  $\beta$ - $\text{AgVO}_3$  show one-dimensional morphologies, and they can be found in the form of nanowires, nanorods or nanotubes. The  $\alpha$ - $\text{AgVO}_3$  phase can be irreversibly transformed into  $\beta$ - $\text{AgVO}_3$  at temperatures up to  $200^\circ\text{C}$ .<sup>2</sup>  $\beta$ - $\text{AgVO}_3$  was obtained in this work and it had monoclinic crystalline system, space group of  $C_m$ , and crystallographic parameters are  $a = 18.106$ ,  $b = 3.579$ ,  $c = 8.043 \text{ \AA}$  and  $\beta = 104.44^\circ$ ,<sup>3</sup> as shown as Fig. S2.



**Fig. S2** The structure of  $\beta$ - $\text{AgVO}_3$ .

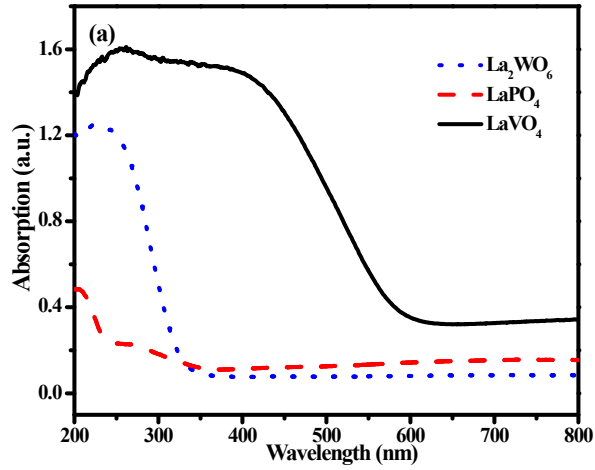


Fig. S3 UV-Vis DRS Spectra of different lanthanum salts.

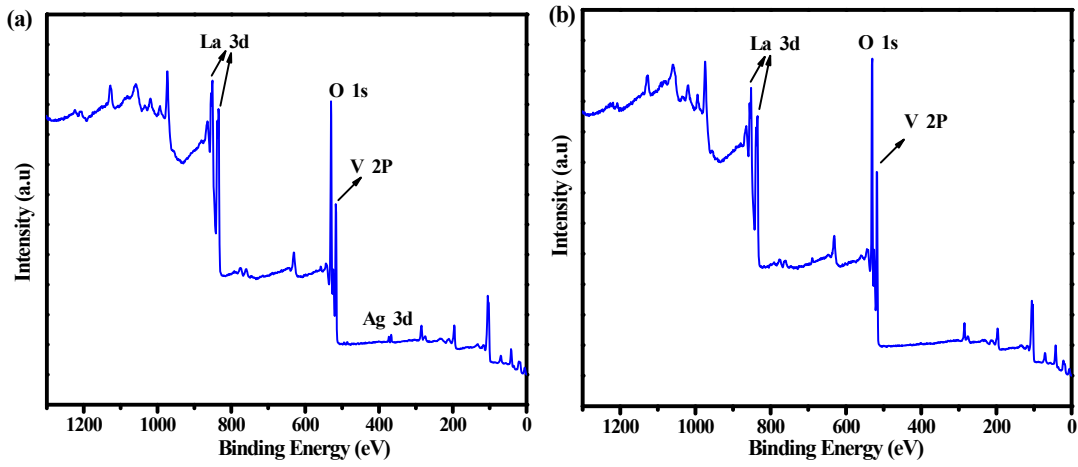


Fig.S4 XPS spectra of LaVO<sub>4</sub> and AgVO<sub>3</sub>/LaVO<sub>4</sub>, (a) survey spectrum of AgVO<sub>3</sub>/LaVO<sub>4</sub>; (b) survey spectrum of pure LaVO<sub>4</sub>.

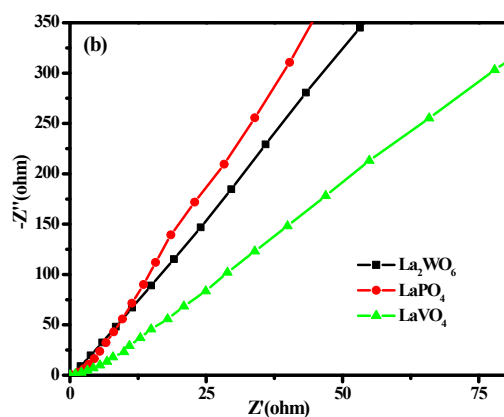


Fig. S5 Electrochemical impedance spectroscopy (EIS) Nyquist plots of different lanthanum salts.

