## **Electronic Supplementary Information**

## Catalytic conversion of methyl oleate to hydrocarbons: Impact of cobalt oxide species integration in SiO<sub>2</sub>-Al<sub>2</sub>O<sub>3</sub>

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Note: This supplementary information contains supplementary figures S1-S13 and supplementary Table S1 and S2

**Fig. S1** EDS mapping of Co, O, Si, Al and EDS spectra of 6% cobalt oxide/ SiO<sub>2</sub>-Al<sub>2</sub>O<sub>3</sub> catalyst.



Fig. S2 EDS mapping of Co, O, Si, Al and EDS spectra of 8% cobalt oxide/  $SiO_2$ -Al<sub>2</sub>O<sub>3</sub> catalyst.



Fig.S3 EDS mapping of Co, O, Si, Al and EDS spectra of 10% cobalt oxide/  $SiO_2\text{-}Al_2O_3$  catalyst



Fig. S4 EDS mapping of Co, O, Si, Al and EDS spectra of 12% cobalt oxide/ SiO<sub>2</sub>-Al<sub>2</sub>O<sub>3</sub> catalyst



Fig. S5 Pore size distribution of cobalt oxide/ SiO<sub>2</sub>-Al<sub>2</sub>O<sub>3</sub> catalyst



Fig. S6 N<sub>2</sub> adsorption/desorption isotherms and BET adsorption plot of the catalysts



Fig. S7 XPS survey spectra of 6% cobalt oxide/ SiO<sub>2</sub>-Al<sub>2</sub>O<sub>3</sub> catalyst



Fig.S8 Ammonia-TPD graphs of cobalt oxide/ SiO<sub>2</sub>-Al<sub>2</sub>O<sub>3</sub> catalyst



Fig. S9 Solid-state NMR spectra of cobalt oxide/SiO<sub>2</sub>-Al<sub>2</sub>O<sub>3</sub> catalyst



Fig.S10 FT-IR spectra of cobalt oxide/SiO<sub>2</sub>-Al<sub>2</sub>O<sub>3</sub> catalyst.



Fig. S11 XRD Spectra of 6% cobalt oxide/ SiO<sub>2</sub>-Al<sub>2</sub>O<sub>3</sub> spent catalyst for one, five and ten cycles



**Fig. S12.** N<sub>2</sub> adsorption/desorption isotherms of 6% cobalt oxide/ SiO<sub>2</sub>-Al<sub>2</sub>O<sub>3</sub> spent catalyst for one, five and ten cycles.





Fig. S13. The Survey and core XPS spectra of spent catalyst

Table S1

Table S1. Reusability test results of 6% Co SiO<sub>2</sub>-Al<sub>2</sub>O<sub>3</sub> catalyst

No	Cycles	Conversion	Selectivity (%)		
			n-C-17	n-C-18	
1	1	100	45	47	
2	3	99	45	45	
3	5	97	43	42	
4	10	94	41	40	

Reactions were performed with methyl oleate (250  $\mu$ L) and catalyst (25 mg) at 10-40 bar H<sub>2</sub> pressure

## Table S2

**Table S2.** Selectivity of lower hydrocarbons from the hydrotreatment reaction of methyl oleatewith prepared  $SiO_2$ -Al<sub>2</sub>O<sub>3</sub> catalyst.

No	Catalyst	Experiment	Selectivity							
			<c-10< th=""><th>C-10</th><th>C-11</th><th>C-12</th><th>C-13</th><th>C-14</th><th>C-15</th><th>C-16</th></c-10<>	C-10	C-11	C-12	C-13	C-14	C-15	C-16
1	6%Co/SiO <sub>2</sub> -Al <sub>2</sub> O <sub>3</sub>	280°C/40 bar/10hrs	-	2	13	16	17	17	14	10
2	6%Co/SiO <sub>2</sub> -Al <sub>2</sub> O <sub>3</sub>	280°C/30 bar/10hrs	-	0.5	0.5	2	1	1	2	5
3	6%Co/SiO <sub>2</sub> -Al <sub>2</sub> O <sub>3</sub>	280°C/20 bar/10hrs	-	-	-	-	3	1	2	8
4	4%Co/SiO <sub>2</sub> -Al <sub>2</sub> O <sub>3</sub>	280°C/20 bar/10hrs	66	31	3	-	-	-	-	-
5	SiO <sub>2</sub> -Al <sub>2</sub> O <sub>3</sub>	280°C/15 bar/10hrs	-	-	-	2	12	16	13	-