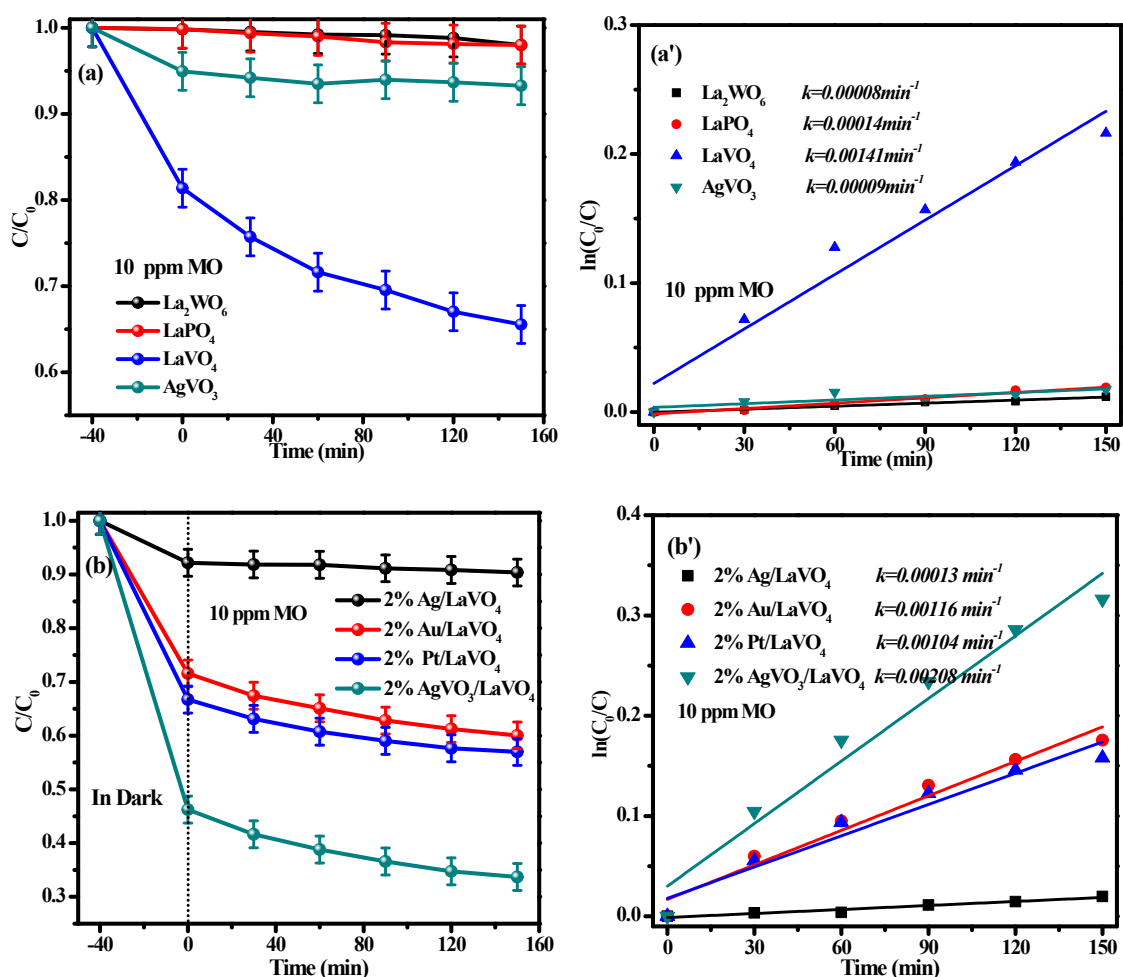


Fig. S6 The degradation and kinetic plots of MO and AO solution with different samples under the visible light irradiation. a and a' correspond to MO degradation of  $\text{La}_2\text{WO}_6$ ,  $\text{LaPO}_4$ ,  $\text{LaVO}_4$  and  $\text{AgVO}_3$ ; b and b' MO degradation of  $\text{Au/LaVO}_4$ ,  $\text{Ag/LaVO}_4$  and  $\text{Pt/LaVO}_4$  catalysts.

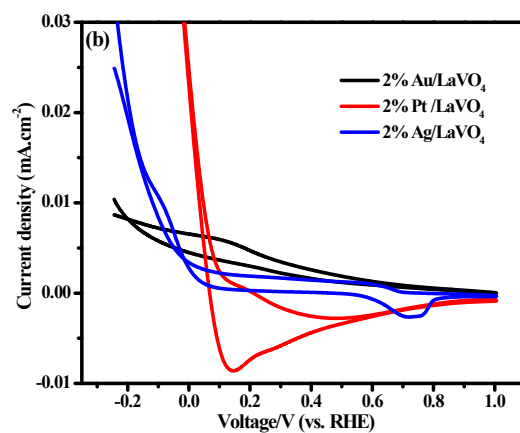


Fig. S7 CV curves over LaVO<sub>4</sub> catalysts supported by Au, Pt And Ag nanoparticles under visible light irradiation.

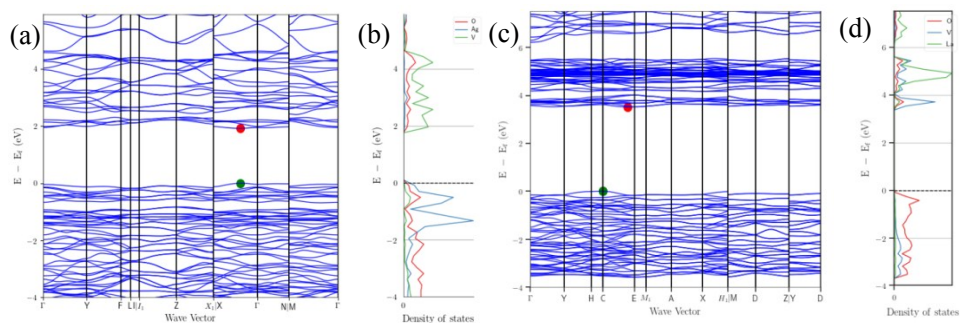


Fig.S8 (a) The projected band structures and (b) PDOS of AgVO<sub>3</sub>, (c) the projected band structures and (d) PDOS of LaVO<sub>4</sub>.

## References

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